Supporting Information

Impact of photosensitizer orientation on the distance dependent photocatalytic activity in zinc phthalocyanine – nanoporous gold hybrid systems

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ESI-1 Supplementary data for the photooxidation of DPBF with H1-3 using different irradiation wavelengths

Fig. S1: UV Vis spectra for the photooxidation of DPBF with **H1-3** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.4 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-2 Supplementary data for the photooxidation of DPBF with H2-3 using different irradiation wavelengths

Fig. S2: UV Vis spectra for the photooxidation of DPBF with **H2-3** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at $\lambda = 415$ nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.4 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.





Fig. S3: UV Vis spectra for the photooxidation of DPBF with **H1-4** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 2.1 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-4 Supplementary data for the photooxidation of DPBF with H2-4 using different irradiation wavelengths

Fig. S4:

UV Vis spectra for the photooxidation of DPBF with **H2-4** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.8 x 10⁻¹⁰ mol as determined from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-5 Supplementary data for the photooxidation of DPBF with H1-5 using different irradiation wavelengths

Fig. S5:

UV Vis spectra for the photooxidation of DPBF with **H1-5** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.5 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.





Fig. S6:

UV Vis spectra for the photooxidation of DPBF with **H2-5** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.5 x 10⁻¹⁰ mol as determined from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-7 Supplementary data for the photooxidation of DPBF with H1-6 using different irradiation wavelengths

Fig. S7: UV Vis spectra for the photooxidation of DPBF with **H1-6** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.5 x 10⁻¹⁰ mol as determined from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-8 Supplementary data for the photooxidation of DPBF with H2-6 using different irradiation wavelengths

Fig. S8:

UV Vis spectra for the photooxidation of DPBF with **H2-6** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.7 x 10⁻¹⁰ mol as determined from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-9 Supplementary data for the photooxidation of DPBF with H1-7 using different irradiation wavelengths

Fig. S9:

UV Vis spectra for the photooxidation of DPBF with **H1-7** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.6 x 10⁻¹⁰ mol as determined from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-10 Supplementary data for the photooxidation of DPBF with H2-7 using different irradiation wavelengths

Fig. S10:

UV Vis spectra for the photooxidation of DPBF with **H2-7** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.4 x 10⁻¹⁰ mol as determined from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-11 Supplementary data for the photooxidation of DPBF with H1-8 using different irradiation wavelengths

Fig. S11: UV Vis spectra for the photooxidation of DPBF with **H1-8** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.3 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-12 Supplementary data for the photooxidation of DPBF with H2-8 using different irradiation wavelengths

Fig. S12: UV Vis spectra for the photooxidation of DPBF with **H2-8** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.2 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.





Fig. S13: UV Vis spectra for the photooxidation of DPBF with **H1-9** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.4 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.





Fig. S14: UV Vis spectra for the photooxidation of DPBF with **H2-9** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.7 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-15 Supplementary data for the photooxidation of DPBF with H1-10 using different irradiation wavelengths

Fig. S15: UV Vis spectra for the photooxidation of DPBF with **H1-10** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.4 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-16 Supplementary data for the photooxidation of DPBF with H2-10 using different irradiation wavelengths

Fig. S16: UV Vis spectra for the photooxidation of DPBF with **H2-10** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.6 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-17 Supplementary data for the photooxidation of DPBF with H1-11 using different irradiation wavelengths

Fig. S17: UV Vis spectra for the photooxidation of DPBF with **H1-11** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.5 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.



ESI-18 Supplementary data for the photooxidation of DPBF with H2-11 using different irradiation wavelengths

Fig. S18: UV Vis spectra for the photooxidation of DPBF with **H2-11** as hybrid photocatalyst employing either a 550 nm cut-on filter (a), a 700 nm bandpass filter (b) or a 550 nm bandpass filter (c) for irradiation. The amount of converted DPBF was determined using Labert Beers law and the extinction coefficinet of 23000 L mol⁻¹ at λ = 415 nm. Turnover numbers (TON) were calculated from the amount of converted DPBF and the illuminated photosensitizer amount of 1.7 x 10⁻¹⁰ mol as determined from ICP-MS. Turnover frequencies (TOF) were obtained by linear regression from the plots of TON vs. reaction time in the linear regime before saturation effects become dominant.

ESI-19 Supplementary data fot the synthesis of the ZnPc derivatives 1 and 2



Scheme S1: Synthesis scheme for the photosensitizers 2,9,16,23-tetrakis(2-propyn-1-yloxy)phthalocyanine zinc(II) (1) and 2,9,16,23-tetrakis(5-hexyn-1-yloxy) phthalocyanine zinc(II) (2).

ESI-20 Supplementary data fot the synthesis of azidoalkylthioacetates 3 - 11



Scheme S2: General synthesis scheme for the azidoalkylthioacetate SAM precursors **3-11** from the corresponding *n*-bromo-1-alkanol via *n*-azidoalkan-1-ol and *n*-azidoalkyl methanesulfonate.