

Electronic Supplementary Information

Palladium-catalyzed synthesis of [60]fullerene-fused furochromenones and further electrochemical functionalization

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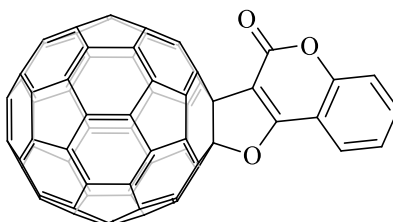
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1. General procedure for the synthesis of compounds 2a–l

A mixture of C₆₀ (0.05 mmol), **1** (0.15 mmol), Pd(OAc)₂ (0.005 mmol), Et₃N (0.15 mmol) and K₂S₂O₈ (0.15 mmol) was dissolved in *ortho*-dichlorobenzene (ODCB) (3 mL). Then the solution was vigorously stirred at 140 °C in a sealed tube and stopped at the designated time. The resulting solution was evaporated in *vacuo* and then separated on a silica gel column with CS₂/CH₂Cl₂ (4:1 v/v) as the eluent to give recovered C₆₀ and the then desired product **2**.

2. Synthesis and spectral data of compounds 2a–l

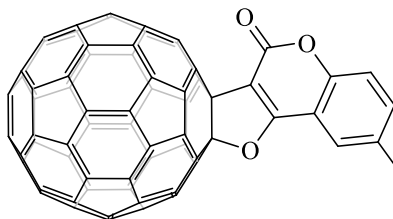
Synthesis and spectral data of compounds 2a



2a

By following the general procedure, the reaction of C₆₀ (36.1 mg, 0.05 mmol) with **1a** (24.4 mg, 0.15 mmol), Pd(OAc)₂ (1.2 mg, 0.005 mmol), Et₃N (21 μL, 0.15 mmol) and K₂S₂O₈ (40.3 mg, 0.15 mmol) at 140 °C for 3 h afforded recovered C₆₀ (20.9 mg, 58%) and **2a** (16.7 mg, 38%): amorphous brown solid; ¹H NMR (400 MHz, CDCl₂CDCl₂) δ 8.18 (dd, *J*₁ = 7.8 Hz, *J*₂ = 1.4 Hz, 1H), 7.77 (ddd, *J*₁ = 8.7 Hz, *J*₂ = 7.4 Hz, *J*₃ = 1.5 Hz, 1H), 7.58 (d, *J* = 8.2 Hz, 1H), 7.52 (td, *J*₁ = 7.7 Hz, *J*₂ = 0.8 Hz, 1H); ¹³C NMR (101 MHz, CDCl₂CDCl₂ with Cr(acac)₃ as the relaxation reagent) (all 2C unless indicated) δ 165.16 (1C, C=O), 158.24 (1C, O-C=C-CO₂), 154.16 (1C, aryl C), 147.10 (1C), 146.25 (1C), 145.75, 145.45, 145.35, 145.12, 145.10, 144.98 (4C), 144.51, 144.35, 144.17, 144.04, 143.54, 143.40, 143.02, 141.73, 141.69, 141.65, 141.57, 141.41, 141.28, 141.23, 141.19, 140.71, 140.21, 139.30, 138.78, 136.96, 134.59, 133.06 (1C, aryl C), 123.87 (1C, aryl C), 123.06 (1C, aryl C), 116.35 (1C, aryl C), 111.29 (1C, aryl C), 105.56 (1C, sp³-C of C₆₀), 102.80 (1C, O-C=C-CO₂), 68.22 (1C, sp³-C of C₆₀); FT-IR ν/cm⁻¹ (KBr) 1725, 1645, 1507, 1466, 1392, 1095, 1024, 883, 742, 550, 524; UV-vis (CHCl₃) λ_{max}/nm (log ε) 252 (5.02), 319 (4.64), 416 (3.36), 463 (3.06); MALDI-TOF MS *m/z* calcd for C₆₉H₄O₃ [M]⁺ 880.0160, found 880.0171.

Synthesis and spectral data of compound 2b

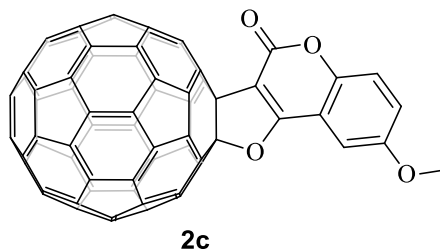


2b

By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1b** (26.4 mg, 0.15 mmol), Pd(OAc)₂ (1.1 mg, 0.005 mmol), Et₃N (21 μL, 0.15 mmol) and K₂S₂O₈ (40.3 mg, 0.15 mmol) at 140 °C for 3 h afforded recovered C₆₀ (22.6 mg, 63%) and **2b** (14.1 mg, 32%):

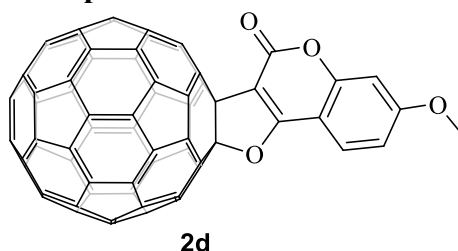
amorphous brown solid; ^1H NMR (400 MHz, $\text{CS}_2/\text{CDCl}_2/\text{CDCl}_2$) δ 7.96 (d, $J = 1.8$ Hz, 1H), 7.57 (dd, $J_1 = 8.6$ Hz, $J_2 = 1.8$ Hz, 1H), 7.47 (d, $J = 8.6$ Hz, 1H), 2.54 (s, 3H); ^{13}C NMR (101 MHz, $\text{CS}_2/\text{CDCl}_2/\text{CDCl}_2$ with $\text{Cr}(\text{acac})_3$ as the relaxation reagent) (all 2C unless indicated) δ 164.87 (1C, $\text{C}=\text{O}$), 157.97 (1C, $\text{O}-\text{C}=\text{C}-\text{CO}_2$), 152.40 (1C, aryl C), 147.11 (1C), 146.24 (1C), 145.88, 145.46, 145.35, 145.24, 145.10, 144.99 (4C), 144.55, 144.33, 144.18, 144.05, 143.52, 143.46, 143.02, 141.77, 141.73, 141.67, 141.64, 141.45, 141.34, 141.25, 141.21, 140.77, 140.27, 139.32, 138.80, 136.94, 134.50, 134.02 (1C, aryl C), 133.72 (1C, aryl C), 122.60 (1C, aryl C), 116.15 (1C, aryl C), 111.07 (1C, aryl C), 105.40 (1C, $\text{sp}^3\text{-C}$ of C_{60}), 102.68 (1C, $\text{O}-\text{C}=\text{C}-\text{CO}_2$), 68.29 (1C, $\text{sp}^3\text{-C}$ of C_{60}), 20.12 (CH_3); FT-IR ν/cm^{-1} (KBr) 2923, 2849, 1725, 1642, 1466, 1371, 1260, 1092, 1018, 933, 883, 844, 794, 550, 524; UV-vis (CHCl_3) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 256 (5.06), 319 (4.68), 416 (3.26), 456 (2.78); MALDI-TOF MS m/z calcd for $\text{C}_{70}\text{H}_6\text{O}_3$ $[\text{M}]^-$ 894.0317, found 894.0318.

Synthesis and spectral data of compound 2c



By following the general procedure, the reaction of C_{60} (36.2 mg, 0.05 mmol) with **1c** (28.9 mg, 0.15 mmol), $\text{Pd}(\text{OAc})_2$ (1.1 mg, 0.005 mmol), Et_3N (21 μL , 0.15 mmol) and $\text{K}_2\text{S}_2\text{O}_8$ (40.4 mg, 0.15 mmol) at 140 $^\circ\text{C}$ for 3 h afforded recovered C_{60} (21.1 mg, 58%) and **2c** (14.9 mg, 33%): amorphous brown solid; ^1H NMR (400 MHz, $\text{CDCl}_2/\text{CDCl}_2$) δ 7.54 (d, $J = 3.0$ Hz, 1H), 7.51 (d, $J = 9.2$ Hz, 1H), 7.34 (dd, $J_1 = 9.2$ Hz, $J_2 = 3.0$ Hz, 1H), 3.96 (s, 3H); ^{13}C NMR (101 MHz, $\text{CDCl}_2/\text{CDCl}_2$ with $\text{Cr}(\text{acac})_3$ as the relaxation reagent) (all 2C unless indicated) δ 164.93 (1C, $\text{C}=\text{O}$), 158.35 (1C, $\text{O}-\text{C}=\text{C}-\text{CO}_2$), 155.28 (1C, aryl C), 148.76 (1C, aryl C), 147.09 (1C), 146.23 (1C), 145.76, 145.44, 145.34, 145.13, 145.08, 144.97 (4C), 144.49, 144.34, 144.16, 144.03, 143.52, 143.40, 143.00, 141.72, 141.69, 141.64, 141.56, 141.39, 141.27, 141.21, 141.17, 140.69, 140.20, 139.28, 138.76, 136.97, 134.58, 121.75 (1C, aryl C), 117.66 (1C, aryl C), 111.50 (1C, aryl C), 105.49 (1C, $\text{sp}^3\text{-C}$ of C_{60}), 104.04 (1C, aryl C), 102.92 (1C, $\text{O}-\text{C}=\text{C}-\text{CO}_2$), 68.29 (1C, $\text{sp}^3\text{-C}$ of C_{60}), 55.29 (CH_3); FT-IR ν/cm^{-1} (KBr) 2917, 2846, 1728, 1654, 1636, 1510, 1466, 1257, 1092, 1018, 886, 800, 547, 524; UV-vis (CHCl_3) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 255 (5.04), 317 (4.66), 413 (3.33), 453 (2.78); MALDI-TOF MS m/z calcd for $\text{C}_{70}\text{H}_6\text{O}_4$ $[\text{M}]^-$ 910.0266, found 910.0266.

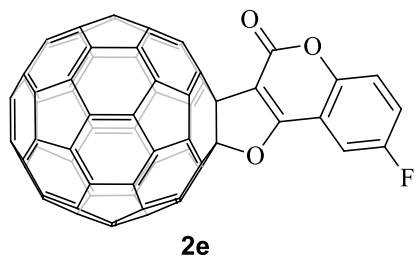
Synthesis and spectral data of compound 2d



By following the general procedure, the reaction of C_{60} (36.0 mg, 0.05 mmol) with **1d** (28.9 mg, 0.15 mmol), $\text{Pd}(\text{OAc})_2$ (1.1 mg, 0.005 mmol), Et_3N (21 μL , 0.15 mmol) and $\text{K}_2\text{S}_2\text{O}_8$ (40.4, 0.15

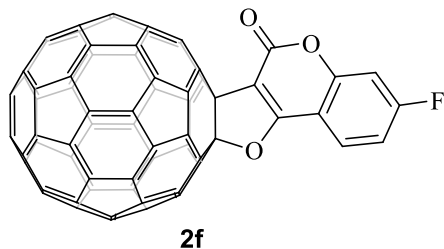
mmol) at 140 °C for 3 h afforded recovered C₆₀ (21.9 mg, 61%) and **2d** (14.4 mg, 32%): amorphous brown solid; ¹H NMR (400 MHz, CDCl₂CDCl₂) δ 8.06 (d, *J* = 8.6 Hz, 1H), 7.10-7.03 (m, 2H), 3.95 (s, 3H); ¹³C NMR (101 MHz, CDCl₂CDCl₂ with Cr(acac)₃ as the relaxation reagent) (all 2C unless indicated) δ 165.39 (1C, C=O), 163.51 (1C, aryl C), 158.59 (1C, O-C=C-CO₂), 156.26 (1C, aryl C), 147.09 (1C), 146.25 (1C), 146.16, 145.44, 145.34, 145.07 (4C), 144.97 (4C), 144.47, 144.36, 144.16, 144.01, 143.56, 143.42, 143.02, 141.80, 141.72, 141.67, 141.63, 141.46, 141.27, 141.22, 141.18, 140.71, 140.21, 139.28, 138.74, 136.99, 134.51, 124.09 (1C, aryl C), 112.41 (1C, aryl C), 105.46 (1C, sp³-C of C₆₀), 104.34 (1C, aryl C), 100.18 (1C, aryl C), 100.03 (1C, O-C=C-CO₂), 68.11 (1C, sp³-C of C₆₀), 55.26 (CH₃); FT-IR ν/cm⁻¹ (KBr) 2917, 2846, 1722, 1654, 1636, 1613, 1510, 1466, 1404, 1274, 1260, 1186, 1154, 1109, 1095, 1021, 892, 833, 818, 794, 550, 527; UV-vis (CHCl₃) λ_{max}/nm (log ε) 256 (5.41), 320 (5.03), 417 (3.41), 465 (2.58); MALDI-TOF MS *m/z* calcd for C₇₀H₆O₄ [M]⁻ 910.0266, found 910.0268.

Synthesis and spectral data of compound **2e**



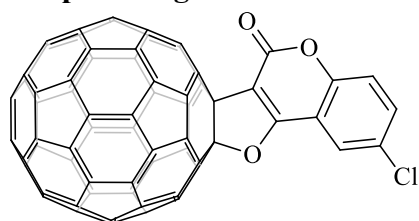
By following the general procedure, the reaction of C₆₀ (36.1 mg, 0.05 mmol) with **1e** (27.0 mg, 0.15 mmol), Pd(OAc)₂ (1.1 mg, 0.005 mmol), Et₃N (21 μL, 0.15 mmol) and K₂S₂O₈ (40.5 mg, 0.15 mmol) at 140 °C for 3.5 h afforded recovered C₆₀ (15.1 mg, 42%) and **2e** (20.1 mg, 45%): amorphous brown solid; ¹H NMR (400 MHz, CS₂ with DMSO-*d*₆ as the external lock solvent) δ 7.79–7.71 (m, 1H), 7.58–7.42 (m, 2H); ¹³C NMR (101 MHz, CS₂ with DMSO-*d*₆ as the external lock solvent and Cr(acac)₃ as the relaxation reagent) (all 2C unless indicated) δ 163.22 (1C, C=O), 157.54 (d, *J* = 247.3 Hz) (1C, aryl C), 155.45 (1C, O-C=C-CO₂), 150.52 (1C, aryl C), 147.09 (1C), 146.20 (1C), 145.62, 145.57, 145.46, 145.36, 145.10, 145.00, 144.99, 144.65, 144.28, 144.18, 144.05, 143.49, 143.47, 142.98, 141.76 (4C), 141.68, 141.40 (6C), 141.26, 141.21, 140.83, 140.29, 139.29, 138.82, 136.89, 134.39, 120.03 (d, *J* = 23.2 Hz) (1C, aryl C), 118.18 (d, *J* = 7.4 Hz), (1C, aryl C), 112.27 (d, *J* = 9.0 Hz) (1C, aryl C), 108.38 (d, *J* = 24.8 Hz) (1C, aryl C), 105.30 (1C, sp³-C of C₆₀), 103.67 (1C, O-C=C-CO₂), 68.36 (1C, sp³-C of C₆₀); FT-IR ν/cm⁻¹ (KBr) 1736, 1651, 1630, 1507, 1466, 1260, 1089, 1024, 930, 886, 821, 794, 553, 527; UV-vis (CHCl₃) λ_{max}/nm (log ε) 255 (5.17), 317 (4.62), 417 (3.53), 459 (3.33); MALDI-TOF MS *m/z* calcd for C₆₉H₃FO₃ [M]⁻ 898.0066, found 898.0075.

Synthesis and spectral data of compound **2f**



By following the general procedure, the reaction of C₆₀ (36.2 mg, 0.05 mmol) with **1f** (26.9 mg, 0.15 mmol), Pd(OAc)₂ (1.1 mg, 0.005 mmol), Et₃N (21 μ L, 0.15 mmol) and K₂S₂O₈ (40.5 mg, 0.15 mmol) at 140 °C for 3 h afforded recovered C₆₀ (21.2 mg, 59%) and **2f** (13.9 mg, 31%): amorphous brown solid; ¹H NMR (400 MHz, CS₂ with DMSO-*d*₆ as the external lock solvent) δ 7.78–7.73 (m, 1H), 7.57–7.43 (m, 2H); ¹³C NMR (101 MHz, CS₂/CDCl₂CDCl₂ with Cr(acac)₃ as the relaxation reagent) (all 2C unless indicated) δ 164.85 (d, *J* = 257.0 Hz) (1C, aryl C), 164.35 (1C, C=O), 157.07 (1C, O-C=C-CO₂), 155.70 (d, *J* = 13.2 Hz) (1C, aryl C), 147.19 (1C), 146.29 (1C), 145.78, 145.54, 145.43, 145.35, 145.16, 145.08, 145.06, 144.69, 144.35, 144.25, 144.12, 143.54, 143.49, 143.05, 141.82 (4C), 141.74, 141.48 (4C), 141.44, 141.31, 141.26, 140.88, 140.31, 139.48, 138.86, 137.02, 134.47, 124.65 (d, *J* = 10.5 Hz) (1C, aryl C), 112.00 (d, *J* = 23.1 Hz) (1C, aryl C), 108.15 (1C, aryl C), 105.62 (1C, sp³-C of C₆₀), 104.16 (d, *J* = 25.5 Hz) (1C, aryl C), 101.91 (1C, O-C=C-CO₂), 68.18 (1C, sp³-C of C₆₀); FT-IR ν /cm⁻¹ (KBr) 1739, 1645, 1513, 1463, 1392, 1260, 1136, 1101, 1021, 892, 809, 550, 524; UV-vis (CHCl₃) λ_{max} /nm (log ϵ) 256 (5.12), 318 (4.77), 417 (3.69), 462 (3.51); MALDI-TOF MS *m/z* calcd for C₆₉H₃FO₃ [M]⁻ 898.0066, found 898.0077.

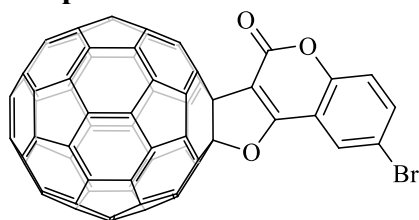
Synthesis and spectral data of compound **2g**



2g

By following the general procedure, the reaction of C₆₀ (36.1 mg, 0.05 mmol) with **1g** (29.5, 0.15 mmol), Pd(OAc)₂ (1.2 mg, 0.005 mmol), Et₃N (21 μ L, 0.15 mmol) and K₂S₂O₈ (40.5 mg, 0.15 mmol) at 140 °C for 3 h afforded recovered C₆₀ (19.0 mg, 53%) and **2g** (14.9 mg, 33%): amorphous brown solid; ¹H NMR (400 MHz, CS₂/CDCl₂CDCl₂) δ 8.16 (d, *J* = 2.5 Hz, 1H), 7.72 (dd, *J*₁ = 8.9 Hz, *J*₂ = 2.5 Hz, 1H), 7.53 (d, *J* = 8.9 Hz, 1H); ¹³C NMR (101 MHz, CS₂/CDCl₂CDCl₂ with Cr(acac)₃ as the relaxation reagent) (all 2C unless indicated) δ 163.89 (1C, C=O), 157.15 (1C, O-C=C-CO₂), 152.51 (1C, aryl C), 147.13 (1C), 146.27 (1C), 145.48, 145.41, 145.37, 145.15 (4C), 145.03 (4C), 144.61, 144.29, 144.21, 144.08, 143.43, 143.03 (4C), 141.78 (4C), 141.71, 141.40, 141.35, 141.28, 141.24 (4C), 140.78, 140.23, 139.36, 138.86, 136.92, 134.56, 132.86 (1C, aryl C), 129.44 (1C, aryl C), 122.40 (1C, aryl C), 117.86 (1C, aryl C), 112.50 (1C, aryl C), 105.72 (1C, sp³-C of C₆₀), 103.68 (1C, O-C=C-CO₂), 68.14 (1C, sp³-C of C₆₀); FT-IR ν /cm⁻¹ (KBr) 1733, 1648, 1630, 1507, 1469, 1374, 1254, 1104, 1027, 886, 818, 794, 721, 700, 547, 524; UV-vis (CHCl₃) λ_{max} /nm (log ϵ) 258 (4.94), 319 (4.56), 417 (3.08), 458 (2.60); MALDI-TOF MS *m/z* calcd for C₆₉H₃³⁵ClO₃ [M]⁻ 913.9771, found 913.9771.

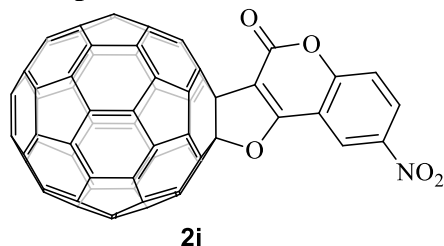
Synthesis and spectral data of compound **2h**



2h

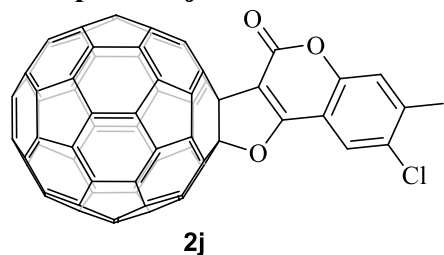
By following the general procedure, the reaction of C₆₀ (36.2 mg, 0.05 mmol) with **1h** (36.3 mg, 0.15 mmol), Pd(OAc)₂ (1.1 mg, 0.005 mmol), Et₃N (21 μL, 0.15 mmol) and K₂S₂O₈ (40.6 mg, 0.15 mmol) at 140 °C for 3.5 h afforded recovered C₆₀ (12.9 mg, 36%) and **2h** (24.9 mg, 52%): amorphous brown solid; ¹H NMR (400 MHz, CDCl₂CDCl₂) δ 8.30 (d, *J* = 2.3 Hz, 1H), 7.85 (dd, *J*₁ = 8.9 Hz, *J*₂ = 2.3 Hz, 1H), 7.47 (d, *J* = 8.9 Hz, 1H); FT-IR ν/cm⁻¹ (KBr) 1728, 1642, 1507, 1469, 1419, 1374, 1265, 1104, 1015, 900, 880, 794, 727, 703, 550, 521; UV-vis (CHCl₃) λ_{max}/nm (log ε) 256 (5.01), 318 (4.61), 417 (3.27), 462 (2.96); MALDI-TOF MS *m/z* calcd for C₆₀H₃⁷⁹BrO₃ [M]⁻ 957.9266, found 957.9269.

Synthesis and spectral data of compound **2i**



By following the general procedure, the reaction of C₆₀ (35.9 mg, 0.05 mmol) with **1i** (31.1 mg, 0.15 mmol), Pd(OAc)₂ (1.1 mg, 0.005 mmol), Et₃N (21 μL, 0.15 mmol) and K₂S₂O₈ (40.4 mg, 0.15 mmol) at 140 °C for 3 h afforded recovered C₆₀ (12.1 mg, 34%) and **2i** (21.8 mg, 47%): amorphous brown solid; ¹H NMR (400 MHz, CS₂ with DMSO-*d*₆ as the external lock solvent) δ 9.01 (s, 1H), 8.61 (d, *J* = 9.0 Hz, 1H), 7.70 (d, *J* = 9.0 Hz, 1H); FT-IR ν/cm⁻¹ (KBr) 1745, 1648, 1630, 1530, 1492, 1466, 1383, 1330, 1260, 1112, 1024, 885, 847, 797, 723, 559, 523; UV-vis (CHCl₃) λ_{max}/nm (log ε) 252 (5.12), 316 (4.73), 412 (3.53), 458 (3.24); MALDI-TOF MS *m/z* calcd for C₆₉H₃NO₅ [M]⁻ 925.0011, found 925.0015.

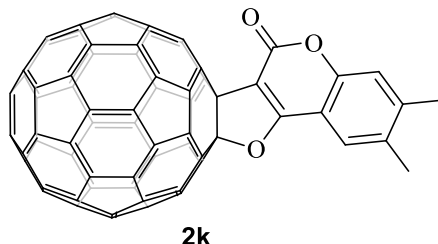
Synthesis and spectral data of compound **2j**



By following the general procedure, the reaction of C₆₀ (36.0 mg, 0.05 mmol) with **1j** (31.5 mg, 0.15 mmol), Pd(OAc)₂ (1.2 mg, 0.005 mmol), Et₃N (21 μL, 0.15 mmol) and K₂S₂O₈ (40.4 mg, 0.15 mmol) at 140 °C for 2.5 h afforded recovered C₆₀ (20.0 mg, 56%) and **2j** (16.4 mg, 35%): amorphous brown solid; ¹H NMR (400 MHz, CS₂/CDCl₂CDCl₂) δ 8.11 (s, 1H), 7.46 (s, 1H), 2.59 (s, 3H); ¹³C NMR (101 MHz, CS₂/CDCl₂CDCl₂ with Cr(acac)₃ as the relaxation reagent) (all 2C unless indicated) δ 164.01 (1C, C=O), 157.41 (1C, O-C=C-CO₂), 152.51 (1C, aryl C), 147.13 (1C), 146.27 (1C), 145.61, 145.48, 145.40, 145.18, 145.13, 145.02 (4C), 144.59, 144.31, 144.21, 144.07, 143.47, 143.45, 143.03, 142.19 (1C, aryl C), 141.78, 141.76, 141.70, 141.43, 141.40, 141.34, 141.28, 141.23, 140.78, 140.25, 139.34, 138.84, 136.94, 134.53, 129.94 (1C, aryl C), 122.55 (1C, aryl C), 118.34 (1C, aryl C), 110.31 (1C, aryl C), 105.66 (1C, sp³-C of C₆₀), 102.82 (1C, O-C=C-CO₂), 68.16 (1C, sp³-C of C₆₀), 20.20 (CH₃); FT-IR ν/cm⁻¹ (KBr) 2917, 2852, 1731, 1648, 1466, 1421, 1374, 1260, 1174, 1106, 1030, 886, 845, 821, 736, 550, 524; UV-vis (CHCl₃) λ_{max}/nm (log ε)

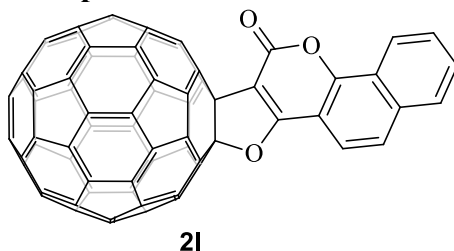
ϵ) 257 (5.01), 319 (4.63), 417 (3.47), 455 (3.27); MALDI-TOF MS m/z calcd for $C_{70}H_5^{35}ClO_3 [M]^-$ 927.9927, found 927.9929.

Synthesis and spectral data of compound 2k



By following the general procedure, the reaction of C_{60} (36.0 mg, 0.05 mmol) with **1k** (28.5 mg, 0.15 mmol), $Pd(OAc)_2$ (1.1 mg, 0.005 mmol), Et_3N (21 μ L, 0.15 mmol) and $K_2S_2O_8$ (40.5 mg, 0.15 mmol) at 140 °C for 3 h afforded recovered C_{60} (13.5 mg, 38%) and **2k** (16.3 mg, 36%): amorphous brown solid; 1H NMR (400 MHz, CS_2 with $DMSO-d_6$ as the external lock solvent) δ 7.82 (s, 1H), 7.30 (s, 1H), 2.54 (s, 3H), 2.51 (s, 3H); ^{13}C NMR (101 MHz, CS_2 with $DMSO-d_6$ as the external lock solvent and $Cr(acac)_3$ as the relaxation reagent) (all 2C unless indicated) δ 163.91 (1C, $C=O$), 156.11 (1C, $O-C=C-CO_2$), 153.01 (1C, aryl C), 147.08 (1C), 146.31, 146.16 (1C), 145.70, 145.44, 145.30, 145.05, 144.98, 144.94, 144.58, 144.35, 144.15, 144.01, 143.58, 143.54, 142.98, 142.59 (1C, aryl C), 141.95, 141.74, 141.72, 141.64, 141.50, 141.41, 141.23, 141.18, 140.83, 140.35, 139.23, 138.75, 136.93, 134.27, 131.99 (1C, aryl C), 122.76 (1C, aryl C), 117.10 (1C, aryl C), 109.11 (1C, aryl C), 104.94 (1C, sp^3 -C of C_{60}), 101.84 (1C, $O-C=C-CO_2$), 68.49 (1C, sp^3 -C of C_{60}), 19.90 (CH_3), 18.69 (CH_3); FT-IR ν/cm^{-1} (KBr) 2922, 2852, 1721, 1648, 1554, 1504, 1469, 1413, 1380, 1260, 1112, 1101, 1024, 895, 839, 792, 721, 550, 523; UV-vis ($CHCl_3$) λ_{max}/nm (log ϵ) 255 (5.02), 319 (4.68), 416 (3.16), 458 (2.48); MALDI-TOF MS m/z calcd for $C_{71}H_8O_3 [M]^-$ 908.0473, found 908.0482.

Synthesis and spectral data of compound 2l

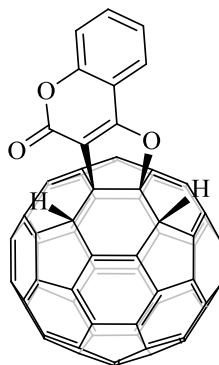


By following the general procedure, the reaction of C_{60} (36.0 mg, 0.05 mmol) with **1l** (31.8 mg, 0.15 mmol), $Pd(OAc)_2$ (1.2 mg, 0.005 mmol), Et_3N (21 μ L, 0.15 mmol) and $K_2S_2O_8$ (40.6 mg, 0.15 mmol) at 140 °C for 3 h afforded recovered C_{60} (18.5 mg, 51%) and **2l** (21.3 mg, 46%): amorphous brown solid; 1H NMR (400 MHz, CS_2 with $DMSO-d_6$ as the external lock solvent) δ 8.68–8.63 (m, 1H), 8.07 (d, J = 8.6 Hz, 1H), 8.00–7.95 (m, 1H), 7.88 (d, J = 8.6 Hz, 1H), 7.77–7.72 (m, 2H); ^{13}C NMR (101 MHz, CS_2 with $DMSO-d_6$ as the external lock solvent and $Cr(acac)_3$ as the relaxation reagent) (all 2C unless indicated) δ 164.94 (1C, $C=O$), 155.79 (1C, $O-C=C-CO_2$), 152.39 (1C, aryl C), 147.08 (1C), 146.18 (1C), 146.16, 145.61, 145.45, 145.33, 145.07, 144.99, 144.96, 144.62, 144.33, 144.16, 144.02, 143.58, 143.53, 142.99, 141.80, 141.76, 141.73, 141.65, 141.49, 141.40, 141.24, 141.20, 140.85, 140.35, 139.29, 138.77, 136.95, 134.76 (1C, aryl C), 134.38, 128.54 (1C, aryl C), 127.23 (1C, aryl C), 126.72 (1C, aryl C), 123.67 (1C, aryl C), 122.46

(1C, aryl C), 122.27 (1C, aryl C), 117.86 (1C, aryl C), 106.56 (1C, aryl C), 105.19 (1C, $\text{sp}^3\text{-C}$ of C_{60}), 102.10 (1C, $\text{O-C}=\text{C-CO}_2$), 68.37 (1C, $\text{sp}^3\text{-C}$ of C_{60}); FT-IR ν/cm^{-1} (KBr) 1728, 1645, 1616, 1466, 1374, 1260, 1104, 1086, 1018, 939, 844, 809, 794, 750, 550, 524; UV-vis (CHCl_3) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 256 (4.96), 319 (4.56), 414 (3.50), 462 (3.28); MALDI-TOF MS m/z calcd for $\text{C}_{73}\text{H}_6\text{O}_3$ $[\text{M}]^-$ 930.0317, found 930.0319.

3. Synthesis and spectral data of compounds 3a–b

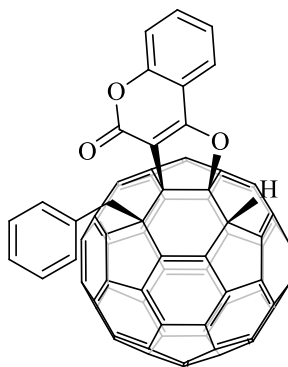
Synthesis and spectral data of compound 3a



3a

20.0 mg (0.023 mmol) of **2a** was electroreduced by controlled potential electrolysis (CPE) at -1.0 V vs saturated calomel electrode (SCE) in 30 mL of ODCB containing 0.1 M *n*-butylammonium perchlorate (TBAP) under an argon atmosphere at 25°C . The electrolysis was terminated when the theoretical number of coulombs required for a full conversion of **2a** to **2a**²⁻ was reached. Then, the dianionic **2a**²⁻ was reacted with trifluoroacetic acid (5 μL , 0.063 mmol) at room temperature for 15 min, and reaction mixture was filtered through a silica gel (200–300 mesh) plug with $\text{CS}_2/\text{CH}_2\text{Cl}_2$ (1:1, v/v) to remove the electrolyte. After evaporation in *vacuo*, the residue was separated by silica gel (400–500 mesh) column with $\text{CS}_2/\text{CH}_2\text{Cl}_2$ (4:1, v/v) to afford **3a** (18.9 mg, 94%). ¹H NMR (400 MHz, CS_2 with $\text{DMSO-}d_6$ as the external lock solvent) δ 8.04 (dd, $J_1 = 7.7$ Hz, $J_2 = 1.5$ Hz, 1H), 7.74 (ddd, $J_1 = 8.8$ Hz, $J_2 = 7.4$ Hz, $J_3 = 1.6$ Hz, 1H), 7.52 (d, $J = 7.9$ Hz, 1H), 7.47 (td, $J_1 = 7.6$ Hz, $J_2 = 1.0$ Hz, 1H), 6.54 (d, $J = 1.8$ Hz, 1H), 6.40 (d, $J = 1.8$ Hz, 1H); ¹³C NMR (101 MHz, CS_2 with $\text{DMSO-}d_6$ as the external lock solvent) (all 1C unless indicated) δ 162.55 ($\text{C}=\text{O}$), 156.26 ($\text{O-C}=\text{C-CO}_2$), 154.45 (aryl C), 149.48, 147.81, 147.41, 147.25, 147.00, 146.44, 146.37, 146.11, 146.08, 145.93, 145.80, 145.33, 145.28, 144.83, 144.69, 144.20, 144.15, 143.90, 143.83, 143.70, 143.67, 143.48, 143.40, 143.36, 143.34, 143.23, 143.13 (2C), 143.12, 143.05, 142.99, 142.84, 142.72, 142.70 (2C), 142.69, 142.51, 142.23, 142.10, 141.72, 141.46, 141.36, 141.30, 141.26, 141.16, 140.73, 140.49, 140.18, 140.14 (2C), 138.91, 138.15, 137.98, 136.08, 135.64, 134.75, 132.14 (aryl C), 123.24 (aryl C), 122.54 (aryl C), 116.40 (aryl C), 111.60 (aryl C), 107.87 ($\text{O-C}=\text{C-CO}_2$), 100.60 ($\text{sp}^3\text{-C}$ of C_{60}), 62.83 ($\text{sp}^3\text{-C}$ of C_{60}), 55.30 ($\text{sp}^3\text{-C}$ of C_{60}), 53.11 ($\text{sp}^3\text{-C}$ of C_{60}); FT-IR ν/cm^{-1} (KBr) 2955, 2920, 2849, 1731, 1645, 1463, 1374, 1262, 1089, 1030, 895, 803, 527; UV-vis (CHCl_3) $\lambda_{\text{max}}/\text{nm}$ ($\log \epsilon$) 258 (5.02), 350 (4.39), 435 (3.70); MALDI-TOF MS m/z calcd for $\text{C}_{69}\text{H}_6\text{O}_3$ $[\text{M}]^-$ 882.0317, found 882.0321.

Synthesis and spectral data of compound 3b



3b

20.0 mg (0.023 mmol) of **2a** was electroreduced by CPE at -1.0V vs SCE in 30 mL of ODCB containing 0.1M TBAP under an argon atmosphere at room temperature. The electrolysis was terminated when the theoretical number of coulombs required for a full conversion of **2a** to **2a**²⁻ was reached. Then, the dianionic **2a**²⁻ was reacted with benzyl bromide (54 μL , 0.46 mmol) at 30 °C for 20 h and reaction mixture was filtered through a silica gel (200–300 mesh) plug with CS₂/CH₂Cl₂ (1:1, v/v) to remove the electrolyte. After evaporation in *vacuo*, the residue was separated by silica gel (400–500 mesh) column with CS₂/CH₂Cl₂ (4:1, v/v) to afford **3b** (17.3 mg, 78%). ¹H NMR (400 MHz, CS₂ with DMSO-*d*₆ as the external lock solvent) δ 7.76–7.69 (m, 2H), 7.58 (d, $J = 7.3$ Hz, 2H), 7.48–7.39 (m, 2H), 6.96 (t, $J = 7.6$ Hz, 2H), 6.48 (t, $J = 7.4$ Hz, 1H), 5.69 (s, 1H), 4.43 (d, $J = 13.2$ Hz, 1H), 4.23 (d, $J = 13.2$ Hz, 1H); ¹³C NMR (101 MHz, CS₂/CDCl₂CDCl₂) δ 163.42 (C=O), 157.32 (O-C=C-CO₂), 153.94 (aryl C), 151.38, 151.08, 148.17, 147.99, 147.27, 146.36, 146.13, 146.03, 145.94, 145.86, 145.68, 145.54, 145.33, 145.20, 145.15, 145.05, 144.90, 144.88, 144.52, 144.03, 143.92, 143.87, 143.75, 143.67, 143.63, 143.58, 143.53, 143.31, 143.26, 143.17 (2C), 143.08, 142.92, 142.80, 142.76, 142.72, 142.45, 142.40, 142.39, 141.99, 141.85, 141.61, 141.54, 141.48, 141.06, 140.71, 140.66, 140.51, 140.46, 139.71, 137.42, 137.31, 136.76, 134.61, 134.32, 133.73, 133.17 (aryl C), 132.23 (aryl C), 129.92 (2C, aryl C), 127.38 (2C, aryl C), 126.02 (aryl C), 123.15 (aryl C), 122.41 (aryl C), 115.99 (aryl C), 111.93 (aryl C), 101.71 (O-C=C-CO₂), 101.48 (sp³-C of C₆₀), 64.65 (sp³-C of C₆₀), 61.42 (sp³-C of C₆₀), 58.68 (sp³-C of C₆₀), 51.36 (CH₂); FT-IR ν/cm^{-1} (KBr) 2961, 2920, 2846, 1733, 1651, 1604, 1495, 1454, 1407, 1318, 1274, 1101, 1036, 895, 812, 759, 700, 555, 526; UV-vis (CHCl₃) $\lambda_{\text{max}}/\text{nm}$ (log ϵ) 258 (5.02), 332 (4.50), 428 (3.63); MALDI-TOF MS m/z calcd for C₇₆H₁₂O₃ [M]⁻ 972.0786, found 972.0788.

Note: Complete removal of petroleum ether/grease from the samples was difficult for the ¹H NMR and ¹³C NMR spectra due to the poor solubility of these compounds. The ¹³C NMR spectra of **2h** and **2i** were unavailable due to their extremely low solubility.

4. NMR spectra of compounds 2a-l and 3a-b

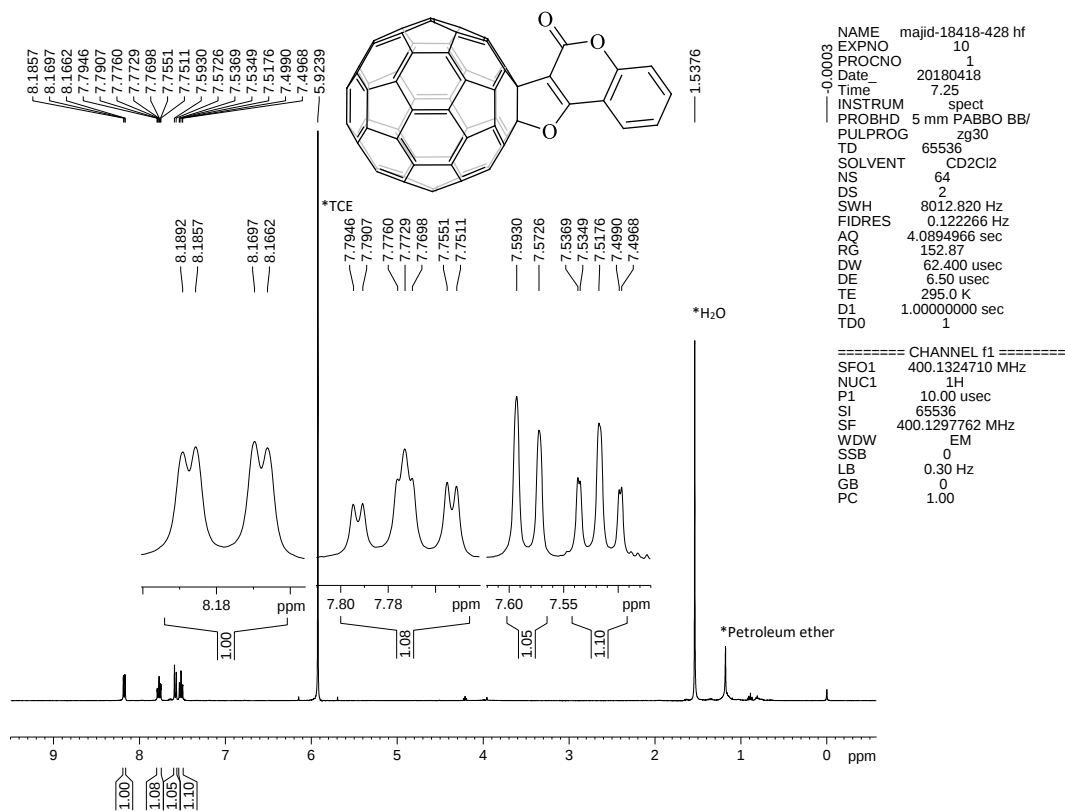


Figure S1 ¹H NMR (400 MHz, CDCl₂CDCl₂) of compound 2a

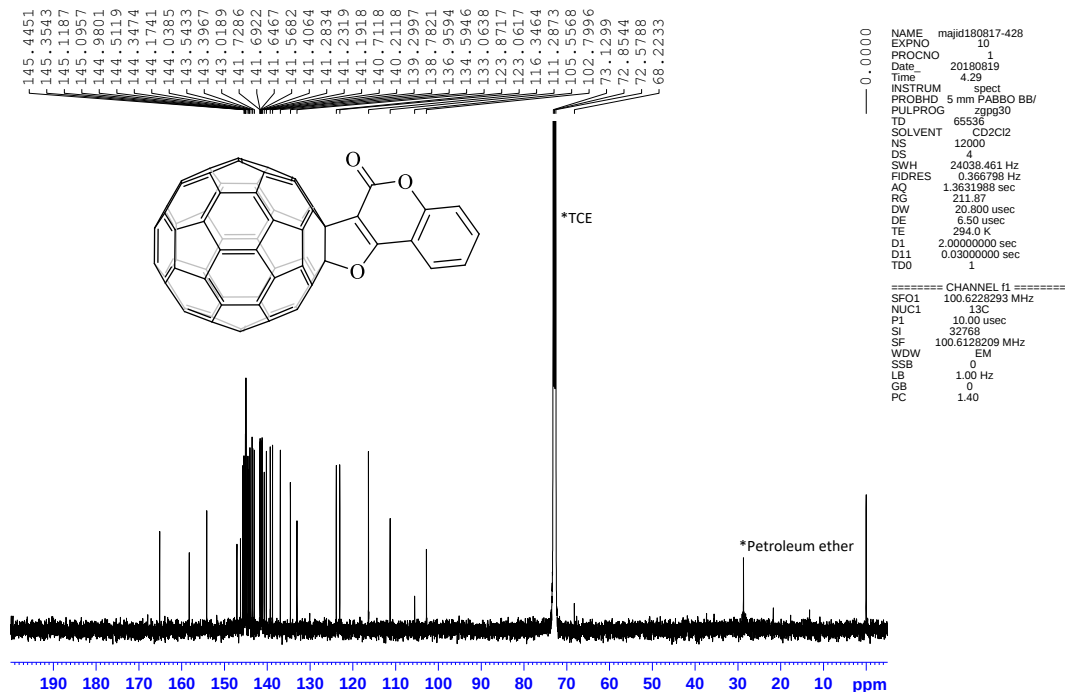


Figure S2 ¹³C NMR (101 MHz, CDCl₂CDCl₂) of compound 2a

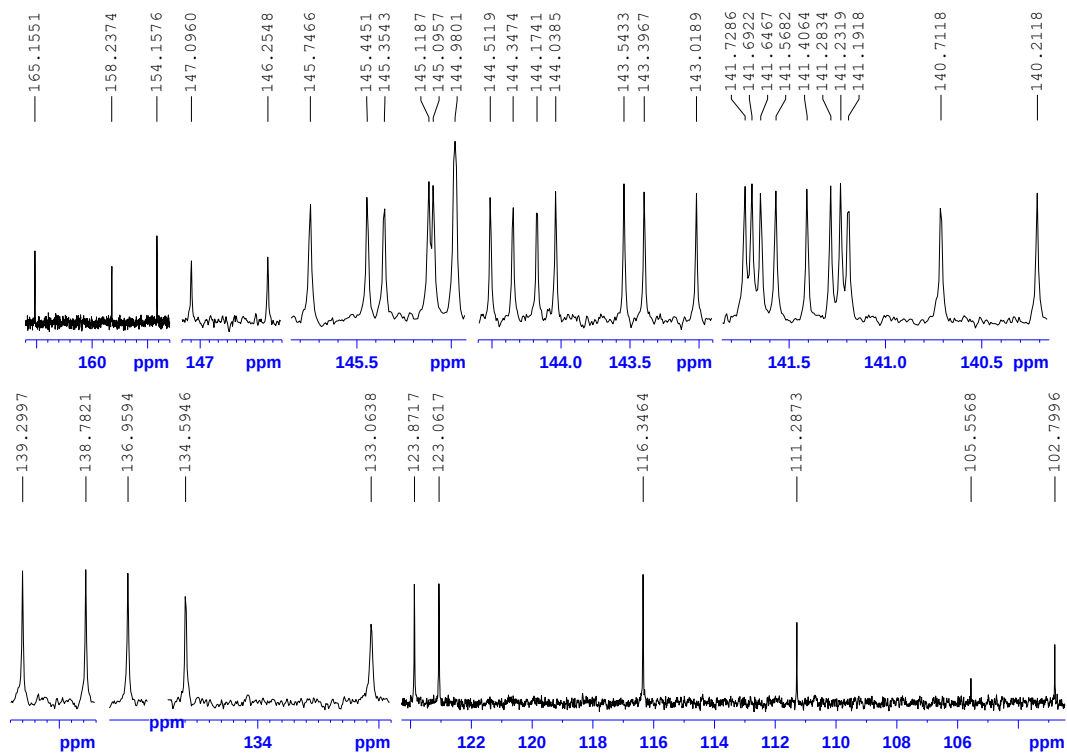


Figure S3 Expanded ^{13}C NMR (101 MHz, $\text{CDCl}_2\text{CDCl}_2$) of compound 2a

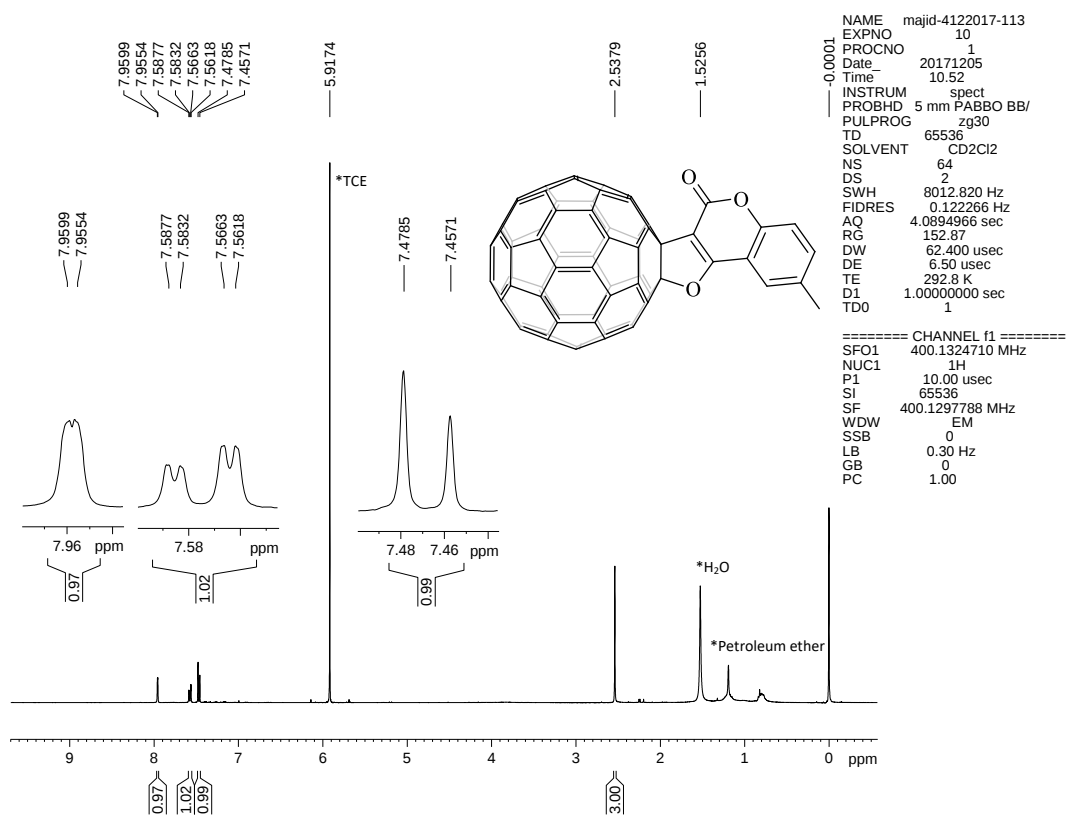
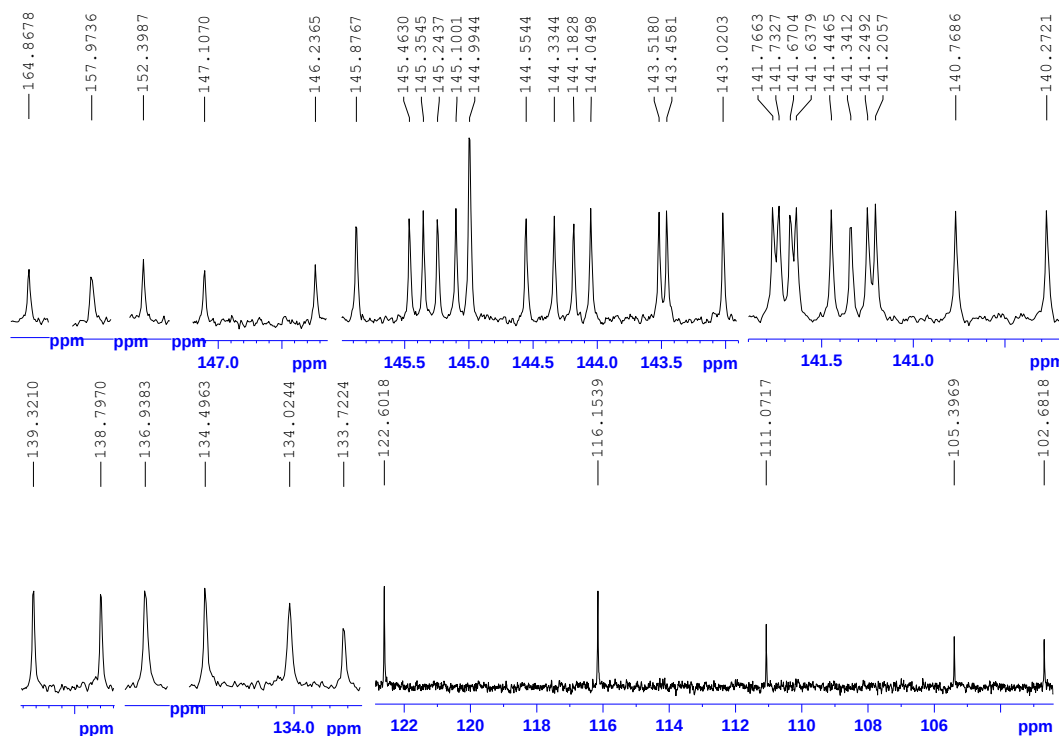
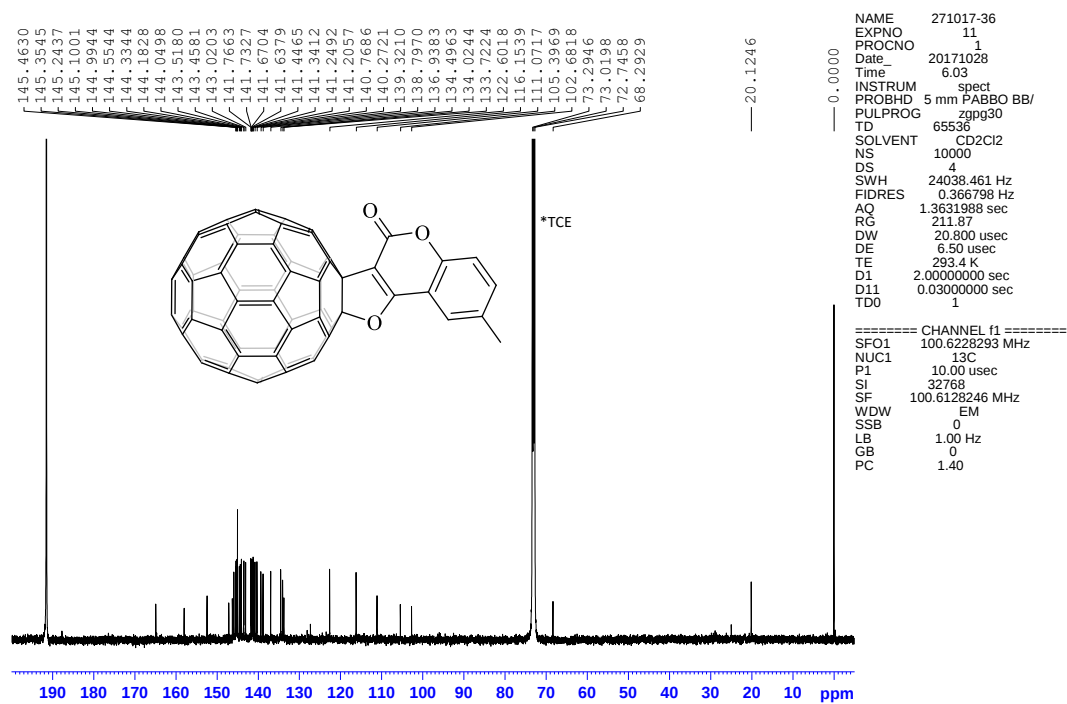


Figure S4 ^1H NMR (400 MHz, $\text{CS}_2/\text{CDCl}_2\text{CDCl}_2$ 2:3) of compound 2b



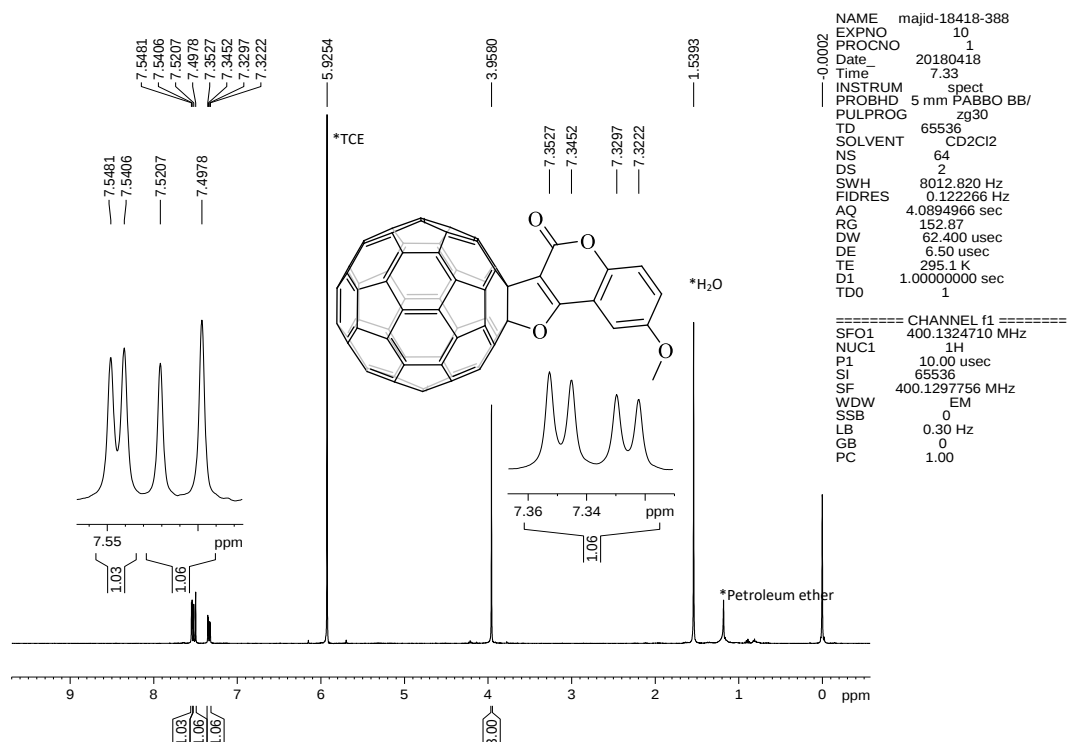


Figure S7 ¹H NMR (400 MHz, CDCl₃) of compound 2c

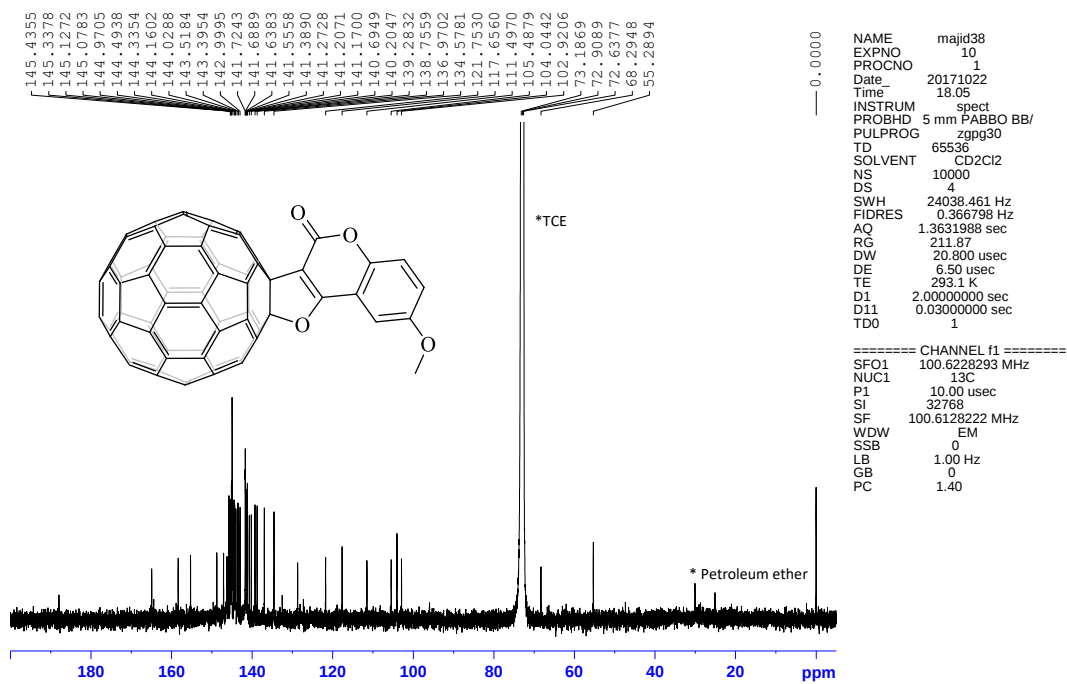


Figure S8 ¹³C NMR (101 MHz, CDCl₃) of compound 2c

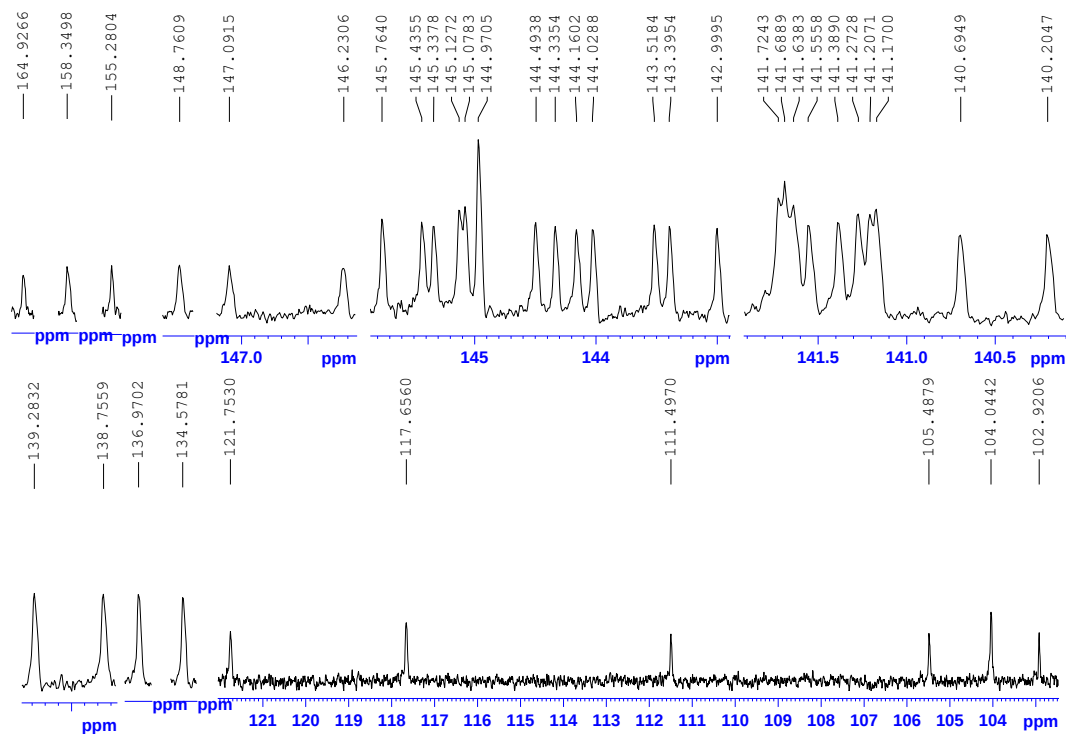


Figure S9 Expanded ^{13}C NMR (101 MHz, $\text{CDCl}_2\text{CDCl}_2$) of compound 2c

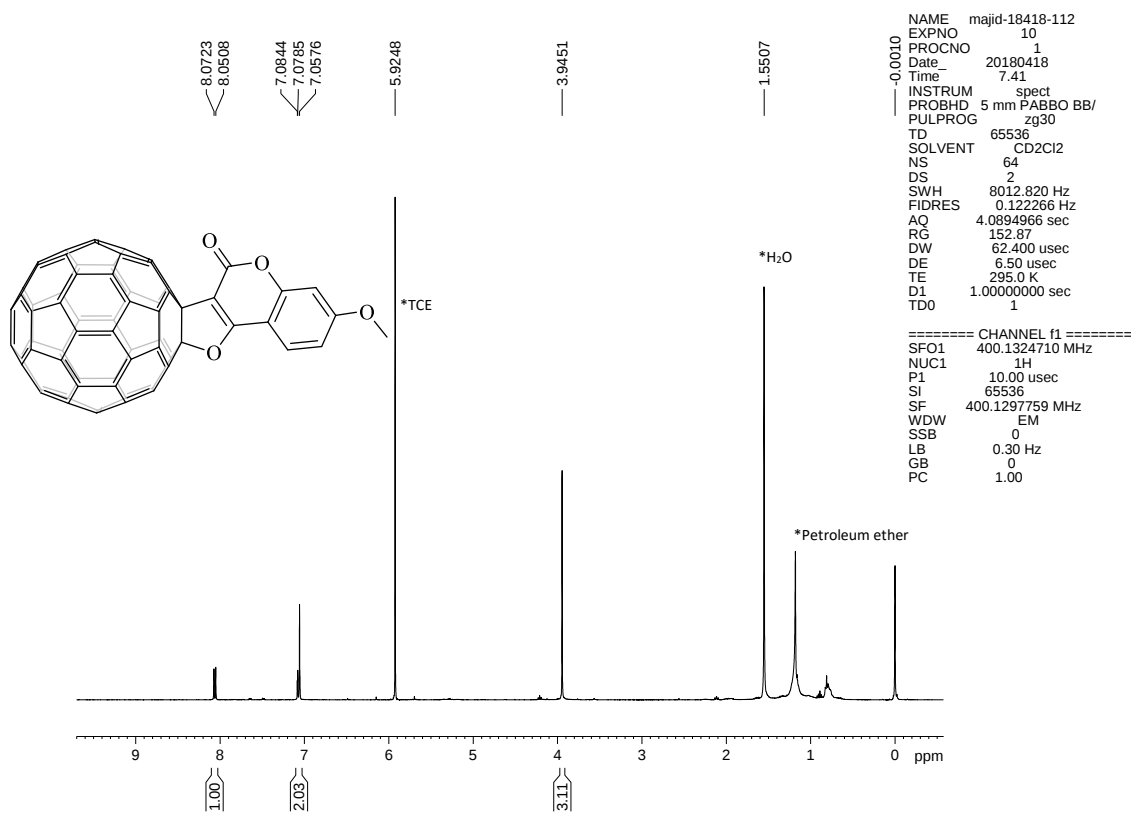


Figure S10 ^1H NMR (400 MHz, $\text{CDCl}_2\text{CDCl}_2$) of compound 2d

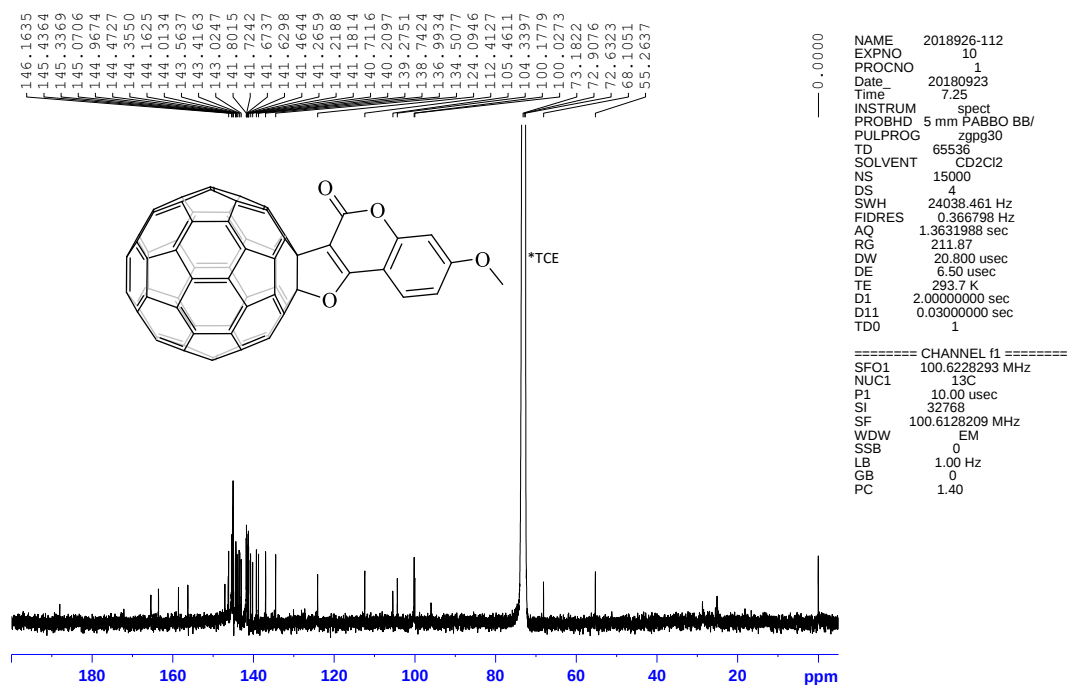


Figure S11 ^{13}C NMR (101 MHz, $\text{CDCl}_2\text{CDCl}_2$) of compound 2d

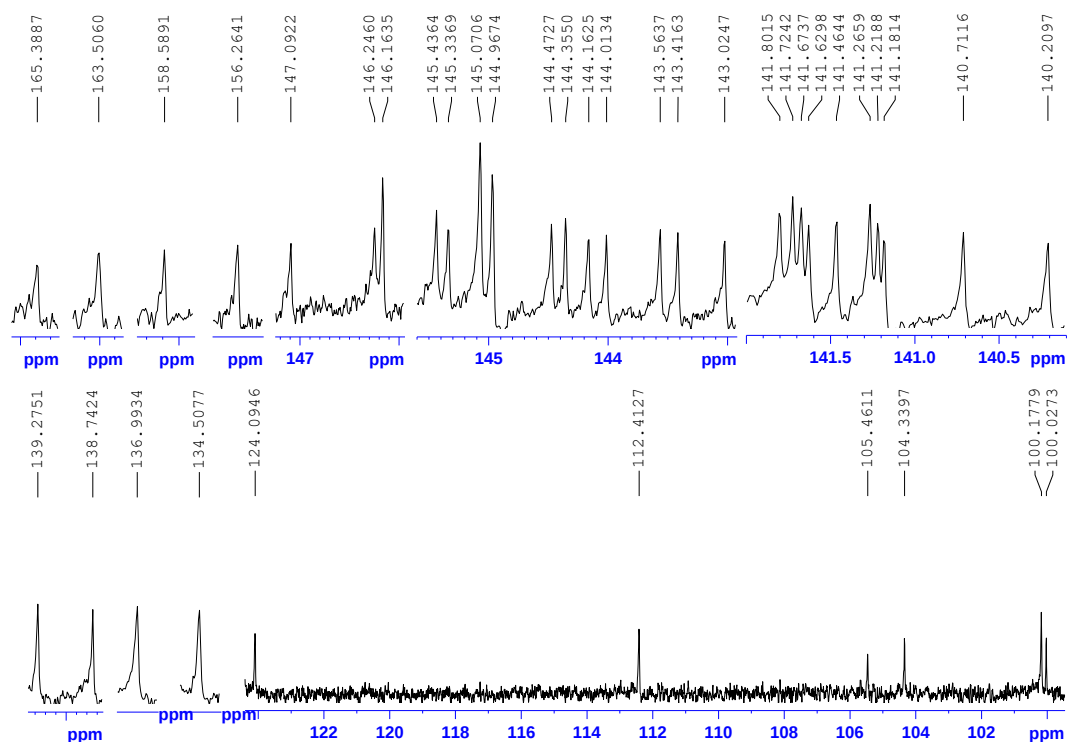


Figure S12 Expanded ^{13}C NMR (101 MHz, $\text{CDCl}_2\text{CDCl}_2$) of compound 2d

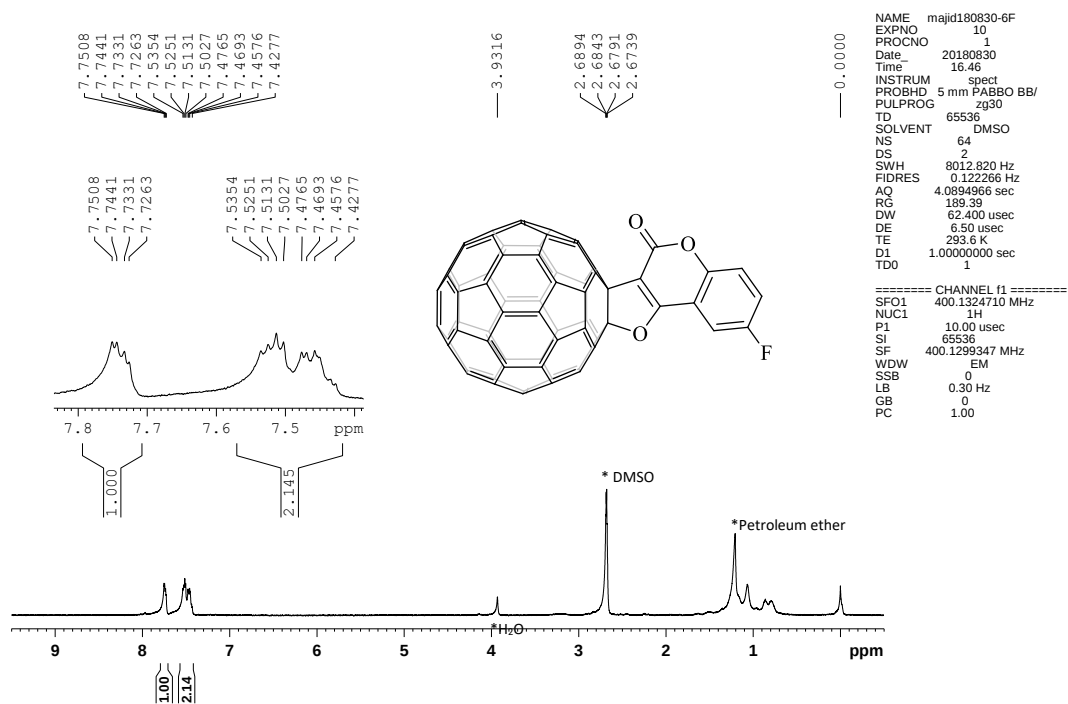


Figure S13 ¹H NMR (400 MHz, CS₂/DMSO-*d*₆) of compound 2e

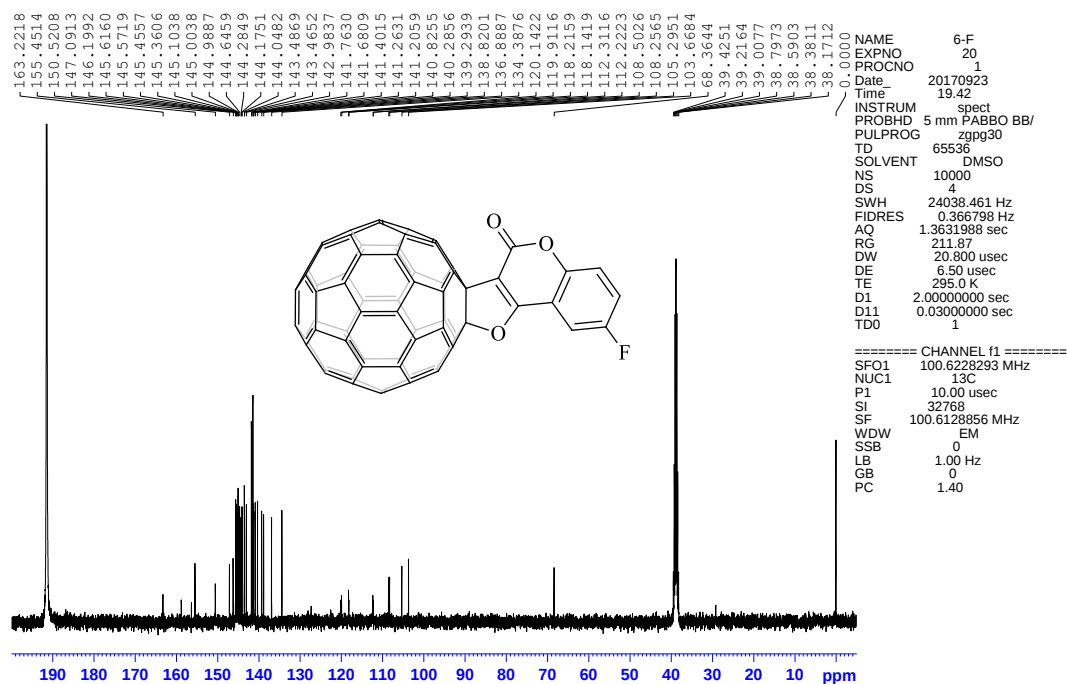


Figure S14 ¹³C NMR (101 MHz, CS₂/DMSO-*d*₆) of compound 2e

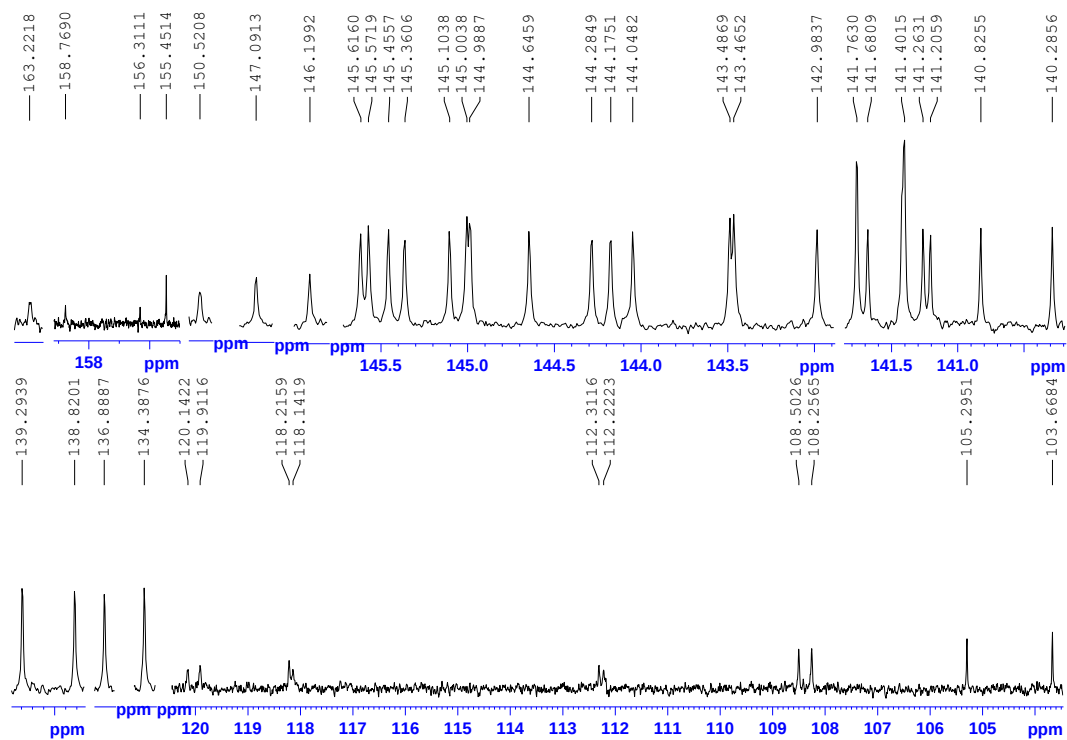


Figure S15 Expanded ^{13}C NMR (101 MHz, $\text{CS}_2/\text{DMSO}-d_6$) of compound 2e

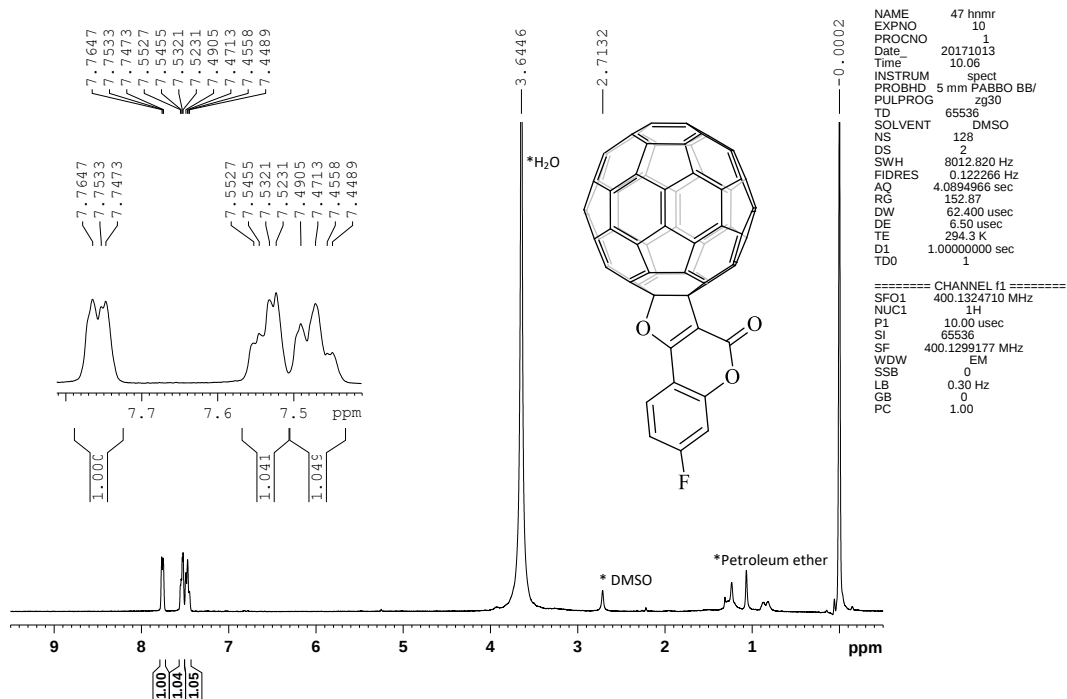


Figure S16 ^1H NMR (400 MHz, $\text{CS}_2/\text{DMSO}-d_6$) of compound 2f

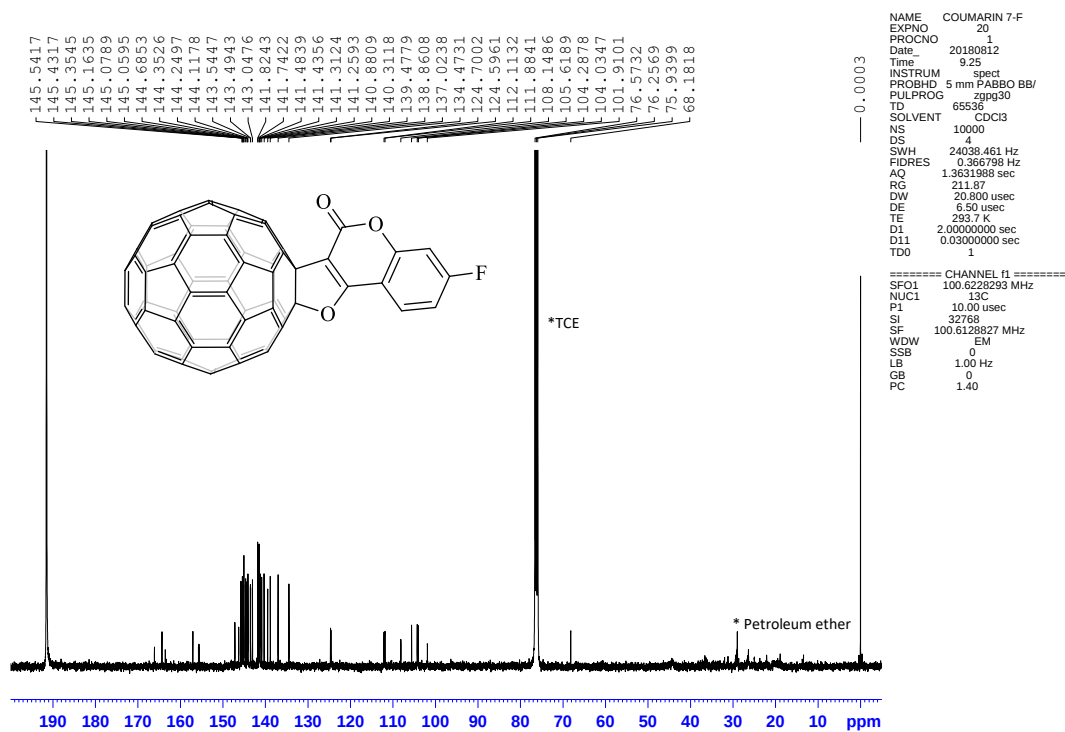


Figure S17 ^{13}C NMR (101 MHz, $\text{CS}_2/\text{CDCl}_3$ 2:3) of compound 2f

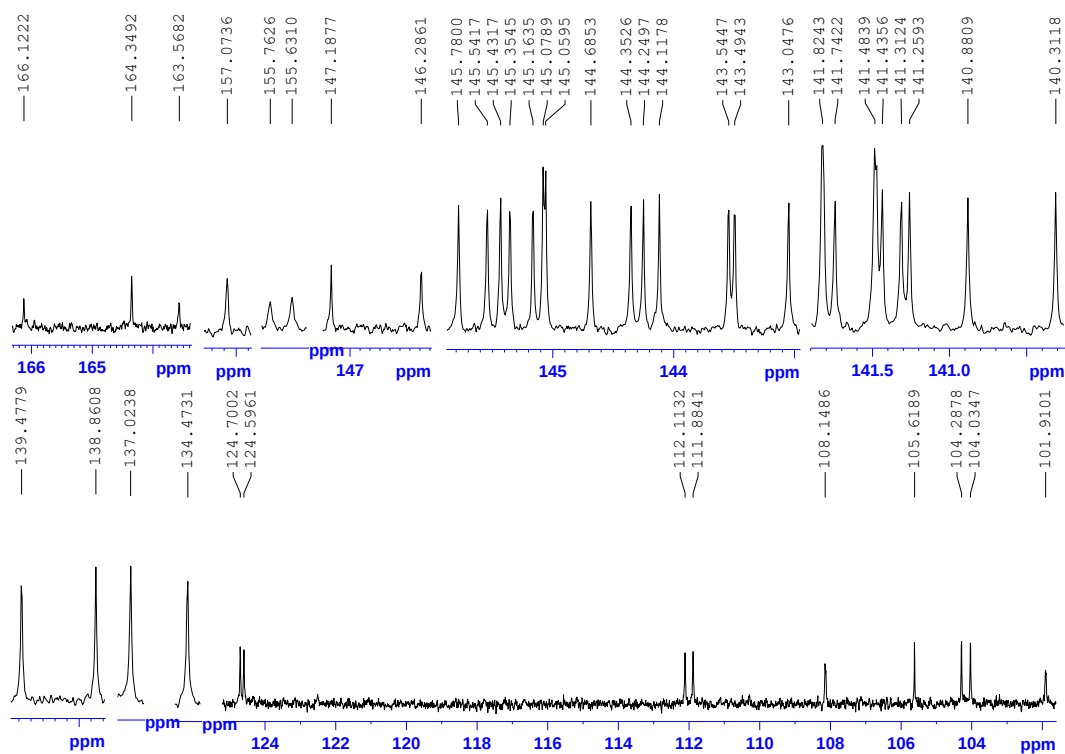


Figure S18 Expanded ^{13}C NMR (101 MHz, $\text{CS}_2/\text{CDCl}_3$ 2:3) of compound 2f

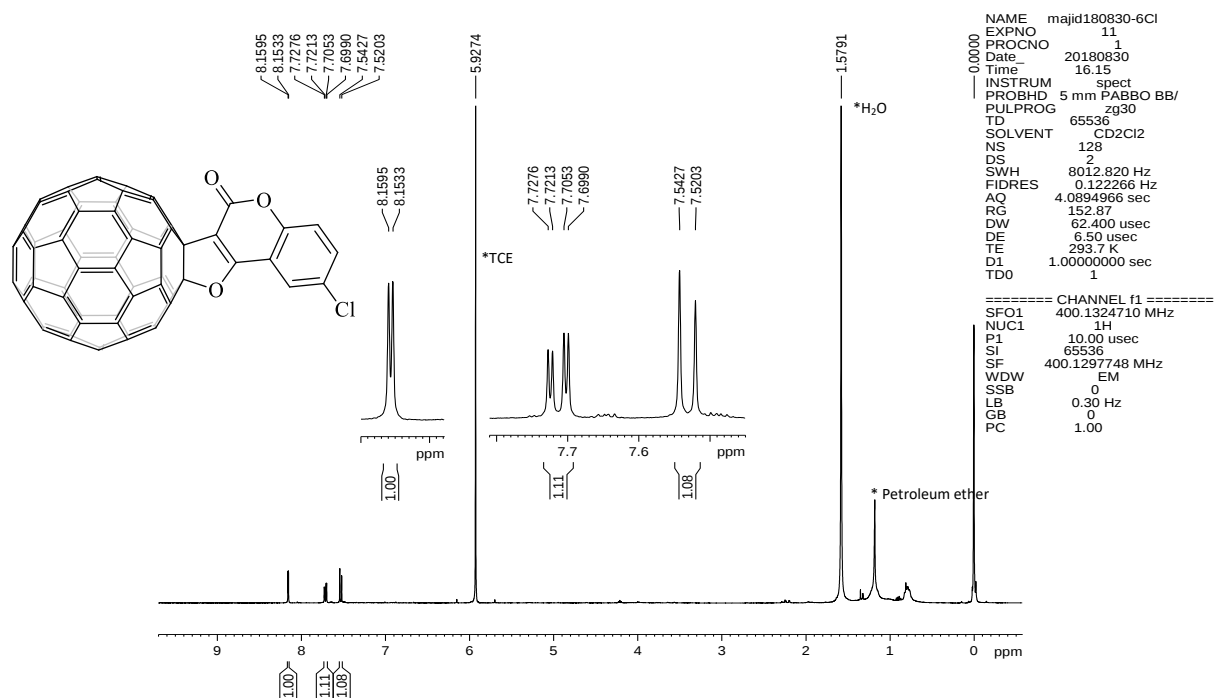


Figure S19 ¹H NMR (400 MHz, CS₂/CDCl₂CDCl₂ 2:3) of compound 2g

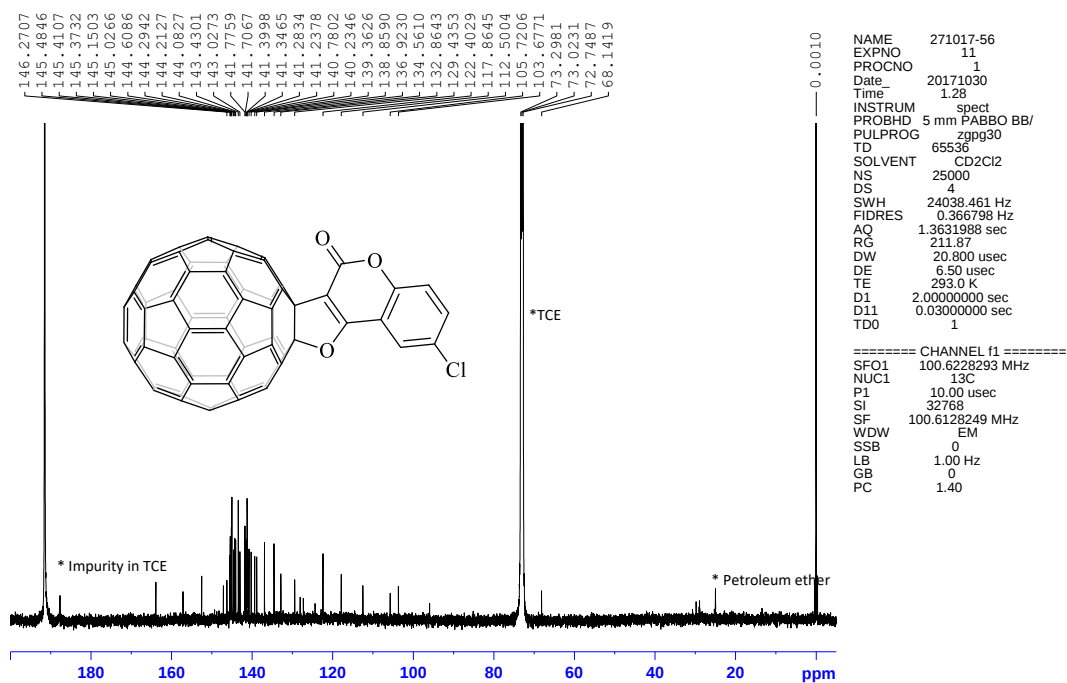


Figure S20 ¹³C NMR (101 MHz, CS₂/CDCl₂CDCl₂ 2:3) of compound 2g

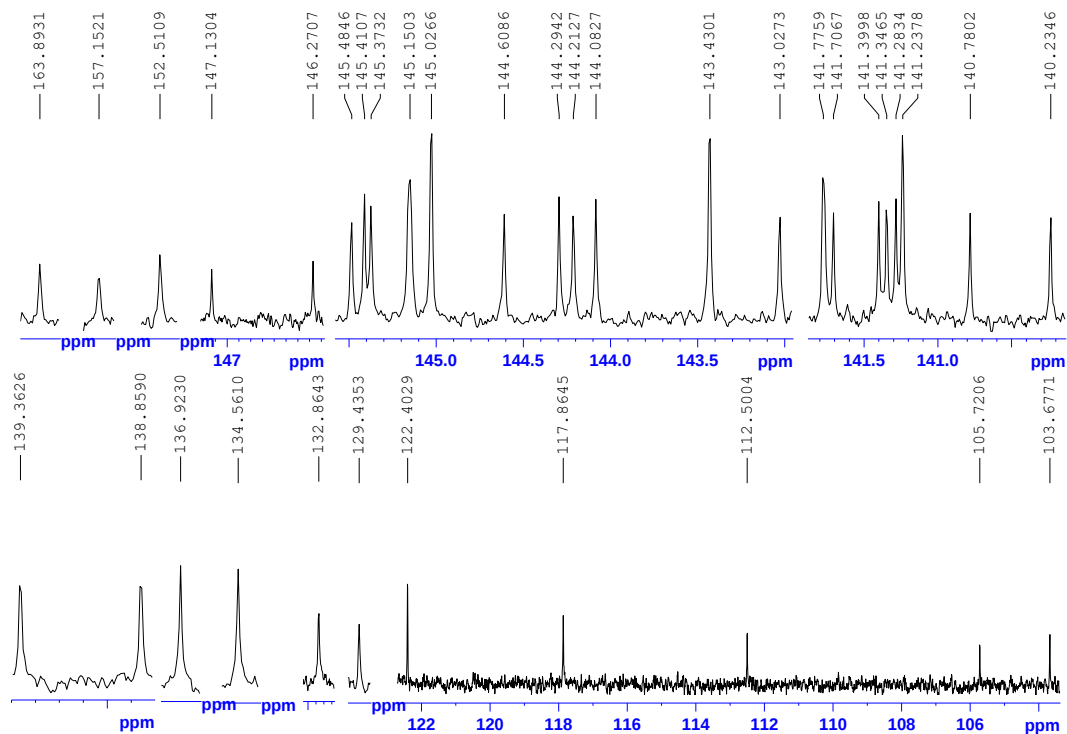


Figure S21 Expanded ^{13}C NMR (101 MHz, $\text{CS}_2/\text{CDCl}_2\text{CDCl}_2$ 2:3) of compound **2g**

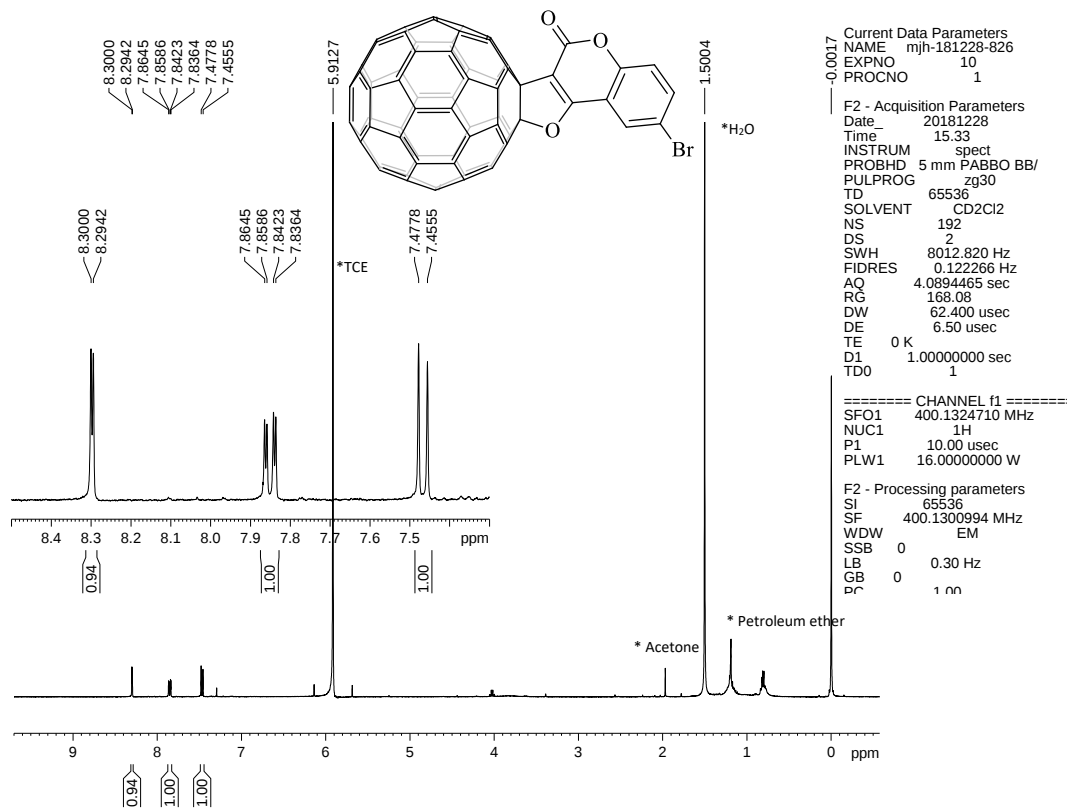


Figure S22 ^1H NMR (400 MHz, $\text{CDCl}_2\text{CDCl}_2$) of compound **2h**

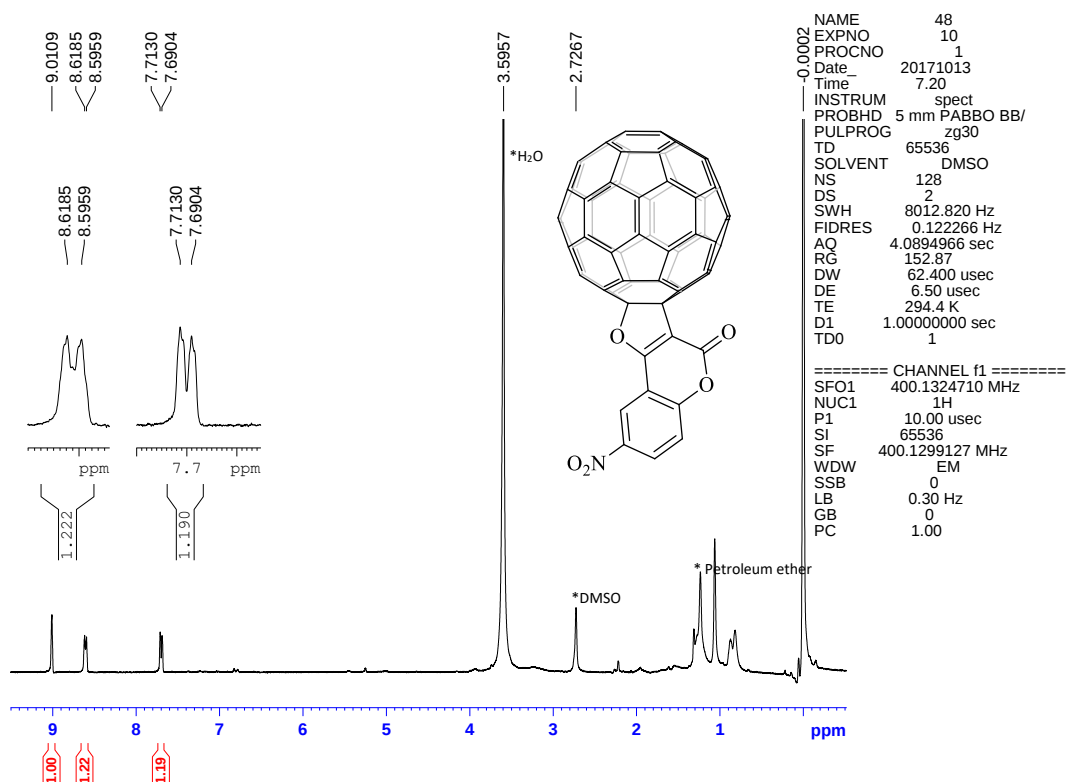


Figure S23 ¹H NMR (400 MHz, CS₂/DMSO-*d*₆) of compound 2i

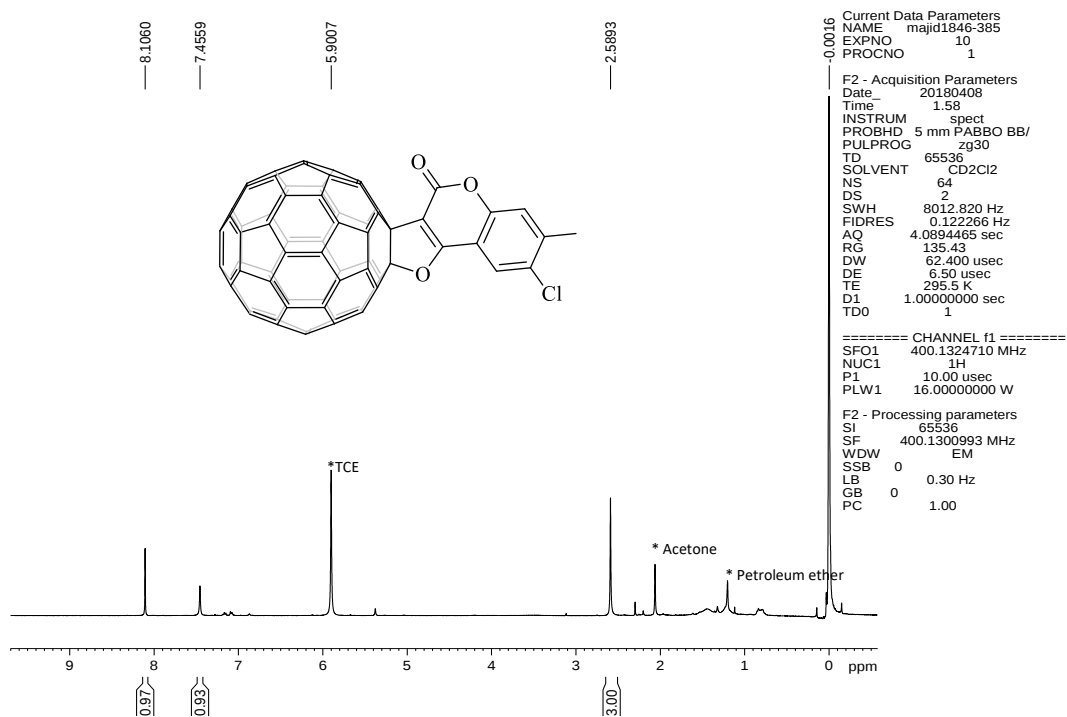


Figure S24 ¹H NMR (400 MHz, CS₂/CDCl₂CDCl₂ 2:3) of compound 2j

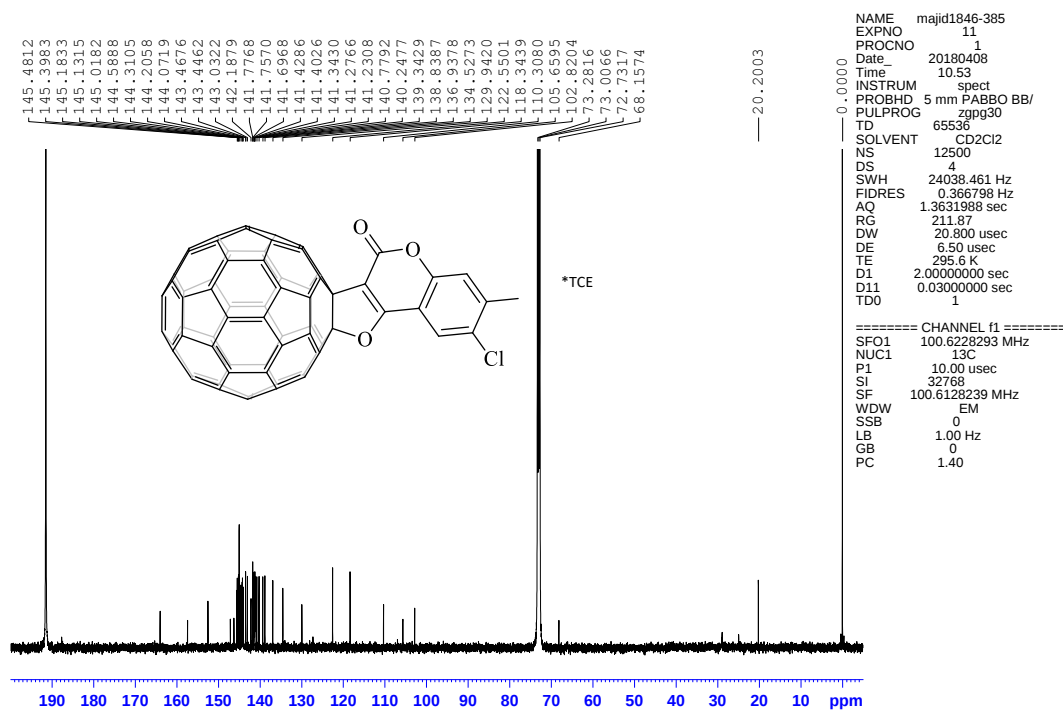


Figure S25 ^{13}C NMR (101 MHz, $\text{CS}_2/\text{CDCl}_2\text{CDCl}_2$ 2:3) of compound 2j

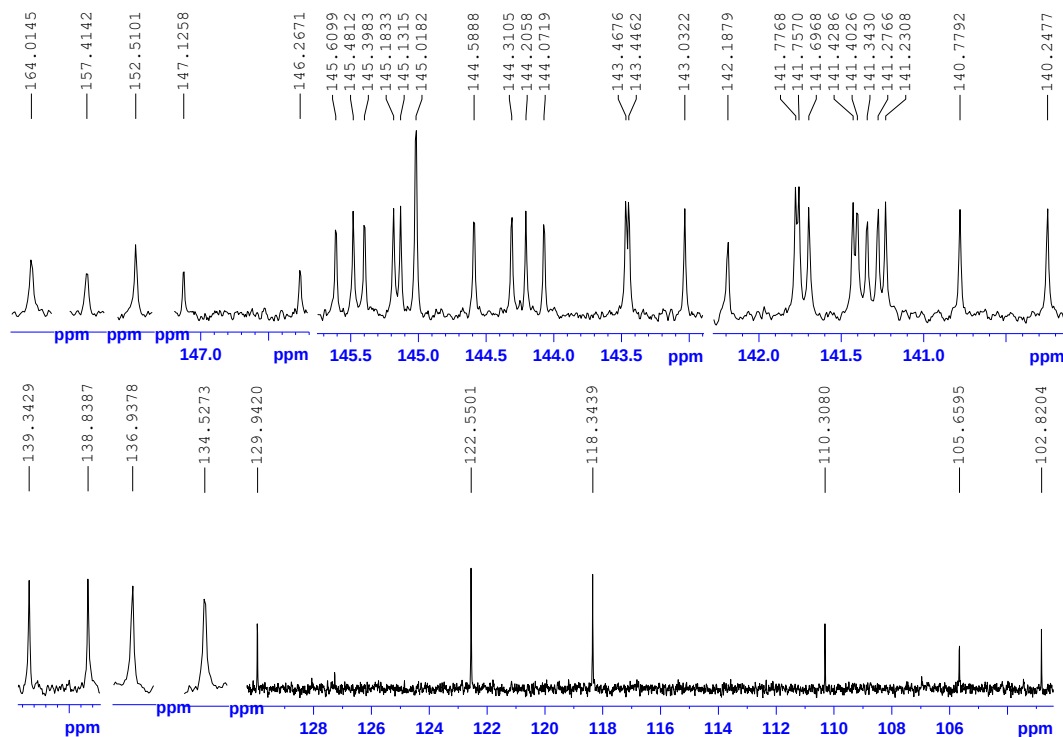
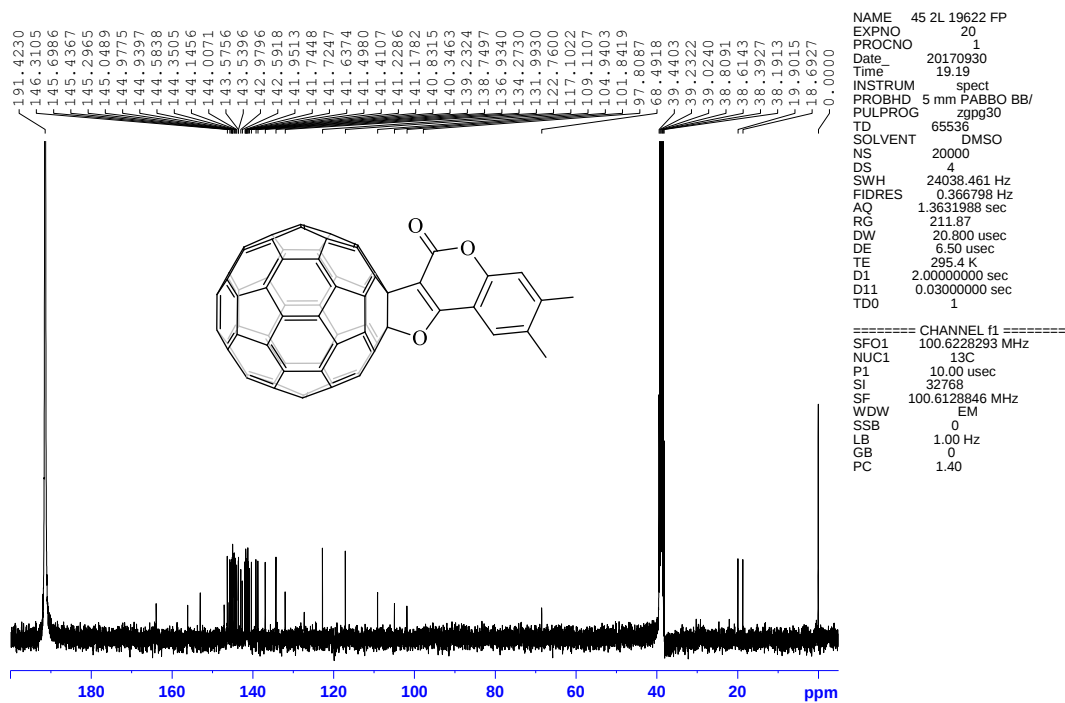
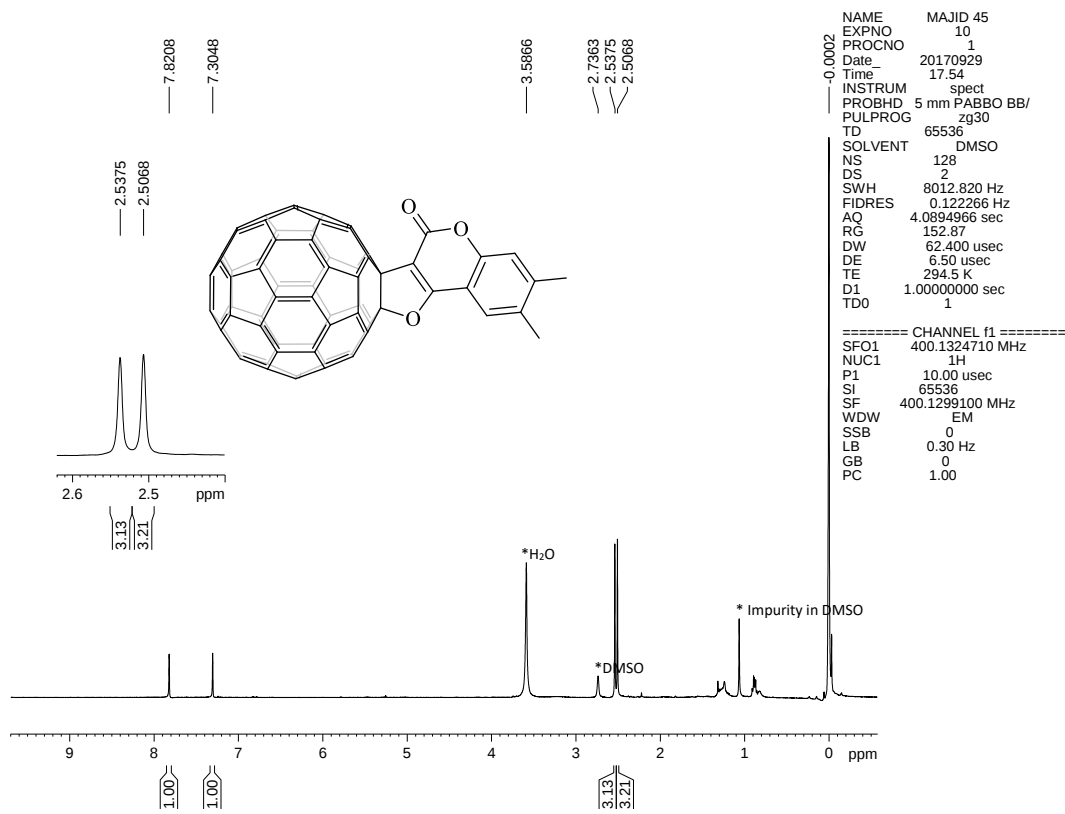


Figure S26 Expanded ^{13}C NMR (101 MHz, 2:3 $\text{CS}_2/\text{CDCl}_2\text{CDCl}_2$) of compound 2j



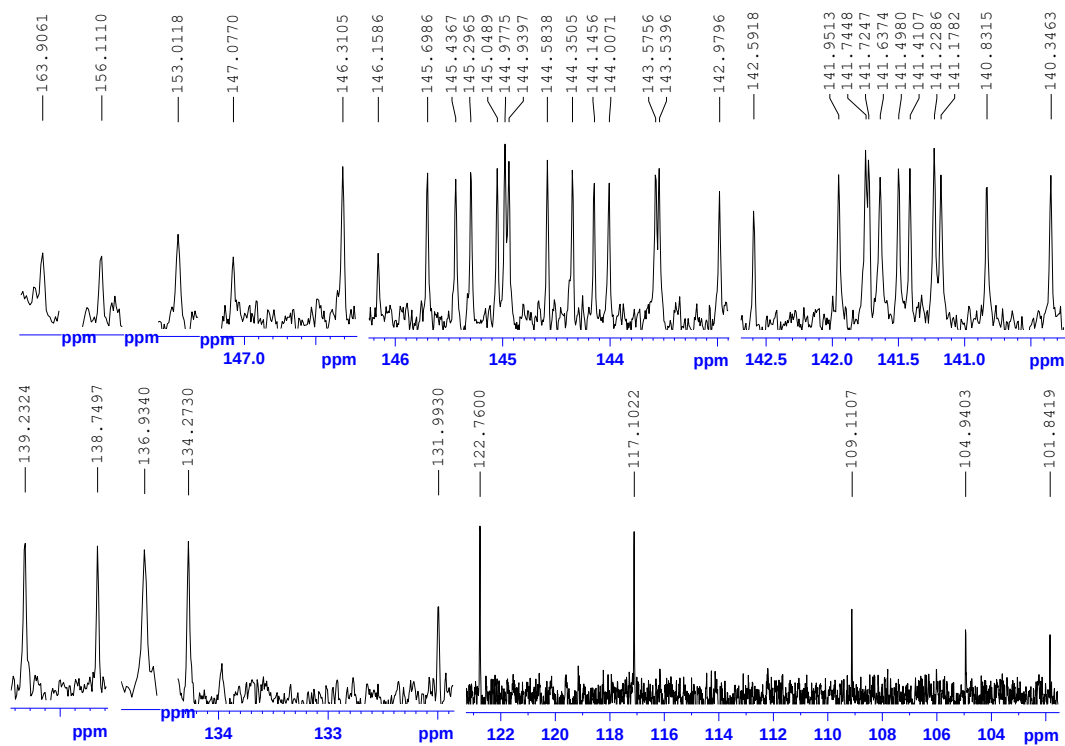


Figure S29 Expanded ^{13}C NMR (101 MHz, $\text{CS}_2/\text{DMSO}-d_6$) of compound 2k

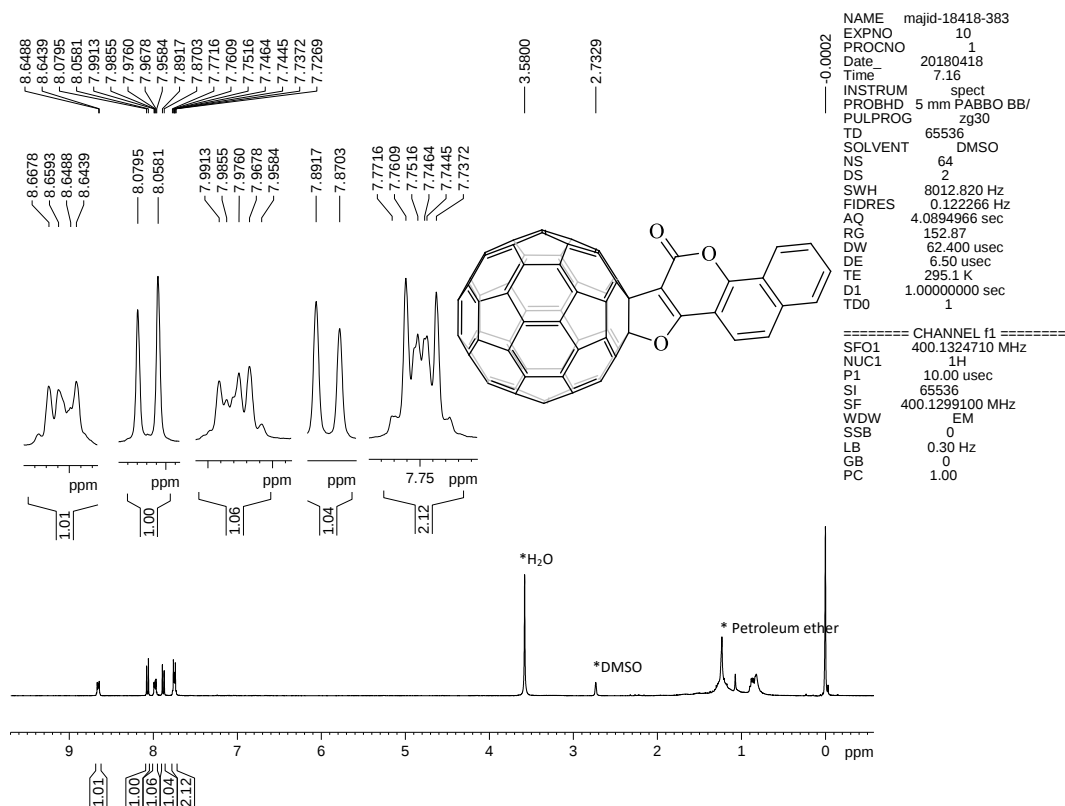


Figure S30 ^1H NMR (400 MHz, $\text{CS}_2/\text{DMSO}-d_6$) of compound 2l

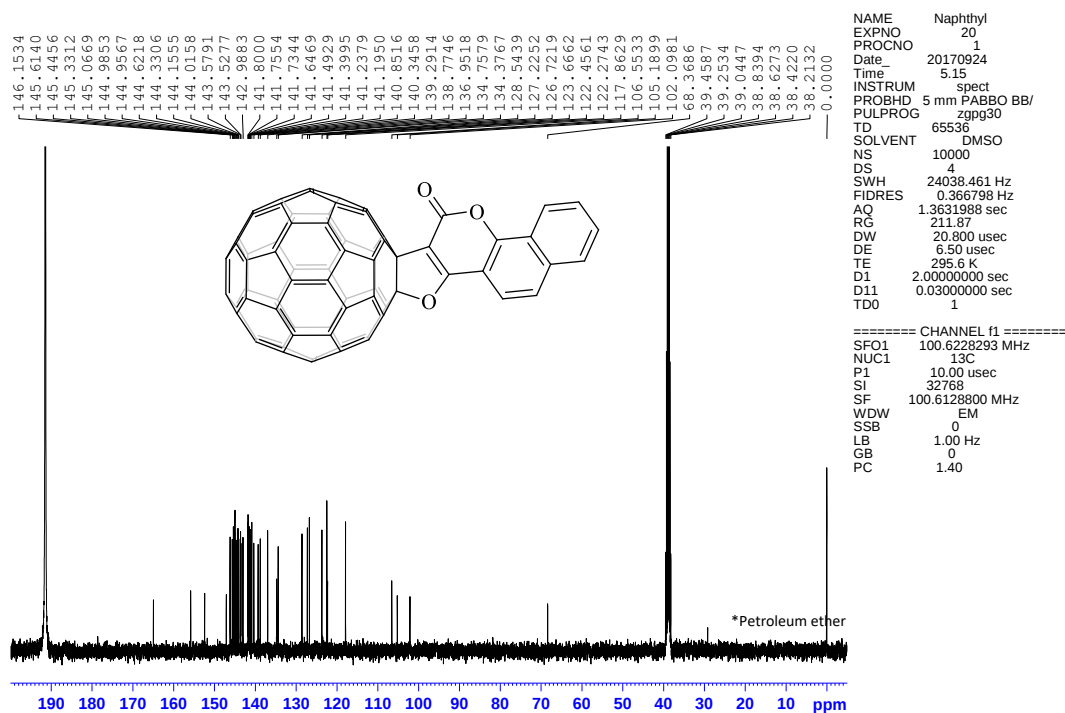


Figure S31 ^{13}C NMR (101 MHz, $\text{CS}_2/\text{DMSO}-d_6$) of compound 2l

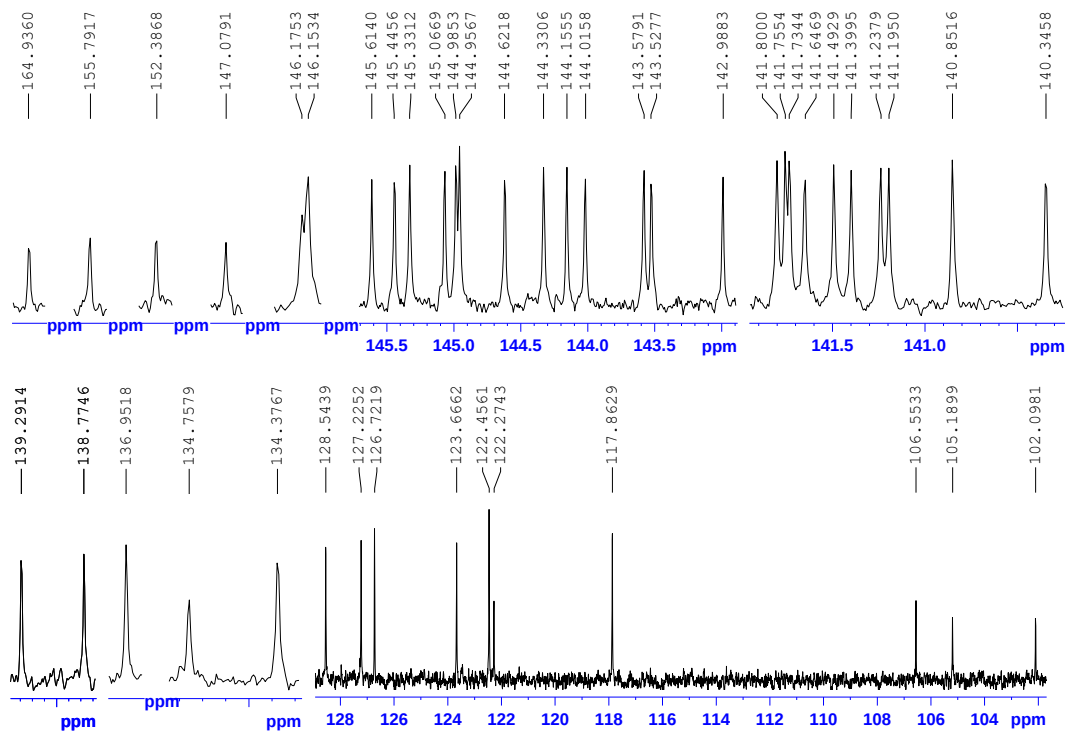


Figure S32 Expanded ^{13}C NMR (101 MHz, $\text{CS}_2/\text{DMSO}-d_6$) of compound 2l

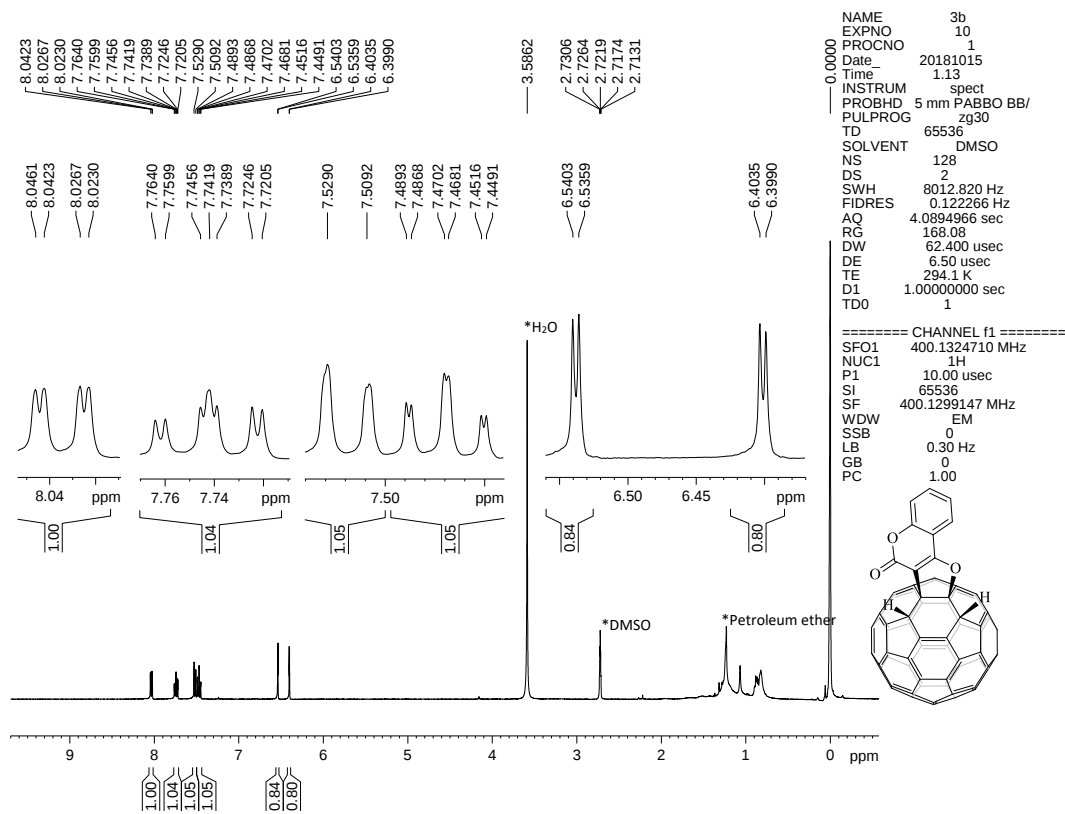


Figure S33 ¹H NMR (400 MHz, CS₂/DMSO-*d*₆) of compound 3a

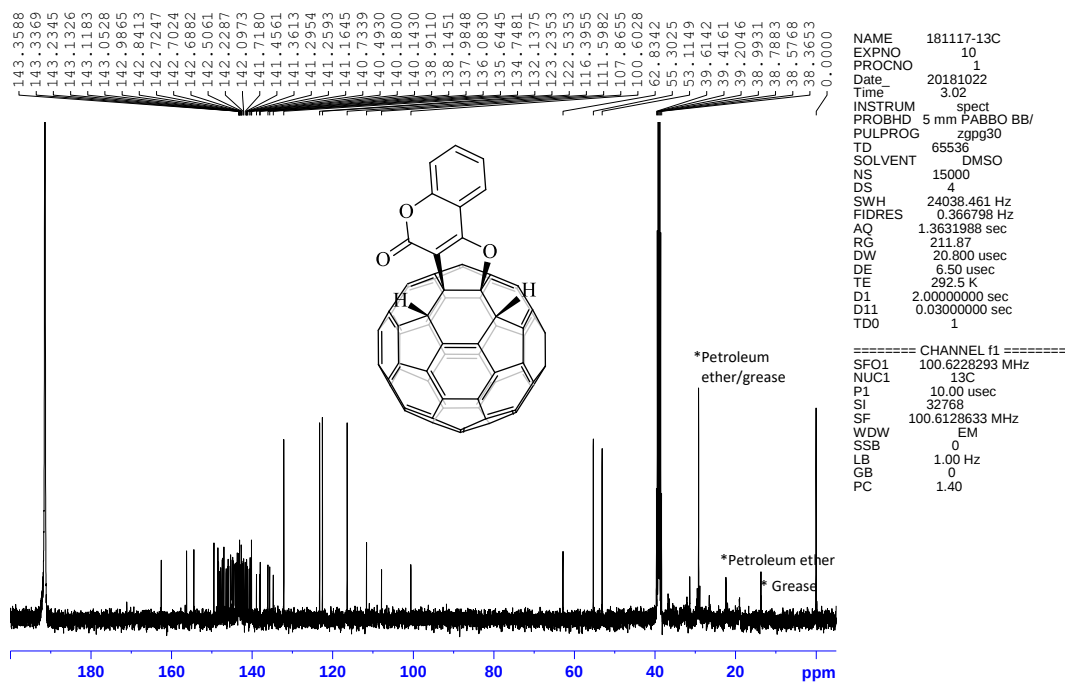


Figure S34 ¹³C NMR (101 MHz, CS₂/DMSO-*d*₆) of compound 3a

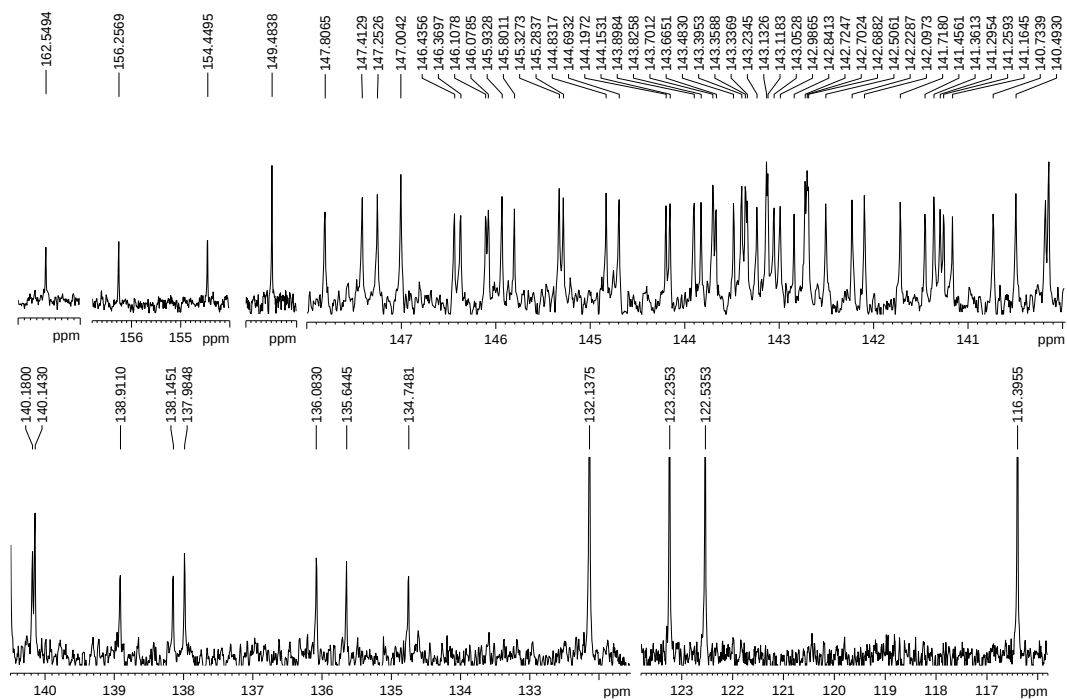


Figure S35 Expanded ^{13}C NMR (101 MHz, $\text{CS}_2/\text{DMSO}-d_6$) of compound 3a

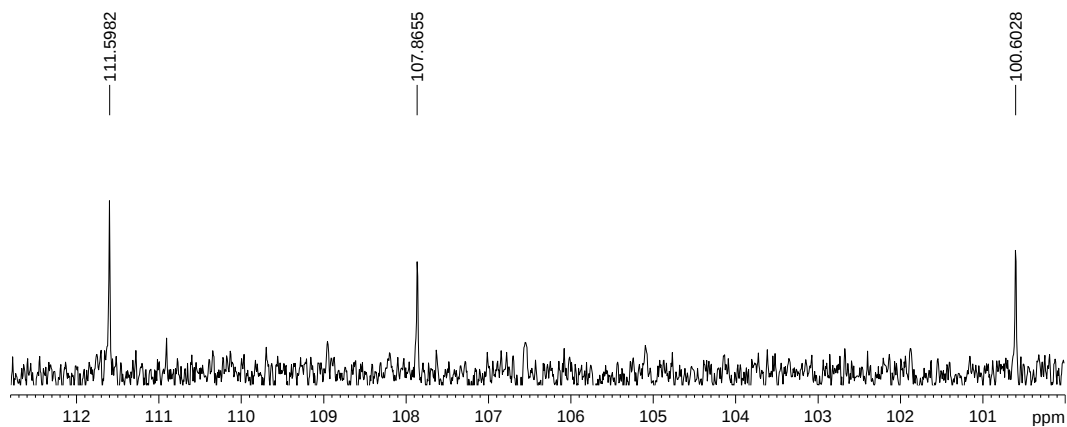


Figure S36 Expanded ^{13}C NMR (101 MHz, $\text{CS}_2/\text{DMSO}-d_6$) of compound 3a

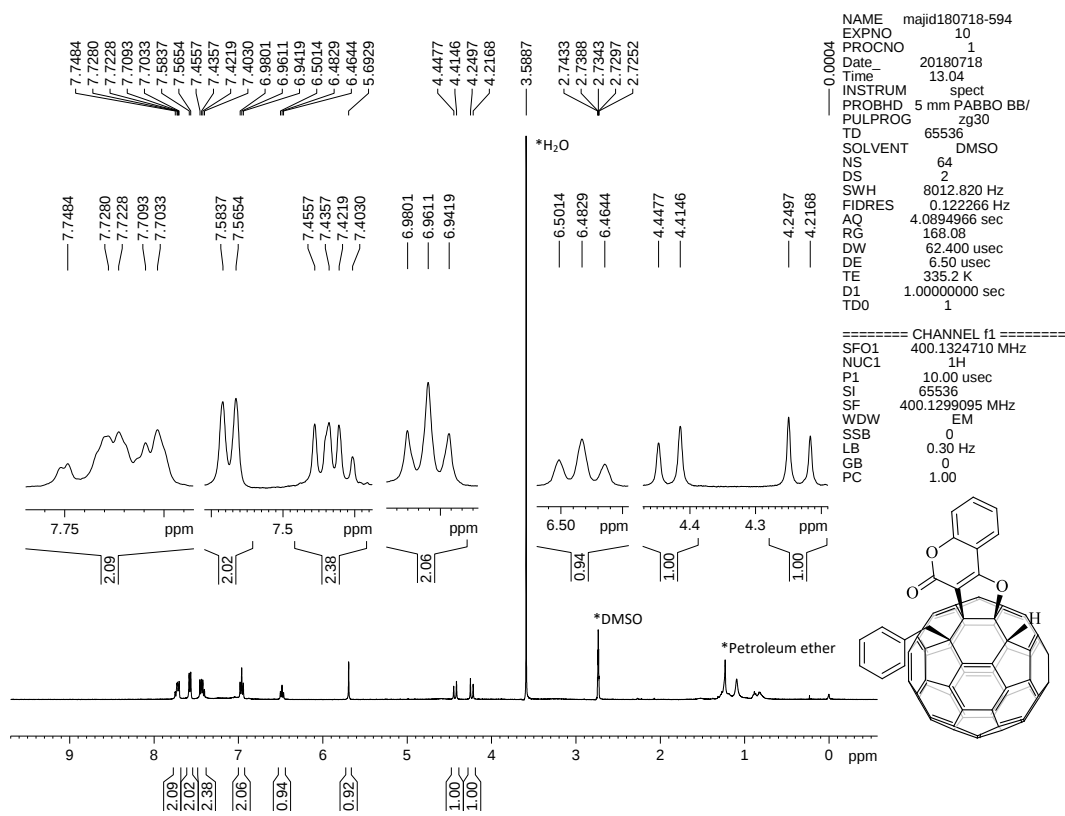


Figure S37 ¹H NMR (400 MHz, CS₂/DMSO-*d*₆) of compound 3b

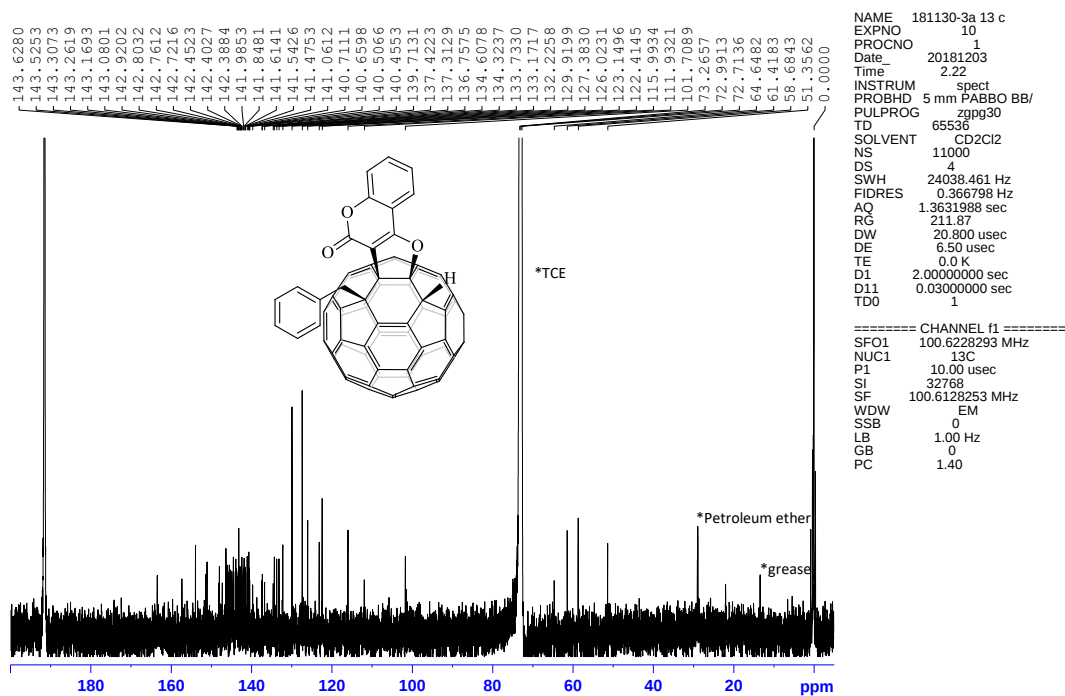


Figure S38 ¹³C NMR (101 MHz, CS₂/CDCl₃ 2:3) of compound 3b

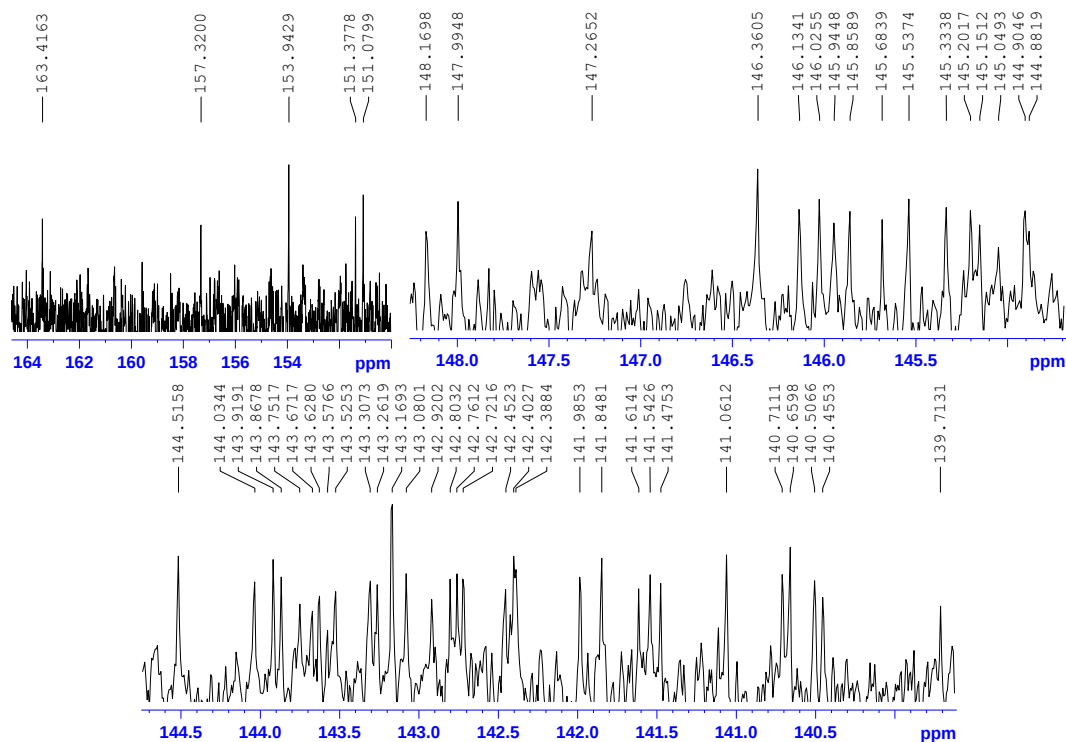


Figure S39 Expanded ^{13}C NMR (101 MHz, $\text{CS}_2/\text{CDCl}_2\text{CDCl}_2$ 2:3) of compound 3b

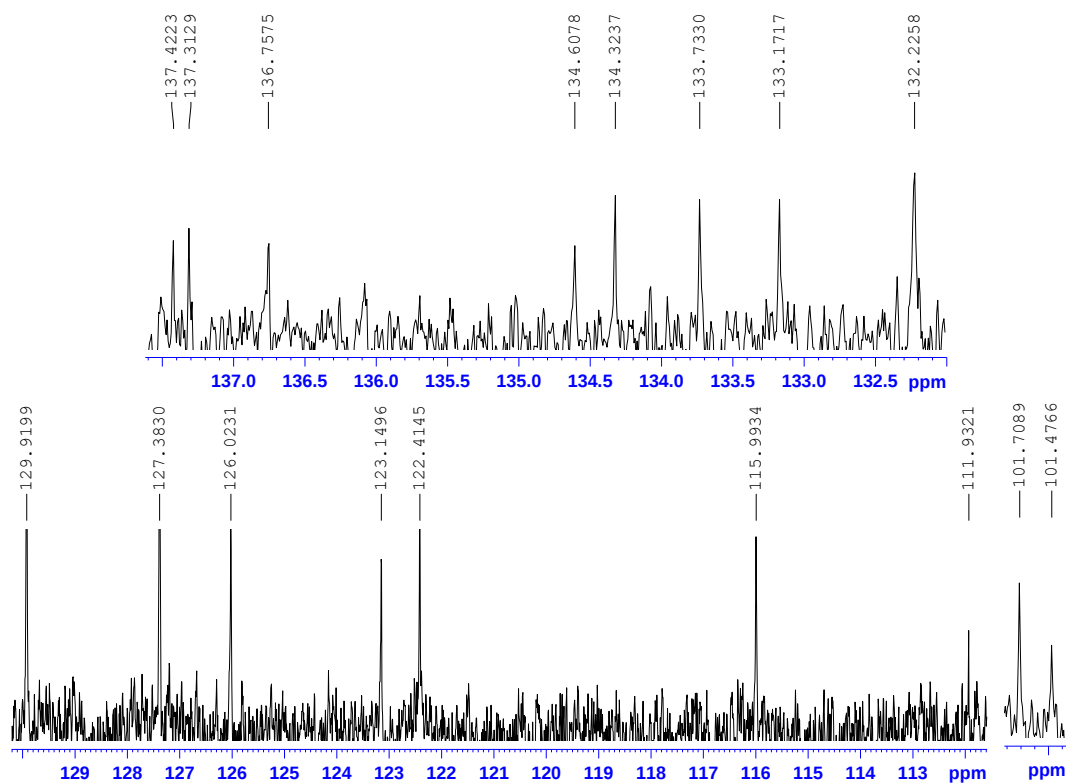


Figure S40 Expanded ^{13}C NMR (101 MHz, $\text{CS}_2/\text{CDCl}_2\text{CDCl}_2$ 2:3) of compound 3b

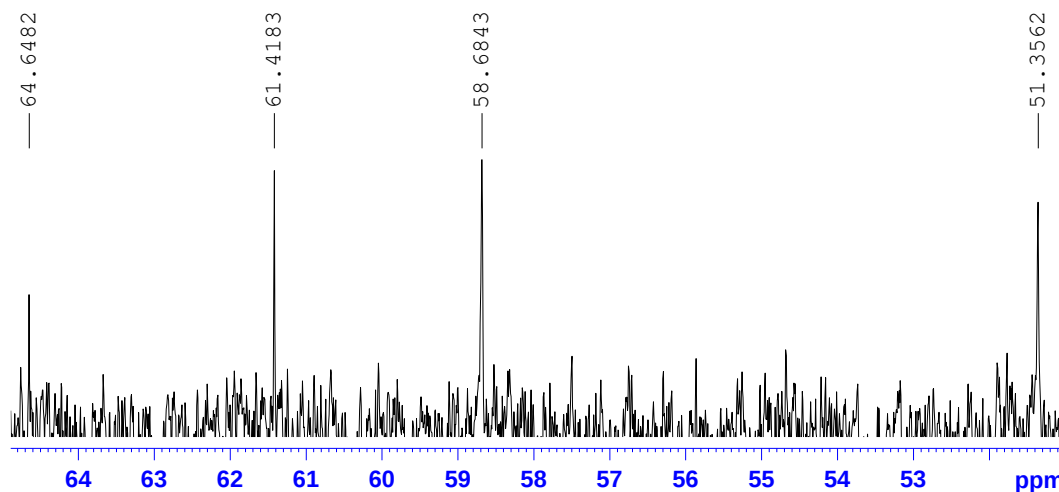


Figure S41 Expanded ^{13}C NMR (101 MHz, 2:3 $\text{CS}_2/\text{CDCl}_3$) of compound 3b

5. UV-vis spectra of compounds 2a–l and 3a–d

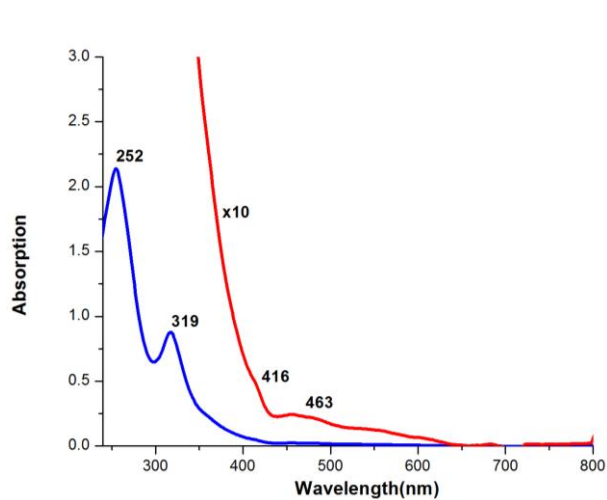


Figure S42 UV-vis absorption of compound 2a

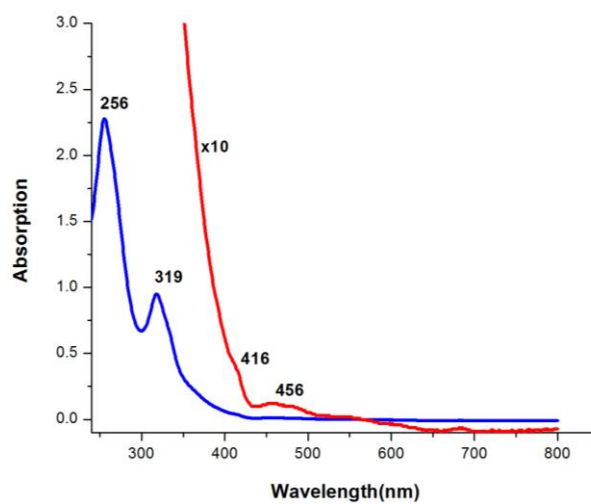


Figure S43 UV-vis absorption of compound 2b

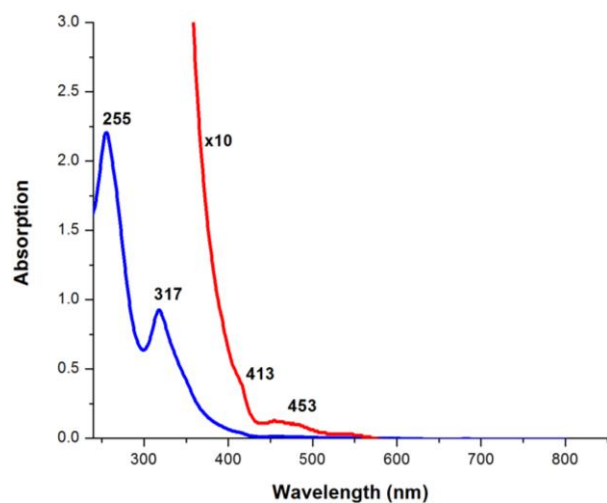


Figure S44 UV-vis absorption of compound 2c

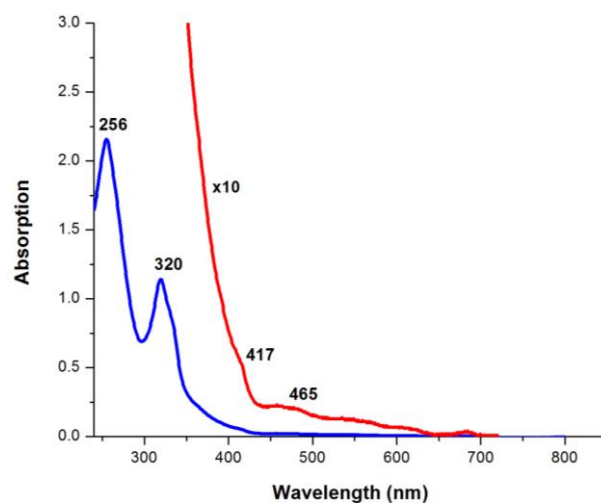


Figure S45 UV-vis absorption of compound 2d

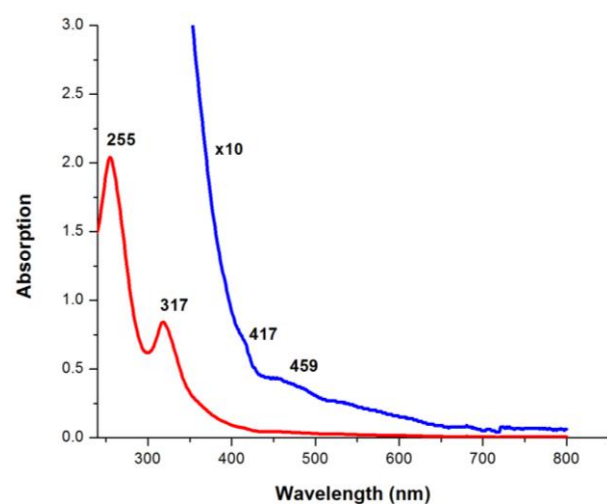


Figure S46 UV-vis absorption of compound 2e

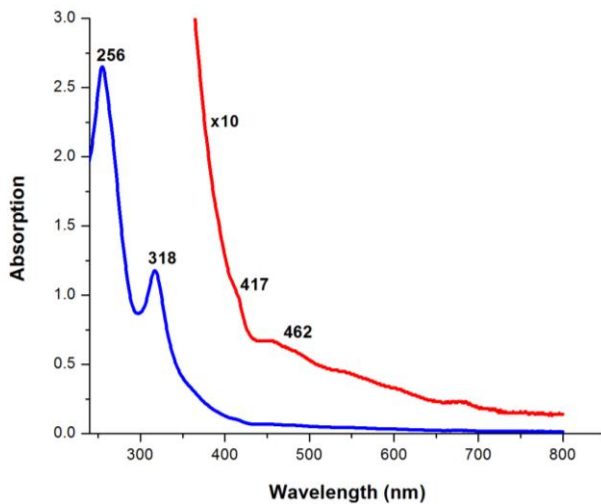


Figure S47 UV-vis absorption of compound 2f

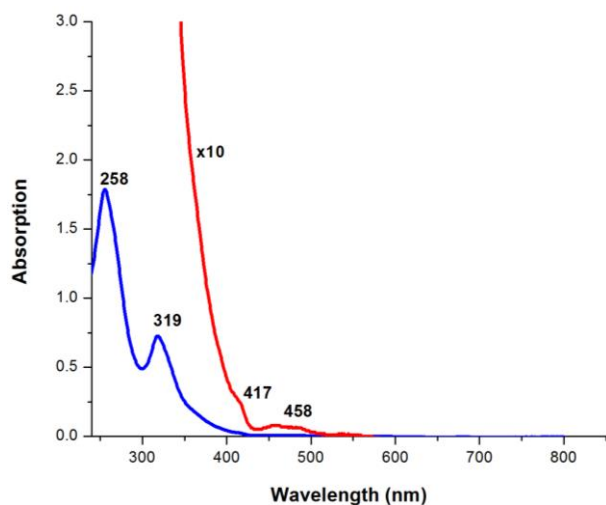


Figure S48 UV-vis absorption of compound 2g

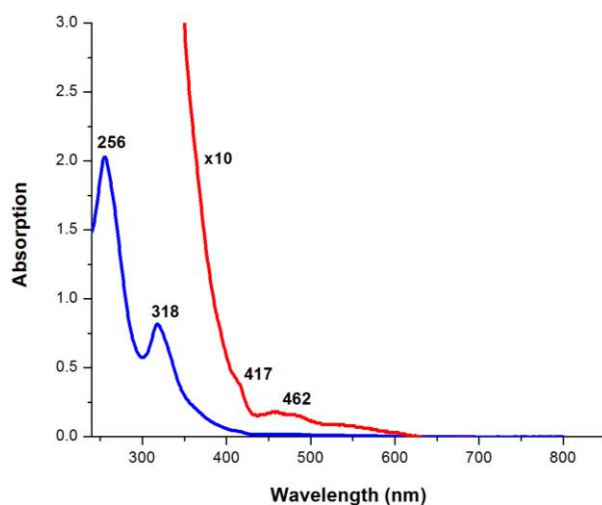


Figure S49 UV-vis absorption of compound 2h

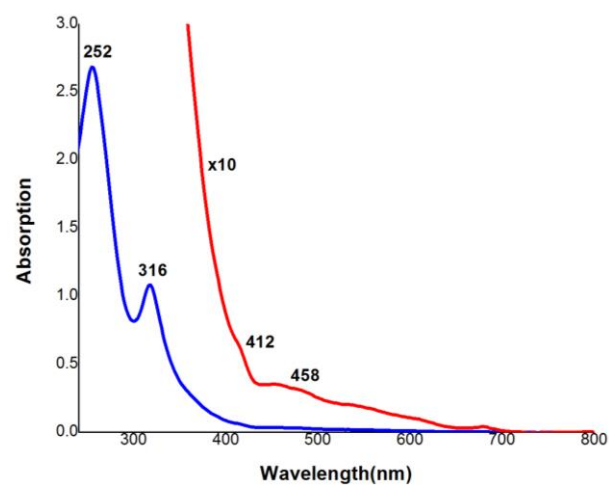


Figure S50 UV-vis absorption of compound 2i

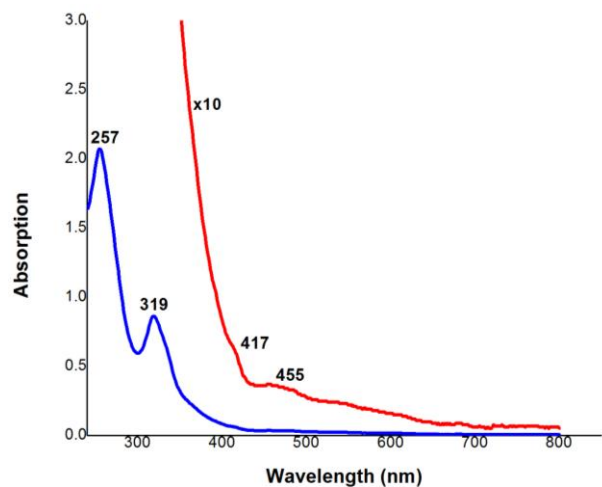


Figure S51 UV-vis absorption of compound 2j

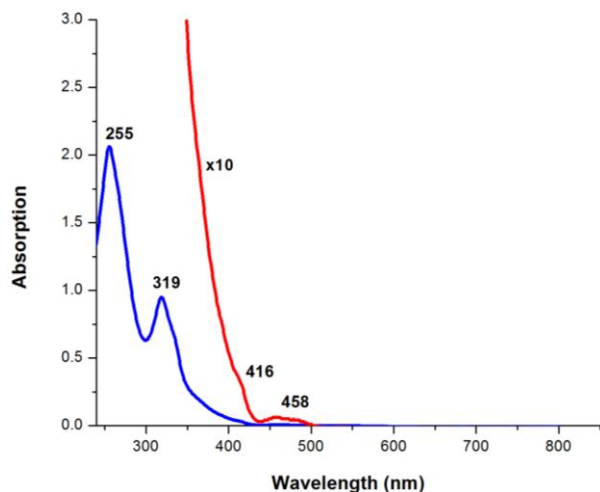


Figure S52 UV-vis absorption of compound 2k

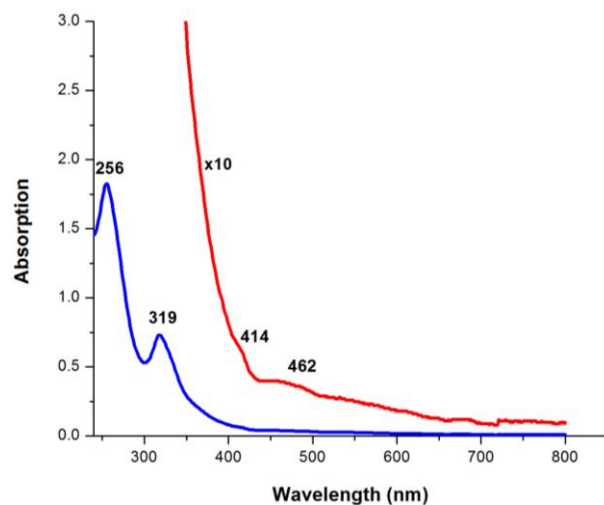


Figure S53 UV-vis absorption of compound 2l

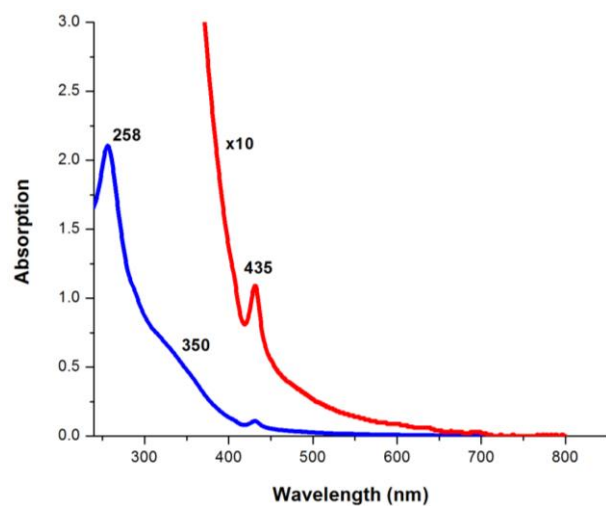


Figure S54 UV-vis absorption of compound 3a

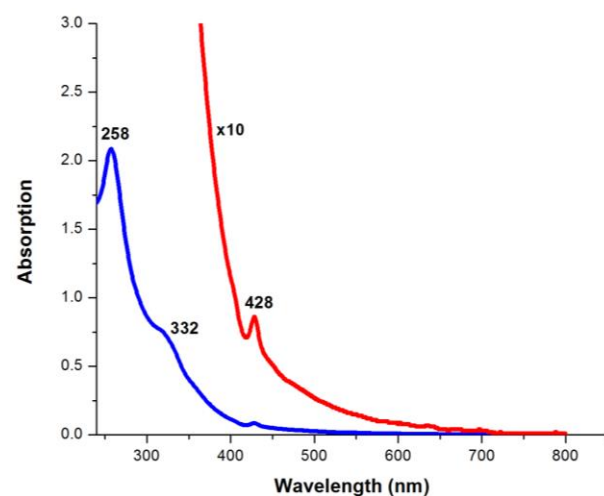


Figure S55 UV-vis absorption of compound 3b

6. CV of compounds 2a–l

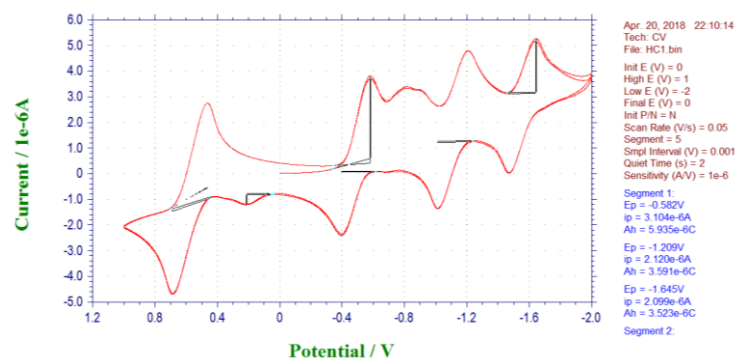


Figure S56 Cyclic voltammogram of compound 2a (scanning rate: 50 mV s⁻¹)

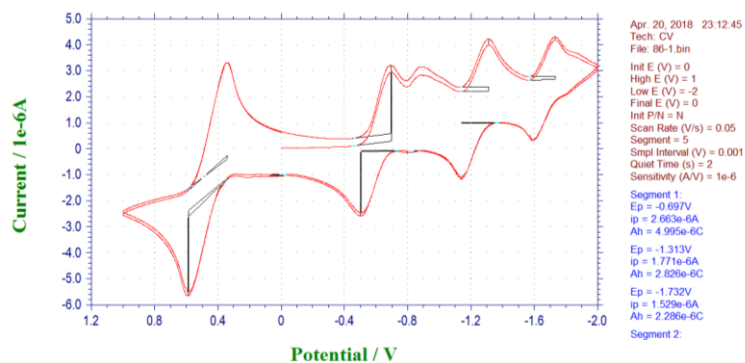


Figure S57 Cyclic voltammogram of compound 2b (scanning rate: 50 mV s⁻¹)

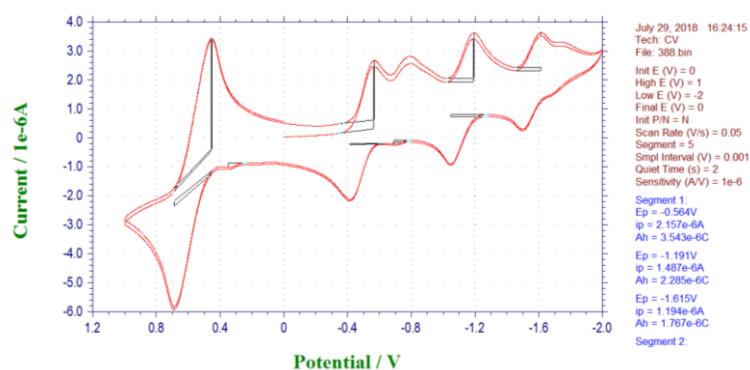


Figure S58 Cyclic voltammogram of compound 2c (scanning rate: 50 mV s⁻¹)

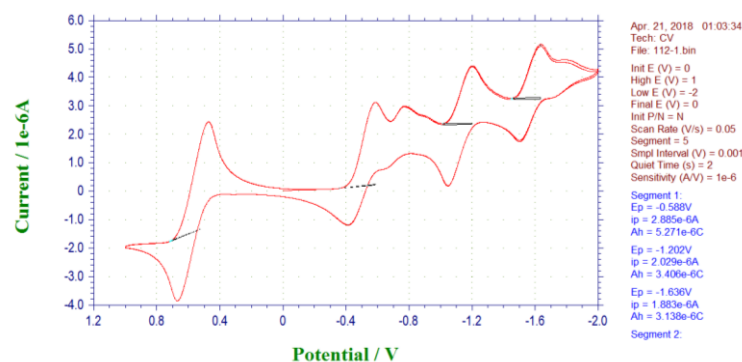


Figure S59 Cyclic voltammogram of compound 2d (scanning rate: 50 mV s⁻¹)

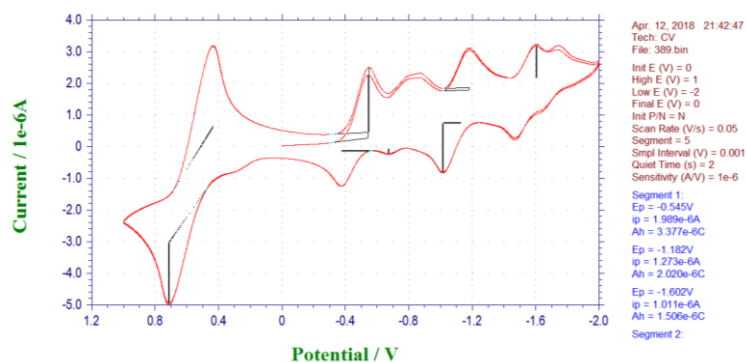


Figure S60 Cyclic voltammogram of compound 2e (scanning rate: 50 mV s⁻¹)

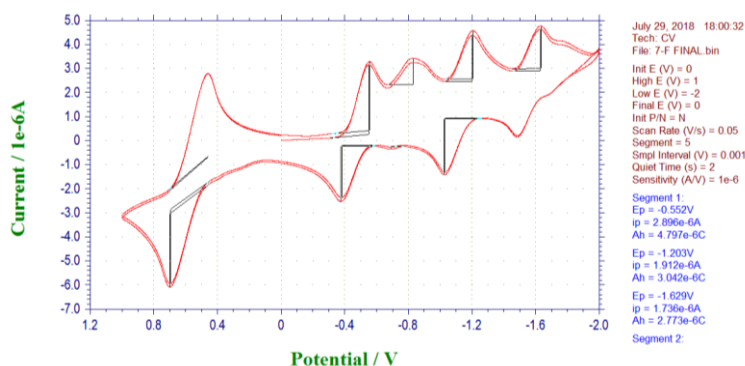


Figure S61 Cyclic voltammogram of compound 2f (scanning rate: 50 mV s⁻¹)

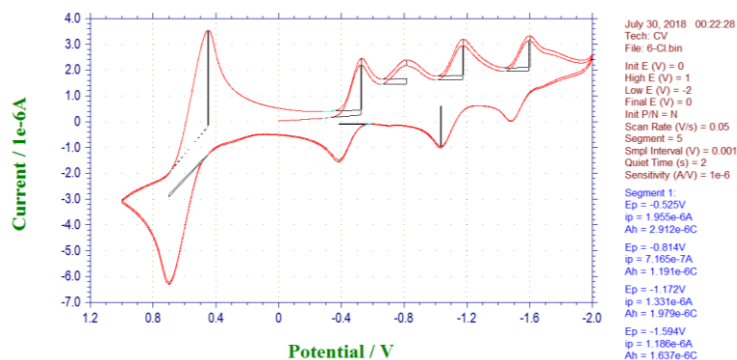


Figure S62 Cyclic voltammogram of compound 2g (scanning rate: 50 mV s⁻¹)

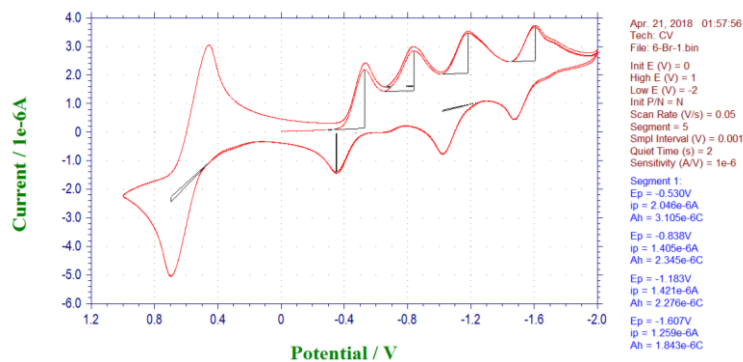


Figure S63 Cyclic voltammogram of compound 2h (scanning rate: 50 mV s⁻¹)

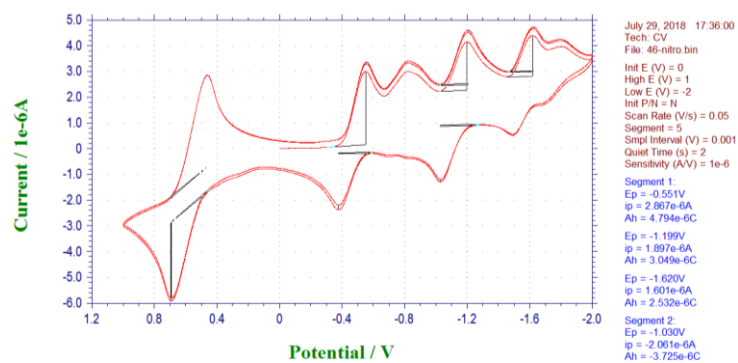


Figure S64 Cyclic voltammogram of compound 2i (scanning rate: 50 mV s⁻¹)

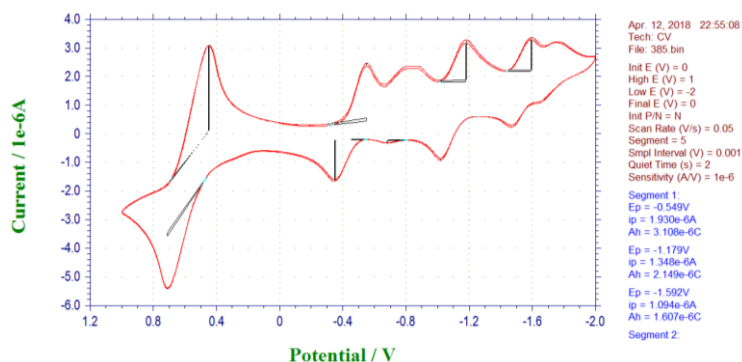


Figure S65 Cyclic voltammogram of compound 2j (scanning rate: 50 mV s⁻¹)

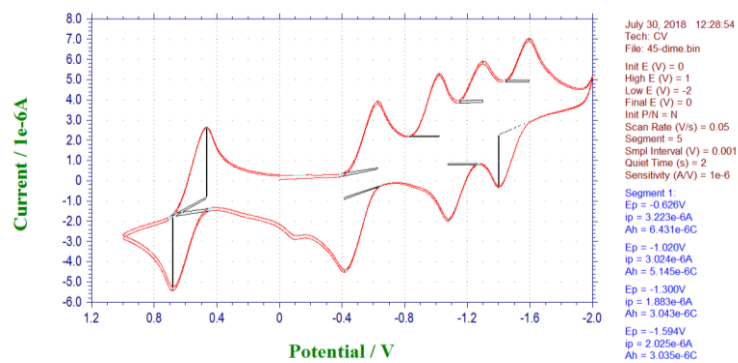


Figure S66 Cyclic voltammogram of compound 2k (scanning rate: 50 mV s⁻¹)

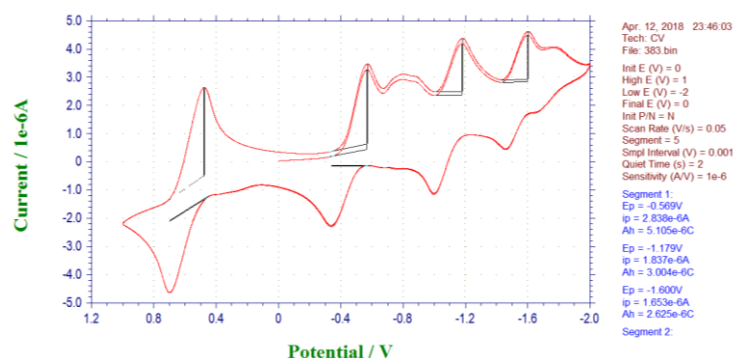


Figure S67 Cyclic voltammogram of compound 2l (scanning rate: 50 mV s⁻¹)

7. X-Ray data of compound 2l

Black block crystals of **2l** were obtained by slow evaporation of a saturated solution in carbon disulfide and *n*-hexane (1:2) at 4 °C. Single-crystal X-ray diffraction data were collected on a diffractometer (Gemini S Ultra, Agilent Technologies) equipped with a CCD area detector using graphite-monochromated Cu K α radiation ($\lambda = 1.54184$ Å) in the scan range $7.08^\circ < 2\theta < 142.47^\circ$. The structure was solved with direct methods using SHELXS-97 and refined with full-matrix least-squares refinement using the SHELXL-97 program within OLEX2.

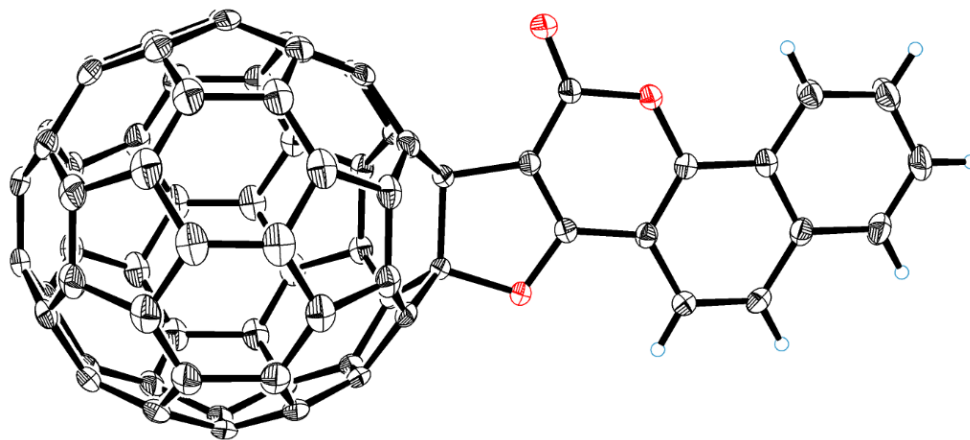


Figure S68 ORTEP diagrams of **2l** with 30% thermal ellipsoids. The carbon disulfide molecule is omitted for clarity.

Table S1. Crystal data and structure refinement for 2l	
Identification code	1864794
Empirical formula	C ₇₄ H ₆ O ₃ S ₂
Formula weight	1006.91
Temperature/K	291(2)
Crystal system	orthorhombic
Space group	Pbca
a/Å	18.1547(2)
b/Å	17.1668(2)
c/Å	24.9689(3)
α /°	90
β /°	90
γ /°	90
Volume/Å ³	7781.76(16)
Z	8
ρ_{calc} /cm ³	1.719
μ /mm ⁻¹	1.800
F(000)	4048.0
Crystal size/mm ³	0.300 × 0.270 × 0.260
Radiation	CuK α (λ = 1.54184)
2 Θ range for data collection/°	7.08 to 142.474
Index ranges	-15 ≤ h ≤ 21, -20 ≤ k ≤ 20, -21 ≤ l ≤ 30
Reflections collected	19182
Independent reflections	7358 [R_{int} = 0.0210, R_{sigma} = 0.0188]
Data/restraints/parameters	7358/0/712
Goodness-of-fit on F ²	1.033
Final R indexes [$I \geq 2\sigma(I)$]	R_1 = 0.0495, wR_2 = 0.1244
Final R indexes [all data]	R_1 = 0.0548, wR_2 = 0.1291
Largest diff. peak/hole / e Å ⁻³	0.62/-0.83