

Electrochemical performance of activated carbon fiber with hydrogen bond-induced high sulphur/nitrogen doping

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Table S1 Elemental contents in S/N-CF, S/N-CF-O and S/N-ACF.

Samples	Surface content (at.%) of different N functionalities								Surface content (at.%) of different S functionalities		
	C	O	N	S	N-6	N-5	N-Q	N-O	S ₁	S ₂	S ₃
S/N-CF	98.14	—	1.25	0.61	0.54	0.18	0.53	—	0.38	0.23	—
S/N-CF-O	90.40	7.37	1.45	0.78	0.43	0.55	0.09	0.38	0.17	0.46	0.15
S/N-ACF	88.77	3.71	4.36	3.16	1.26	1.75	0.67	0.68	1.30	1.28	0.58

Note: N-6 is pyridinic nitrogen; N-5 is pyrrolic nitrogen; N-Q is graphitic nitrogen; S₁ is thiophene-S2p_{3/2}; S₂ is thiophene-S2p_{1/2}; S₃ is the sulfur oxide.

Table S2 Comparison of nitrogen and sulfur content of recently reported carbons with S/N co-doping.

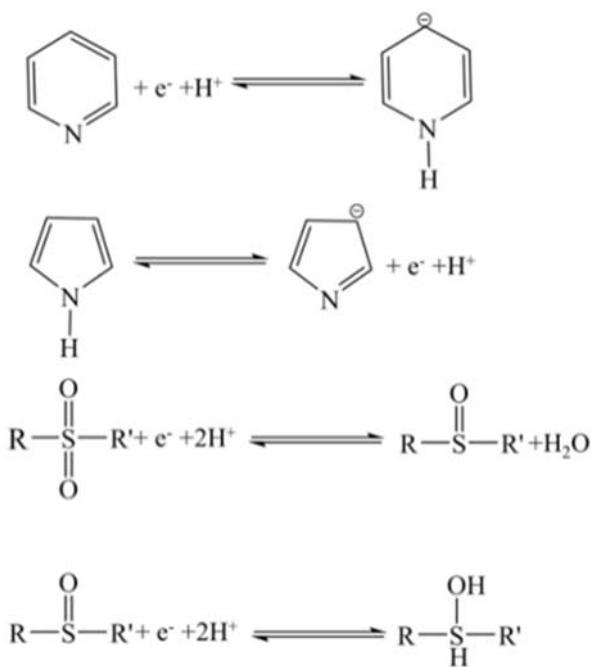
Samples	methods	N (At. %)	S (At. %)	Ref.
N/S Co-Doped Graphene Hydrogel	Graphene, (NH ₄) ₂ SO ₄ and melamine at 600 °C	2.86	0.07	¹
N/S Self-Doped Activated Carbon	Pretreated elm flowers at 700 °C for 1h	2.61	0.8	²
N/S Co-Doped Graphene Hydrogel	Hydrothermal treatment of TU and GO	2.5	1.2	³
Porous Carbon	Glucose, NaHCO ₃ and TU at 700 °C for 2h	3.42	0.2	⁴
S-RGO	H ₂ SO ₄ , H ₂ O ₂ and graphite at 1000 °C for 2h	-	1.47	⁵
CFC-750-N-S	LiNO ₃ , Na ₂ S ₂ O ₃ and CFC at 1000 °C for 2h	2.72	1.84	⁶
N/S co-doped carbon nanosheets	C15 powders and TU at 800 °C for 3h	1.7	2.6	⁷
S/N-ACF	TU and CF-OH at 1000 °C for 2h	4.36	3.16	This work

Table S3 Summary of calculated results: the binding energy (E_b), the distance of TU to C₃₂ or hydrogen bond (d).

Configurations	Binding site	E_b (eV)	d (Å)
TU/C ₃₂	N	0.73	3.08
TU/C ₃₂ -O	H	0.79	1.78
TU/C ₃₂ -OH	H	1.46	1.86
	S		1.40

Table S4 Fitting values of the equivalent circuit elements for S/N-CF, S/N-CF-O and S/N-ACF.

Samples	R_o (Ω)	R_{ct} (Ω)	CPE		W_o		
			CPE _T	CPE _P	W_R	W_T	W_P
S/N-CF	2.36	0.82	0.33	0.83	2.97	2.01	0.48
S/N-CF-O	4.15	4.83	0.35	0.81	1.94	1.32	0.47
S/N-ACF	1.36	0.72	0.51	0.97	1.35	1.20	0.49

**Fig. S1** Redox reaction of S, N compounds.

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