

**Poly(Ethylene Glycol)-Poly(Propylene Glycol)-Poly(Ethylene Glycol) and Polyvinylidene Fluoride Blend Doped with the Oxydianiline Based Thiourea Derivatives as a Novel and Modest Gel Electrolyte System for Dye-Sensitized Solar Cell Applications**

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## 1. Synthesis of organic additives:

### 1.1. Synthesis of OPMT

4-Methoxyphenyl isothiocyanate (2.2 mmol) with 4,4'-oxydianiline (1 mmol) was stirred in anhydrous methanol at room temperature for 2-3 days. The product obtained was white precipitate. It was then filtered, washed with methanol and dried under vacuum to obtain **1,1'-(oxybis(4,1-phenylene))bis(3-(4-methoxyphenyl)thiourea)** (**OPMT**) as white solid with 78% yield. The relevant scheme is given in the Figure S1. <sup>1</sup>H NMR (400 MHz, DMSO d<sub>6</sub>) δ 9.62 (s, 2H), 9.58 (s, 2H), 7.46 (d, *J* = 8.8 Hz, 4H), 7.33 (d, *J* = 8.8 Hz, 4H), 6.99 (d, *J* = 8.8 Hz, 4H), 6.92 (d, *J* = 8.8 Hz, 4H), 3.75 (s, 6H); <sup>13</sup>C NMR (100 MHz, DMSO d<sub>6</sub>) δ 180.56, 157.07, 154.06, 135.37, 132.61, 126.63, 126.43, 118.90, 114.15, 55.71. HRMS (ESI): calcd. For C<sub>28</sub>H<sub>26</sub>N<sub>4</sub>O<sub>3</sub>S<sub>2</sub> [M+H]<sup>+</sup> 530.1446 found to be 531.1519.

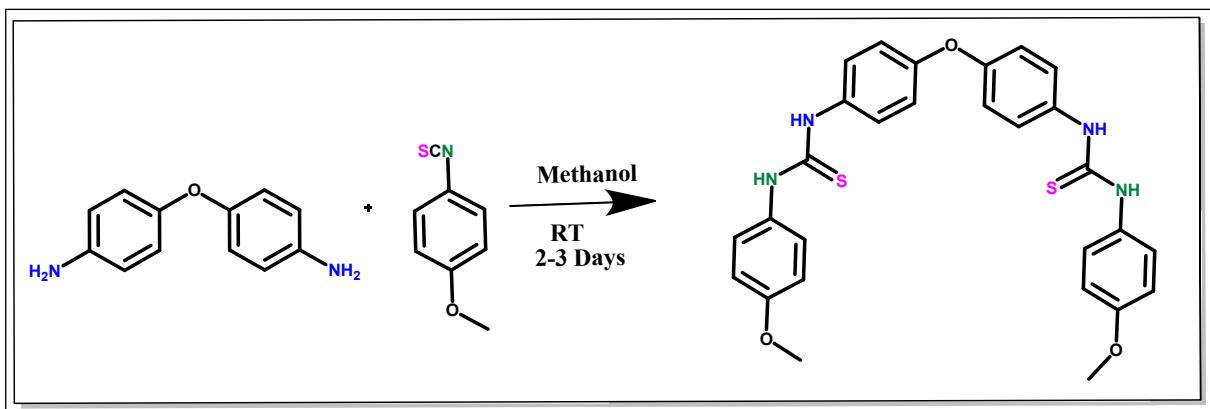
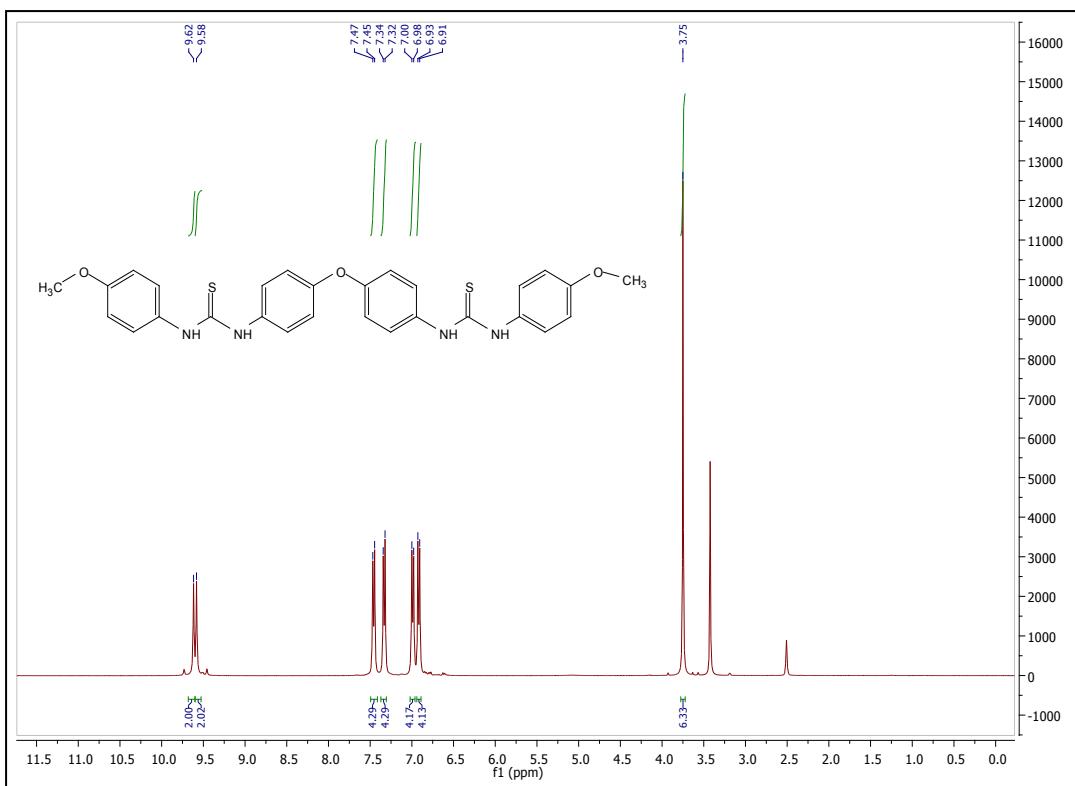
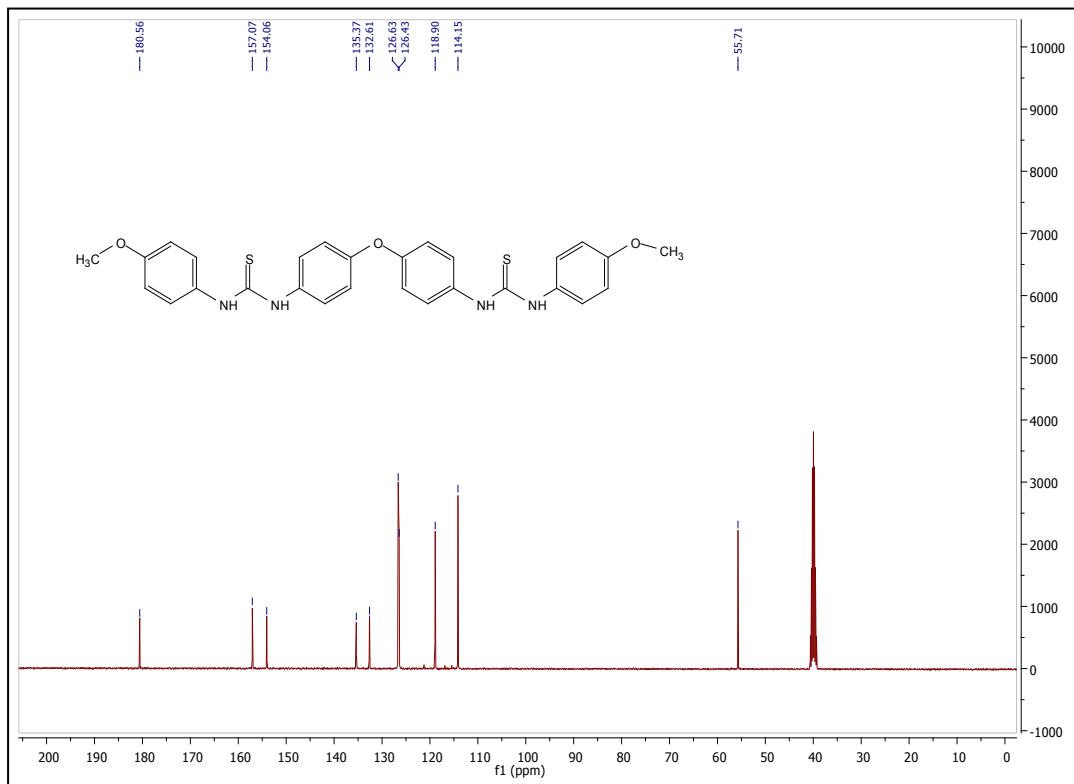


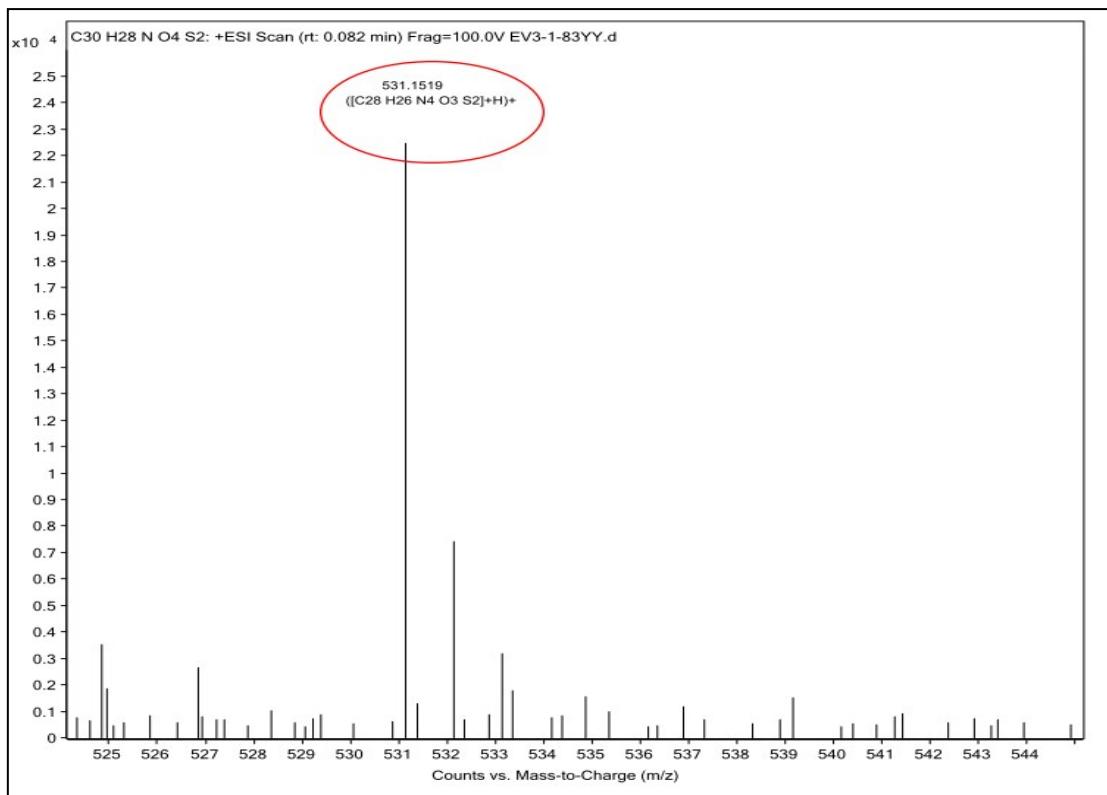
Figure S1. Scheme for the synthesis of organic additive-1 (OPMT)



**Figure S2.**  $^1\text{H}$  NMR spectrum of OPMT



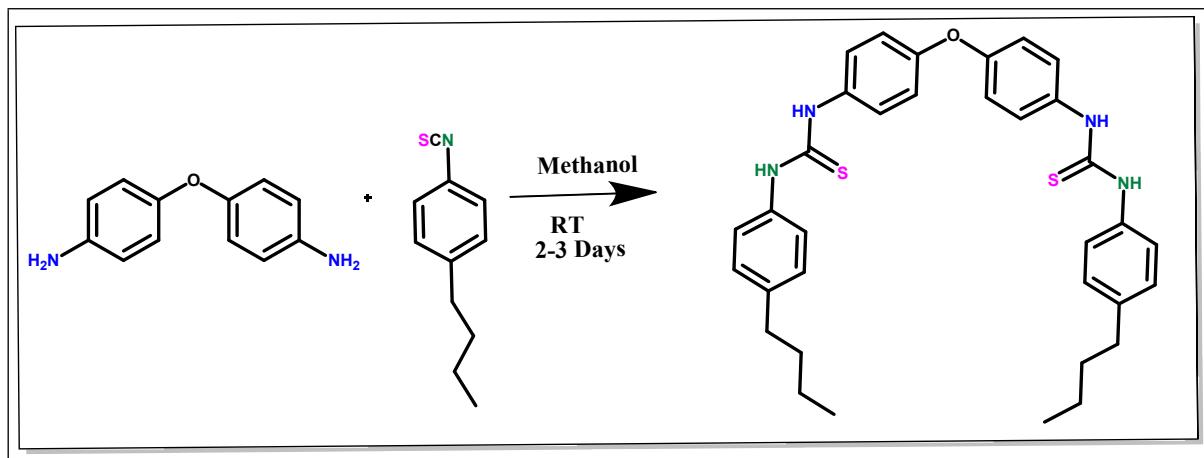
**Figure S3.**  $^{13}\text{C}$  NMR spectrum of OPMT



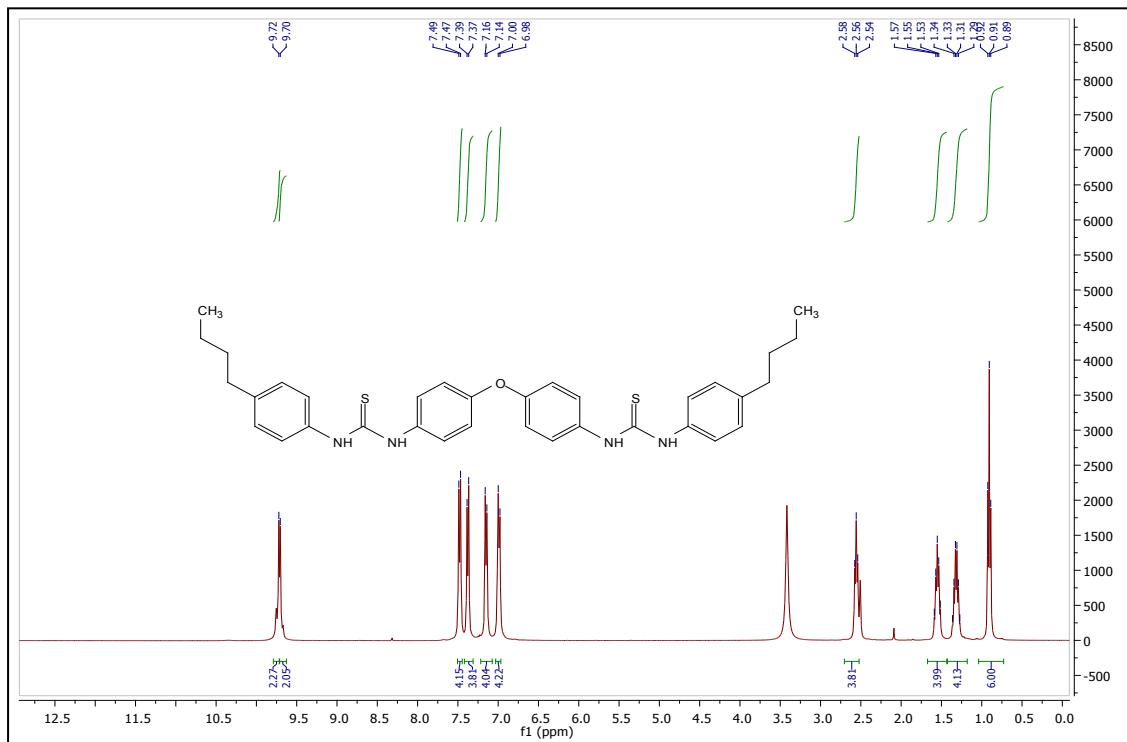
**Figure S4. HRMS spectrum of OPMT**

## 1.2. Synthesis of OPBT

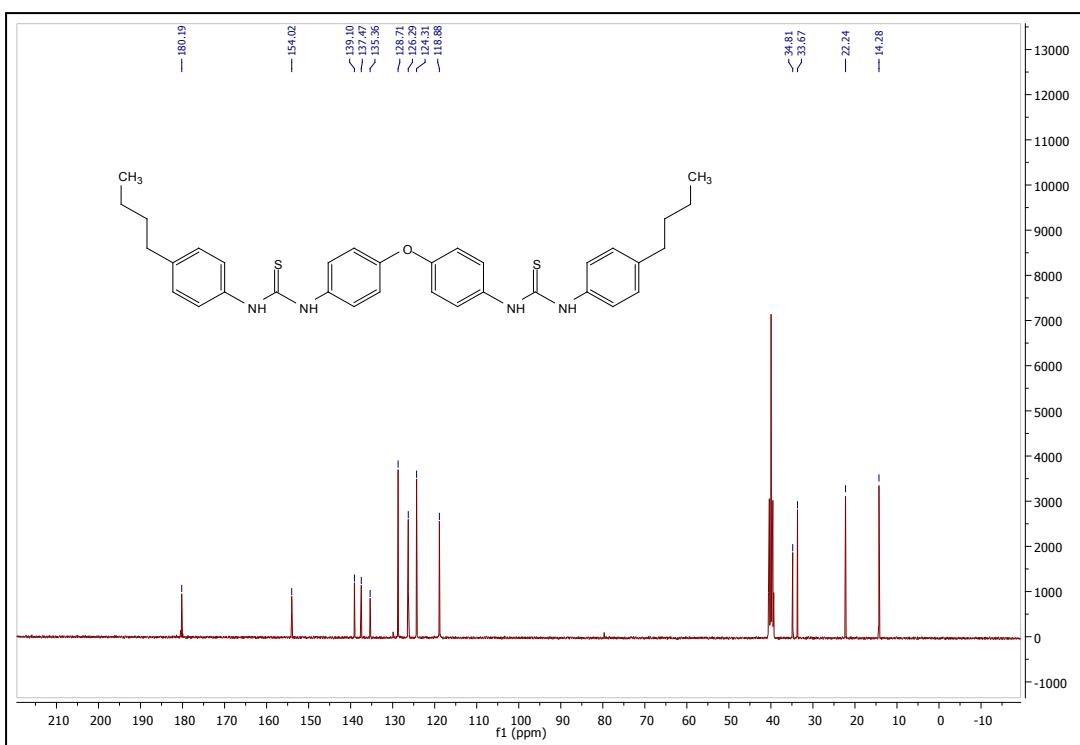
4-n-Butylphenyl isothiocyanate (2.2 mmol) with 4,4'-oxydianiline (1 mmol) was stirred in anhydrous methanol at room temperature for 2-3 days. The product obtained was white precipitate which was then filtered, washed with methanol and dried under vacuum to obtain **1,1'-(oxybis(4,1-phenylene))bis(3-(4-butylphenyl)thiourea)(OPBT)** as a white solid with 75% yield. The relevant scheme is given in the Figure S5.<sup>1</sup>H NMR (400 MHz, DMSO d<sub>6</sub>) δ 9.72 (s, 2H), 9.70 (s, 2H), 7.48 (d, J = 8.6 Hz, 4H), 7.38 (d, J = 8.1 Hz, 4H), 7.15 (d, J = 8.1 Hz, 4H), 6.99 (d, J = 8.6 Hz, 4H), 2.55 (m, 4H), 1.57 – 1.43 (m, 4H), 1.32 (m, 4H), 0.91 (t, J = 7.3 Hz, 6H), <sup>13</sup>CNMR (100MHz, DMSO d<sub>6</sub>) δ 180.19, 154.02, 139.10, 137.47, 135.36, 128.71, 126.29, 124.31, 118.88, 34.81, 33.67, 22.24, 14.28; HRMS (ESI):calcd. For C<sub>34</sub>H<sub>38</sub>N<sub>4</sub>OS<sub>2</sub> [M+H]<sup>+</sup> 582.2487 found to be 583.2560.



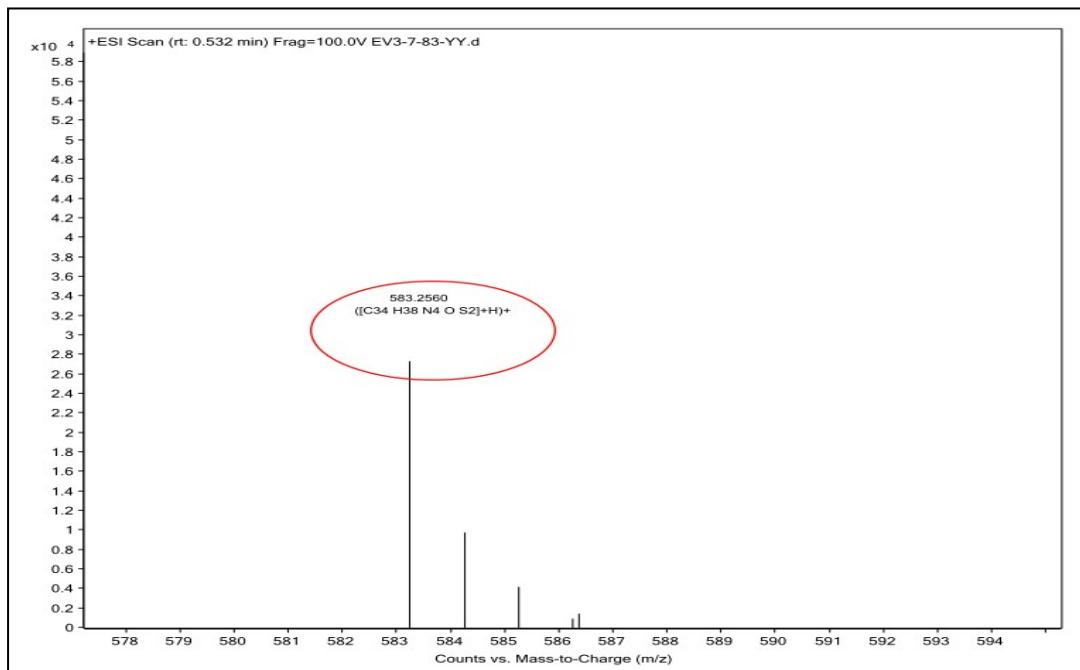
**Figure S5. Scheme for the synthesis of organic additive-2 (OPBT)**



**Figure S6. <sup>1</sup>H NMR spectrum of OPBT**



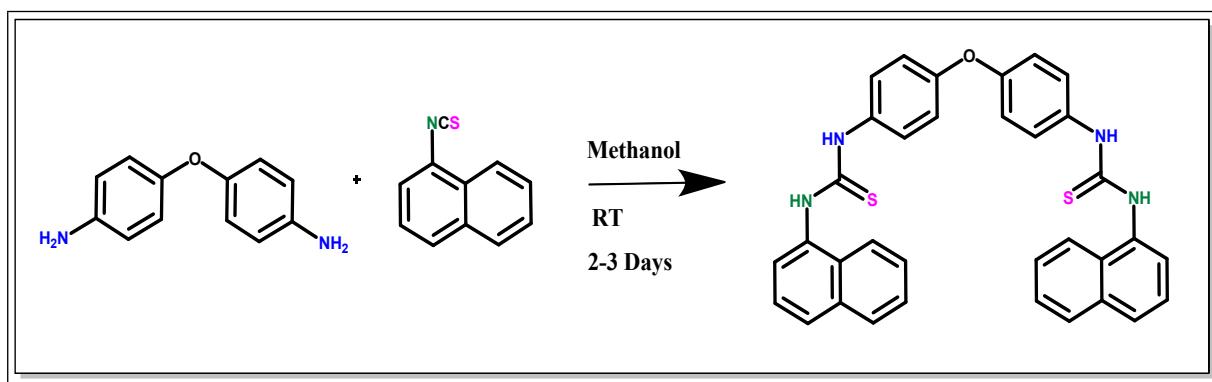
**Figure S7.**  $^{13}\text{C}$  NMR spectrum of OPBT



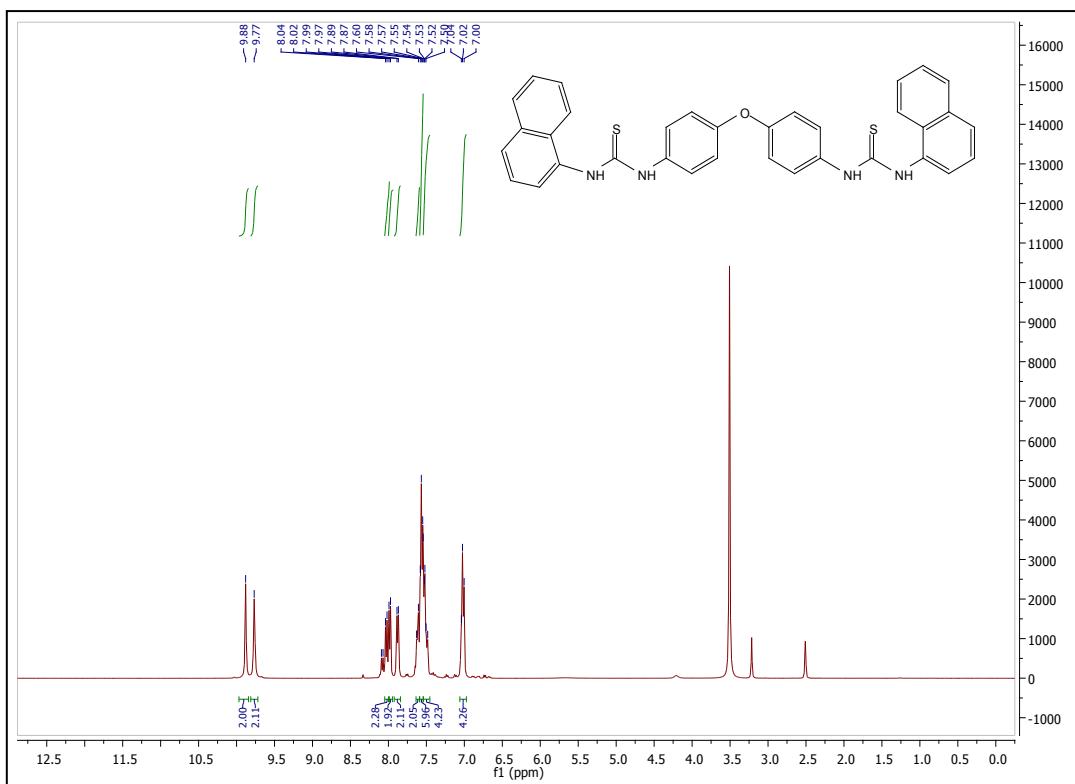
**Figure S8.** HRMS spectrum of OPBT

### 1.3. Synthesis of OPNT

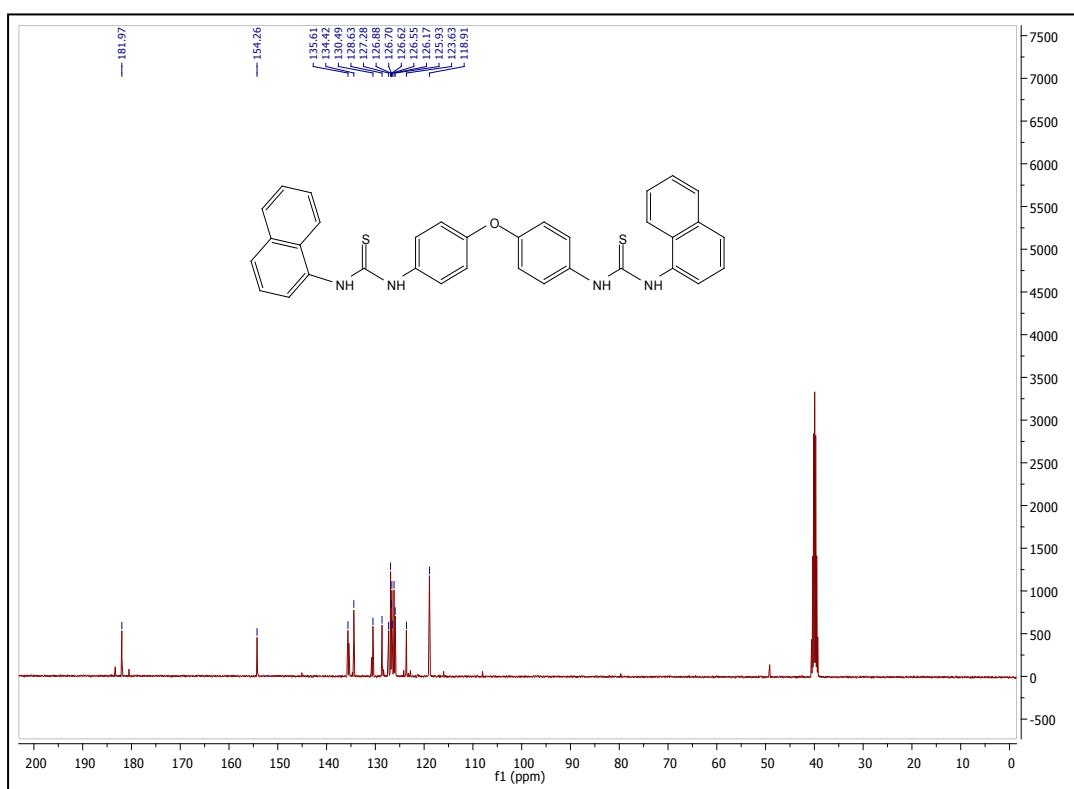
1-Naphthyl isothiocyanate (2.2 mmol) with 4,4'-oxydianiline (1 mmol) was stirred in anhydrous methanol at room temperature for 2-3 days to obtain the product as white precipitate. It was then filtered, washed with methanol and dried under vacuum to obtain **1,1'-(oxybis(4,1-phenylene))bis(3-(naphthalen-1-yl)thiourea)(OPNT)** as white solid with 70% yield. The relevant scheme is given in the Figure S9.  $^1\text{H}$  NMR (400 MHz, DMSO  $\text{d}_6$ )  $\delta$  9.88 (s, 2H), 9.77 (s, 2H), 8.05 – 7.99 (m, 2H), 7.98 (d,  $J$  = 7.9 Hz, 2H), 7.88 (d,  $J$  = 7.3 Hz, 2H), 7.62 (d,  $J$  = 9.8 Hz, 2H), 7.58-7.54 (m, 6H), 7.53-7.48 (m, 4H), 7.02 (t,  $J$  = 7.3 Hz, 4H);  $^{13}\text{C}$  NMR (100 MHz, DMSO  $\text{d}_6$ )  $\delta$  181.97, 154.26, 135.61, 134.42, 130.49, 128.63, 127.28, 126.88, 126.70, 126.62, 126.55, 126.17, 125.93, 123.63, 118.91; HRMS (ESI): calcd. For  $\text{C}_{34}\text{H}_{26}\text{N}_4\text{OS}_2[\text{M}+\text{H}]^+$  570.1548 found to be 571.1621.



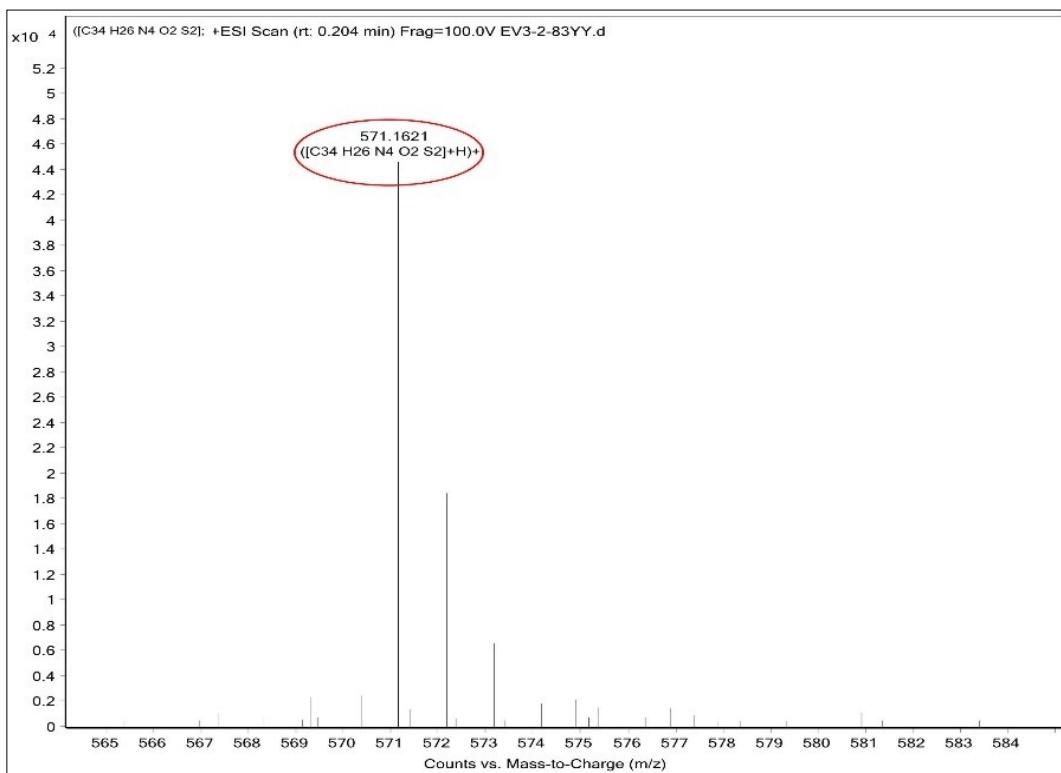
**Figure S9. Scheme for the synthesis of organic additive-3 (OPNT)**



**Figure S10.**  $^1\text{H}$  NMR spectrum of OPNT



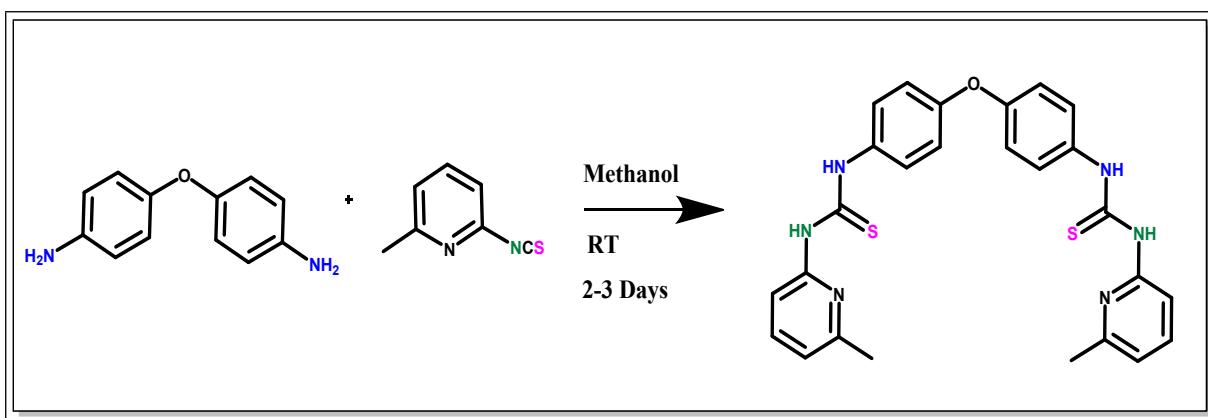
**Figure S11.**  $^{13}\text{C}$  NMR spectrum of OPNT



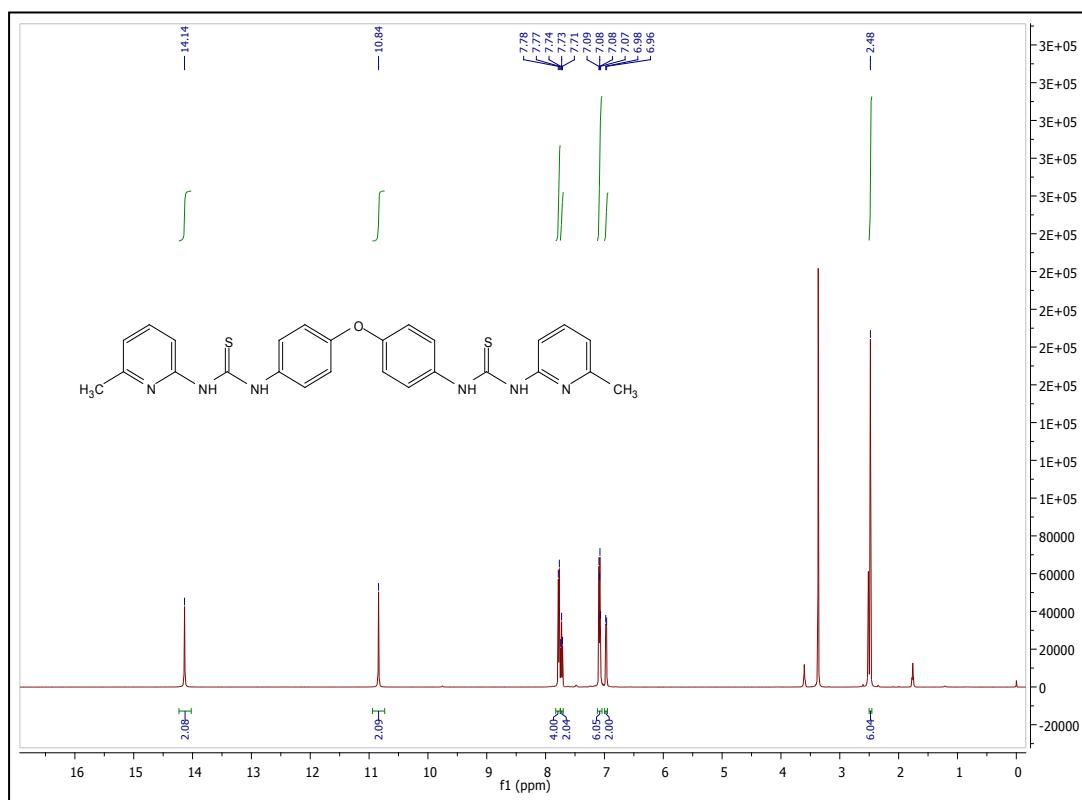
**Figure S12. HRMS spectrum of OPNT**

#### 1.4. Synthesis of OPPT

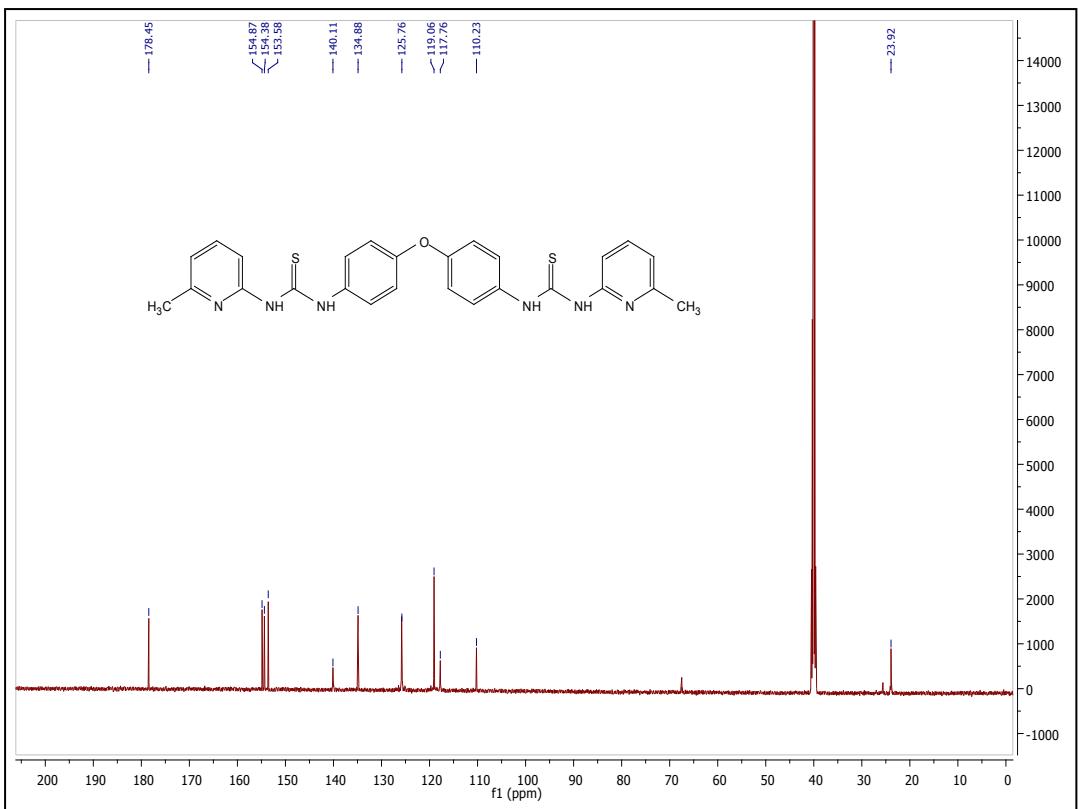
2-isothiocyanato-6-methylpyridine was synthesized by already reported procedure.<sup>48</sup> 2-isothiocyanato-6-methylpyridine (2.2 mmol) with 4,4'-oxydianiline (1 mmol) was stirred in anhydrous methanol at room temperature for 2-3 days to obtain the product as white precipitate. It was then filtered, washed with methanol and dried under vacuum to obtain **1,1'-(oxybis(4,1-phenylene))bis(3-(6-methylpyridin-2-yl)thiourea)** (**OPPT**) with 85% yield. The relevant scheme is given in the Figure S13. <sup>1</sup>H NMR (500 MHz, DMSO d<sub>6</sub>) δ 14.14 (s, 2H), 10.84 (s, 2H), 7.78 (d, J = 8.8 Hz, 4H), 7.73 (t, J = 7.9 Hz, 2H), 7.09-7.07 (m, 6H), 6.97 (d, J = 7.4 Hz, 2H), 2.48 (s, 6H); <sup>13</sup>C NMR (125 MHz, DMSO d<sub>6</sub>) δ 178.45, 154.87, 153.58, 140.11, 134.88, 125.76, 119.06, 117.76, 110.23, 54.38, 23.92. HRMS (ESI): calcd. For C<sub>26</sub>H<sub>24</sub>N<sub>6</sub>OS<sub>2</sub> [M+H]<sup>+</sup> 500.1453 found to be 501.1526.



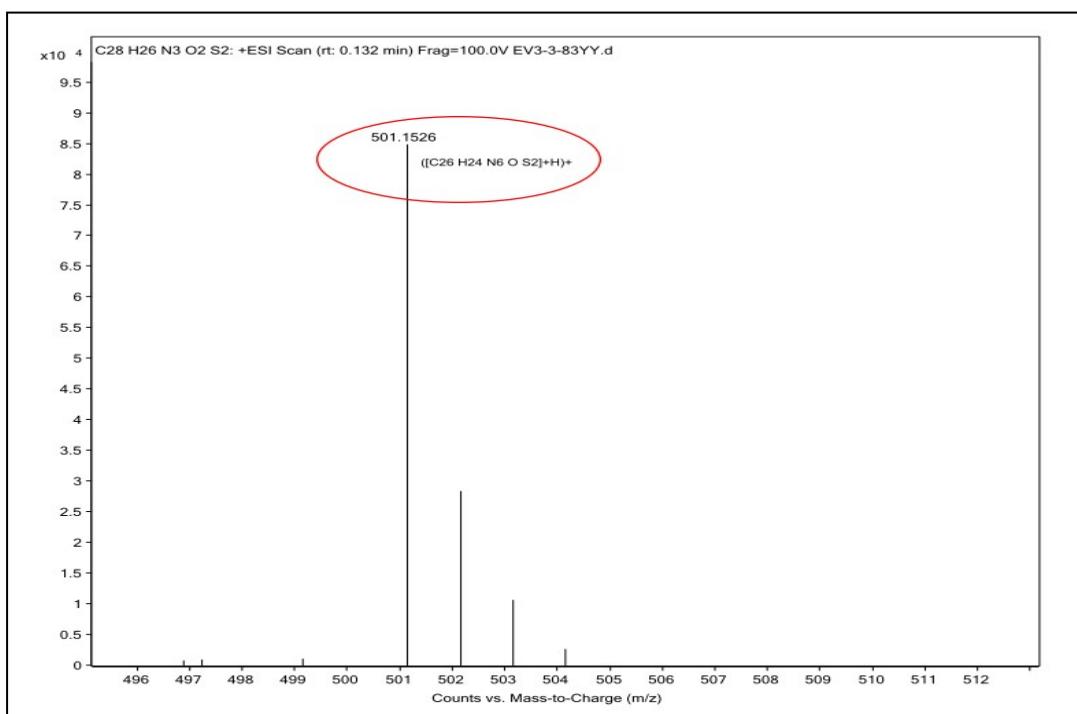
**Figure S13.** Scheme for the synthesis of organic additive-4 (OPPT)



**Figure S14.** <sup>1</sup>H NMR spectrum of OPPT



**Figure S15.**  $^{13}\text{C}$  NMR spectrum of OPPT



**Figure S16. HRMS spectrum of OPPT**

**Table S1: Statistical analysis of the efficiency of the fabricated DSSC with gel polymer electrolyte system (Device-1)**

S. No	Gel polymer electrolyte system	$J_{sc}$ (mA cm <sup>-2</sup> )	$V_{oc}$ (mV)	FF	$\eta$ %
1	Device-1	10.4	755	0.51	4
2	Device-1	10.5	746	0.51	3.99
3	Device-1	10.3	756	0.51	3.97
Average(Mean)		10.4	752.3333	0.51	3.986667
Standard Deviation		0.08167	4.4969	0	0.0123

**Table S2: Statistical analysis of the efficiency of the fabricated DSSC with gel polymer electrolyte system (Device-2)**

S.No	Gel polymer electrolyte system	$J_{sc}$ (mA cm <sup>-2</sup> )	$V_{oc}$ (mV)	FF	$\eta$ %
1	Device-2	13.6	785	0.52	5.55
2	Device-2	13.5	787	0.52	5.52
3	Device-2	13.7	778	0.52	5.54
Average(Mean)		13.6	783.3333	0.52	5.536667
Standard Deviation		0.08167	3.8584	0	0.0123

**Table S3: Statistical analysis of the efficiency of the fabricated DSSC with gel polymer electrolyte system (Device-3)**

S.No	Gel polymer electrolyte system	$J_{sc}$ (mA cm <sup>-2</sup> )	$V_{oc}$ (mV)	FF	$\eta$ %
1	Device-3	11.1	765	0.51	4.33
2	Device-3	11.2	758	0.51	4.32
3	Device-3	11.3	752	0.51	4.33
Average(Mean)		11.2	758.3333	0.51	4.326667
Standard Deviation		0.08167	5.3124	0	0.00471

**Table S4: Statistical analysis of the efficiency of the fabricated DSSC with gel polymer electrolyte system (Device-4)**

S.No	Gel polymer electrolyte system	$J_{sc}$ (mA cm <sup>-2</sup> )	$V_{oc}$ (mV)	FF	$\eta$ %
1	Device-4	12.2	775	0.52	4.91
2	Device-4	12.1	776	0.52	4.88
3	Device-4	12.1	773	0.52	4.86
Average(Mean)		12.13333	774.6667	0.52	4.883333
Standard Deviation		0.04711	1.2471	0	0.0649

**Table S5: Statistical analysis of the efficiency of the fabricated DSSC with gel polymer electrolyte system (Device-5)**

S.No	Gel polymer electrolyte system	$J_{sc}$ (mA cm <sup>-2</sup> )	$V_{oc}$ (mV)	FF	$\eta$ %
1	Device-5	14	790	0.52	5.75
2	Device-5	14.1	784	0.52	5.74
3	Device-5	14.2	777	0.52	5.73
Average(Mean)		14.1	783.6667	0.52	5.74
Standard Deviation		0.08167	5.3124	0	0.00816

**Table S6: Statistical analysis of the efficiency of the fabricated DSSC with gel polymer electrolyte system (Device-6)**

S.No	Gel polymer electrolyte system	$J_{sc}$ (mA cm <sup>-2</sup> )	$V_{oc}$ (mV)	FF	$\eta$ %
1	Device-6	12.6	780	0.51	5.01
2	Device-6	12.5	781	0.51	4.98
3	Device-6	12.4	782	0.51	4.94
Average(Mean)		12.5	781	0.51	4.976667
Standard Deviation		0.08167	0.8164	0	0.0283

**Table S7: Photovoltaic parameters of the fabricated DSSC using prepared GPE's utilized for the stability test.**

Device	J <sub>sc</sub> (mA)	V <sub>oc</sub> (mV)	ff	η (%)
Device-1	10.4	755	0.51	4 %
Device-2	13.6	785	0.52	5.55%
Device-3	11.1	765	0.51	4.33%
Device-4	12.2	775	0.52	4.91%
Device-5	14	790	0.52	5.75%
Device-6	12.6	780	0.51	5.01%