Supporting Information

A Highly Green Approach towards Aromatic Nitro Group Substitutions: Catalyst Free Reactions of Nitroimidazoles with Carbon Nucleophiles in Water

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Figure S1. ¹H NMR Spectrum (600 MHz, DMSO- d_6) of compound **1**.



Figure S2. ¹³C NMR spectrum (151 MHz, DMSO-*d*₆) of Compound 1.



Figure S3. ¹H NMR Spectrum (600 MHz, DMSO- d_6) of compound **2**.



Figure S4. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) compound of **2**.



Figure S5. ¹H NMR Spectrum (600 MHz, DMSO- d_6) of compound **3**.



Figure S6. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) of compound **3**.



Figure S7. ¹H NMR Spectrum (600 MHz, DMSO- d_6) of compound 4.



Figure S8. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) of compound **4**.



Figure S9. ¹H NMR Spectrum (600 MHz, DMSO- d_6) of compound **5**.



Figure S10. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) of compound **5**.



Figure S11. ¹H NMR Spectrum (600 MHz, DMSO- d_6) of compound **6**.



Figure S12. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) of compound of compound **6**.



Figure S13. ¹H NMR Spectrum (600 MHz, DMSO-*d*₆) of compound **7**



Figure S14. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) of compound 7.



Figure S15. ¹H NMR Spectrum (600 MHz, DMSO-*d*₆) of compound **8**.



Figure S16. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) of compound **8**.



Figure: S17. ¹H NMR Spectrum (600 MHz, DMSO- d_6) of compound **9**.



Figure S18. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) of compound **9**.



Figure S19. ¹H NMR Spectrum (600 MHz, DMSO- d_6) of compound **10**.



Figure S20. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) of compound **10**.



Figure S21. ¹H NMR Spectrum (600 MHz, DMSO- d_6) of compound **11**.



Figure S22. ¹³C NMR Spectrum (151 MHz, DMSO- d_6) of compound 11.

Structural Analysis with X-Ray

For X-ray measurements, Single crystals of **1**, **3** and **7** were mounted on a MiTeGen loop with grease and examined on a Bruker D8 Venture APEX di \Box ractometer equipped with Photon 100 CCD area detector at 296 (2) K using graphite-monochromated Mo-K_a radiation ($\lambda = 0.71073$ Å). Data were collected using the APEX-II software,¹ integrated using SAINT² and corrected for absorption using a multi-scan approach (SADABS).³ Final cell constants were determined from full least squares refinement of all observed reflections. The structure was solved using intrinsic phasing (SHELXT).⁴ All non-H atoms were located in subsequent difference maps and refined anisotropically with SHELXL-2014/7.⁵ H-atoms were added at calculated positions and refined with a riding model. The structures of **1**, **3** and **7** have been deposited with the CCDC (CCDC deposition numbers 1995006–1995008).

(1) APEX-II, Bruker AXS, Madison, WI, USA.

- (2) SAINT, Bruker AXS, Madison, WI, USA.
- (3) SADABS, Bruker AXS, Madison, WI, USA.
- (4) G. M. Sheldrick, Acta Crystallogr., Sect. A: Found. Adv., 2015, 71, 3.
- (5) SHELXTL, Bruker AXS, Madison, Wisconsin, USA, 2015.

Crystal Structures of the Representative Compounds



Figure S23: (A) Crystal Structure of compound **1** (B) hydrogen bonding in compound **1**, Hbonds are shown in blue.



Figure S24. Crystal Structures of Compound **3.** H-bond with lattice water molecule is shown in blue.



Figure S25. Crystal Structure of compound **5** (B) hydrogen bonding in compound **5**, Hbonds are shown in blue.









Figure S28. From left to right initial and final progress of **3**.



Figure S29. From left to right initial and final images of **4**.



Figure S31. From left to right initial and final images of **6**.



Figure S32. From left to right initial and final images of **7**.





Figure S33. From left to right initial and final images of **8**.



Figure S34 From left to right initial and final images of 9.



Figure S35. From left to right initial and final images of **10**.

S30



Figure S36. From left to right initial and final images of **11**.