Photocatalyst-, metal- and additive- free, direct C-H arylation of quinoxalin-2(1*H*)-ones with aryl acyl peroxides

induced by visible light †

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Supporting Information

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1. Ineffective transformations



2. Fluorescence quenching experiment

The fluorescence quenching experiments were carried out, and the results were shown in Figure 1. In a typical experiment, to a 2.0 mL solution of 1-methyl-3-phenylquinoxalin-2(1H)-one (**3aa**) in acetone (2.0×10⁻⁶ mol/L) was added different amounts of BPO (**2a**). It was found that the emission intensity of 1-methyl-3-phenylquinoxalin-2(1H)-one has been decreased along with the increasing of the amount of BPO.

3. Computational details

Geometry optimizations of all stationary points were carried out at the M06-2X¹/6-311G(d,p) level. Vibrational frequency calculations were performed to identify that each of the species was a local minimum (no imaginary frequencies) or a transition state (one imaginary frequency). To confirm that each transition state connects the desired reactants and products along the reaction path, intrinsic reaction coordinate (IRC) calculations were conducted at the same level of theory. In order to obtain more accurate energies, single point energies were computed at the M06-2X/def2-TZVPP² level. The solvent effect of acetone was considered with SMD solvation model.³ All calculations were performed using the Gaussian 16 C.01 package.⁴

Α			
0	-2.43321200	-1.02213500	-0.00066200
С	-1.70639000	-0.00001000	0.00002800
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Cartesian coordinates of the optimized structures

B

С	0.00000000	1.39583600	0.00000000
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Н	-1.78479300	2.91613200	-2.12073900
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С	2.50705300	0.39046500	0.55727300
С	2.96781800	-0.13893500	1.76778900
С	3.35146500	0.37170400	-0.55995100
С	4.24465200	-0.67579400	1.86407400
Н	2.31303300	-0.12658100	2.63061200
С	4.62410100	-0.18302300	-0.46234700
Н	3.00255400	0.76656400	-1.50253400
С	5.07585300	-0.70407700	0.74696300
Н	4.59117700	-1.07413300	2.81060100
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С	-1.91605000	-1.10246900	-1.54132300
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С	-3.49949300	-2.37309400	-0.24596300
Н	-4.03732600	-0.77887400	-1.58883500
Н	-2.67387900	-3.89880700	1.02601500
Н	-4.52369400	-2.57533600	0.04644200

[1,2]-H shift

We have also calculated the [1,2]-H shift step and the structure of the transition state is shown in the following figure. The reaction barrier is calculated to be 36.9 kcal/mol.



Cartesian	coordinates	of the	transition	state

С	-4.33804200	-0.79431800	0.13678600
С	-3.56199200	0.35821700	0.20042700
С	-2.17669600	0.28734900	0.03298200
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С	-2.37807100	-2.11730200	-0.26312900
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С	-2.00632200	2.72983000	0.30624800
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Н	-2.49316600	2.74801600	1.28385000
Н	-1.23628000	3.49278700	0.27120700
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С	2.04679800	-0.16964000	-0.11000700
С	2.48700700	-1.36725000	0.46892300
С	3.00164400	0.76702100	-0.53396800
С	3.84477600	-1.61615500	0.63322100
Н	1.75635700	-2.09690300	0.79465100
С	4.35618300	0.51069400	-0.36770000
Н	2.67775300	1.69026400	-0.99339200
С	4.78548000	-0.67912800	0.21706700
Н	4.16698000	-2.54428900	1.09151200
Н	5.08205300	1.24266700	-0.70324600
Н	5.84409000	-0.87394700	0.34426100

4. References

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5. Copies of NMR spectra

3aa ¹H NMR





3ab ¹H NMR



3ab ¹³C NMR



3ac ¹H NMR



3ac ¹³C NMR



3ad ¹H NMR



3ad ¹³C NMR



3ae ¹H NMR



3ae ¹³C NMR



3af ¹H NMR



3af ¹³C NMR



3ag ¹H NMR



3ag ¹³C NMR





-44.79

3ah ¹H NMR



3ah ¹³C NMR





3ai ¹H NMR



3ai ¹³C NMR



3aj ¹H NMR



3aj ¹³C NMR



3ak ¹H NMR



3ak ¹³C NMR



3al ¹H NMR



3al ¹³C NMR





3am ¹H NMR



3am ¹³C NMR





3an ¹H NMR



3an ¹³C NMR





3ao ¹H NMR



3ao ¹³C NMR



3ap ¹H NMR



3ap ¹³C NMR



3aq ¹H NMR



3aq ¹³C NMR



3ar ¹H NMR



3ar ¹³C NMR



-29.45







3as ¹³C NMR







3at ¹³C NMR



-29.51





3au ¹³C NMR



3av ¹H NMR



3av ¹³C NMR

91	8 2 1 2 28
4 0	~ 0 0 0 0 04
12 12	5 5 5 5 5 E

29.29 20.68 19.24



3aw ¹H NMR



-29.85

3aw ¹³C NMR

54.41	54.38	54.28	50.63	50.51	45.90	35.51	30.72	30.64	30.57	30.56	29.54	29.32	28.16	17.94	b2:23	02.11
1	7	-	T.	-	-	-	-	-	-	-	-	Tr	5	2	1	5





3ax ¹³C NMR



3ba ¹H NMR



3ba ¹³C NMR



3bb ¹H NMR



3bb ¹³C NMR



3bc ¹H NMR



3bc ¹³C NMR



3bd ¹H NMR



3bd ¹³C NMR



3be ¹H NMR



3be ¹³C NMR



3bf ¹H NMR



3bf ¹³C NMR



3bg ¹H NMR



3bg ¹³C NMR





3bh ¹H NMR



3bh ¹³C NMR



3bi ¹H NMR



3bi ¹³C NMR





3bj ¹H NMR



3bj ¹³C NMR



3bk ¹H NMR



3bk ¹³C NMR





3bl ¹³C NMR





3bm ¹³C NMR





3bn ¹H NMR



3bn ¹³C NMR



3bo ¹H NMR



3bo ¹³C NMR



3bp ¹H NMR



3bp ¹³C NMR



3bq ¹H NMR



3bq ¹³C NMR



7¹H NMR



7¹³C NMR





6. GC-MS spectra of benzoic acid

