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Supplementary Material

Synthesis of Narrow-Band Curled Carbon Nitride Nanosheets with High Specific Surface Area for Hydrogen Evolution from Water Splitting by Low-Temperature Aqueous Copolymerization to Form Copolymer

Wenbo Liu, Zhendong Zhang, Deguang Zhang, Runwei Wang, Zongtao Zhang*, Shilun Qiu

a State Key Laboratory of Inorganic Synthesis and Preparative Chemistry, College of Chemistry, Jilin University, Changchun 130012, China

E-mail: zzhang@jlu.edu.cn



Fig. S1 SEM image of the B-C₃N₄.



Fig. S2 TEM image of the B-C₃N₄.



Fig. S3 SEM image of the copolymer after thermal polycondensation.



Fig. S4 TEM image of $NS-C_3N_4$ with hierarchical porous structure.



Fig. S5 XRD image of copolymer after thermal polycondensation.



Fig. S6 thermogravimetric analysis-differential thermal analysis (TGA-DTA) measurement of copolymer without intercalation and copolymer after intercalation.



Fig. S7 UV–vis di□use-reflectance spectrum and wavelength-dependent AQE of NS-C₃N₄.



Fig. S8 VB XPS spectra of B-C₃N₄ and NS-C₃N₄.

Sample	Condition	BET (m^2/g)	HER ($\mu mol/g \cdot h$)	Times to B-	Ref
				C_3N_4	
$g-C_3N_4$ (w-N ₂)	Wet nitrogen and	211.2	1113.48 (3% Pt)	6	1
	reflux				
CN650NS	650 °C 4 h in air,	37.245	2627.8 (0.5 Pt)	24.6	2
	ultrasound 4 h,520				
	°C under hydrogen				
	argon mixture				
HC-CN	Ni-foam as the	39.24	808.5 (3% Pt)	20	3
	template				
NCN	vapor deposition	118	926 (3% Pt)	14	4
	method				
g-C ₃ N ₄ (580)-T	Thermal oxidation	92.8	1391(3% Pt)	—	5
	etching				
CN-2	Radio frequency	—	1227.8 (1% Pt)	—	6
	(RF) plasma				
	treatment				
NS-C ₃ N ₄	Thermal exfoliation	60.962	4061.8 (1% Pt)	37.5	This work
	copolymer				

copolymerTable S1. Summary of photocatalytic activity of g-C3N4 nanosheets with improved
photocatalytic activity.

Sample	C1s At. %	N1s At. %	O1s At. %
B-C ₃ N ₄	54.66	40.17	5.17
NS-C ₃ N ₄	54.96	42.2	2.84

Table S2. Atomic ratio of elements in $B-C_3N_4$ and $NS-C_3N_4$.

sample	CB (eV)	VB (eV)	Bandgap (eV)
B-C ₃ N ₄	-1.14	1.5	2.64
NS-C ₃ N ₄	-0.97	1.15	2.12

Table S3. Electronic band structure of $B-C_3N_4$ and $NS-C_3N_4$.

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