Supplementary information

Comparative study on different strategies for synthesizing all-silica DD3R zeolite crystals with uniform morphology and size

Anna Peng,^a Xinqing Lu,^{*a,b} Rui Ma,^{a,b} Yanghe Fu,^{a,b} Shuhua Wang^c and Weidong Zhu^{*a,b,c}

^a Zhejiang Engineering Laboratory for Green Syntheses and Applications of Fluorine-Containing Specialty Chemicals, Institute of Advanced Fluorine-Containing Materials, Zhejiang Normal University, 321004 Jinhua, People's Republic of China
^b Key Laboratory of the Ministry of Education for Advanced Catalysis Materials, Institute of Physical Chemistry, Zhejiang Normal University, 321004 Jinhua, People's Republic of China
^c National Engineering Technology Research Center of Fluoro-Materials, Zhejiang Juhua Technology Center Co., Ltd., 324004 Quzhou, People's Republic of China

Corresponding Authors

E-mails: xinqinglu@zjnu.cn (X. Lu); weidongzhu@zjnu.cn (W. Zhu)



Fig. S1 SEM image of the "amorphous" DD3R seeds.



Fig. S2 SEM image of the "amorphous" ZSM-58 seeds.



Fig. S3 XRD patterns of the DD3R crystals synthesized using "amorphous" DD3R seeds with different amounts.



Fig. S4 XRD patterns of the DD3R crystals synthesized using 0.1 wt.% of "amorphous" DD3R seeds with different KF/SiO₂ molar ratios.



Fig. S5 XRD patterns of the DD3R crystals synthesized using 0.1 wt.% of "amorphous" DD3R seeds with different gel molar compositions: 47 1-ADA : $100 \text{ SiO}_2 : 100 \text{ KF} : 6000 \text{ H}_2\text{O}$ (a) and 47 1-ADA : $100 \text{ SiO}_2 : 100 \text{ KF} : 8000 \text{ H}_2\text{O} : 2 \text{ KOH}$ (b).



Fig. S6 TGA curves of the synthesized DD3R samples represented in Table 2.