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Supporting Information for

Near-infrared and metal-free tetra(butylamino)phthalocyanine

nanoparticles for dual modal cancer phototherapy

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Fig. S1 ¹H NMR spectra of 3-butylaminophthalonitrile (top) and 4butylaminophthalonitrile (bottom) in $CDCl_3$. The signal at 7.26 ppm was attributed to the residual $CHCl_3$.



Fig. S2 MALDI-TOF mass spectra for α -Pc and β -Pc.



Fig. S3 ¹H NMR spectra of α -Pc (top) in CDCl₃/[D₆]DMSO (6:1) and *B*-Pc (bottom) in [D₆]DMSO (with the signals due to residues of CHCl₃, H₂O and DMSO denoted as *,[#] and [&], respectively).



Fig. S4 Pictures of **HSA-** α **-Pc** NP (a) and **HSA-** θ **-Pc** NP (b) in water (both at 80 μ M, Pc equiv) after standing for different time. The DLS analysis of **HSA-** α **-Pc** NP (c) and **HSA-** θ **-Pc** NP (d) dispersed in aqueous solution. The DLS stability of HSA- α **-Pc** NP (e) and **HSA-** θ **-Pc** NP (f) in various media for five days, and the data were based on the results of 3 independent experiments.



Fig. S5 (a) The UV-vis spectra of α -Pc in V(CHCl₃) :V(DMSO) = 95:5 solution at different concentrations. (b) The calibration curve of α -Pc at 786 nm. The absorbance data were presented as mean ± SD (n = 3). (c) The procedure of calculating the amounts of α -Pc in NP aqueous solution by using organic solvent extraction.



Fig. S6 (a) The UV-vis spectra of **\beta-Pc** in V(CHCl₃) : V(DMSO) = 95:5 solution at different concentrations. (b) The calibration curve of **\beta-Pc** at 742 nm. The absorbance data were presented as mean ± SD (n = 3). (c) The procedure of calculating the amounts of **\beta-Pc** in NP aqueous solution by using organic solvent extraction.



Fig. S7 The zeta potentials of **HSA**- α -**Pc** NP (a) and **HSA**- β -**Pc** NP (b) in water solution at 25°C (both at 80 μ M, Pc equiv).



Fig. S8 Photothermal property of **HSA-** α **-Pc** NP in aqueous solution under 808 nm laser irradiation. (a) Photothermal effect and the laser was switched off after irradiation for 15 min (80 μ M Pc equiv, 1.5 W·cm⁻²). The inset presents the plot of cooling time (15 min) *vs* the negative natural logarithm of the temperature driving force obtained from the cooling stage. (b) Temperature variations for 6 cycles of the (1.5 W·cm⁻², 15 min per cycle) at the concentration 20 μ M (Pc equiv).



Fig. S9 Photothermal property of **HSA-***B***-Pc** NP dispersed in water under 808 nm laser irradiation. (a) Temperature elevation in dependence of light intensity at the concentration of 80 μ M (Pc equiv). (b) Concentration-dependent temperature elevation with an intensity of 0.9 W·cm⁻². (c) Photothermal effect and the laser was switched off after irradiation for 15 min (80 μ M, Pc equiv., 1.5 W·cm⁻²). (d) Plot of cooling time (15 min) *vs* the negative natural logarithm of the temperature driving force obtained from the cooling stage. (e) Temperature variations for 6 cycles of the (1.5 W·cm⁻², 15 min per cycle) at the concentration 20 μ M (Pc equiv).



Fig. S10 Singlet oxygen generation ability of the **HSA-** α **-Pc** and **HSA-** β **-Pc** NPs. (a) Timedependent UV-vis absorption of DPBF/DMF (1.33 mM, 62 μ L) at 420 nm with **HSA-** α -**Pc** and **HSA-** β **-Pc** (both at 20 μ M Pc equiv, 2.0 mL) in aqueous solution under 808 nm laser irradiation (1.5 W·cm⁻²) for 5 min. (b) Comparison of the decay rate of DPBF, **HSA-** α -**Pc** and **HSA-** β **-Pc**.



Fig. S11 Fluorescence spectra of α -Pc (5 μ M, CH₂Cl₂, excited at 300 nm) and θ -Pc (5 μ M, CH₂Cl₂, excited at 309 nm) as well as HSA- α -Pc and HSA- θ -Pc NPs (both at 20 μ M, Pc equiv) in water upon irradiated at 405 nm.



Fig. S12 *In vitro* CLSM images of MCF-7 cells treated with **HSA**- α -**Pc** and **HSA**- β -**Pc** NPs (both at 20 μ M, Pc equiv) for 2 h, using a green channel for phthalocyanine (λ_{ex} = 405 nm, 440-480 nm), scale bar 25 μ m.



Fig. S13 (Left) Thermal imageries of the nude mice. (Right) Body weight of the mice in various groups during the treatment. The data are presented as the mean \pm SD (n = 4).

Table	S1.	^{1}H	NMR	spectroscopic	data	(δ)	for	3-butylaminophthalonitrile,	4-
butyla	mino	phth	naloniti	rile, α-Ρc and β -	Pc.				

Compound (Solvent)	δ			
3-butylaminophthalonitrile	7.47-7.43 (t, 1H,-C ₆ H ₃ -), 7.01-6.99 (d, 1H,-C ₆ H ₃ -), 6.91-6.89 (d			
(CDCl ₃)	1H, -C ₆ H ₃ -), 4.81 (s, 1H, -NH-), 3.25-3.21 (t, 2H, -CH ₂ -), 1.70-1.6			
	(m, 2H, -CH ₂ -), 1.48-1.43 (m, 2H, -CH ₂ -), 1.00-0.96 (t, 3H, -CH ₃)			
4-butylaminophthalonitrile	7.51-7.49 (d, 1H,-C ₆ H ₃ -), 6.83 (s, 1H,-C ₆ H ₃ -), 6.75-6.73 (d, 1H, -			
(CDCl ₃)	C ₆ H ₃ -), 5.34 (s, 1H, -NH-), 3.19-3.16 (t, 2H, -CH ₂ -), 1.68-1.60 (m,			
	2H, -CH ₂ -), 1.45-1.40 (m, 2H, -CH ₂ -), 1.00-0.96 (t, 3H, -CH ₃).			
α-Pc	7.80-6.58 (-Рс _{с6н3} -), 3.64-3.38 (4Н, -NH-), 2.20-1.78 (-CH ₂ -),			
(CDCl ₃ /[D ₆]DMSO (6:1))	1.18-0.79 (-CH ₂ CH ₂ CH ₃), -1.95 (-Pc _{NH} -).			
<i>6</i> -Pc	8.95-8.82 (4H, -Рс _{сбнз} -), 8.37-8.20 (4H, -Рс _{сбнз} -), 7.38-7.34 (4H,			
([D ₆]DMSO)	-Рс _{с6н3} -), 6.96-6.79 (4Н, -NH-), 3.44-3.30 (-NH <u>CH₂-), 1.81-0.93</u>			
	(28H, -CH ₂ CH ₂ CH ₃).			

Table S2. Summary of recent typical Pc-nanomaterials for PTT/PDT synergistic phototherapy in

 living cells by a single light source.

comple	light course (M/cm^{-2})	η	PS	MTT	ref.
sample		(%)	(µM)	(%)	
Pc@HSNs	730 nm (1.5)	37	~575	30	1
NanoPcTB	655 nm (2.5)		6	20	2
ZnPc NW	808 nm (3.0)		~205	50	3
O ₂ @PFH@HMoS _x -HSA/AIPc	670 nm (1.0)	40	~255	25	4
GR-TSCuPc	650 nm (3.0)		~15	40	5

SWNHs-TSCuPc	650 nm (3.0)		~15	15	6
ZnPc NPs	650 nm (0.7)	31	20	~15	7
НSА- <i>в</i> -Рс	808 nm (1.5)	54	20	31	this work
HSA-α-Pc	808 nm (1.5)	56	20	20	this work

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