

## Supplementary Information

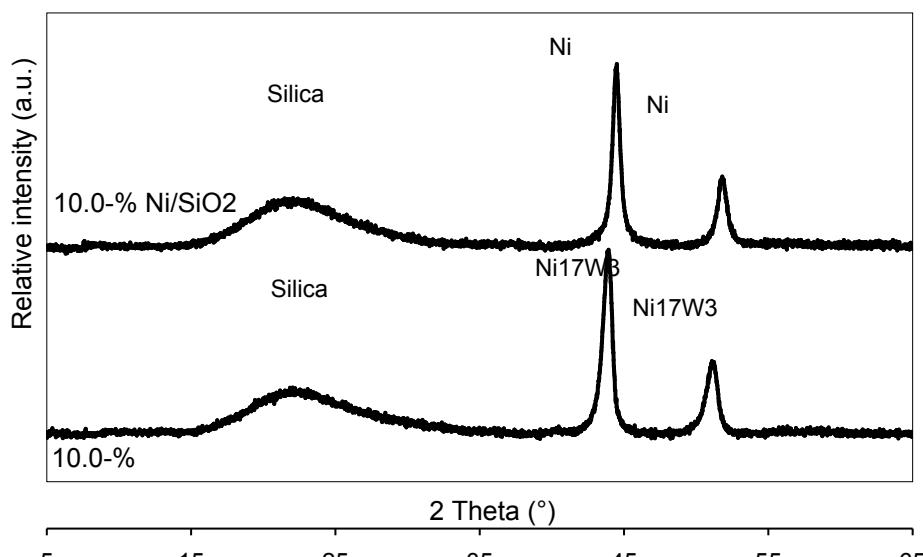


Figure S1: X-ray diffractograms of Ni/SiO<sub>2</sub> and Ni<sub>17</sub>W<sub>3</sub>/SiO<sub>2</sub> catalysts. Both catalysts are loaded with 10 wt.-% metal.

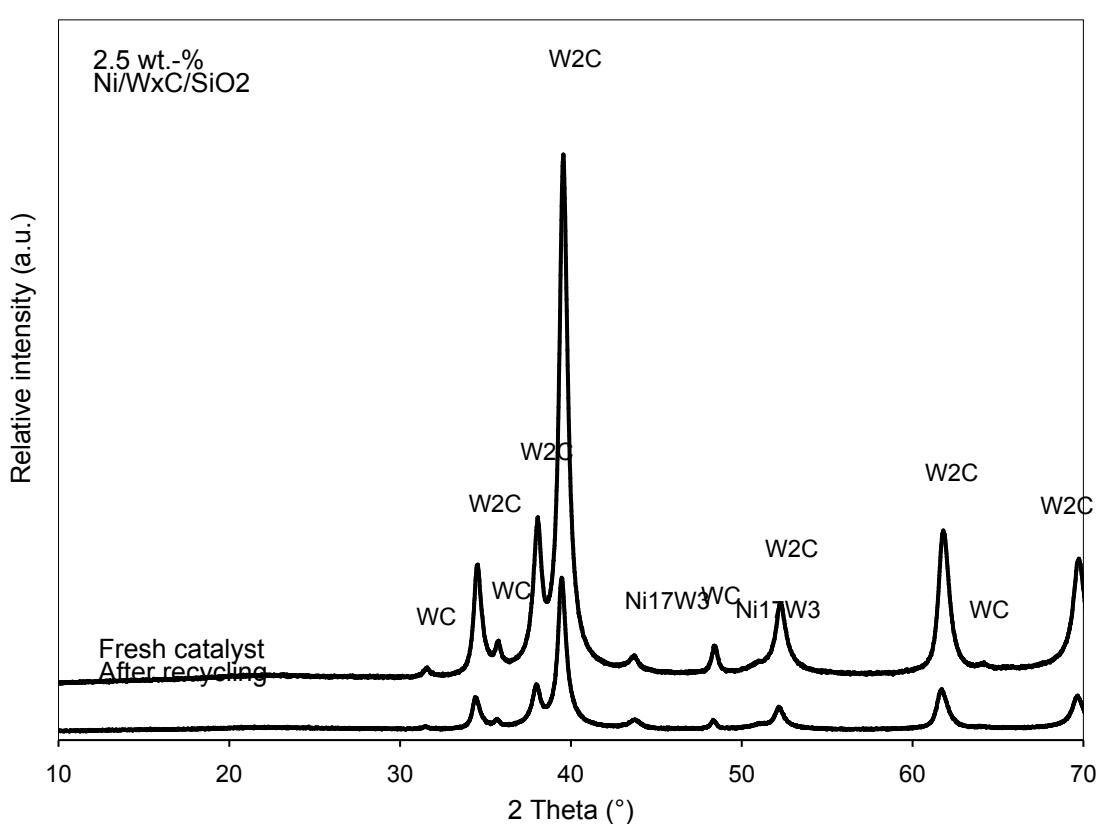


Figure S2: X-ray diffractograms of the 2.5 wt.-% Ni/W<sub>x</sub>C/SiO<sub>2</sub> catalysts directly after synthesis and after the recycling catalysis tests with a total of 26 h under reaction conditions.

*Table S1: Textural properties of catalysts as determined by isothermal nitrogen adsorption (BET method).*

Entry	Catalyst	Surface area (m <sup>2</sup> /g)	Pore volume (mL/g)	Pore radius (nm)
1	W <sub>x</sub> C/SiO <sub>2</sub>	96,84	0,52	1,69
2	0.5 %-Ni/W <sub>x</sub> C/SiO <sub>2</sub>	81,66	0,42	1,69
3	1.0 %-Ni/W <sub>x</sub> C/SiO <sub>2</sub>	77,26	0,44	1,70
4	2.5 %-Ni/W <sub>x</sub> C/SiO <sub>2</sub>	75,66	0,45	1,68
5	4.0 %-Ni/W <sub>x</sub> C/SiO <sub>2</sub>	73,85	0,45	1,68
6	9.0 %-Ni/W <sub>x</sub> C/SiO <sub>2</sub>	70,38	0,48	1,68
7	10.0 %-Ni/SiO <sub>2</sub>	151,47	1,24	1,69
8	10.0 %-Ni <sub>17</sub> W <sub>3</sub> /SiO <sub>2</sub>	149,18	1,29	1,69
9	2.5 %-Ni/W <sub>x</sub> C/SiO <sub>2</sub> after recycling	83,79	0,55	1,69

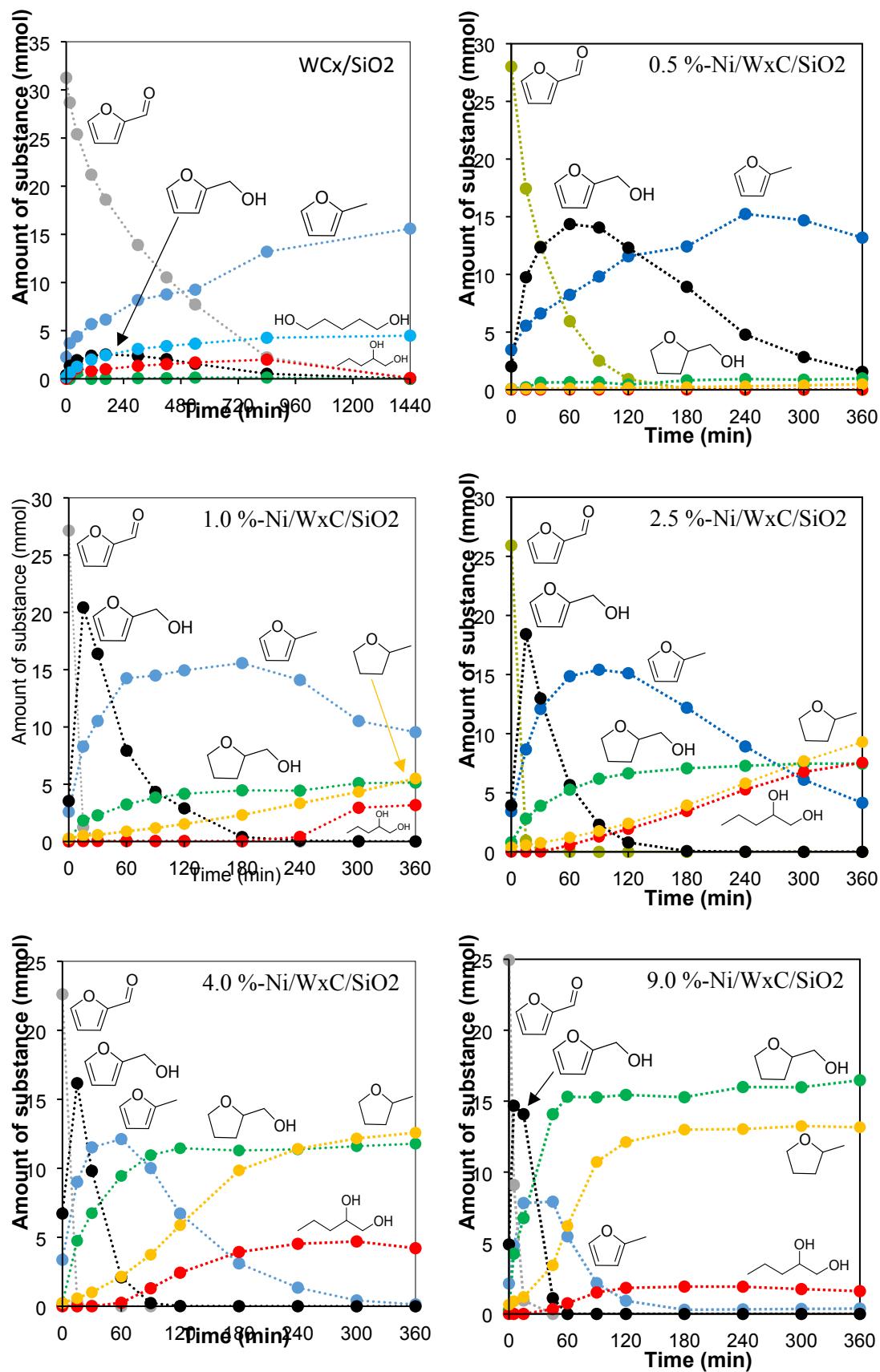


Figure S3: Reaction progress during catalytic conversion of furfural for tungsten carbide catalysts with differing nickel loadings from 0.0 to 9.0 wt.-% in the catalyst precursor. The reaction proceeds at 200 °C under 65 bar H<sub>2</sub>, with 0.50 g catalyst and 3 mL furfural in isopropyl alcohol as solvent. The total reaction time for W<sub>x</sub>C/SiO<sub>2</sub> (without Ni) is 24 h, else it is 6 h.

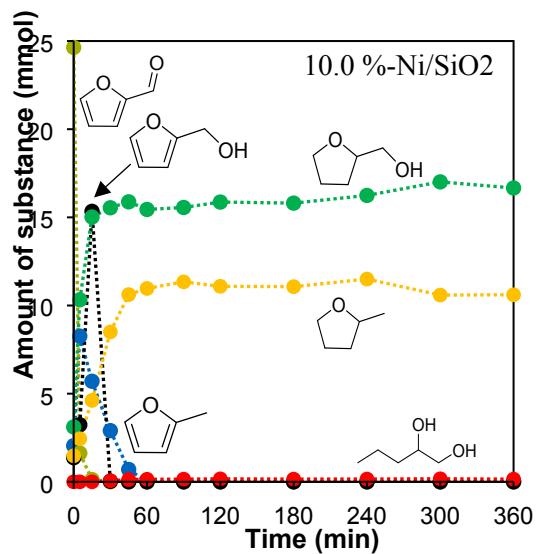


Figure S4: Reaction progress during catalytic conversion of furfural over a  $\text{Ni/SiO}_2$  catalyst. The reaction proceeds at  $200\text{ }^\circ\text{C}$  under 65 bar  $\text{H}_2$ , with 0.50 g catalyst and 3 mL furfural in isopropyl alcohol as solvent.

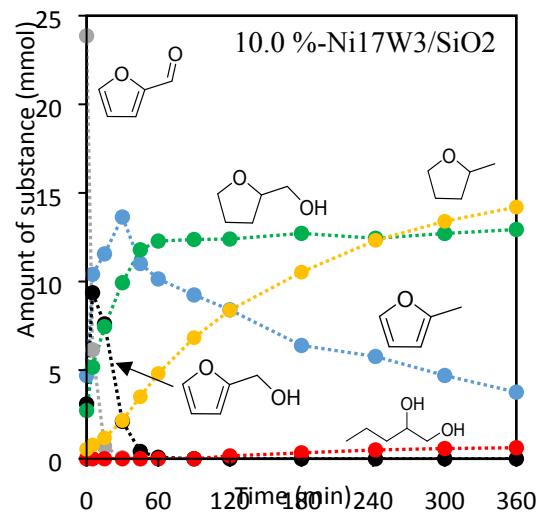


Figure S5: Reaction progress during catalytic conversion of furfural over a  $\text{Ni}_{17}\text{W}_3/\text{SiO}_2$  catalyst. The reaction proceeds at  $200\text{ }^\circ\text{C}$  under 65 bar  $\text{H}_2$ , with 0.50 g catalyst and 3 mL furfural in isopropyl alcohol as solvent.

Formulas for the calculation of conversion, yield and selectivity.

$$\text{Conversion} = 1 - \frac{n_{(\text{Furfural, out})}}{n_{(\text{Furfural, in})}}$$

$$\text{Yield}_{(\text{Product})} = \frac{n_{(\text{Product, out})}}{n_{(\text{Furfural, in})}}$$

$$\text{Selectivity}_{(\text{Product})} = \text{Conversion} \times \text{Yield}_{(\text{Product})}$$