

## Supplementary Information

# Phosphorus/Nitrogen Co-Doped and Bimetallic MOFs-Derived Cathode for All-Solid-State Rechargeable Zinc-Air Battery

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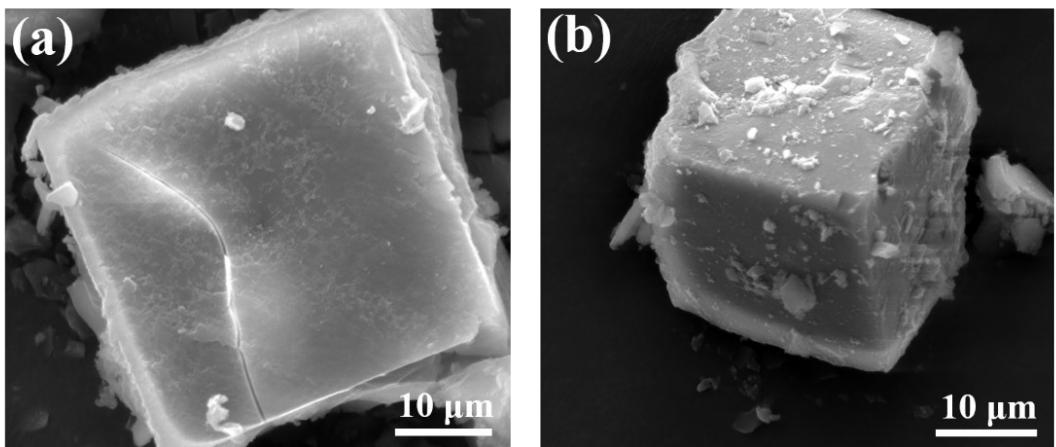
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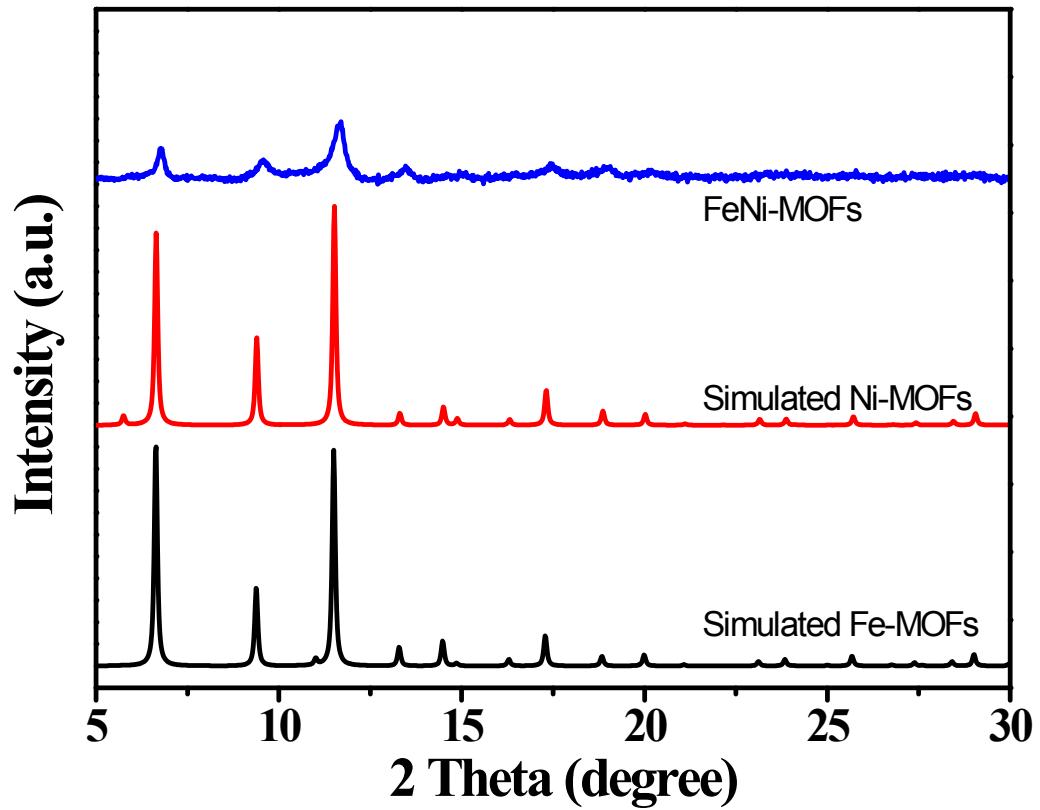
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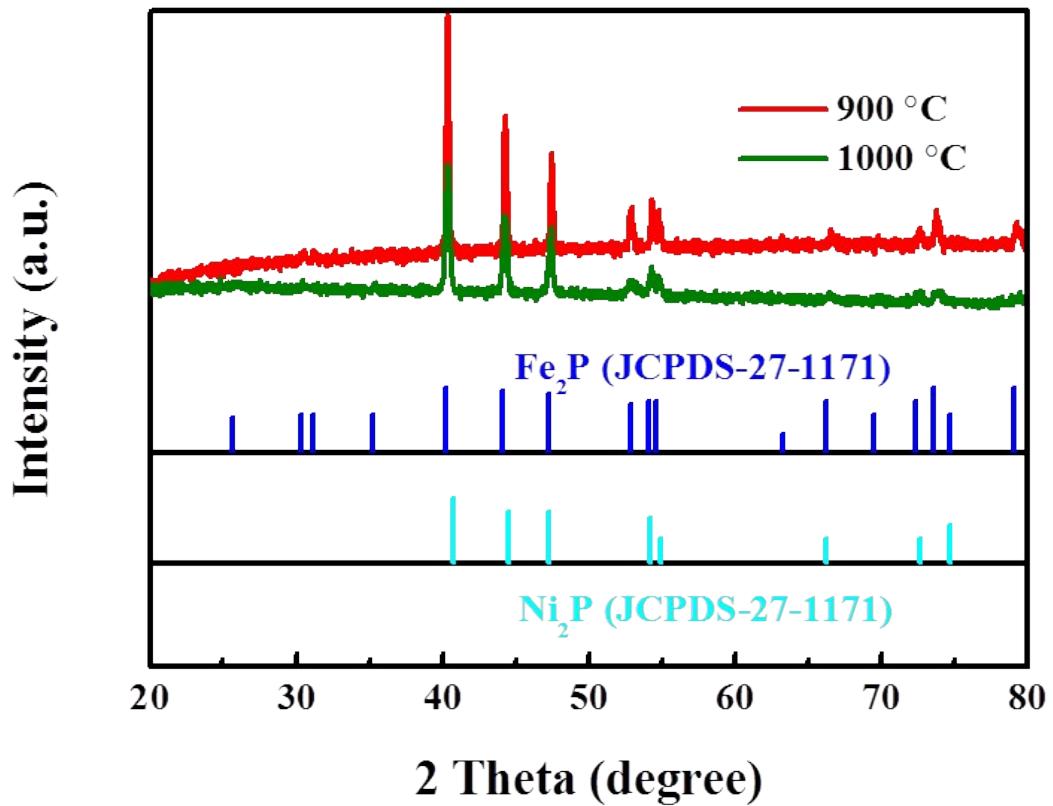
<sup>1</sup>These authors contributed equally to this work.



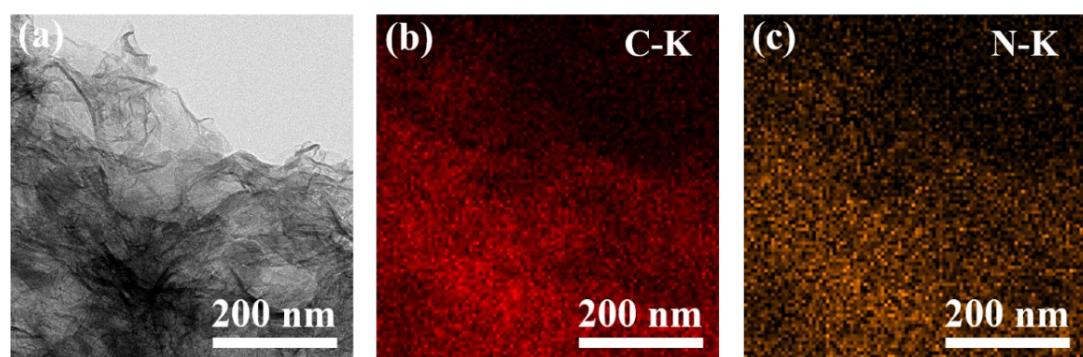
**Fig. S1.** SEM images of FeNi-MOFs. (a) Top view and (b) Side view.



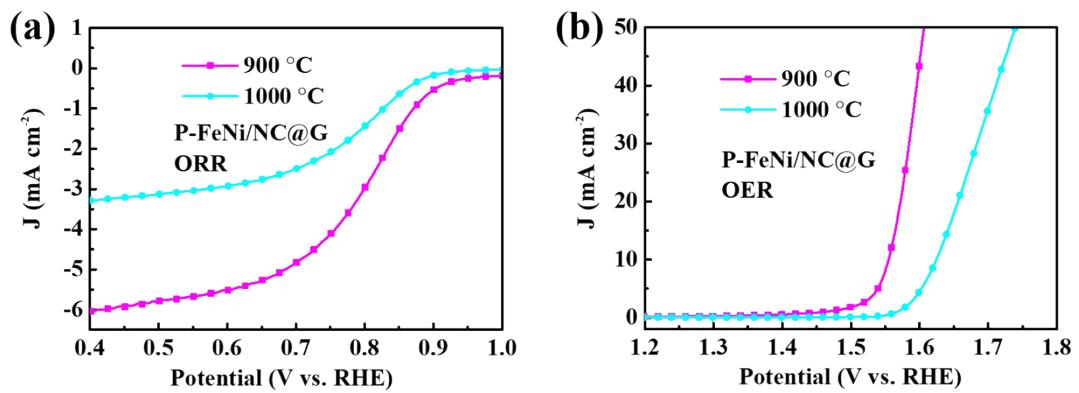
**Fig. S2.** The XRD pattern of FeNi-MOFs.



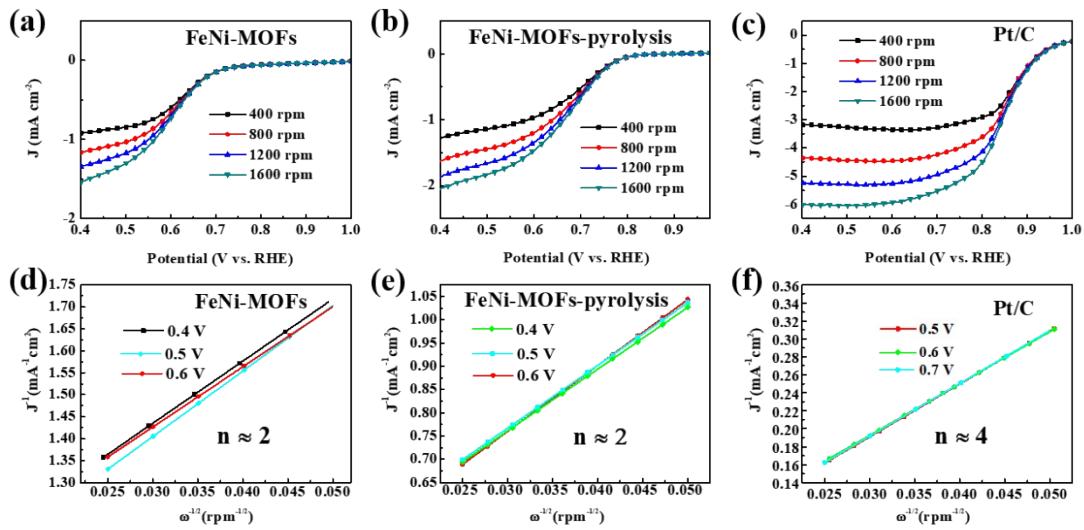
**Fig. S3.** The XRD patterns of P-FeNi/NC@G synthesized at different temperature.



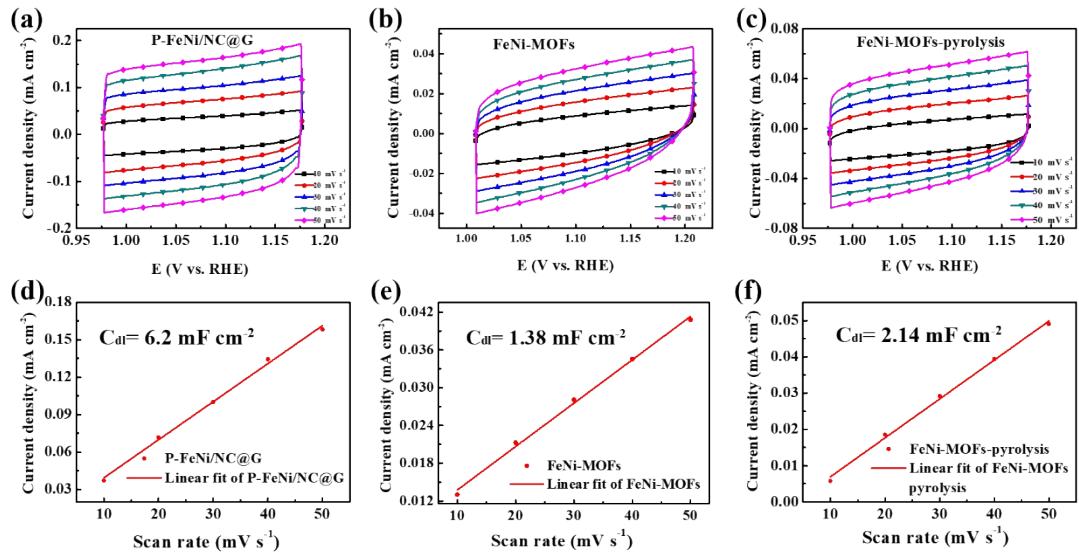
**Fig. S4.** TEM image and the corresponding EDS elemental mapping images of C, N elements.



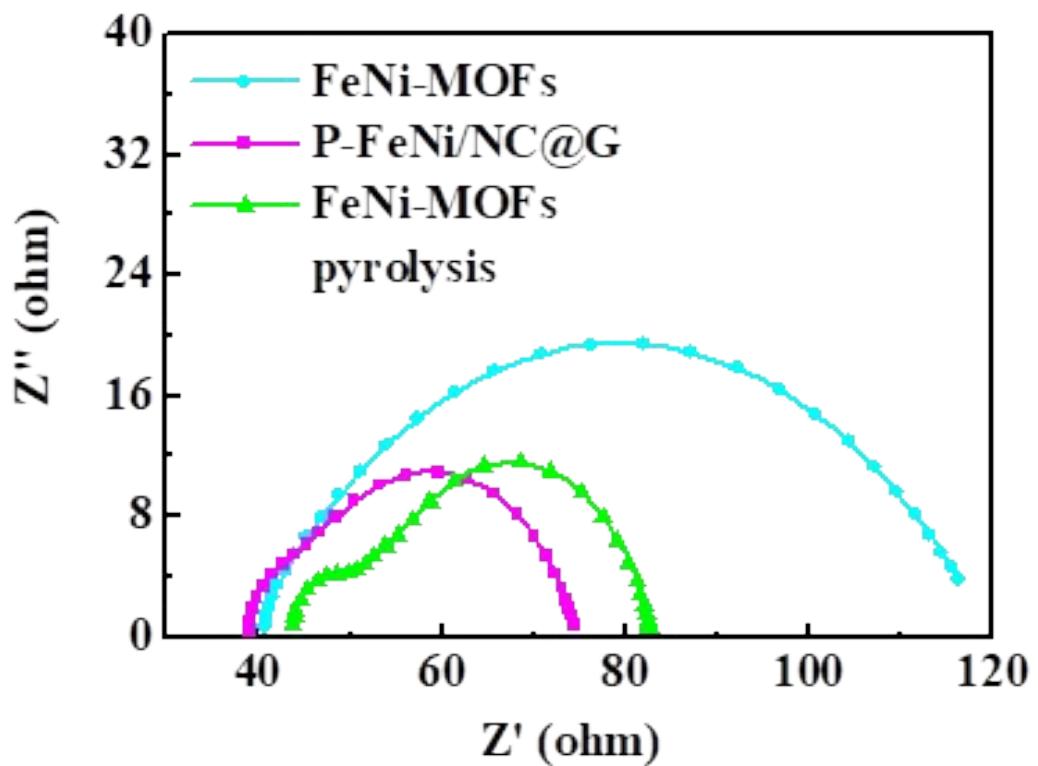
**Fig. S5.** (a) ORR and (b) OER polarization curves tested in  $\text{O}_2$ -saturated 0.1 M KOH at 1600 rpm. The P-FeNi/NC@G catalysts are synthesized at different pyrolysis temperature of 900 °C and 1000 °C in argon atmosphere.



**Fig. S6.** ORR polarization curves of (a) FeNi-MOFs, (b) FeNi-MOFs-pyrolysis and (c) commercial Pt/C tested in O<sub>2</sub>-saturated 0.1 M KOH at different Rotation speeds. (d)-(f) The corresponding K-L plots at different potentials and the calculated electron transfer number per oxygen molecule (n).

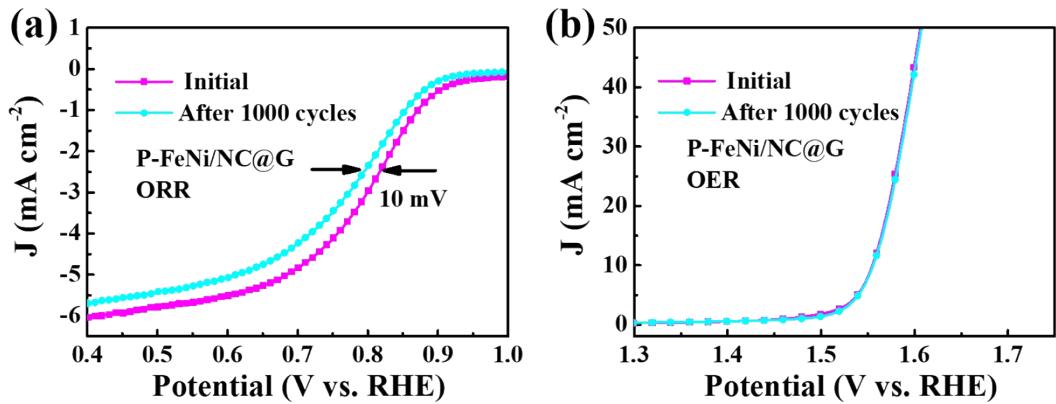


**Fig. S7.** CV curves in the range of 0-0.2 V at scan rates from 10 to 50 mV s<sup>-1</sup> and corresponding current density-scan rates curves. (a) and (d) for P-FeNi/NC@G, (b) and (e) for FeNi-MOFs, (c) and (f) for FeNi-MOFs-pyrolysis.



**Fig. S8.** Nyquist plots of FeNi-MOFs, FeNi-MOFs-pyrolysis, and P-FeNi/C@G

catalysts.



**Fig. S9.** The (a) ORR and (b) OER LSV curves of P-FeNi/NC@G catalyst recorded before and after 1000 CV cycles.

**Table S1.** The  $\Delta E$  value of different catalysts.<sup>#</sup>

Catalysts	$E_{1/2}$ (V)	$E_{j=10}$ (V)	$\Delta E$ (V)
FeNi-MOFs	0.64	1.65	1.01
FeNi-MOFs-pyrolysis	0.71	1.60	0.89
Pt/C	0.84	1.86	1.02
RuO <sub>2</sub>	0.60	1.56	0.96

<sup>#</sup> $E_{1/2}$  is the half over-potential of ORR,  $E_{j=10}$  is the potential of OER at a current density of 10 mA cm<sup>-2</sup>,  $\Delta E$  is the potential difference between  $E_{j=10}$  and  $E_{1/2}$ .

**Table S2.** List of properties of recently reported all-solid-state ZABs based on other cathodic catalysts.

Catalysts	$E_{1/2}$ (V)	$E_{j=10}$ (V)	$\Delta E$ (V)	Power density (mW cm <sup>-2</sup> )	Reference
P-FeNi/NC@G	0.81	1.54	0.73	159	<b>This work</b>
Co <sub>2</sub> P@CNF	0.803	1.691	0.88	121	S1
CoNi-MOF/RGO	0.718	1.548	0.83	97	S2
Hybrid nanosheets	0.79	1.72	0.93	—	S3
FeN <sub>x</sub> -PNC	0.86	1.635	0.775	118	S4
Fe-CoN <sub>4</sub> @NC	0.83	1.62	0.79	105	S5
FeNi@NCNTs/CC	0.77	1.48 (1M KOH)	0.71	—	S6
Co <sub>3</sub> O <sub>4-x</sub> HoNPs@HPNCS-60	0.83	1.57	0.74	94.1	S7
NiCo/PFC	0.79	1.63	0.84	—	S8
N-GCNT/FeCo <sub>3</sub>	0.92	1.73	0.81	97.6	S9
EA-Ni-900	0.78	1.55	0.77	65	S10
CC-AC	0.72	1.59	0.87	52.3	S11
FeNi-LDH@3DG/CNTS	0.71	1.61	0.9	—	S12
Fe-Ni <sub>2</sub> P@N, P-CNS	0.75	1.62	0.87	—	S13
N-CoSe <sub>2</sub> /3D Ti <sub>3</sub> C <sub>2</sub> TX	0.79	1.54	0.75	142	S14
FeNi-NC	0.83	1.61	0.78	80.8	S15
NiFe/N-CNT	0.75	1.52	0.77	105.4	S16

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