

Construction of PdO-decorated double-shell ZnSnO₃ hollow microspheres for n-propanol detection at low temperature

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The n-propanol gas was derived by a static liquid-gas method.^{1,2} The relationship between the volume of n-propanol solution and the concentration of n-propanol gas could be calculated by the following formula:

$$C = [(22.4 \times d \times \rho \times V_1) / (M \times V_2)] \times 1000$$

Where C (ppm) is the target gas concentration, d is the purity of n-propanol solution, ρ (g/mL) is the density of n-propanol solution, V_1 (μ L) is the injection volume of n-propanol solution, V_2 (L) is the volume of the glass chamber and M (g/mol) is the molecular weight of n-propanol solution. The n-propanol solution is purchased from Aladdin Chemistry Co. Ltd. (Shanghai, China).

References

- [1] L. Li, P. F. Cheng, Y. L. Wang, L. P. Xu, B. Zhang, C. Lv, J. Ma, Y. Zhang, Sb-doped three-dimensional ZnFe₂O₄ macroporous spheres for N-butanol chemiresistive gas sensors, *Sensors. Actuators B: Chem.*, 2020, **320**, 128384.

[2] P. Sun, W. N. Wang, Y. P. Liu, Y. F. Sun, J. Ma, G. Y. Lu, Hydrothermal synthesis of 3D urchin-like α -Fe₂O₃ nanostructure for gas sensor, *Sensors. Actuators B: Chem.*, 2012, **173**, 52-57.

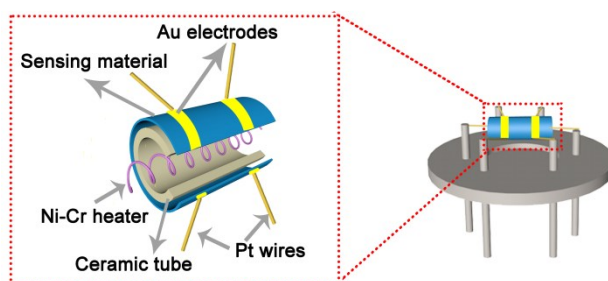


Fig. S1. Schematic structure of the overall gas sensor.

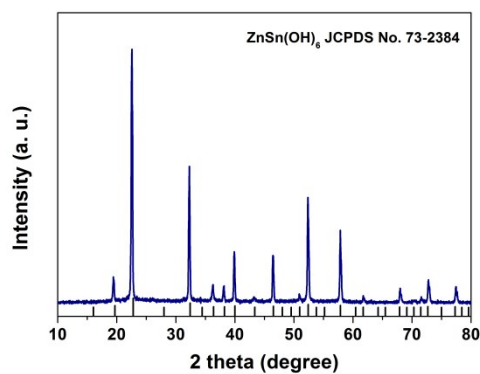


Fig. S2. The XRD pattern of ZnSn(OH)₆ precursor.

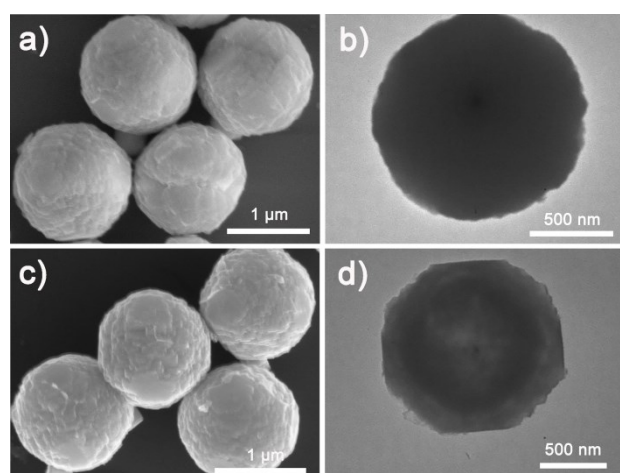


Fig. S3. The FESEM images of (a) ZnSn(OH)_6 precursor and (c) ZnSnO_3 ; the TEM images of (b) ZnSn(OH)_6 precursor and (d) ZnSnO_3 .

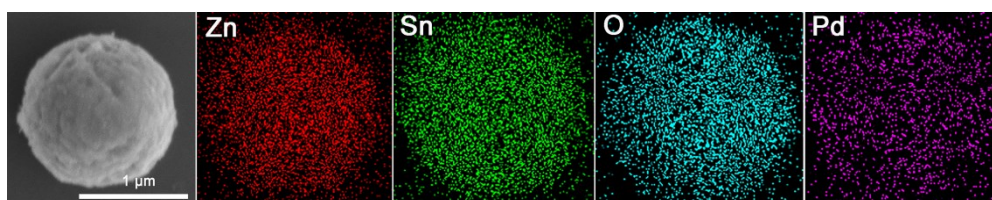


Fig. S4. The elemental mapping image of 4 wt% PdO-loaded ZnSnO_3 samples.

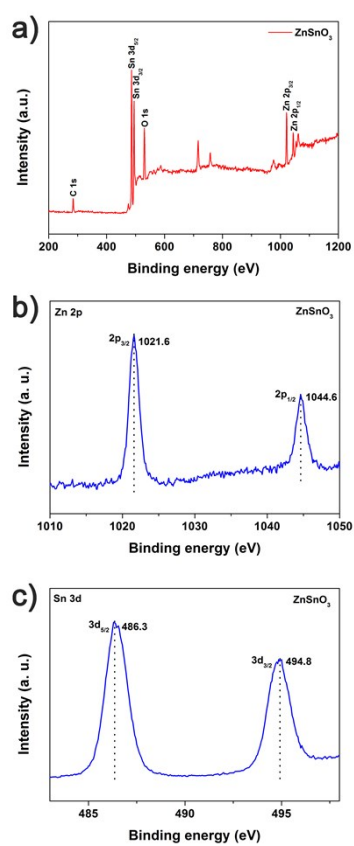


Fig. S5. The XPS spectrum of pure ZnSnO₃: (a) overall spectrum; (b) Zn 2p and (c) Sn 3d.

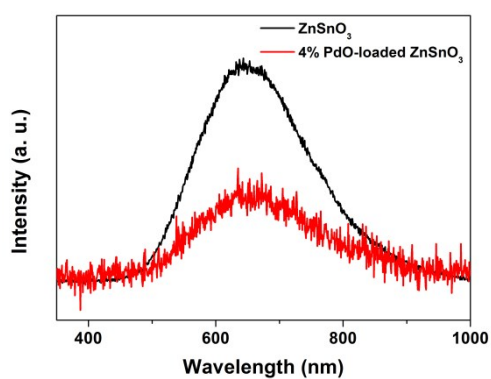


Fig. S6. The room-temperature PL spectrum of pure ZnSnO₃ and 4 wt% PdO-loaded ZnSnO₃ samples.

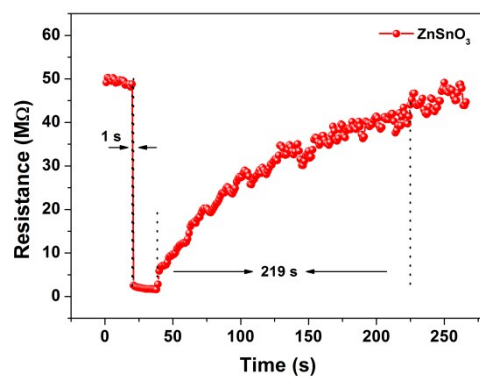


Fig. S7. The response/recovery speed of pure ZnSnO₃ to 100 ppm n-propanol at 200°C.

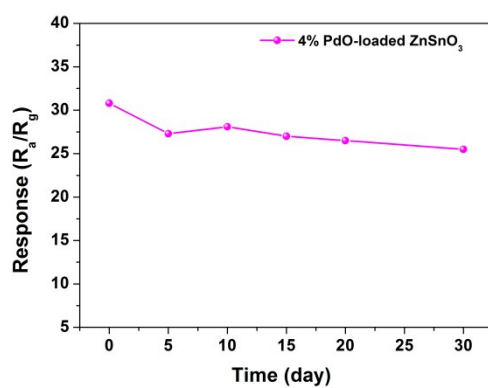


Fig. S8. The long-term stability of the sensor based on 4 wt% PdO-loaded ZnSnO₃ to 100 ppm n-propanol at 140°C.