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Ultrafast-charging and long cycle-life anode materials of TiO₂-bronze/nitrogen-doped graphene nanocomposites for high-performance lithium-ion batteries

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Fig. S1 Schematic diagram of experimental steps for the synchronous preparation of TNG nanocomposites via the hydrothermal method and followed by heat-treatment process.



Fig. S2 Powder XRD patterns measured in a range of 2θ from 10° to 60° of graphite, GO, rGO, and nitrogen-doped graphene.

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Fig. S3 the high magnified curve fitting of C 1s XPS spectra



Fig. S4 SEM images of (a) nitrogen-doped graphene, and TNG products, (b) TNG-3h, (c) TNG-6h, (d) TNG-9h, (e) TNG-12h, and (f) TNG-24h, which were prepared from different hydrothermal reaction times.



Fig. S5 the XRD patterns with reference lines for peak assignment: (a) $NaTi_3O_6(OH)(H_2O)$ after the hydrothermal reaction, (b) $H_2Ti_3O_7$ after ion-exchanging, and (c) $TiO_2(B)$ after heat treatment.



Fig. S6 Thermogravimetric analysis curves of as-prepared TNG products

Table S1. Discharge-charge capacities and coulombic efficiencies of TNG nanocomposite
anodes at a current rate of 1C.

Sample	Phase and morphology	Cycle	Discharge	Charge	Coulombic
		number	(mAh g ⁻¹)	$(mAh g^{-1})$	efficiency (%)
TNG-3h	Anatase nanoparticles on NG	1 st cycle	474.01	391.66	82.63
		2 nd cycle	403.94	374.43	92.69
		3 rd cycle	385.63	364.23	94.45
TNG-6h	Anatase nanosheets on NG	1 st cycle	347.19	288.49	83.09
		2 nd cycle	300.03	280.34	93.44
		3 rd cycle	289.26	274.09	94.76
TNG-9h	TiO ₂ (B) nanotubes on NG	1 st cycle	430.13	361.20	83.97
		2 nd cycle	375.98	354.43	94.27
		3 rd cycle	381.38	364.83	95.66
TNG-12h	TiO ₂ (B) nanorods on NG	1 st cycle	400.99	342.24	85.35
		2 nd cycle	357.07	338.41	94.77
		3 rd cycle	350.68	336.96	96.09
TNG-24h	TiO ₂ (B) nanorods on NG	1 st cycle	634.54	527.42	83.12
		2 nd cycle	550.14	507.79	92.30
		3 rd cycle	525.57	494.07	94.01