Supporting Information

One-Pot Preparation of Hierarchical Cu₂O Hollow Spheres for

Improved Visible-Light Photocatalytic Properties

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Fig. S1 TEM images of the as-prepared product at different magnification.



Fig. S2 The XPS survey spectrum (a), high resolution Cu 2p XPS (b) and Auger Cu LMM spectra(c) of the as-prepared product.



Fig. S3 EDS patterns and elemental contents (inset) from bulk (a) and surface (b) of the as-

prepared product.



Fig. S4 XRD patterns of the as-prepared products at the reaction time of 45, 60 and 90 min.



Fig. S5 SEM (a-c) and TEM (d-f) images of the as-prepared products at the different temperatures: (a and d) 150 °C, (b and e) 160 °C, and (c and f) 180 °C.



Fig. S6 SEM (a-e) and TEM images (f-j) of the as-prepared products using the different [BMIM]BF₄ concentration: (a and f) 0.0125 mol·L⁻¹, (b and g) 0.025 mol·L⁻¹, (c and h) 0.075 mol·L⁻¹, (d and i) 0.1 mol·L⁻¹, and (e and j) 0.15 mol·L⁻¹.

Table S1 Comparison of the photocatalytic activities of various Cu_2O catalysts for the

degradation of MO.

Photocatalyst	Light source	Additive	Photocatalytic condition	Degradation activity	Degradation rate / (mmol ·g ⁻¹ ·h ⁻¹)	Ref.
Cu ₂ O hollow spheres	250 W xenon arc lamp (λ > 420 nm)	None	Photocatalyst: 10 mg MO solution: 50 mL 20 mg·L ⁻¹	more than 90 % (20 min) completely degraded (30 min)	0.825	This study
rhombic dodecahedral Cu ₂ O	500 W xenon lamp (λ = 400-700 nm)	None	Photocatalyst: 15 mg MO solution: 100 mL 15 mg·L ⁻¹	88.3 % (30 min)	0.540	1
Cu ₂ O nanowires	500 W xenon lamp (420 nm $<\lambda$ <780 nm)	None	Photocatalyst: 4 mg MO solution: 100 mL 10 mg·L ⁻¹	73 % (120 min)	0.279	2
Cu ₂ O cubes	500 W xenon lamp	None	Photocatalyst: 71.54 mg MO solution: 90 mL 30 mg·L ⁻¹	~20 % (90 min)	0.0154	3
Cu ₂ O rhombic dodecahedra	500 W xenon lamp	None	Photocatalyst: 71.54 mg MO solution: 90 mL 30 mg·L ⁻¹	almost completely degraded (90 min)	0.0769	3
Cu ₂ O nanowire polyhedra	500 W Xe lamp (380 $nm < \lambda < 700$ nm)	H ₂ O ₂	Photocatalyst: 30 mg MO solution: 50 mL 20 mg·L ⁻¹	complete degraded (40 min)	0.153	4
oxygen- deficient Cu ₂ O nanoparticles	cold white LED (16 mW·cm ⁻²)	None	Photocatalyst : 2 mg MO solution: 20 mL 5×10 ⁻ ⁵ mol·L ⁻¹	~98 % (20 min)	1.47	5
Cu ₂ O hierarchical nanoclusters	150 W tungsten- halogen lamp	None	Photocatalyst: 0.01 g MO solution: 100 mL 10 mg \cdot L ⁻¹ and 40 mg \cdot L ⁻¹	completely degraded (12 min for 10 mg \cdot L ⁻¹ MO solution and 65 min for 40 mg \cdot L ⁻¹)	1.528 (10 mg·L ⁻¹) 1.128 (40 mg·L ⁻¹)	6
octahedral, truncated octahedral, cuboctahedral , truncated cubic and cubic Cu ₂ O microcrystals	500 W xenon lamp (λ>400 nm)	None	Photocatalyst: 0.025 g MO solution: 50 mL 15 mg·L ⁻¹	100 % for octahedral $Cu_2O(3 h)$ 76 % for truncated octahedral $Cu_2O(3 h)$ 28 % for cuboctahedral $Cu_2O(3 h)$ 36 % for truncated cubic $Cu_2O(3 h)$ 17 % for cubic $Cu_2O(3 h)$ h)	0.0306 (octahedral Cu2O) 0.0232 (truncated octahedral Cu2O) 0.00855 (cuboctahedral Cu2O) 0.0100 (truncated cubic Cu2O) 0.00519 (cubic Cu2O)	7



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Fig. S7 Mass spectrum of the MO solution under the 60-min visible-light irradiation over the hierarchical Cu₂O hollow spheres.



Fig. S8 PL spectra of Cu₂O hollow spheres and Cu₂O solid spheres at an excitation wavelength of

436 nm.

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