

## Supplementary

### Oxide modified Al for removal of methyl orange and methyl blue in aqueous solution

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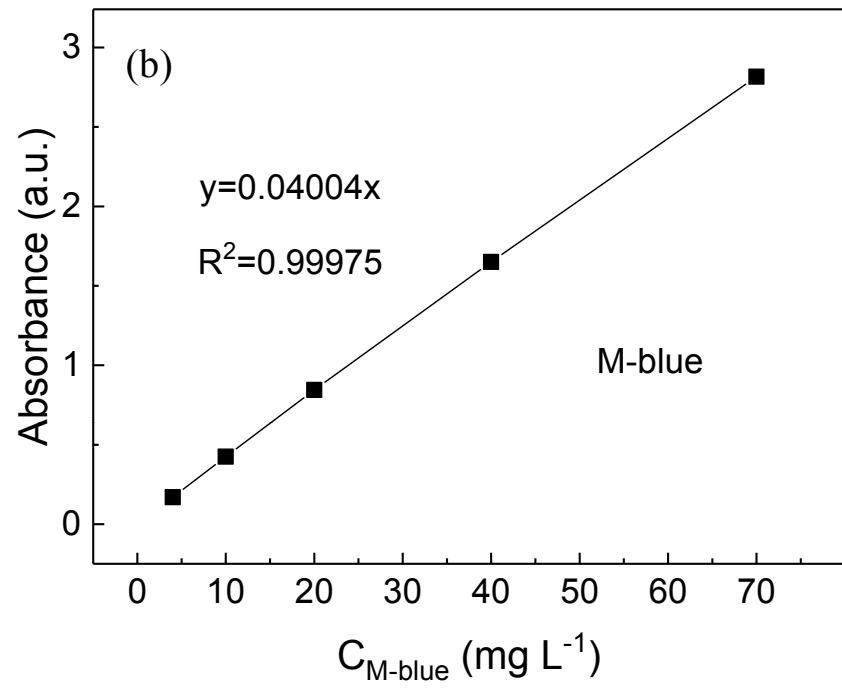
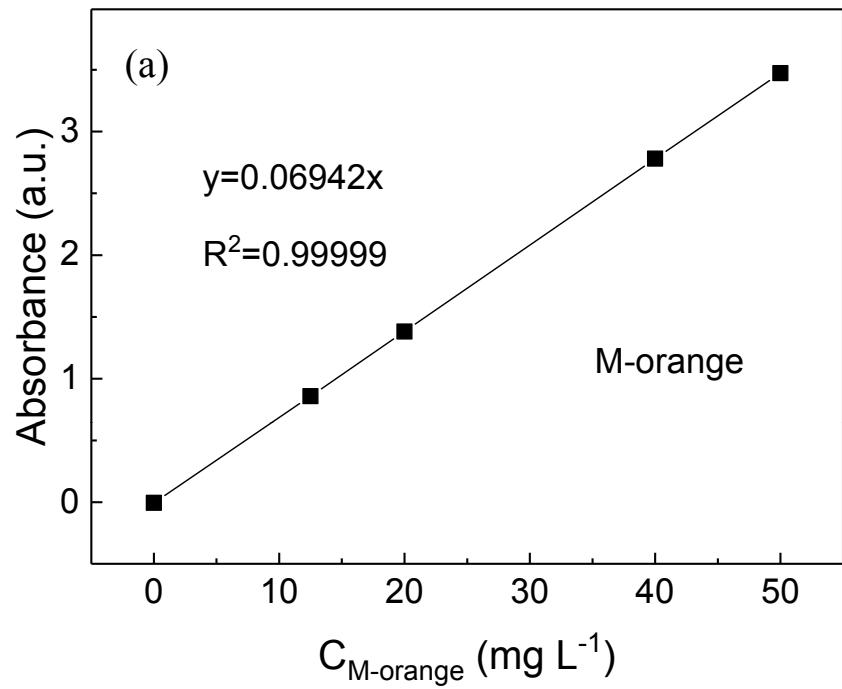
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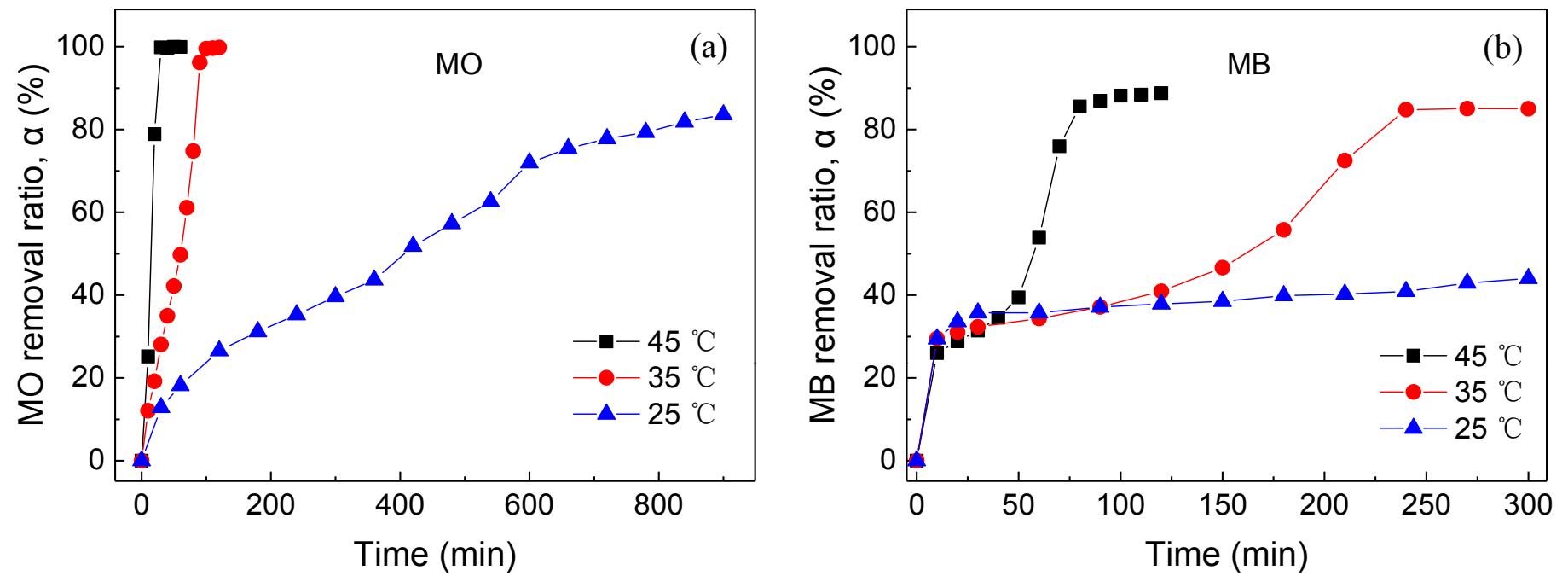
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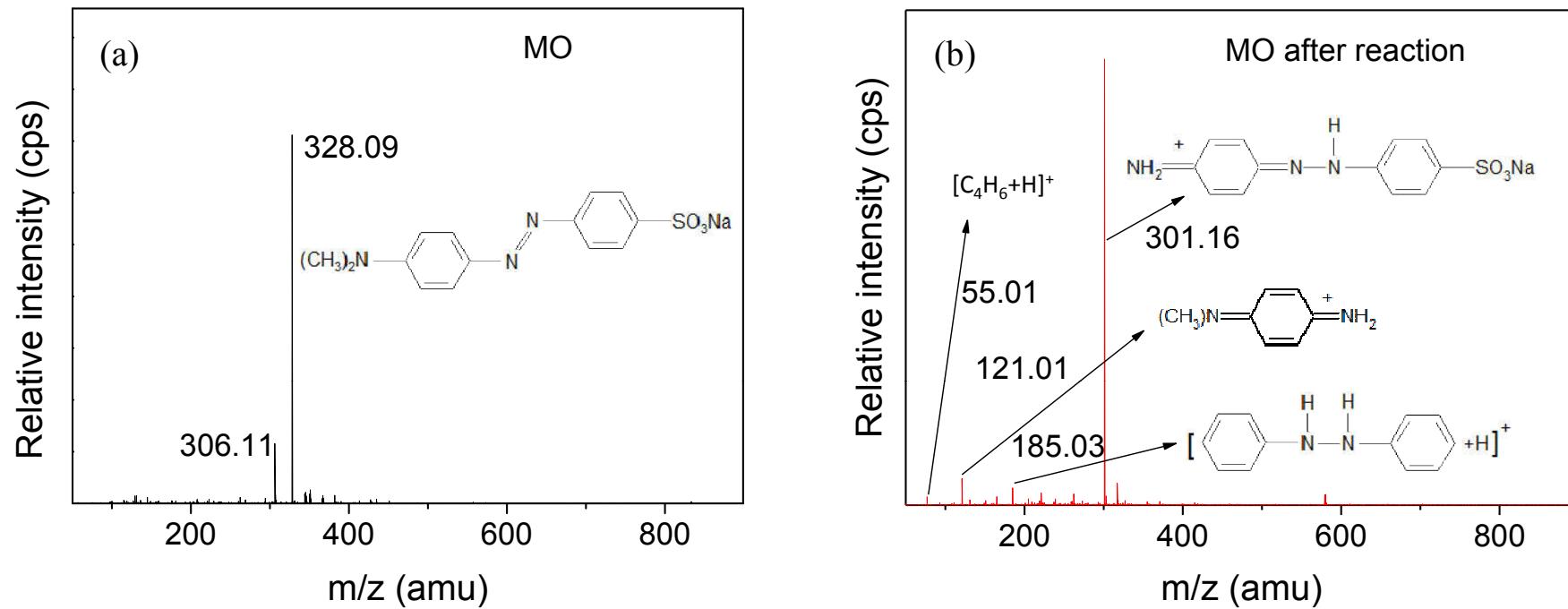
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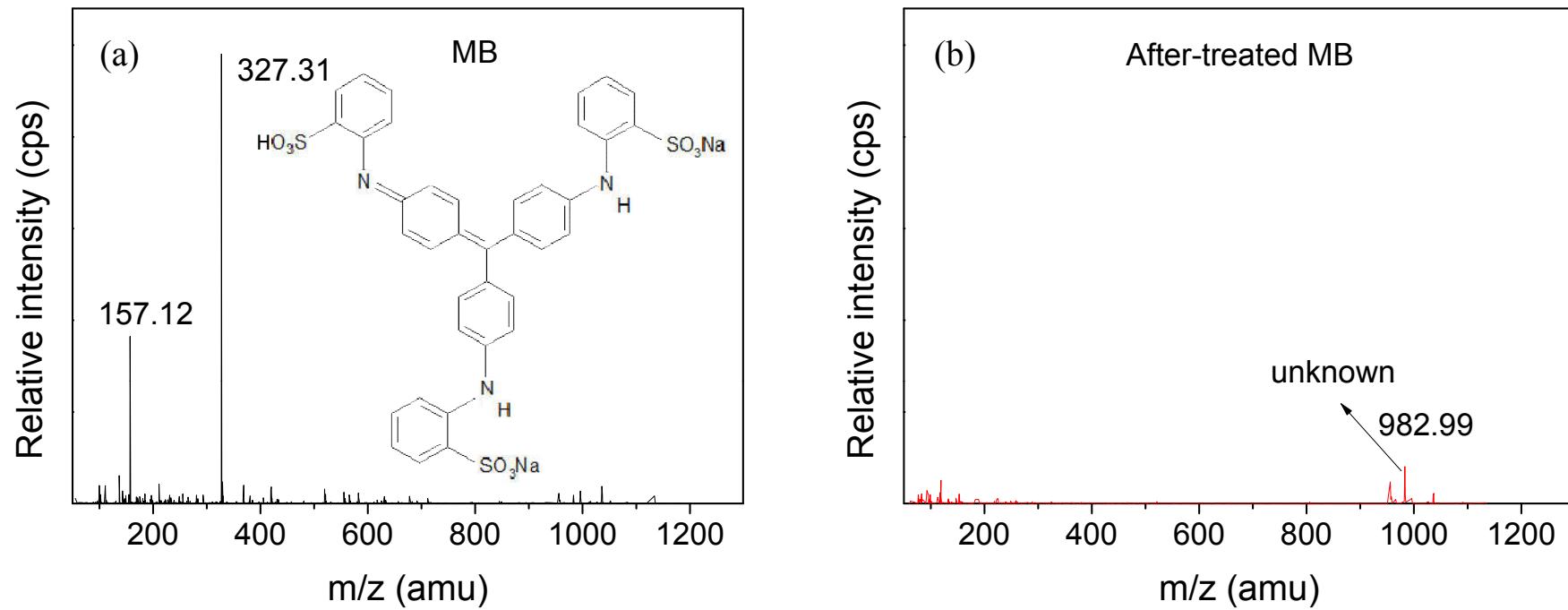
**Fig. S1** Linear fitting of the main absorption peak heights of (a) M-orange ( $\lambda_{\max} \sim 464$  nm) and (b) M-blue ( $\lambda_{\max} \sim 314$  nm) to their concentrations in aqueous solutions.



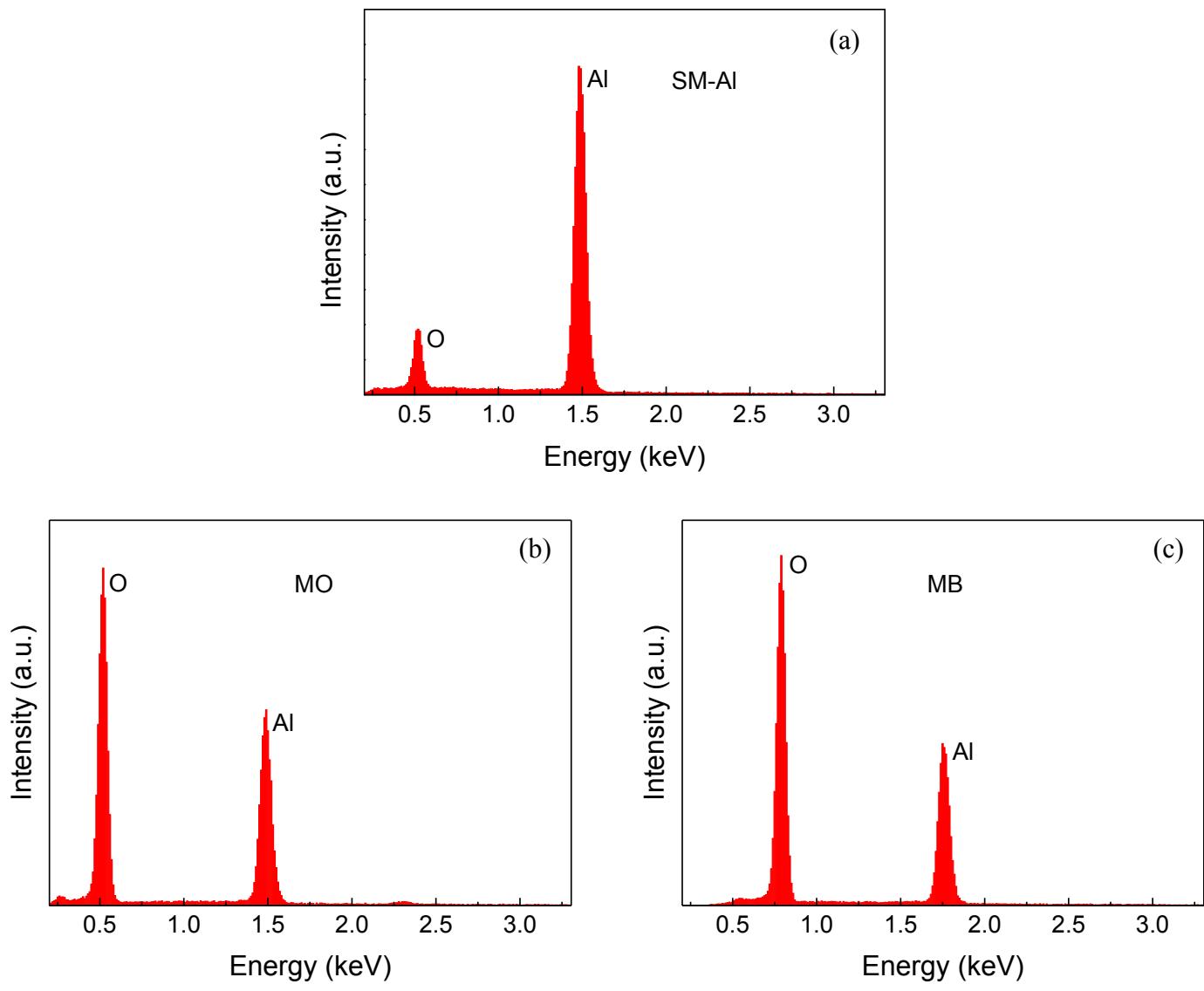
**Fig. S2** Dependence of (a) M-orange and (b) M-blue removal in aqueous solutions on reaction time at different temperature using 100 nm Al powder ( $C_{M\text{-orange}} = C_{M\text{-blue}} = 20 \text{ mg L}^{-1}$ , Al dosage =  $1 \text{ g L}^{-1}$ ).



**Fig. S3** LC/MS spectra of M-orange solution in the positive ion mode (a) before and (b) after reaction using surface modified 7.29  $\mu\text{m}$  Al powder ( $C_{\text{M-orange}} = 20 \text{ mg L}^{-1}$ , Al dosage = 1 g L<sup>-1</sup>, T = 45 °C).



**Fig. S4** LC/MS spectra of M-blue solution in the negative ion mode (a) before and (b) after reaction using surface modified 7.29  $\mu\text{m}$  Al powder ( $C_{\text{M-blue}} = 20 \text{ mg L}^{-1}$ , Al dosage = 1 g  $\text{L}^{-1}$ ,  $T = 45^\circ\text{C}$ ).



**Fig. S5.** EDS spectra of (a) surface modified  $7.29 \mu\text{m}$  Al powder, that in (a) after reaction with MO (b) or MB (c) in aqueous solutions ( $C_{\text{M-orange}} = C_{\text{M-blue}} = 20 \text{ mg L}^{-1}$ , Al dosage =  $1 \text{ g L}^{-1}$ , T =  $45^\circ\text{C}$ ).