Confinement of PMo₁₂ in Hollow SiO₂-PMo₁₂@rGO Nanospheres for High Performance Lithium Storage

Hanbin Hu, Xueying Jia, Jiaxin Wang, Wei Chen,* Lei He* and Yu-Fei Song*

State Key Laboratory of Chemical Resource Engineering, Beijing Advanced Innovation Center for Soft Matter Science and Engineering, Beijing University of Chemical Technology, Beijing 100029, P. R. China. Email: songyf@mail.buct.edu.cn, helei@mail.buct.edu.cn, chenw@mail.buct.edu.cn; Tel/Fax: +86 10-64431832.

List of contents:

Fig. S1 SEM images (a) PMMA spheres and (b) PMMA@SiO₂ spheres.
Fig. S2 FT-IR spectra of PMMA, PMMA@SiO₂, SiO₂-NH₂, PMo₁₂ and SiO₂-PMo₁₂.
Fig. S3 (a) XRD patterns of SiO₂-PMo₁₂@rGO and SiO₂@rGO. (b) TG curves of SiO₂-PMo₁₂@rGO, rGO and PMo₁₂.

Fig. S4 High-resolution XPS spectrum of P 2p peak of the SiO₂-PMo₁₂@rGO
Fig. S5 High-resolution XPS spectra of Mo 3d and P 2p peaks of PMo₁₂.
Fig. S6 The charge-discharge profiles for different cycles of SiO₂-PMo₁₂@rGO under 100 mA g⁻¹.

Fig. S7 (a) The CV of the SiO₂@rGO; (b) The charge-discharge profiles for different cycles of SiO₂@rGO under 100 mA g⁻¹; (c) Rate performance of SiO₂@rGO electrode;
(d) Cycling performance and CE of SiO₂@rGO under 100 mA g⁻¹ within a voltage of 0.01-3.0 V.

Fig. S8 (a) The CV of the SiO₂-PMo₁₂. (b) The charge-discharge profiles for different cycles of SiO₂-PMo₁₂ under 100 mA g^{-1} . (c) Rate performance of SiO₂-PMo₁₂ electrode. (d) Cycling performance and CE of SiO₂-PMo₁₂ under 100 mA g^{-1} within a voltage of 0.01-3.0 V.

Fig. S9 TEM image of SiO₂-PMo₁₂@rGO after 100 cycles at 100 mA g⁻¹.

Table S1. Comparison of SiO_2 -PMo₁₂@rGO with other reported SiO₂- and POM-based electrodes as LIBs anode materials.



Fig. S1 SEM images of (a) PMMA spheres and (b) PMMA@SiO₂ spheres.



Fig. S2 FT-IR spectra of PMMA, PMMA@SiO₂, SiO₂-NH₂, PMo₁₂ and SiO₂-PMo₁₂, respectively.



Fig. S3 (a) XRD patterns of SiO_2 -PMo₁₂@rGO and SiO_2 @rGO. (b) TG curves of SiO_2 -PMo₁₂@rGO, rGO and PMo₁₂.



Fig. S4 High-resolution XPS spectrum of P 2p peak of the SiO₂-PMo₁₂@rGO.



Fig. S5 High-resolution XPS spectra of Mo 3d and P 2p peaks of PMo₁₂.



Fig. S6 The charge-discharge profiles for different cycles of SiO_2 -PMo₁₂@rGO under 100 mA g⁻¹.



Fig. S7 (a) The CV of the SiO₂@rGO; (b) The charge-discharge profiles for different cycles of SiO₂@rGO under 100 mA g^{-1} ; (c) Rate performance of SiO₂@rGO electrode; (d) Cycling performance and CE of SiO₂@rGO under 100 mA/g within a voltage of 0.01-3.0 V.



Fig. S8 (a) The CV of the SiO₂-PMo₁₂. (b) The charge-discharge profiles for different cycles of SiO₂-PMo₁₂ under 100 mA g^{-1} . (c) Rate performance of SiO₂-PMo₁₂ electrode. (d) Cycling performance and CE of SiO₂-PMo₁₂ under 100 mA g^{-1} within a voltage of 0.01-3.0 V.



Fig. S9 TEM image of SiO₂-PMo₁₂@rGO after 100 cycles at 100 mA g⁻¹.

Electrode materials	Current density (mA g ⁻¹)	Capacity (mA h g ⁻¹)	Cycles	Ref.
Hollow SiO ₂ @CN	100	800	100	1
C/SiO ₂ /C	300	300	70	2
SiO ₂ /C hollow spheres	100	624	100	3
H-SiO ₂ /C	200	564	400	4
CoW-POM-Cu foam	100	737	100	5
Py-Anderson-CNTs	0.5 mA cm ⁻²	665	100	6
PMo12-PPy/RGO	100	1000	50	7
Hollow SiO2-PMo12@rGO	100	720	100	This work

Table S1. Comparison of SiO_2 -PMo₁₂@rGO with other reported SiO₂- and POM-based electrodes as LIBs anode materials.

References

- 1. T.T. Xiao, W.F. Zhang, T. Xu, J.X. Wu and M.D. Wei, Hollow SiO₂ microspheres coated with nitrogen doped carbon layer as an anode for high performance lithium-ion batteries, *Electrochim. Acta*, 2019, **306**,106-112.
- Z.Q. Gu, X.H. Xia, C. Liu, X. Hu, Y.X. Chen, Z.Y. Wang and H.B. Liu, Yolk structure of porous C/SiO₂/C composites as anode for lithium-ion batteries with quickly activated SiO₂, *J. Alloy. Compd.*, 2018, **757**, 265-272.
- 3. X.L. Liu, Y.X. Chen, H.B. Liu and Z.Q. Liu, SiO₂@C hollow sphere anodes for lithium-ion batteries, *J. Mater. Sci. Technol.* 2017, **33**, 239-245.
- Y. Jiang, D.B. Mu, S. Chen, B.R. Wu, Z.K. Zhao, Y.Z. Wu, Z.P. Ding and F. Wu, Hollow silica spheres with facile carbon modification as an anode material for lithium-ion batteries, J. *Alloy. Compd.* 2018, **744**, 7-14.
- K. Sun, H.Q. Li, H.J. Ye, F.Q. Jiang, H. Zhu and J. Yin, 3D-Structured Polyoxometalate Microcrystals with Enhanced Rate Capability and Cycle Stability for Lithium-Ion Storage, *ACS Appl. Mater. Interfaces*, 2018, 10, 18657-18664.
- L. Huang, J. Hu, Y. Ji, C. Streb and Y.-F. Song, Pyrene-Anderson-modified CNTs as anode materials for lithium-ion batteries, *Chem. Eur. J.*, 2015, 21, 18799-18804.
- M. Zhang, T. Wei, A.M. Zhang, S.L. Li, F.C. Shen, L.Z. Dong, D.S. Li and Y.Q. Lan, Polyoxomolybdate-polypyrrole/reduced graphene oxide nanocomposite as high-capacity electrodes for lithium storage, *ACS Omega*, 2017, 2, 5684-5690.