

Supporting Information

Synthesis and characterization of a supported Pd complex on volcanic pumice laminates textured by cellulose for facilitating Suzuki–Miyaura cross-coupling reactions

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Table S1. A brief list of the applied materials and equipment.

Material / instrument	Brand
Pumice powder	Was purchased from a market in Tehran, IRAN
Hydrochloric acid	Merck, 37%
Cellulose powder	Sigma Aldrich (fibers, medium)
Palladium chloride	Sigma Aldrich (99%)
Potassium hydroxide	Merck, pellets for analysis
Potassium carbonate	Sigma Aldrich (ACS reagent)
Sodium borohydride	Sigma Aldrich ($\geq 98.0\%$)
Triphenylphosphine	Sigma Aldrich (for synthesis)
Solvents	Merck
Aryl halide derivatives	Sigma Aldrich
Phenylboronic acid	Sigma Aldrich (95.0%)
Silica gel	Sigma Aldrich (for column chromatography, 60)
Ball mill	Retsch PM-100, Retsch GmbH & amp, Germany
Furnace	Muffle Furnace (Omron E5CC)
Heater-stirrer	HEIDOLPH Magnetic Stirrer with Heating
Ultrasound cleaner bath	KQ-250 DE
Oven	Memmert (UN30)
Crucible	EISCO (porcelain)
Glassware	Isolab
TLC plate	Merck (0.2 mm, 60 F254 aluminium sheets)
FT-IR spectrometer	Shimadzu IR-470 (KBr pellets)
EDX spectrometer	Numerix DXP-X10P
VSM	Lakeshore 7407
TGA	STA504
FESEM	Sigma-Zeiss microscope with attached camera
TEM	Philips CM-12
BET	Micromeritics ASAP 2010
XPS	ESCALAB Xi ρ , and Thermo Scientific
Melting point measurement apparatus	Capillary melting point
NMR spectrometer	Bruker FT-NMR

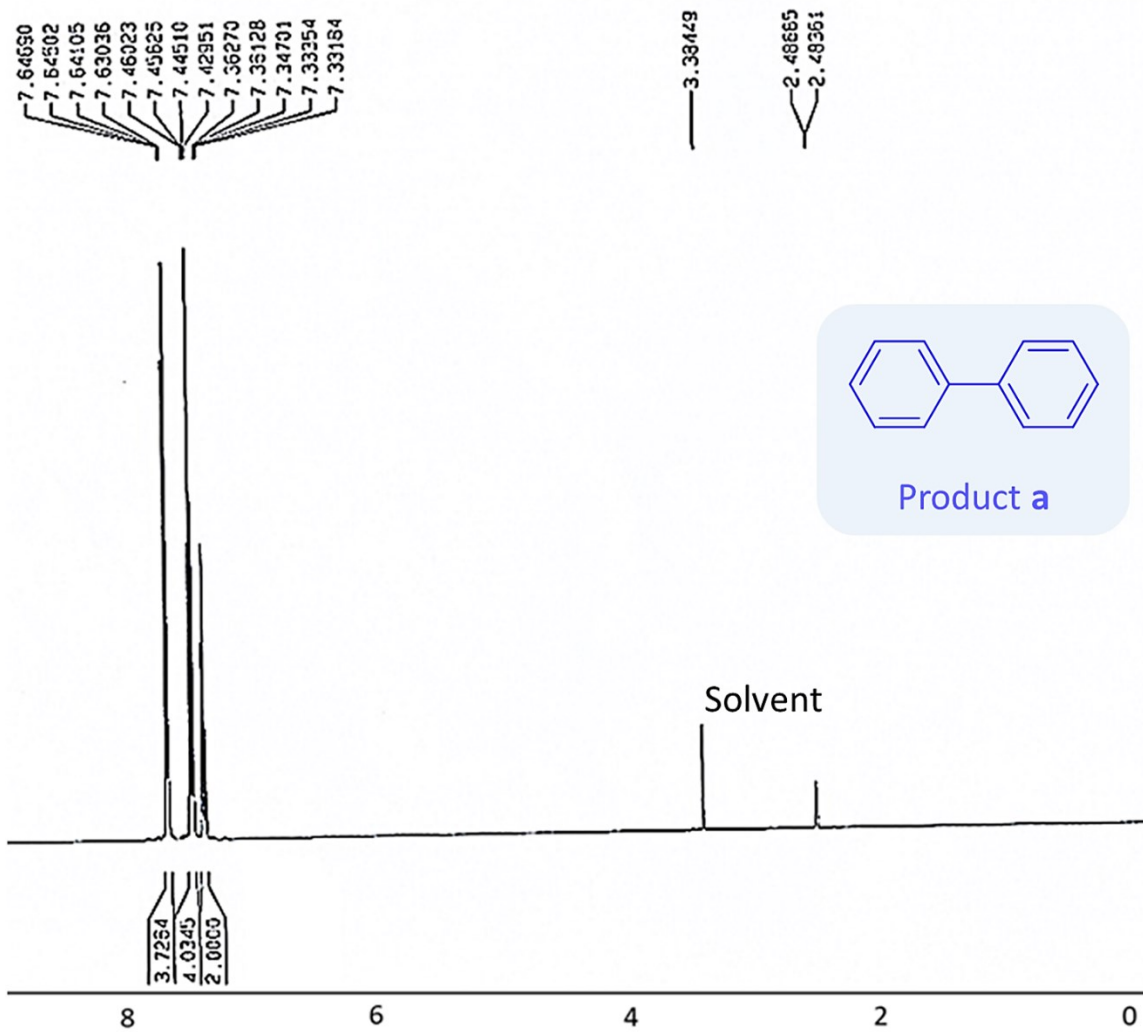


Figure S1. ¹H-NMR spectrum of biphenyl (a).

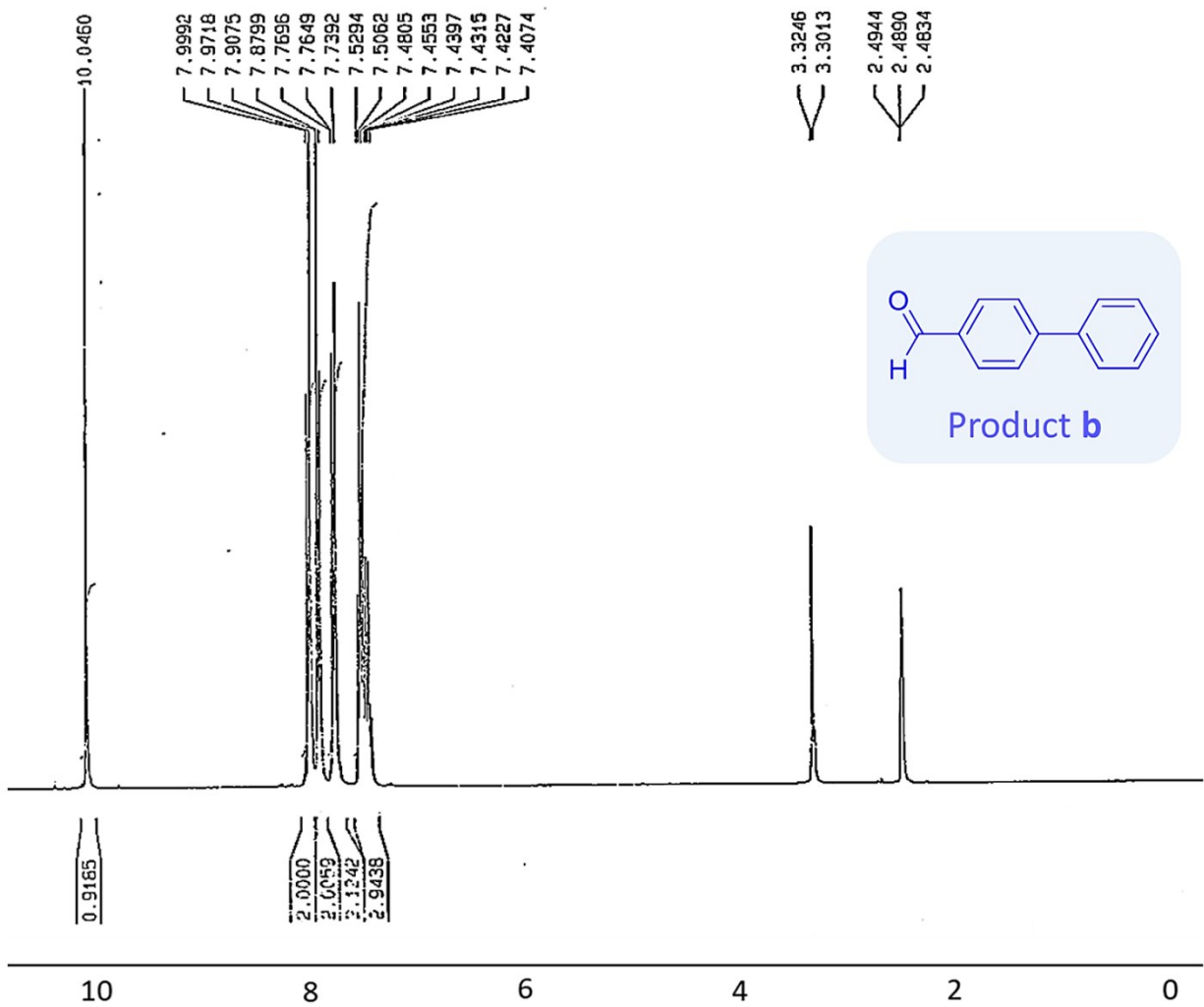


Figure S2. $^1\text{H-NMR}$ spectrum of 4-carbaldehyde-biphenyl (**b**).

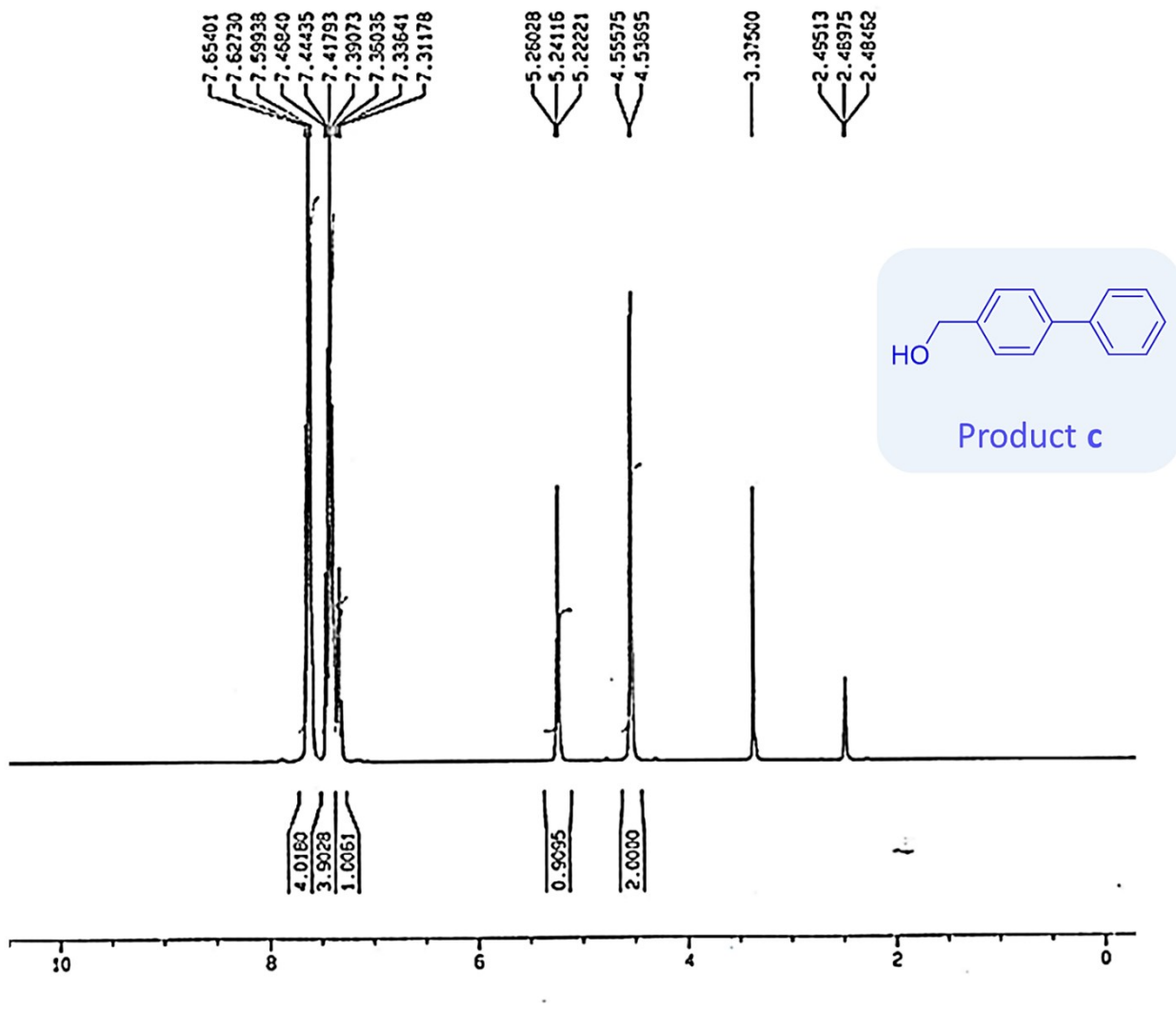


Figure S3. ¹H-NMR spectrum of 4-methanol-biphenyl (c).

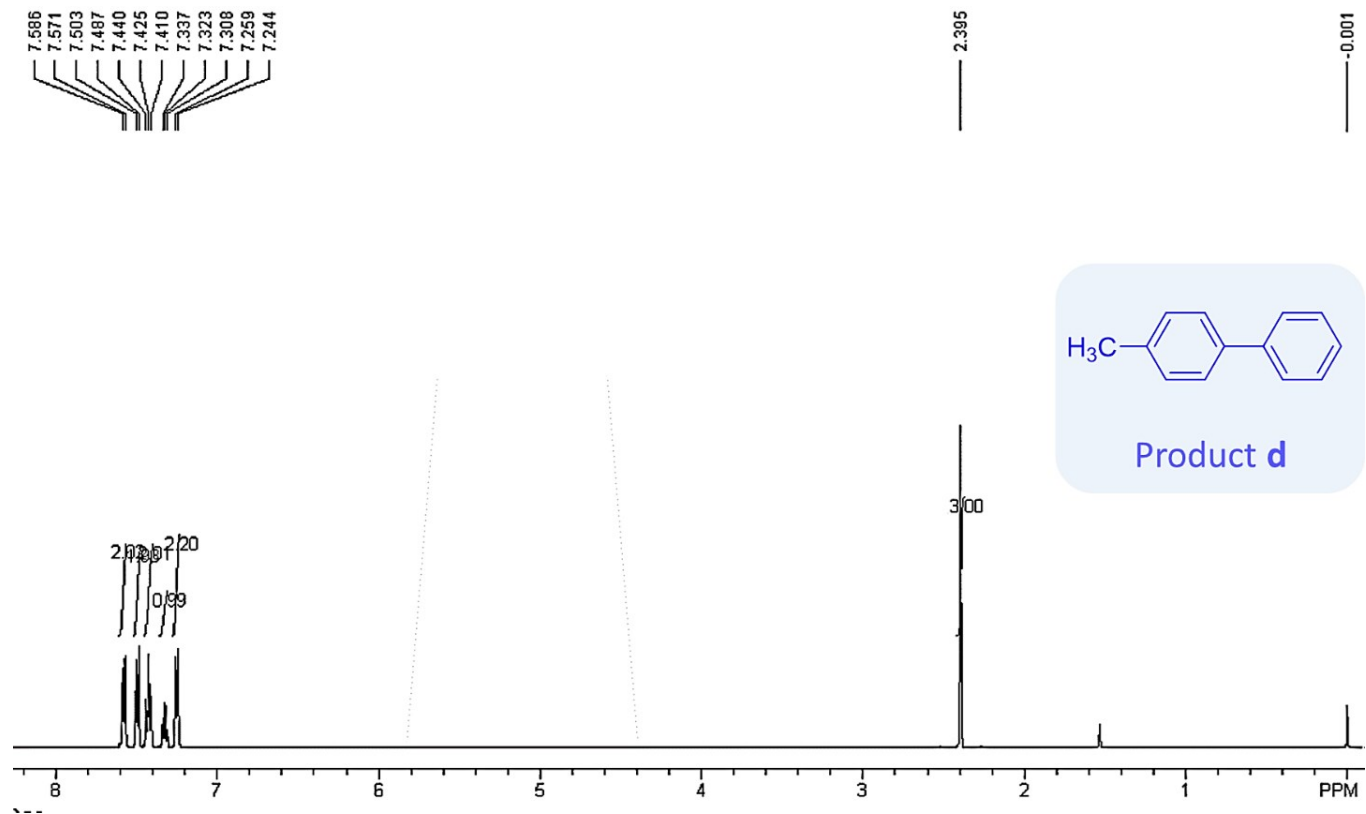


Figure S4. ¹H-NMR spectrum of 4-methyl-biphenyl (d).

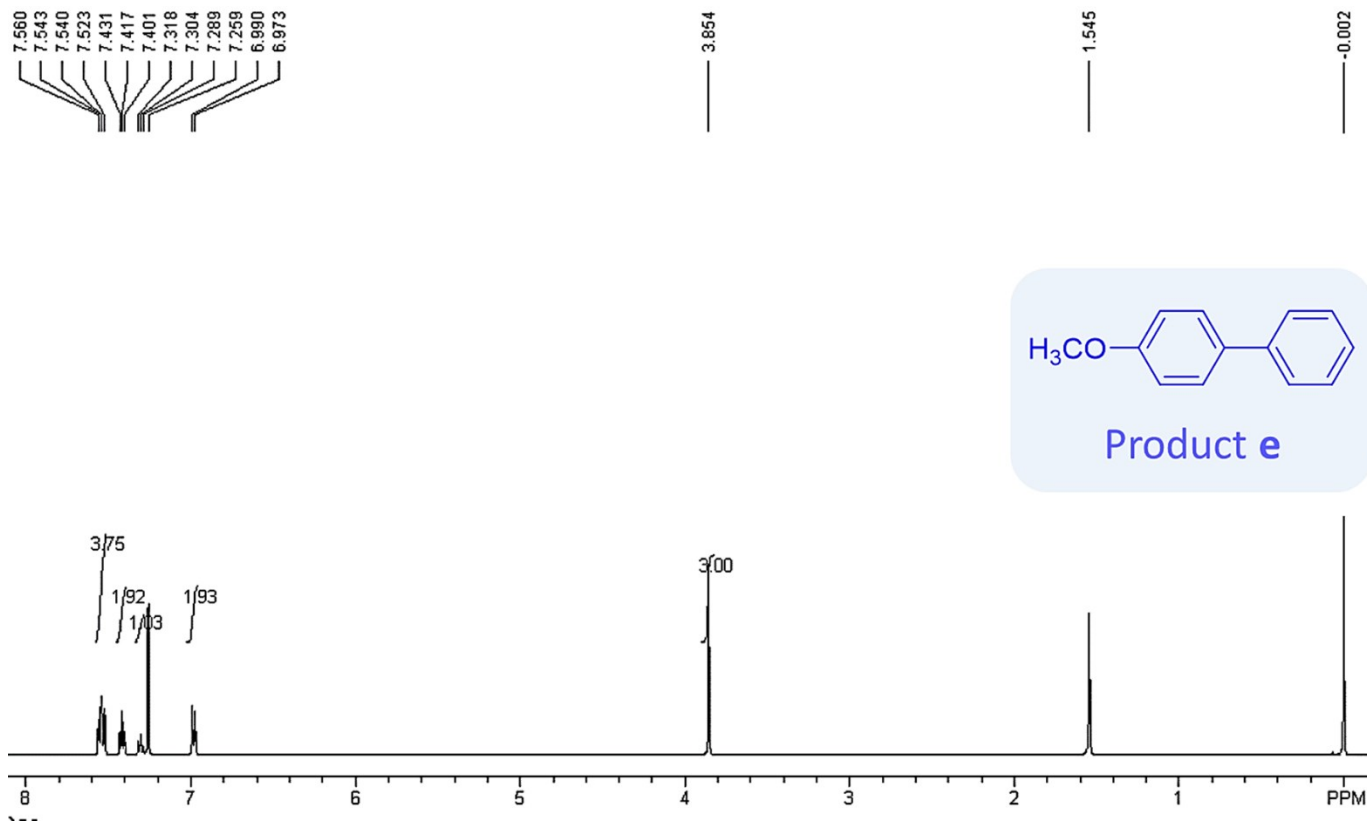


Figure S5. ¹H-NMR spectrum of 4-methoxy-biphenyl (e).

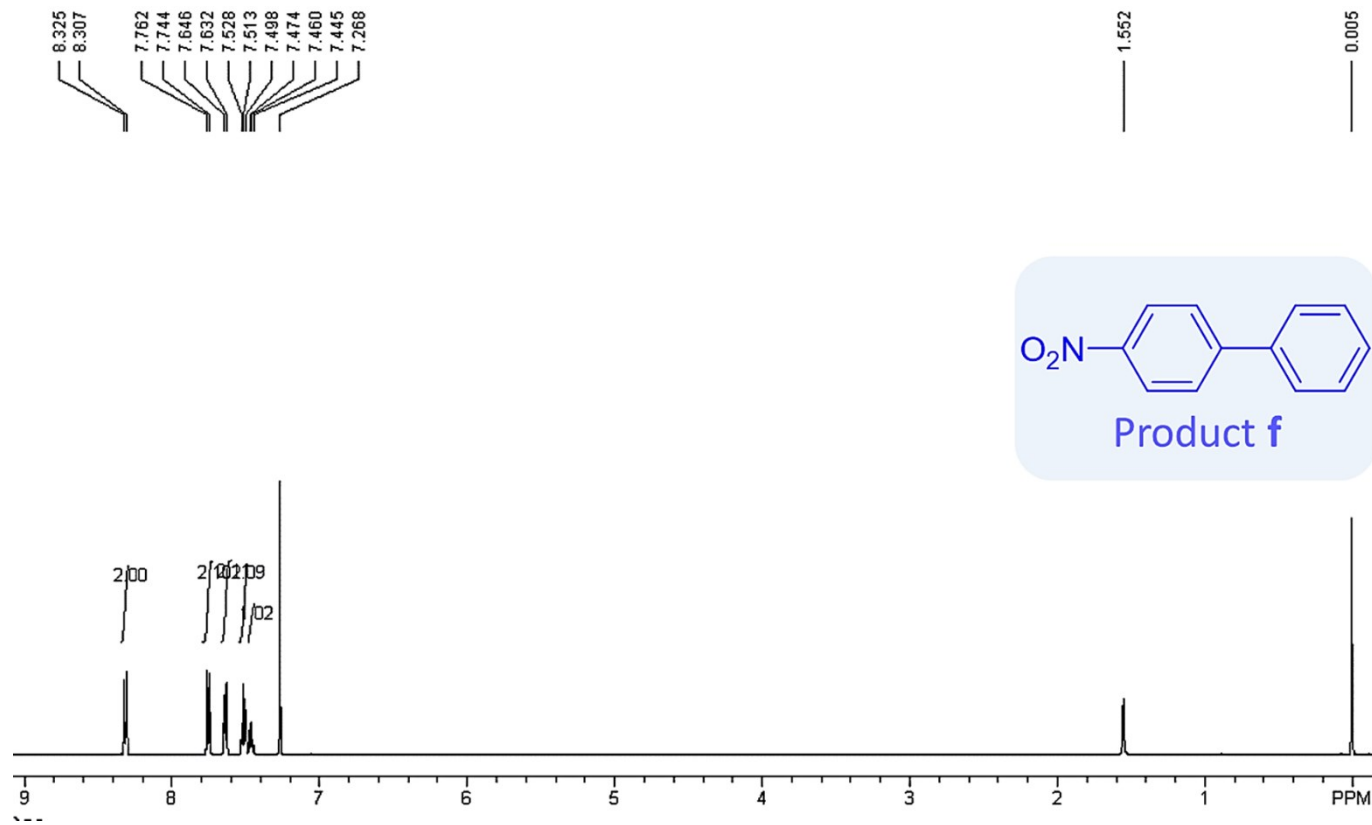


Figure S6. ¹H-NMR spectrum of 4-nitro-biphenyl (**f**).

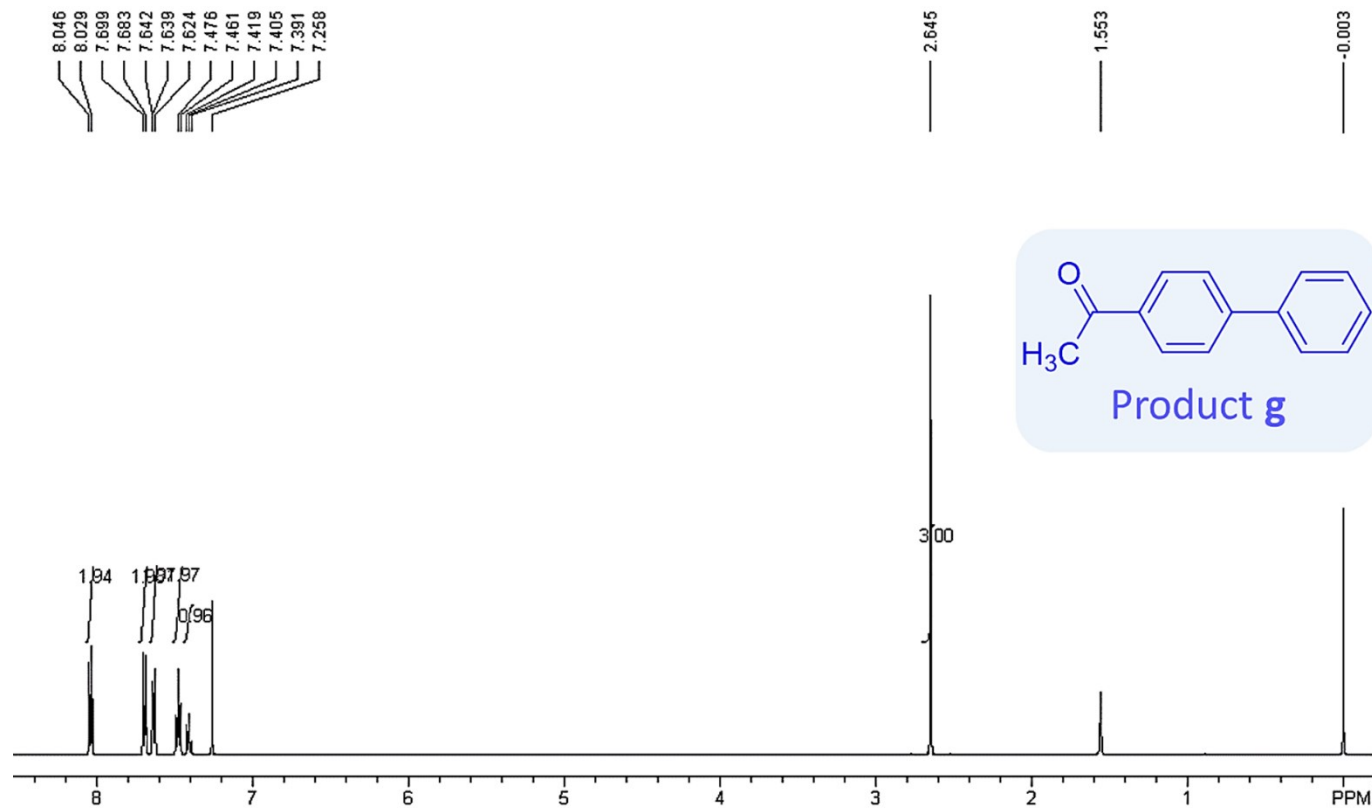


Figure S7. ¹H-NMR spectrum of 1-biphenyl-4-yl-ethanone (g).

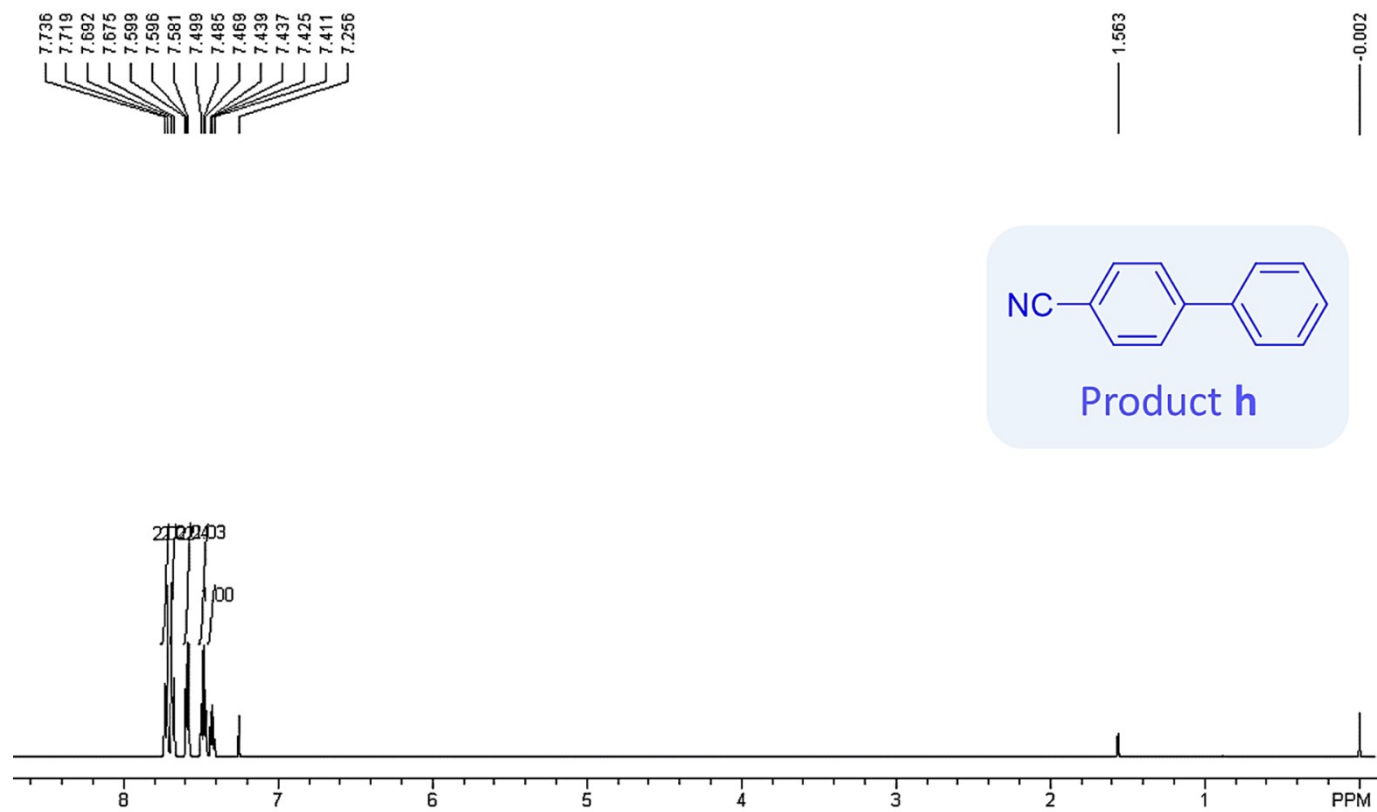


Figure S8. $^1\text{H-NMR}$ spectrum of biphenyl-4-carbonitrile (**h**).

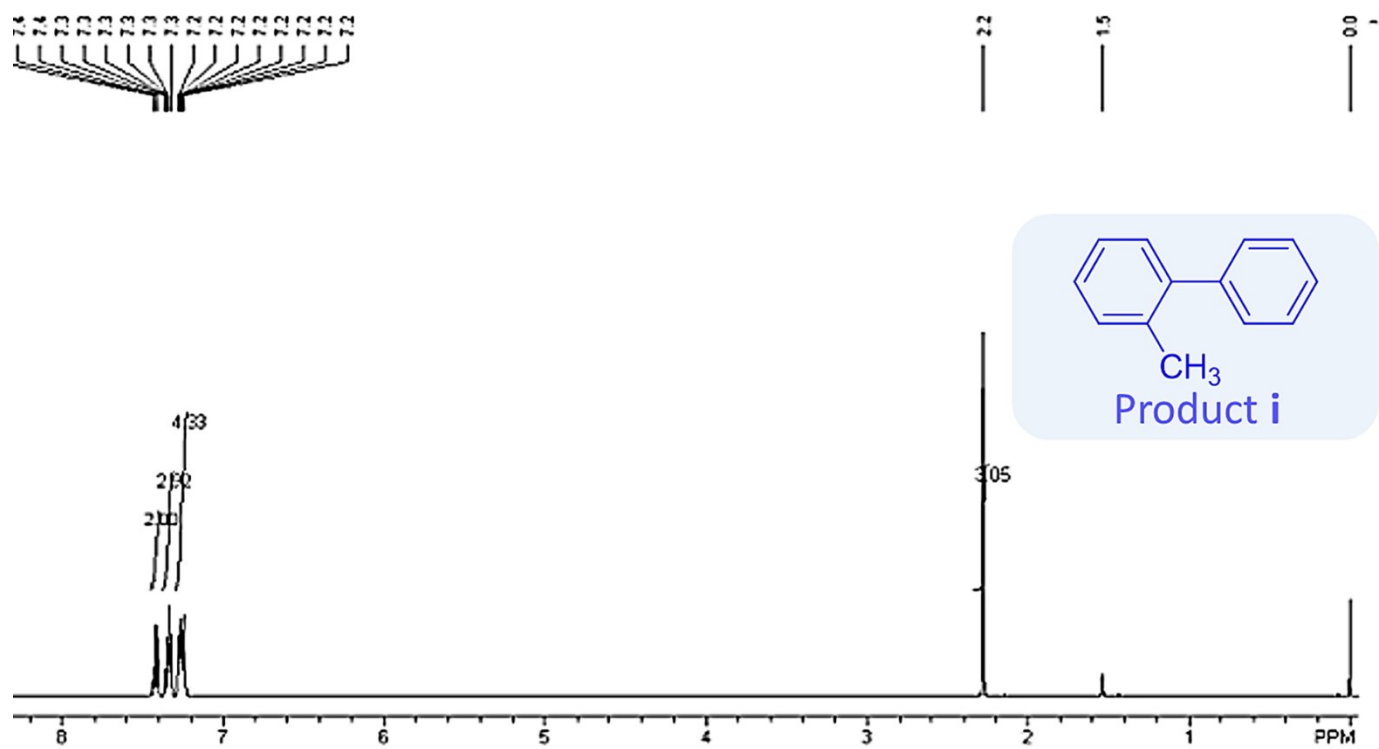


Figure S9. $^1\text{H-NMR}$ spectrum of 2-methyl-biphenyl (i).

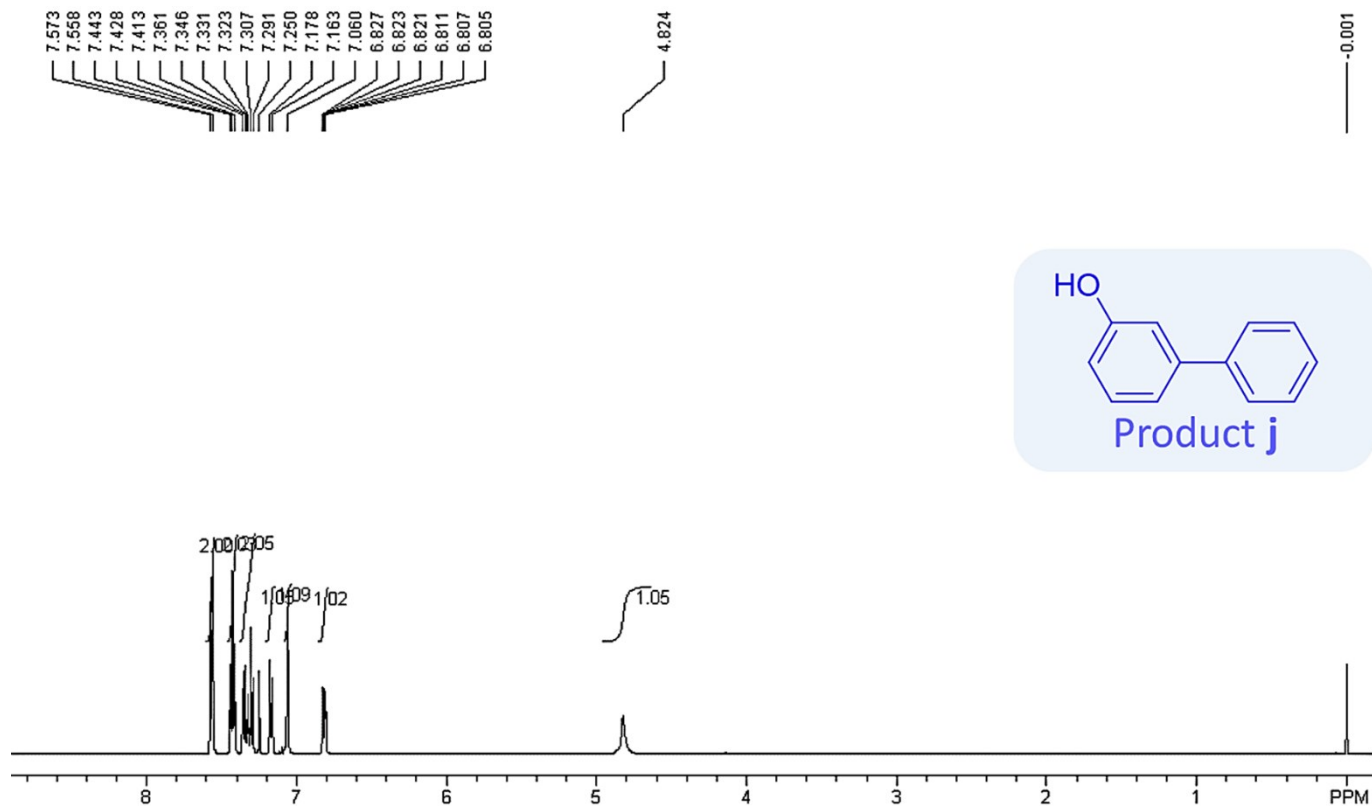


Figure S10. ¹H-NMR spectrum of biphenyl-3-ol (**j**).

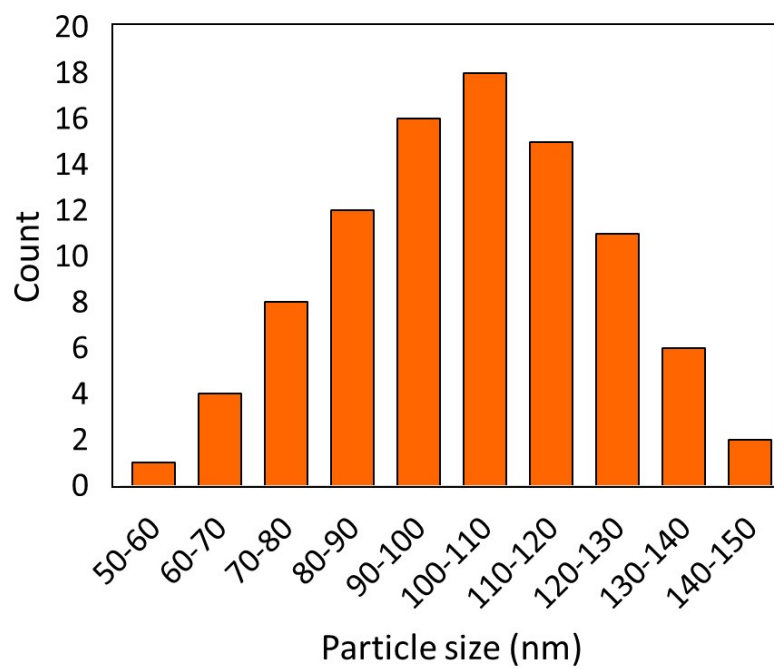


Figure S11. Size distribution diagram of the formed Pd nanoparticles.

Calculations of mol% of VPMP@CLS-Pd catalyst:

4-Iodonitrobenzene (as reactant): 1.0 mmol = 0.249 g, and VPMP@CLS-Pd (as catalyst): 0.01 g were used.

From EDX analysis (Figure 2b), 3.5 wt% of the total weight of catalyst is related to Pd nanoparticles.

$3.5 \times 0.01 \text{ g} / 100 = 0.00035 \text{ g}$ (pure weight of Pd nanoparticles in 0.01 g of catalyst)

$\Rightarrow (0.00035 / 0.249) \times 100 = \mathbf{0.14 \text{ wt\%}}$ (weight percentage of the applied catalyst)

$0.00035 \text{ g (Pd)} = 0.00328 \text{ mmol (Pd)}$

$\Rightarrow (0.00328 \text{ mmol of Pd} / 1 \text{ mmol of reactant}) \times 100 = \mathbf{0.33 \text{ mol\%}}$