

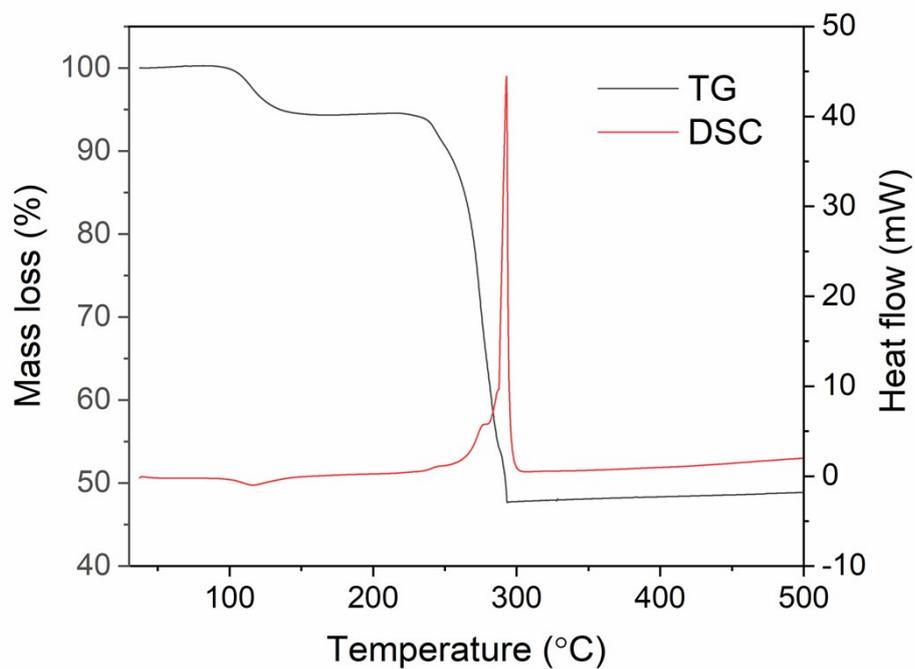
# **Argentophilicity Induced Anomalous Thermal Expansion Behavior in 2D Silver Squarate**

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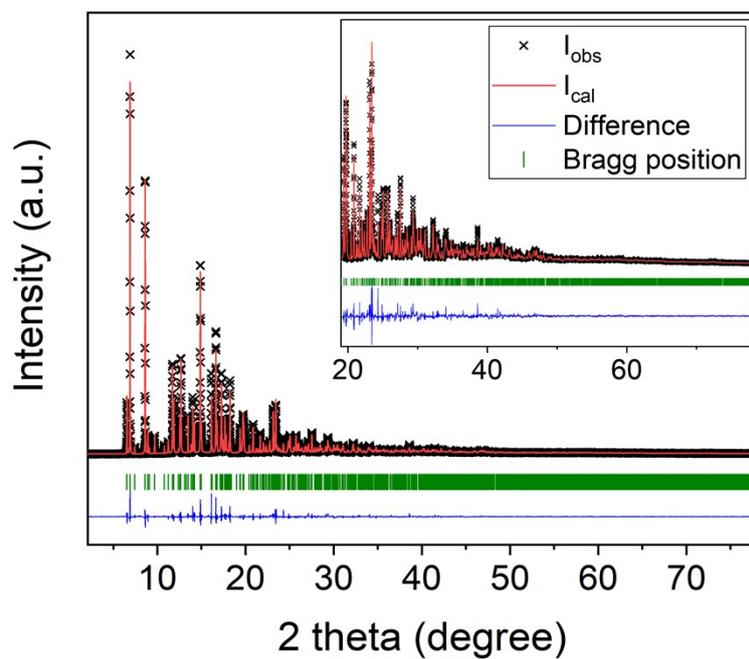
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**Table S1** Crystallographic data of compound Ag\_sq

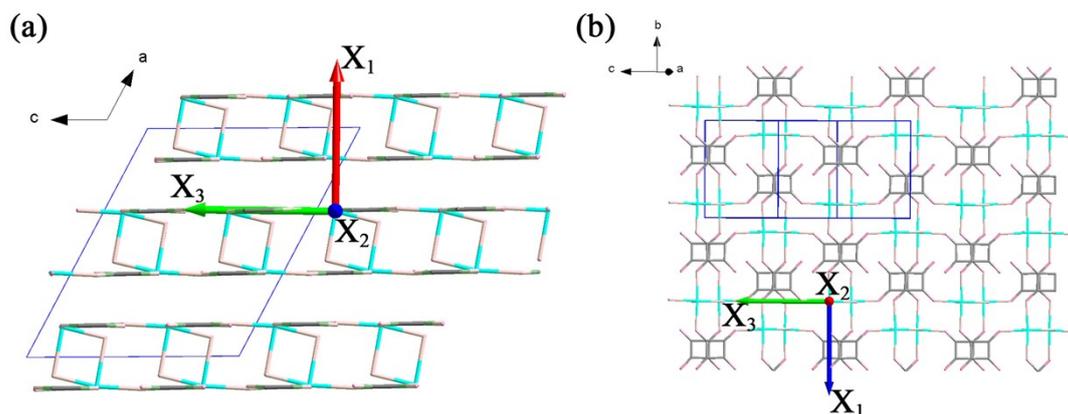
Name	Ag_sq
Empirical formula	C <sub>4</sub> H <sub>3</sub> O <sub>5</sub> Ag
Formula weight	238.93
Crystal system	Monoclinic
Space group	<i>C2/c</i>
a/Å	13.358(3)
b/Å	8.2296(9)
c/Å	11.1054(16)
α/°	90
β/°	118.508(17)
γ/°	90
Volume/ Å <sup>3</sup>	1072.8(4)
Z	8
ρ <sub>cal</sub> g/cm <sup>3</sup>	2.934
2θ range	6.53~70.1
Independent reflections	2016
Goodness-of-fit on F <sup>2</sup>	1.267
Final R indexes [I ≥ 2σ(I)]	R <sub>1</sub> =0.1023 wR <sub>2</sub> =0.2693
Final R indexes [all data]	R <sub>1</sub> =0.1059 wR <sub>2</sub> =0.2782
CCDC	2014485



**Figure S1** TG-DSC curves of Ag\_sq in air.



**Figure S2** Rietveld refinement of the high resolution synchrotron X-ray powder diffraction pattern of Ag\_sq at 300K ( $R_p = 5.05\%$ ,  $wR_p = 7.52\%$ )



**Figure S3** Illustration of the three principal axes

**Table S2** Some representative 2D materials with NTE.

Name	CTE (MK <sup>-1</sup> )	Temperature range (K)	Reference
Ag_sq	-6.1(2)	100~300	This work
Graphene	-8.0	200~400	1
Silicene	-5.3	-	2
Germanene	-2.1	-	2
Blue phosphorene	-4.2	-	2
Boron nitride	-2.9 (at 293K)	0~770	3

**Table S3** Proton conductivities of some representative compounds

Name	Conductivity (S cm <sup>-1</sup> )	Condition	References
Ag_sq	$1.4 \times 10^{-4}$	80°C, 99% R. H.	This work
Ferrous Oxalate Dihydrate	$1.3 \times 10^{-4}$	25°C, 98% R. H.	4
(N <sub>2</sub> H <sub>5</sub> )[CeEu(C <sub>2</sub> O <sub>4</sub> ) <sub>4</sub> (N <sub>2</sub> H <sub>5</sub> )]·4H <sub>2</sub> O	$3.42 \times 10^{-4}$	RT, 98% R. H.	5
Cu <sub>4</sub> (L) <sub>2</sub> (OH) <sub>2</sub> (DMF) <sub>2</sub>			
L = 5-sulfoisophthalic acid	$7.4 \times 10^{-4}$	95°C, 95% R. H.	6
monosodium salt			
β-PCMOF2	$1.3 \times 10^{-3}$	85°C, 90% R. H.	7
PCMOF10	$3.55 \times 10^{-2}$	70°C, 95% R. H.	8

Ca-PiPhA-NH <sub>3</sub>	$6.6 \times 10^{-2}$	24°C, 98% R. H	9
LiCl@[Ca(C <sub>4</sub> O <sub>4</sub> )(H <sub>2</sub> O)]	$1.8 \times 10^{-2}$	25°C, 40% R. H	10

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