

Supplementary Data

Synthesis and application of tuneable carbon-silica composites from the microwave pyrolysis of waste paper for selective recovery of gold from acidic solutions

Konstantina Sotiriou,^a Nontipa Supanchaiyamat,^b Tengyao Jiang,^a Intuorn Janekarn,^b Andrea Muñoz García,^a Vitaliy L. Budarin,^a Duncan J. MacQuarrie^a and Andrew J. Hunt ^{b*}

^a Green Chemistry Centre of Excellence, Department of Chemistry, University of York, Heslington, York, YO10 5DD, UK

^b Materials Chemistry Research Center, Department of Chemistry and Center of Excellence for Innovation in Chemistry, Faculty of Science, Khon Kaen University, Khon Kaen, 40002, Thailand

*Corresponding authors email: andrew@kku.ac.th

NITROGEN ADSORPTION/ DESORPTION POROSIMETRY

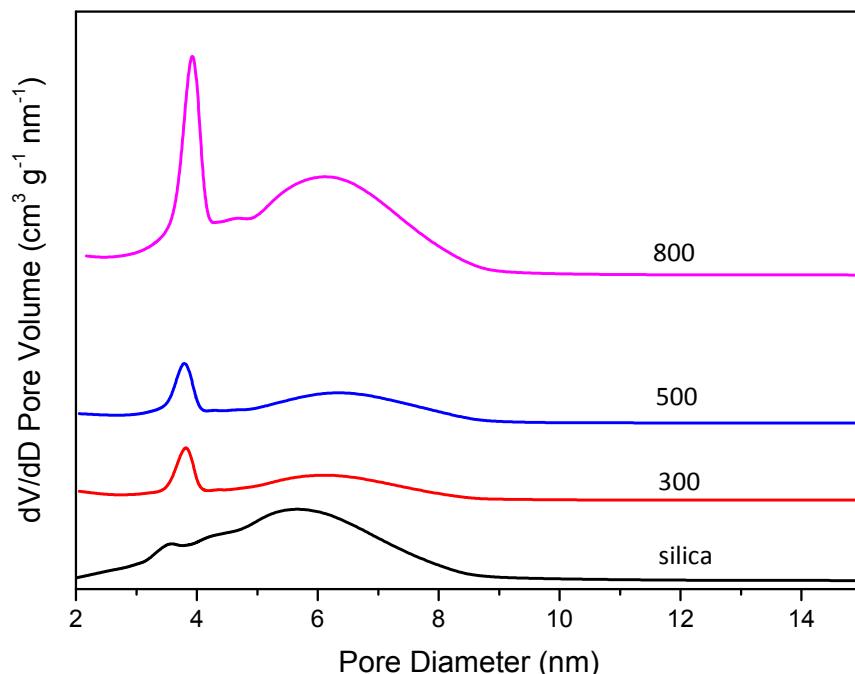


Fig. S1 Pore size distribution curves of silica gel K60 and CSCs

INFRARED DATA

Table S1 Adsorption bands observed in IR spectra of CSCs

Functional Group	Characteristic Adsorption	Notes
O-H stretch	3200-3550 cm ⁻¹	Due to the presence of carboxylic acids
Alkyl C-H stretch	2950-2850 cm ⁻¹	Aliphatic chains of bio-oil
C=O stretch	1790-1680 cm ⁻¹	Due to the presence of carboxylic acids, esters, ketones, aldehydes.
Si-O-Si asymmetric stretch	1150-100 cm ⁻¹	Due to interaction of silanol groups and bio-oil.
Si-O-Si symmetric stretch	900-700 cm ⁻¹	Due to interaction of silanol groups and bio-oil.

X-RAY PHOTOEMISSION SPECTROSCOPY DATA

Table S2 Carbon nature and corresponding binding energies for peak deconvolution of C1s spectra of CSCs

Carbon Nature	Binding Energy/ eV
C=C sp ²	284-284.6
C-C sp ³	284.6-285.1
C- O (C-O-C, C-O-H)	285.5-286.9
C=O	287.1-287.7
C-O-Si	288-290

Table S3 Gold nature and corresponding binding energies for peak deconvolution of Au4f spectra of CSCs

Element	Nature	Binding Energy / eV
Au (4f 7/2)	Au(0)	84.2 ± 0.4
Au (4f 5/2)	Au(0)	87.9 ± 0.4
Au (4f 7/2)	Au(I)	85.8 ± 0.4
Au (4f 5/2)	Au(I)	86.2 ± 0.4
Au (4f 7/2)	Au(III)	86.4 ± 0.2
Au (4f 5/2)	Au(III)	90.1 ± 0.2

Au4f XPS Spectra

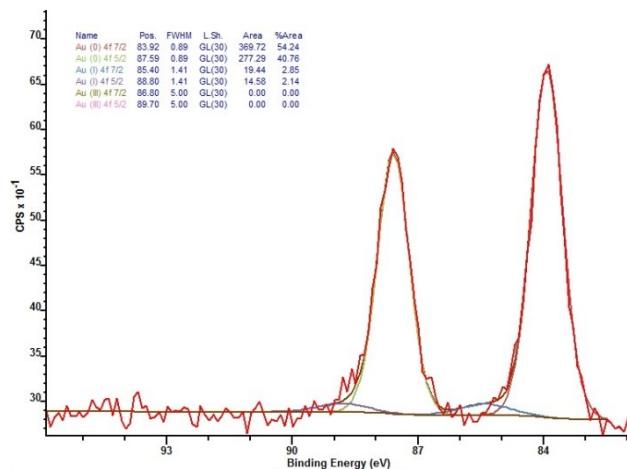


Fig. S2 Au4f spectrum for CSC300 after adsorption in 300 mg L^{-1} of AuCl_3

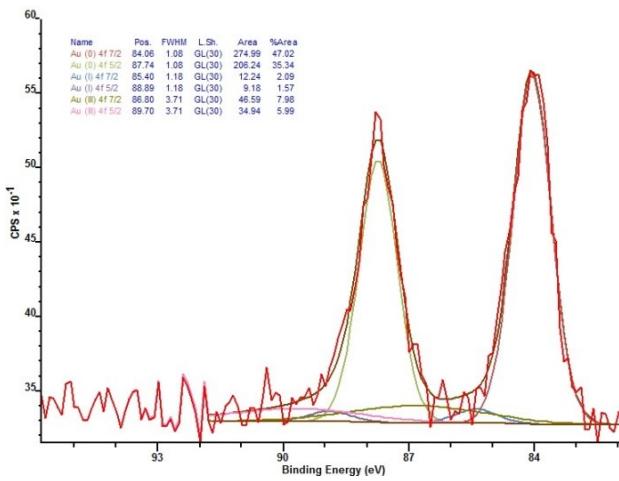


Fig. S3 Au4f spectrum for CSC300 after adsorption in 50 mg L⁻¹ of AuCl₃

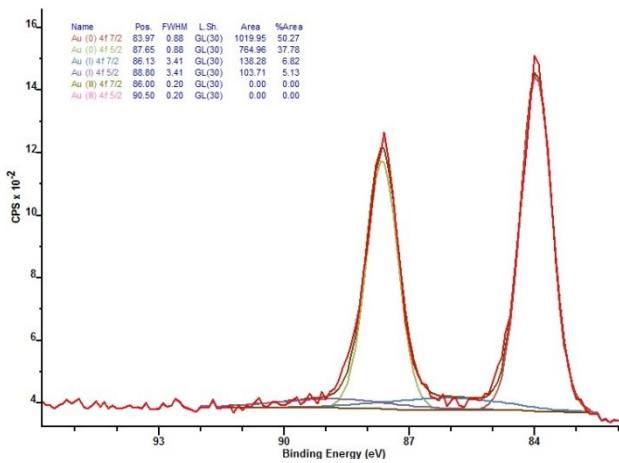


Fig. S4 Au4f spectrum for CSC500 after adsorption in 500 mg L⁻¹ of AuCl₃

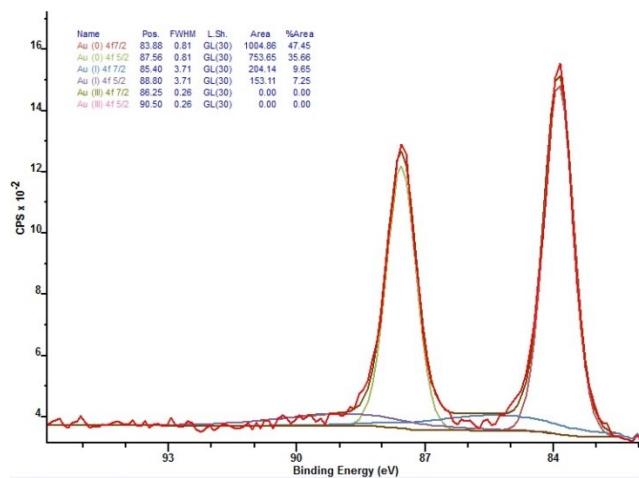


Fig. S5 Au4f spectrum for CSC500 after adsorption in 300 mg L⁻¹ of AuCl₃

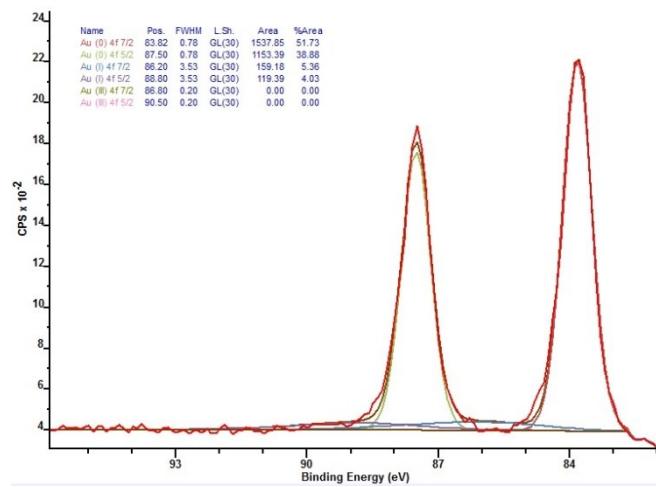


Fig. S6 Au4f spectrum for CSC500 after adsorption in 100 mg L⁻¹ of AuCl₃

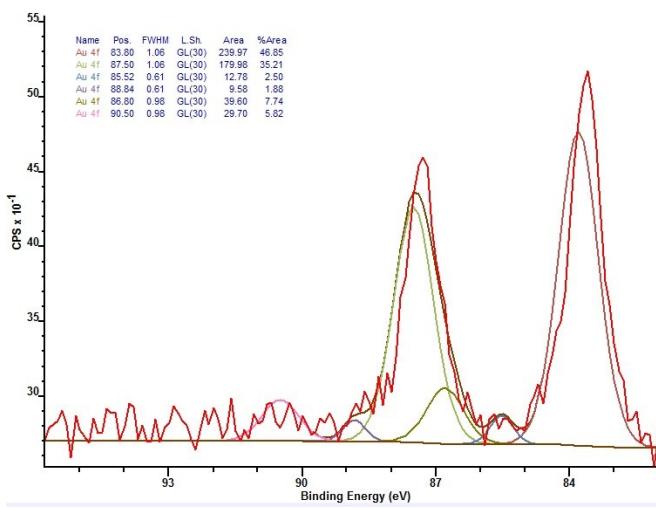


Fig. S7 Au4f spectrum for CSC500 after adsorption in 50 mg L⁻¹ of AuCl₃

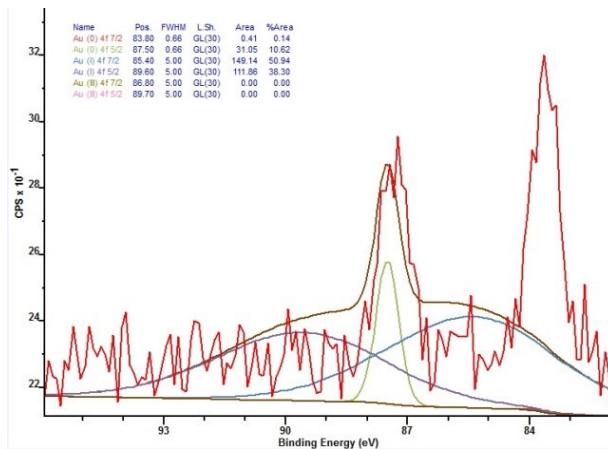


Fig. S8 Au4f spectrum for CSC800 after adsorption in 300 mg L⁻¹ of AuCl₃

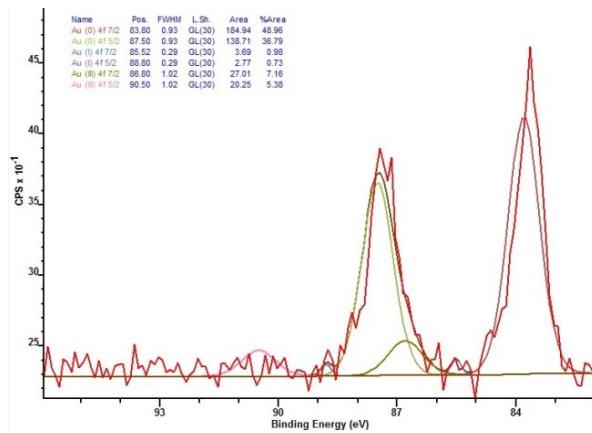


Fig. S9 Au4f spectrum for CSC800 after adsorption in 50 mg L⁻¹ of AuCl₃

POWDER X-RAY DIFFRACTION

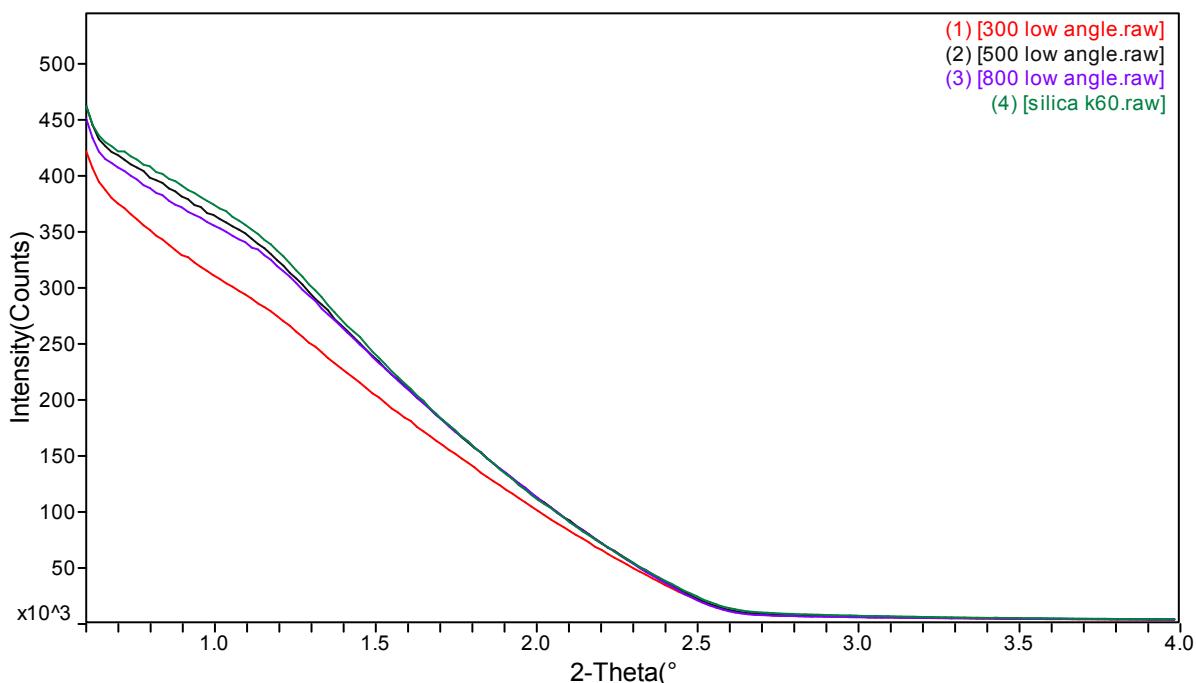


Fig. S10 Small Angle XRD patterns of silica gel K60 and CSCs

SCANNING ELECTRON MICROSCOPY- ENERGY DISPERSIVE X-RAY ANALYSIS

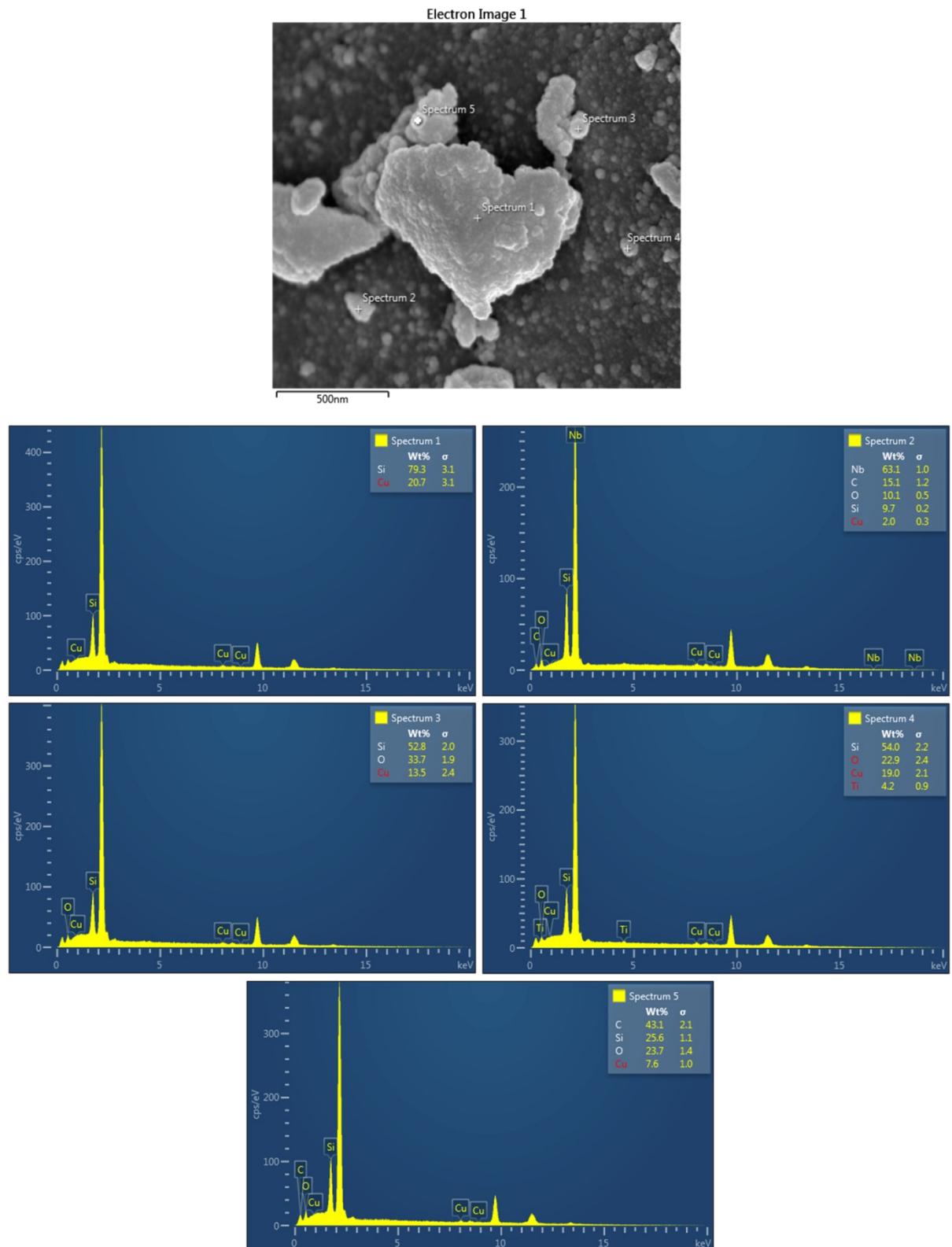


Fig. S11 SEM- EDX of CSC300

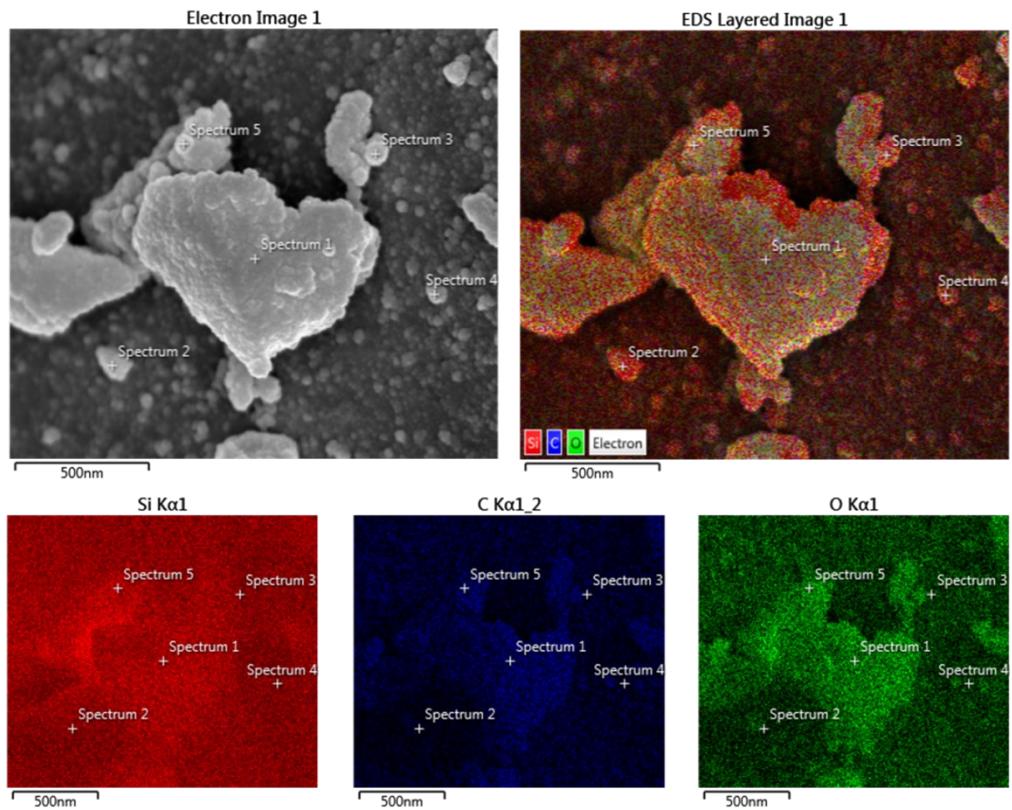


Fig. S12 SEM- EDX mapping of C, Si and O for CSC300

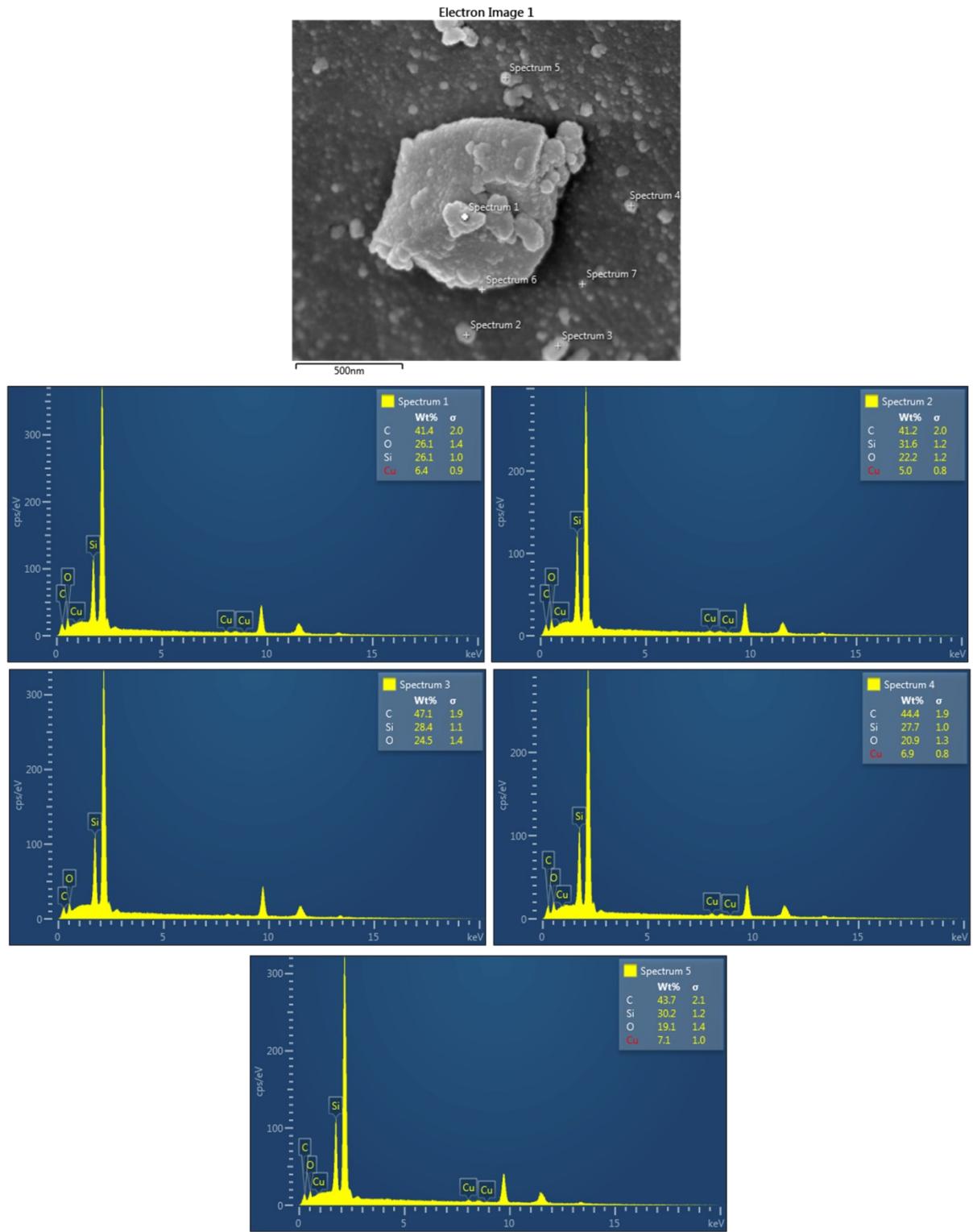


Fig. S13 SEM- EDX of CSC500

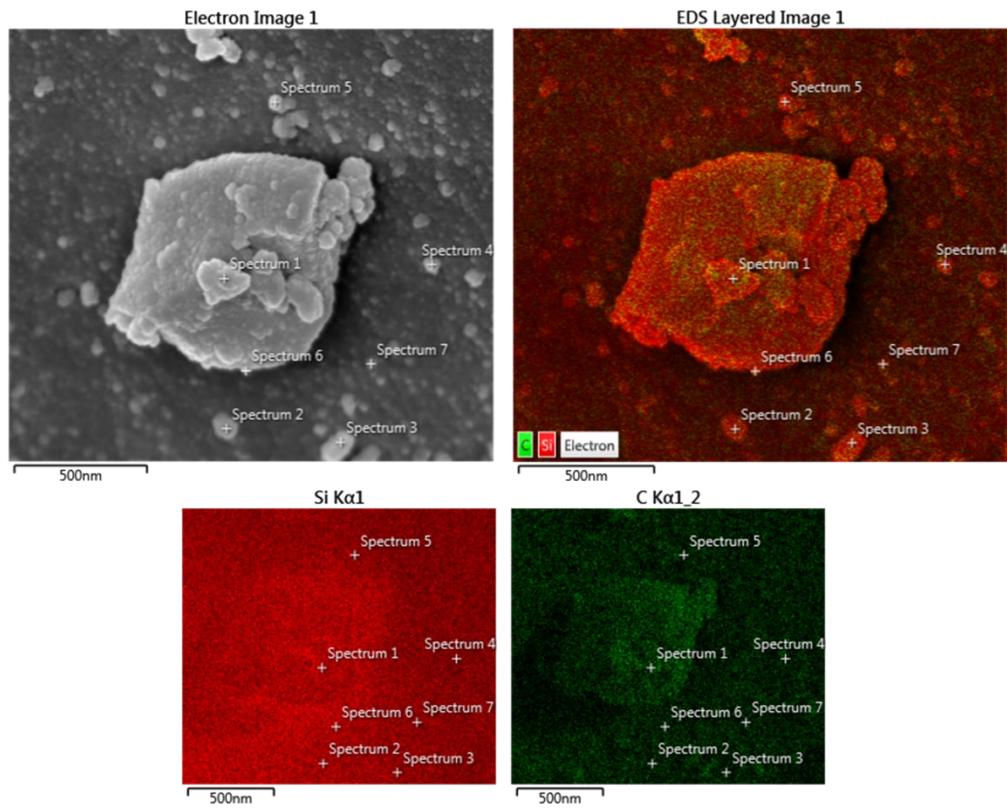


Fig. S14 SEM- EDX mapping of C and Si for CSC500

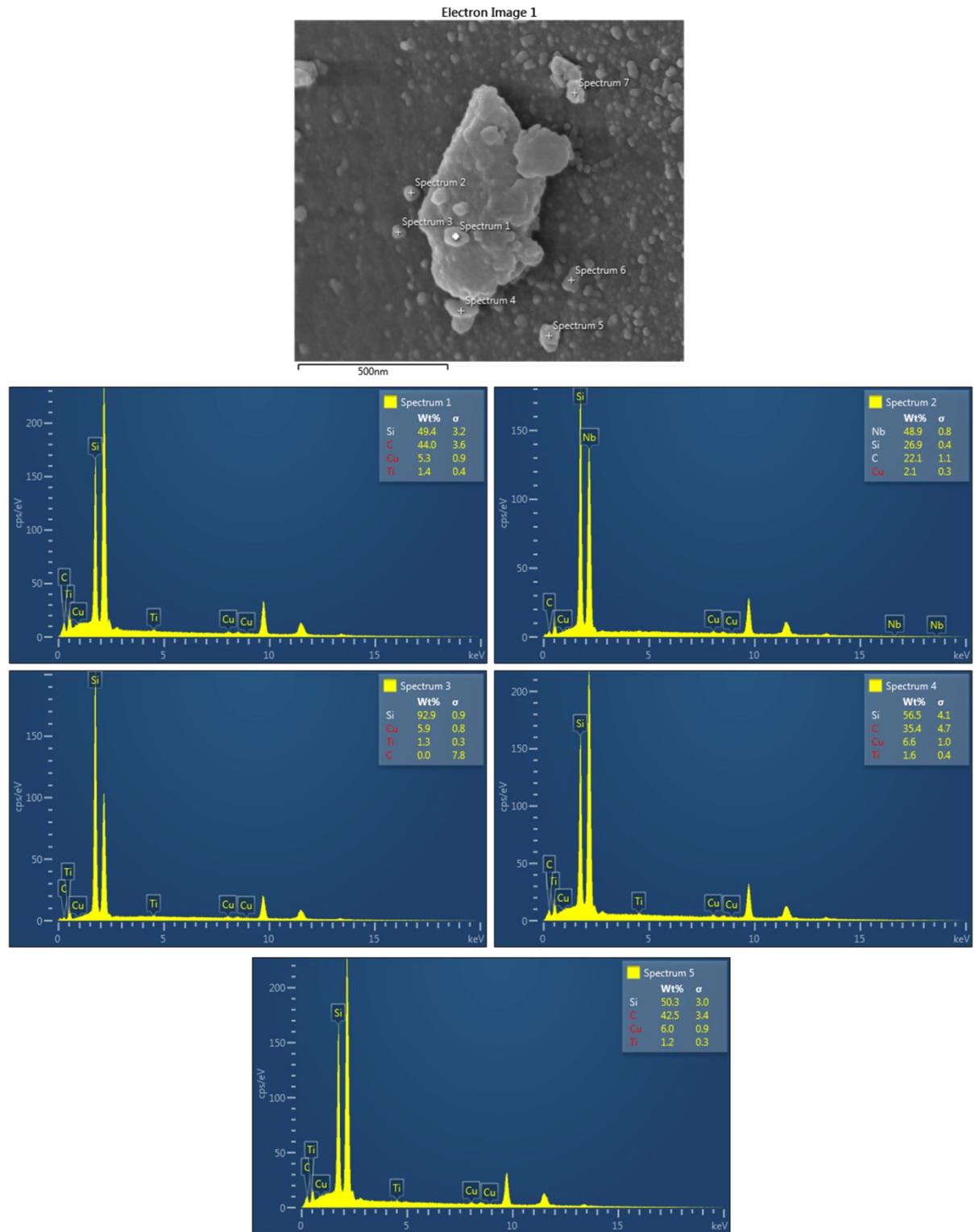


Fig. S15 SEM- EDX of CSC800

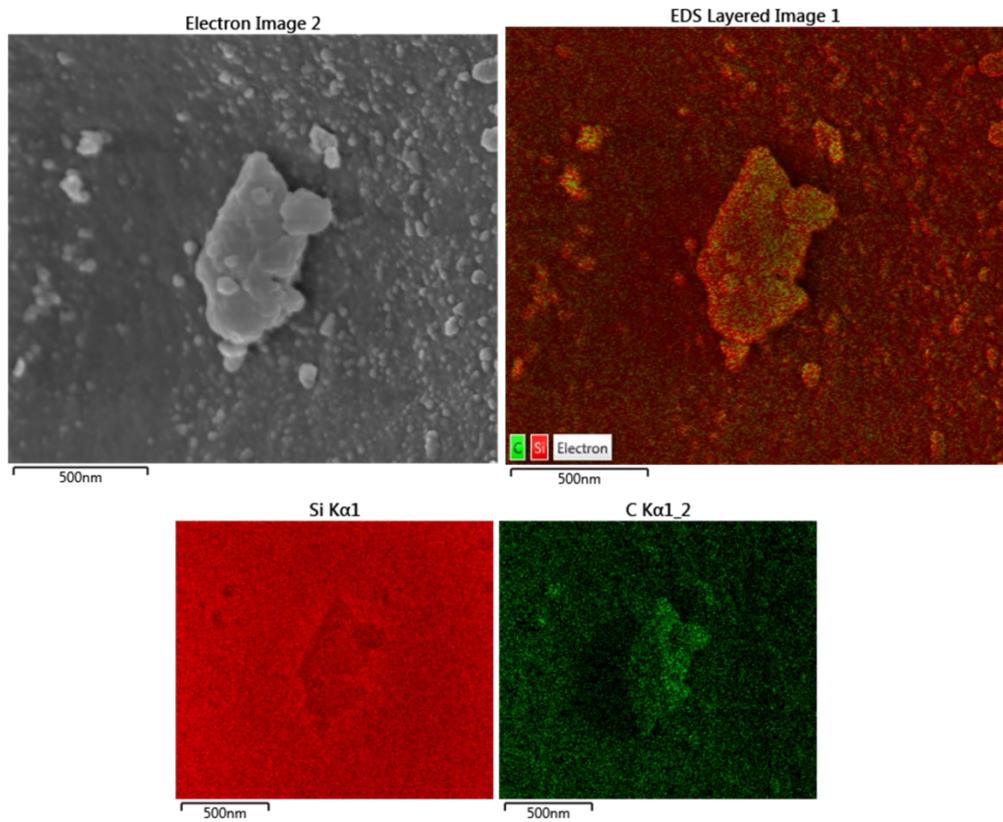
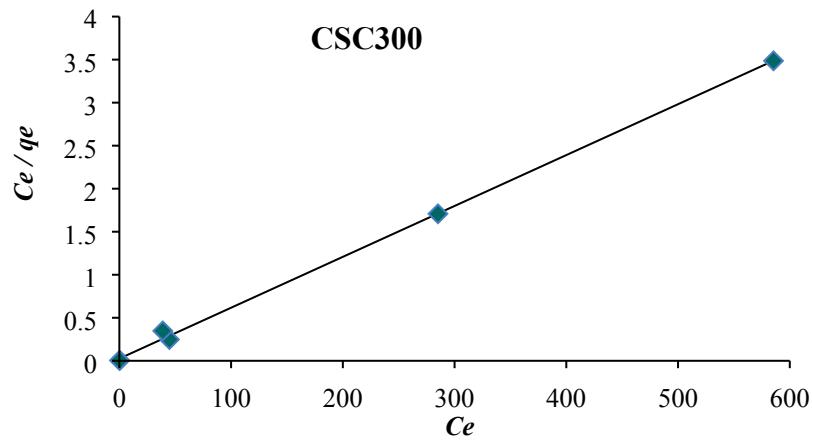


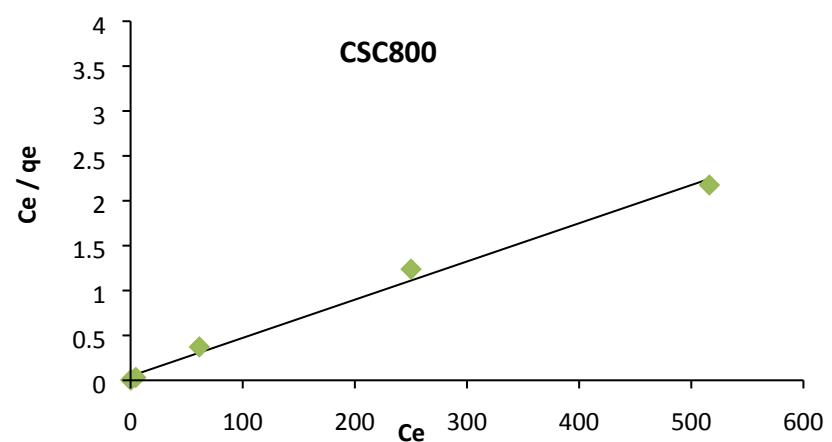
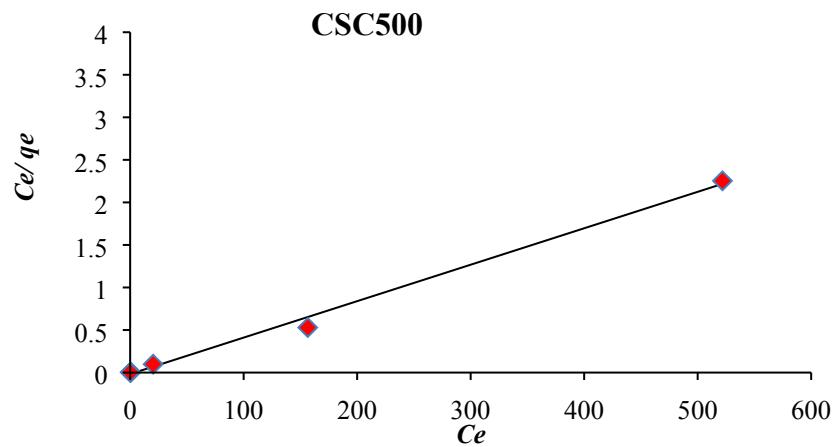
Fig. S16 SEM- EDX mapping of C and Si for CSC800

GOLD ADSORPTION DATA

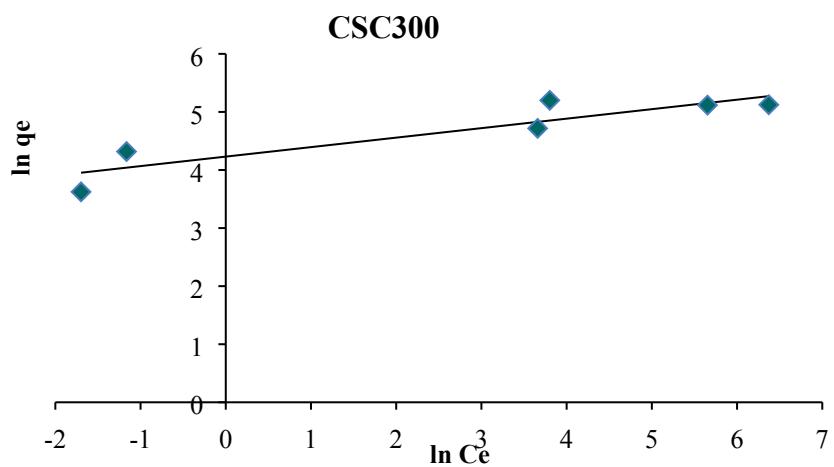
Gold Selectivity of CSCs

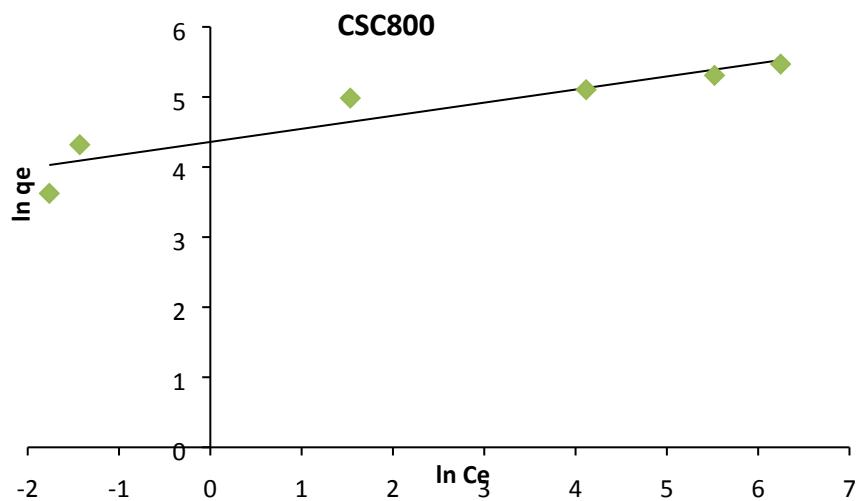
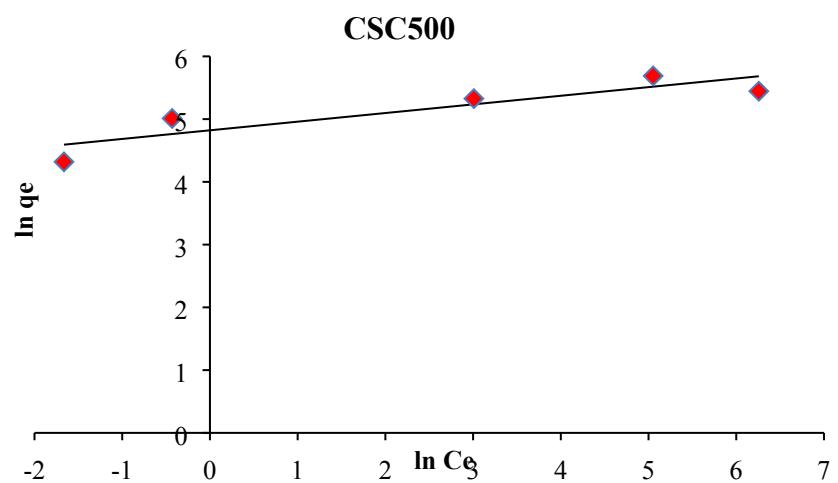
Linear model fittings: Langmuir model



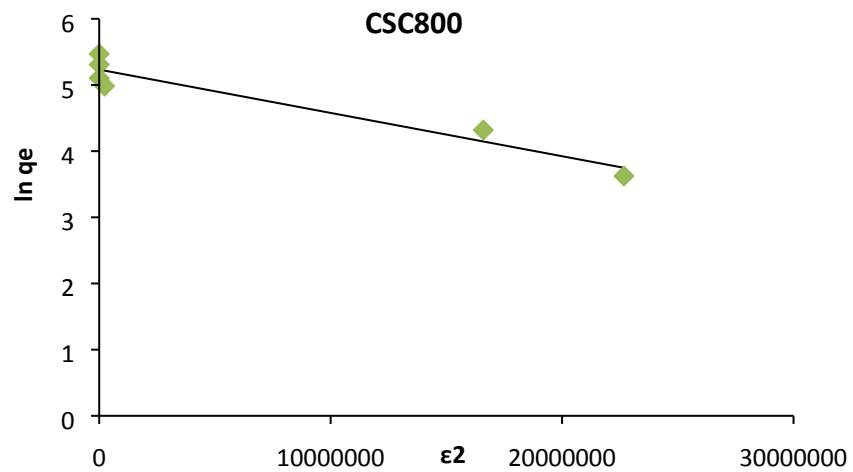
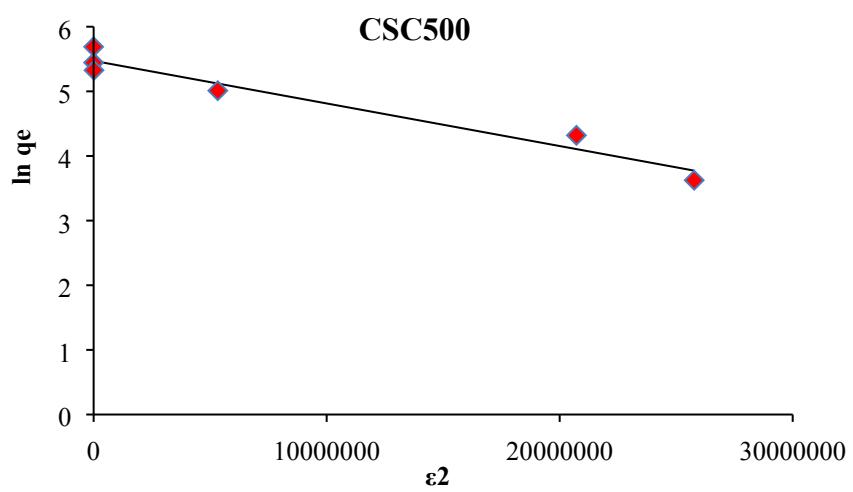
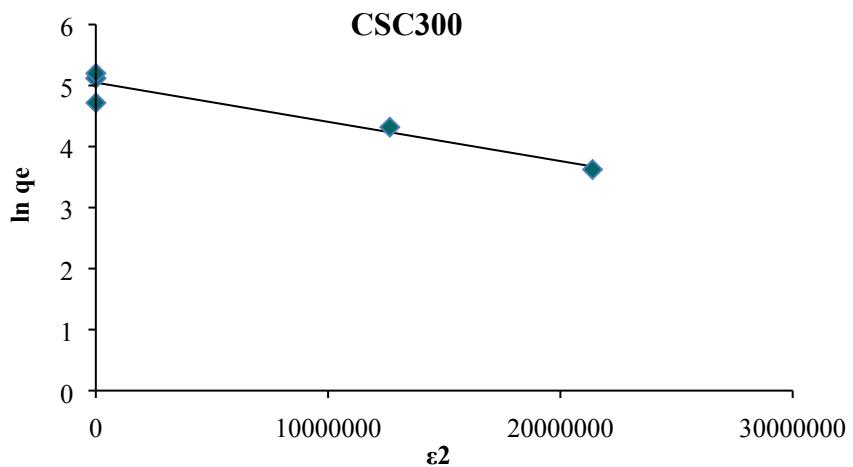


Linear model fittings: Freundlich model





Linear model fittings: D-R model



MICROWAVE PYROLYSIS SETUP

