

Supporting Information

Biradical Oxo-molybdenum Complex Containing a Semiquinone and *O*-aminophenol Benzoxazole-Based Ligands: Cyclohexene and Sulfide Catalytic Oxidation

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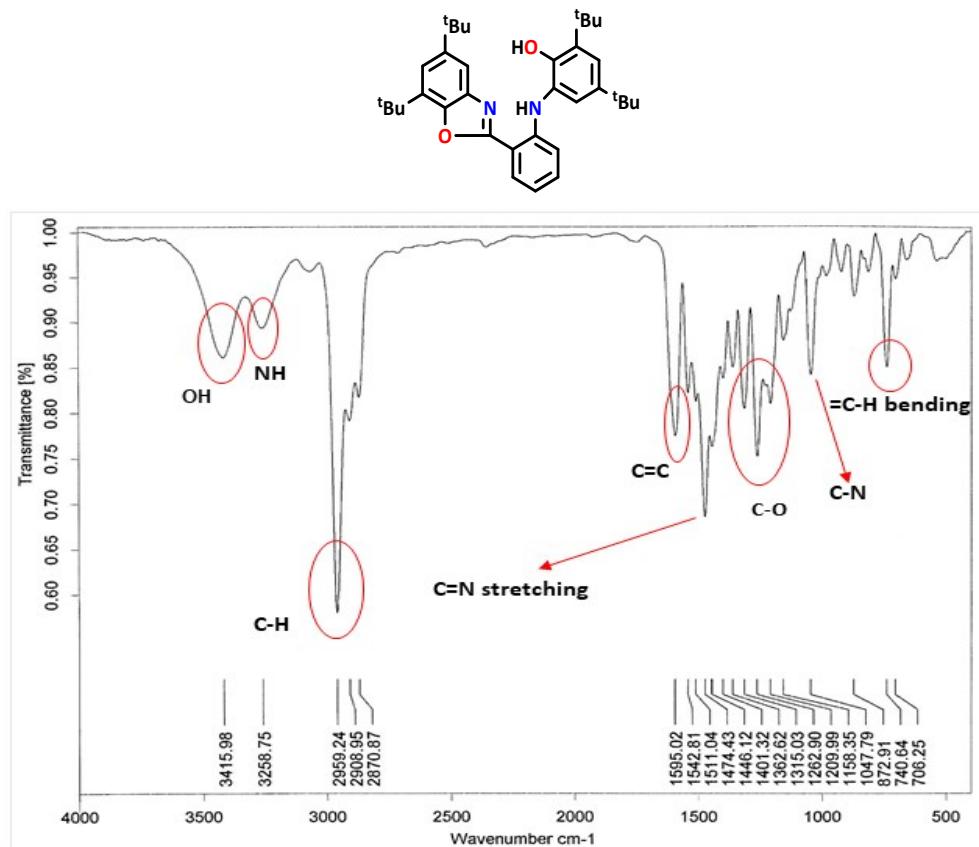


Figure S1 IR spectrum of $\text{H}_2\text{L}^{\text{BAP}}$

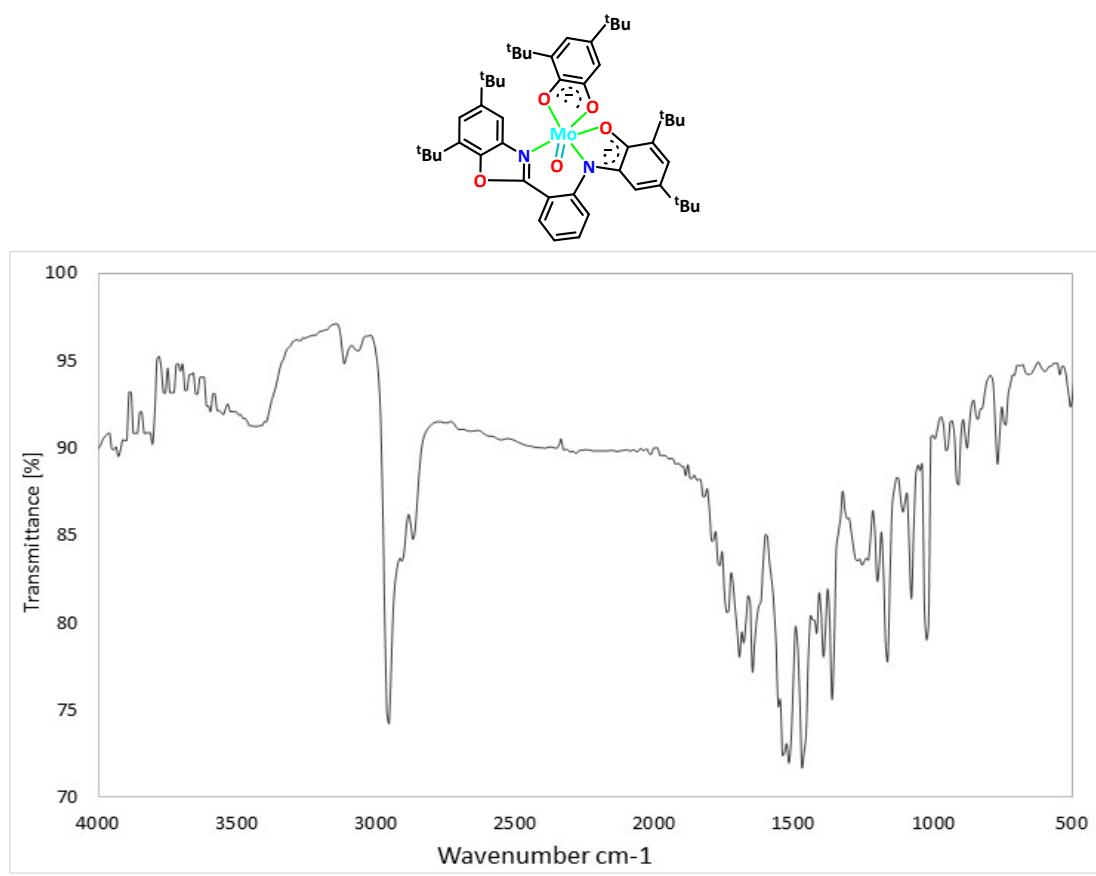


Figure S2 IR spectrum of $\text{MoOL}^{\text{BIS}}\text{L}^{\text{SQ}}$

Table S1 Selected bond lengths [\AA] and angles [$^\circ$] for MoOL^{BISL_{SQ}}

Mo1-O4	1.684(4)	O4-Mo1-O1	108.18(18)
Mo1-O1	1.956(4)	O4-Mo1-N3	92.28(19)
Mo1-N3	1.986(4)	O1-Mo1-N3	86.50(16)
Mo1-N1	2.039(4)	O4-Mo1-N1	108.4(2)
Mo1-O3	2.090(4)	O1-Mo1-N1	78.77(16)
Mo1-N2	2.150(4)	N3-Mo1-N1	157.50(18)
O1-C1	1.340(7)	O4-Mo1-O3	159.76(18)
C1-C2	1.404(8)	O1-Mo1-O3	87.33(16)
C1-C6	1.418(8)	N3-Mo1-O3	75.35(15)
C2-C3	1.383(9)	N1-Mo1-O3	86.96(17)
C2-C20	1.549(8)	O4-Mo1-N2	89.04(18)
C3-C4	1.419(9)	O1-Mo1-N2	158.34(17)
C4-C5	1.386(9)	N3-Mo1-N2	106.35(15)
C4-C24	1.529(9)	N1-Mo1-N2	83.43(16)
C5-C6	1.392(8)	O3-Mo1-N2	79.46(16)
C6-N1	1.398(7)	C1-O1-Mo1	114.8(3)
N1-C7	1.407(7)	O1-C1-C2	123.5(5)
C7-C12	1.409(7)	O1-C1-C6	115.7(5)
C7-C8	1.424(8)	C2-C1-C6	120.8(5)
C8-C9	1.370(8)	C3-C2-C1	117.2(5)
C9-C10	1.368(9)	C3-C2-C20	122.2(6)
C10-C11	1.370(8)	C1-C2-C20	120.5(6)
C11-C12	1.411(8)	C2-C3-C4	123.1(6)
C12-C13	1.454(7)	C5-C4-C3	118.6(6)
C13-N2	1.306(6)	C5-C4-C24	122.5(6)
C13-O2	1.350(6)	C3-C4-C24	118.9(6)
N2-C14	1.404(7)	C4-C5-C6	120.1(5)
C14-C15	1.374(7)	C5-C6-N1	127.3(5)
C14-C19	1.393(7)	C5-C6-C1	120.2(5)
C15-C16	1.377(8)	N1-C6-C1	112.5(5)
C16-C17	1.423(7)	C6-N1-C7	122.5(5)
C16-C28	1.523(7)	C6-N1-Mo1	111.0(3)
C17-C18	1.388(8)	C7-N1-Mo1	126.2(4)
C18-C19	1.382(7)	N1-C7-C12	121.8(5)
C18-C32	1.535(7)	N1-C7-C8	121.1(5)
C19-O2	1.395(6)	C12-C7-C8	117.1(5)
C20-C23	1.508(10)	C9-C8-C7	121.1(6)
C20-C21	1.515(9)	C10-C9-C8	121.5(6)
C20-C22	1.532(9)	C9-C10-C11	119.3(6)
C24-C26B	1.514(10)	C10-C11-C12	121.4(6)
C24-C25B	1.516(10)	C7-C12-C11	119.5(5)
C24-C25	1.517(9)	C7-C12-C13	121.7(5)
C24-C27	1.518(9)	C11-C12-C13	118.3(5)
C24-C26	1.519(9)	N2-C13-O2	113.9(4)
C24-C27B	1.525(10)	N2-C13-C12	127.9(5)
C28-C30	1.534(9)	O2-C13-C12	118.1(4)
C28-C31	1.537(9)	C13-N2-C14	106.0(4)
C28-C29	1.546(8)	C13-N2-Mo1	123.4(3)
C32-C33	1.512(8)	C14-N2-Mo1	130.0(3)
C32-C34	1.532(8)	C15-C14-C19	121.5(5)
C32-C35	1.551(9)	C15-C14-N2	131.0(5)
O3-C36	1.316(6)	C19-C14-N2	107.5(5)
N3-C41	1.312(7)	C14-C15-C16	117.5(5)
C36-C37	1.393(8)	C15-C16-C17	119.4(5)
C36-C41	1.427(7)	C15-C16-C28	122.6(5)
C37-C38	1.387(8)	C17-C16-C28	117.9(5)
C37-C42	1.543(8)	C18-C17-C16	124.3(5)
C38-C39	1.404(8)	C19-C18-C17	113.3(5)
C39-C40	1.367(8)	C19-C18-C32	122.1(5)
C39-C46	1.550(8)	C17-C18-C32	124.5(5)
C40-C41	1.403(7)	C18-C19-C14	123.8(5)
C42-C44	1.514(10)	C18-C19-O2	129.0(5)
C42-C45	1.525(10)	C14-C19-O2	107.2(5)
C42-C43	1.528(10)	C13-O2-C19	105.4(4)
C46-C48	1.510(8)	C23-C20-C21	111.4(7)
C46-C47B	1.514(10)	C23-C20-C22	107.9(7)
C46-C49B	1.515(10)	C21-C20-C22	107.0(6)
C46-C48B	1.518(10)	C23-C20-C2	109.7(6)
C46-C47	1.522(8)	C21-C20-C2	109.3(5)
C46-C49	1.525(9)	C22-C20-C2	111.5(6)

C40-C41-C36	120.6(6)	C26B-C24-C25B	110(2)
C44-C42-C45	108.3(7)	C25-C24-C27	109.6(12)
C44-C42-C43	108.0(6)	C25-C24-C26	105.4(12)
C45-C42-C43	109.6(6)	C27-C24-C26	110.8(14)
C44-C42-C37	112.8(6)	C26B-C24-C27B	104(2)
C45-C42-C37	108.6(6)	C25B-C24-C27B	114(3)
C43-C42-C37	109.5(6)	C26B-C24-C4	117.2(15)
C47B-C46-C49B	105(2)	C25B-C24-C4	108.1(16)
C47B-C46-C48B	118(2)	C25-C24-C4	109.4(10)
C49B-C46-C48B	105(2)	C27-C24-C4	114.6(9)
C48-C46-C47	109.0(9)	C26-C24-C4	106.5(9)
C48-C46-C49	107.6(11)	C27B-C24-C4	102.9(12)
C47-C46-C49	109.3(11)	C16-C28-C30	110.8(5)
C48-C46-C39	112.7(7)	C16-C28-C31	108.8(5)
C47B-C46-C39	112.7(14)	C30-C28-C31	109.7(6)
C49B-C46-C39	109.2(11)	C16-C28-C29	111.7(5)
C48B-C46-C39	106.7(12)	C30-C28-C29	108.6(6)
C47-C46-C39	109.1(7)	C31-C28-C29	107.2(5)
C49-C46-C39	109.1(9)	C33-C32-C34	108.7(5)
C40-C41-C36	120.6(6)	C33-C32-C18	110.6(5)
C44-C42-C45	108.3(7)	C34-C32-C18	111.1(5)
C44-C42-C43	108.0(6)	C33-C32-C35	110.5(5)
C45-C42-C43	109.6(6)	C34-C32-C35	108.7(5)
C44-C42-C37	112.8(6)	C18-C32-C35	107.3(5)
C45-C42-C37	108.6(6)	C36-O3-Mo1	115.3(3)
C43-C42-C37	109.5(6)	C41-N3-Mo1	118.3(3)
C47B-C46-C49B	105(2)	O3-C36-C37	126.2(5)
C47B-C46-C48B	118(2)	O3-C36-C41	113.6(5)
C49B-C46-C48B	105(2)	C37-C36-C41	120.2(5)
C48-C46-C47	109.0(9)	C38-C37-C36	116.7(5)
C48-C46-C49	107.6(11)	C38-C37-C42	122.3(6)
C47-C46-C49	109.3(11)	C36-C37-C42	121.0(5)
C48-C46-C39	112.7(7)	C37-C38-C39	124.0(6)
C47B-C46-C39	112.7(14)	C40-C39-C38	118.9(5)
C49B-C46-C39	109.2(11)	C40-C39-C46	121.9(5)
N3-C41-C40	125.3(5)	C38-C39-C46	119.2(5)
N3-C41-C36	114.2(5)	C39-C40-C41	119.4(5)

Symmetry transformations used to generate equivalent atoms

Crystal information of complex MoOL^{BIS}L^{SQ}

Symmetry transformations used to generate equivalent atoms:

The O4-Mo1-O3 angle of 75.54(15)° is the smallest, since it involves two donor groups of the L^{SQ} ligand. The valence angles between coordination bonds to Mo1 involving O1, N1 and N2 of L^{BIS} are much smaller than those expected for the octahedral sphere (Table 2), indicating the strain caused by its three-dentate coordination. The hydrophobic interactions between L^{BIS} and L^{SQ}, in particular those formed by *tBu* substituents at C2 and C16 (Figure 1) result in positioning of the latter in such a way, that the angles between bonds formed by L^{SQ} and L^{BIS} ligands range from O3-Mo1-N2 79.43(15) to O4-Mo1-N2 106.36(15)°, while the trans angle O4-Mo1-N1 is 157.65(18)°.

Conformation of the L^{BIS} ligand reflects the strain resulting from its three-dentate coordination. The C1-C6-N1-C7 and C6-N1-C7-C8 torsion angles are 153.6(5) and -24.0(9)°, while C7-C12-C13-N2 angle is 14.7(9)°. In the conformation of L^{BIS}, the dihedral angles between the phenolic and the central phenyl and benzoxazole rings are 43.6(3)° and 53.1(3)°, respectively. The central phenyl ring forms the dihedral angle of 15.9(3)° to the benzoxazole ring plane. The dihedral angles between the semiquinone ring and phenolate, central phenyl and benzoxazole rings are 55.9(3), 80.8(3) and 83.5(3)°, respectively.

The Conformation of the chelate ring Mo1-O1-C1-C6-N1 is an envelope on Mo1, Mo1-O3-C36-C41-O4 is twisted on O4-Mo1, Mo1-N1-C7-C12 ligands. -C13-N2 is an envelope on Mo1.

Some intra-molecular C-H···X interactions are found involving C-H groups of L^{BIS} and L^{SQ}, and O1, O2 of L^{BIS} and O3 of L^{SQ}. Also, analysis of the crystal packing also revealed the inter-molecular interactions C12···C48[1+x,y,z] of 3.451(15) and C13···C48[1+x,y,z] of 3.355(15) Å.

Table S2 Torsion angles [°] for MoOL^{BIS}L^{SQ}

Mo(1)-O(1)-C(1)-C(2)	-161.4(5)
Mo(1)-O(1)-C(1)-C(6)	17.6(6)
O(1)-C(1)-C(2)-C(3)	177.1(6)
C(6)-C(1)-C(2)-C(3)	-1.8(9)
O(1)-C(1)-C(2)-C(20)	-1.8(9)
C(6)-C(1)-C(2)-C(20)	179.3(6)
C(1)-C(2)-C(3)-C(4)	-0.6(10)
C(20)-C(2)-C(3)-C(4)	178.3(7)
C(2)-C(3)-C(4)-C(5)	1.2(11)
C(2)-C(3)-C(4)-C(24)	-178.5(7)
C(3)-C(4)-C(5)-C(6)	0.7(10)
C(24)-C(4)-C(5)-C(6)	-179.7(7)
C(4)-C(5)-C(6)-N(1)	178.9(6)
C(4)-C(5)-C(6)-C(1)	-3.0(9)
O(1)-C(1)-C(6)-C(5)	-175.4(5)
C(2)-C(1)-C(6)-C(5)	3.6(9)
O(1)-C(1)-C(6)-N(1)	3.0(7)
C(2)-C(1)-C(6)-N(1)	-178.0(5)
C(5)-C(6)-N(1)-C(7)	-27.8(9)
C(1)-C(6)-N(1)-C(7)	154.0(5)
C(5)-C(6)-N(1)-Mo(1)	157.4(5)
C(1)-C(6)-N(1)-Mo(1)	-20.8(6)
C(6)-N(1)-C(7)-C(12)	156.5(6)
Mo(1)-N(1)-C(7)-C(12)	-29.6(8)
C(6)-N(1)-C(7)-C(8)	-24.5(9)
Mo(1)-N(1)-C(7)-C(8)	149.4(5)
N(1)-C(7)-C(8)-C(9)	178.8(6)
C(12)-C(7)-C(8)-C(9)	-2.1(10)
C(7)-C(8)-C(9)-C(10)	0.5(11)
C(8)-C(9)-C(10)-C(11)	1.1(11)
C(9)-C(10)-C(11)-C(12)	-1.1(11)
N(1)-C(7)-C(12)-C(11)	-178.8(6)
C(8)-C(7)-C(12)-C(11)	2.1(9)
N(1)-C(7)-C(12)-C(13)	-7.2(9)
C(8)-C(7)-C(12)-C(13)	173.7(6)
C(10)-C(11)-C(12)-C(7)	-0.6(10)
C(10)-C(11)-C(12)-C(13)	-172.5(6)
C(7)-C(12)-C(13)-N(2)	15.1(9)
C(11)-C(12)-C(13)-N(2)	-173.2(6)
C(7)-C(12)-C(13)-O(2)	-160.9(5)
C(11)-C(12)-C(13)-O(2)	10.9(8)
O(2)-C(13)-N(2)-C(14)	1.1(6)
C(12)-C(13)-N(2)-C(14)	-175.0(6)
O(2)-C(13)-N(2)-Mo(1)	-171.0(3)
C(12)-C(13)-N(2)-Mo(1)	12.9(8)
C(13)-N(2)-C(14)-C(15)	179.3(6)
Mo(1)-N(2)-C(14)-C(15)	-9.4(9)
C(13)-N(2)-C(14)-C(19)	-1.2(6)
Mo(1)-N(2)-C(14)-C(19)	170.1(4)
C(19)-C(14)-C(15)-C(16)	0.0(9)
N(2)-C(14)-C(15)-C(16)	179.5(6)
C(14)-C(15)-C(16)-C(17)	-1.8(8)
C(14)-C(15)-C(16)-C(28)	-179.5(6)
C(15)-C(16)-C(17)-C(18)	2.4(9)
C(28)-C(16)-C(17)-C(18)	-179.8(5)
C(16)-C(17)-C(18)-C(19)	-0.9(9)
C(16)-C(17)-C(18)-C(32)	-179.8(5)
C(17)-C(18)-C(19)-C(14)	-1.0(8)
C(32)-C(18)-C(19)-C(14)	177.9(5)
C(17)-C(18)-C(19)-O(2)	-179.8(6)
C(32)-C(18)-C(19)-O(2)	-0.8(9)
C(15)-C(14)-C(19)-C(18)	1.6(9)
N(2)-C(14)-C(19)-C(18)	-178.0(5)
C(15)-C(14)-C(19)-O(2)	-179.5(5)
N(2)-C(14)-C(19)-O(2)	0.9(6)
N(2)-C(13)-O(2)-C(19)	-0.5(6)
C(12)-C(13)-O(2)-C(19)	176.0(5)
C(18)-C(19)-O(2)-C(13)	178.6(6)
C(14)-C(19)-O(2)-C(13)	-0.3(6)
C(3)-C(2)-C(20)-C(23)	123.7(7)

C(1)-C(2)-C(20)-C(23)	-57.4(8)
C(3)-C(2)-C(20)-C(21)	-113.9(7)
C(1)-C(2)-C(20)-C(21)	65.0(8)
C(3)-C(2)-C(20)-C(22)	4.2(10)
C(1)-C(2)-C(20)-C(22)	-176.9(6)
C(5)-C(4)-C(24)-C(26B)	152.5(19)
C(3)-C(4)-C(24)-C(26B)	-28(2)
C(5)-C(4)-C(24)-C(25B)	-82(2)
C(3)-C(4)-C(24)-C(25B)	98(2)
C(5)-C(4)-C(24)-C(25)	-123.3(11)
C(3)-C(4)-C(24)-C(25)	56.3(12)
C(5)-C(4)-C(24)-C(27)	0.3(15)
C(3)-C(4)-C(24)-C(27)	179.9(12)
C(5)-C(4)-C(24)-C(26)	123.2(11)
C(3)-C(4)-C(24)-C(26)	-57.1(12)
C(5)-C(4)-C(24)-C(27B)	39(2)
C(3)-C(4)-C(24)-C(27B)	-142(2)
C(15)-C(16)-C(28)-C(30)	-132.9(6)
C(17)-C(16)-C(28)-C(30)	49.3(8)
C(15)-C(16)-C(28)-C(31)	106.4(7)
C(17)-C(16)-C(28)-C(31)	-71.3(7)
C(15)-C(16)-C(28)-C(29)	-11.7(8)
C(17)-C(16)-C(28)-C(29)	170.5(5)
C(19)-C(18)-C(32)-C(33)	56.6(7)
C(17)-C(18)-C(32)-C(33)	-124.5(6)
C(19)-C(18)-C(32)-C(34)	177.4(5)
C(17)-C(18)-C(32)-C(34)	-3.7(8)
C(19)-C(18)-C(32)-C(35)	-63.9(7)
C(17)-C(18)-C(32)-C(35)	115.0(6)
Mo(1)-O(3)-C(36)-C(37)	-170.1(4)
Mo(1)-O(3)-C(36)-C(41)	9.0(6)
O(3)-C(36)-C(37)-C(38)	174.9(5)
O(3)-C(36)-C(37)-C(42)	-5.4(9)
C(41)-C(36)-C(37)-C(42)	175.6(5)
C(36)-C(37)-C(38)-C(39)	1.1(9)
C(42)-C(37)-C(38)-C(39)	-178.6(6)
C(37)-C(38)-C(39)-C(40)	2.6(10)
C(37)-C(38)-C(39)-C(46)	-176.0(6)
C(38)-C(39)-C(40)-C(41)	-3.2(8)
C(46)-C(39)-C(40)-C(41)	175.4(5)
Mo(1)-N(3)-C(41)-C(40)	162.9(4)
Mo(1)-N(3)-C(41)-C(36)	-17.4(6)
C(39)-C(40)-C(41)-N(3)	179.9(5)
C(39)-C(40)-C(41)-C(36)	0.2(8)
O(3)-C(36)-C(41)-N(3)	4.7(7)
C(37)-C(36)-C(41)-N(3)	-176.2(5)
O(3)-C(36)-C(41)-C(40)	-175.5(5)
C(37)-C(36)-C(41)-C(40)	3.6(8)
C(38)-C(37)-C(42)-C(44)	1.0(9)
C(36)-C(37)-C(42)-C(44)	-178.6(6)
C(36)-C(37)-C(42)-C(45)	61.4(8)
C(38)-C(37)-C(42)-C(43)	121.4(7)
C(36)-C(37)-C(42)-C(43)	-58.3(8)
C(40)-C(39)-C(46)-C(48)	-3.2(11)
C(38)-C(39)-C(46)-C(48)	175.4(9)
C(40)-C(39)-C(46)-C(47B)	154(2)
C(38)-C(39)-C(46)-C(47B)	-27(2)
C(40)-C(39)-C(46)-C(49B)	37.7(16)
C(38)-C(39)-C(46)-C(49B)	-143.7(15)
C(40)-C(39)-C(46)-C(48B)	-75(2)
C(38)-C(39)-C(46)-C(48B)	103(2)
C(40)-C(39)-C(46)-C(47)	-124.4(9)
C(38)-C(39)-C(46)-C(47)	54.2(10)
C(40)-C(39)-C(46)-C(49)	116.3(10)
C(38)-C(39)-C(46)-C(49)	-65.1(10)

Symmetry transformations used to generate equivalent atoms

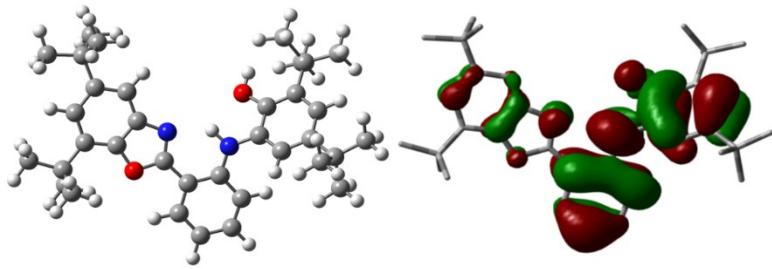


Figure S3 Optimized structure and HOMO (hydrogen atoms omitted for clarity) of compound H_2L^{BAP}

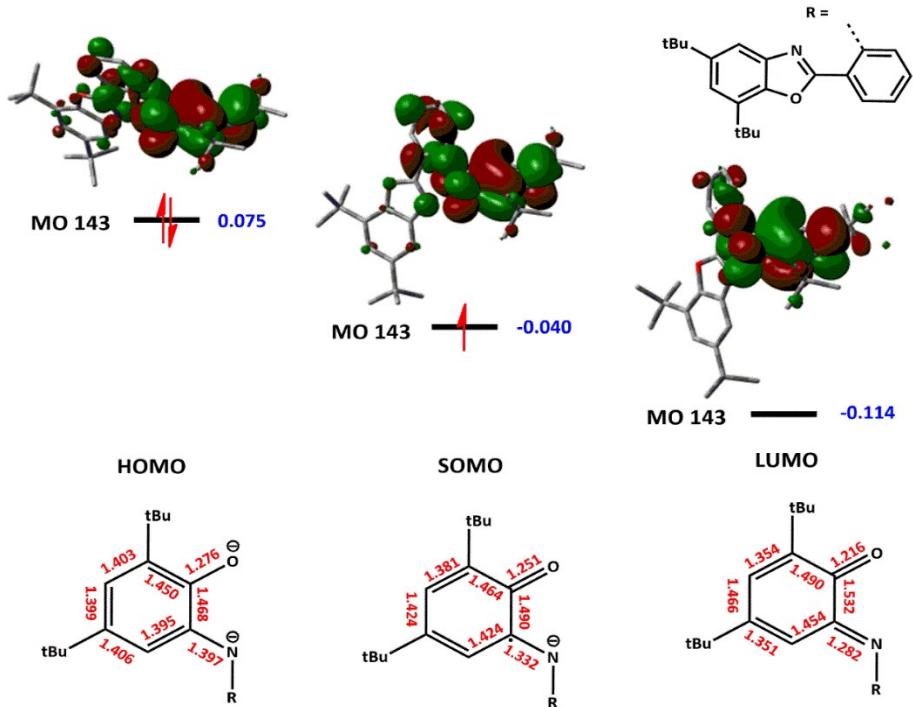


Figure S4 Selected bond lengths, MOs (hydrogen atoms omitted for clarity) and energies of optimized $(L^{BIS})^n^-$ ligands ($n = 2, 1, 0$)

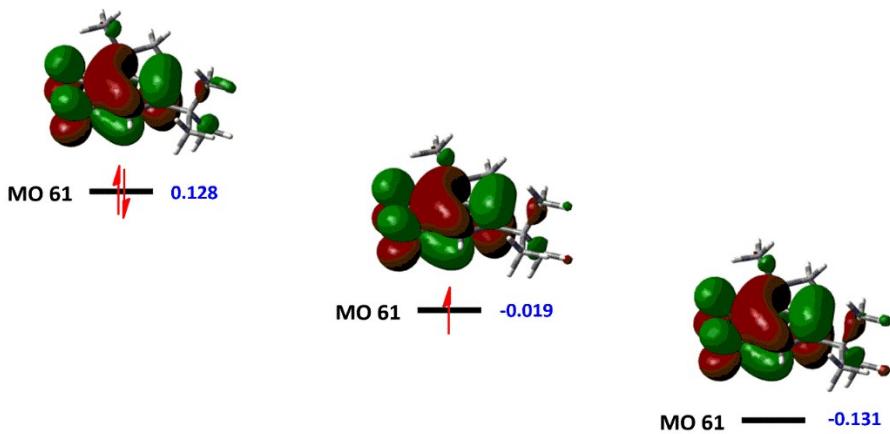
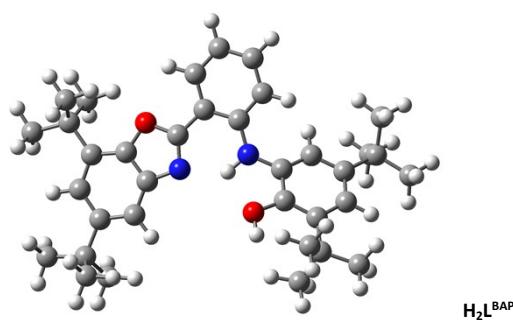


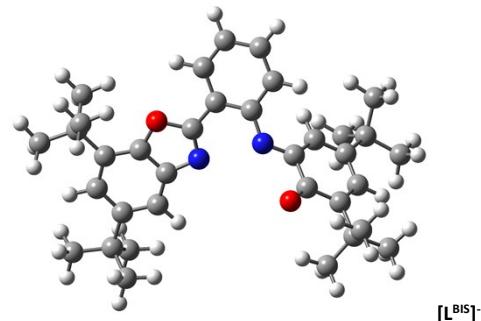
Figure S5 Selected bond lengths, MOs (hydrogen atoms omitted for clarity) and energies of optimized SQ^n^- ligands ($n = 2, 1, 0$)

Table S3 Coordinates of the optimized compounds



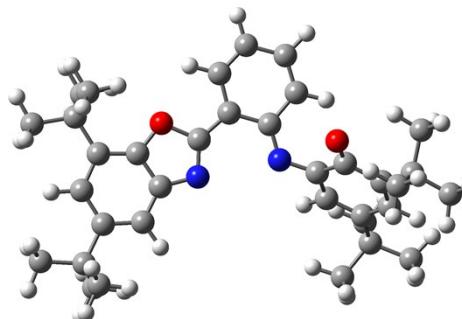
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H	6.14650768	1.26811463	0.42558201
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C	3.61298923	-0.93912059	0.57760356
H	3.33259443	-1.89009488	1.01057817
C	2.66182293	-0.25467303	-0.17474145
N	1.35974130	-0.73814662	-0.39097836
C	0.96270855	-2.02908168	-0.65154026
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C	-0.81185697	-3.70110564	-0.90043367
H	-1.86672826	-3.94155221	-0.87694779
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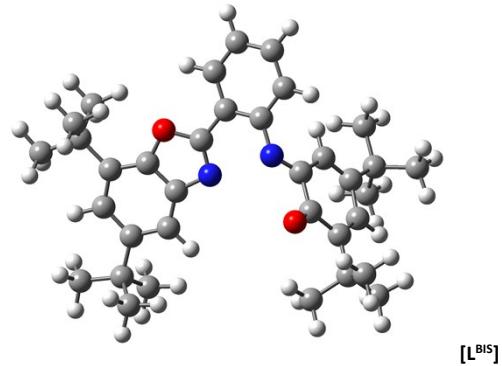
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[L^{BIS}]²⁻

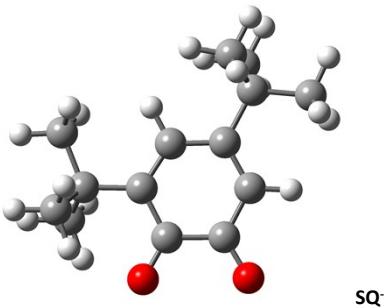
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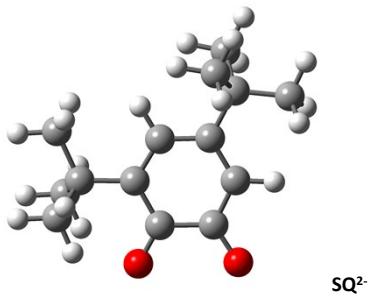
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H	6.51382855	0.12093301	2.01329985
C	3.98334262	-2.03675043	3.02501077
H	4.43309034	-2.23173353	4.00205491
H	2.89875409	-2.01690254	3.15919493
H	4.23362591	-2.87750582	2.37265474
C	4.14500821	0.41839481	3.47047605
H	4.54377066	1.39033514	3.17111957
H	3.06001038	0.51365417	3.56351915
H	4.55207867	0.18574789	4.45887486
C	-5.57015200	-1.95603110	-1.35068274
H	-4.62518032	-2.46878406	-1.53565729
H	-6.37413359	-2.69685397	-1.40327374
H	-5.72566947	-1.23040393	-2.15386828
C	-5.37641794	-2.31207026	1.14433546
H	-4.42468958	-2.83399603	1.03365215
H	-5.39423137	-1.84200981	2.13166064
H	-6.17747799	-3.05692690	1.10883279
C	-6.96409113	-0.61181994	0.23044086
H	-7.18587006	0.12632620	-0.54510664
H	-7.73562283	-1.38488393	0.17985522
H	-7.05003211	-0.12389837	1.20523088
C	-3.41088842	4.42142296	-0.60584460
H	-2.33742983	4.29769190	-0.76484393
H	-3.92083728	4.10894648	-1.52132022
H	-3.60676389	5.48765311	-0.45356760
C	-5.41028015	3.89485187	0.79386617
H	-5.99251535	3.61983191	-0.08981441
H	-5.82530365	3.36843646	1.65783843
H	-5.55503664	4.96570214	0.96104025
C	-3.17248717	4.07995694	1.88238627
H	-3.50837492	3.51964858	2.75947250
H	-2.09117887	3.95101729	1.79811570
H	-3.36900408	5.14196030	2.06105804

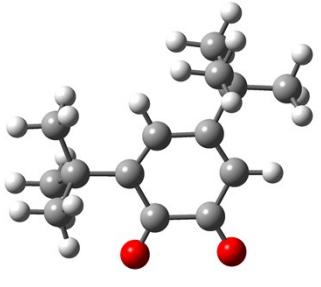


O	-2.39519574	1.87581672	-0.00027606
C	-1.26362746	1.33288691	-0.00010270
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C	0.19576034	-0.64756076	0.00003581
H	0.31395954	-1.72407182	0.00017762
C	1.38495756	0.14511707	-0.00006452
C	1.25580679	1.51523284	-0.00024154
H	2.11511075	2.17603046	-0.00029972
C	-0.02159571	2.18763308	-0.00040776
O	-0.09904972	3.43835345	-0.00035453
C	-2.33079811	-1.02202261	0.00013795
C	2.74989860	-0.56902523	0.00007290
C	3.92998214	0.42062512	0.00030896
H	3.91427496	1.06411597	-0.88324300
H	3.91408979	1.06391837	0.88400044
H	4.87705192	-0.13065783	0.00034895
C	2.88032554	-1.45808800	1.26021178

H	2.09118306	-2.21224584	1.30400652
H	3.84676485	-1.97763558	1.27119261
H	2.80756871	-0.84966574	2.16631378
C	2.88070050	-1.45791883	-1.26014754
H	3.84716228	-1.97742863	-1.27093122
H	2.09160269	-2.21210151	-1.30426173
H	2.80817239	-0.84937905	-2.16618901
C	-1.98393605	-2.52417319	0.00019893
H	-2.91042343	-3.10906961	0.00030786
H	-1.41047025	-2.81263995	-0.88644453
H	-1.41032719	-2.81253405	0.88678441
C	-3.18189583	-0.74218927	-1.26286806
H	-3.47485073	0.30658907	-1.29535488
H	-2.60936507	-0.97382027	-2.16745633
H	-4.08428614	-1.36762336	-1.26255364
C	-3.18174494	-0.74205349	1.26321528
H	-2.60910417	-0.97358200	2.16776010
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H	-4.08413214	-1.36749233	1.26307824

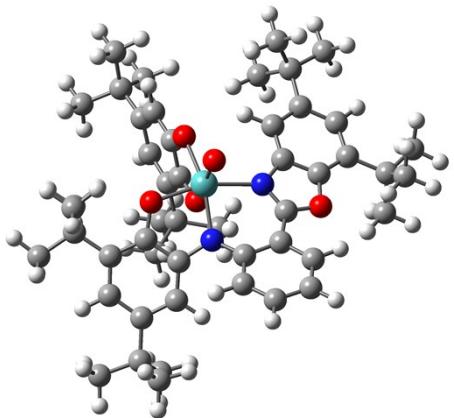


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C	-1.07683700	-0.11867400	-0.00224100
C	0.21932100	-0.66538600	0.00711700
H	0.33672900	-1.74315800	0.00921300
C	1.38095300	0.13015000	0.00084600
C	1.21831900	1.52007000	-0.00672900
H	2.07284600	2.19019200	-0.00889700
C	-0.05204900	2.18149800	0.00234100
O	-0.14492700	3.44954800	-0.00139300
C	-2.32195500	-1.03320100	-0.00072900
C	2.76587700	-0.55103600	-0.00245200
C	3.92326600	0.46621500	-0.04198300
H	3.87371700	1.09142900	-0.93866000
H	3.89714900	1.13048600	0.82662800
H	4.88798500	-0.05596600	-0.03826300
C	2.94808000	-1.40924900	1.27779900
H	2.19633300	-2.20024400	1.33621600
H	3.94404900	-1.88345000	1.31315500
H	2.83631800	-0.78443900	2.16911400
C	2.91247100	-1.47130600	-1.23855600
H	3.89301200	-1.96159600	-1.25744100
H	2.14397900	-2.24921100	-1.24339200
H	2.79907700	-0.88724200	-2.15809800
C	-1.97916400	-2.53724400	-0.01140000
H	-2.90833600	-3.12324100	-0.00729800
H	-1.40995600	-2.81875200	-0.90415900
H	-1.39494700	-2.83065700	0.86633100
C	-3.18735400	-0.75084400	-1.25526500
H	-3.44218900	0.30955500	-1.28482600
H	-2.62803900	-1.00729200	-2.16242600
H	-4.11004100	-1.35372600	-1.24048700
C	-3.16912900	-0.75988000	1.26745600
H	-2.59863100	-1.02536400	2.16518800
H	-3.41806600	0.30216400	1.30763600
H	-4.09542000	-1.35573900	1.26086500



SQ

O	-2.31261439	1.93502369	-0.00030962
C	-1.24716675	1.35756320	-0.00016229
C	-1.10003518	-0.12692049	-0.00000590
C	0.16037651	-0.61790034	0.00004679
H	0.28885706	-1.69132382	0.00013373
C	1.40512388	0.17326049	-0.00000071
C	1.34267480	1.52255337	-0.00006182
H	2.22045063	2.15478056	-0.00007434
C	0.05956383	2.22038299	-0.00008302
O	-0.03544990	3.43063152	-0.00005436
C	-2.35516351	-1.00949413	0.00005129
C	2.72578623	-0.60427959	0.00004970
C	3.94926645	0.32862913	0.00006371
H	3.97275492	0.96813094	-0.88600880
H	3.97272519	0.96814438	0.88612711
H	4.86386166	-0.26959919	0.00008452
C	2.79359777	-1.49389368	1.26588416
H	1.97842147	-2.21930955	1.31108270
H	3.73305808	-2.05366247	1.27327650
H	2.75457824	-0.88532852	2.17318754
C	2.79366819	-1.49394378	-1.26574636
H	3.73313735	-2.05369861	-1.27307345
H	1.97850583	-2.21937413	-1.31095388
H	2.75468155	-0.88541531	-2.17307581
C	-2.00170648	-2.50828005	0.00012722
H	-2.92470420	-3.09326531	0.00017824
H	-1.43225719	-2.79793651	-0.88810412
H	-1.43221936	-2.79783578	0.88836722
C	-3.19728547	-0.71692363	-1.26677991
H	-3.53434202	0.31822731	-1.29363489
H	-2.62009267	-0.91993720	-2.17387912
H	-4.07858747	-1.36518149	-1.27901834
C	-3.19726274	-0.71679451	1.26686795
H	-2.62005061	-0.91970791	2.17397747
H	-3.53433049	0.31835538	1.29362188
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[MoO(L^{BIS})(SQ)] (S = 1)

Mo	0.41281587	-0.38180539	-1.63721253
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O	2.12905552	0.64337890	-1.68871383
C	3.14824161	0.14998243	-0.96642917

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 C 5.28460976 0.24644103 0.09981191
 H 6.20951152 0.76944617 0.28437793
 C 5.07233816 -0.98191614 0.74257538
 C 3.88281154 -1.66533504 0.47603005
 H 3.67587973 -2.61177910 0.95761299
 C 2.94263763 -1.13342751 -0.40493746
 N 1.71192812 -1.72182323 -0.77719510
 C 1.50774314 -3.09353096 -0.72599690
 C 2.56914749 -3.99951002 -0.95196042
 H 3.54219918 -3.59838596 -1.19731974
 C 2.38542465 -5.36815700 -0.89593519
 H 3.22611088 -6.02386492 -1.09446932
 C 1.12738414 -5.90929620 -0.60411337
 H 0.98320398 -6.98141782 -0.55405915
 C 0.06137539 -5.05269147 -0.41405346
 H -0.92484762 -5.45364691 -0.22207116
 C 0.21679928 -3.65255674 -0.48932670
 C -0.97203471 -2.84155730 -0.35000411
 N -1.18967007 -1.60079295 -0.73145626
 C -2.53992814 -1.33160666 -0.46112577
 C -3.32846486 -0.21584610 -0.71155738
 H -2.89866538 0.65210295 -1.19030860
 C -4.67733674 -0.27904155 -0.34804332
 C -5.17400927 -1.45330015 0.25461119
 H -6.21550567 -1.47861581 0.53315170
 C -4.41054560 -2.60205505 0.51000161
 C -3.08335950 -2.47696789 0.11343702
 O -2.07585781 -3.42307921 0.18126615
 O 0.22122929 0.57486606 0.45589040
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 C -0.77710221 3.88627186 1.46573781
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 C 4.60800110 2.21001894 -1.42642569
 C 6.09033282 -1.59010315 1.72663849
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 C -2.94250413 6.02093524 1.13337425
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 H 0.62260245 4.03762005 3.93010386
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 H 2.30689826 2.59525230 2.67350151
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 H -0.28652527 0.46511520 4.37304090
 C 4.51211770 2.05613925 -2.96366487
 H 5.26165139 1.34724046 -3.32815467

H 3.52718711 1.70218372 -3.26661434
 H 4.69657050 3.02046655 -3.44836800
 C 3.56431863 3.24503438 -0.94306536
 H 2.55152443 2.93239959 -1.19354534
 H 3.62747431 3.38162997 0.14059277
 H 3.75337695 4.21390575 -1.41693431
 C 6.00695470 2.76425091 -1.09593604
 H 6.80385097 2.09403919 -1.43073533
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 H -3.11848415 -4.34422467 2.25120464
 H -4.27809283 -3.37341544 3.16825852
 H -4.57192957 -5.09994205 2.91332830
 C -6.45616808 -3.73593445 1.52034597
 H -7.08006223 -3.55433692 0.64109543
 H -6.80365115 -4.66293608 1.98378069
 H -6.62412534 -2.92696873 2.23641535
 C -5.56656296 1.25766129 -2.13640855
 H -4.56098088 1.48790441 -2.49434616
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 C -7.04588500 0.69755257 -0.20407354
 H -7.49479897 -0.13943704 -0.74608487
 H -7.13908756 0.50506588 0.86839922
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 C -5.05341723 2.15783880 0.16548648
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[MoO(L^{BIS})(SQ)] (S = 0)

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 C 3.12966800 0.06004800 -0.90395000
 C 4.35928600 0.70194200 -0.66335600
 C 5.21832100 0.07992400 0.25144500
 H 6.15460400 0.56754200 0.47193100
 C 4.92382900 -1.12304600 0.91376900
 C 3.72112400 -1.76060500 0.60565600
 H 3.44763200 -2.69009600 1.08816200
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 N 1.63274400 -1.75552300 -0.77527600
 C 1.43949000 -3.12922900 -0.82365400
 C 2.51504300 -4.01468200 -1.06201200
 H 3.49667500 -3.59720000 -1.23703300
 C 2.33042300 -5.38333700 -1.10379400
 H 3.17856500 -6.02613600 -1.31195300
 C 1.06161500 -5.94246500 -0.89827100
 H 0.92005700 -7.01573700 -0.92794200
 C -0.01330100 -5.10379600 -0.68425000
 H -1.00478300 -5.51515000 -0.54764400

C	0.14521400	-3.70259500	-0.66126200
C	-1.02426500	-2.88572700	-0.46200500
N	-1.19045700	-1.60821000	-0.72706500
C	-2.51963900	-1.30113200	-0.40394700
C	-3.25059900	-0.12804000	-0.52938600
H	-2.78068400	0.76155800	-0.92261300
C	-4.59606200	-0.15916200	-0.15054800
C	-5.14439500	-1.36322500	0.33821300
H	-6.18283800	-1.36531900	0.62914400
C	-4.43829800	-2.56909800	0.46579800
C	-3.10988400	-2.47072700	0.06639200
O	-2.14613200	-3.46554700	0.02927300
O	0.20037400	0.57828100	0.37594300
C	-0.19626200	1.81845800	0.39462000
C	-0.30315500	2.65347100	1.54435100
C	-0.70221000	3.96273800	1.31353700
H	-0.76432100	4.62855100	2.16315400
C	-1.02990700	4.50557100	0.03774800
C	-0.96011900	3.67374600	-1.06560800
H	-1.20008900	4.00293100	-2.06712400
C	-0.55487900	2.34036100	-0.88869400
O	-0.51460800	1.45527500	-1.86079800
C	-5.05688000	-3.87635600	0.98063400
C	-5.43966400	1.12490600	-0.28432400
C	4.71383700	2.03070100	-1.35563600
C	5.86657300	-1.74617000	1.96156500
C	0.04946800	2.11606000	2.94060000
C	-1.43396300	5.98642900	-0.06229800
C	-1.75969900	6.40223200	-1.50837000
H	-0.89896700	6.27810400	-2.17089100
H	-2.59338100	5.82634700	-1.91943300
H	-2.04458900	7.45768100	-1.53112600
C	-2.68668600	6.24920000	0.80629300
H	-2.51031700	6.01868400	1.85944600
H	-2.97794000	7.30239000	0.74130800
H	-3.53009900	5.64160800	0.46715600
C	-0.26956100	6.87332000	0.44010300
H	-0.53899600	7.93155100	0.36332700
H	-0.02183600	6.66854600	1.48410100
H	0.63145400	6.70719000	-0.15645400
C	-0.16131700	3.17210300	4.04079100
H	0.08832600	2.73516100	5.01159500
H	0.48232000	4.04517600	3.90045600
H	-1.19979500	3.51258500	4.09005000
C	1.53588000	1.68669100	2.96760600
H	1.74044500	0.91052900	2.23068700
H	2.18760600	2.54014200	2.75861800
H	1.79471600	1.29922200	3.95847500
C	-0.84697200	0.89836700	3.26925600
H	-1.90255600	1.18607600	3.28559900
H	-0.71610700	0.10492900	2.53410800
H	-0.58999900	0.50164100	4.25673600
C	4.64875100	1.84982800	-2.89130400
H	5.36611300	1.09280900	-3.22247100
H	3.65354300	1.54655700	-3.21511500
H	4.90019800	2.79235100	-3.38823700
C	3.71979100	3.13270600	-0.91912500
H	2.69526100	2.87369400	-1.18259800
H	3.76796400	3.29122600	0.16221900
H	3.97237100	4.07793300	-1.41063000
C	6.13409400	2.50805600	-0.99842300
H	6.89729800	1.78061300	-1.28968800
H	6.34451200	3.43822200	-1.53330700
H	6.24328600	2.71425800	0.07014200
C	7.16281200	-0.93500300	2.14414900
H	7.79682900	-1.42251900	2.88991700
H	7.73668300	-0.86888600	1.21565100
H	6.96265900	0.08042100	2.49623400
C	6.25829000	-3.17806300	1.52751600
H	6.92916300	-3.63002300	2.26507600
H	5.38478400	-3.82781900	1.43359600

H 6.77326000 -3.16597600 0.56264100

Table S5 Experimental and selected list of calculated absorption properties of the MoOL^{BISL^{SQ}} complex

Exp. λ_{max} /nm	Tr ^a	Composition	Energy/eV (nm)	Oscillator strength	Assignment
450-542	5	HOMO→LUMO+2 (75%) HOMO→LUMO+3 (13%)	2.32 (534)	0.03262	LMCT
	8	HOMO-3→LUMO (43%) HOMO-1→LUMO+1 (23%) HOMO-1→LUMO+2 (10%)	2.59 (479)	0.1213	LMCT
	9	HOMO→LUMO+2 (13%) HOMO→LUMO+3 (72%)	2.71 (457)	0.1095	LMCT
C	5.14364800	-1.80862700	3.32774800		
H	4.85808400	-0.80743200	3.66163900		
H	4.23554900	-2.41421600	3.27775300		
H	5.79817600	-2.24871500	4.08714500		
C	-4.98035600	-4.95202700	-0.12905800		
H	-3.94996000	-5.16128800	-0.42067900		
H	-5.42639600	-5.88656100	0.22411100		
H	-5.52707000	-4.63157500	-1.02011300		
C	-4.27962400	-4.35818300	2.22861000		
H	-3.22811300	-4.54554000	2.00568500		
H	-4.33031900	-3.61504900	3.02912100		
H	-4.71487400	-5.28926400	2.60391400		
C	-6.53389100	-3.69725800	1.37410300		
H	-7.15018500	-3.39202800	0.52419300		
H	-6.92714600	-4.64917600	1.74011700		
H	-6.65730400	-2.96121300	2.17304100		
C	-5.43000800	1.59128000	-1.75915900		
H	-4.41997400	1.80784300	-2.11289300		
H	-5.85724700	0.82649400	-2.41382200		
H	-6.02464900	2.50324700	-1.86937100		
C	-6.90508900	0.91969800	0.14103200		
H	-7.40884400	0.16629400	-0.47104600		
H	-6.98987500	0.62589800	1.19092700		
H	-7.45128200	1.85877400	0.01973800		
C	-4.83061600	2.23100800	0.60824100		
H	-4.83514600	1.93061400	1.65984900		
H	-3.80021500	2.46014300	0.32987600		
H	-5.41462800	3.15202300	0.51687600		

Table S4 Comparison of experimental and theoretical structural parameters of complex MoOL^{BISL^{SQ}}

Structural parameters (Å and °)	Experimental (X-ray)	Theoretical (DFT) S = 1	Theoretical (DFT) S = 0
Mo1-O1	1.955(4)	1.992	1.995
Mo1-O3	2.091(4)	2.305	2.169
Mo1-O4	1.993(4)	2.136	2.017
Mo1-O5	1.686(4)	1.700	1.705
Mo1-N1	2.037(4)	2.053	2.058
Mo1-N2	2.148(4)	2.207	2.196
O5-Mo1-O1	108.2(2)	106.9	108.1
O5-Mo1-O4	92.1(2)	96.0	91.3
O5-Mo1-N1	108.4(2)	107.5	108.4
O5-Mo1-O3	159.8(2)	164.2	159.7
O5-Mo1-N2	89.0(2)	94.5	91.1

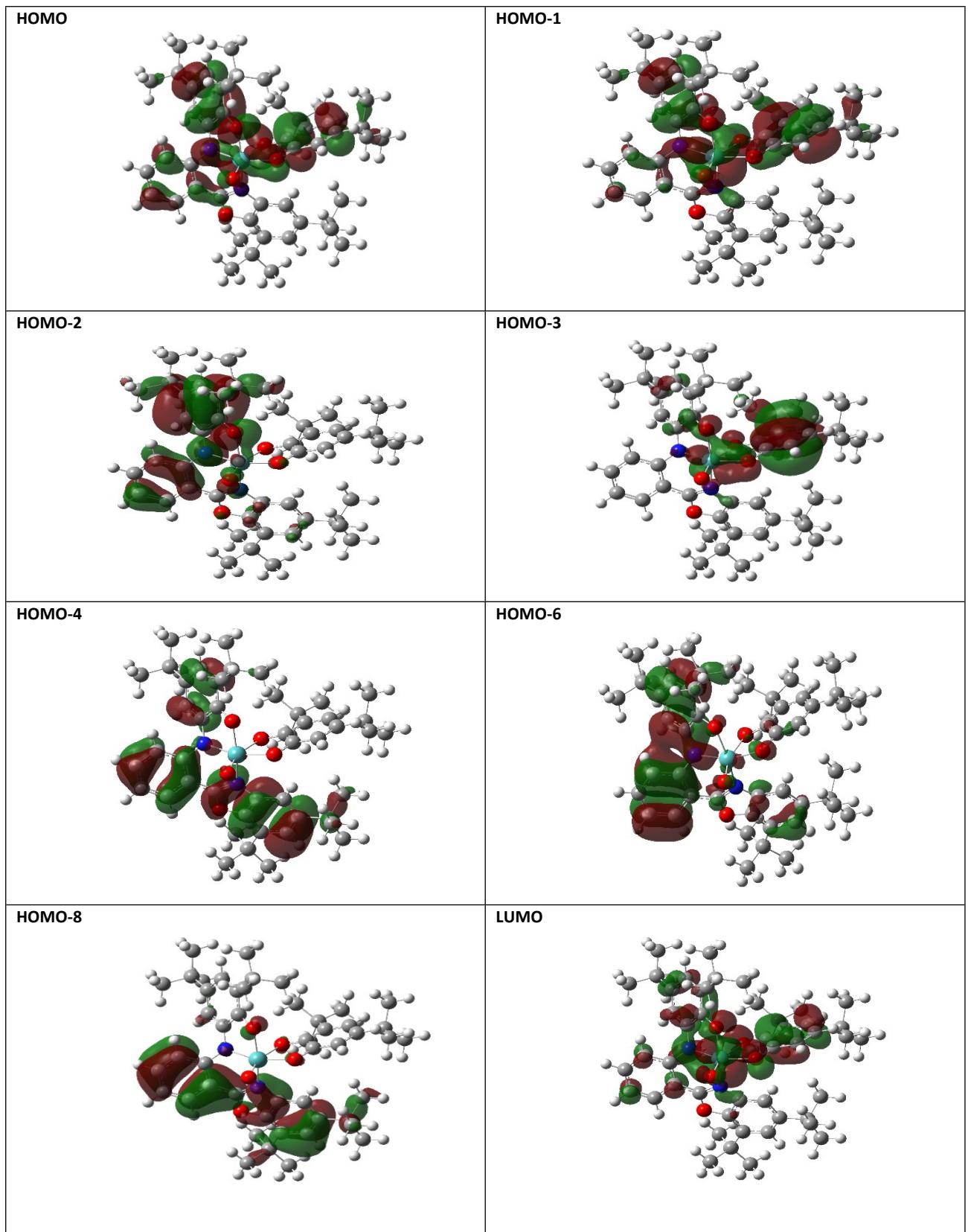
344-380	13	HOMO-6→LUMO (13%) HOMO-2→LUMO+1 (46%) HOMO-1→LUMO+2 (24%)	3.11 (398)	0.0665	LMCT
	15	HOMO-2→LUMO+2 (86%) HOMO-3→LUMO+1 (15%)	3.29 (377)	0.0461	LMCT
	17	HOMO-1→LUMO+2 (12%) HOMO-1→LUMO+3 (54%)	3.45 (359)	0.0504	LMCT
296	25	HOMO-4→LUMO+2 (70%)	3.95 (313)	0.1889	LLCT
	32	HOMO-4→LUMO+3 (81%)	4.24 (292)	0.2419	LLCT
246	45	HOMO-6→LUMO+3 (25%) HOMO-2→LUMO+4 (30%)	4.72 (263)	0.1471	LLCT
	47	HOMO-8→LUMO+1 (63%)	4.79 (257)	0.0681	LLCT

^aTr = transition number as obtained in the TD-DFT calculation output.

Table S6 Energies (eV) and composition (%) of some frontier molecular orbitals of MoOL^{BIS}L^{SQ} complex

Orbital	Energy	Mo	O	L ^{BIS}	L ^{SQ}
LUMO+4	-0.64	6	0	93	1
LUMO+3	-1.68	45	13	39	4
LUMO+2	-2	36	11	49	4
LUMO+1	-2.31	48	18	31	4
LUMO	-3.31	41	1	23	35
HOMO	-5.07	4	1	50	45
HOMO-1	-5.89	25	1	34	40
HOMO-2	-5.98	1	2	96	1
HOMO-3	-6.39	9	0	8	83
HOMO-4	-6.51	0	0	99	0
HOMO-6	-7.07	1	1	96	3
HOMO-8	-7.74	0	0	99	1

Table S7 Molecular orbitals corresponding to TD-DFT excitations for the MoOL^{BIS}L^{SQ} complex



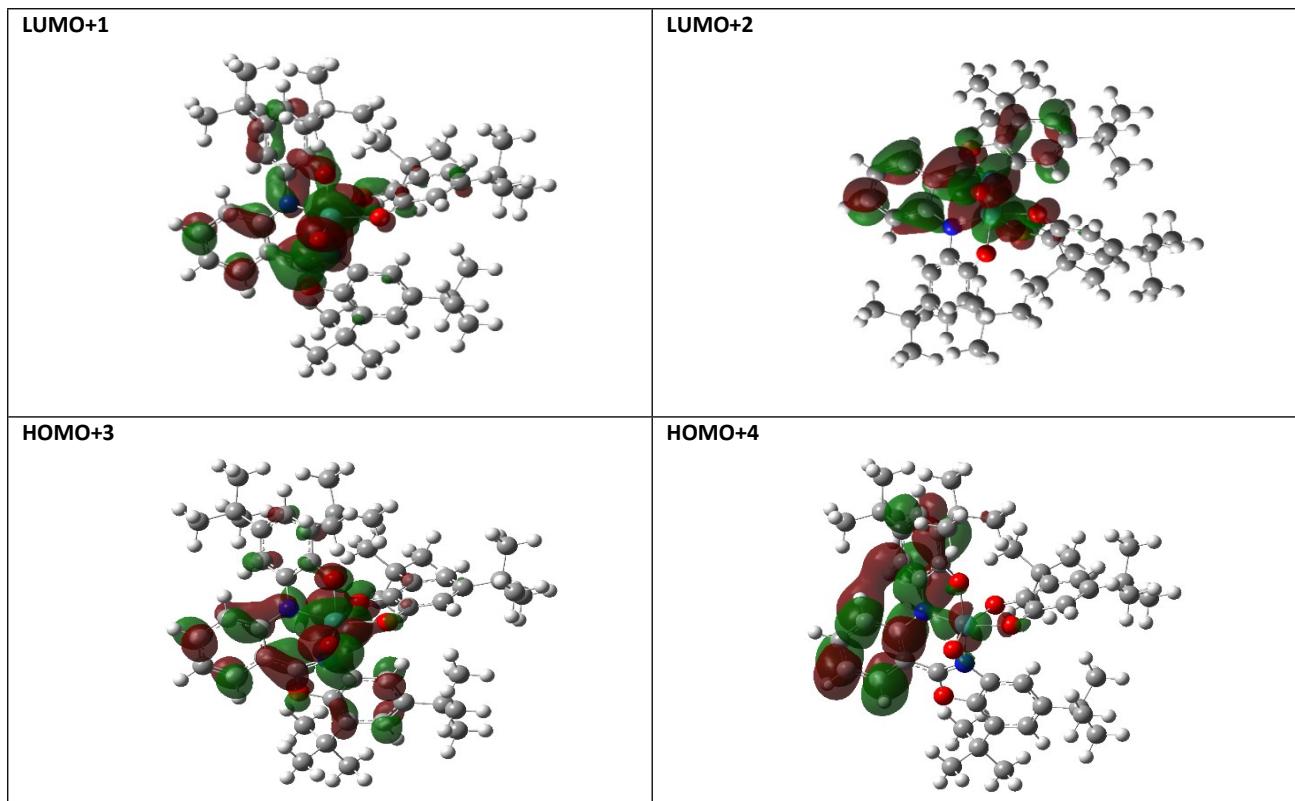


Table S8 The result of blank test for oxidation of phenyl sulfide with some catalytic systems

Entry	Catalytic system	Time (h)	Conversion (%)
1	No catalyst/H ₂ O ₂	8	-
1	H ₂ L ^{BIS} /H ₂ O ₂	8	31
2	MoO ₂ (acac) ₂ /H ₂ O ₂	8	35
3	MoO ₂ (acac) ₂ / H ₂ L ^{BIS}	8	20
4	MoO ₂ (acac) ₂ / H ₂ L ^{BIS} / H ₂ O ₂	8	39
5	MoOL ^{BIS} L ^{SQ} /Sulfide	2	-

Reactions condition: Catalyst (2 mol%), H₂O₂ (3 equiv.), acetone as solvent (2 mL), R T.

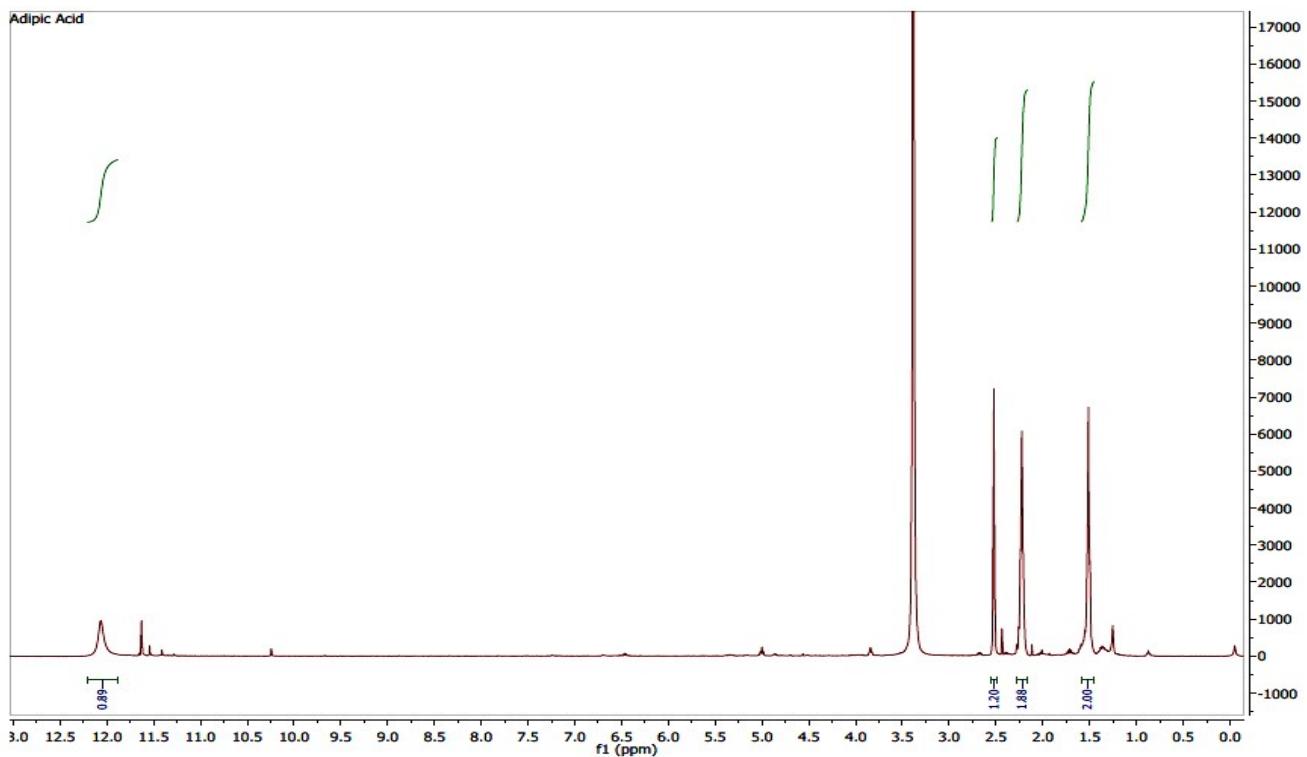
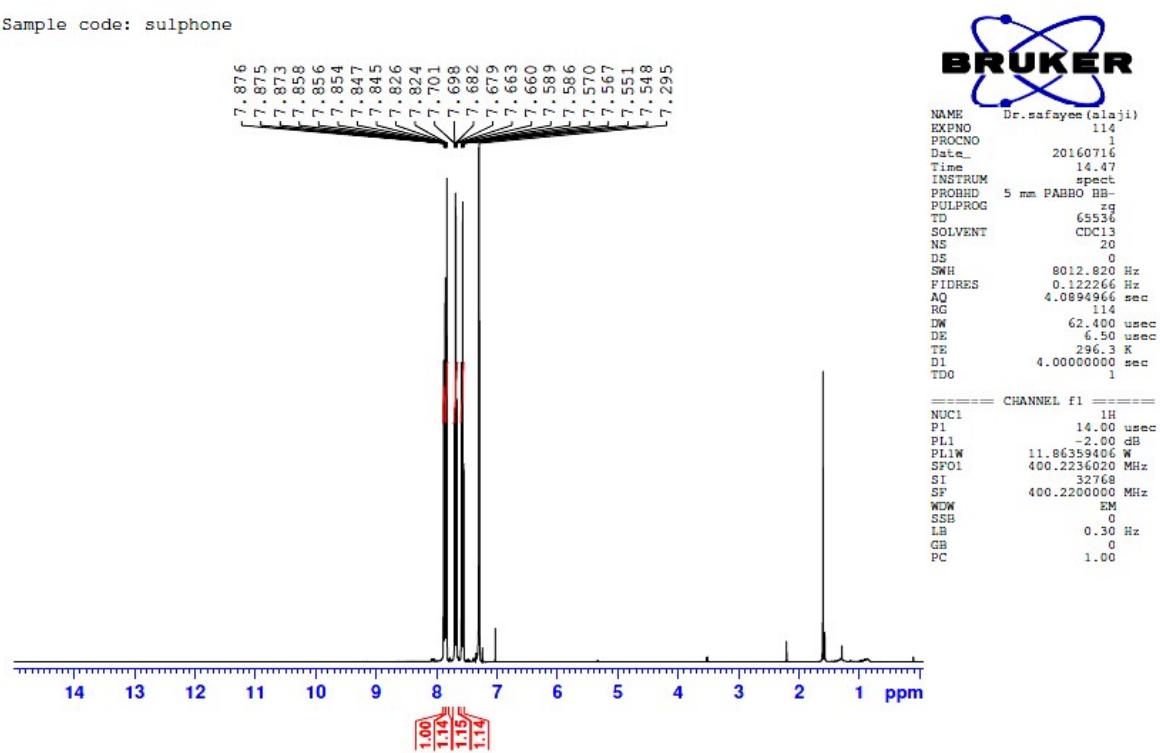


Figure S6 ^1H NMR spectrum of adipic acid

Sample code: sulphone



Sample code: sulphone

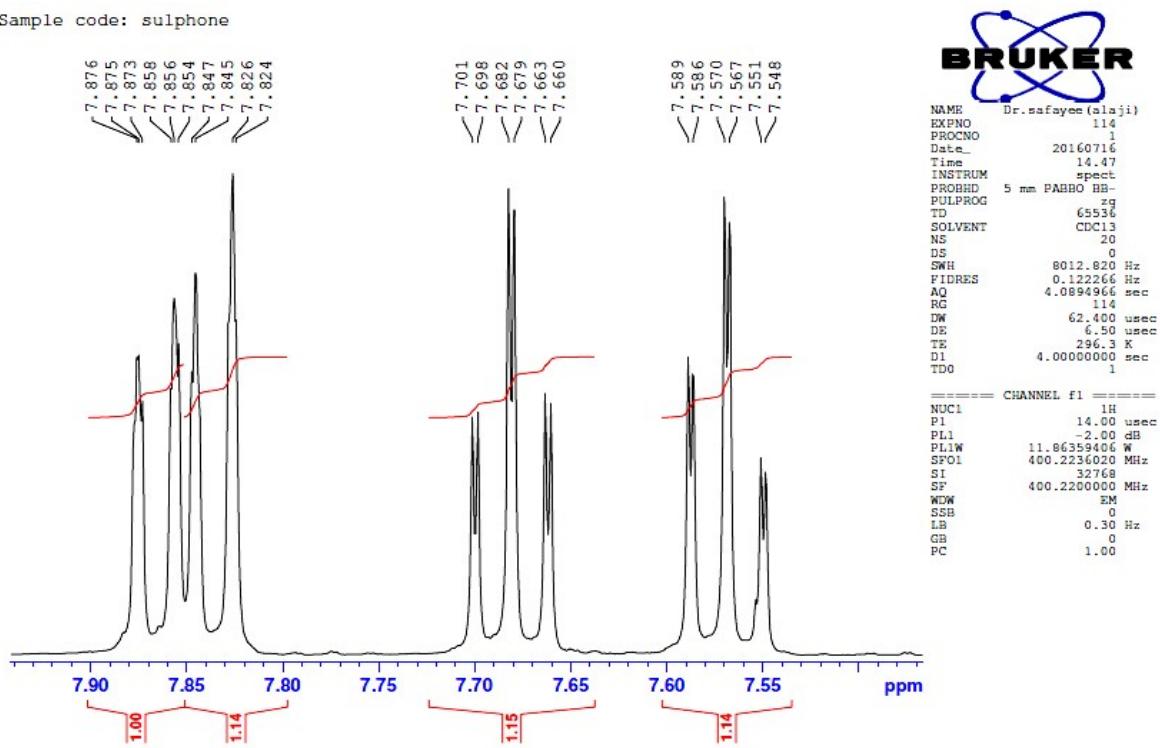
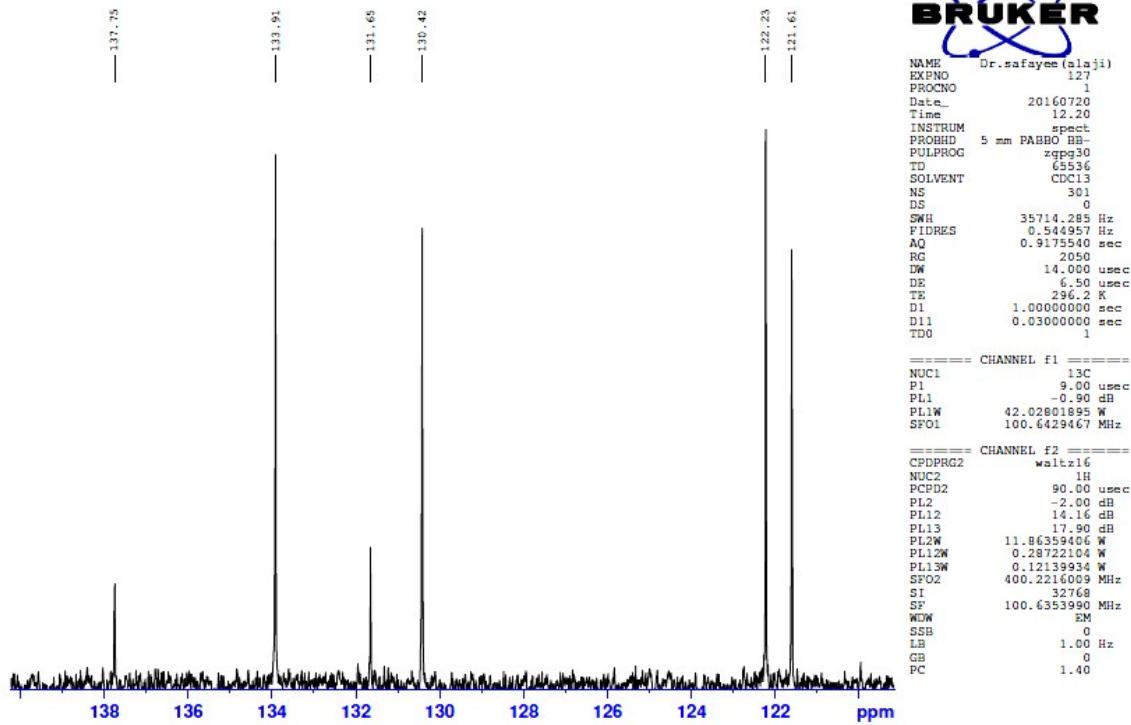


Figure S7 ^1H NMR spectrum of dibenzothiophene sulfone

Sample code:sulfone



Sample code:sulfone

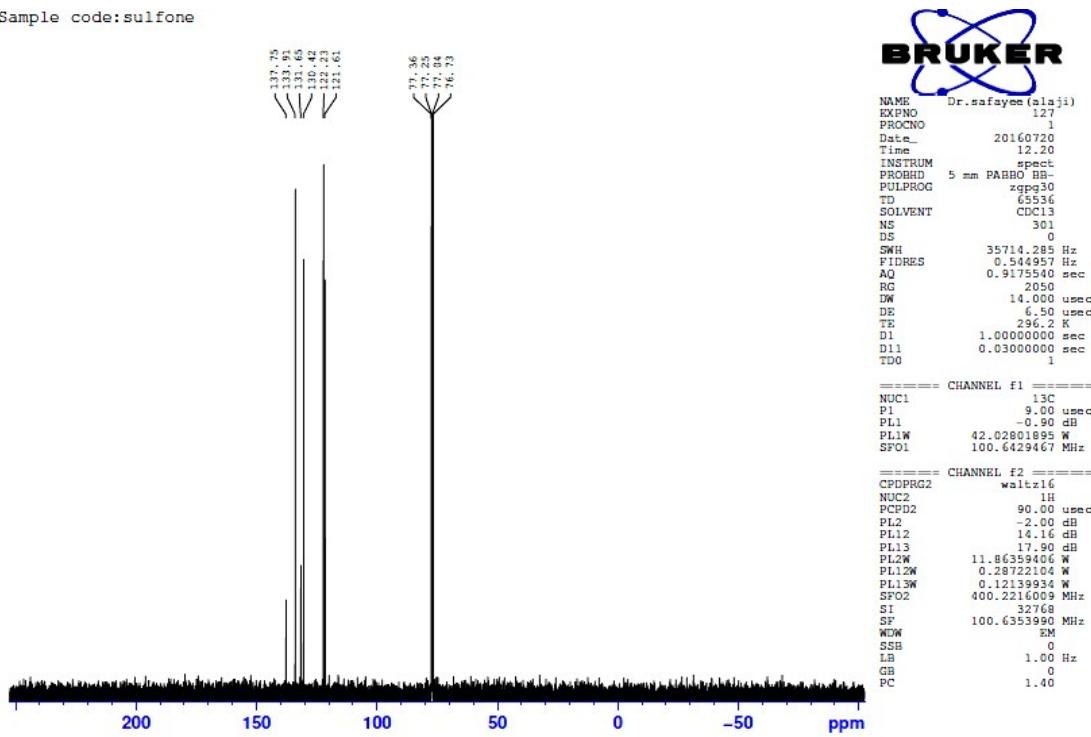


Figure S8 ^{13}C NMR spectrum of dibenzothiophene sulfone

