Electronic Supplementary Information

Crystal growth control of rod-shaped ε-Fe₂O₃ nanocrystals

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Contents

§1. Crystal structure analyses. ----- Fig. S1 S2 Table S1 S2 Fig. S2 S3 §3. Magnetic hysteresis loops at room temperature. ----- Fig. S4 **S6** §4. Raman spectra. ----- Fig. S5 **S**7 §5. Rectangular-type rod-shaped ε -Fe₂O₃ crystal and its crystallographic direction. **S**8 Fig. S6 **§6.** Surface energy in ε-Fe₂O₃.----- **Table S3** S9 §7. Schematic of the lattice planes, values of surface energy, and number of FeO₆-broken bond per area. ----- Fig. S7 S10

Page

§1. Crystal structure analyses.



Fig. S1. Phase fractions obtained by Rietveld analyses for samples 1–7.

Table S1. Lattice parameters of ε -Fe₂O₃ in 1–7 and refinement parameters obtained by Rietveld analyses of the XRPD patterns.

	Orthorhombic : $Pna 2_1$				Refinement p	parameters
Sample	Lattice parameters			Crystal size		
	a (Å)	b (Å)	c (Å)	(nm)	$R_{\rm wp}$ (%)	S
1	5.064(8)	8.729(14)	9.610(9)	4.4	0.77	1.30
2	5.087(2)	8.787(3)	9.489(3)	5.9	0.71	1.17
3	5.0906(12)	8.7777(19)	9.4924(15)	5.5	0.75	1.29
4	5.0864(4)	8.7856(8)	9.4741(5)	9.8	0.62	1.21
5	5.0895(5)	8.7860(10)	9.4705(9)	15.2	0.74	1.29
6	5.0896(5)	8.7873(9)	9.4714(9)	16.8	0.88	1.65
7	5.0927(3)	8.7859(5)	9.4732(5)	26.6	1.17	1.99



Fig. S2. Phase diagram of the phase ratio versus the sintering temperature for iron oxides.

§2. Particle sizes.

Sample	Particle size for long axis	Particle size for short axis	Aspect ratio	
Sample	$d_{\rm long}$ (nm)	$d_{\rm short}$ (nm)	$d_{\rm long}$ / $d_{\rm short}$	
1	11.9 ± 3.5	8.4 ± 1.9	1.4	
2	22.2 ± 8.1	12.5 ± 3.2	1.8	
3	23.0 ± 9.3	12.3 ± 4.1	1.9	
4	31.7 ± 13.0	15.8 ± 4.7	2.0	
5	50.0 ± 25.2	21.3 ± 5.9	2.3	
6	70.1 ± 44.0	24.0 ± 7.6	2.9	
7	93.2 ± 105	28.2 ± 12.2	3.3	

Table S2. Particle size of 1–7 obtained by the TEM images.

The values of the errors are the standard deviation.



Fig. S3. Particle size distributions of the long-axis (gray) and the short-axis (orange) for **1** (a), **2** (b), **3** (c), **4** (d), **5** (e), **6** (f), and **7** (g). Blue lines are the fitted curve by the log-normal distribution function.

§3. Magnetic hysteresis loops at room temperature.



Fig. S4. Magnetic hysteresis loops measured at room temperature up to 7 Tesla for 1–7.

§4. Raman spectra.



Fig. S5. Raman spectra measured at room temperature for 1–7. Black lines indicate the reported peak positions of α -Fe₂O₃.³⁸ Bottom spectrum is a reference of ε -Ga^{III}_{0.48}Fe^{III}_{1.52}O₃, which was previously reported.^{S1} Spectra of 1–7 do not show any peaks attributed to the divalent iron. (i.e., Raman spectra of 1–7 indicate iron is in the trivalent state).

^{S1} S. Ohkoshi, M. Yoshikiyo, Y. Umeta, M. Komine, R. Fujiwara, H. Tokoro, K. Chiba, T. Soejima, A. Namai, Y. Miyamoto, T. Nasu, Phonon mode calculation, Far- and Mid-infrared, and Raman spectra of ε-Ga_{0.5}Fe_{1.5}O₃, *J. Phys. Chem. C*, 2017, **121**, 5812–5819.

§5. Rectangular-type rod-shaped ε-Fe₂O₃ nanocrystal and its crystallographic direction.



Fig. S6. Schematic of a rectangular-type rod-shaped ε -Fe₂O₃ nanocrystal and its crystallographic direction. Red, blue, and gray spheres indicate Fe at octahedral sites, Fe at tetrahedral sites, and O, respectively.

§6. Surface energy in ε-Fe₂O₃.

Lattice plane	$G = G = (-2)^{*}$	N _{hkl}		
Lattice plane	Surface energy α_{hkl} (nm)*	Number of FeO ₆ -broken bond per area (nm^{-2})		
(1 0 0)	19.2	14.4		
(0 1 0)	18.7	16.6		
(0 0 1)	13.4	8.9		
(1 1 0)	19.0	15.2		
(0 1 1)	11.9	7.1		
(1 0 1)	19.7	15.1		
(1 1 1)	17.8	13.4		

Table S3. Surface energy and the ratio of the coordination bonds to the broken bonds in ϵ -Fe₂O₃.

*Relative surface energy, number of broken bonds per area assuming that the Fe and O sites have equal binding energies.

§7. Schematic of the lattice planes, values of surface energy, and number of FeO₆-broken bond per area.



Fig. S7. Schematic of the lattice planes (green plane), values of surface energy (α_{hkl} , nm⁻²), and number of FeO₆-broken bond per area (N_{hkl} , nm⁻²) for the rod-shaped ε -Fe₂O₃ nanocrystals. Red, blue, and gray spheres in the crystal structure indicate the Fe at octahedral site, Fe at tetrahedral site, and O, respectively.