

Supporting information

A unique dual fluorescence emission in solid-state from small molecule based on phenanthrocarbazole with AIE luminogen as single-molecule white emissive material

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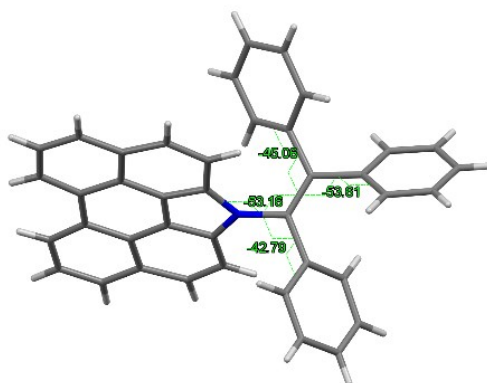
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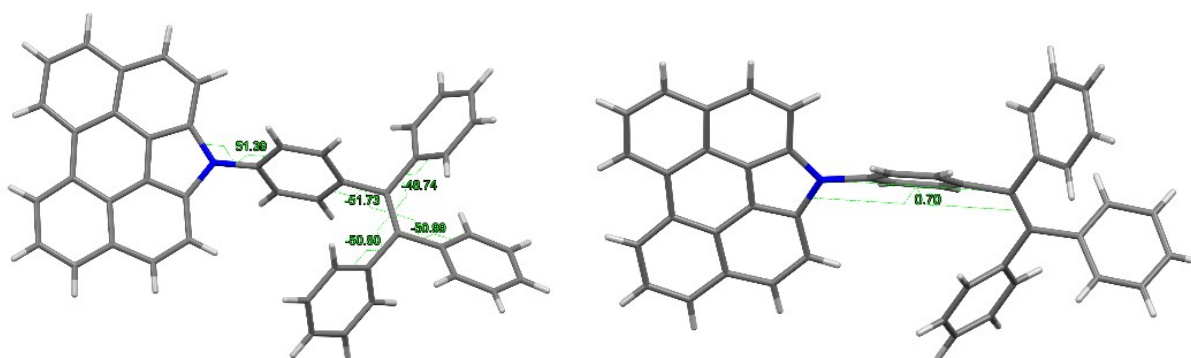
I Crystal Data and Packing

Table S1 Crystal data and structure refinement for PCTrPE and PCTPE .		
Identification code	PCTrPE	PCTPE
CCDC Deposition Number	1967746	1967946
Empirical formula	C ₄₀ H ₂₅ N	C ₄₆ H ₂₉ N
Formula weight	519.61	595.70
Temperature/K	298.27	298.(2)
Crystal system	orthorhombic	orthorhombic
Space group	P2 ₁ 2 ₁ 2 ₁	Pbca
a/Å	9.9849(8)	9.0957(2)
b/Å	10.6568(9)	18.8991(6)
c/Å	26.012(2)	36.9438(11)
$\alpha/^\circ$	90	90
$\beta/^\circ$	90	90
$\gamma/^\circ$	90	90
Volume/Å ³	2767.9(4)	6350.7(3)
Z	4	8
$\rho_{\text{calc}}/\text{g/cm}^3$	1.247	1.246

μ/mm^{-1}	0.072	0.071
F(000)	1088.0	2496.0
Crystal size/ mm^3	$0.45 \times 0.33 \times 0.25$	$0.450 \times 0.200 \times 0.120$
Radiation	$\text{MoK}\alpha$ ($\lambda = 0.71073$)	$\text{MoK}\alpha$ ($\lambda = 0.71073$)
2θ range for data collection/ $^\circ$	4.37 to 57.378	5 to 50.8
Index ranges	$-11 \leq h \leq 13, -14 \leq k \leq 14, -34 \leq l \leq 35$	$-10 \leq h \leq 10, -22 \leq k \leq 22, -44 \leq l \leq 44$
Reflections collected	26512	41276
Independent reflections	7140 [$R_{\text{int}} = 0.0526, R_{\text{sigma}} = 0.0595$]	5811 [$R_{\text{int}} = 0.0526, R_{\text{sigma}} = 0.0315$]
Data/restraints/parameters	7140/0/375	5811/0/425
Goodness-of-fit on F^2	1.032	1.031
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0508, wR_2 = 0.1014$	$R_1 = 0.0534, wR_2 = 0.1283$
Final R indexes [all data]	$R_1 = 0.0851, wR_2 = 0.1125$	$R_1 = 0.0802, wR_2 = 0.1452$
Largest diff. peak/hole / $e \text{ \AA}^{-3}$	0.22/-0.15	0.22/-0.23

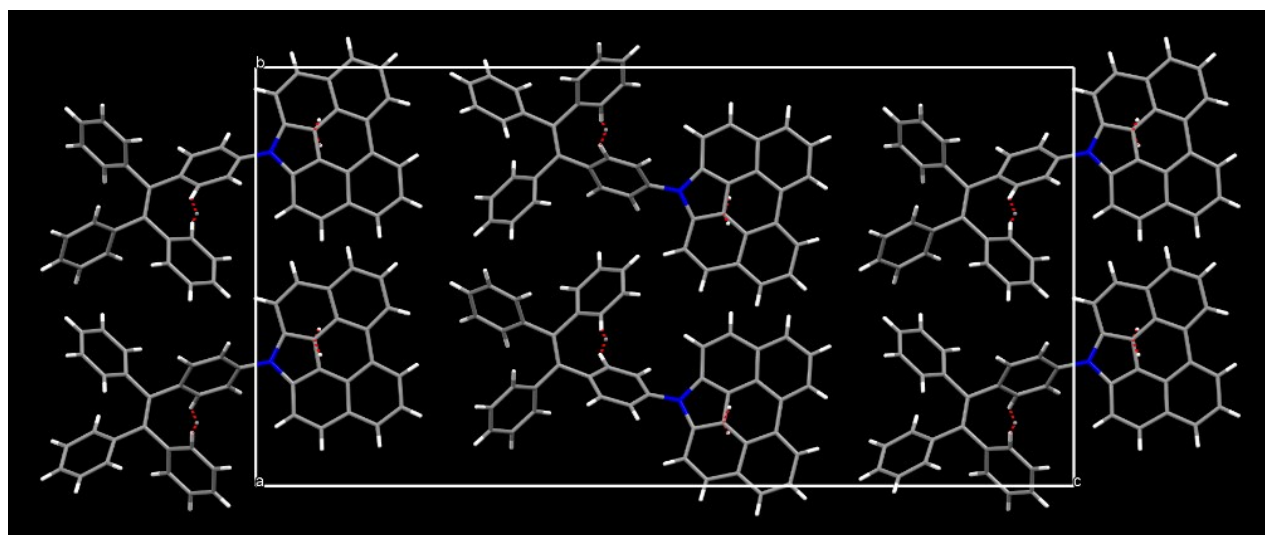


PCTrPE

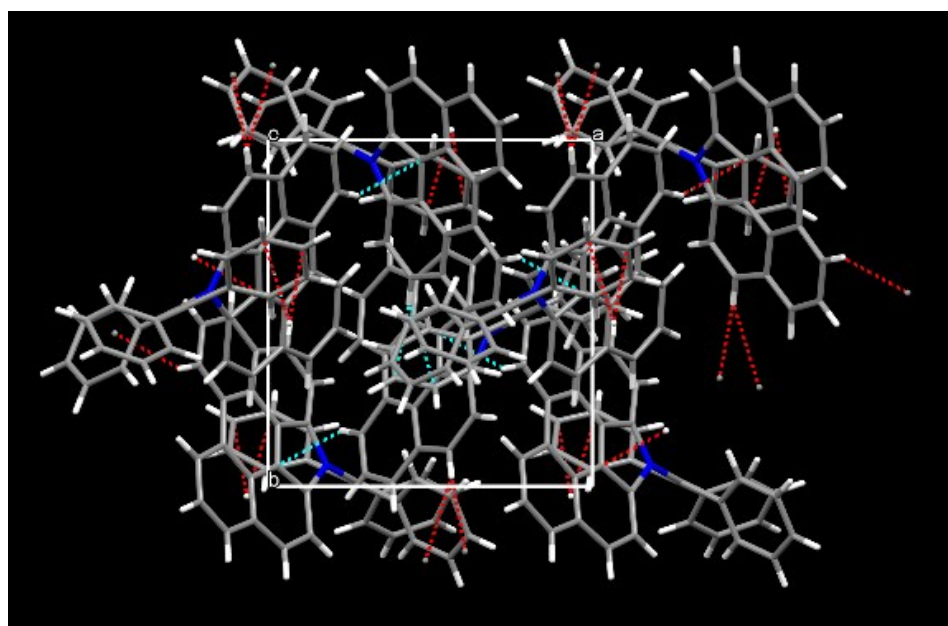


PCTPE

Figure S1. The dihedral angles (ϕ) in the molecules of **PCTrPE** and **PCTPE**.



PCTPE



PCTrPE

Figure S2. Crystal packing along *a*-axis of **PCTPE** and along *c*-axis of **PCTrPE**.

II Additional Spectra and Data

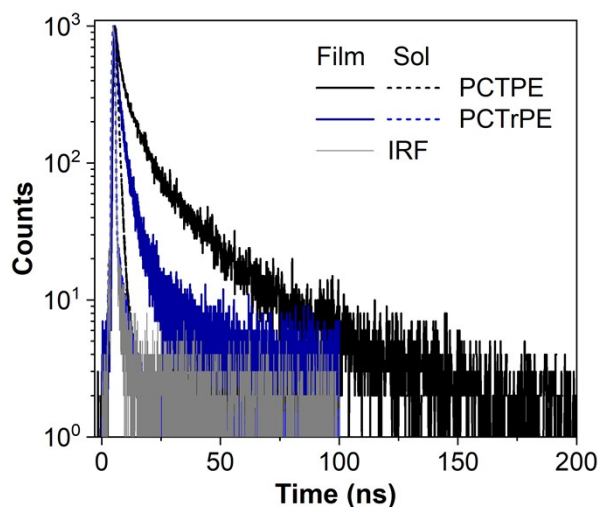


Figure S3. Transient PL spectra of **PCTPE** and **PCTrPE** in toluene solution and thin films coated on fused silica substrates.

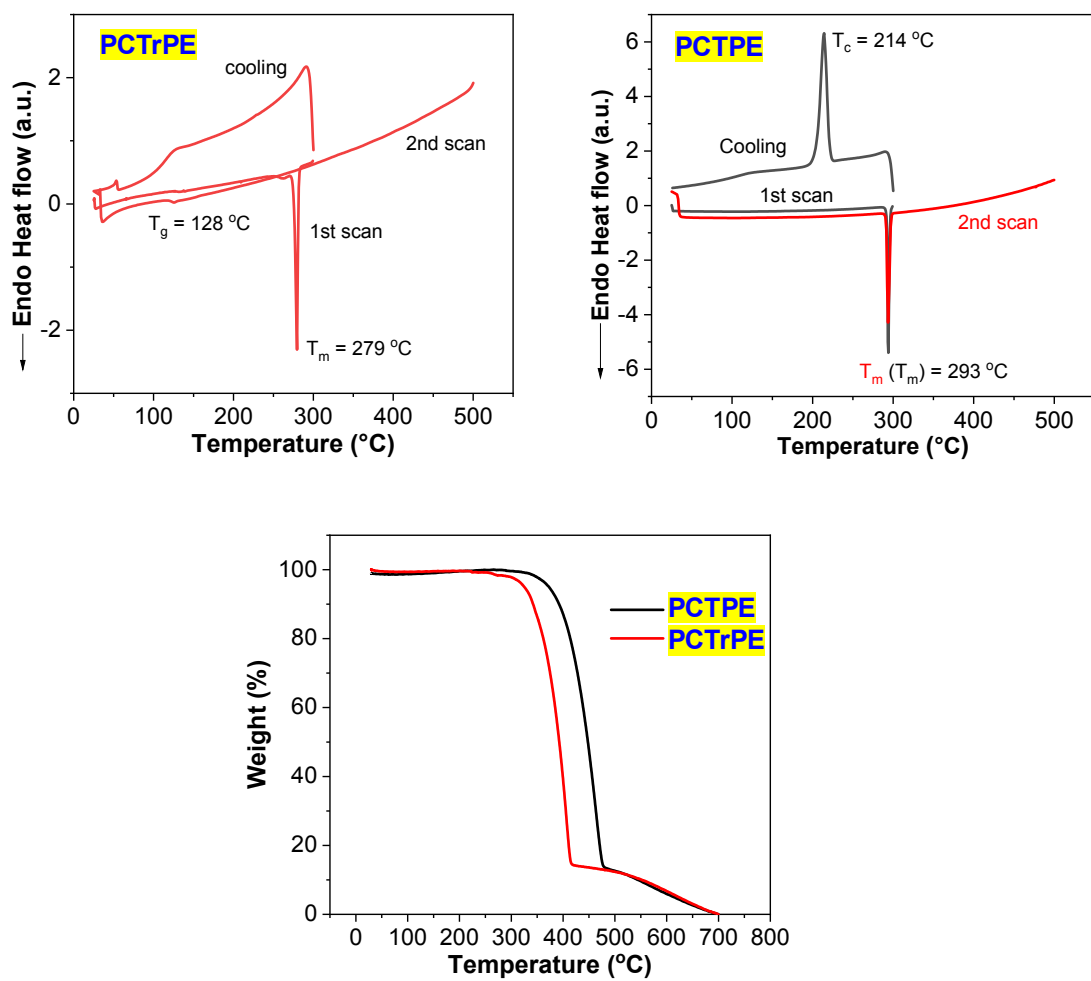


Figure S4. (Top) DSC (1st heating (thin line), cooling (dot line) and 2nd heating scans (thick line)) and (Bottom) TGA traces of **PCTPE** and **PCTrPE** analysed at heating rate of 10 °C min^{-1} under N_2 flow.

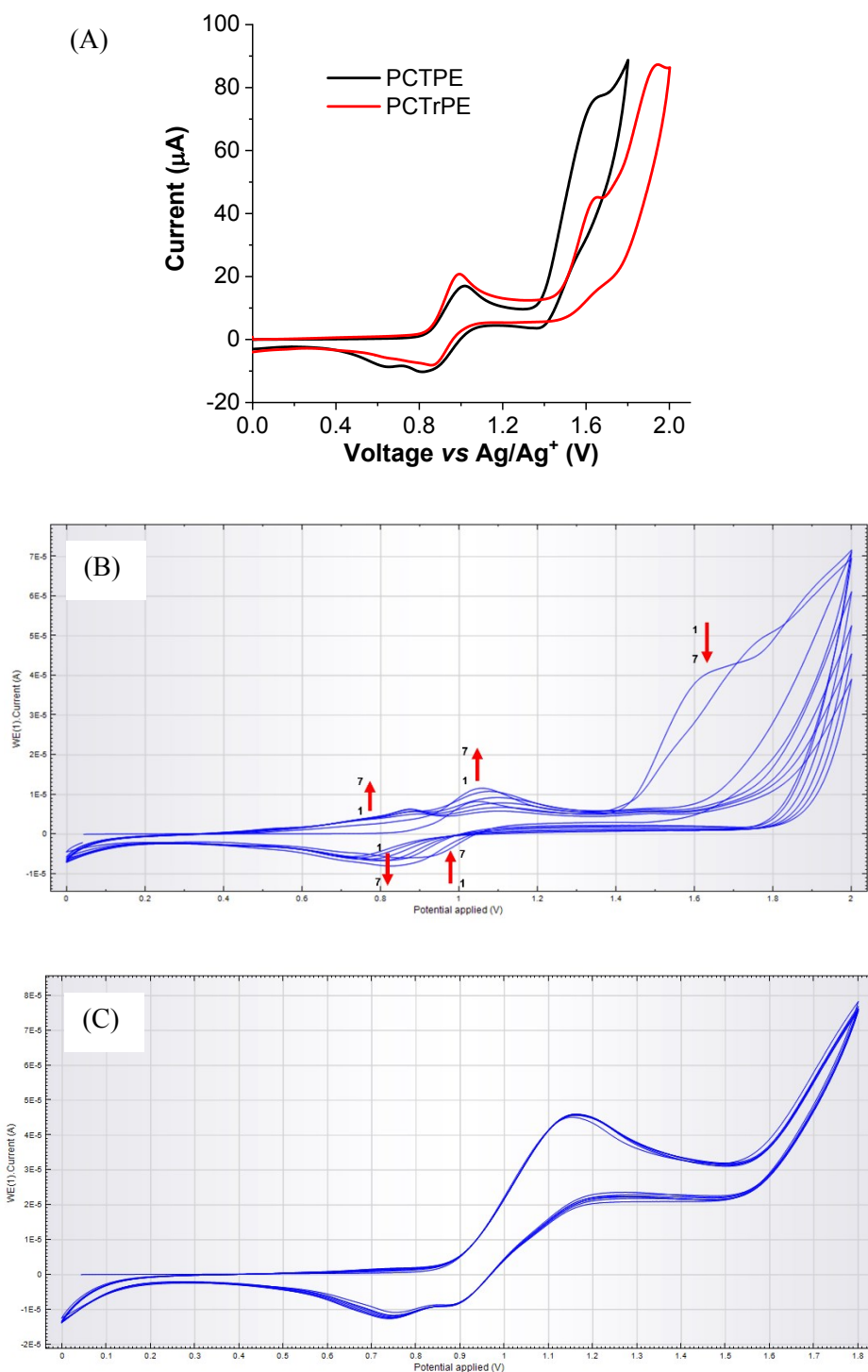


Figure S5. (A) CV plots of **PCTPE** and **PCTrPE** and (B and C) 7-scan CV plots of **PCTPE** and **PCTrPE** , respectively, measured in CH_2Cl_2 containing $n\text{-Bu}_4\text{NPF}_6$ as a supporting electrolyte at a scan rate of 50 mV s^{-1} under Ar flow. The red arrows on figure (B) showed the increasing and decreasing of working electrode's current from the 1st to 7th scans.

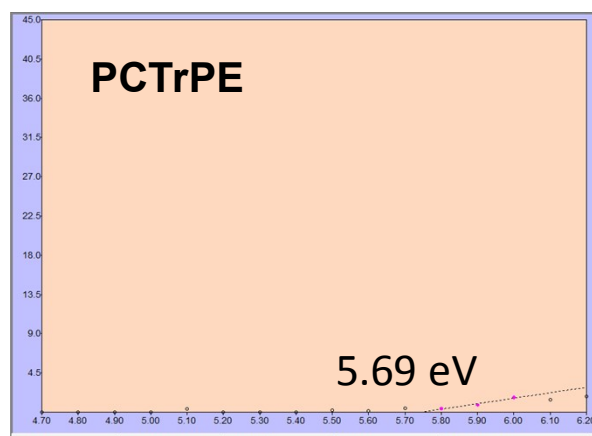
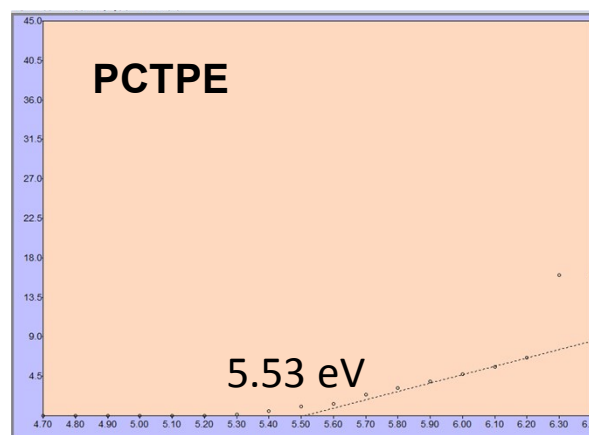


Figure S6 Photoemission yield spectroscopy in air (AC-2) spectra of **PCTPE** and **PCTrPE**.

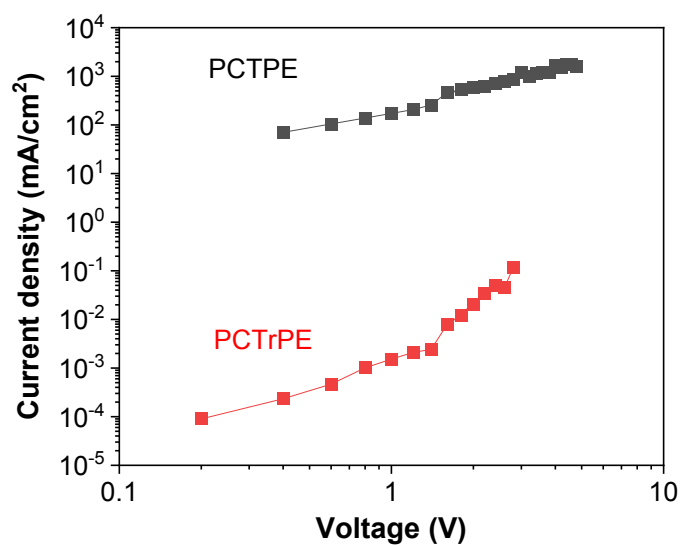


Figure S7. Current density-voltage (J-V) plot of the hole only devices: (ITO/PEDOT:PSS/**PCTPE** or **PCTrPE** (170 nm)/MoO₃/Al).

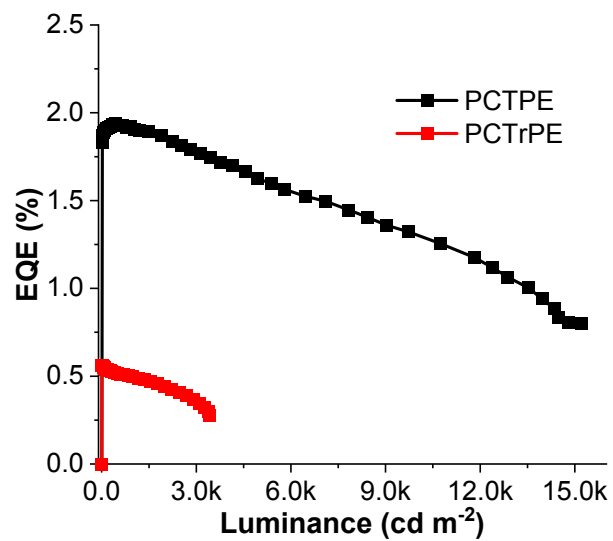
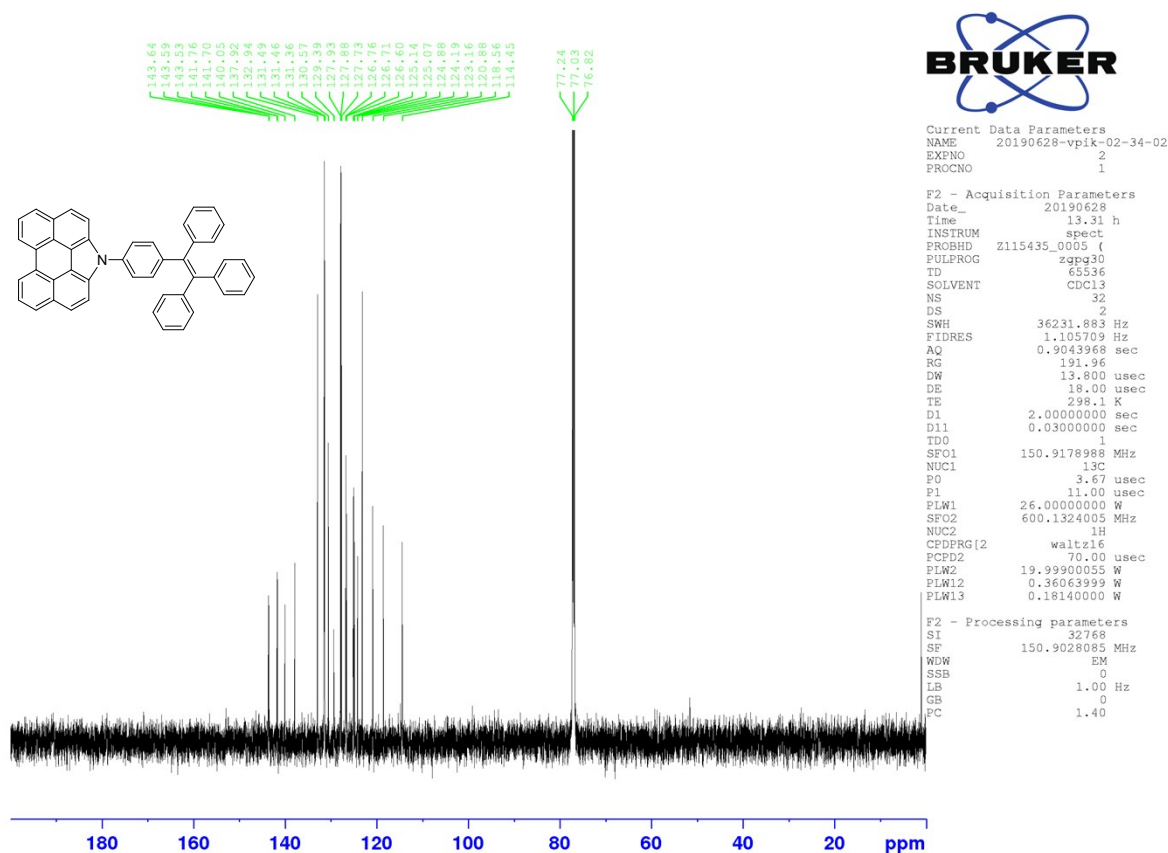
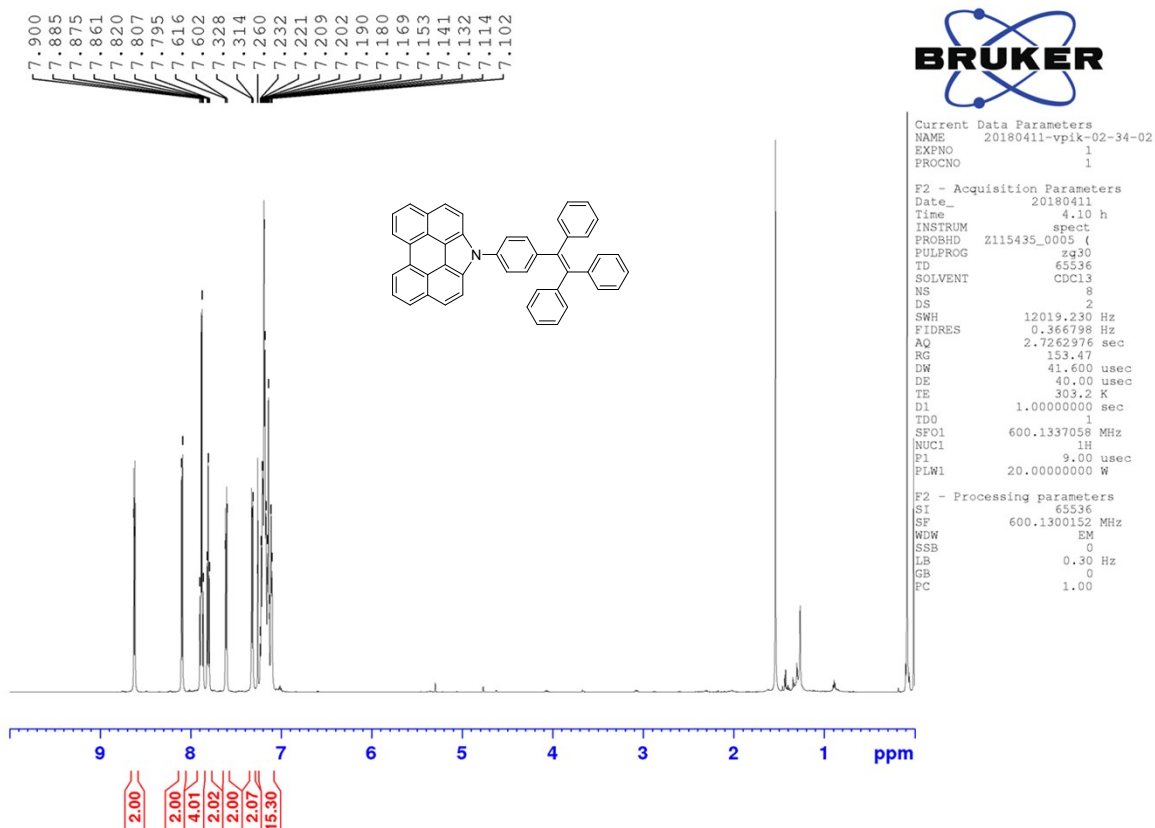
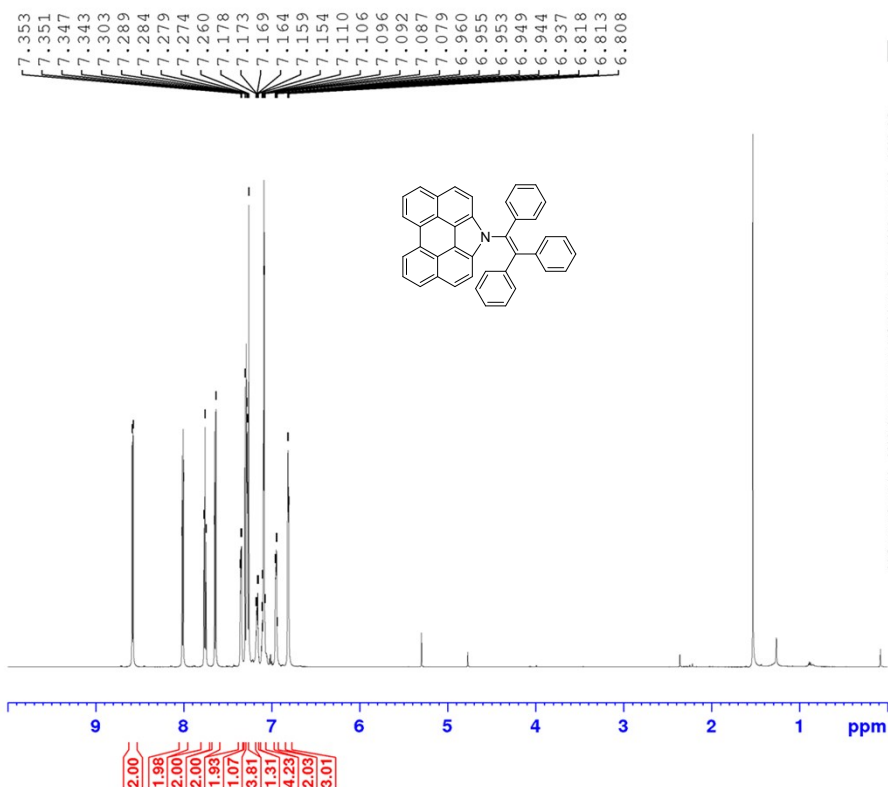


Figure S8. EQE-luminance plots of the fabricated OLEDs.

III Copies of NMR spectra and mass spectra

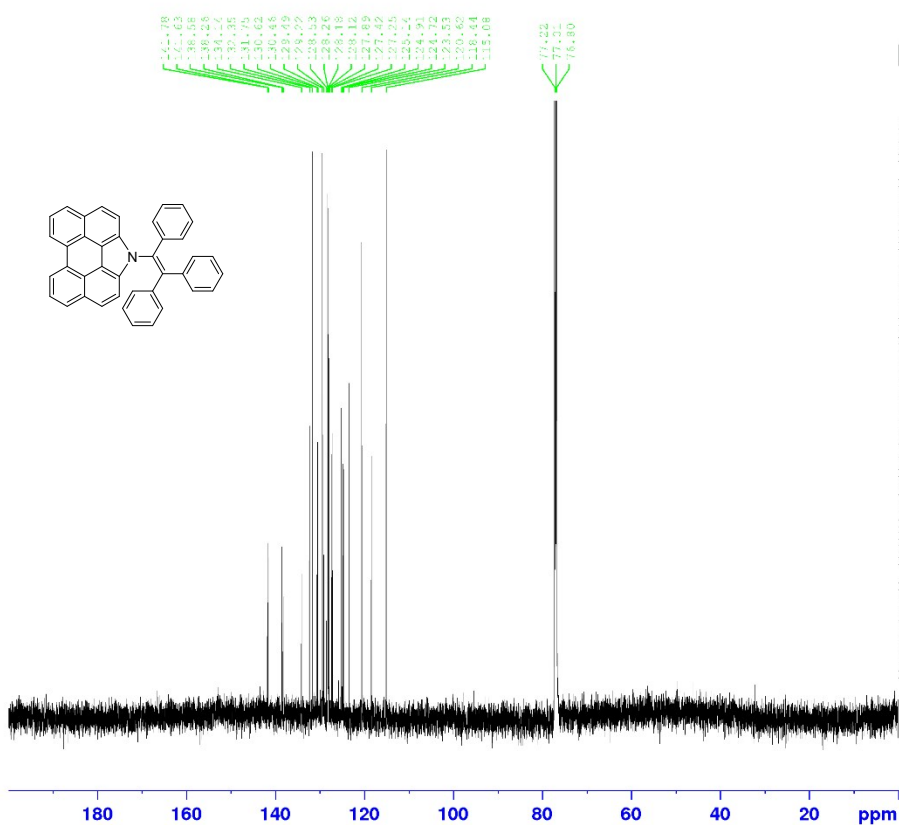


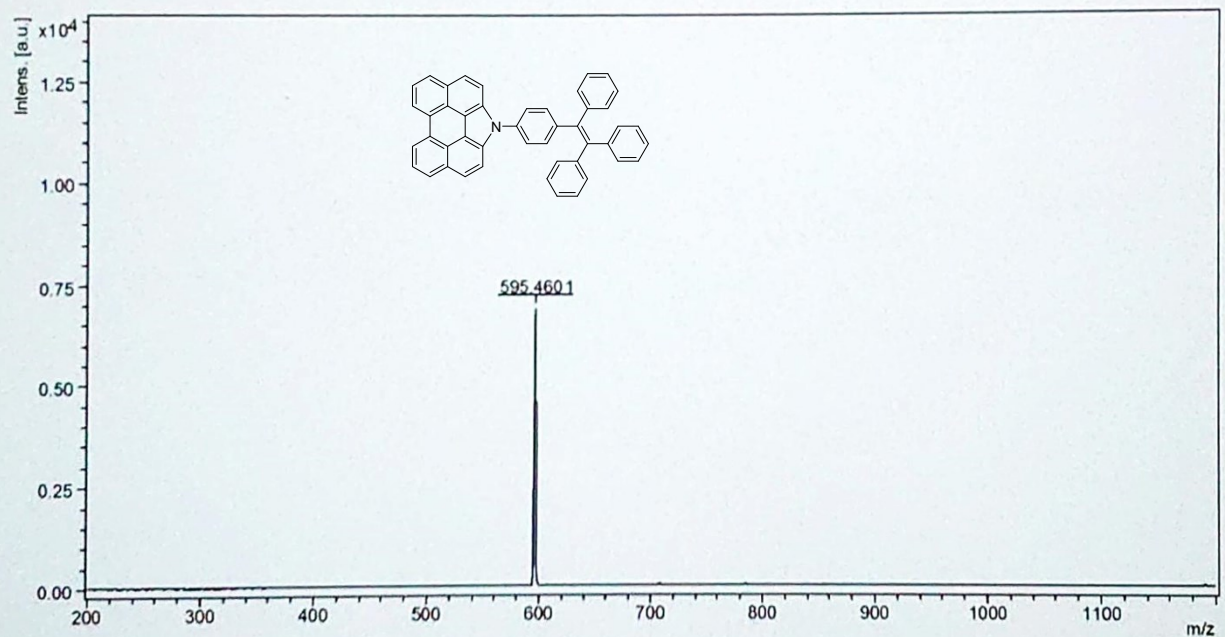


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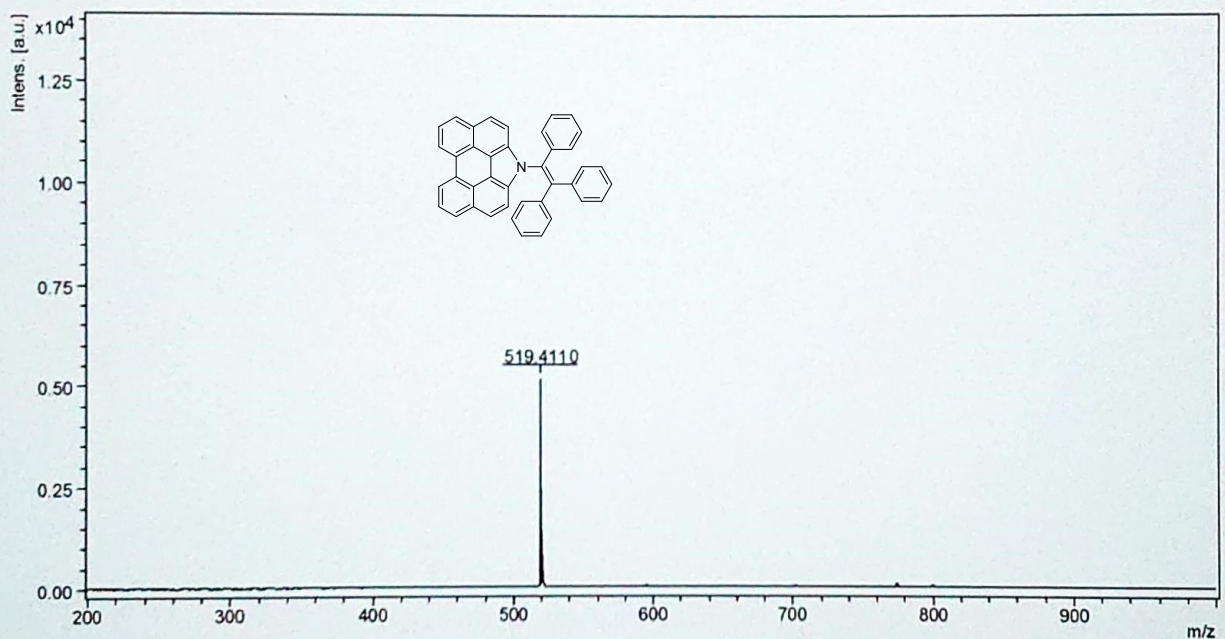




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