

Hierarchical ultrathin defect-rich $\text{CoFe}_2\text{O}_4@\text{BC}$ nanoflowers synthesized via temperature-regulated strategy for an outstanding hydrogen evolution reaction

Bashir Adegbemiga Yusuf^a, Meng Xie^{b*}, Waleed Yaseen^a, Jimin Xie^a, Yuanguo Xu^{a*}

^a School of Chemistry and Chemical Engineering, Jiangsu University, Zhenjiang 212013, China

^b School of Pharmacy, Jiangsu University, Zhenjiang, 212013, PR Chin

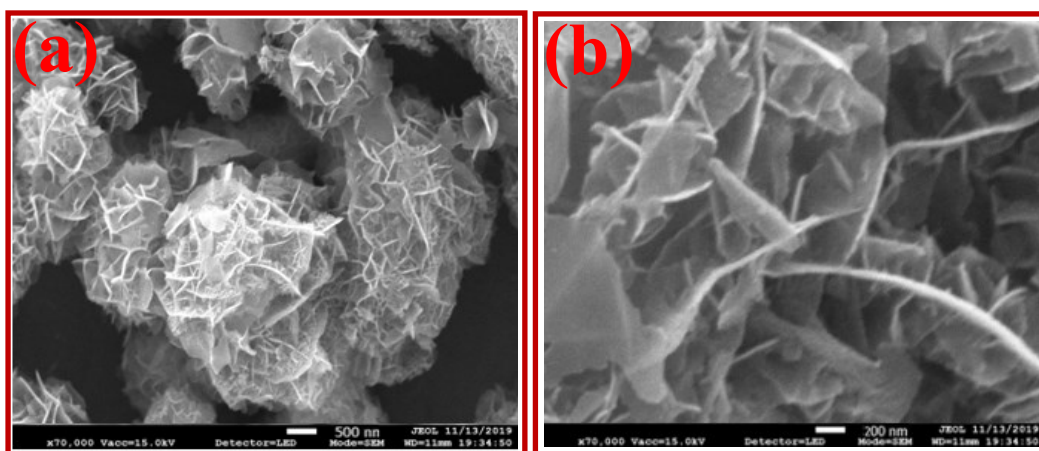
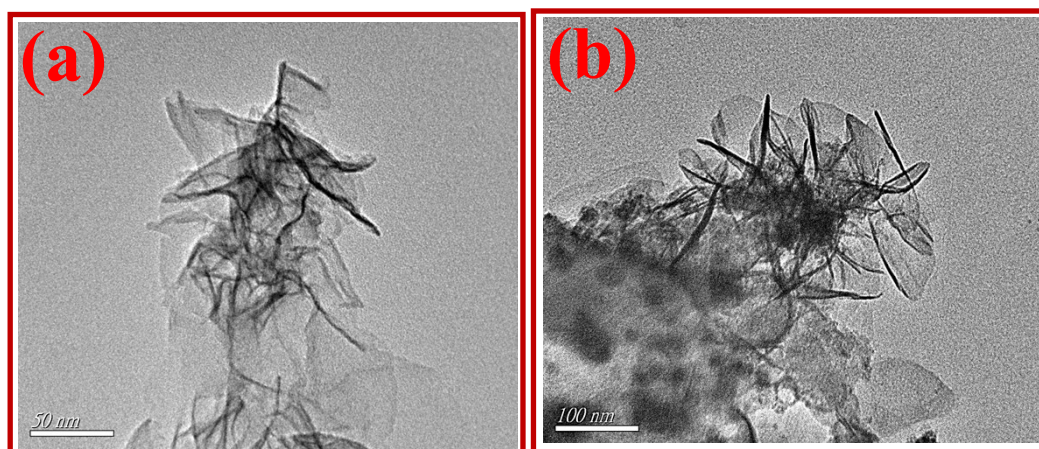


Fig S1. (a and b) SEM images of $\text{CoFe}_2\text{O}_4@\text{BC}$ 500°C



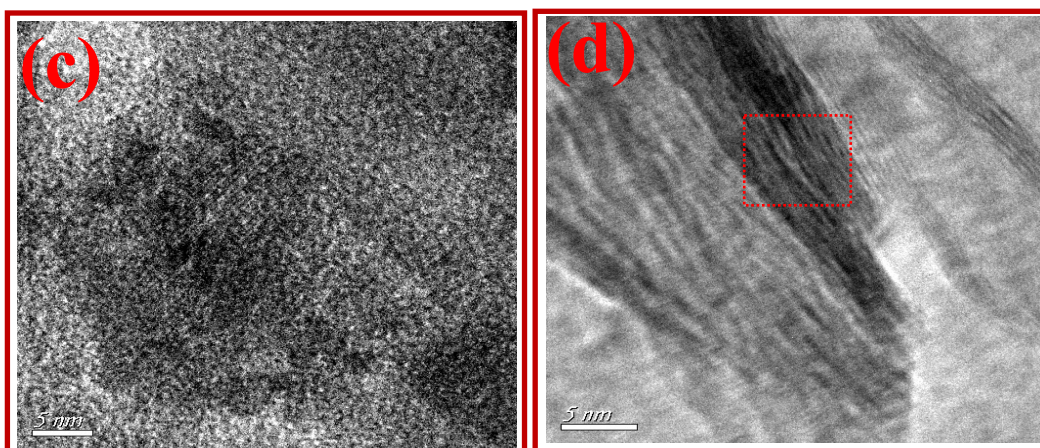


Fig. S2. (a-b) TEM images and (c and d) HR-TEM image of nanoflowers and the red region (*inset: carbon shell*) of the as-prepared $\text{CoFe}_2\text{O}_4@\text{BC}$ 500°C.

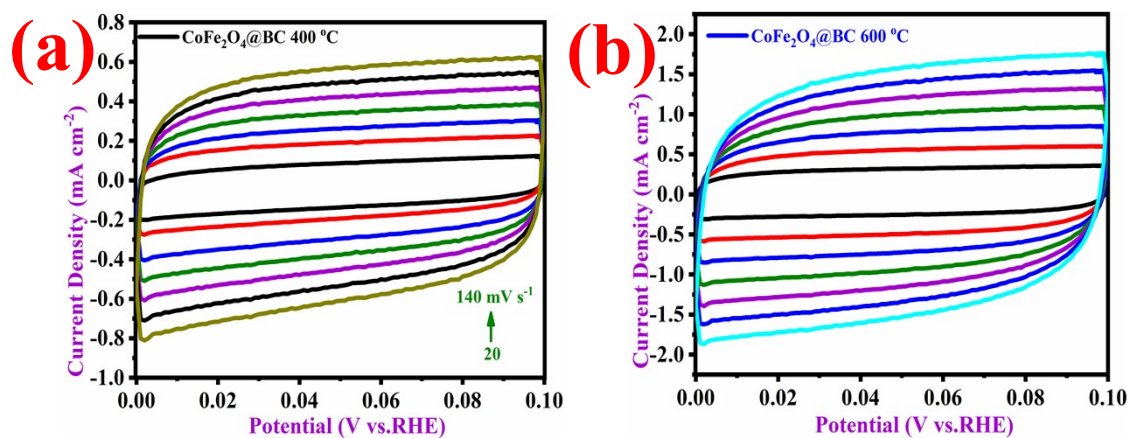


Fig S3. Cyclic voltammograms for (a) $\text{CoFe}_2\text{O}_4@\text{BC}$ 400°C and (b) $\text{CoFe}_2\text{O}_4@\text{BC}$ 600°C

Table S1. Overpotential at a current density 10 mA cm⁻² of the as-prepared CoFe₂O₄@BC 500°C and CoFe₂O₄@BC 600°C catalyst and other reported catalysts for HER.

Electrocatalyst	Electrolyte	η_{10} (mV)	Nanostructures	Synthetic routes	Ref
CoFe ₂ O ₄ @BC 500°C	1 M KOH	58	Nanoflowers	Chemical reduction and calcination	This work
CoFe ₂ O ₄ @BC 600°C		82			
NiCo ₂ S ₄ /CC	1 M KOH	263	Nanosheet	Hydrothermal treatment	[1]
Co-NRCNTs	1 M KOH	270	Carbon nanotube	Calcination process	[2]
NiCoP/NF	1 M KOH	350	Nanocube	Pyrolysis	[3]
Ni@B-C 500°C	1 M KOH	176	Nanosheet	Calcination treatment	[4]
CoS ₂ /CC	1 M KOH	462	Nanowire	Hydrothermal method	[5]
Ni ₂ / ₃ Fe ₁ / ₃ -rGO	1 M KOH	560	Nanosheet	Homogeneous precipitation method	[6]
Ni ₃ Fe@BC-500°C	1 M KOH	330	Nanoparticle	Chemical reduction and pyrolysis	[7]

Co-30Ni-B	1 M KOH	170	Nanoparticle	Chemical reduction treatment	[8]
Co-Ni-B@NF	1 M KOH	205	Nanoparticle	Electroless plating method	[9]
Co-B	1 M KOH	250	Nanoparticle	Chemical reduction method	[10]

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