

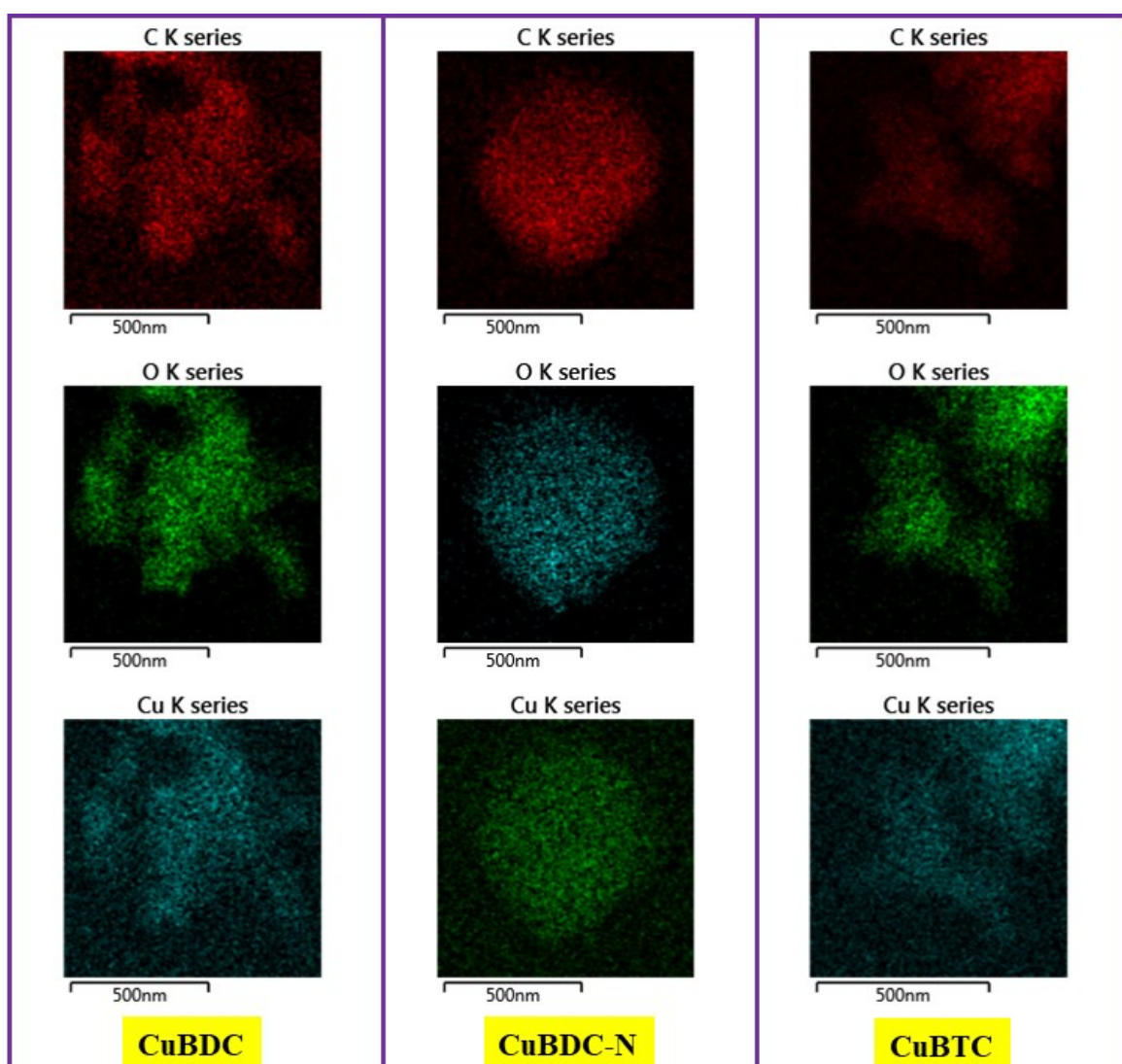
Chemisorption of hydrogen sulfide over copper-based metal-organic frameworks:

Methanol and UV-assisted regeneration

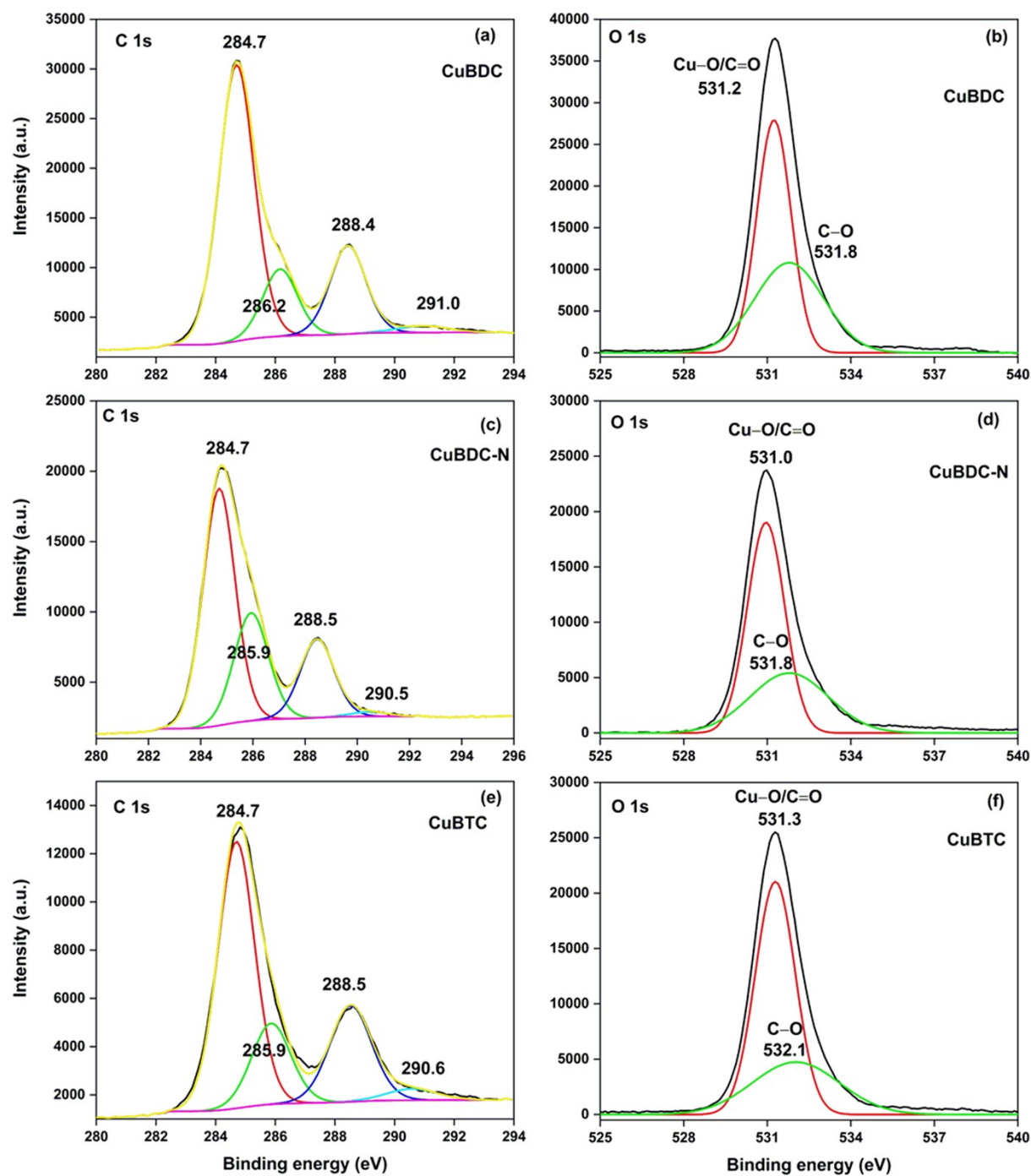
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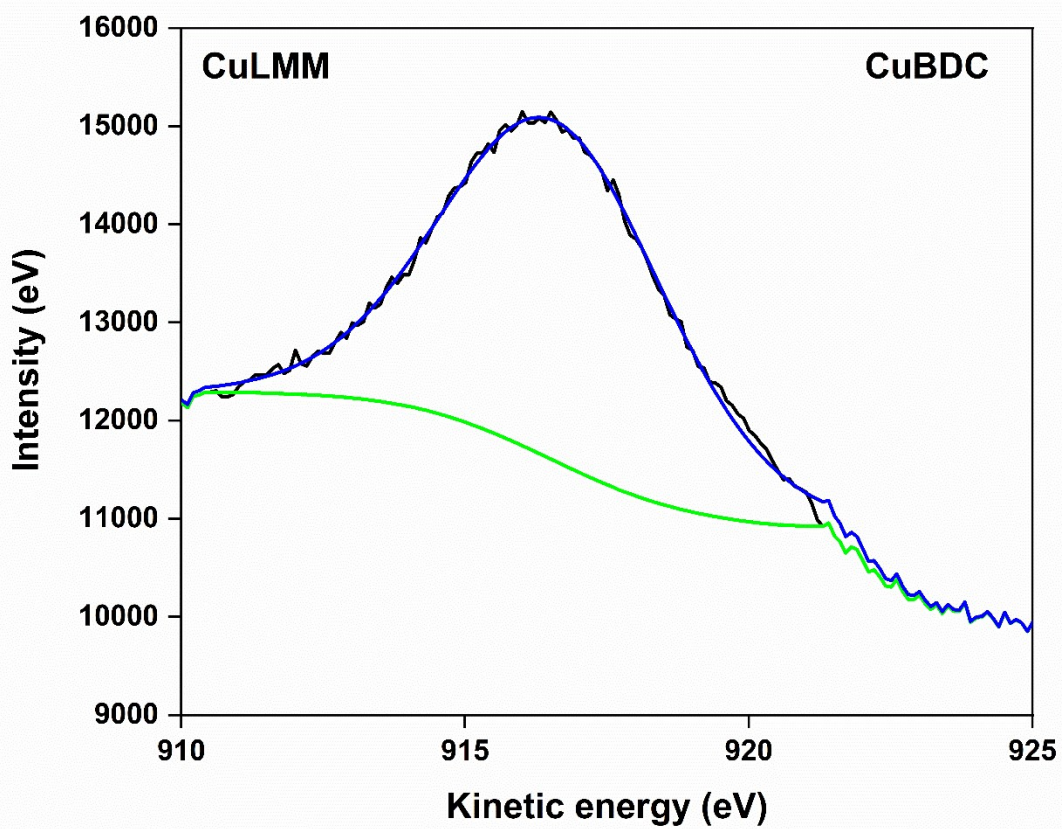
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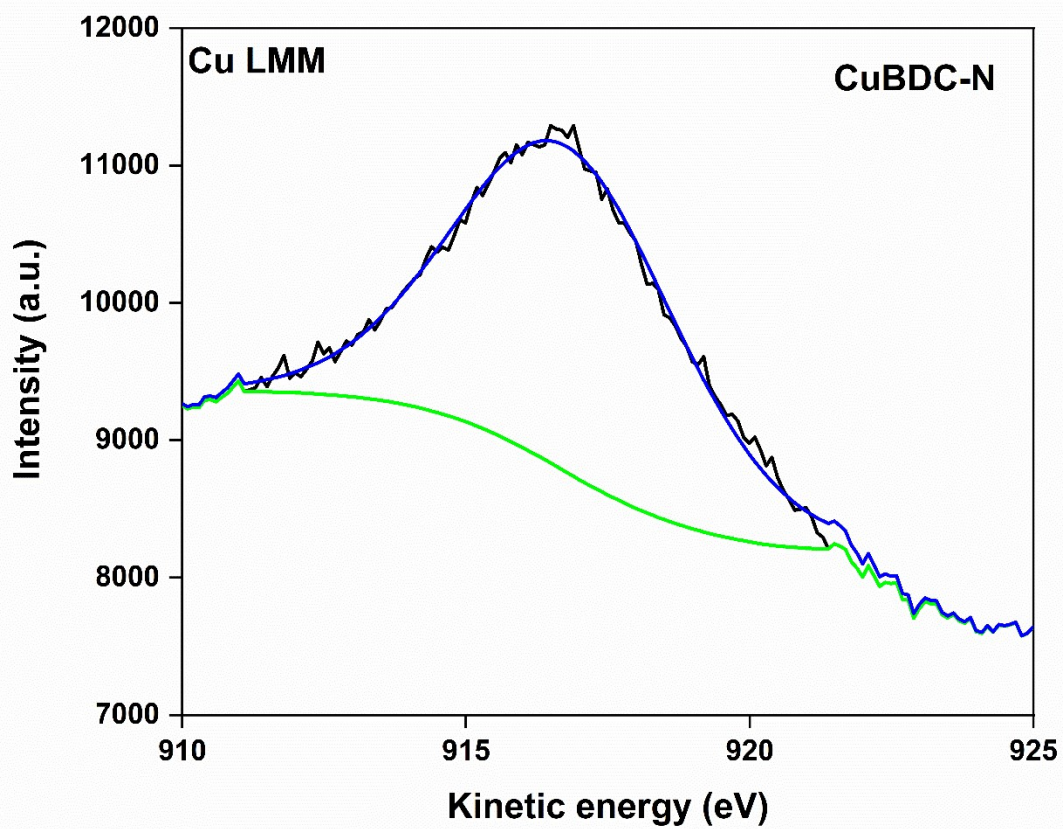
S. Figure 1. 2D elemental mapping of Cu-MOFs.



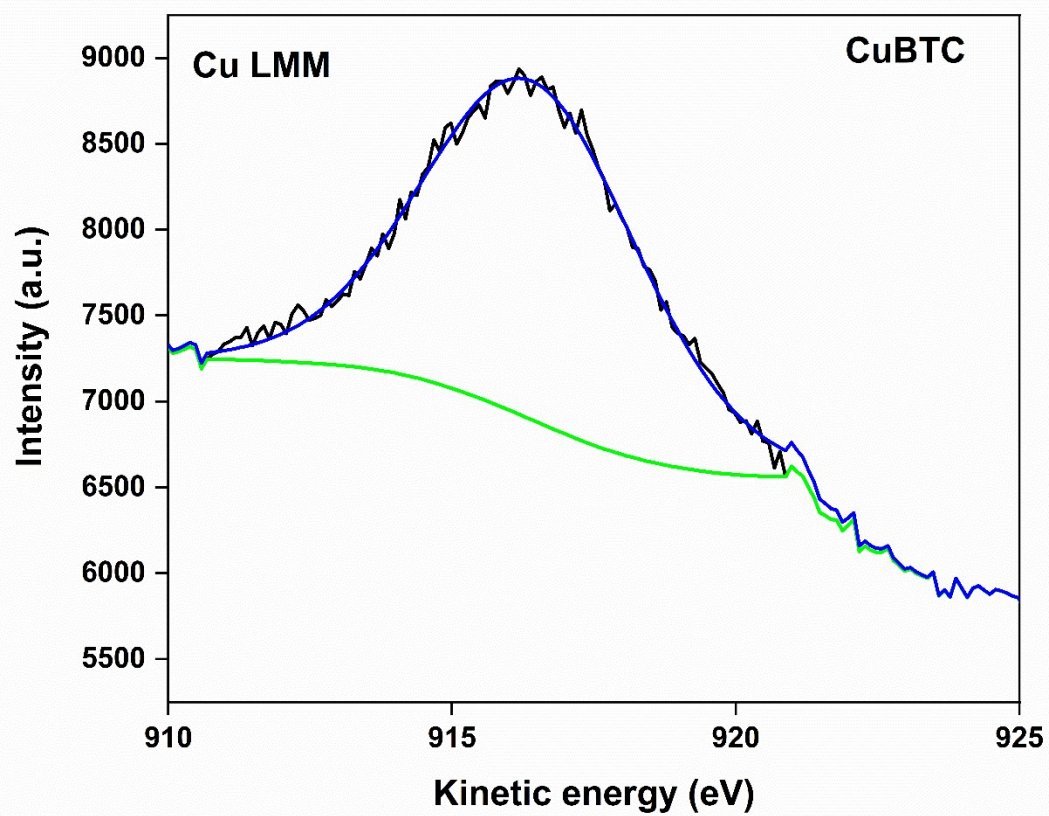
S. Figure 2. HRXPS C 1s spectra of (a) CuBDC; (c) CuBDC-N; (e) CuBTC; HRXPS O 1s spectra of (b) CuBDC; (d) CuBDC-N; (f) CuBTC.



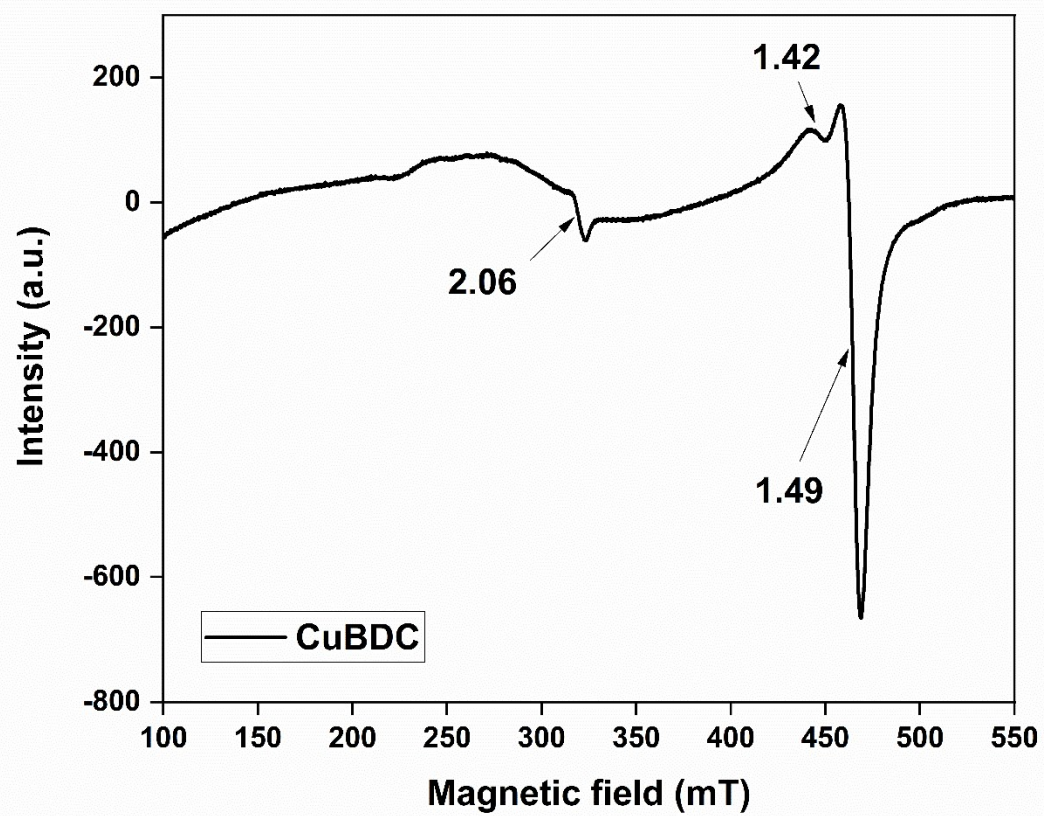
S. Figure 3a. HRXPS Cu LMM spectrum of CuBDC.



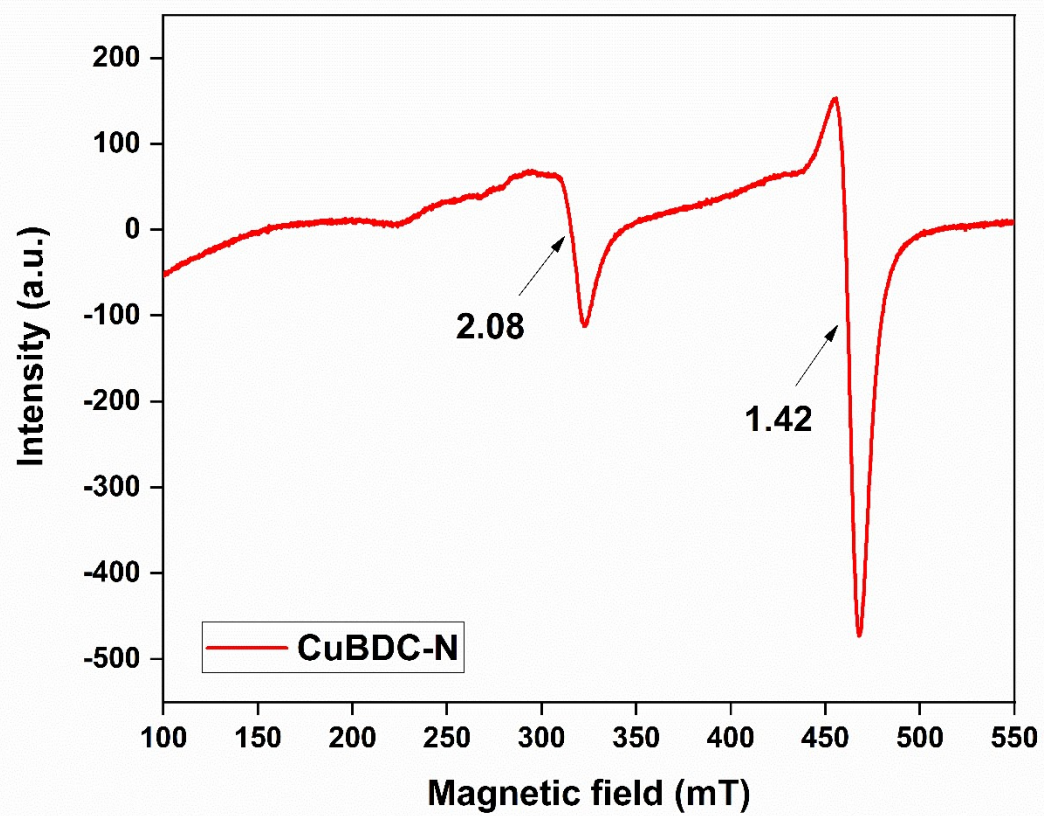
S. Figure 3b. HRXPS Cu LMM spectrum of CuBDC-N.



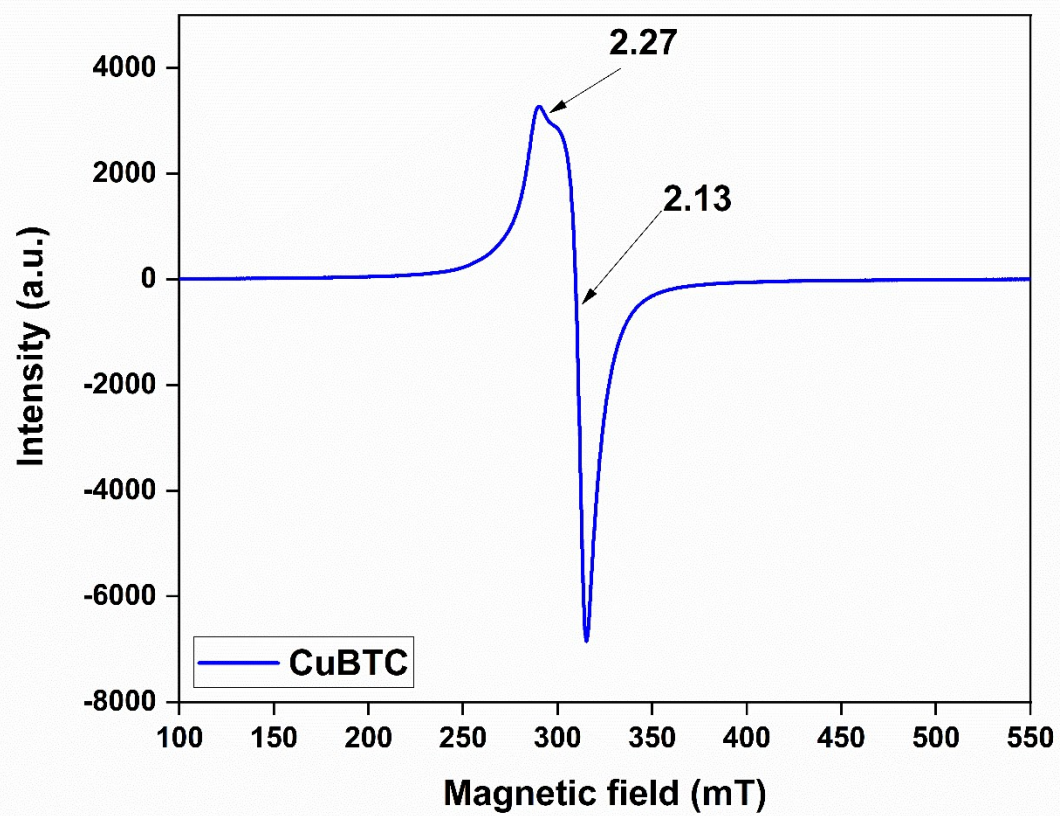
S. Figure 3c. HRXPS Cu LMM spectrum of CuBTC.



S. Figure 4a. ESR spectrum of CuBDC recorded at 25 °C.



S. Figure 4b. ESR spectrum of CuBDC-N recorded at 25 °C.



S. Figure 4c. ESR spectrum of CuBTC recorded at 25 °C.

S. Table 1. Appearance and yield of synthesized Cu-MOFs

MOF	Color	Yield (g)
CuBDC	Light blue	14.07
CuBDC-N	Dark green	13.53
CuBTC	Light blue	11.61

S. Table 2. Miller indices and corresponding diffraction angle of CuBDC

2θ (degree)	hkl	d (Å)	Area%	FWHM
10.16	110	8.70	92.8	0.203
12.08	020	7.32	43.6	0.291
13.58	11-1	6.52	12.2	0.220
17.12	-201	5.17	100.0	0.253
17.70	111	5.01	18.4	0.239
20.40	220	4.35	25.2	0.273
24.82	131	3.58	53.7	0.197

S. Table 3. Miller indices and corresponding diffraction angle of CuBDC-N

2θ (degree)	hkl	d (Å)	Area%	FWHM
10.28	110	8.60	74.1	0.139
11.82	001	7.48	46.0	0.180
13.14	11-1	6.73	14.4	0.148
16.80	-201	5.27	100.0	0.149
18.02	111	4.92	23.3	0.182
20.70	220	4.29	20.6	0.138
24.71	131	3.60	46.1	0.121

S. Table 4. Miller indices and corresponding diffraction angle of CuBTC

2θ (degree)	hkl	d (Å)	Area%	FWHM
9.46	022	9.34	87.5	0.241
11.62	222	7.61	87.4	0.184
13.42	004	6.59	100.0	0.215
19.02	044	4.66	88.8	0.189
25.96	355	3.43	89.0	0.222

S. Table 5. Crystallite size and lattice parameters of Cu-MOFs

Cu-MOF	<i>D</i> (nm)	Lattice parameters
CuBDC	31.7	Monoclinic, space group = $C2/m$; $a = 11.37 \text{ \AA}$, $b = 14.62 \text{ \AA}$, $c = 7.72 \text{ \AA}$, $\beta = 108.49^\circ$
CuBDC-N	46.6	Monoclinic, space group = $C2/m$; $a = 11.23 \text{ \AA}$, $b = 14.96 \text{ \AA}$, $c = 8.02 \text{ \AA}$, $\beta = 111.67^\circ$
CuBTC	38.2	Cubic, space group = $Fm-3m$; $a = 26.36 \text{ \AA}$

S. Table 6. The peak-fitting results of C 1s high resolution signals of CuBDC, CuBDC-N and CuBTC.

Samples	Assignment	E_B (eV)	FWHM (eV)	At. %
CuBDC	C1s $C=C$	284.7	1.4	62.6
	C1s $C-C$, $C-H$	286.2	1.4	14.9
	C1s $O-C=O$	288.4	1.4	20.4
	C1s $COOH$	291.0	2.0	2.1
CuBDC-N	C1s $C=C$	284.7	1.4	55.0
	C1s $C-C$, $C-H$, $C-N$	285.9	1.5	26.5
	C1s $O-C=O$	288.5	1.5	18.5
	C1s $COOH$	290.5	1.6	1.2
CuBTC	C1s $C=C$	284.7	1.4	55.7

	C1s _{C-C, C-H}	285.9	1.5	18.1
	C1s _{O-C=O}	288.5	1.6	22.9
	C1s _{COOH}	290.6	2.0	3.3

S. Table 7. Curve fitting parameters from CuLMM Auger Signals for CuBDC, CuBDC-N and CuBTC samples.

Samples	BE (eV)	KE (eV)	α' (eV)	Assignment
CuBDC	570.6	916.6	1849.0	Cu ⁺
			1850.6	Cu ²⁺
CuBDC-N	570.0	916.8	1849.3	Cu ⁺
			1851.0	Cu ²⁺
CuBTC	570.6	916.4	1848.7	Cu ⁺
			1850.7	Cu ²⁺

S. Table 8. Elemental composition of Cu-MOFs using TEM-EDS analysis.

Element	CuBDC		CuBDC-N		CuBTC	
	Wt.%	At.%	Wt.%	At.%	Wt.%	At.%
C	46.99	67.50	58.64	76.67	56.94	77.55
O	22.44	24.20	17.85	17.52	14.86	15.20
Cu	30.58	8.30	23.51	5.81	28.20	7.26

S. Table 9. Surface elemental composition (atomic%) of Cu-MOFs using XPS analysis.

Element	CuBDC	CuBDC-N	CuBTC
C	65.66	63.14	60.48
Cu	5.93	4.98	6.60
O	28.41	24.46	32.92
N	-	7.42	-

S. Table 10. Surface elemental composition (atomic%) of fresh, spent, and regenerated CuBTC using XPS analysis.

Element	CuBTC	Spent CuBTC	Reg. CuBTC
C	60.48	50.67	55.05
O	32.92	38.08	34.92
Cu	6.60	7.33	8.53
S	-	3.91	1.49