## Structural effect of oxazolone derivatives on the initiating abilities of the dye-borate photoredox systems in radical polymerization under visible light

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**Fig. S1.** UV-Vis absorption spectra of **1**, **2** and **3** in toluene (the concentration was  $1.8 \times 10^{-5}$  M for **1**,  $5.8 \times 10^{-6}$  M for **2** and  $3.6 \times 10^{-6}$  M for **3**).



Fig. S2. Normalized fluorescence spectra of 1, 2 and 3 in toluene ( $\lambda_{ex}$  = 470 nm).



Fig. S3. Normalized electronic absorption and fluorescence spectra of 1 in ethyl acetate ( $\lambda_{ex}$  = 470 nm).



Fig. S4. Normalized electronic absorption and fluorescence spectra of 2 in ethyl acetate ( $\lambda_{ex}$  = 470 nm).



Fig. S5. Normalized electronic absorption and fluorescence spectra of 3 in ethyl acetate ( $\lambda_{ex}$  = 470 nm).



**Fig. S6.** UV-Vis absorption spectra recorded at different irradiation times of the oxazolone derivatives (left panel) and the dye-borate (right panel) in EtOAc upon exposure to the DPSS laser 473 nm; under air; **B6** concentration was 0.0016 M for **1** and 0.029 M for **2** and **3**. From the top: dye **1**, dye **2** and dye **3**.



Fig. S7. Fluorescence spectra recorded at without and with the presence of B6 in EtOAc (the dye concentration was  $5 \times 10^{-6}$  M; Ex = 465 nm).



Fig. S8. Laser flash photolysis spectra of the dyes in toluene recorded after irradiation with laser pulses of 355 nm. Delay times shown in legends.



**Fig. S9.** Laser flash photolysis spectra of the dye **3** (A = 0.53 at 355 nm; 2.5 mL) in toluene recorded after irradiation with laser pulses of 355 nm (plot on the left) and in the presence of **B6** (200  $\mu$ L 0.05 M) (plot on the right). Delay times: 2.0  $\mu$ s 2.8  $\mu$ s 4.0  $\mu$ s 6.0  $\mu$ s and 10.0  $\mu$ s.



**Fig. S10.** Laser flash photolysis spectra of the dyes **2** and **3** (A = 0.5 at 355 nm; 3 mL) in toluene recorded after irradiation with laser pulses of 355 nm in the presence of N-phenylglycine (**NPG**) (0.1 M) or trimethylammonium phenyltriethylborate (**B6**) (0.028 M). Delay times marked in Figure.



Fig. S11. Kinetic curves for the decay of absorbance (540 nm) of 3 in toluene in the presence of the electron donors marked in Figure.