

Supporting Information for

**Ceria regulated surface properties of Mn/SAPO-34 for improved
NH₃-SCR at low temperature**

Qizhi Chen^{a,c}, Yong Yang^b, Hang Luo^b, Zuohua Liu^b, Zhangfa Tong^c, Changyuan Tao^b, Jun

Du^{b*}

(a. School of Chemistry and Chemical Engineering, Guangxi University, Nanning 530004, China

*(b. College of Chemistry and Chemical Engineering, Chongqing University, Chongqing 401331,
China)*

c. Guangxi Huiyuan Manganese Industry Co., Ltd.

**Corresponding authors: College of Chemistry and Chemical Engineering, Chongqing
University, Chongqing 401331, China.*

E-mail addresses: dujune@cqu.edu.cn (Jun Du),

Mailing address for correspondence:

Dr. Du, Jun (Prof.)

College of Chemistry and Chemical Engineering, Chongqing University

No.55 Daxuecheng South Rd., Shapingba

Chongqing 401331, China

Tel: +86-23-65678923

Fax: +86-23-65678923

E-mail: dujune@cqu.edu.cn

1. Nitrogen adsorption and desorption for Texture analysis

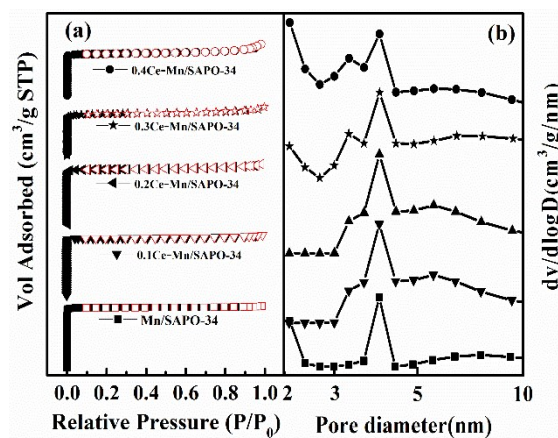


Fig. S1 N₂ adsorption-desorption isotherms (a) and pore distribution (b) of Mn/SAPO-34 and Ce-Mn/SAPO-34.

Table S1 BET surface area and pore structures of Mn/SAPO-34 and Ce-Mn/SAPO-34 catalysts.

Samples	S_{BET} (m ² g ⁻¹) ^a	Pore volume (cm ³ g ⁻¹) ^b	Pore diameter (nm) ^c
Mn/SAPO-34	416.57	0.2117	3.66
0.1Ce-Mn/SAPO-34	408.50	0.1975	3.46
0.2Ce-Mn/SAPO-34	394.78	0.1880	3.45
0.3Ce-Mn/SAPO-34	375.75	0.1692	3.35
0.4Ce-Mn/SAPO-34	370.07	0.1634	3.32

^a BET Specific surface area

^b T-Plot desorption pore volume

^c The most probable pore diameter

2. BET analysis of aged 0.3Ce-Mn/SAPO-34 catalyst

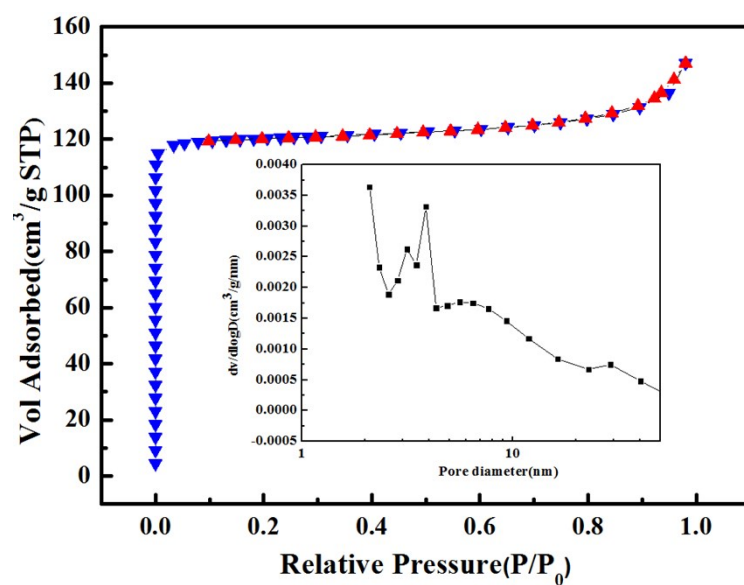


Fig. S2 N₂ adsorption-desorption isotherms and pore distribution of aged 0.3Ce-Mn/SAPO-34 catalyst.

Table S2 BET surface area and pore structures of aged Ce-Mn/SAPO-34 catalyst

Samples	S_{BET} (m ² g ⁻¹) ^a	Pore volume (cm ³ g ⁻¹) ^b	Pore diameter (nm) ^c
0.3Ce-Mn/SAPO-34 aged	359.21	0.0514	1.97

^a BET Specific surface area

^b T-Plot desorption pore volume

^c The most probable pore diameter

3. Surface acid amount of Mn/SAPO-34 and Ce-Mn/SAPO-34 catalysts based NH₃-TPD

Table S3 Surface acidity of Mn/SAPO-34 and Ce-Mn/SAPO-34 catalysts

Samples	Weak acid amount (mmol g ⁻¹)	Strong acid amount (mmol g ⁻¹)	The total acid amount (mmol g ⁻¹)
Mn/SAPO-34	0.203	0.104	0.307
0.1Ce-Mn/SAPO-34	0.195	0.09	0.285
0.2Ce-Mn/SAPO-34	0.148	0.079	0.227
0.3Ce-Mn/SAPO-34	0.143	0.078	0.221
0.4Ce-Mn/SAPO-34	0.100	0.035	0.135

Table.S4 the relationship between the ratio of Ce/Mn and the catalytic activity or selectivity

The atomic concentration ratio of Ce/Mn	The NO conversion in 200°C	The N ₂ selectivity in 200°C
(%)	(%)	(%)
0.00	77.5	93.4
9.32	80.4	94.8
27.30	85.7	96.1
49.05	90.3	98.5
73.16	86.2	96.4

Table. S5 the SCR performance on the catalysts about Ce and Mn in the literatures.

	Catalyst	NO conversion (%)	N ₂ selectivity (%)	N ₂ O concentration (ppm)	reference
1	Ce-Mn/SAPO-34	90.3	98.5	0.2	This work
2	MnCeO _x @Z5	80.2	89.4	23	[S.1]
3	Mn ₄ Co ₁ O _x -HS	100	66.5	41.5	[S.2]
4	CeCu-SSZ-39-750	55	46.8	--	[S.3]
5	6% Cu/SSZ-13	99.4	--	2.8	[S.4]
6	MnCoCeO _x	98.5	98.8	--	[S.5]
7	Mn-Ce nanorods	70.5	89.5	--	[S.6]
8	CeO ₂ -N	73.4	98.5	--	[S.7]
9	11.2Fe11Mn/SBA-15	90	--	--	[S.8]
10	Mn-W/Ce-Cl-C	98	90	--	[S.9]
11	Mn10/W5Ti	97	86	--	[S.10]
12	5W-5Nb/CZ	90.8	--	0.5	[S.11]
13	Ce _{0.2} W _{0.1} TiO _x	79.5	--	2	[S.12]
14	Mn-VO _x /CeO ₂	92	68	--	[S.13]
15	MnO _x /TiO ₂	90	--	--	[S.14]
16	Hollow MnO _x @PrO _x	92	95	--	[S.15]
17	Cu/SAPO-34	60.5	--	4	[S.16]
18	H-Sn/CeO ₂	61	98	--	[S.17]
19	Mn(25%)Ce(28%)/CNTs-S	89.2	--	--	[S.18]
20	MnCo ₂ Nb _{0.9} O _x	98.2	80.5	24.8	[S.19]

- [S.1] R. Yan, S. Lin, Y. Li et al. J Hazard Mater, 2020, 396, 122592
- [S.2] C. Yang, X. Tang, H. Yi. ChemiStrySelect 2020, 5, 9409
- [S.3] Y. Wang, G. Li, S. Zhang. Chem Eng J, 2020, 393, 124782
- [S.4] L. Jiang, Y. Cai, M. Jin. Journal of ChemiStry, 2020, 2020, 1
- [S.5] X. Wang, R. Duanc, W. Liu. Appl Surf Sci. 2020, 510, 145517
- [S.6] C. Yang, J. Yang, Q. Jiao. Ceram Int, 2020,46, 4394
- [S.7] X. Sun, R. Guo, S. Liu. Appl Surf Sci, 2018, 462, 187
- [S.8] G. Lia, B. Wang, H. Wang. Catal Commun, 2018, 108, 82
- [S.9] X. Wang, K. Zhang, W. Zhao. Ind. Eng. Chem. ReS. 2017, 56, 14980
- [S.10] J. Shia, Z. Zhang, M. Chen. Fuel, 2017,210, 783
- [S.11] H. Xu, Q. Lin, Y. Wang. RSC Adv, 2017, 7, 47570
- [S.12] Y. Geng, H. Huang, X. Chen. RSC Adv, 2016, 6, 64803
- [S.13] Z. Lian, F. Li, H. He. RSC Adv., 2015,5, 37675
- [S.14] X. Gao, L. Li, L. Song. RSC Adv., 2015, 5, 29577
- [S.15] S. Cheng, J. Shao, B. Huang. RSC Adv., 2020, 10, 13855
- [S.16] Y. Duan, J. Wang, T. Yu. RSC Adv., 2015, 5, 14103
- [S.17] C. Fang, L. Shi, H. Li. RSC Adv, 2016,6, 78727

[S.18] D. Zhang, L. Zhang, C. Fang. RSC Adv, 2013, 3, 882

[S.19] T. Xu, C. Wang, X. Wu. RSC Adv., 2016,6, 97004