

Supplementary Information

Synthesis of Fe₃O₄ nanocomposites for efficient separation of ultra-small oil droplets from hexadecane–water emulsions

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1. Characterization of Fe₃O₄@PDDA

PDDA-modified Fe₃O₄ NPs were prepared using previously reported methods [1]. The XRD diagram for Fe₃O₄@PDDA (Fig. S1) shows that the material has diffractions of (220), (311), (400), (511), (440) and is consistent with the standard card (JCPDS NO. 19-0629), indicating the cubic spinel structure of Fe₃O₄.

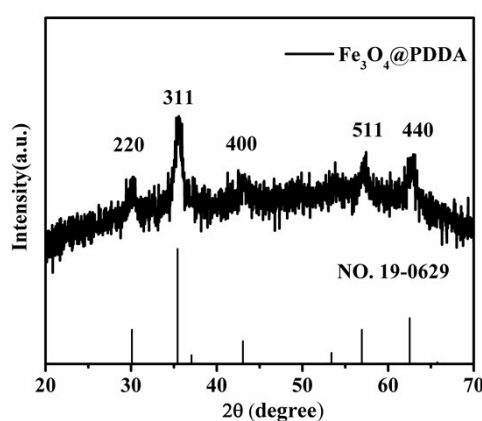


Fig. S1 XRD patterns of Fe₃O₄@PDDA samples.

The TEM image (Fig. S2A) shows that the Fe₃O₄@PDDA NPs are agglomerate and non-spherical. The SEM image (Fig. S2B) confirmed that the samples had

particle sized about 20–40 nm. The results of DLS (Fig. S3) showed that the average particle size of Fe₃O₄@PDDA NPs was 138.4 (±11.5) nm.

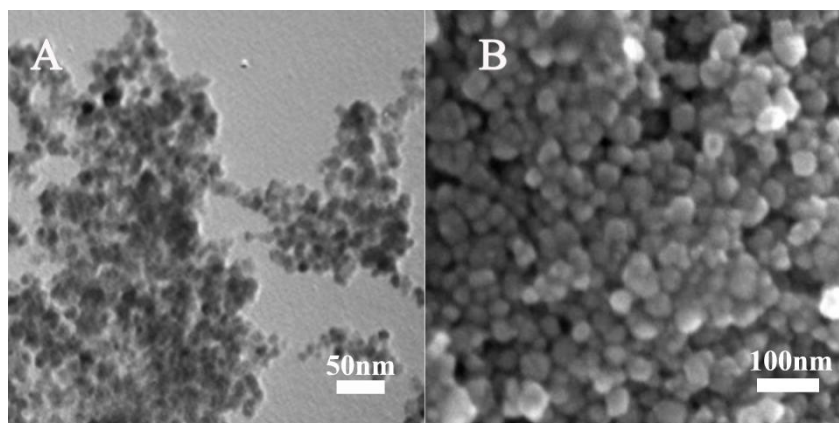


Fig. S2 TEM image and SEM image of Fe₃O₄@PDDA samples

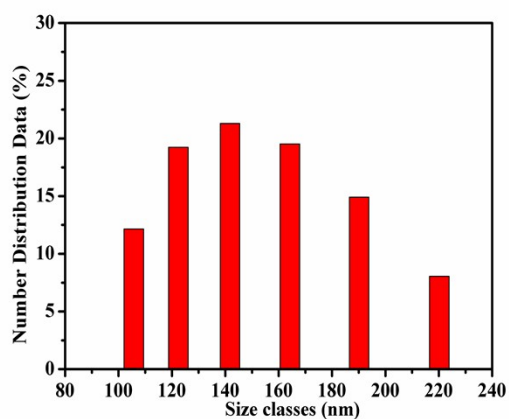


Fig. S3 Particle size distribution of Fe₃O₄@PDDA NPs

2. Particle size distribution of emulsion

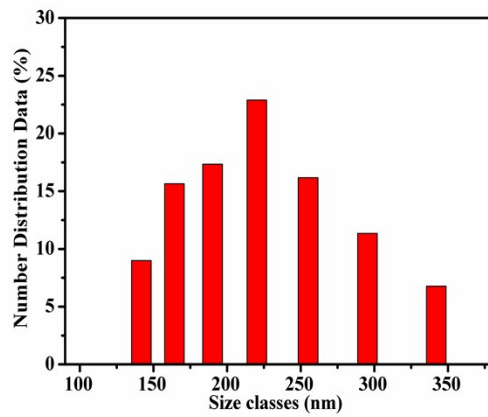


Fig. S4 Particle size distribution of hexadecane-water emulsion

3. Oil-water separation performance of magnetic nanoparticles (MNPs)

Fig. S5 shows that E_S reaches equilibrium when the shaking time was 150 min. For the hexadecane-water emulsion without MNPs, the change of E_S is almost zero with the change of shaking time, which also shows that the emulsion is stable in the range of shaking time.

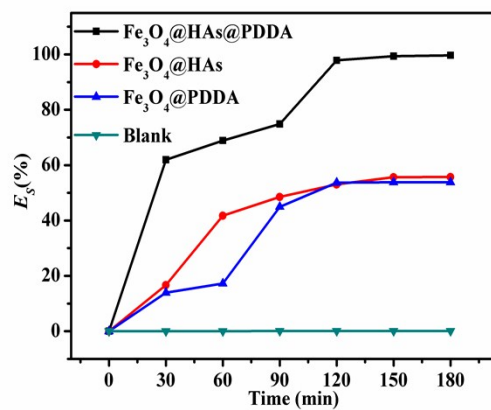


Fig. S5 The effect of magnetic nanoparticle shaking time on E_S in O/W emulsion

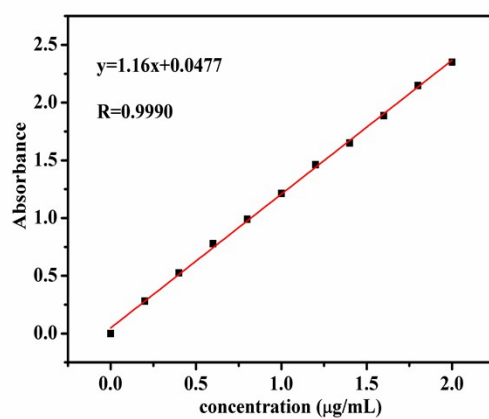


Fig. S6 Standard curve of hexadecane emulsion

References:

- 1 J. Chen, Y. Ju and H. Chen, *NANO*, 2019, **14**, 1950019-1950033.