

Supporting Information

Facile synthesis of flower-like hierarchical N-doped Nb₂O₅/C nanostructures with efficient photocatalytic activity under visible light

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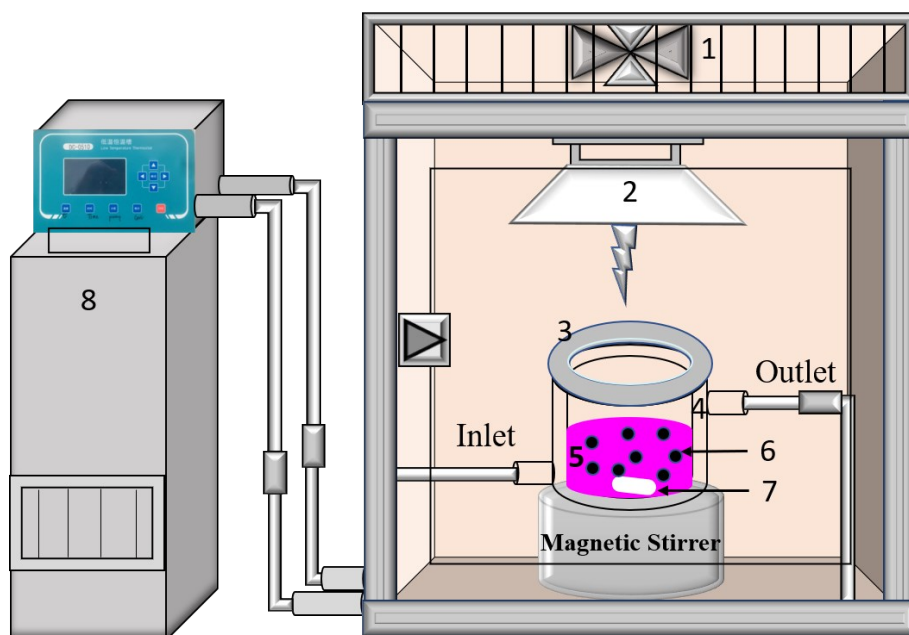
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Table of Contents

SUPPORTING FIGURES AND TABLES.....	1
Supporting Scheme S1	2
Supporting Figure S1	2

Supporting Figure S2	3
Supporting Figure S3	4
Supporting Figure S4	4
Supporting Figure S5	5
Supporting Figure S6	5
Supporting Figure S7	6
Supporting Figure S8	6
Supporting Table S1	7
Supporting Table S2	7
References	10

SUPPORTING FIGURES



Scheme S1. Schematic illustration for the photocatalytic reactor used in the photodegradation of RhB under visible-light illumination (1. Suction fan, 2. Xe lamp 400 W (300-800 nm), 3. Cut-off filter ($\lambda > 400$ nm), 4. Cooling Jacket, 5. 200 mL quartz vessel, 6. photocatalyst, 7. Magnet bar, and 8. Low-Temperature Thermostat.).

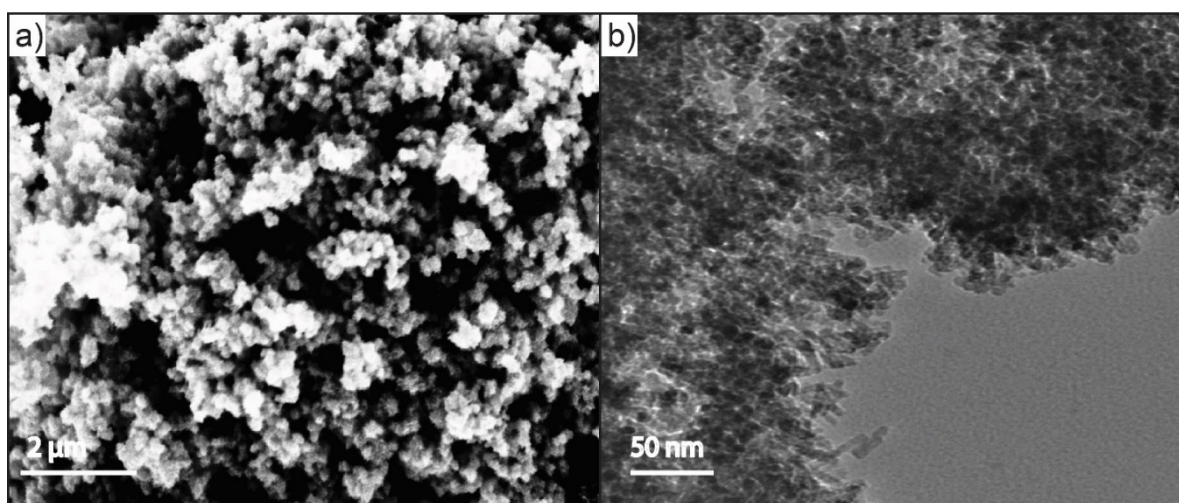


Fig. S1. (a) SEM and (b) TEM images of pure NBO at calcination temperature 350 °C.

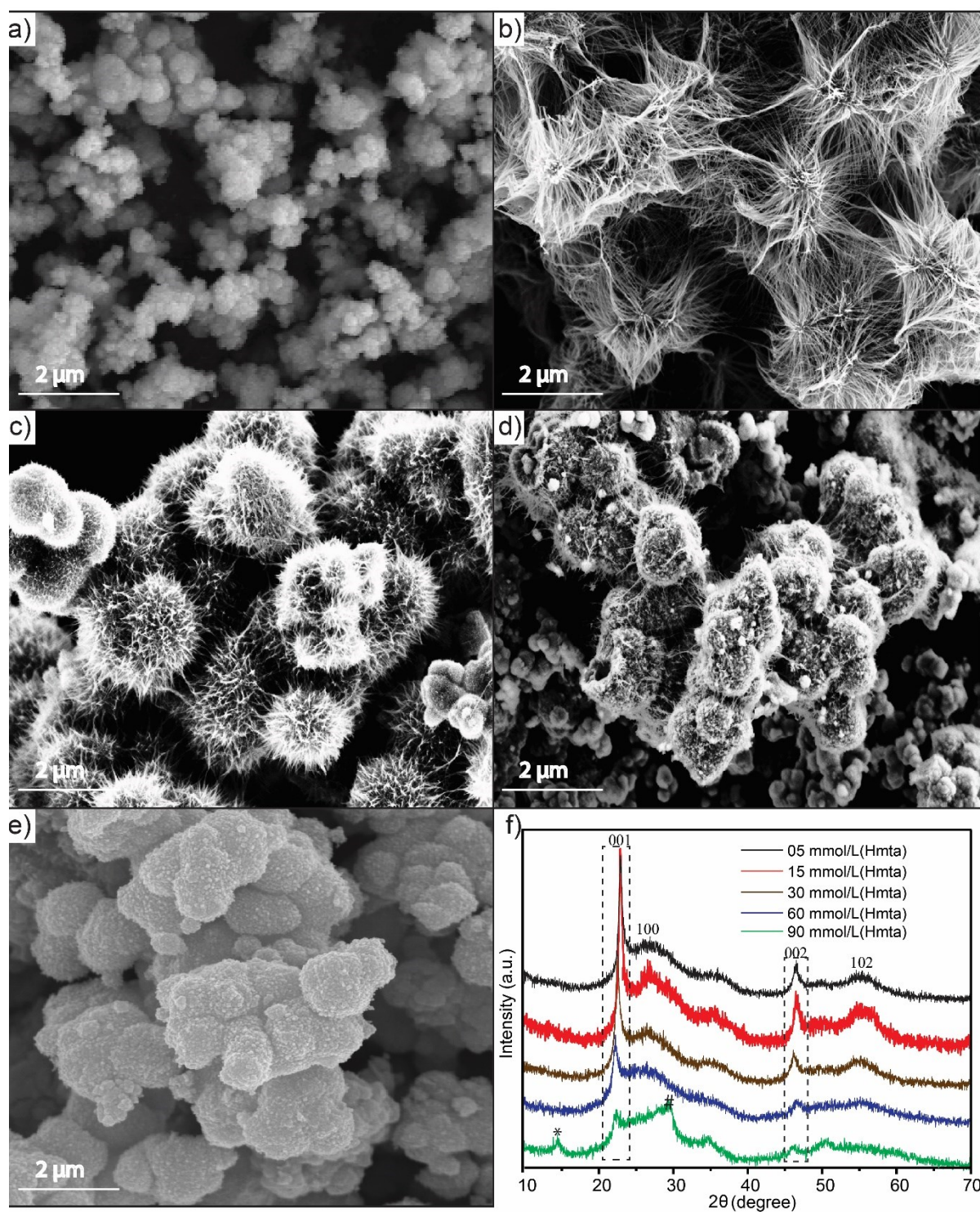


Fig. S2. SEM images and XRD patterns of NBO nanostructures synthesized by using different concentrations of Hmta: (a) 5 mmol·L⁻¹, (b) 15 mmol·L⁻¹, (c) 30 mmol·L⁻¹, (d) 60 mmol·L⁻¹, (e) 90 mmol·L⁻¹.

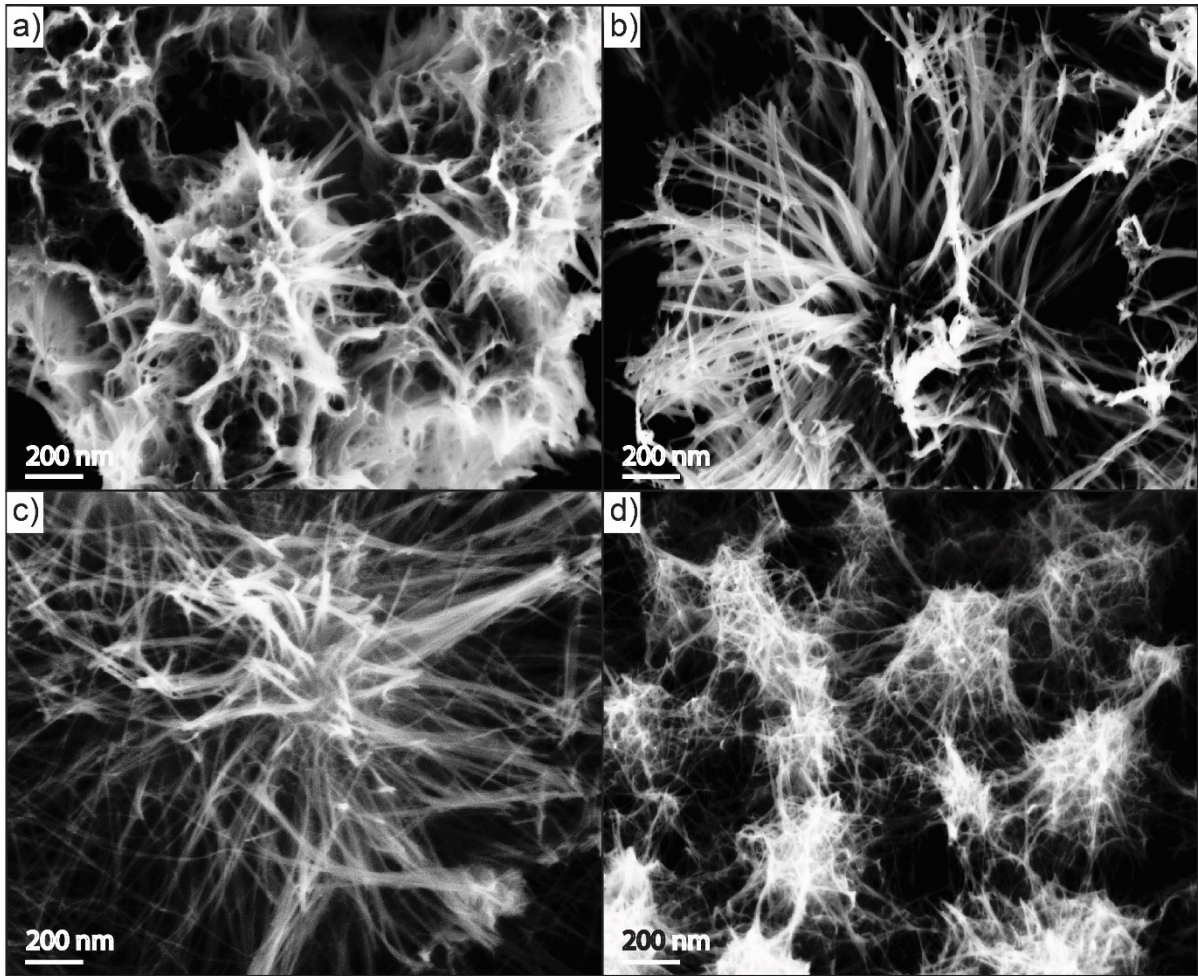


Fig. S3. SEM images of N-NBO/C nanostructures synthesized at different hydrothermal temperatures: (a) 140 °C, (b) 160 °C, (c) 180 °C, and (d) 200 °C.

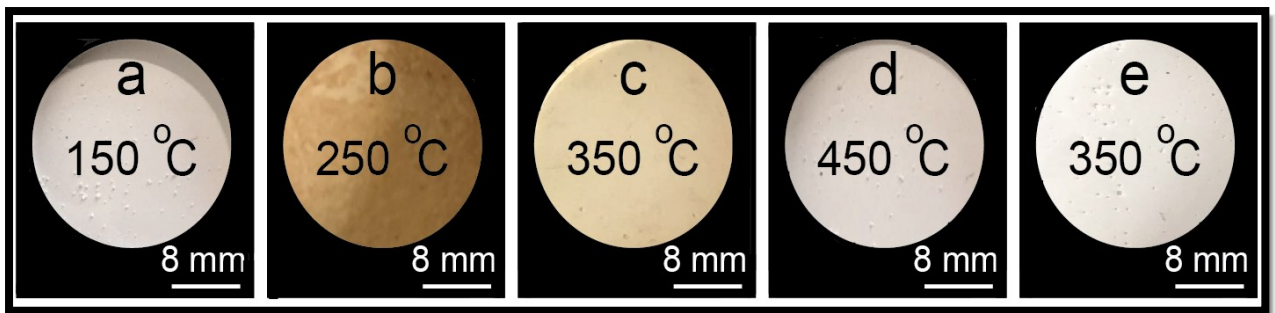


Fig. S4. Camera images of NBO nanostructures powder at different calcination temperatures of (a) 150 °C, (b) 250 °C, (c) 350 °C, (d) 450 °C, and (e) Pure NBO at 350 °C.

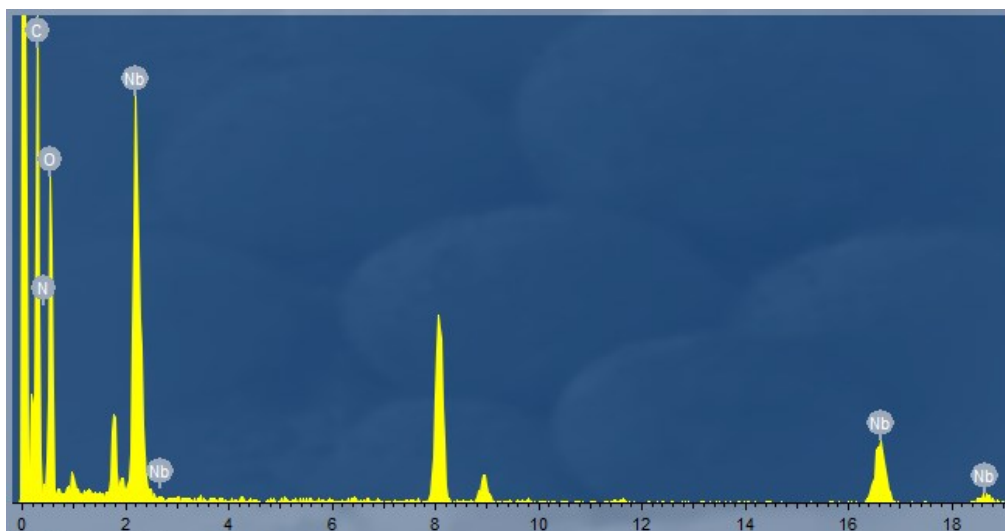


Fig. S5. EDX analysis of C, N, Nb, and O elements for N-NBO/C

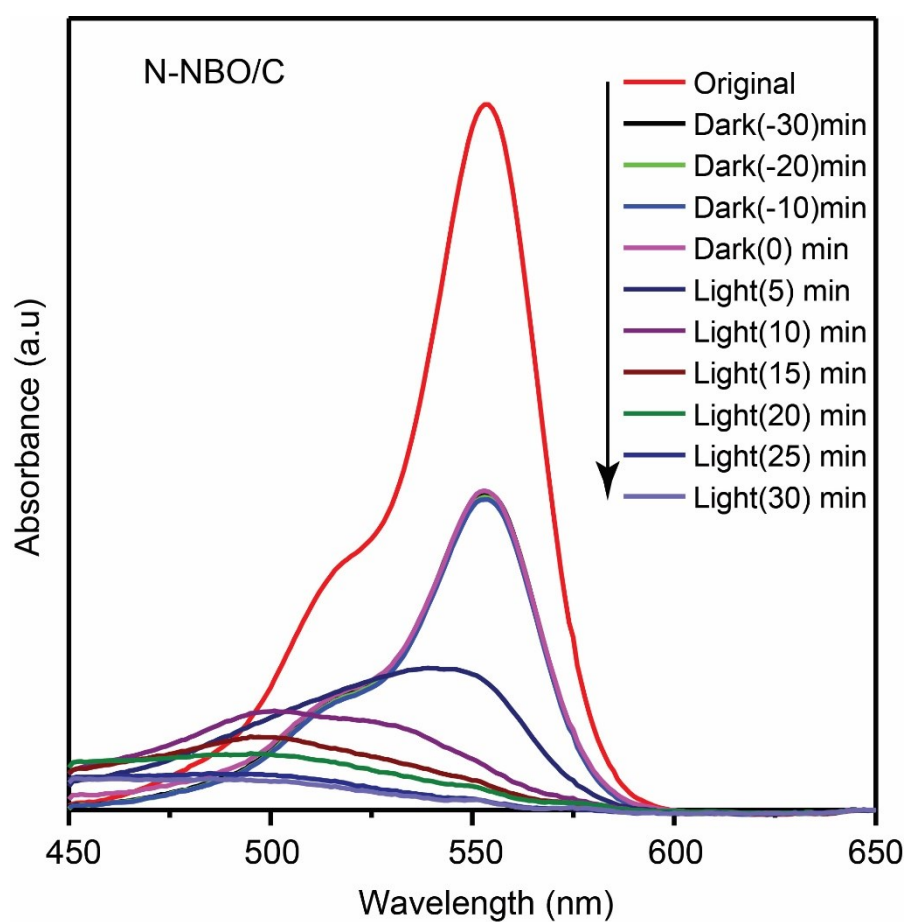


Fig. S6. UV-vis absorption spectra changes of the RhB absorbance in aqueous solution over N-NBO/C

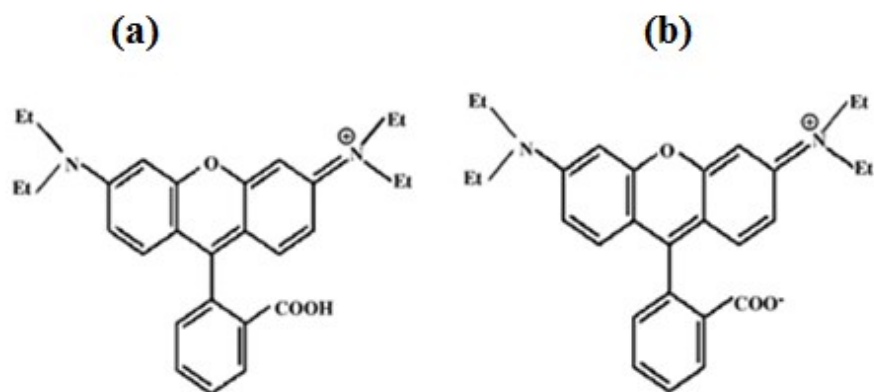


Fig. S7. Chemical structure of RhB, (a) cationic and (b) zwitterionic form.

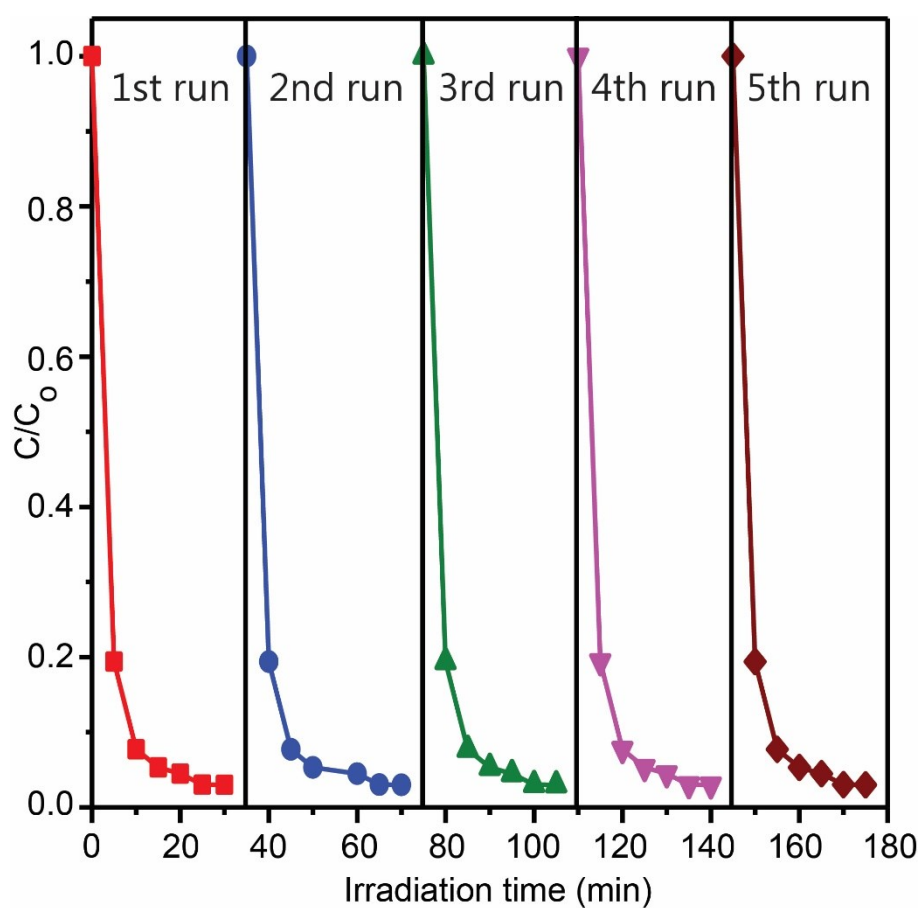


Fig. S8. The number of cycles for the photodegradation of 30 mg/L RhB in the presence of (1 g/L) N-NBO/C.

SUPPORTING TABLES

Table S1. EDX &XPS analysis of C, N, Nb, and O elements for N-NBO/C

EDX			XPS		
Element	Percentage by weight	Atomic percentage	Element	PP At. %	Atomic %
C K	37.27	64.56	C 1s	19.19	25.37
N K	0.39	0.58	N 1s	2.45	3.94
O K	19.37	25.20	O 1s	60.58	52.95
Nb K	42.98	9.63	Nb 1s	17.79	17.74
Total	100 %	100 %	Total	100 %	100 %

Table S2. Comparison of the photocatalysts conditions and conversation rate (%) between some reported catalysts for RhB degradation in aqueous solution and this work.

Photocatalyst conditions and conversation rate Photocatalysts	Light intensity (W)	RhB concentration	Surface area m ² /g	Catalyst amount (g)/100 mL	Conversation rate (%)	References
N-NBO/C	400	30 (mg/L)	260.37	0.1	98.5 after 30 mint	This work
Ag ₃ VO ₄ /TiO ₂ /GR	500	10 ppm	86	0.1	94 after 40 mint	¹
NBO-NNF	500	2 mM	--	0.1	99 after 90 mint	²
C-N/Nb ₂ O ₅ nanonets	400	10 (mg/L)	229	0.1	99 after 30 mint	³
C-Nb ₂ O ₅	500	5 (mg/L)	187	0.1	100 in 3 hours	⁴
C, N co-modified Nb ₂ O ₅ nanoneedles	300	10 (mg/L)	85.6	0.1	98 after 40 mint	⁵
HNb ₃ O ₈ -N	300	12 (mg/L)	11.1	0.3	100 after 50 mint	⁶
2D layered CdS@Ti ₃ C ₂ @TiO ₂	300	20 (mg/L)	42.38	0.05	100 in 2 hours	⁷
N-Nb ₂ O ₅	300	10 (mg/L)	14	0.3	98 after 30 mint	⁸

References

- 1 J. Wang, P. Wang, Y. Cao, J. Chen, W. Li, Y. Shao, Y. Zheng and D. Li, *Appl. Catal.*

- B Environ.*, 2013, **136–137**, 94–102.
- 2 F. Idrees, C. Cao, R. Ahmed, F. Butt, S. Butt, M. Tahir, M. Tanveer, I. Aslam and Z. Ali, *Sci. Adv. Mater.*, 2015, **7**, 1298–1303.
 - 3 F. A. Qaraah, S. A. Mahyoub, M. E. Hafez and G. Xiu, *RSC Adv.*, 2019, **9**, 39561–39571.
 - 4 S. Ge, H. Jia, H. Zhao, Z. Zheng and L. Zhang, *J. Mater. Chem.*, 2010, **20**, 3052.
 - 5 J. Xue, R. Wang, Z. Zhang and S. Qiu, *Dalt. Trans.*, 2016, **45**, 16519–16525.
 - 6 X. Li, N. Kikugawa and J. Ye, *Adv. Mater.*, 2008, **20**, 3816–3819.
 - 7 Q. Liu, X. Tan, S. Wang, F. Ma, H. Znad, Z. Shen, L. Liu and S. Liu, *Environ. Sci. Nano*, 2019, **6**, 3158–3169.
 - 8 X. Wang, G. Chen, C. Zhou, Y. Yu and G. Wang, *Eur. J. Inorg. Chem.*, 2012, **2012**, 1742–1749.