

Electronic Supplementary Information (ESI)

Polymorphism of $\text{Au}_{11}(\text{PR}_3)_7\text{Cl}_3$ clusters: Understanding C-H $\cdots\pi$ interaction and C-H $\cdots\text{Cl}$ -C van der Waals interaction on cluster assembly by surface modification

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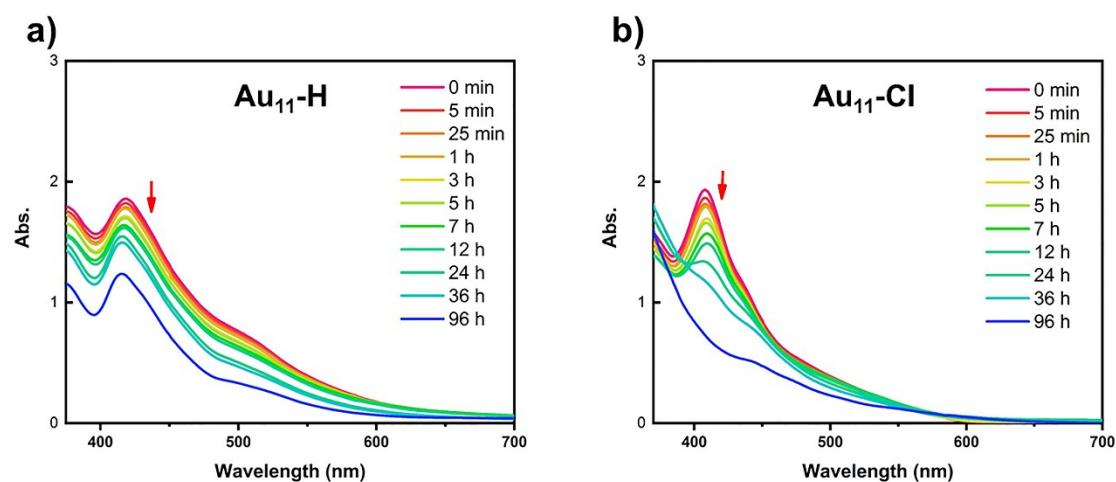


Figure S1. UV-vis spectra confirming the thermal stability of **Au₁₁-H** (a), **Au₁₁-Cl** (b) over time.

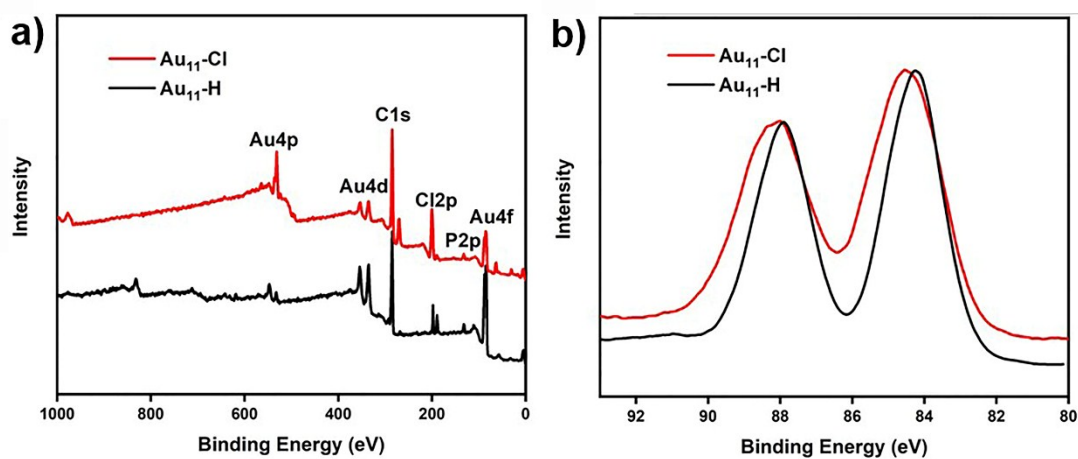


Figure S2. a) X-ray photoelectron spectra of **Au₁₁-Cl** and **Au₁₁-H**; b) XPS of Au4f in the **Au₁₁-Cl** and **Au₁₁-H**.

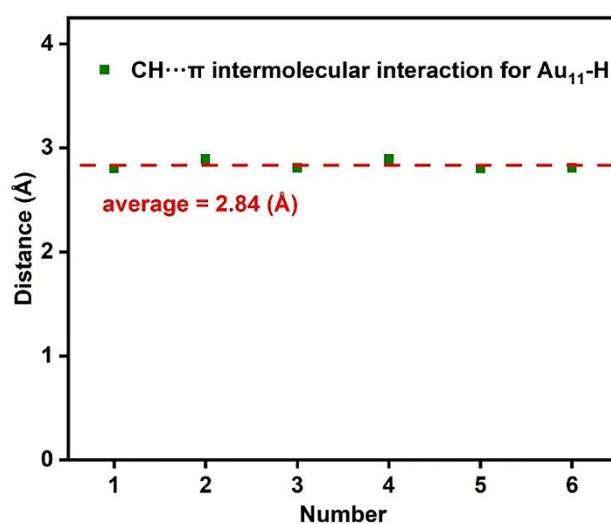


Figure S3. The statistical histogram of the CH... π interactions for **Au₁₁-H**.

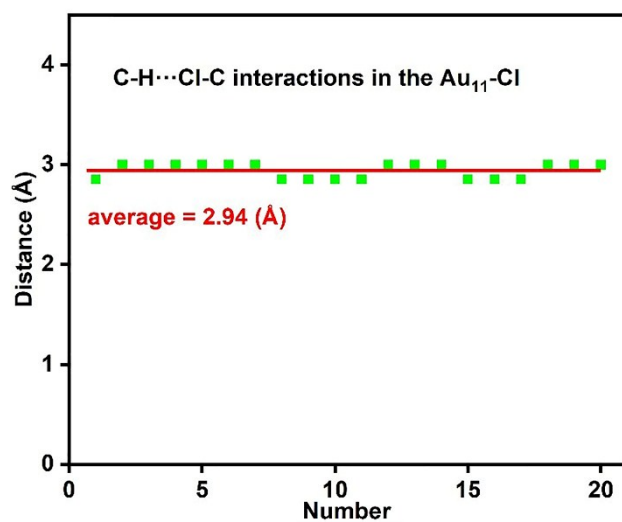


Figure S4. The statistical histogram of C-H...Cl-C interactions for Au₁₁-Cl.

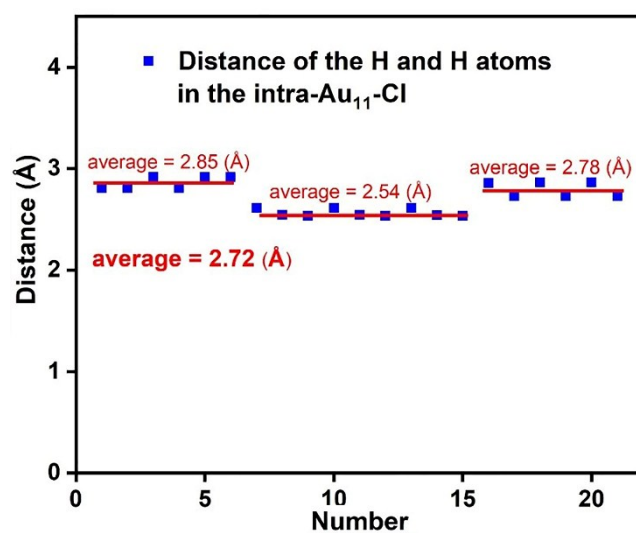


Figure S5. The statistical histogram of the intra-D_{HH} in the Au₁₁-Cl.

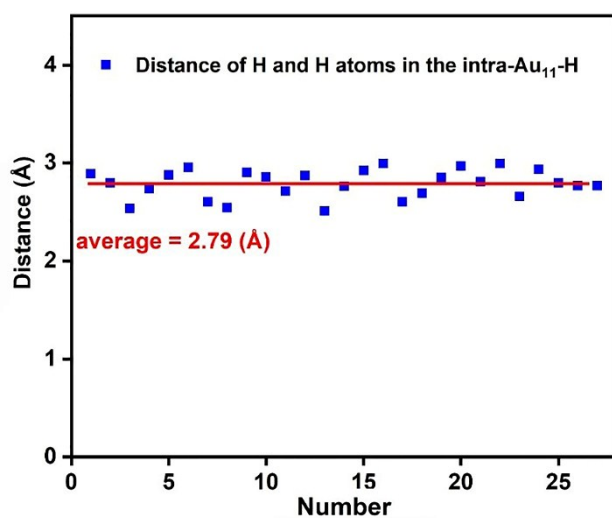


Figure S6. The statistical histogram of the intra-D_{HH} in the Au₁₁-H.

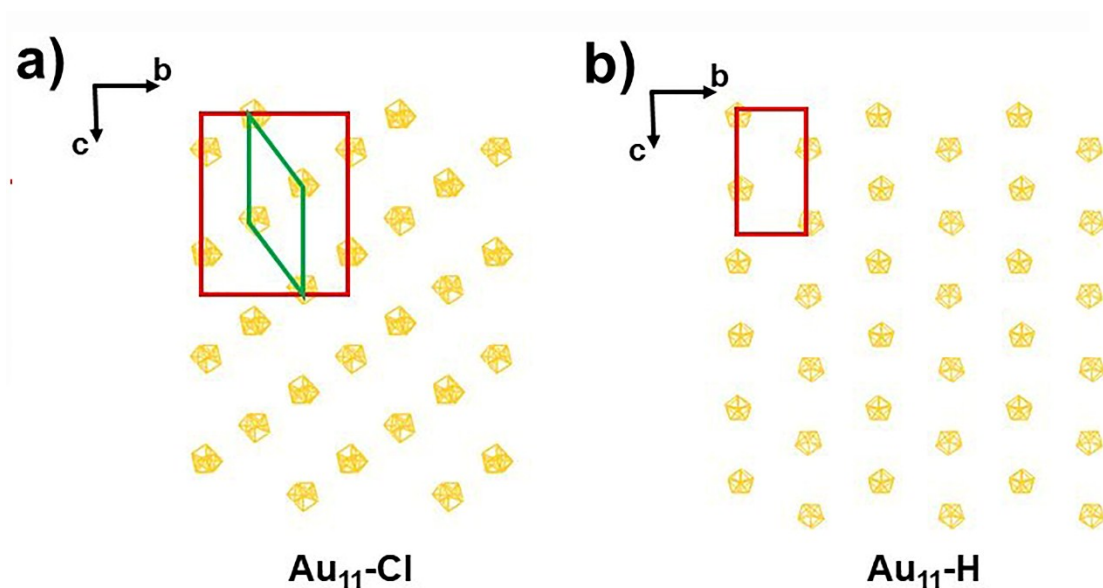


Figure S7. The superlattice of the **$\text{Au}_{11}\text{-Cl}$** (T) and **$\text{Au}_{11}\text{-H}$** (M) view from the a-axis.

Table S1. Crystal data and structure refinement for the **$\text{Au}_{11}\text{-Cl}$** and **$\text{Au}_{11}\text{-H}^a$** .

Identification code	$\text{Au}_{11}(\text{PPh}_3)_7\text{Cl}_3$	$\text{Au}_{11}(p\text{-ClPPh}_3)_7\text{Cl}_3$
Empirical formula	$\text{C}_{129}\text{H}_{112}\text{Au}_{11}\text{Cl}_4 \text{ P}_7$	$\text{C}_{126}\text{H}_{84}\text{Au}_{11}\text{Cl}_{24}\text{P}_7$
Formula weight	4187.41 g mol ⁻¹	4832.200 g mol ⁻¹
Crystal system	Monoclinic	Trigonal
Space group	P2(1)/n	R-3
Length a	17.864(3) Å	27.949(4) Å
Length b	25.801(5) Å	27.949(4) Å
Length c	26.912(5) Å	34.439(5) Å
Angle alpha	90.00°	90.000°
Angle beta	91.809(3)°	90.000°
Angle gamma	90.00°	120.000°
volume	12398(4) Å ³	23297.7(80) Å ³
Formula units Z	4	6
Density	2.243	2.066

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References:

1. L. C. McKenzie, T. O. Zaikova and J. E. Hutchison, *J Am Chem Soc*, 2014, **136**, 13426-13435.