

Electronic Supplementary Information (ESI)

Synthesis of carboxy-cyclobutane Isomers combining amide bond and self-assembly of coordination polymers in solid state : Controlling the reaction site of [2+ 2] cycloaddition by introducing a substituent group

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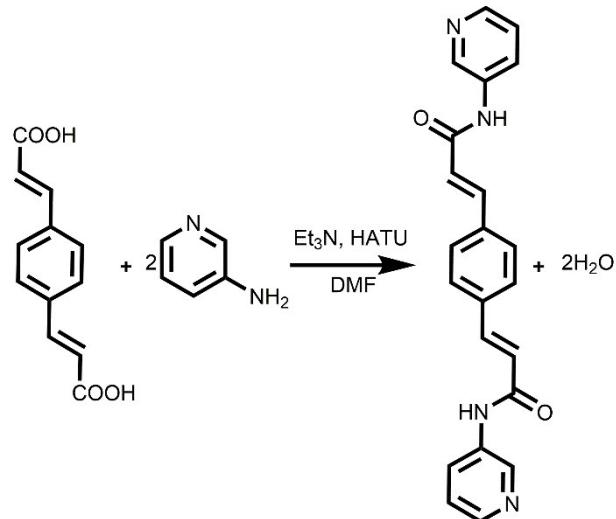
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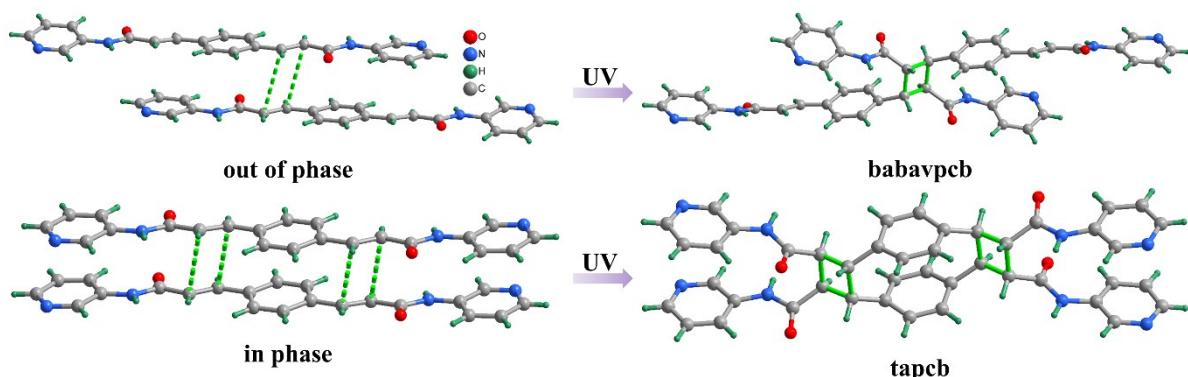
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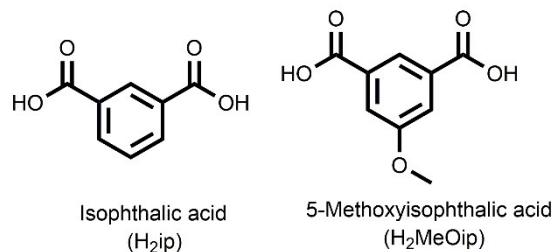
Experimental



Scheme S1 Synthesis of pbpa.



Scheme S2 Two different alignments of pbpa ligands, which result in the formation of responding photoproducts under UV irradiation.



Scheme S3 Structures of isophthalic acid and 5-Methoxyisophthalic acid, the carboxylic acids used in this work.

Table S1 Crystallographic data and structure refinement summary for **pbpa** and **1-2a**.

Compound	pbpa	1	1a	2	2a
CCDC number	2031703	2031704	2031705	2031706	2031707
Empirical formula	C ₂₂ H ₂₂ N ₄ O ₄	C ₆₀ H ₅₄ Cd ₂ N ₈ O ₁₇	C ₆₀ H ₅₄ Cd ₂ N ₈ O ₁₇	C ₃₁ H ₃₀ CdN ₄ O ₁₀	C ₃₁ H ₃₀ CdN ₄ O ₁₀
Formula weight	406.43	1383.91	1383.91	730.99	730.99
Temperature/K	110(2)	293.15	293(2)	100(2)	150.0
Crystal system	Monoclinic	Triclinic	Triclinic	Triclinic	Triclinic
Space group	P ₂ 1/n	P ₁	P ₁	P ₁	P ₁
a/Å	4.6257(3)	10.1695(13)	10.310(3)	10.3164(3)	10.4338(3)
b/Å	22.0943(15)	10.4856(14)	11.101(3)	11.5498(3)	11.3666(3)
c/Å	10.1647(6)	14.2567(18)	13.809(3)	13.9452(4)	13.9471(4)
α/°	90	87.806(2)	106.513(3)	113.1120(10)	113.5200(10)
β/°	102.715(2)	73.590(2)	90.714(3)	92.1050(10)	92.3600(10)
γ/°	90	85.321(2)	95.475(3)	96.5960(1)	94.6500(10)
Volume/Å ³	1013.37(11)	1453.3(3)	1507.1(7)	1512.03(7)	1506.77(7)
Z	2	1	1	2	2
ρ _{calcg/cm³}	1.332	1.581	1.525	1.606	1.611
μ/mm ⁻¹	0.094	0.811	0.783	0.788	0.791
F(000)	428.0	702.0	702.0	744.0	744.0
Crystal size/mm ³	0.19×0.15×0.12	0.20×0.16×0.12	0.26×0.20×0.16	0.17×0.14×0.11	0.15×0.12×0.09
Crystal shape	Needle	Block	Block	Block	Block
Radiation	MoKα (λ = 0.71073)	MoKα (λ = 0.71073)	MoKα (λ = 0.71073)	MoKα (λ = 0.71073)	MoKα (λ = 0.71073)
2θ range for data collection/°	4.502–55.032	4.868–56.844	3.078–56.278	5.156–55.096	4.862–52.828
Index ranges	-5 ≤ h ≤ 6, -28 ≤ k ≤ 28, -12 ≤ l ≤ 13	-13 ≤ h ≤ 13, -13 ≤ k ≤ 13, -19 ≤ l ≤ 18	-13 ≤ h ≤ 13, -14 ≤ k ≤ 14, -18 ≤ l ≤ 18	-13 ≤ h ≤ 13, -14 ≤ k ≤ 15, -18 ≤ l ≤ 18	-12 ≤ h ≤ 13, -13 ≤ k ≤ 14, -17 ≤ l ≤ 17
Reflections collected	8162	13886	14241	18512	17422
Independent reflections	2245 [R _{int} = 0.0516, R _{sigma} = 0.0529]	6667 [R _{int} = 0.0463, R _{sigma} = 0.0828]	6817 [R _{int} = 0.0658, R _{sigma} = 0.1198]	6881 [R _{int} = 0.0390, R _{sigma} = 0.0479]	6147 [R _{int} = 0.0477, R _{sigma} = 0.0595]
Data/restrains/parameters	2245/0/197	6667/1/403	6817/0/509	6881/0/569	6147/18/433
Goodness-of-fit on F ²	1.071	0.976	0.968	1.075	1.136
Final R indexes [I>=2σ(I)]	R ₁ = 0.0474, wR ₂ = 0.0914	R ₁ = 0.0464, wR ₂ = 0.0888	R ₁ = 0.0584, wR ₂ = 0.1267	R ₁ = 0.0400, wR ₂ = 0.0916	R ₁ = 0.0514, wR ₂ = 0.1232
Final R indexes [all data]	R ₁ = 0.0814, wR ₂ = 0.1086	R ₁ = 0.0772, wR ₂ = 0.1024	R ₁ = 0.1327, wR ₂ = 0.1679	R ₁ = 0.0441, wR ₂ = 0.0941	R ₁ = 0.0724, wR ₂ = 0.1368
Largest diff. peak/hole / e·Å ⁻³	0.18/-0.23	0.98/-0.89	0.72/-1.07	1.06/-0.93	1.80/-0.86

Table S2 Selected Bond Distances (\AA) and Angles (deg) for **1-2a**.

1			
Cd1—O3 ⁱ	2.225 (3)	O3 ⁱ —Cd1—O1	96.33 (10)
Cd1—N4 ⁱⁱ	2.280 (3)	N4 ⁱⁱ —Cd1—O1	101.12 (10)
Cd1—N1	2.295 (3)	N1—Cd1—O1	94.41 (10)
Cd1—O1	2.340 (3)	O3 ⁱ —Cd1—O2	151.19 (10)
Cd1—O2	2.427 (2)	N4 ⁱⁱ —Cd1—O2	90.60 (9)
Cd1—O3W	2.496 (6)	N1—Cd1—O2	89.93 (10)
Cd1···O3W	2.566 (6)	O1—Cd1—O2	54.87 (9)
O3—Cd1 ⁱⁱⁱ	2.225 (3)	O3 ⁱ —Cd1—O3W	126.50 (13)
N4—Cd1 ^{iv}	2.280 (3)	N4 ⁱⁱ —Cd1—O3W	80.79 (17)
O3 ⁱ —Cd1—N4 ⁱⁱ	95.25 (10)	N1—Cd1—O3W	80.81 (17)
O3 ⁱ —Cd1—N1	93.24 (10)	O1—Cd1—O3W	136.99 (12)
N4 ⁱⁱ —Cd1—N1	161.34 (11)	O2—Cd1—O3W	82.27 (13)
1a			
Cd1—O3 ⁱ	2.239 (4)	O3 ⁱ —Cd1—O1	89.38 (14)
Cd1—N1	2.294 (4)	N1—Cd1—O1	93.41 (14)
Cd1—N4 ⁱⁱ	2.300 (4)	N4 ⁱⁱ —Cd1—O1	101.74 (15)
Cd1—O1	2.341 (4)	O3 ⁱ —Cd1—O2	144.32 (16)
Cd1—O2	2.416 (3)	N1—Cd1—O2	89.06 (14)
Cd1—O3W	2.454 (8)	N4 ⁱⁱ —Cd1—O2	89.66 (14)
Cd1···O3W	2.746 (1)	O1—Cd1—O2	54.99 (13)
O3—Cd1 ⁱⁱⁱ	2.239 (4)	O3 ⁱ —Cd1—O3W	110.1 (2)
N4—Cd1 ^{iv}	2.300 (4)	N1—Cd1—O3W	82.7 (3)
O3 ⁱ —Cd1—N1	95.65 (14)	N4 ⁱⁱ —Cd1—O3W	78.9 (3)
O3 ⁱ —Cd1—N4 ⁱⁱ	96.70 (15)	O1—Cd1—O3W	160.39 (19)
N1—Cd1—N4 ⁱⁱ	160.49 (16)	O2—Cd1—O3W	105.6 (2)
2			
Cd1—N1	2.274 (3)	N1—Cd1—O2 ⁱⁱ	90.58 (9)
Cd1—N4 ⁱ	2.278 (3)	N4 ⁱ —Cd1—O2 ⁱⁱ	87.59 (9)
Cd1—O1	2.309 (2)	O1—Cd1—O2 ⁱⁱ	134.48 (8)
Cd1—O2 ⁱⁱ	2.326 (2)	N1—Cd1—O3 ⁱⁱⁱ	95.19 (9)
Cd1—O3 ⁱⁱⁱ	2.415 (2)	N4 ⁱ —Cd1—O3 ⁱⁱⁱ	91.29 (9)
Cd1—O4 ⁱⁱⁱ	2.419 (2)	O1—Cd1—O3 ⁱⁱⁱ	138.97 (8)
O2—Cd1 ⁱⁱ	2.326 (2)	O2 ⁱⁱ —Cd1—O3 ⁱⁱⁱ	86.51 (8)
O4—Cd1 ^{iv}	2.419 (2)	N1—Cd1—O4 ⁱⁱⁱ	93.64 (9)
O3—Cd1 ^{iv}	2.415 (2)	N4 ⁱ —Cd1—O4 ⁱⁱⁱ	91.98 (9)
N1—Cd1—N4 ⁱ	173.14 (9)	O1—Cd1—O4 ⁱⁱⁱ	85.07 (8)
N1—Cd1—O1	87.28 (9)	O2 ⁱⁱ —Cd1—O4 ⁱⁱⁱ	140.41 (8)
N4 ⁱ —Cd1—O1	89.29 (9)	O3 ⁱⁱⁱ —Cd1—O4 ⁱⁱⁱ	53.90 (7)
2a			

Cd1—O1	2.272 (3)	O1—Cd1—C30 ⁱⁱ	118.01 (13)
Cd1—O2 ⁱ	2.325 (3)	O2 ⁱ —Cd1—O4 ⁱⁱ	84.93 (12)
Cd1—O4 ⁱⁱ	2.398 (3)	O2 ⁱ —Cd1—O3 ⁱⁱ	138.86 (11)
Cd1—O3 ⁱⁱ	2.414 (4)	O2 ⁱ —Cd1—C30 ⁱⁱ	111.84 (13)
Cd1—N4 ⁱⁱⁱ	2.288 (4)	O4 ⁱⁱ —Cd1—O3 ⁱⁱ	54.15 (11)
Cd1—N1	2.288 (4)	O4 ⁱⁱ —Cd1—C30 ⁱⁱ	26.98 (12)
O2—Cd1 ⁱ	2.325 (3)	O3 ⁱⁱ —Cd1—C30 ⁱⁱ	27.18 (12)
O4—Cd1 ^{iv}	2.398 (3)	N4 ⁱⁱⁱ —Cd1—O2 ⁱ	85.39 (13)
O3—Cd1 ^{iv}	2.414 (4)	N4 ⁱⁱⁱ —Cd1—O4 ⁱⁱ	90.37 (12)
N4—Cd1 ^v	2.288 (4)	N4 ⁱⁱⁱ —Cd1—O3 ⁱⁱ	90.28 (13)
O1—Cd1—O2 ⁱ	130.08 (12)	N4 ⁱⁱⁱ —Cd1—N1	173.76 (14)
O1—Cd1—O4 ⁱⁱ	144.98 (12)	N1—Cd1—O2 ⁱ	89.54 (13)
O1—Cd1—O3 ⁱⁱ	90.86 (12)	N1—Cd1—O4 ⁱⁱ	92.80 (13)
O1—Cd1—N4 ⁱⁱⁱ	91.44 (12)	N1—Cd1—O3 ⁱⁱ	95.93 (14)
O1—Cd1—N1	89.04 (13)		

Symmetry codes, for 1: (i) $x, y-1, z$; (ii) $x-1, y-1, z-1$; (iii) $x, y+1, z$; (iv) $x+1, y+1, z+1$; for 1a: (i) $x-1, y, z$; (ii) $x+1, y-1, z+1$; (iii) $x+1, y, z$; (iv) $x-1, y+1, z-1$; (v) $-x-1, -y+3, -z$; for 2: (i) $x+1, y+2, z+1$; (ii) $-x+1, -y+2, -z+2$; (iii) $x-1, y, z$; (iv) $x+1, y, z$; (v) $x-1, y-2, z-1$; for 2a: (i) $-x+1, -y, -z$; (ii) $x+1, y, z$; (iii) $x-1, y-2, z-1$; (iv) $x-1, y, z$; (v) $x+1, y+2, z+1$; (vi) $-x+2, -y+2, -z+1$; (vii) $-x+1, -y+1, -z$.

Results and Discussion

Crystal structures

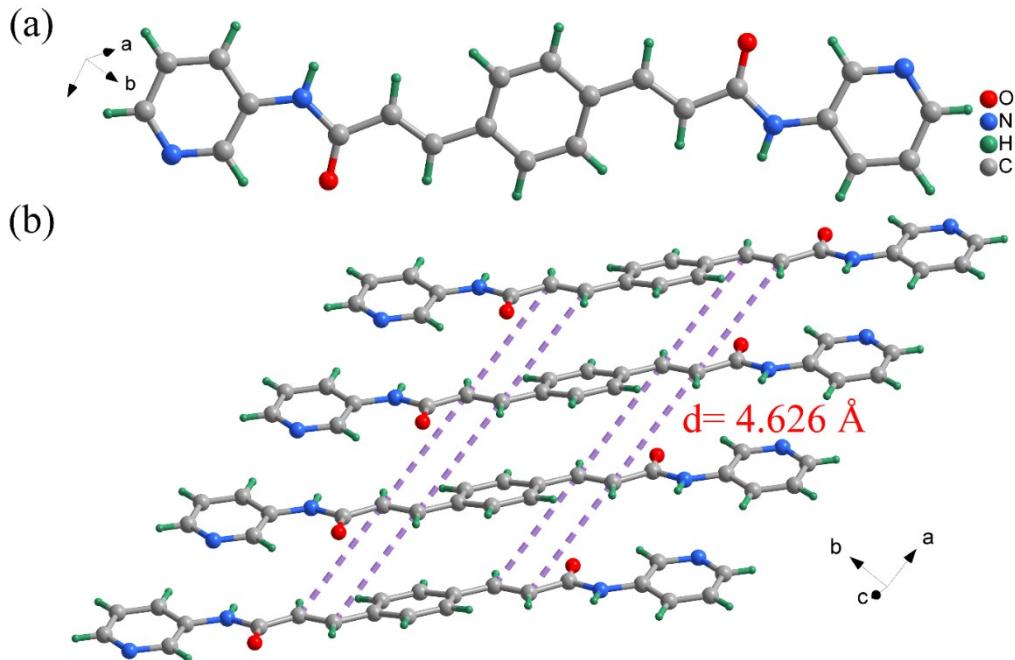


Fig. S1 (a) Crystal structure of pbpa. (b) Distance between adjacent olefins in pbpa crystals. The non-coordinated solvent molecules are omitted.

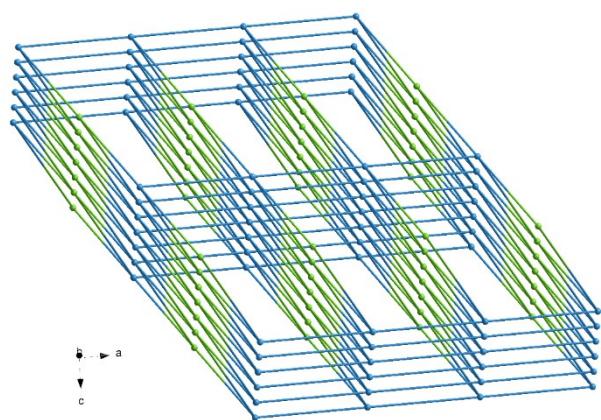


Fig. S2 Binoda fsc topology with point symbol of $\{4^4\cdot6^{10}\cdot8\}\{4^4\cdot6^2\}$ in **1a**.

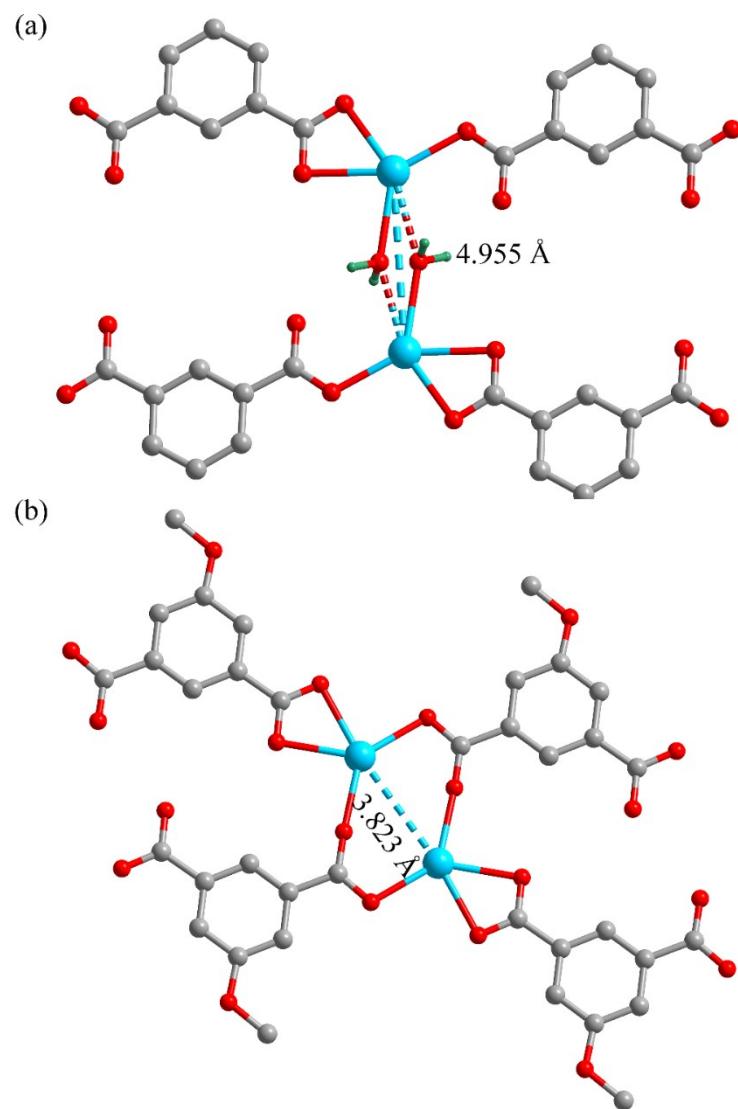


Fig. S3 Cd···Cd distance and component of the 1D carboxyl chain in **1** (a) and **2** (b).

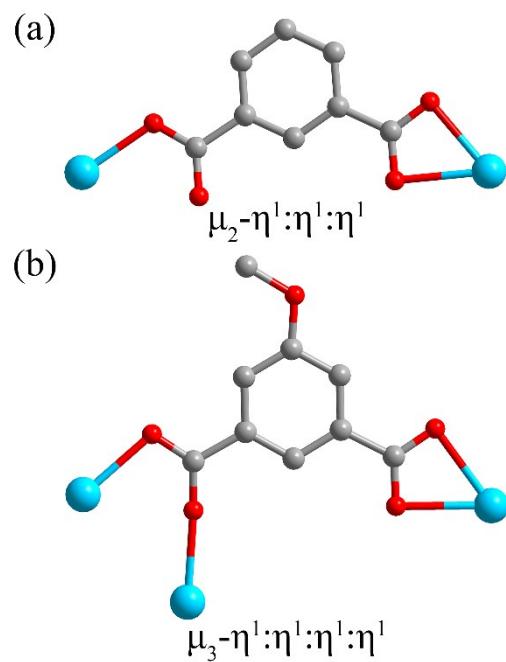


Fig. S4 Coordination modes of the carboxyl ligands in **1** (a) and **2** (b).

PXRD

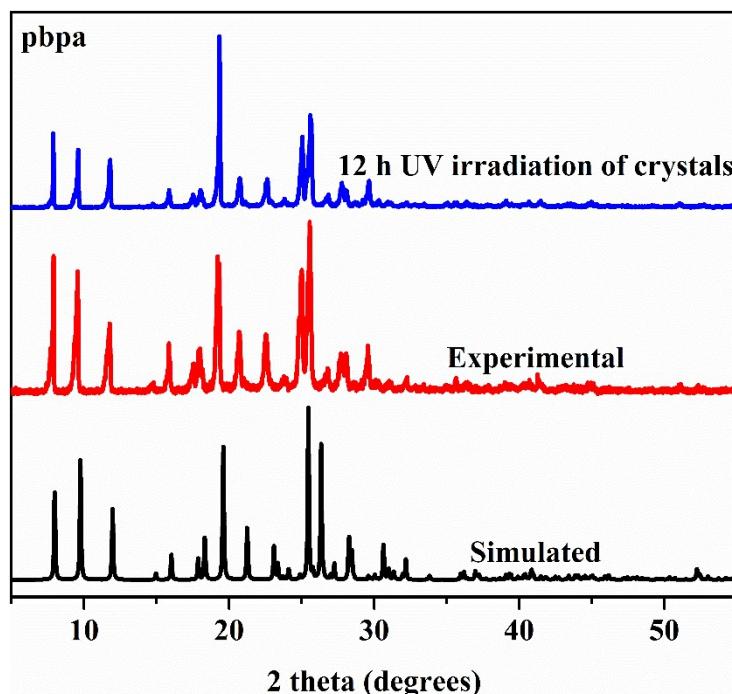


Fig. S5 PXRD patterns of pbpa.

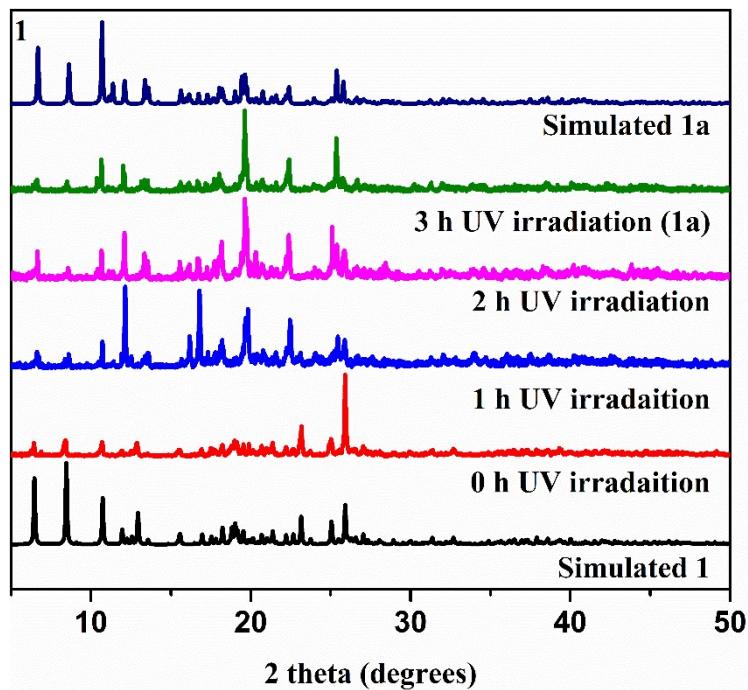


Fig. S6 PXRD patterns of 1-1a.

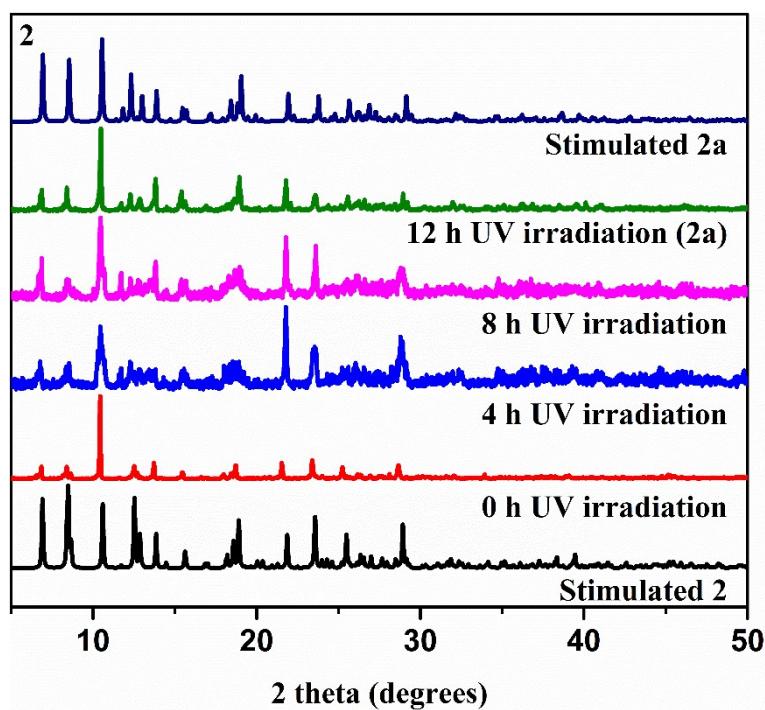


Fig. S7 PXRD patterns of 2-2a.

NMR spectra

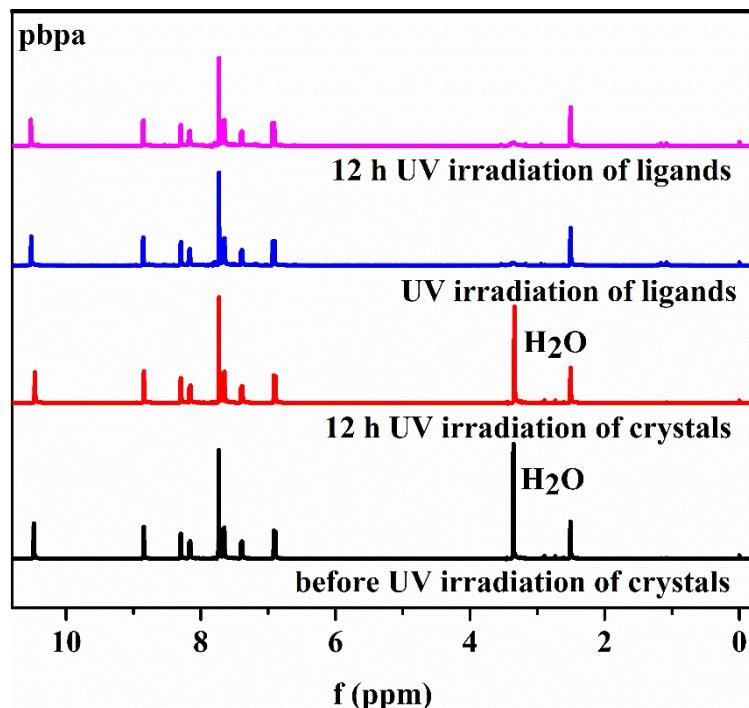


Fig. S8 ¹H-NMR patterns of pbpa ligands and crystals in DMSO-*d*₆ subjected to UV irradiation at 0 and 12 h.

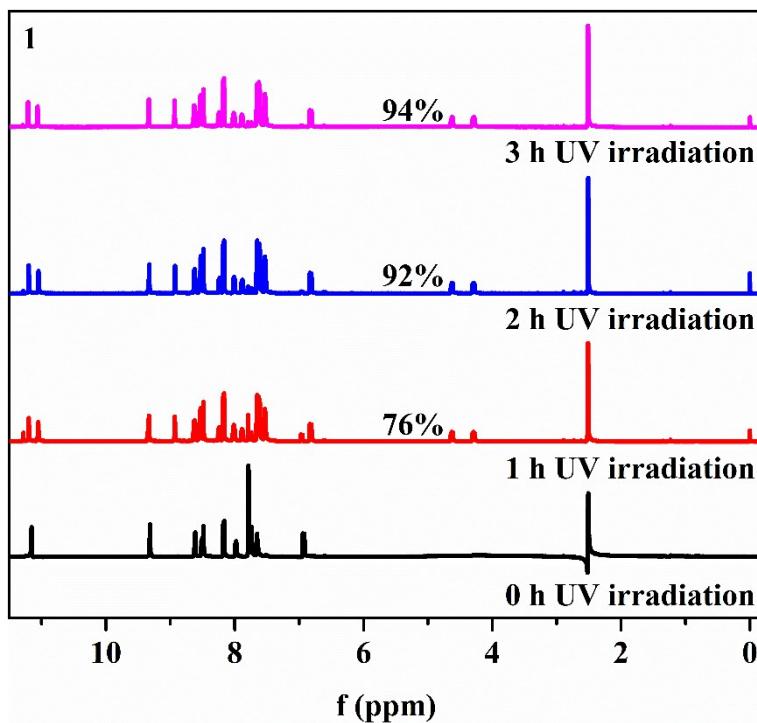


Fig. S9 ¹H-NMR patterns of 1 in DMSO-*d*₆ subjected to UV irradiation at same interval of time.

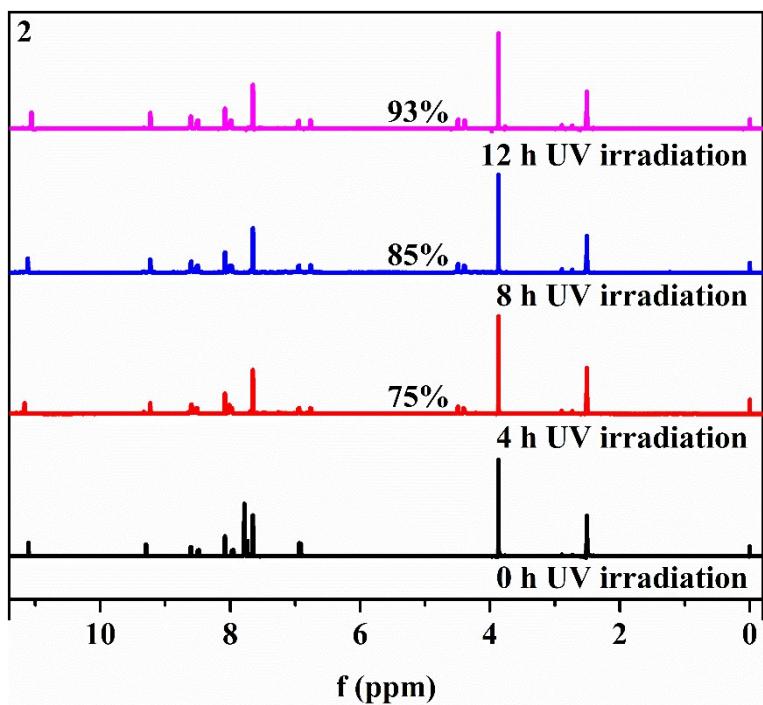


Fig. S10 ¹H-NMR patterns of **2** in DMSO-*d*₆ subjected to UV irradiation at same interval of time.

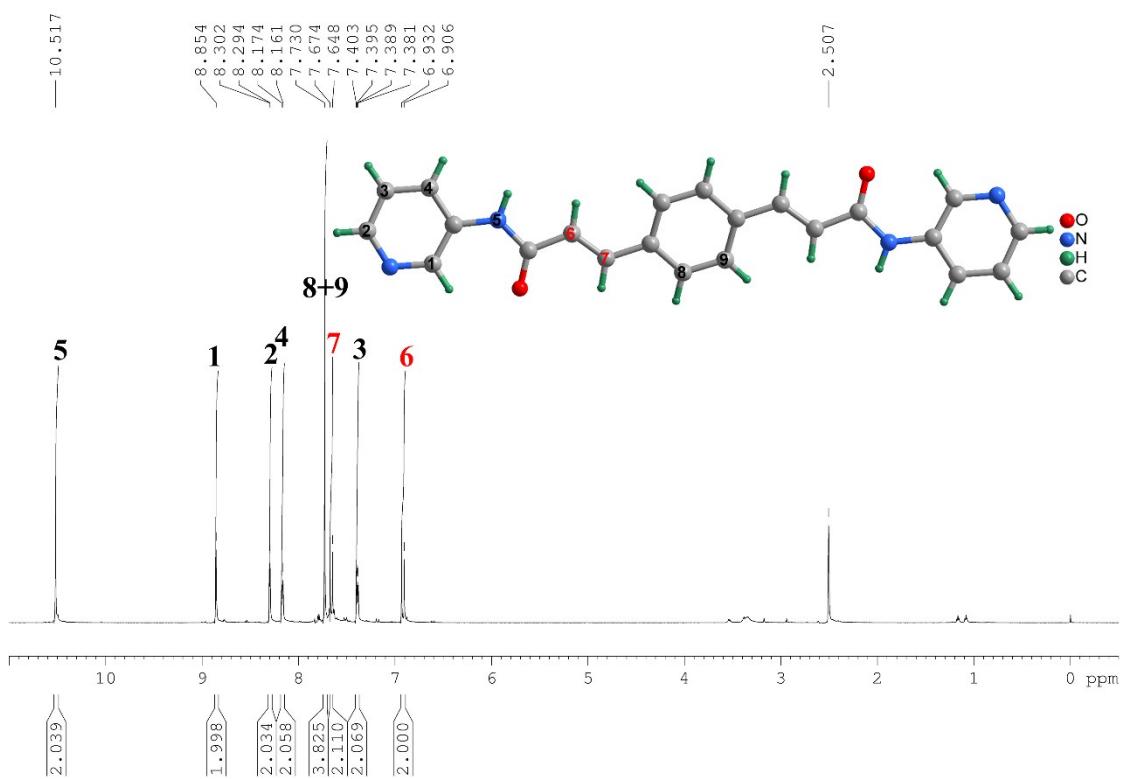


Fig. S11 ¹H-NMR spectrum of pbpa in DMSO-*d*₆.

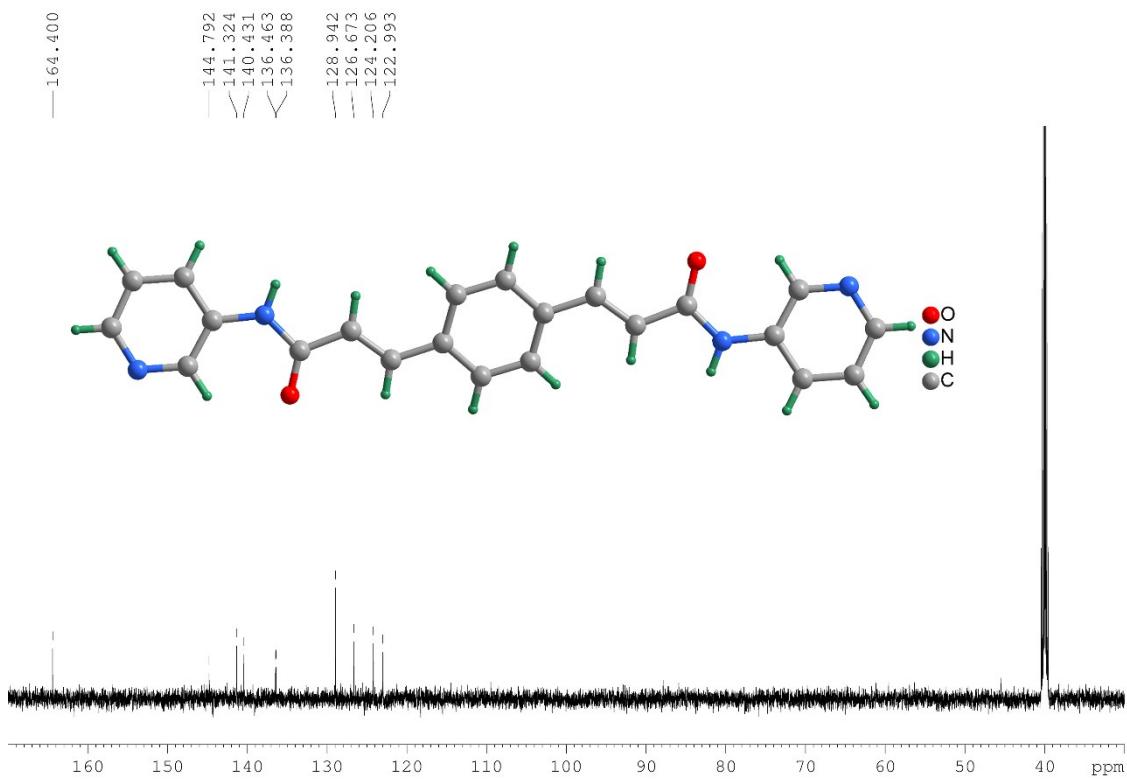


Fig. S12 ¹³C-NMR spectrum of pbpa in DMSO-*d*₆.

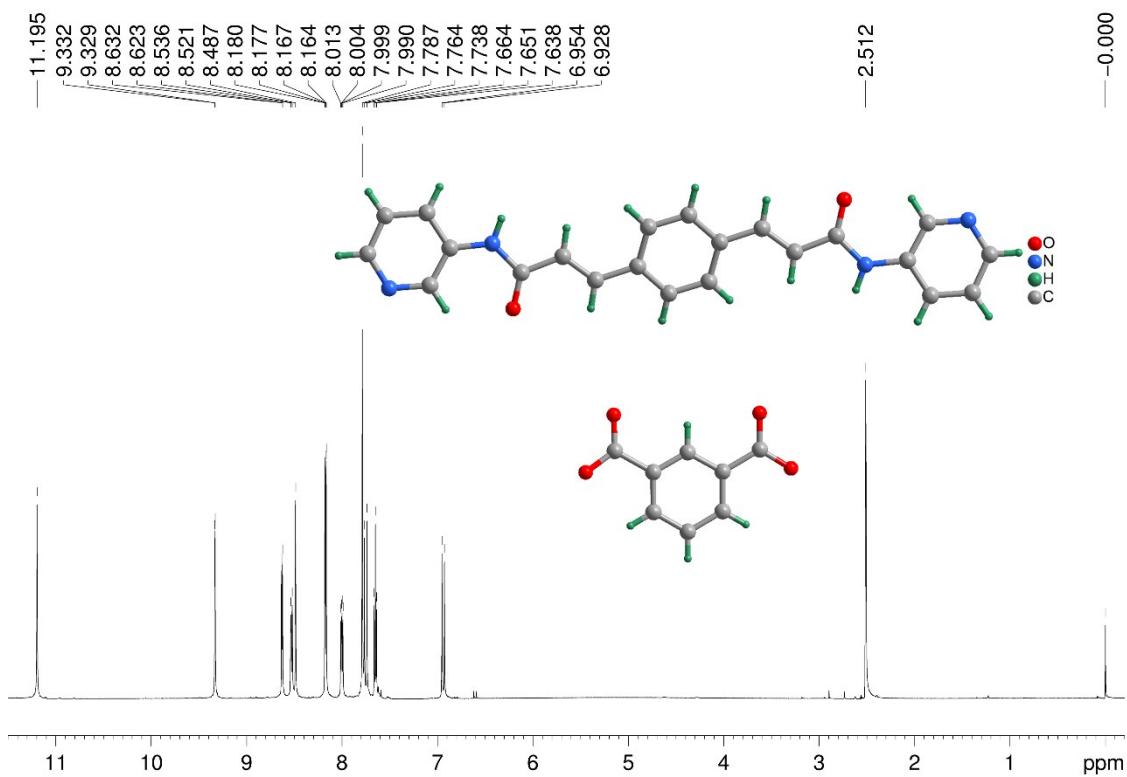


Fig. S13 ¹H-NMR spectrum of 1 in DMSO-*d*₆.

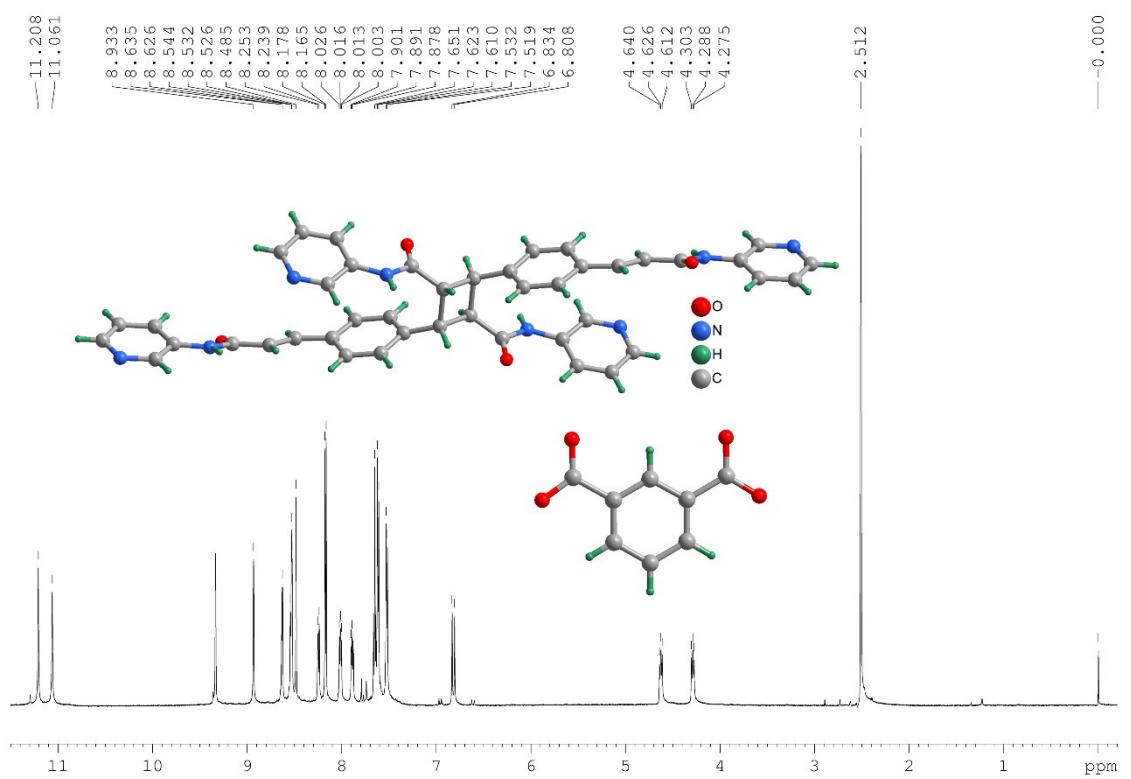


Fig. S14 ^1H -NMR spectrum of **1a** in $\text{DMSO}-d_6$.

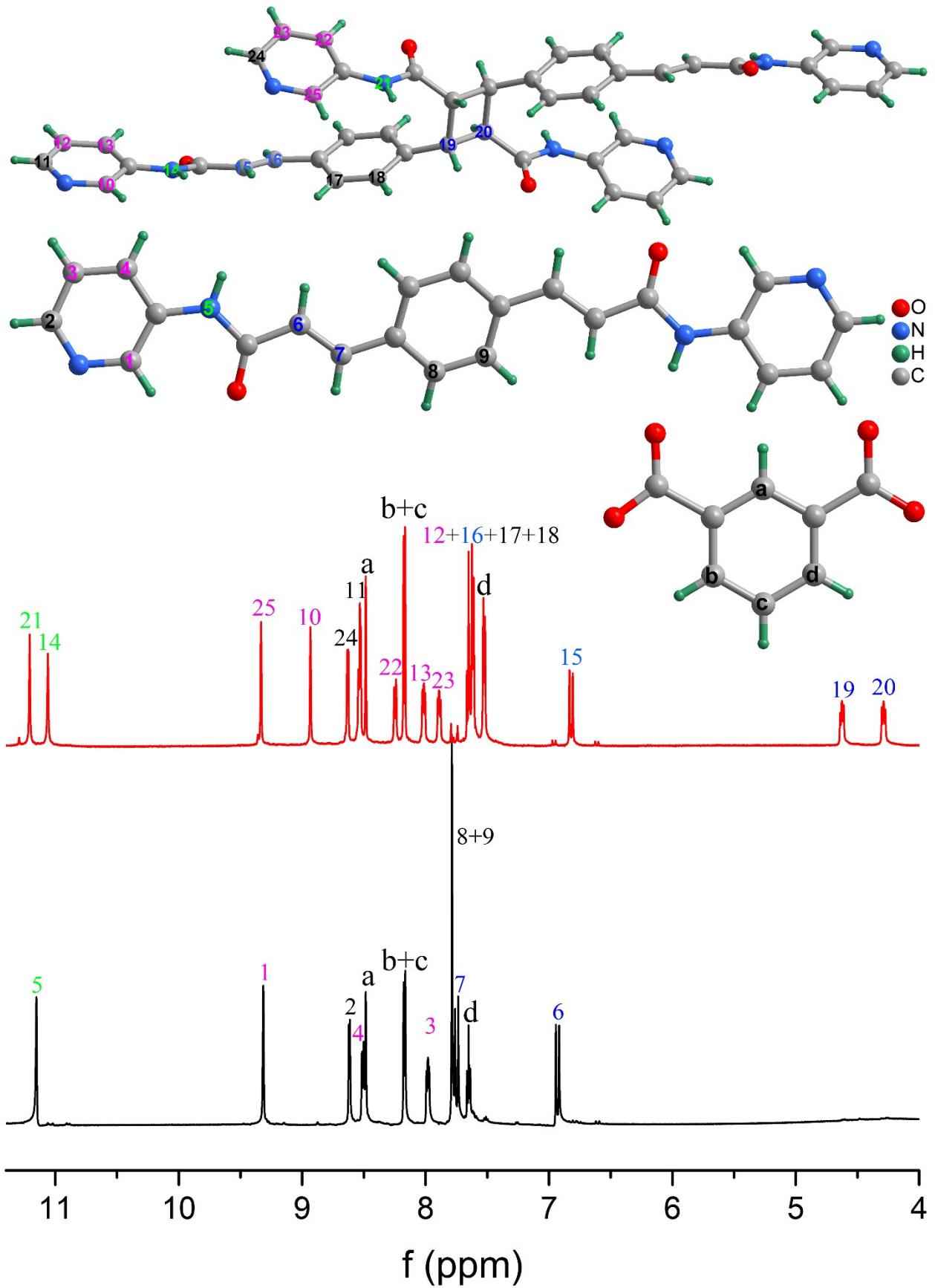


Fig. S15 A comparison of the ¹H-NMR spectrum of **1** and **1a** in $\text{DMSO}-d_6$.

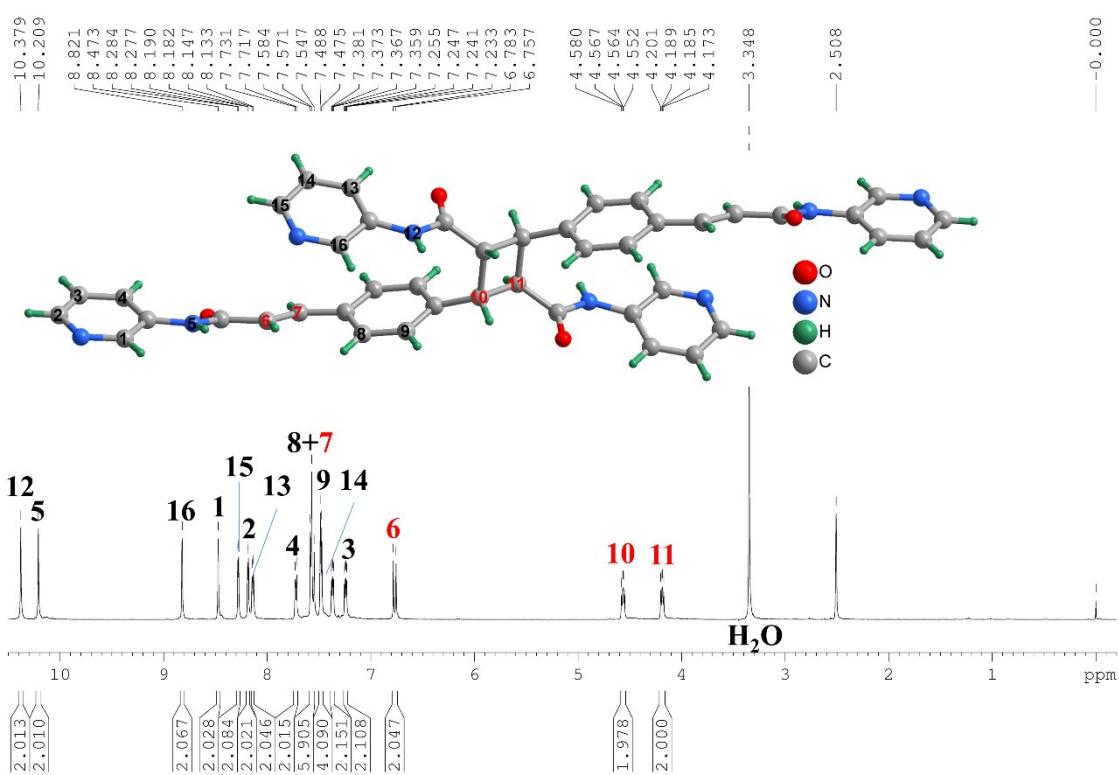


Fig. S16 ^1H -NMR spectrum of **babavpcb** in $\text{DMSO}-d_6$.

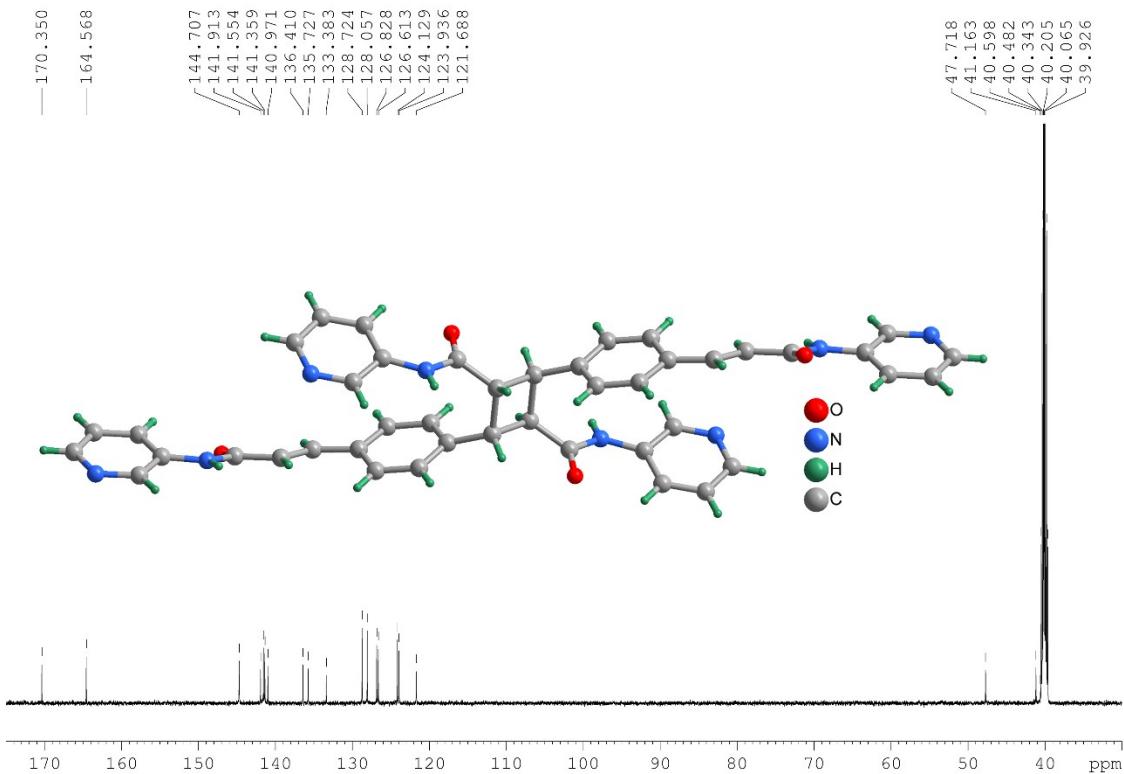


Fig. S17 ^{13}C -NMR spectrum of **babavpcb** in $\text{DMSO}-d_6$.

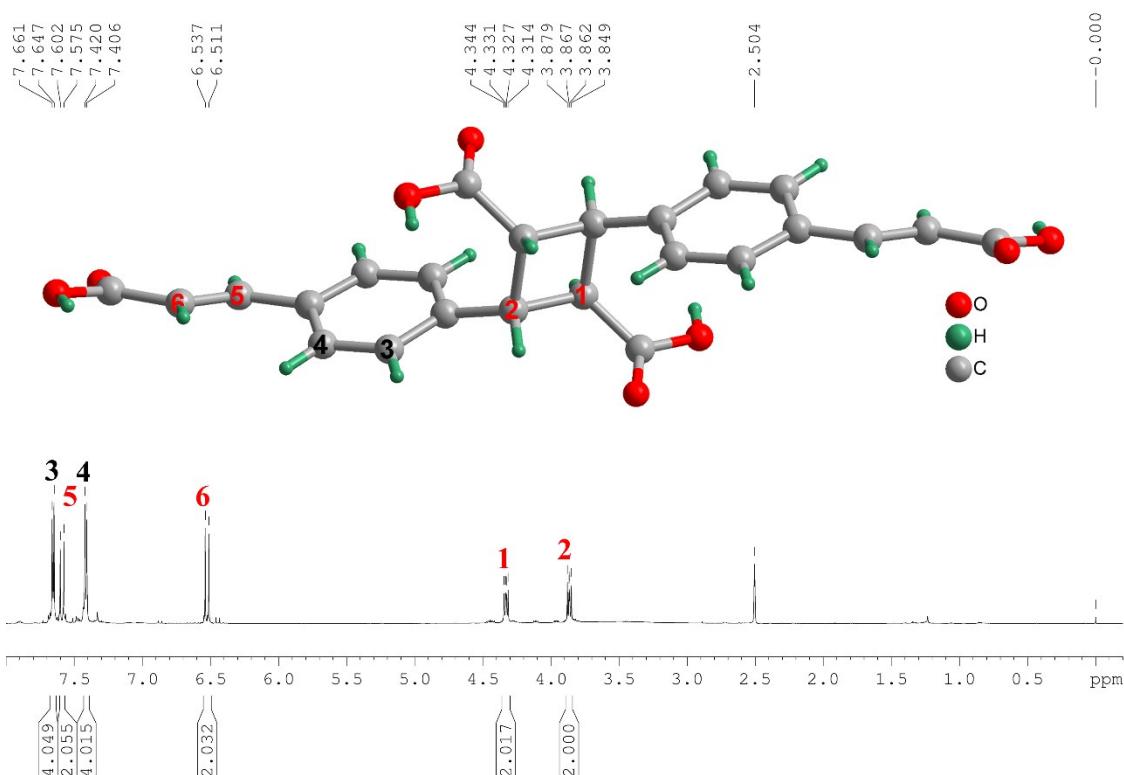


Fig. S18 ^1H -NMR spectrum of **bcbcvpcb** in $\text{DMSO}-d_6$.

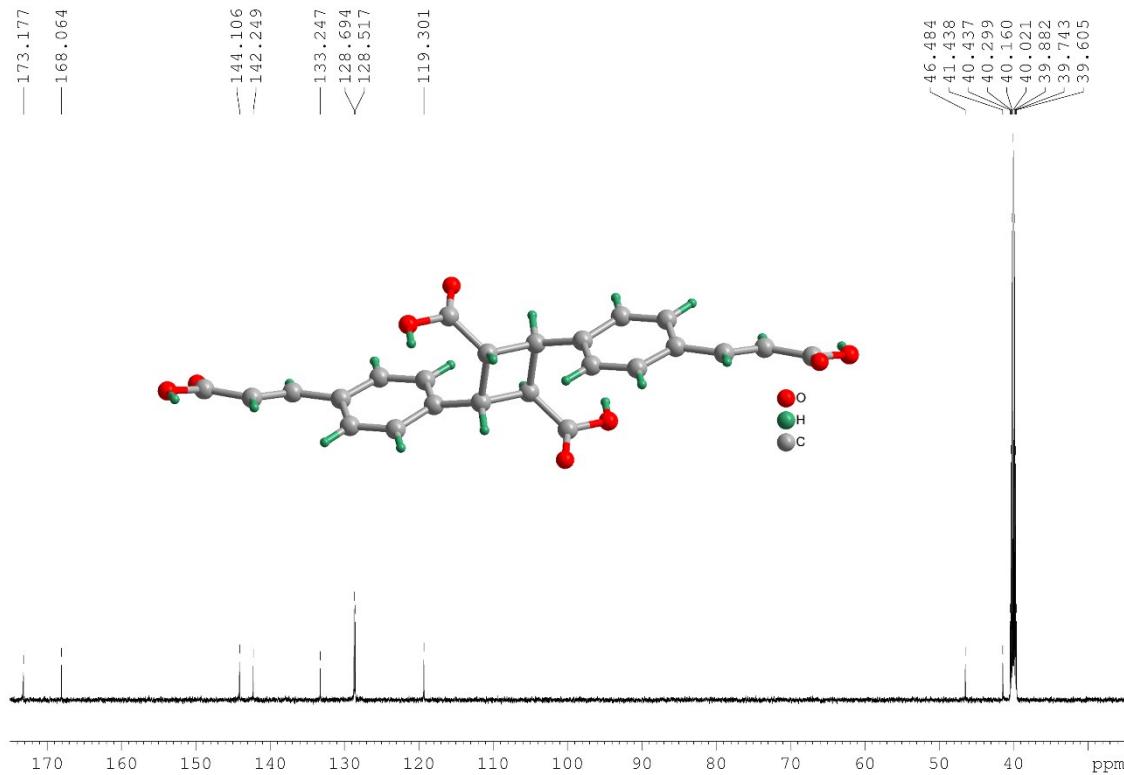


Fig. S19 ^{13}C -NMR spectrum of **bcbcvpcb** in $\text{DMSO}-d_6$.

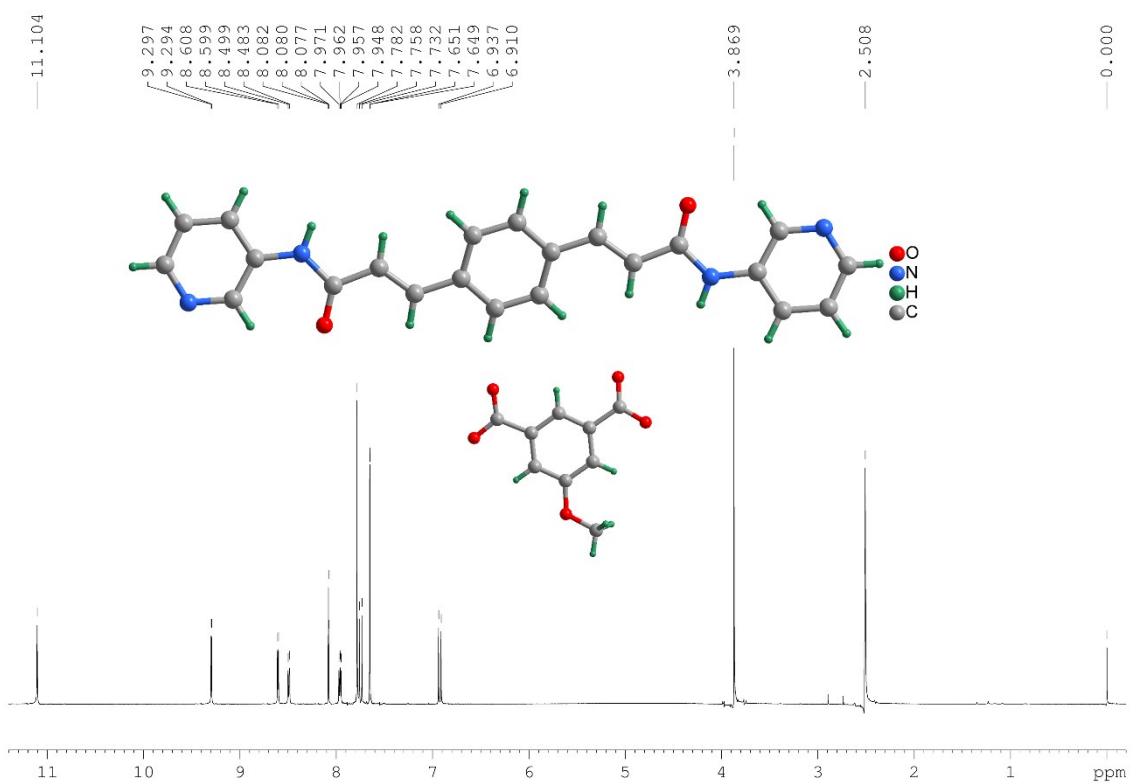


Fig. S20 ^1H -NMR spectrum of **2** in $\text{DMSO}-d_6$.

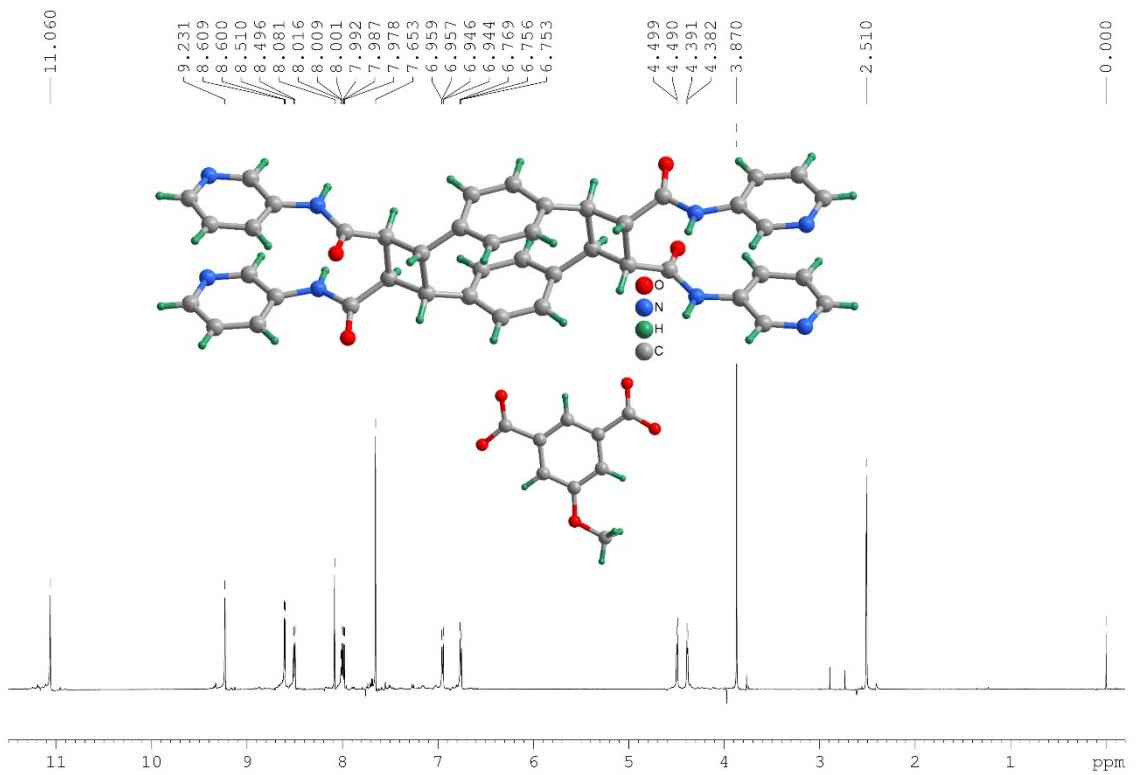


Fig. S21 ^1H -NMR spectrum of **2a** in $\text{DMSO}-d_6$.

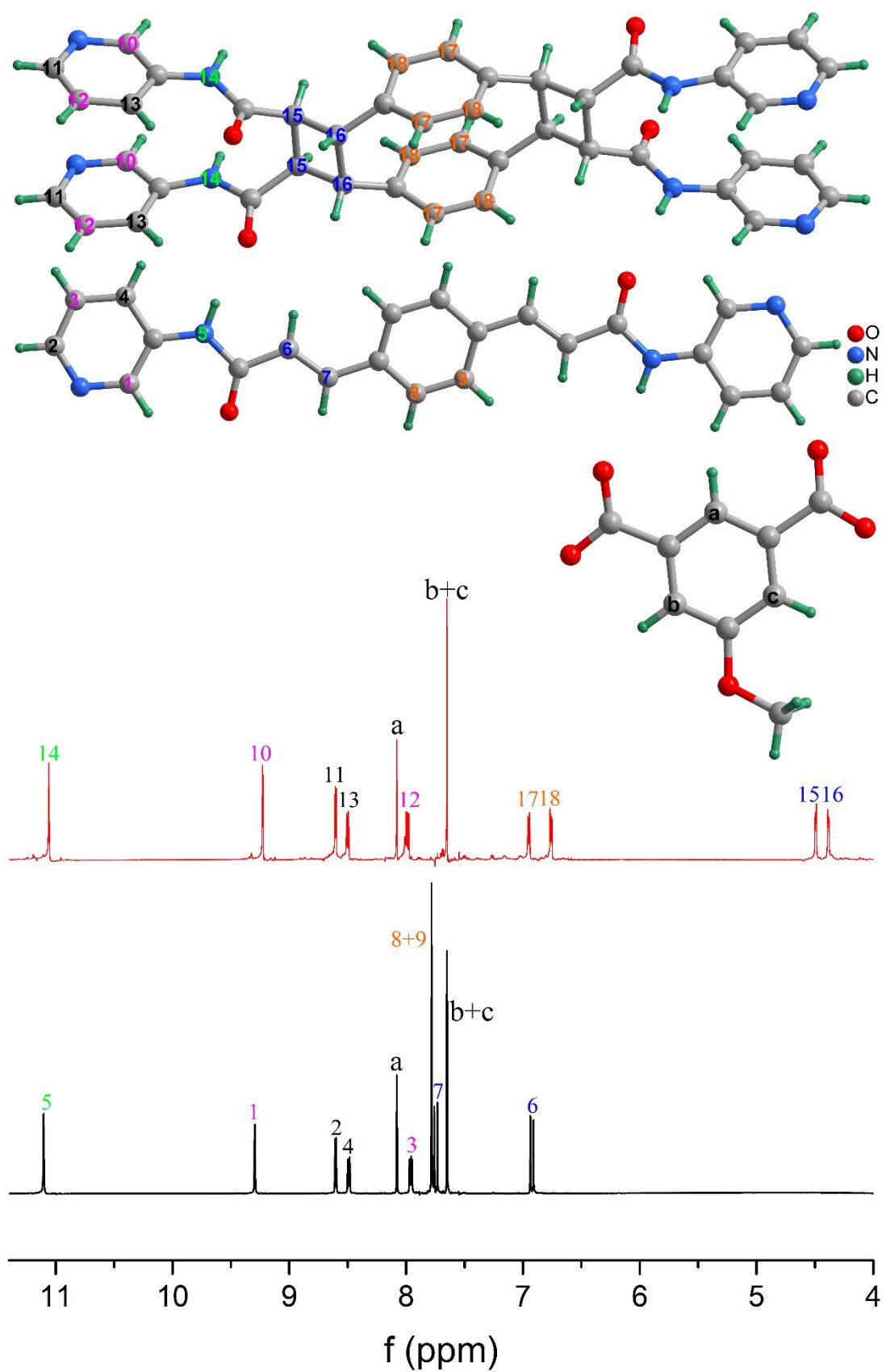


Fig. S22 A comparison of the ^1H -NMR spectra of **2** and **2a** in $\text{DMSO}-d_6$.

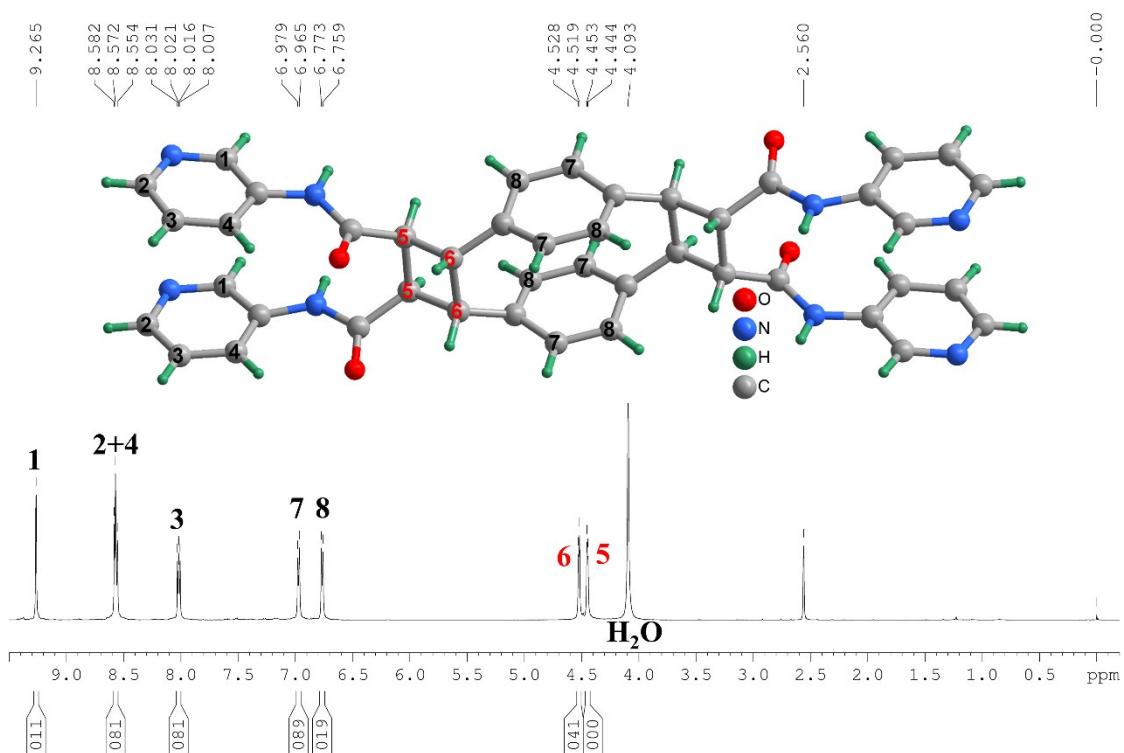


Fig. S23 ¹H-NMR spectrum of tapcp in DMSO-*d*₆ with 100 μ L DCl to dissolve the sample.

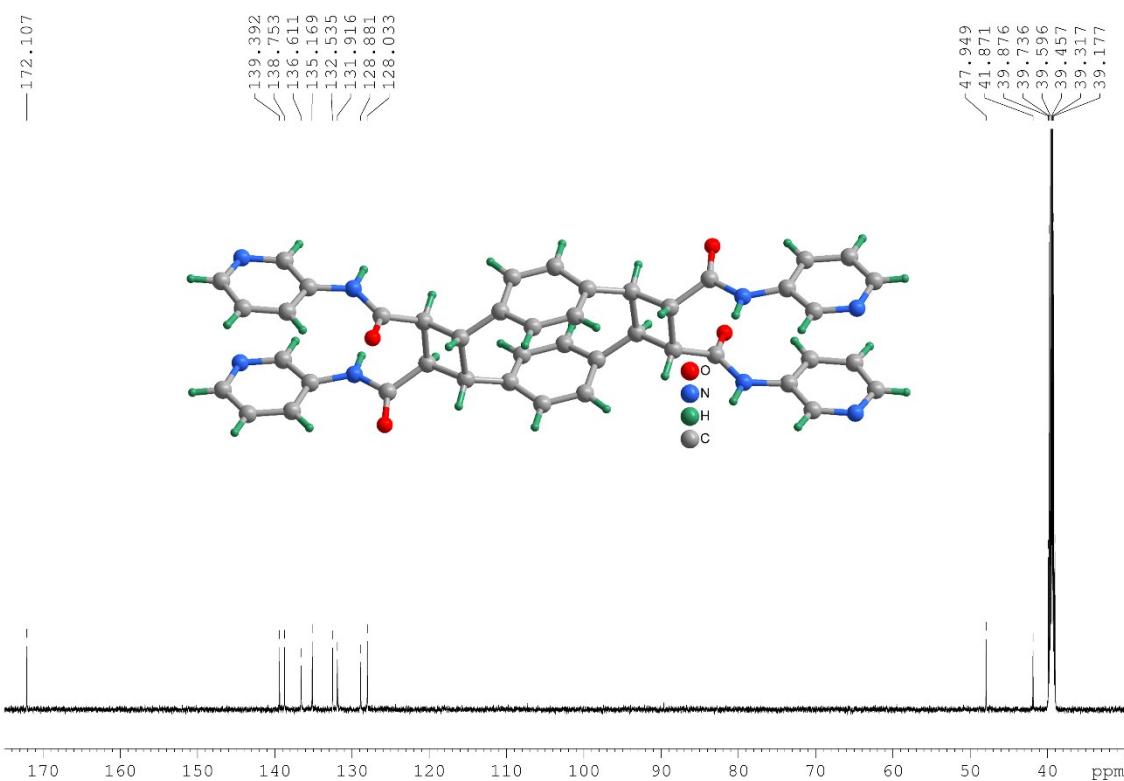
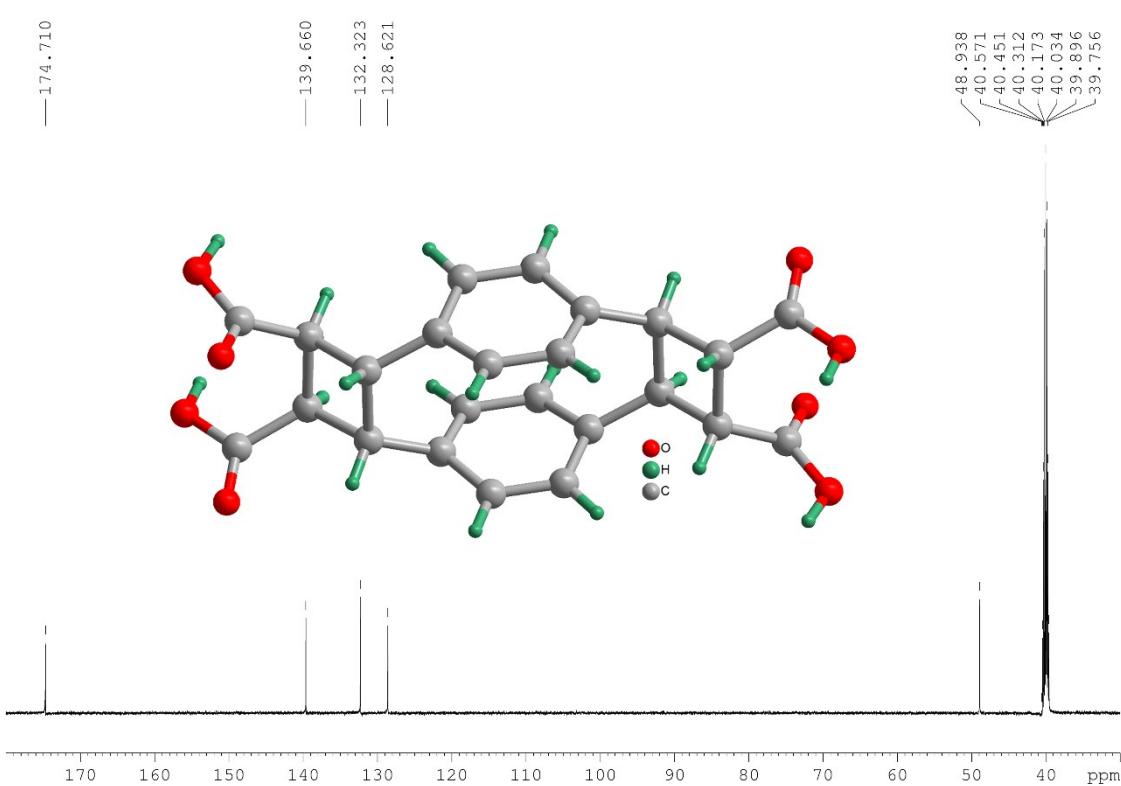
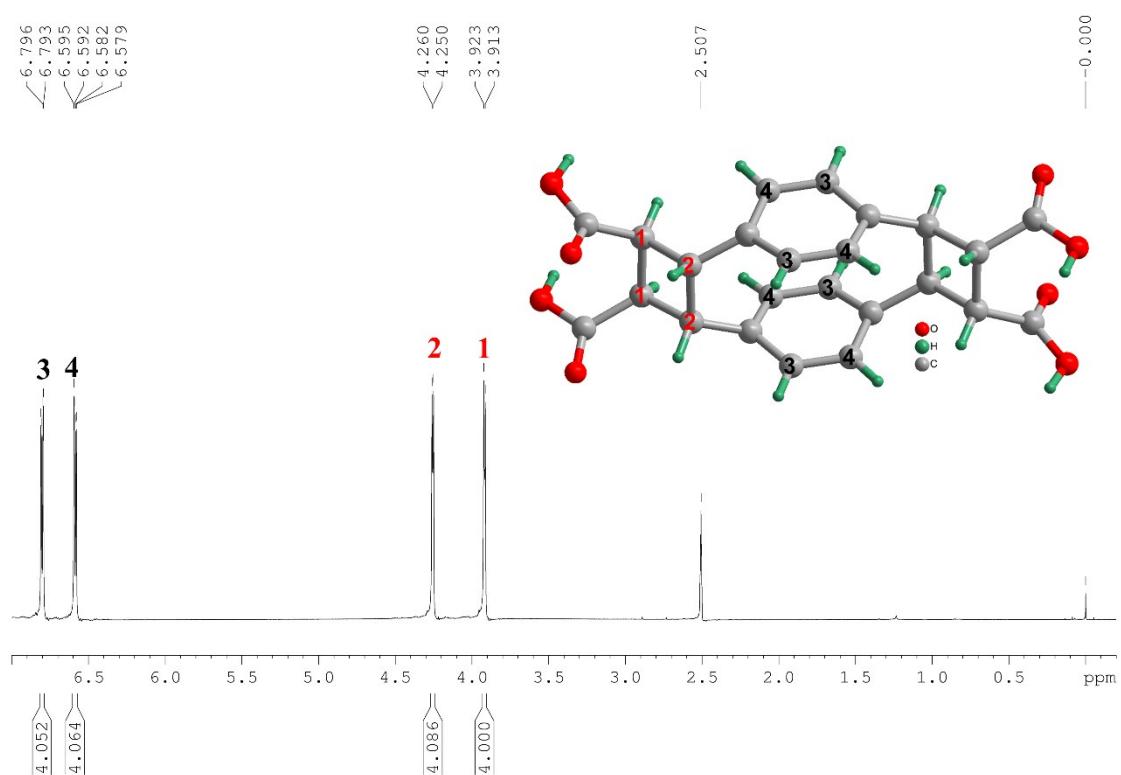


Fig. S24 ¹³C-NMR spectrum of tapcp in DMSO-*d*₆.



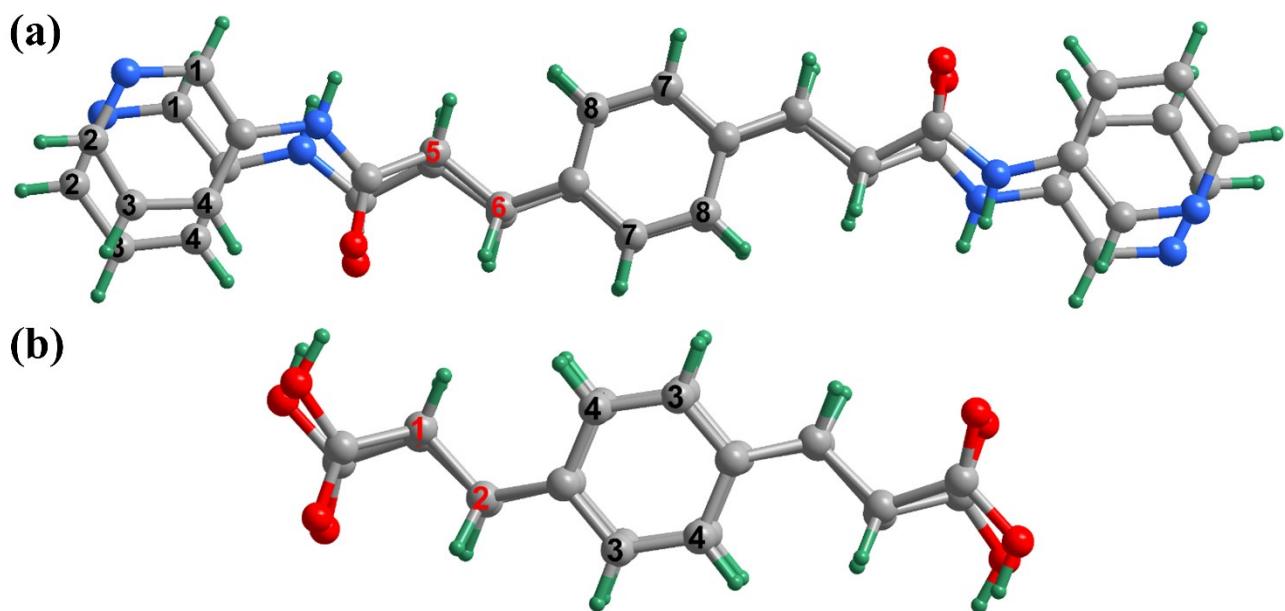


Fig. S27 Molecular Structures of **tapcp** and **tcpcp**.

Mass spectra

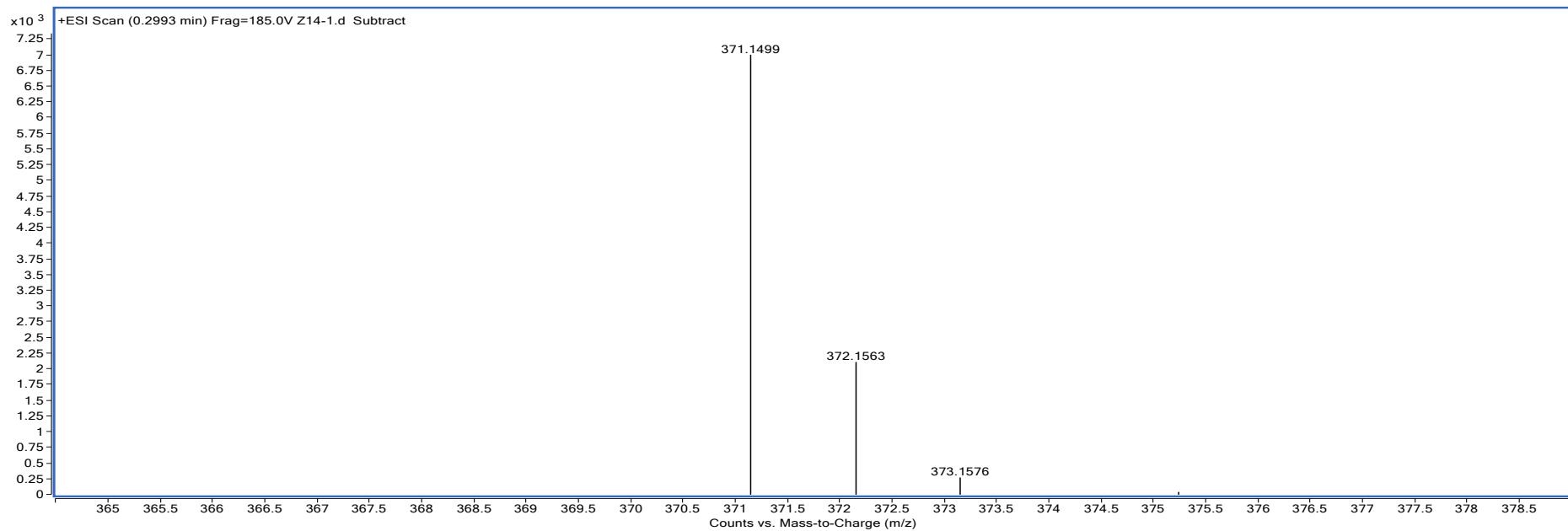


Fig. S28 Positive-ion ESI mass spectrum of pbpa in DMSO.

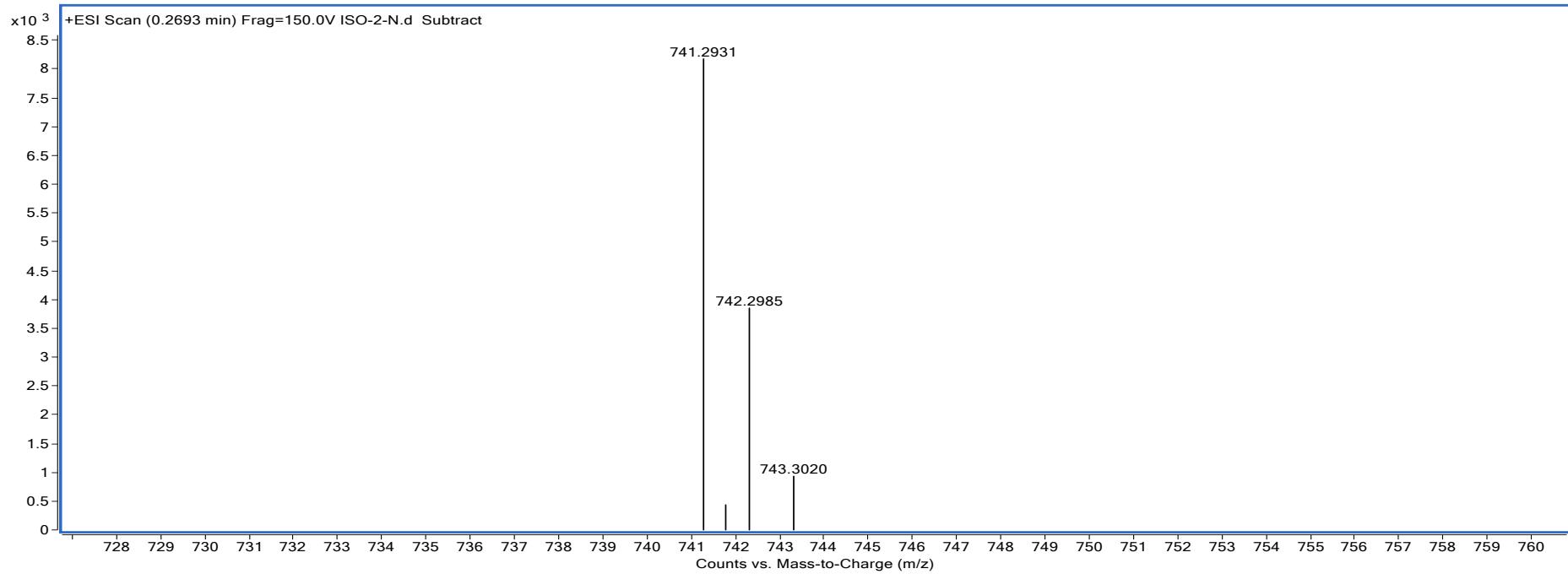


Fig. S29 Positive–ion mass spectrum of babavpcb in DMSO.

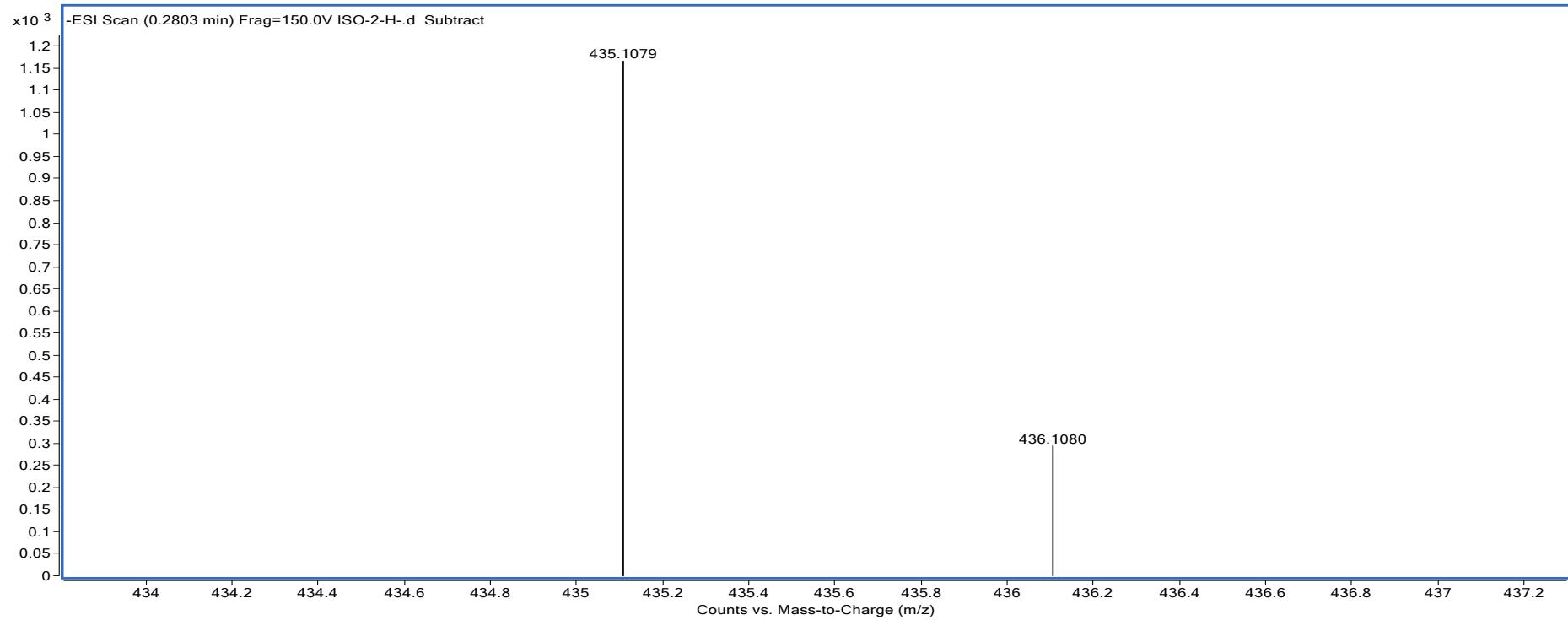


Fig. S30 Negative-ion mass spectrum of bcbcvpcb in DMSO.

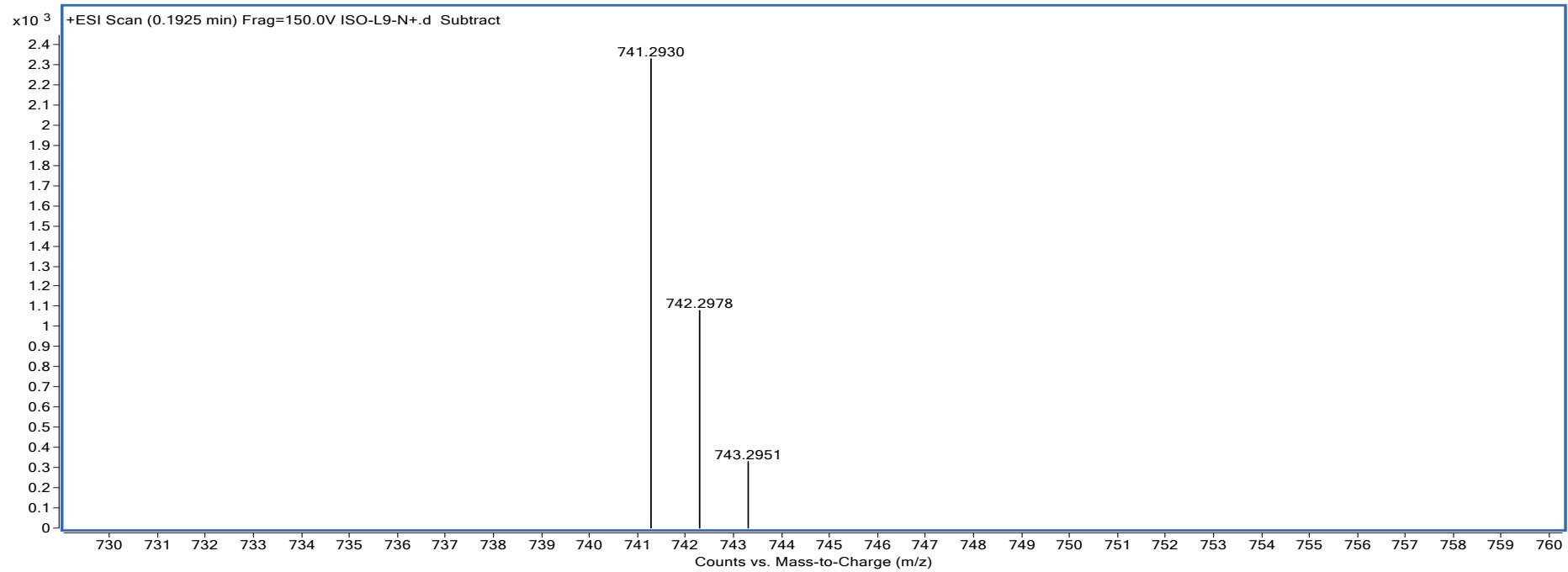


Fig. S31 Positive–ion mass spectrum of tapcp in DMSO.

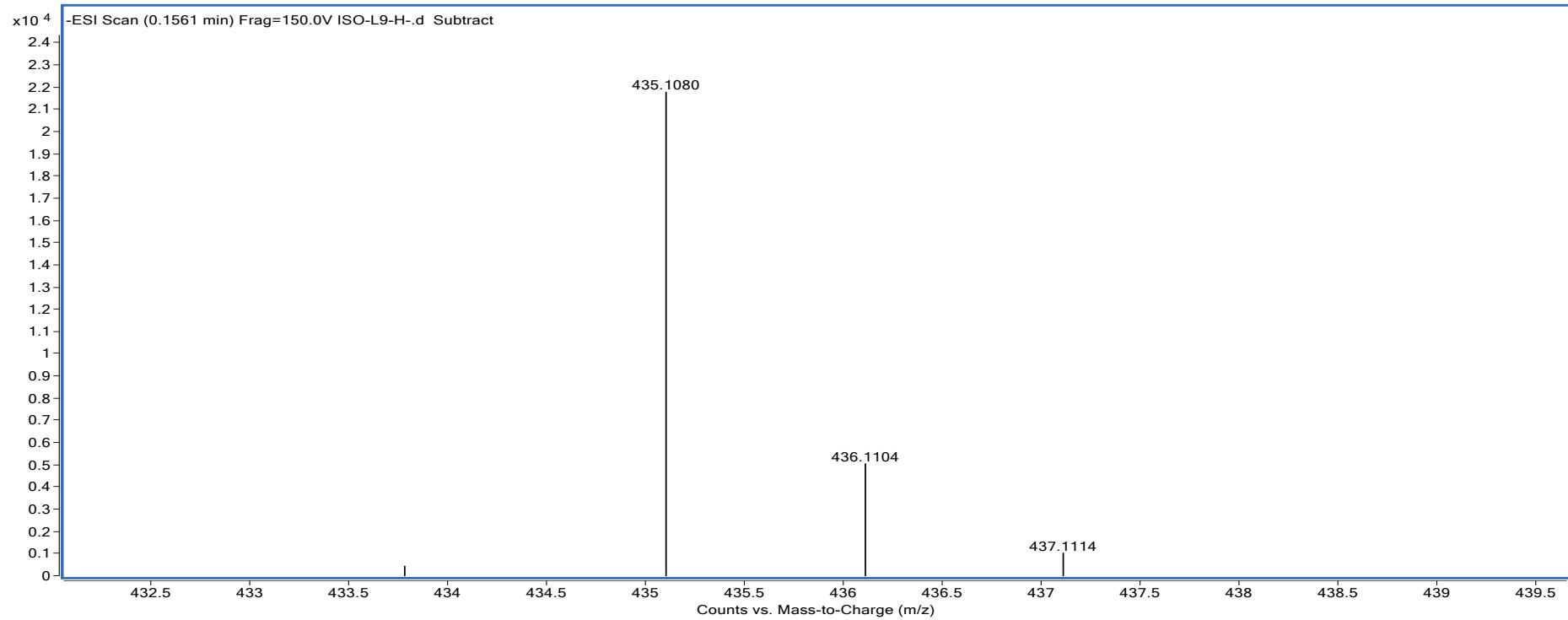


Fig. S32 Negative-ion mass spectrum of tcpcp in DMSO.

IR spectra

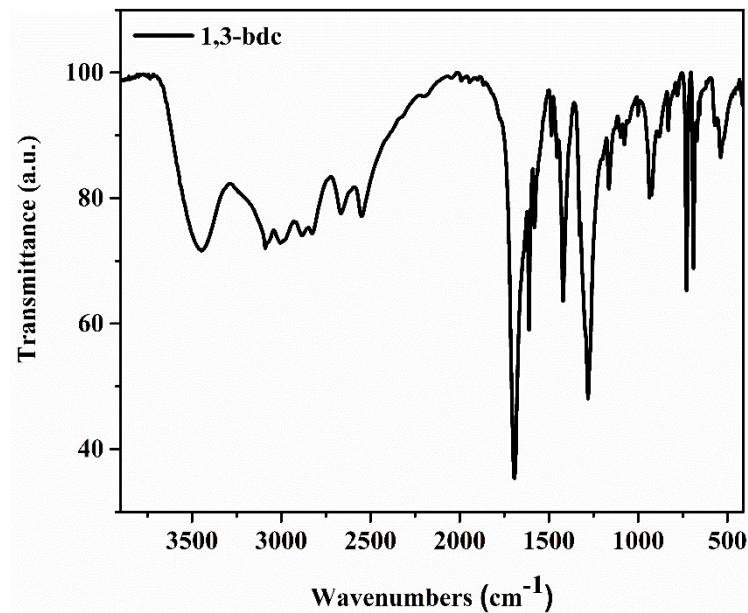


Fig. S33 IR spectra of **1,3-bdc**.

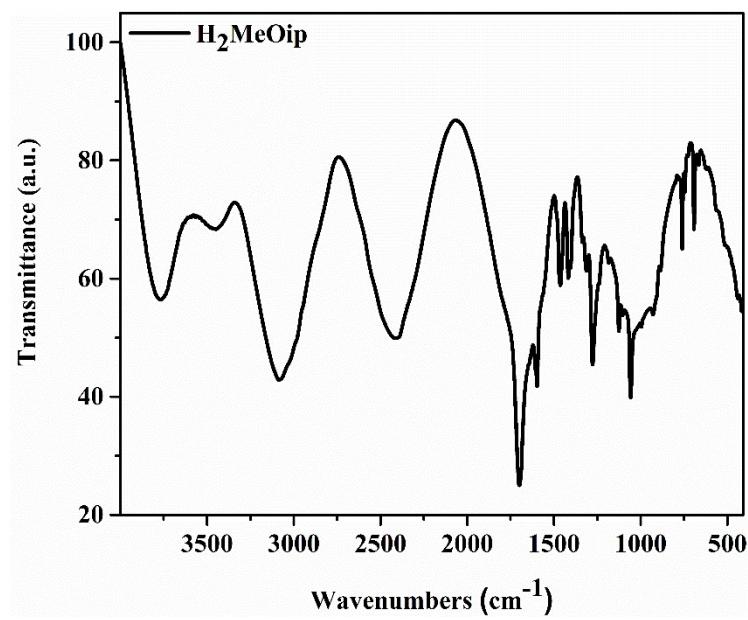


Fig. S34 IR spectra of H_2MeOip .

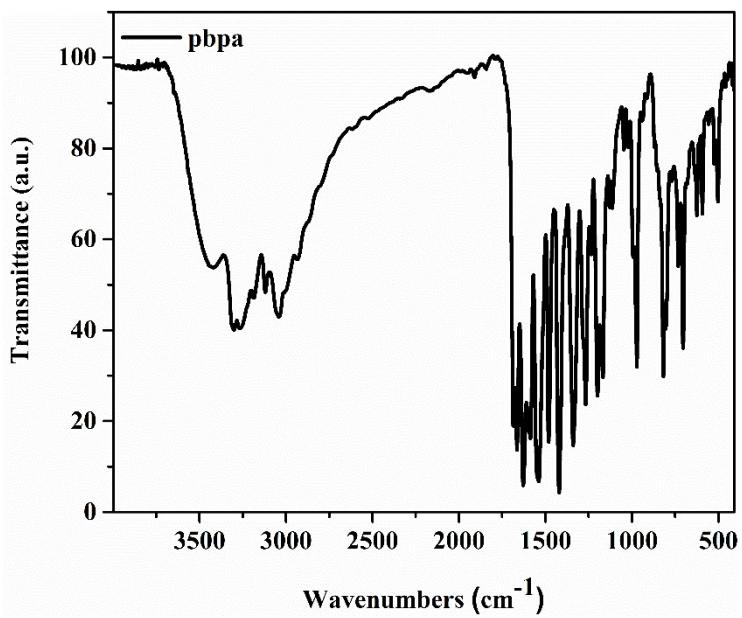


Fig. S35 IR spectra of pbpa.

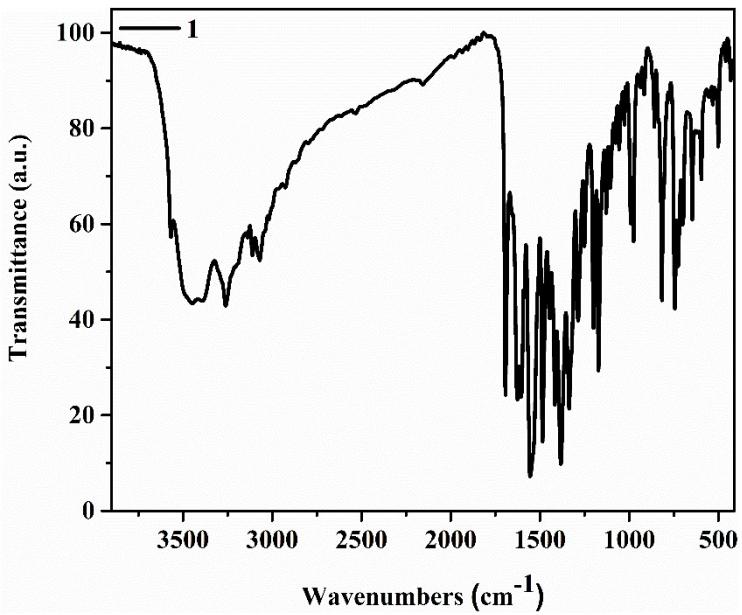


Fig. S36 IR spectra of 1.

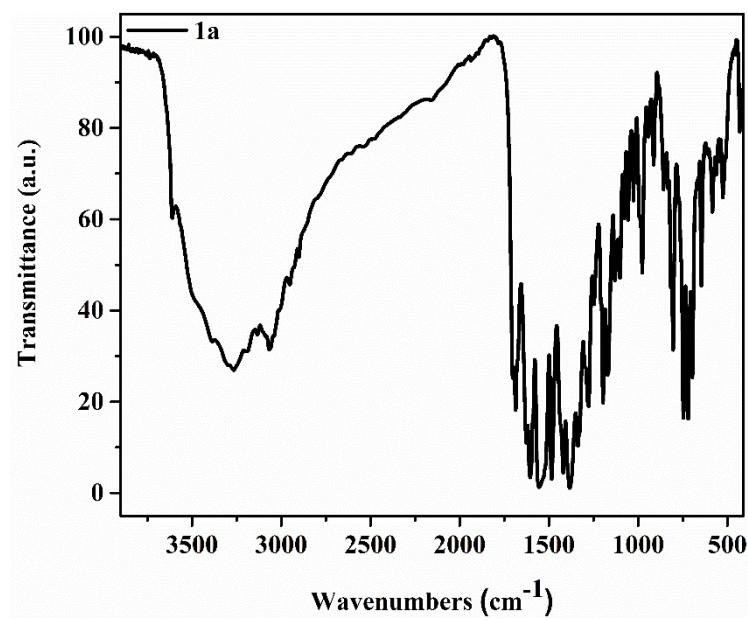


Fig. S37 IR spectra of **1a**.

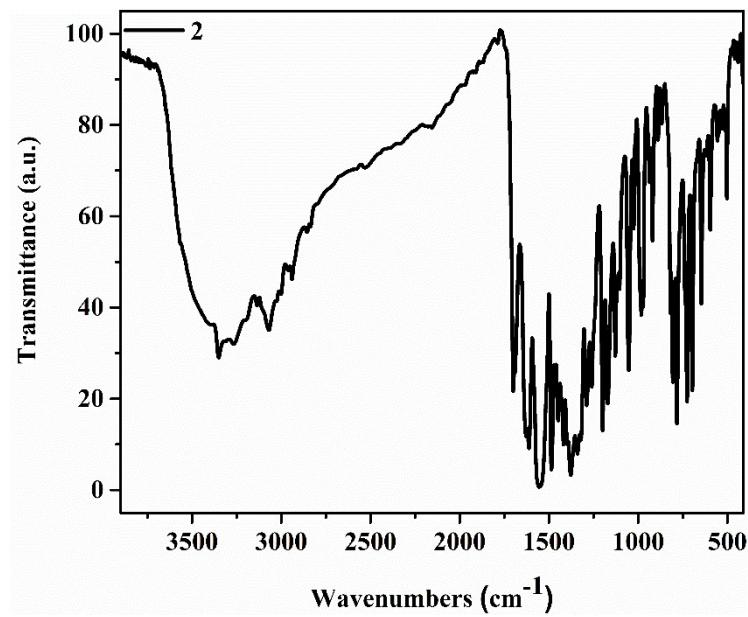


Fig. S38 IR spectra of **2**.

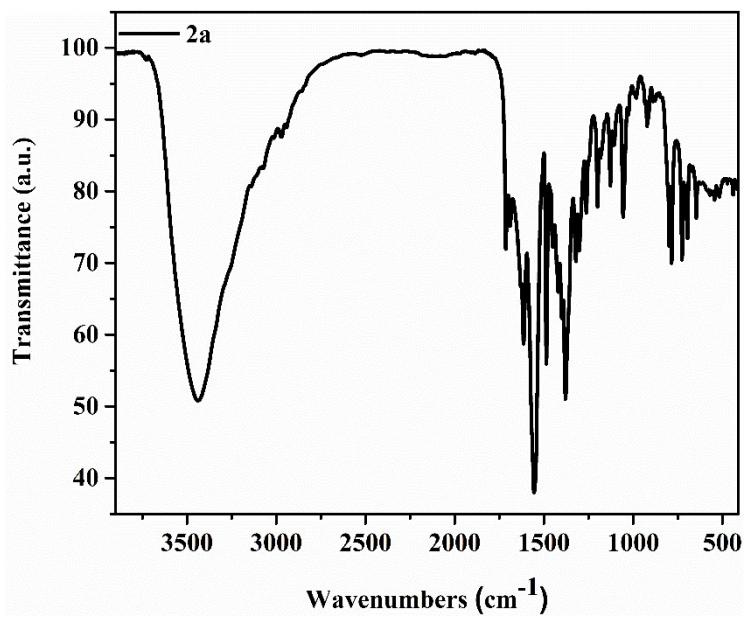


Fig. S39 IR spectra of **2a**.

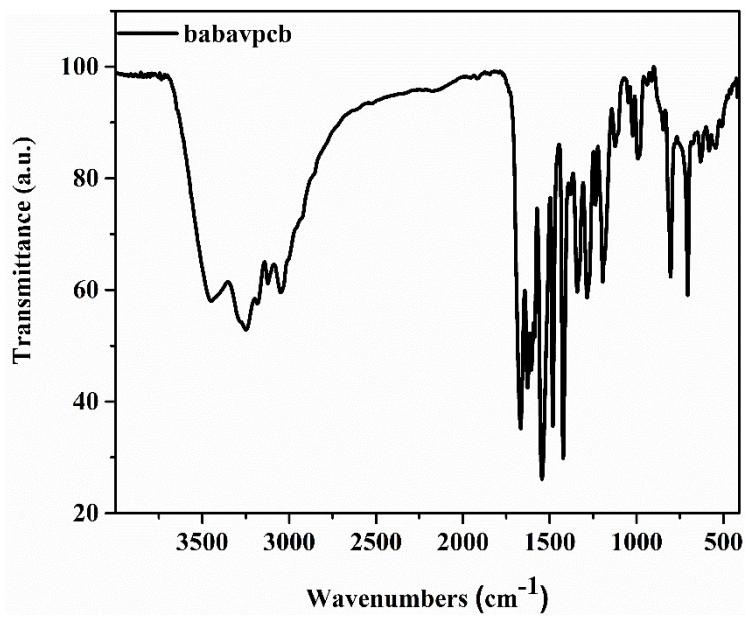


Fig. S40 IR spectra of **babavpcb**.

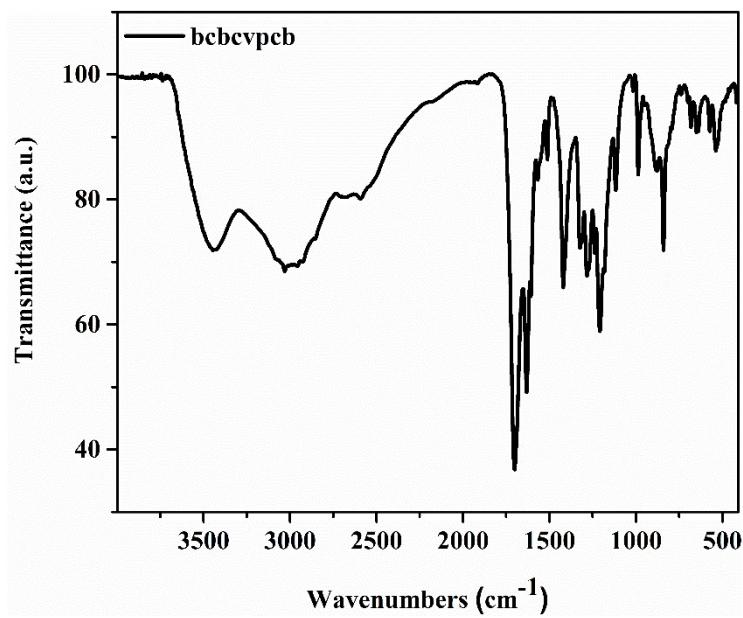


Fig. S41 IR spectra of **bcbcvpcb**.

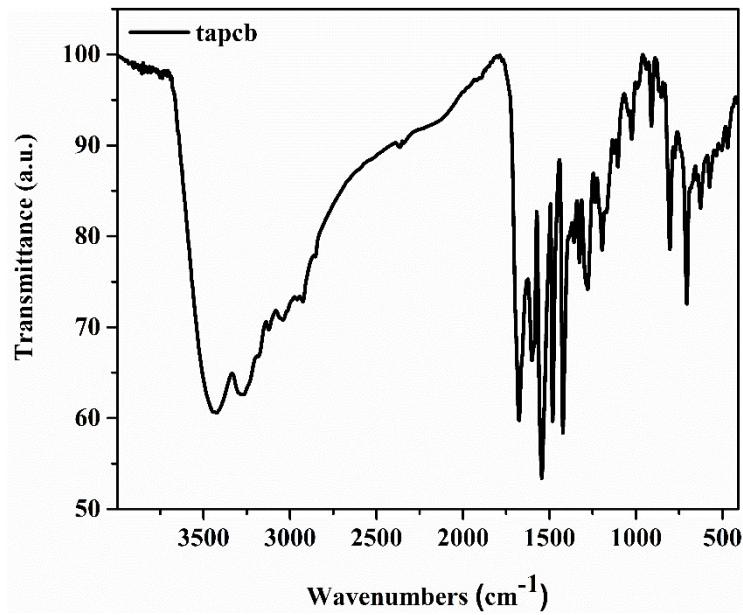


Fig. S42 IR spectra of **tapcb**.

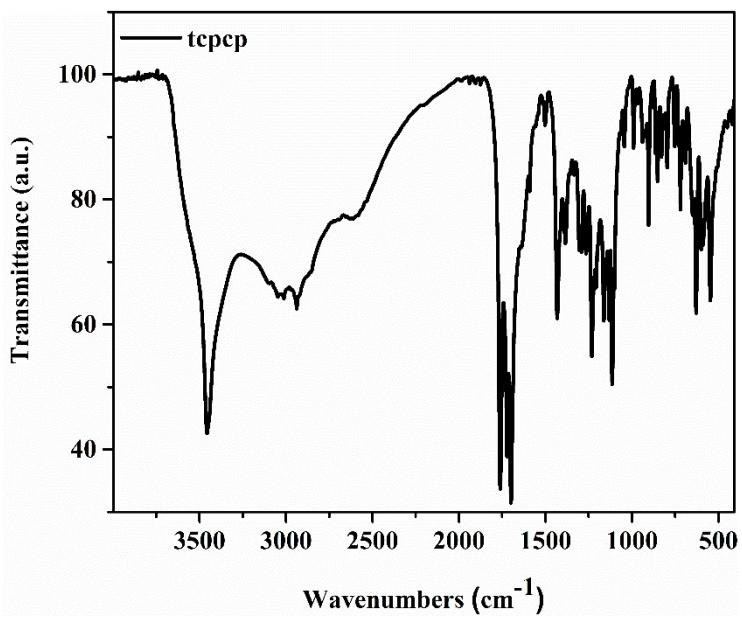


Fig. S43 IR spectra of tpcpb.

Thermo-gravimetric analysis (TGA)

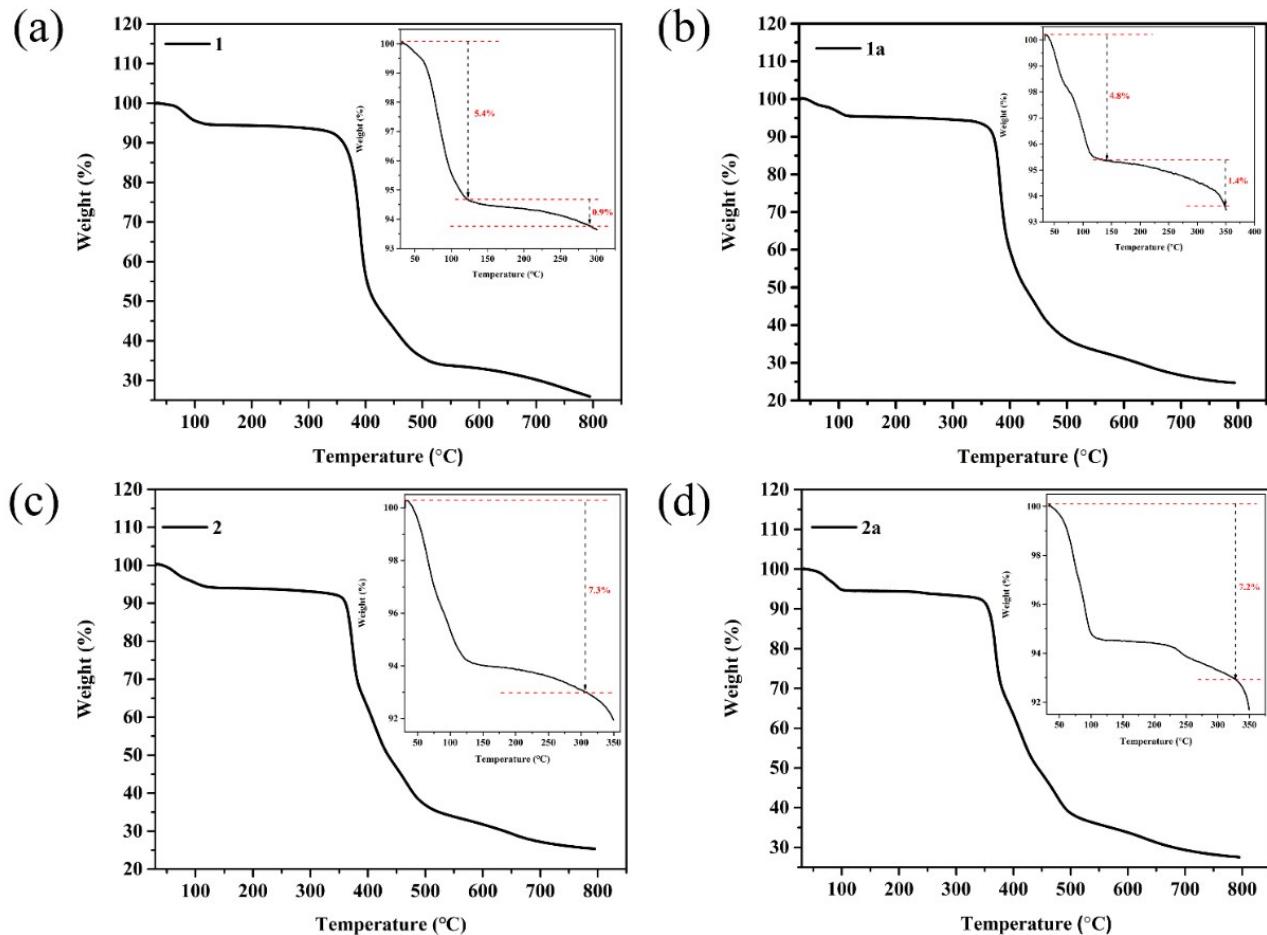


Fig. S44 TGA of 1–2a.

Solid-state UV-visible absorbance spectra

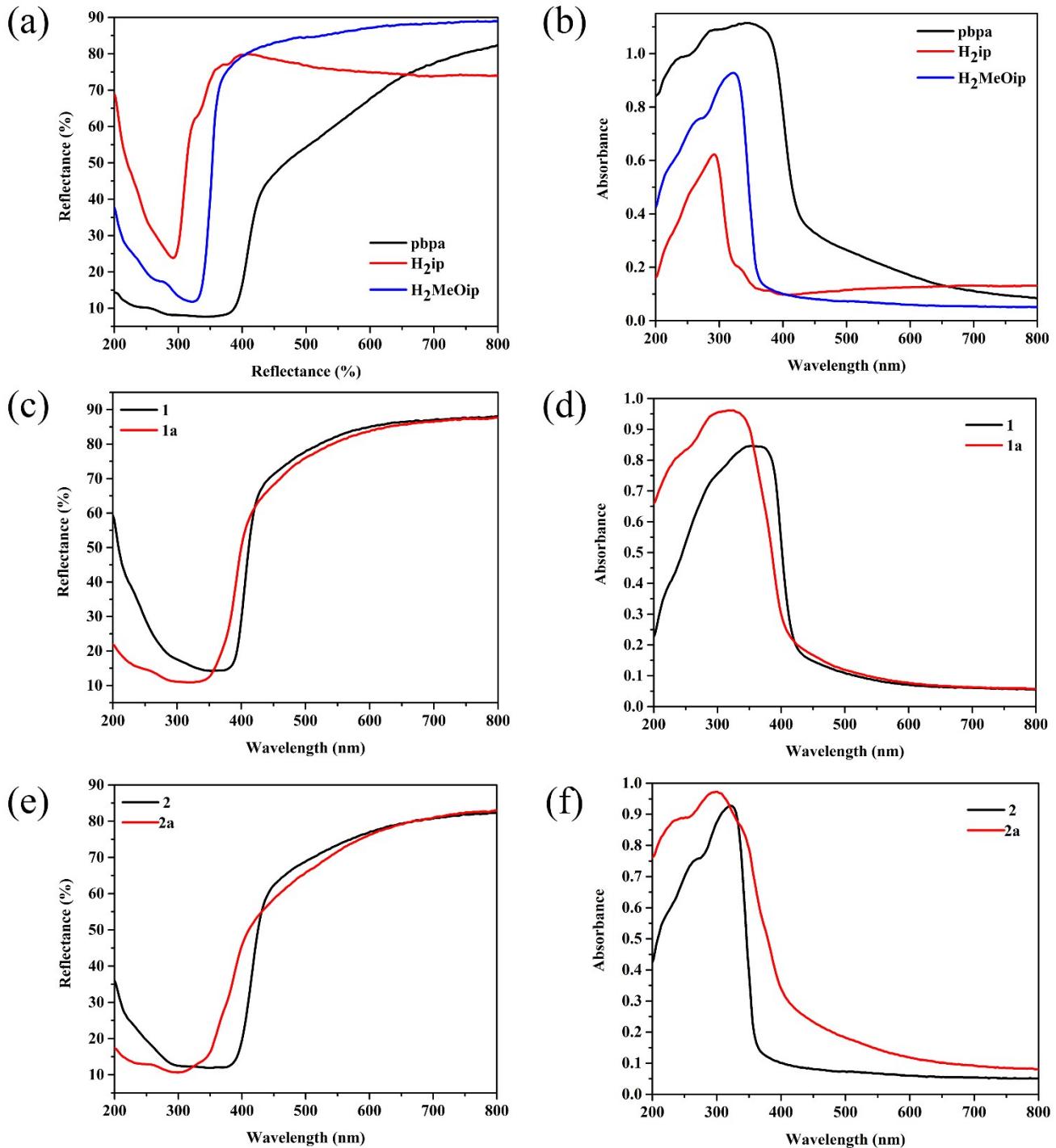


Fig. S45 Solid-state UV-visible absorbance.

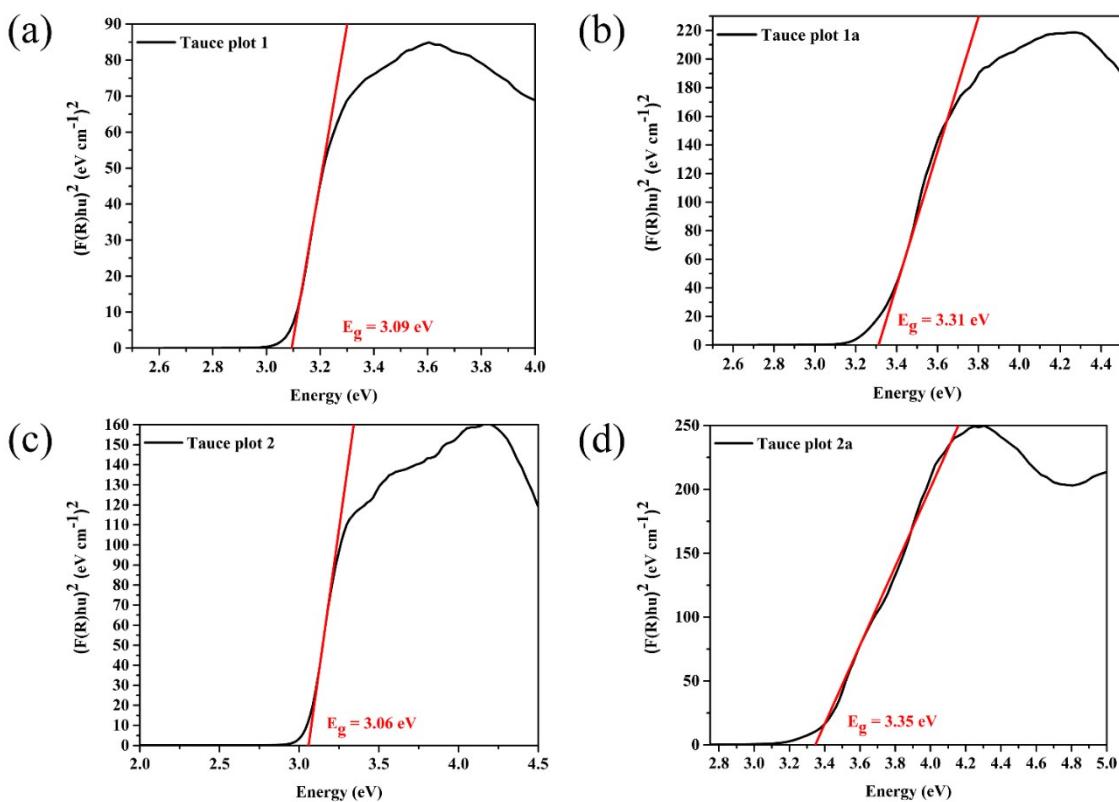


Fig. S46 the direct optical band gap.

Solid-state photoluminescence spectra

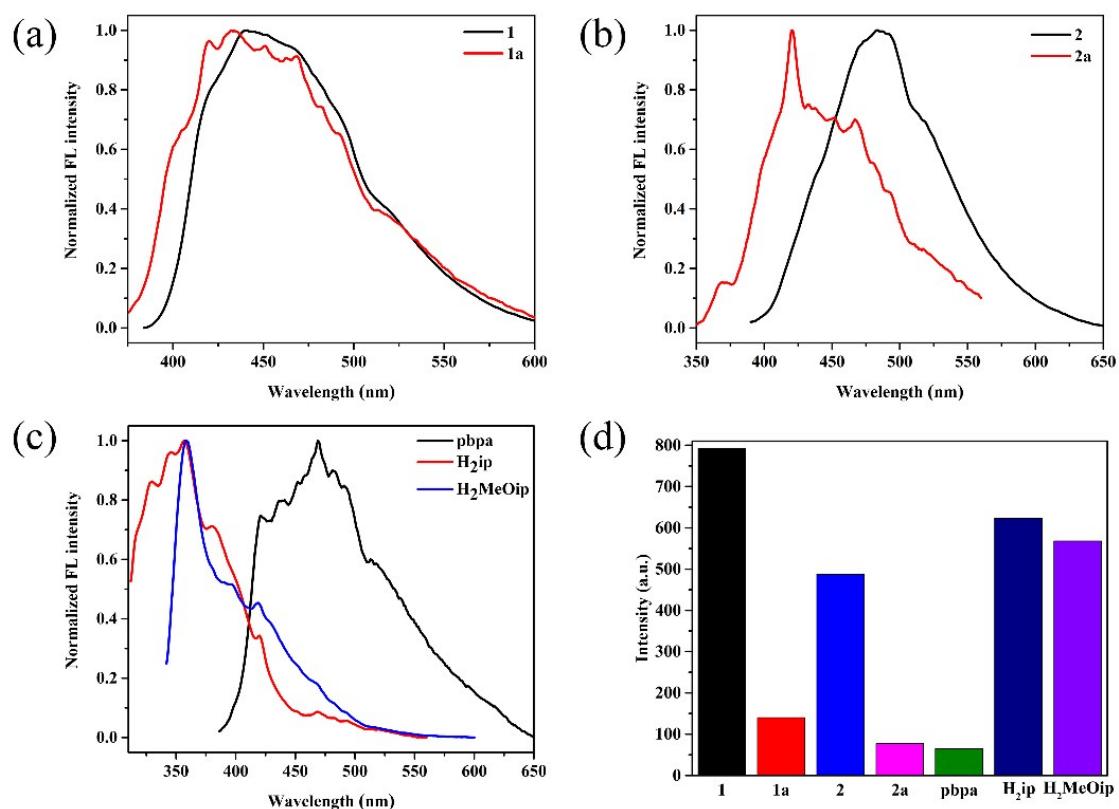


Fig. S47 Normalized fluorescence spectra and fluorescence intensity.