

## Preparation of Spiro[indole-3,5'-isoxazoles] via Grignard Conjugate Addition/Spirocyclization Sequence

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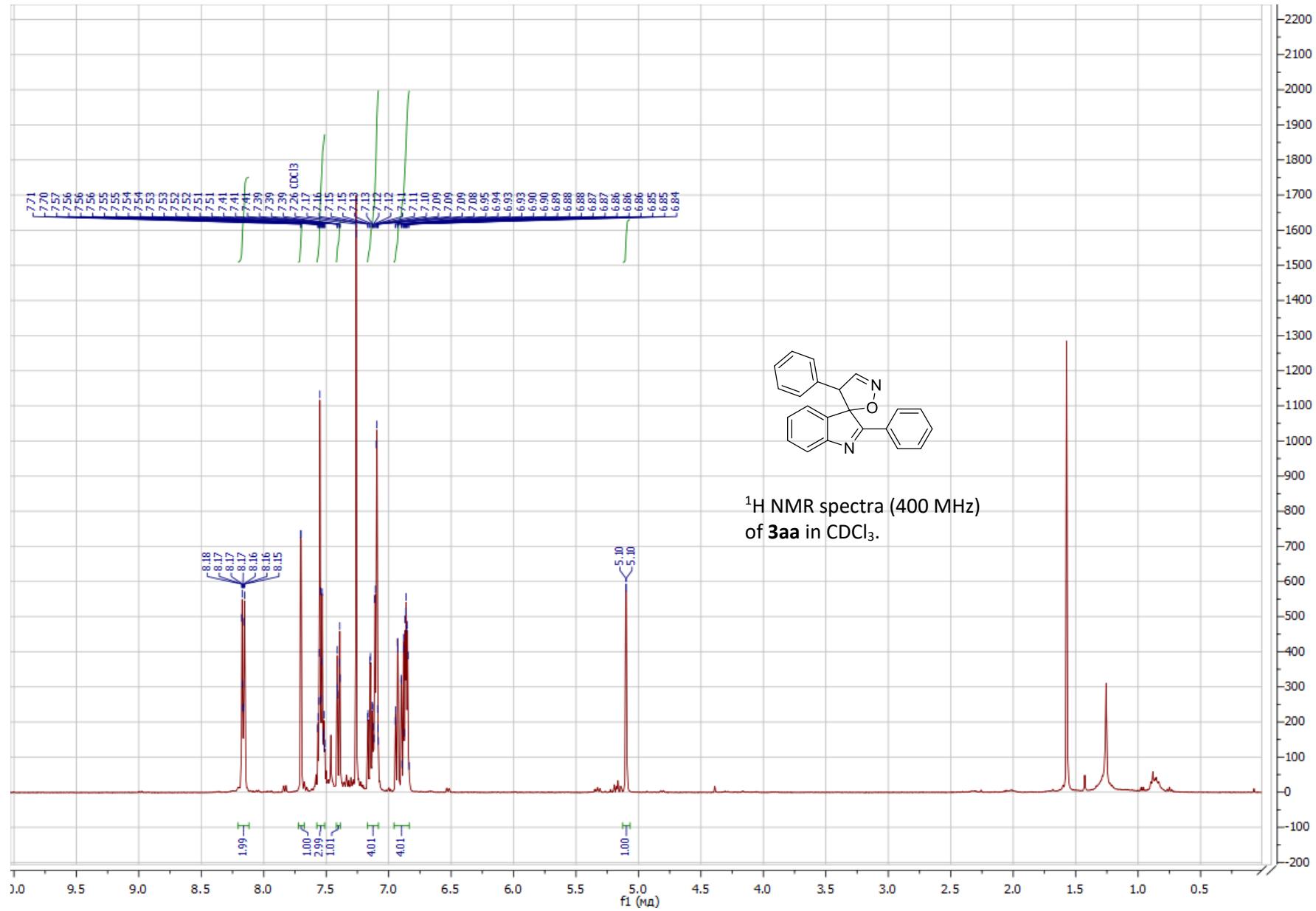
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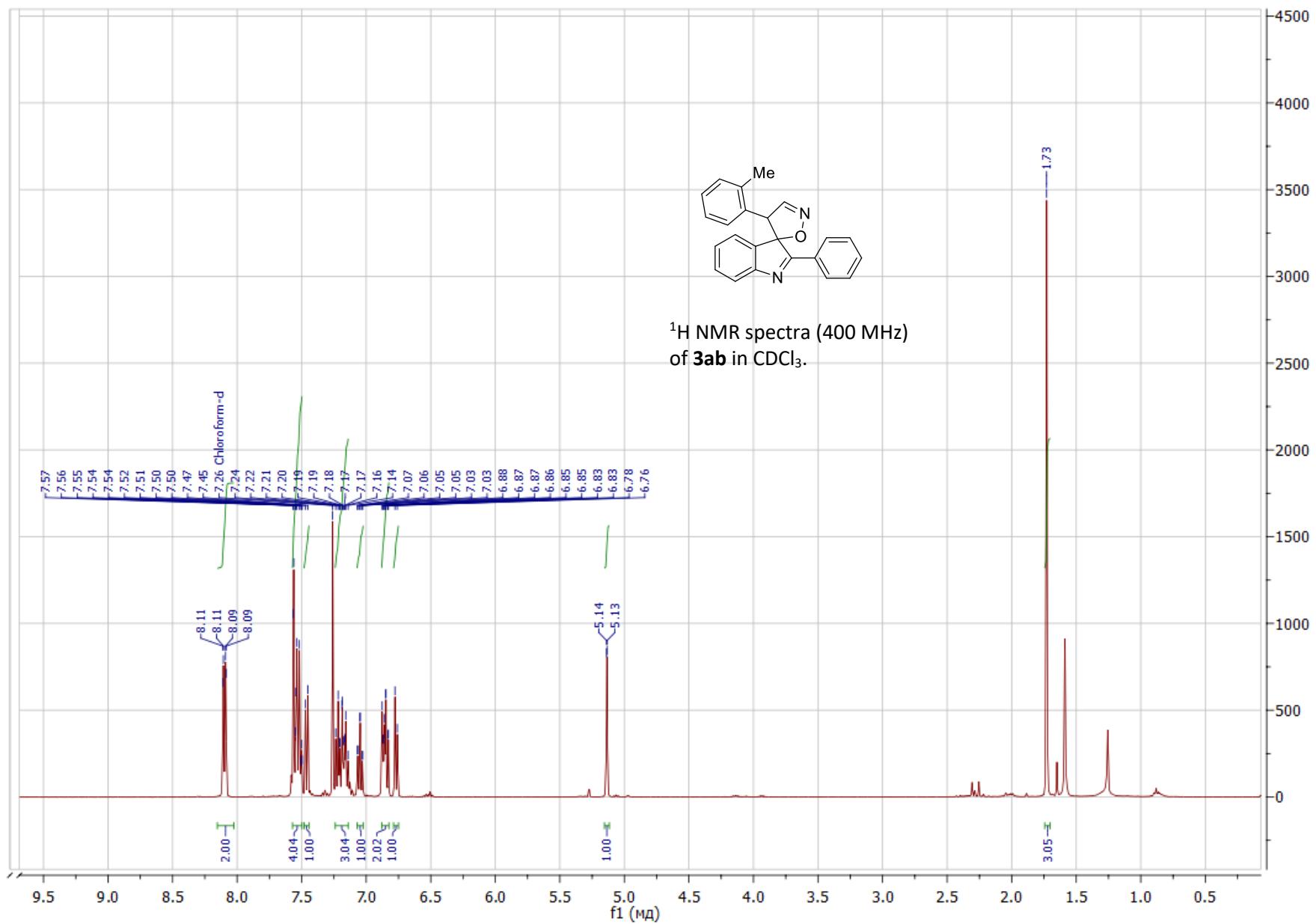
## Supporting Information

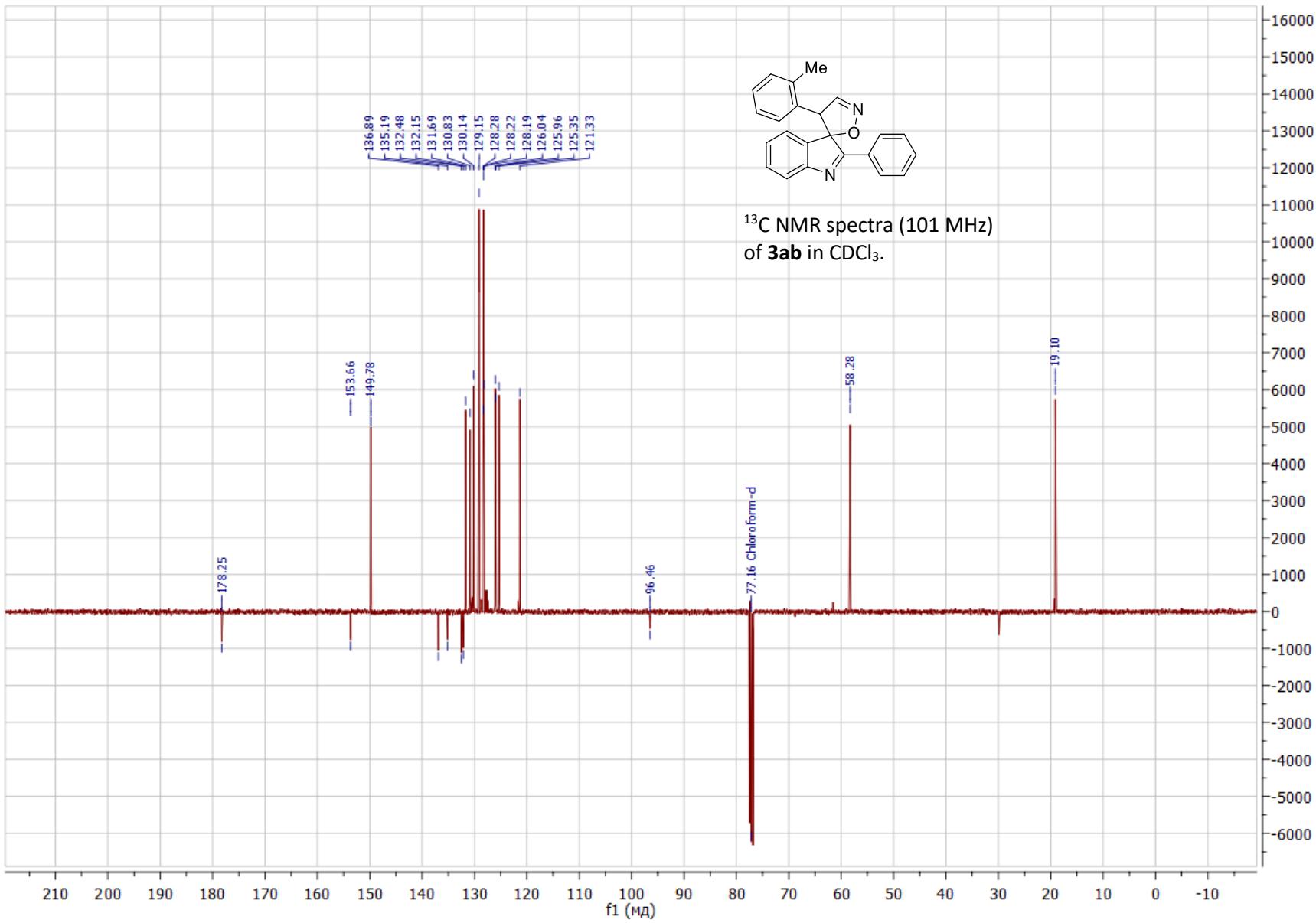
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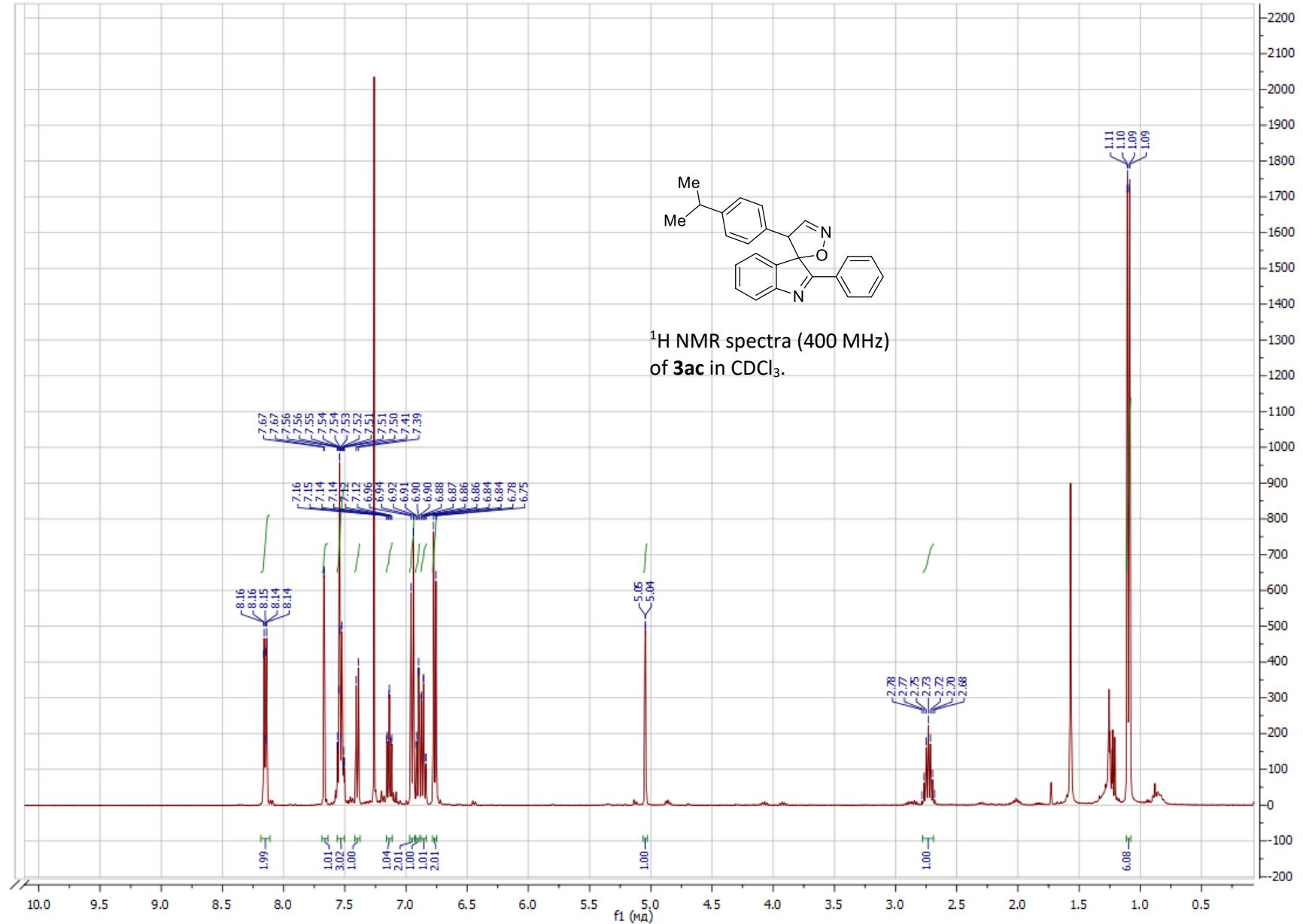
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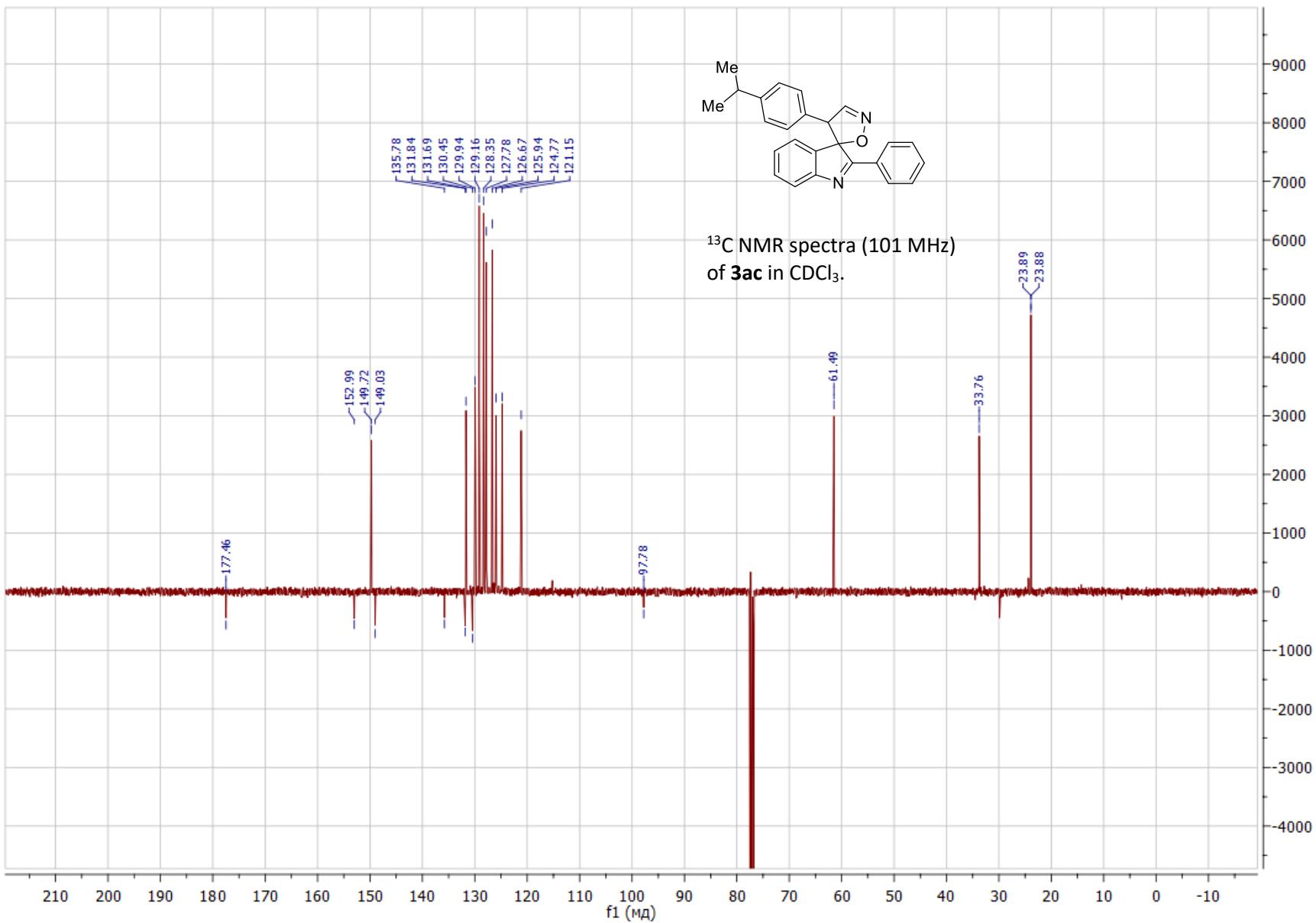
## <sup>1</sup>H and <sup>13</sup>C NMR Spectral Charts

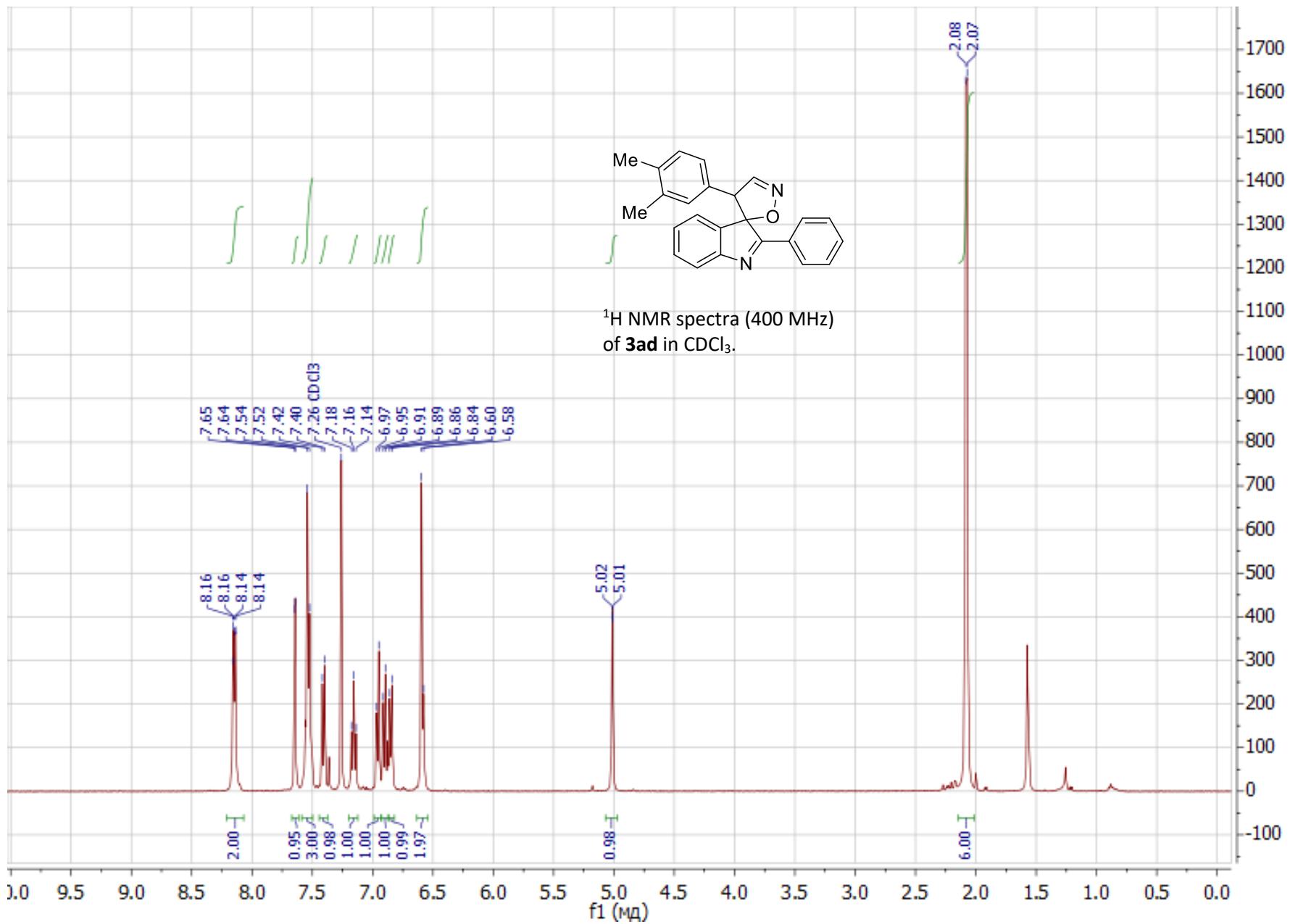


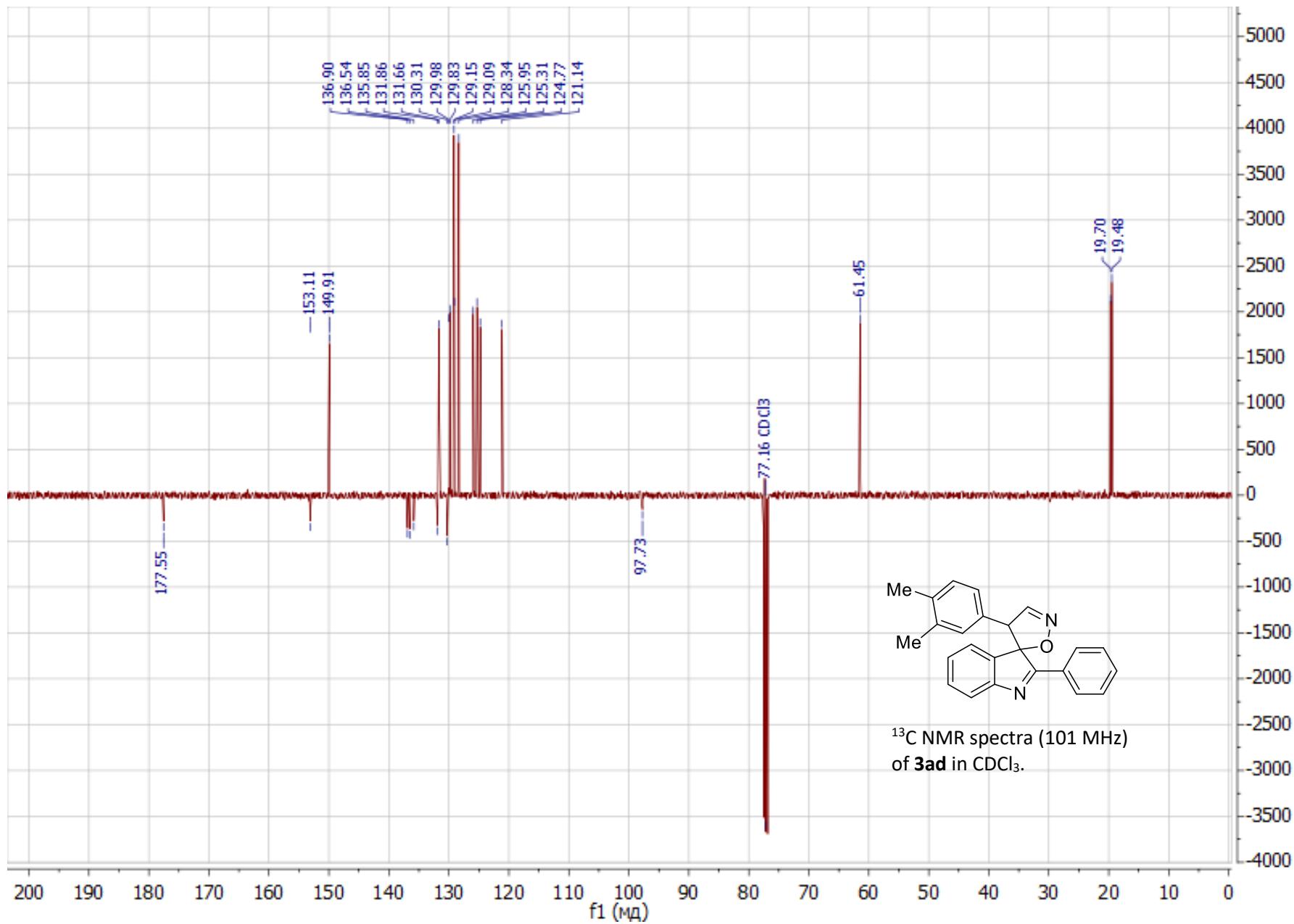


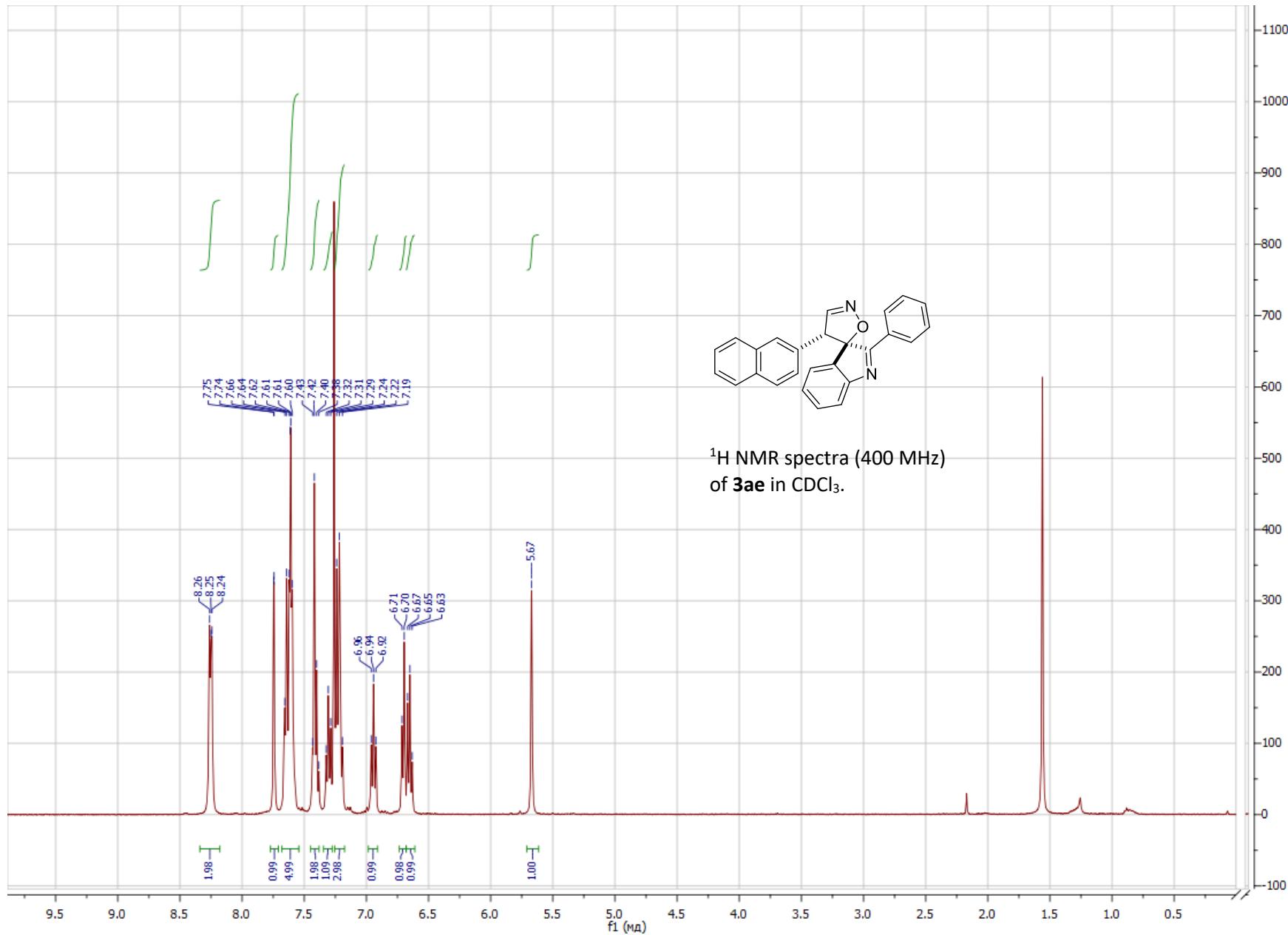


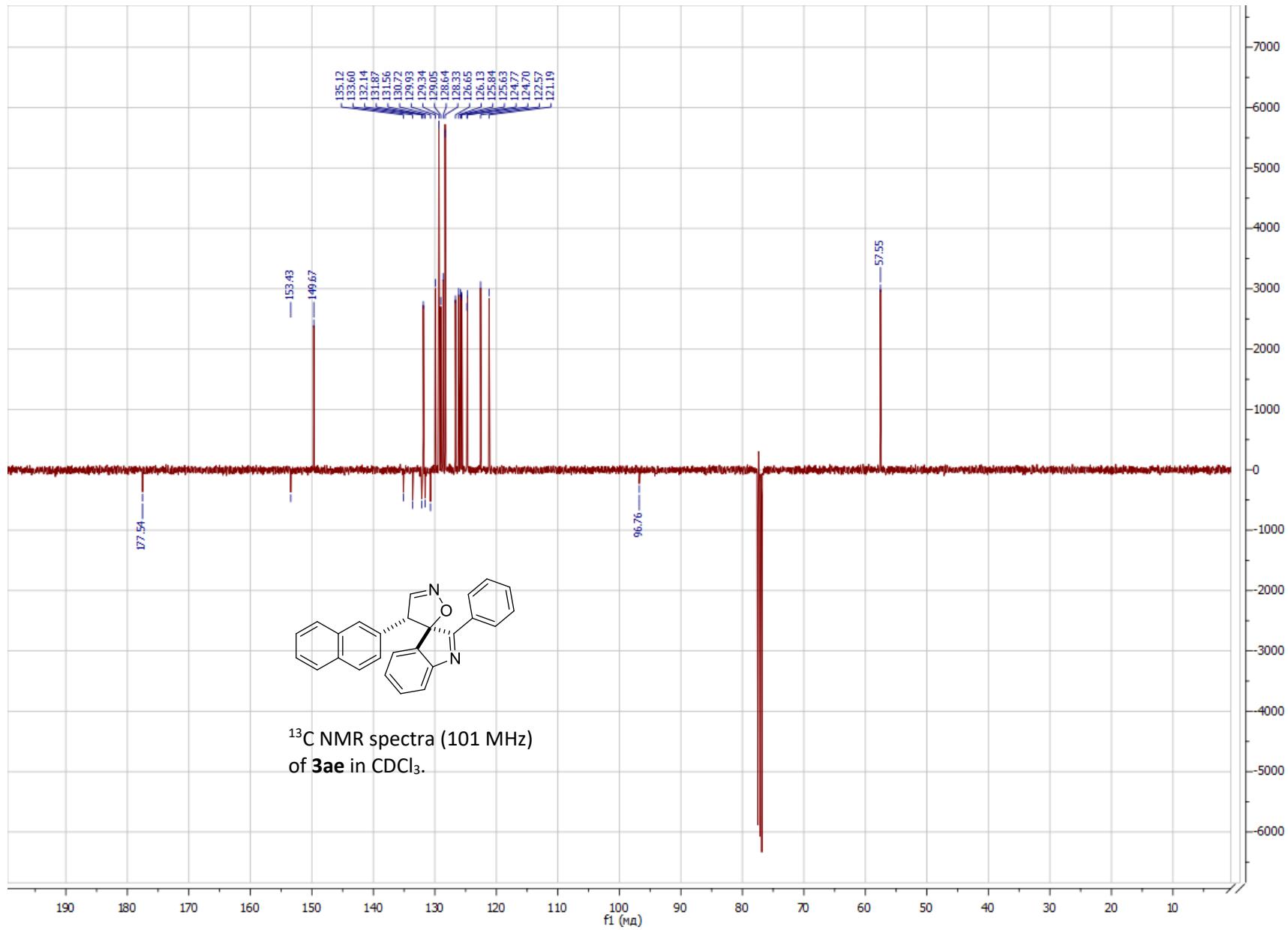


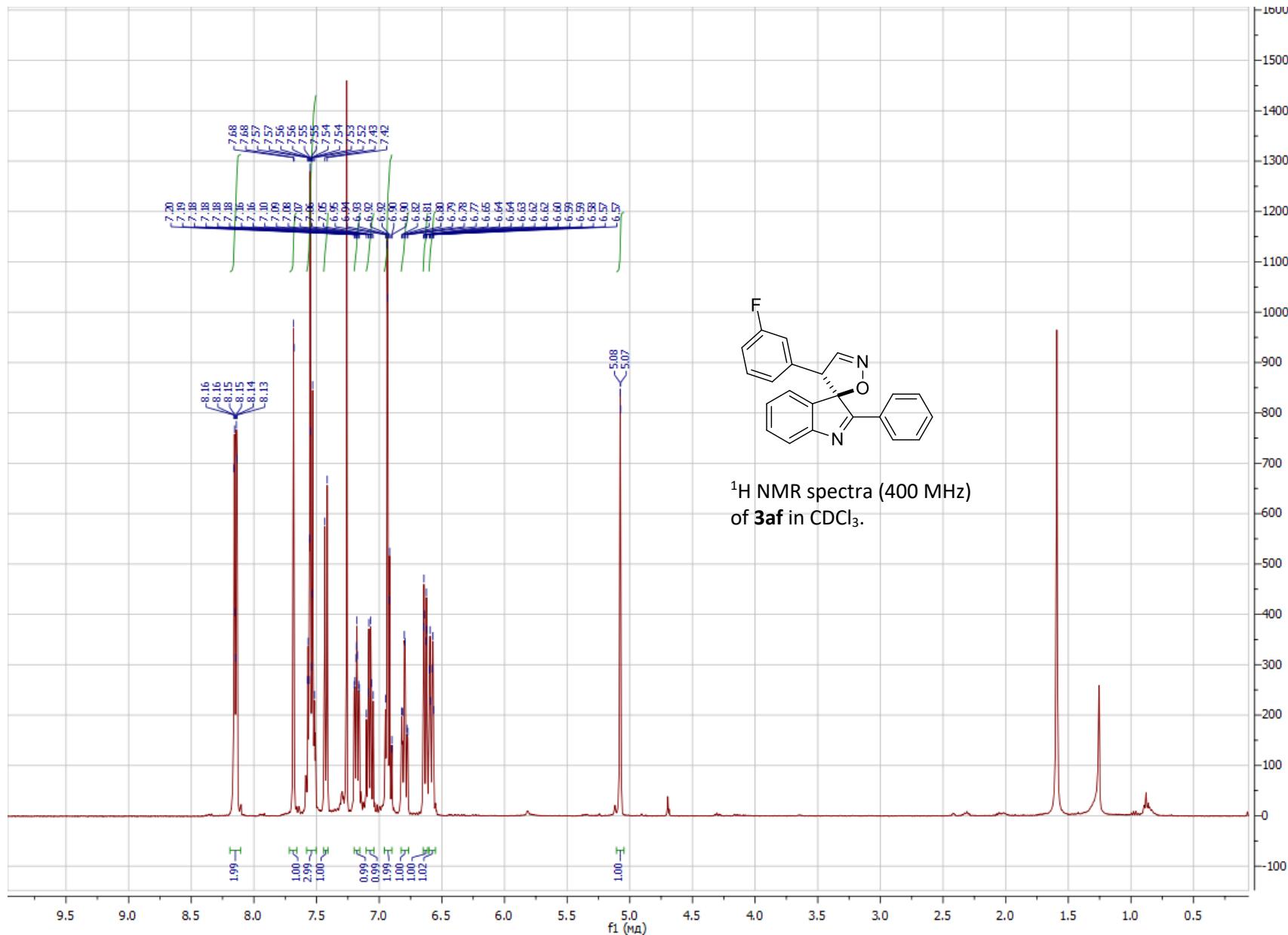


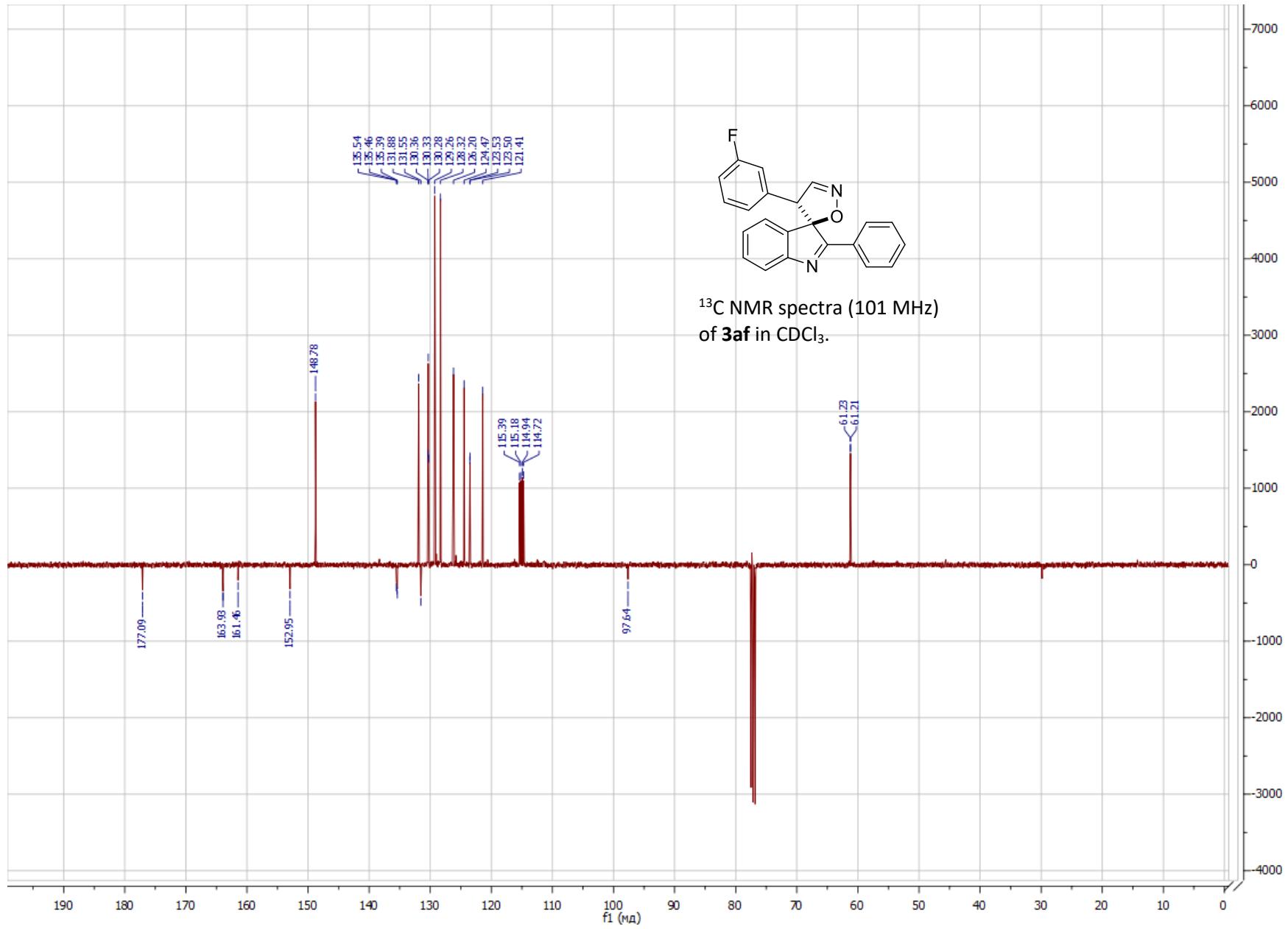


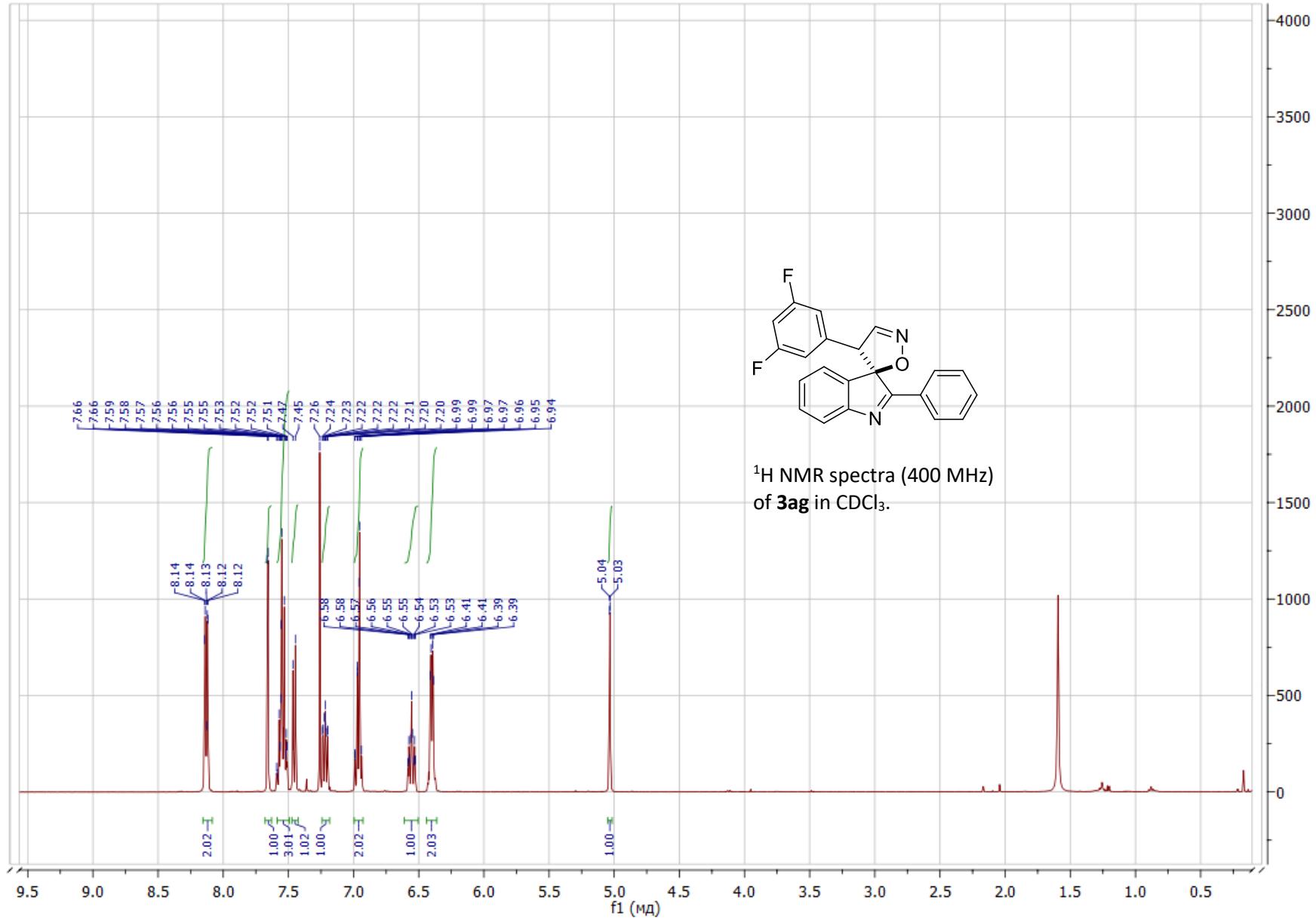


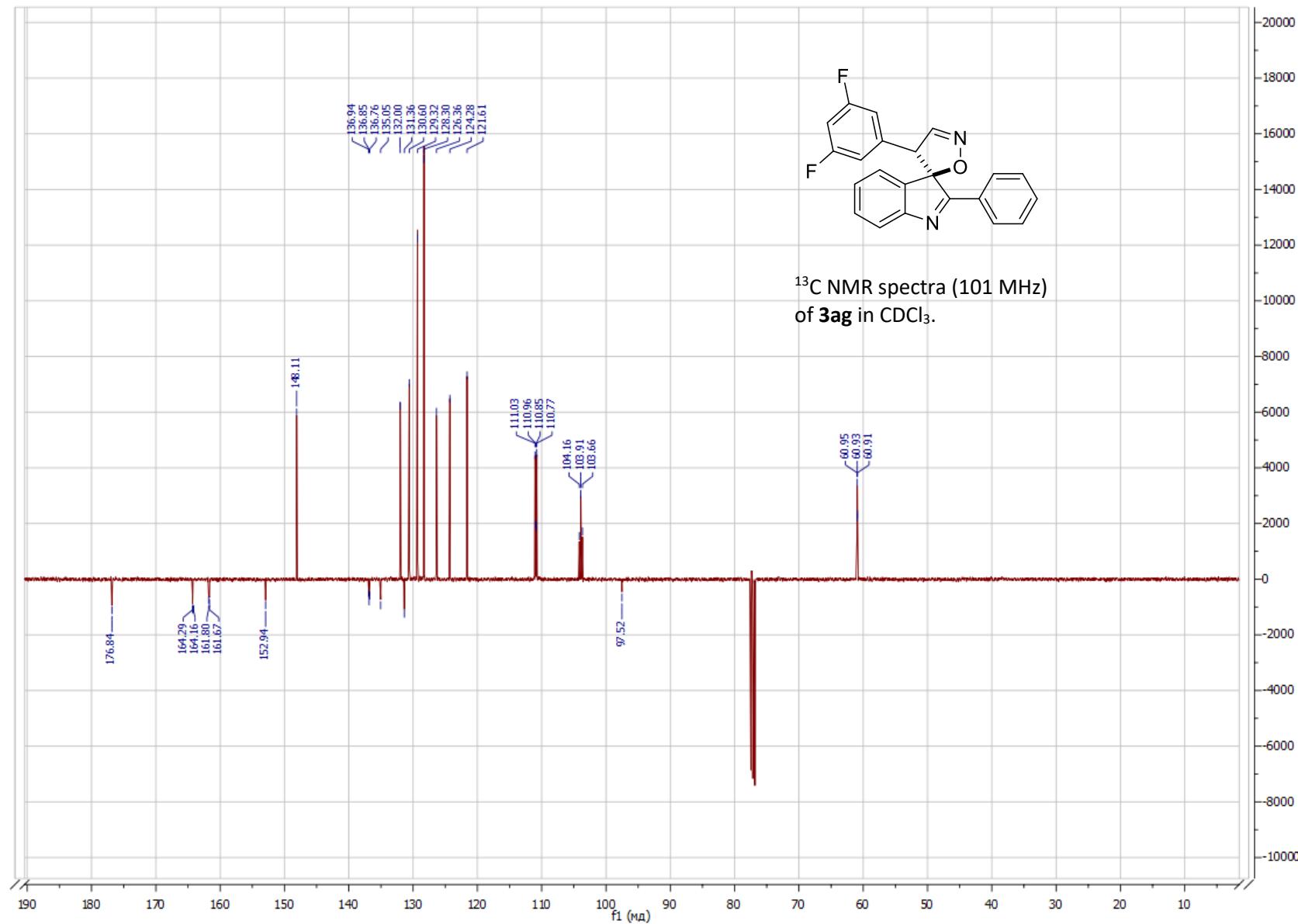


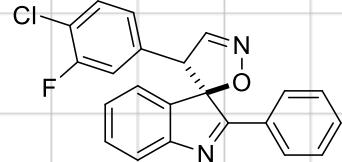
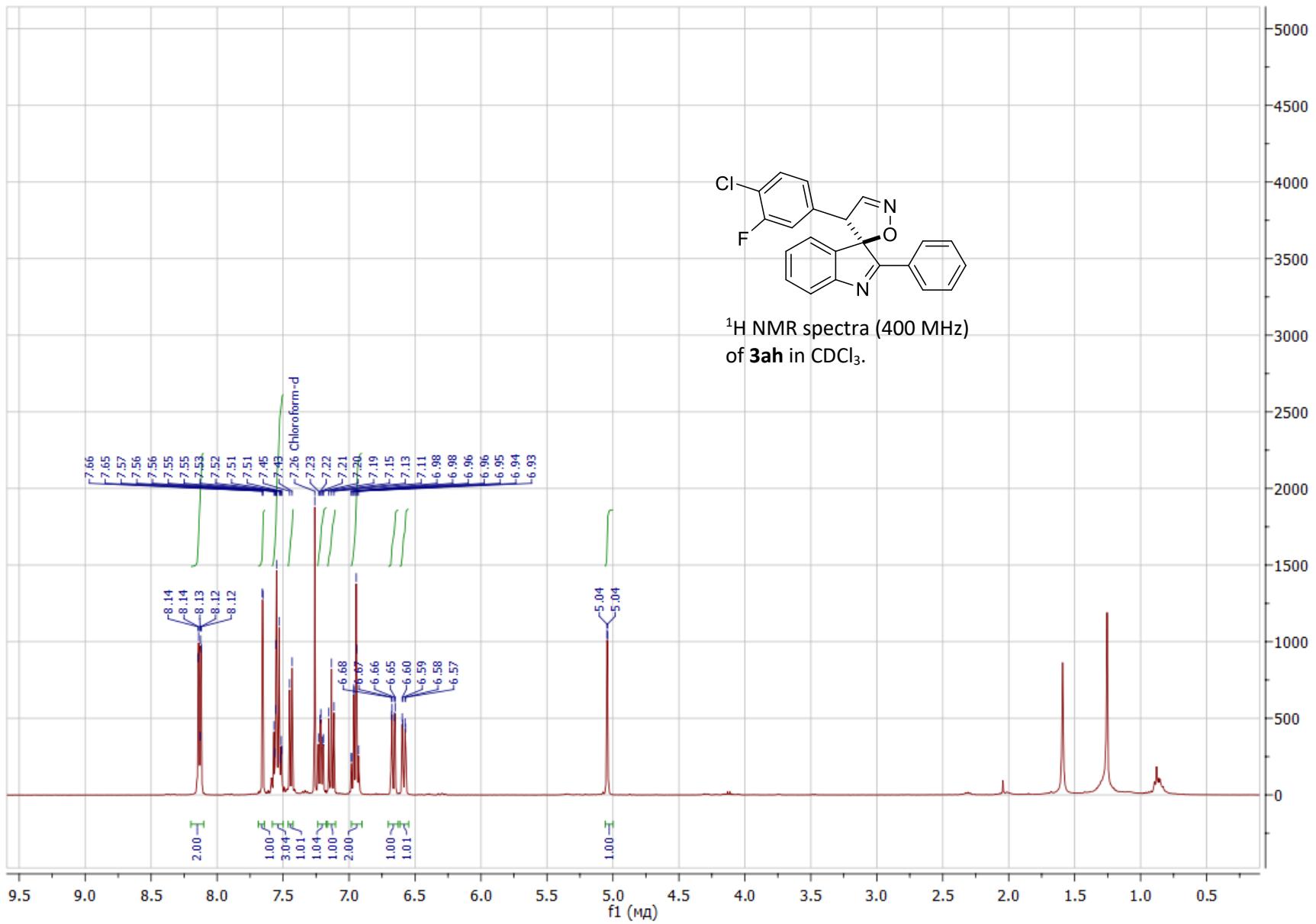




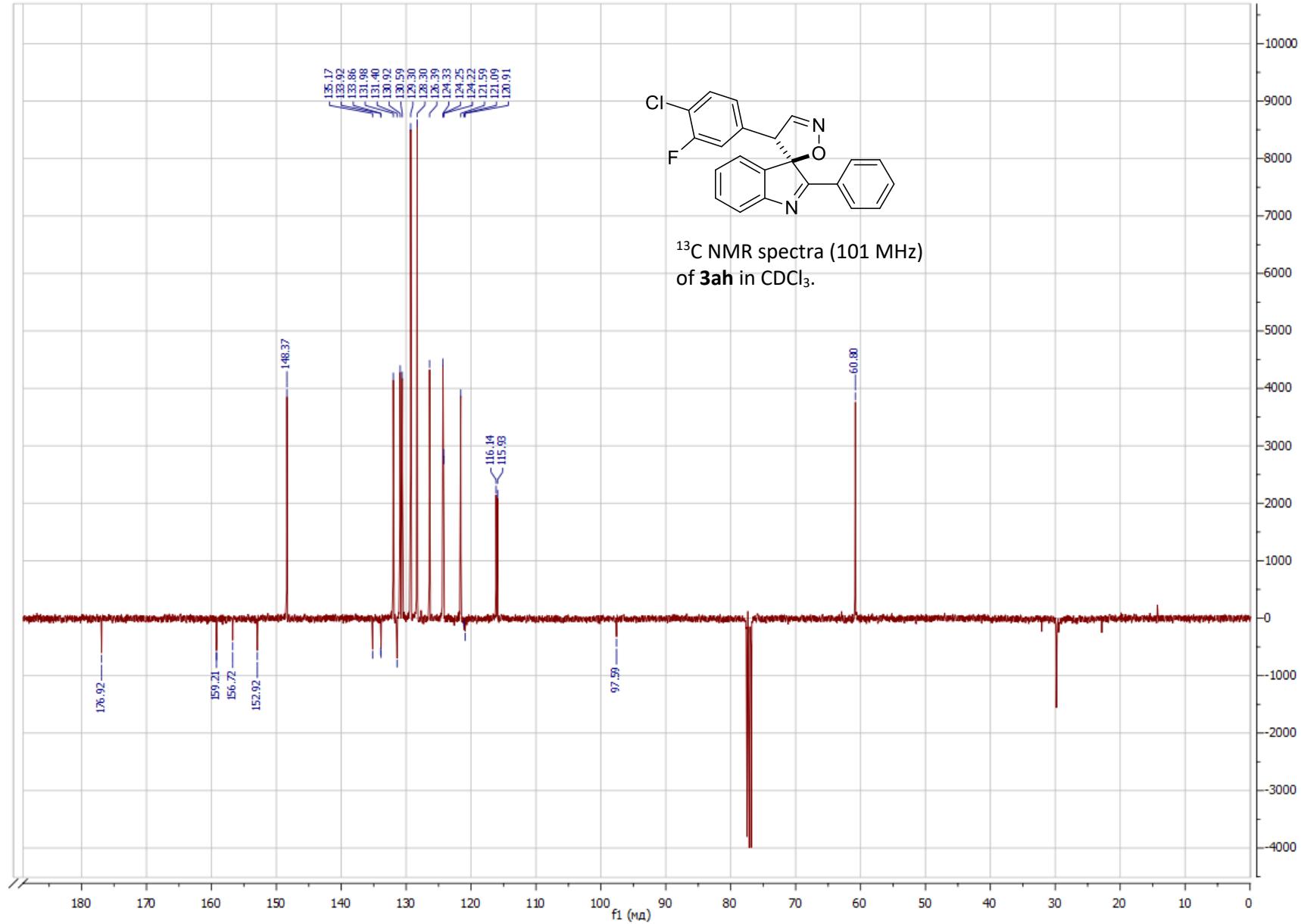


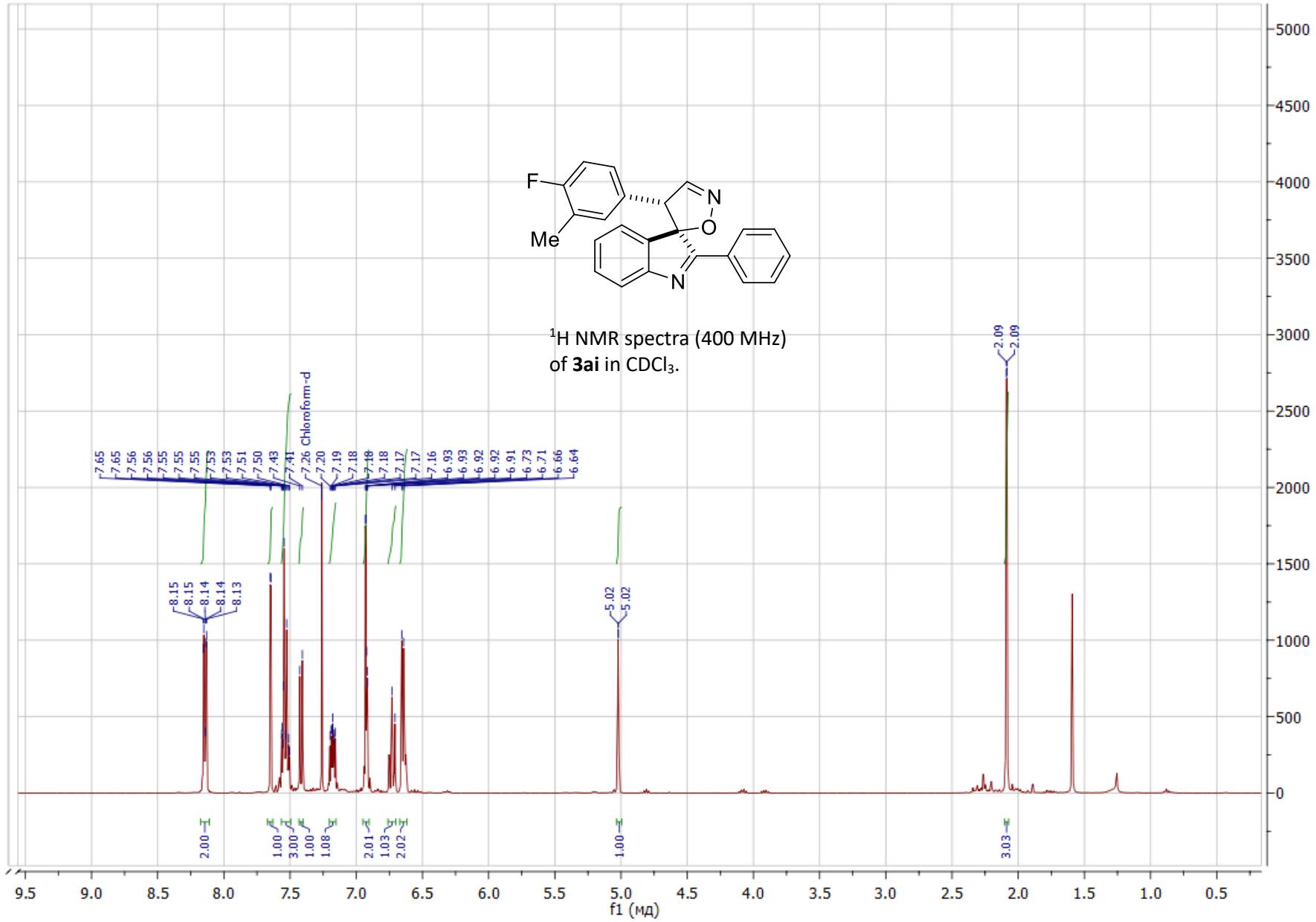


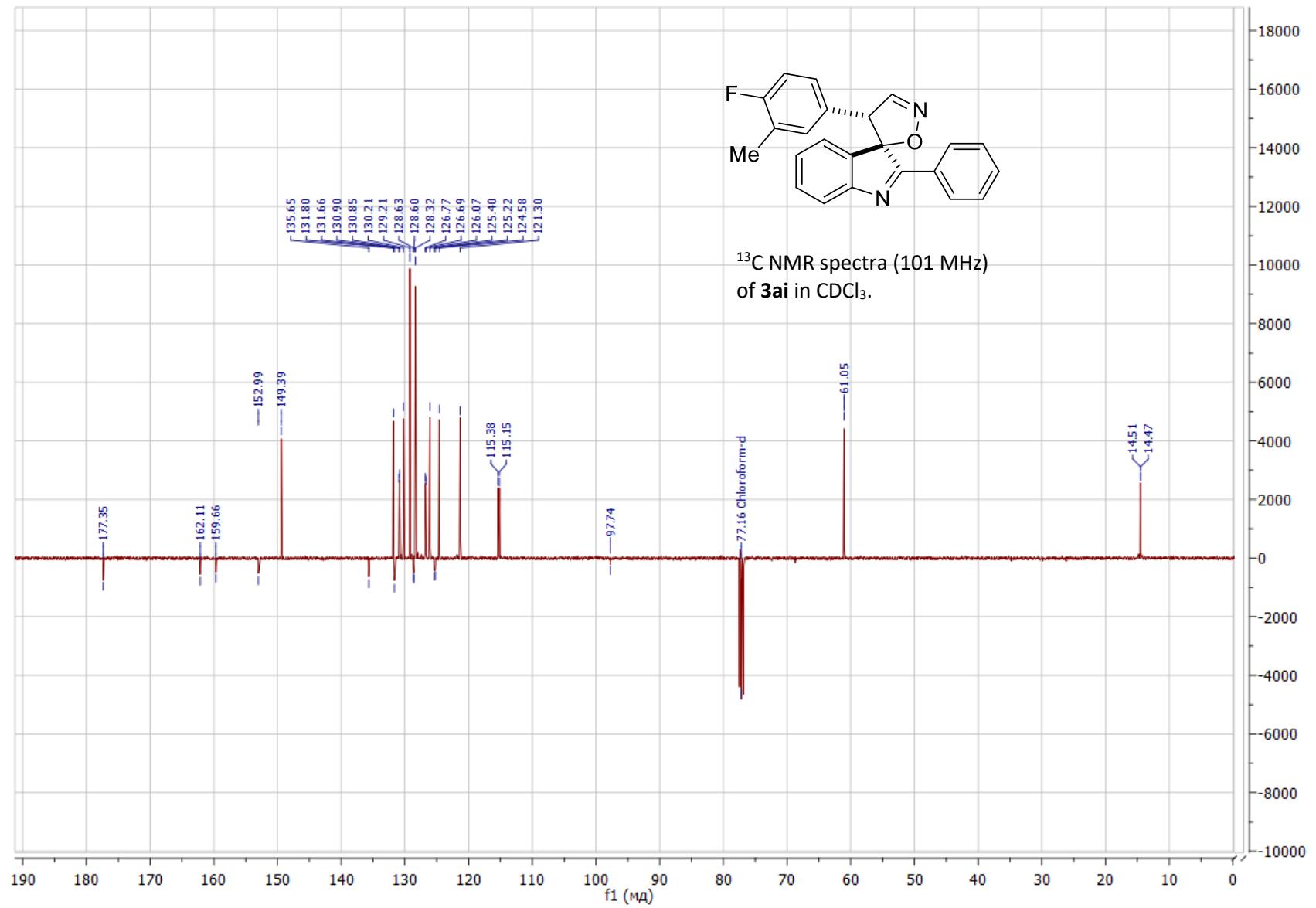


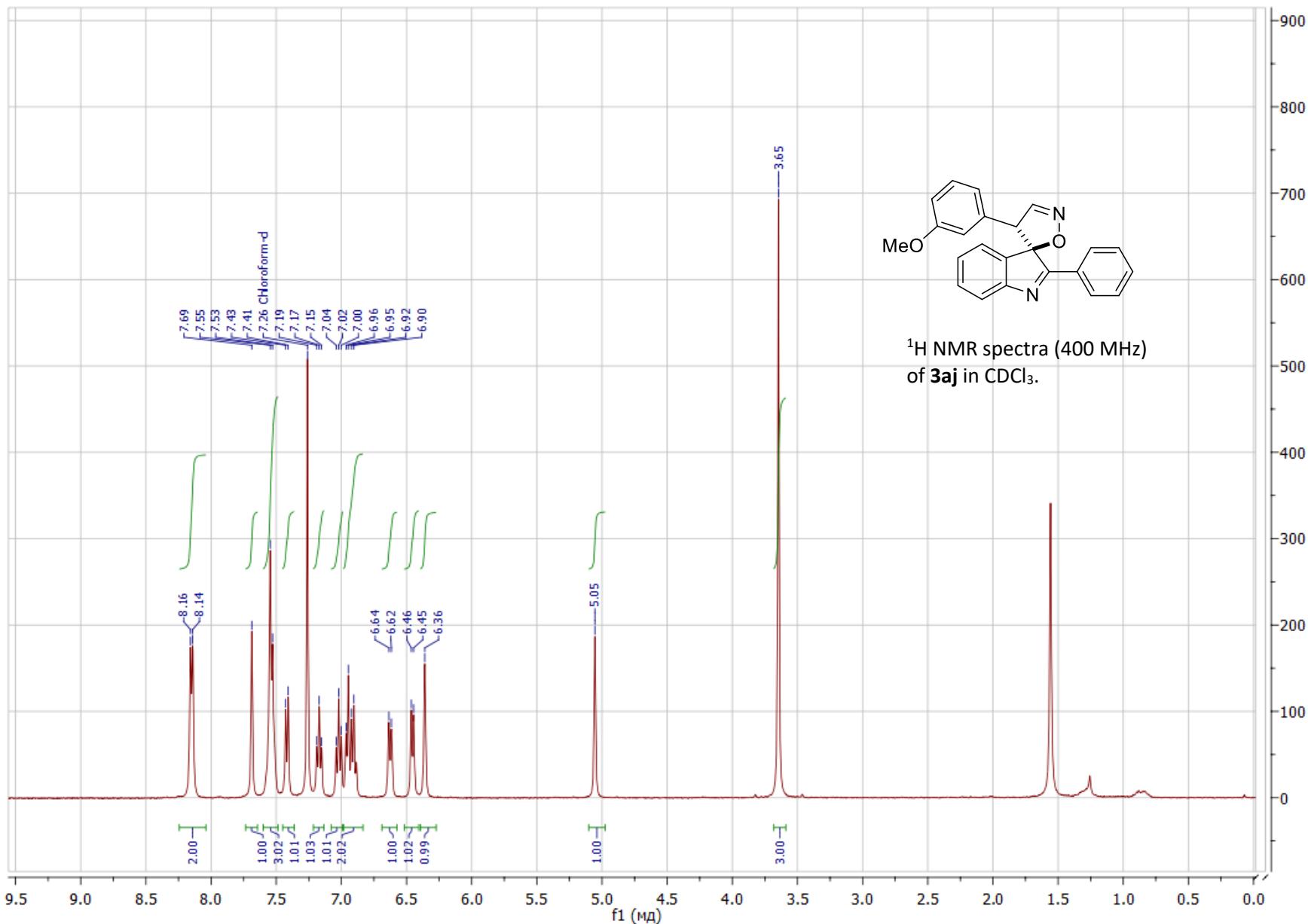


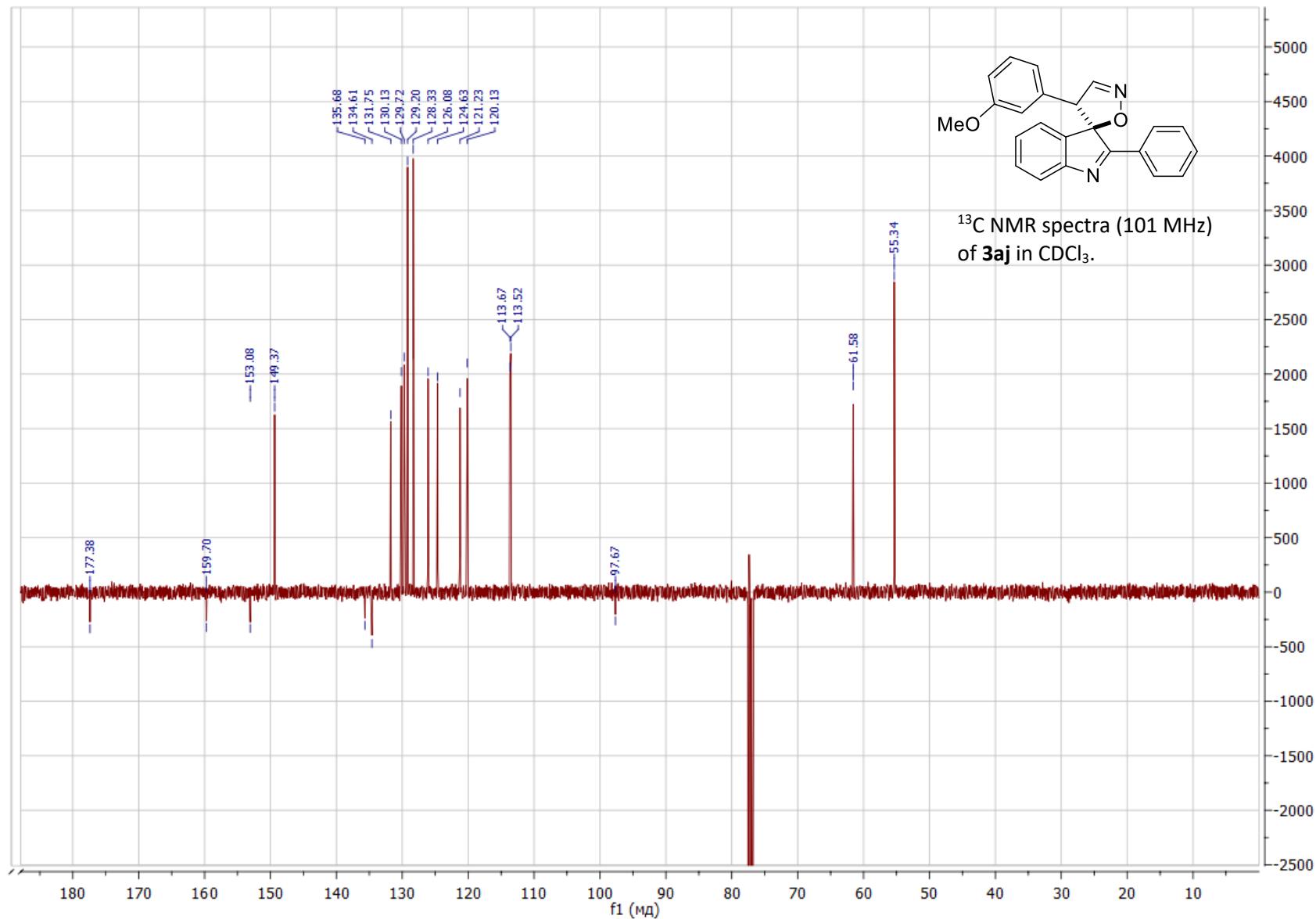
<sup>1</sup>H NMR spectra (400 MHz) of **3ah** in CDCl<sub>3</sub>.

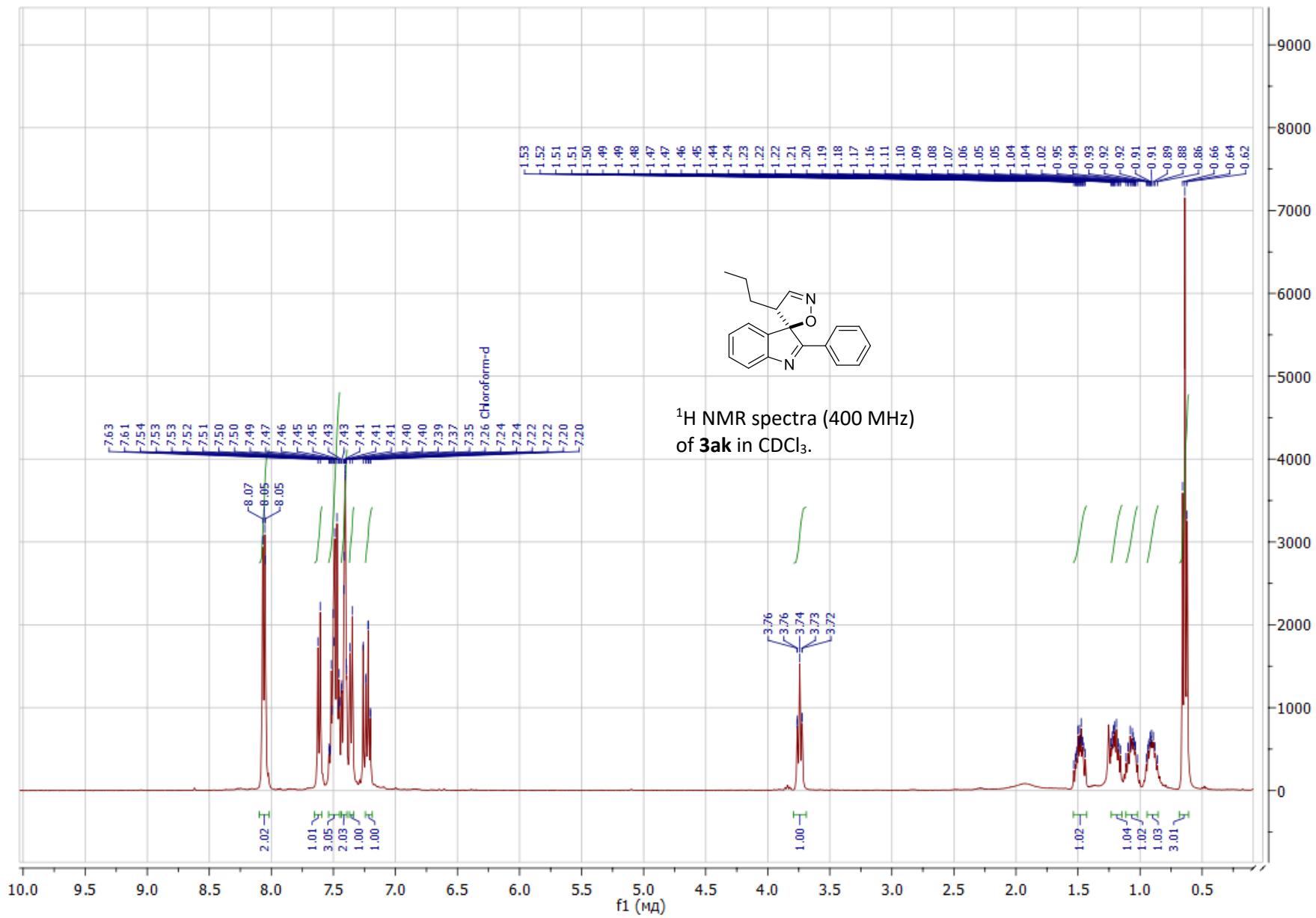


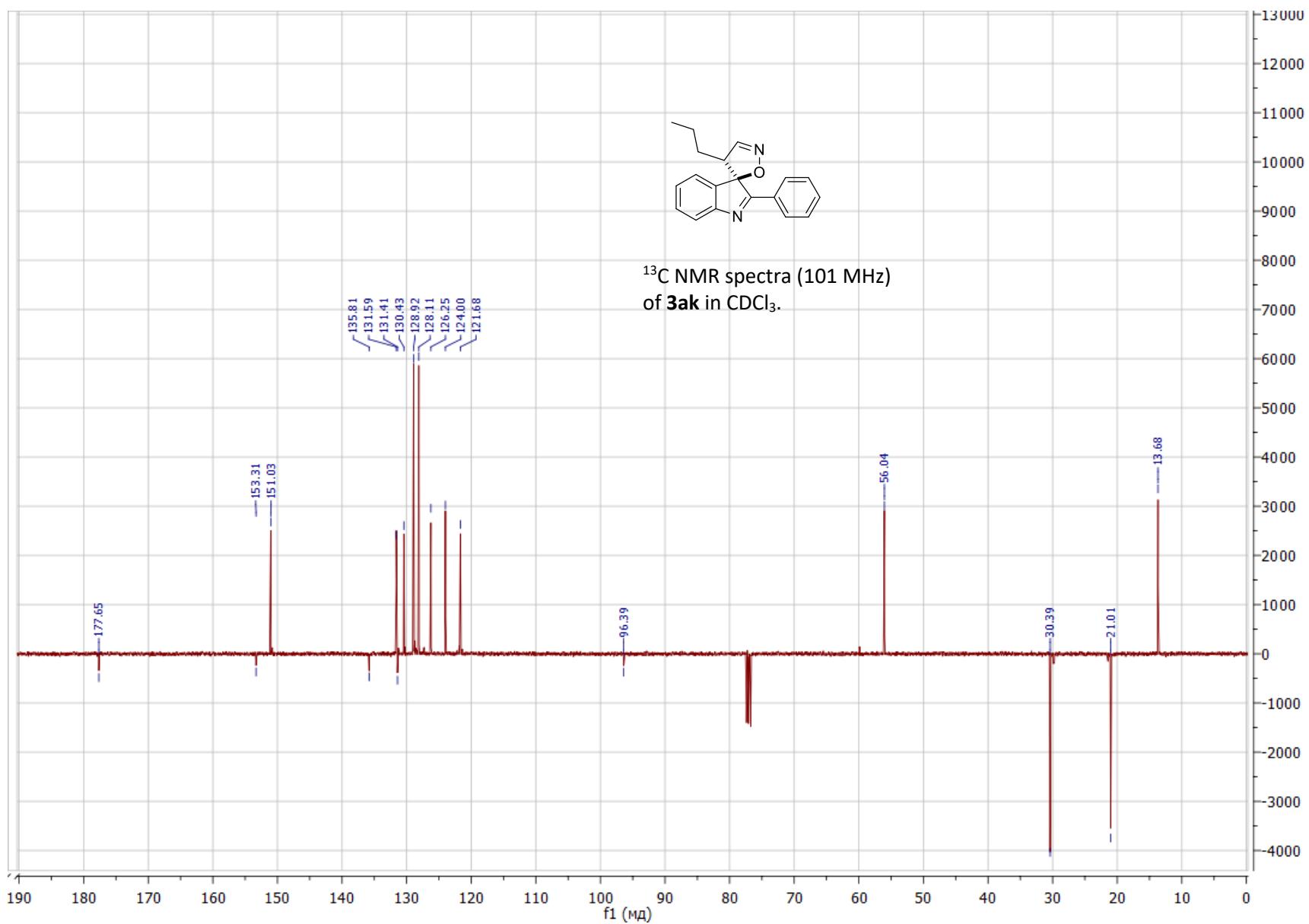


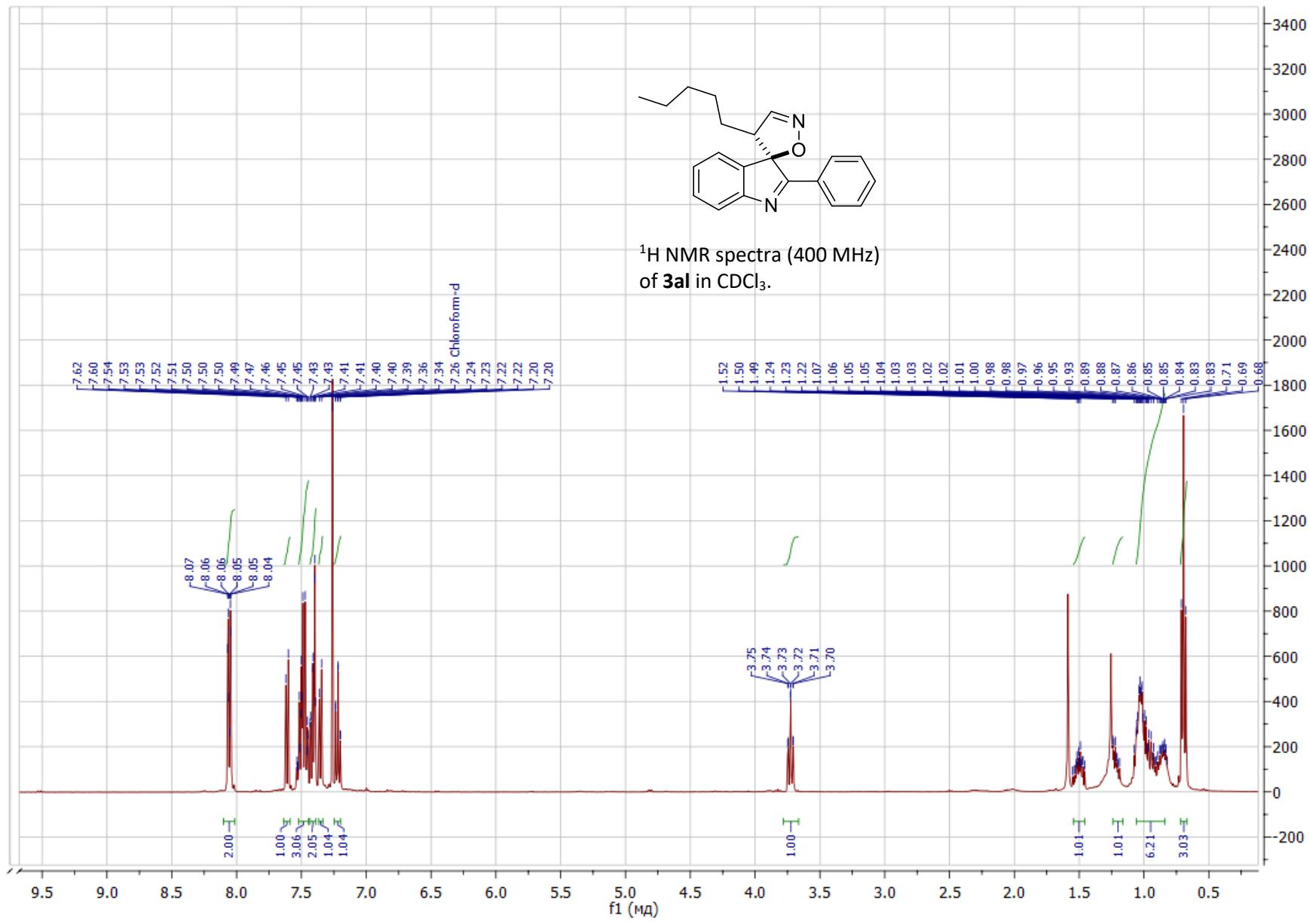


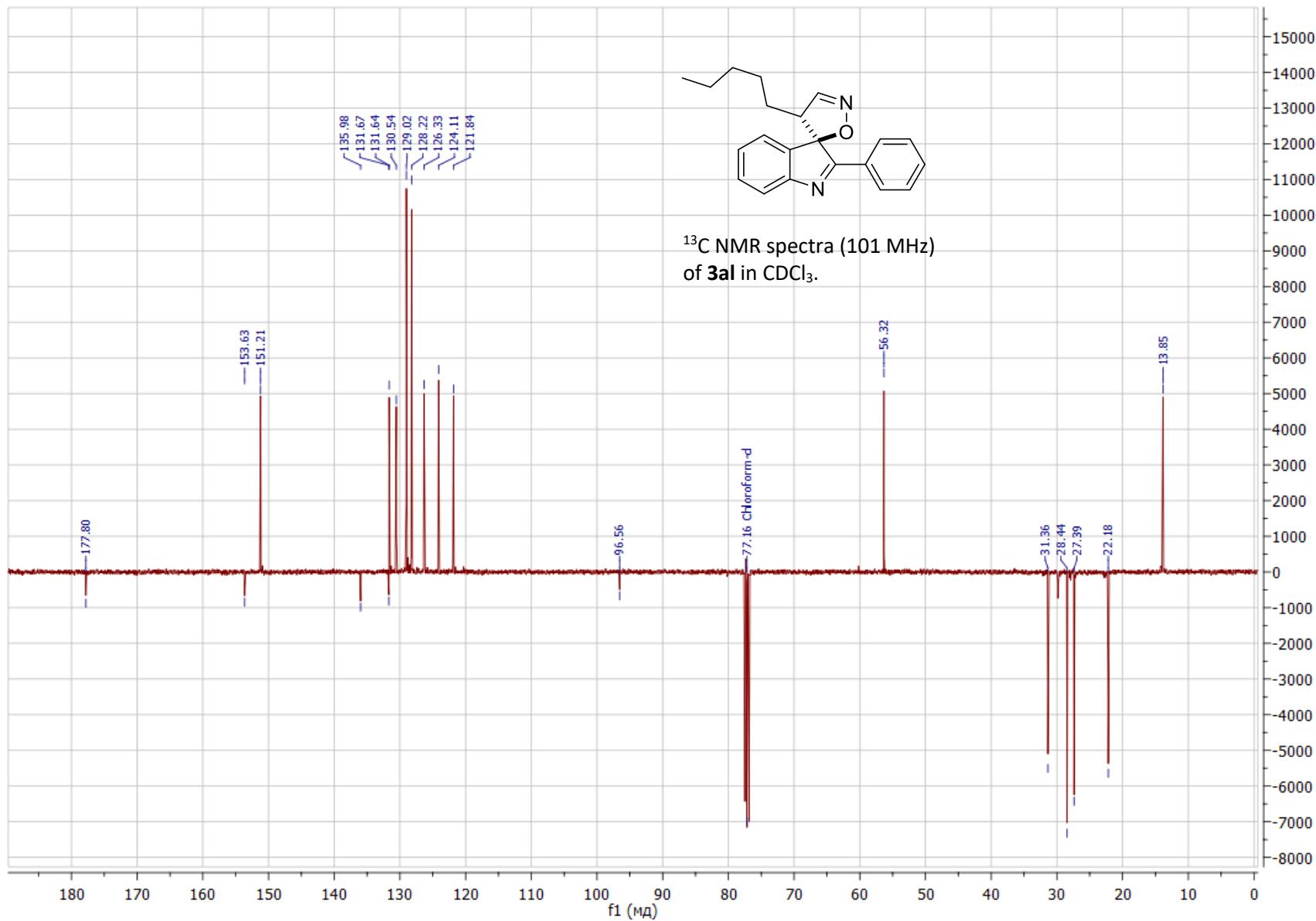


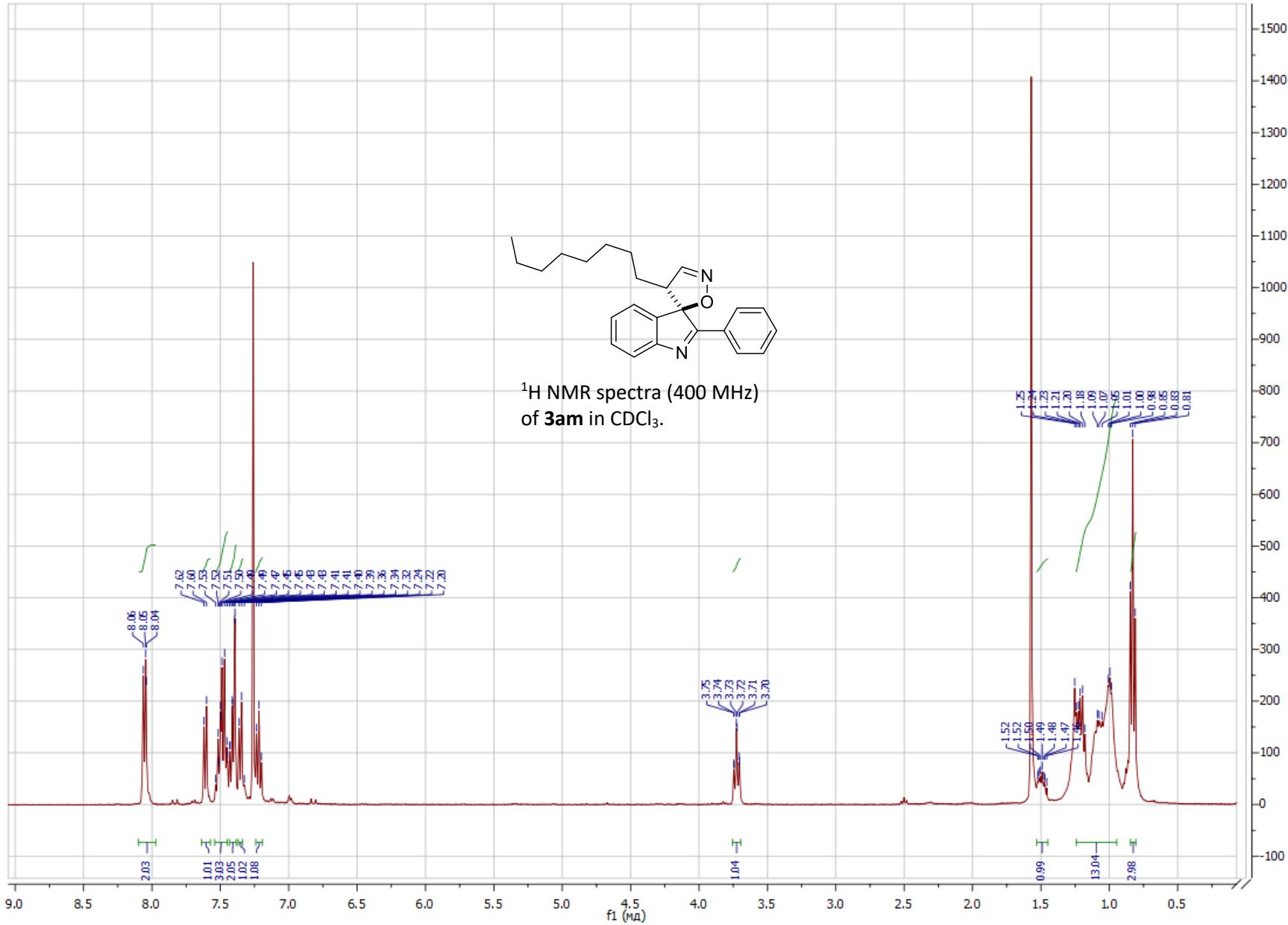


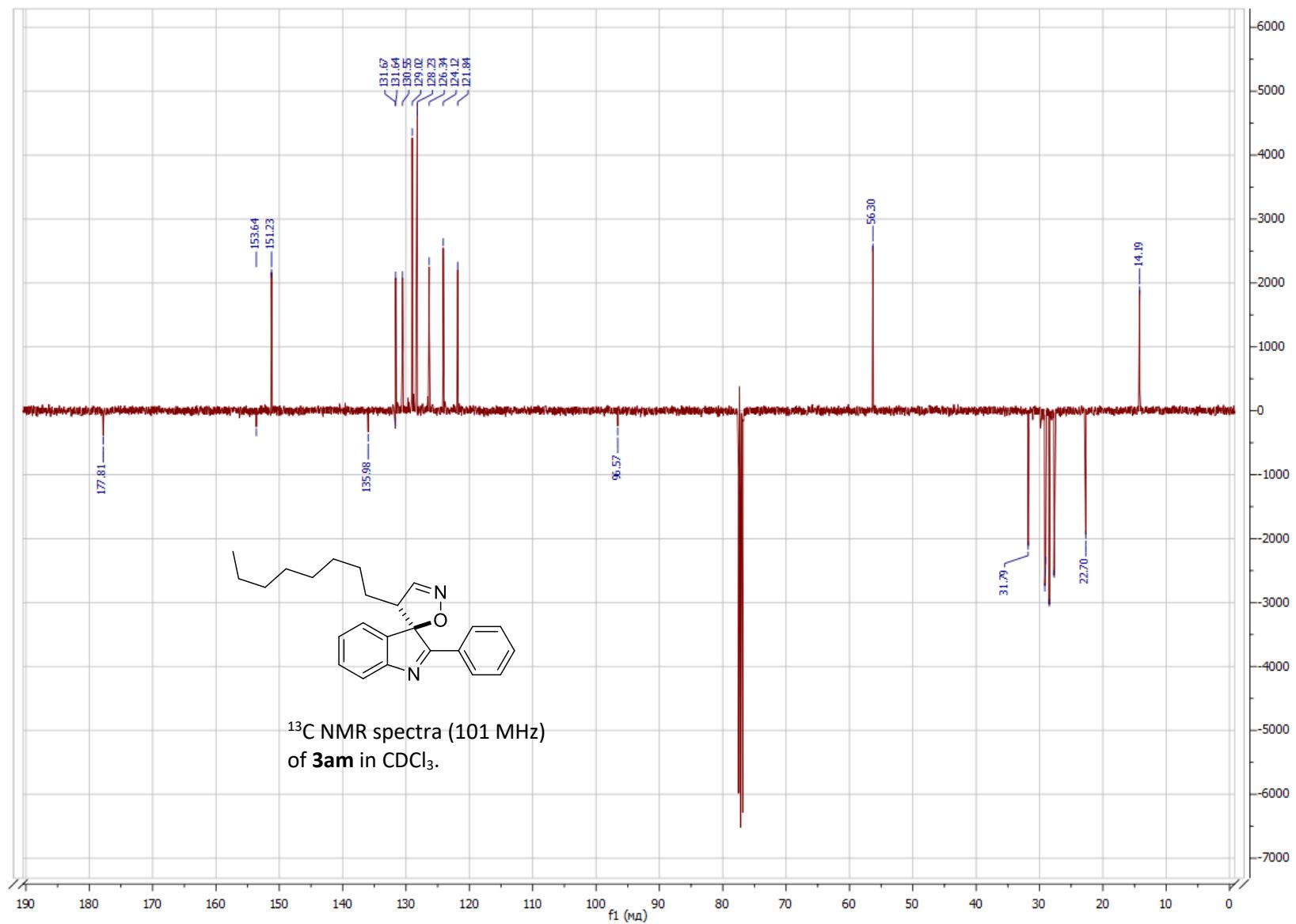


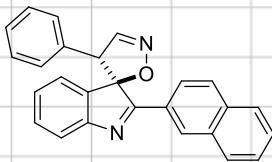
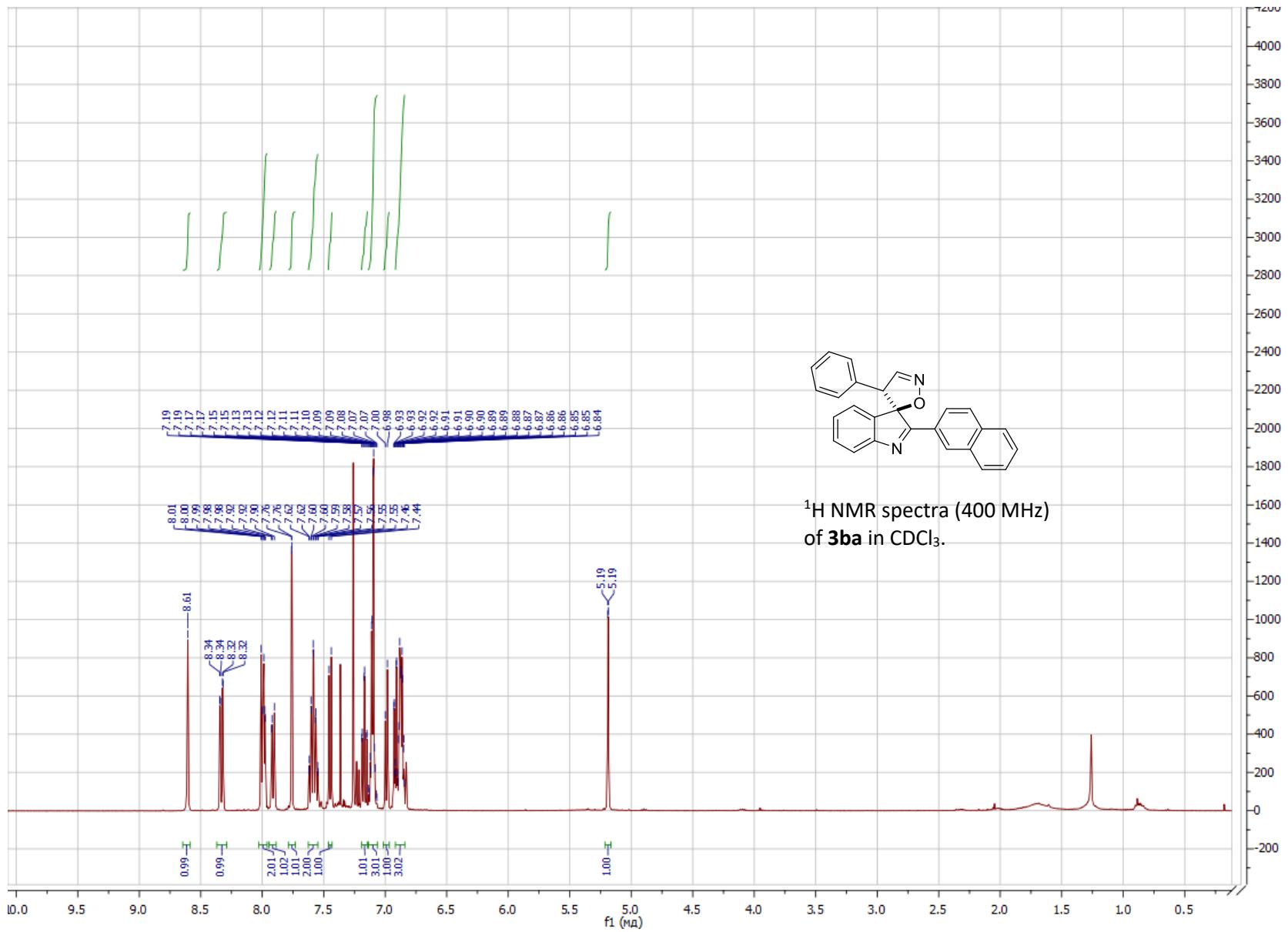




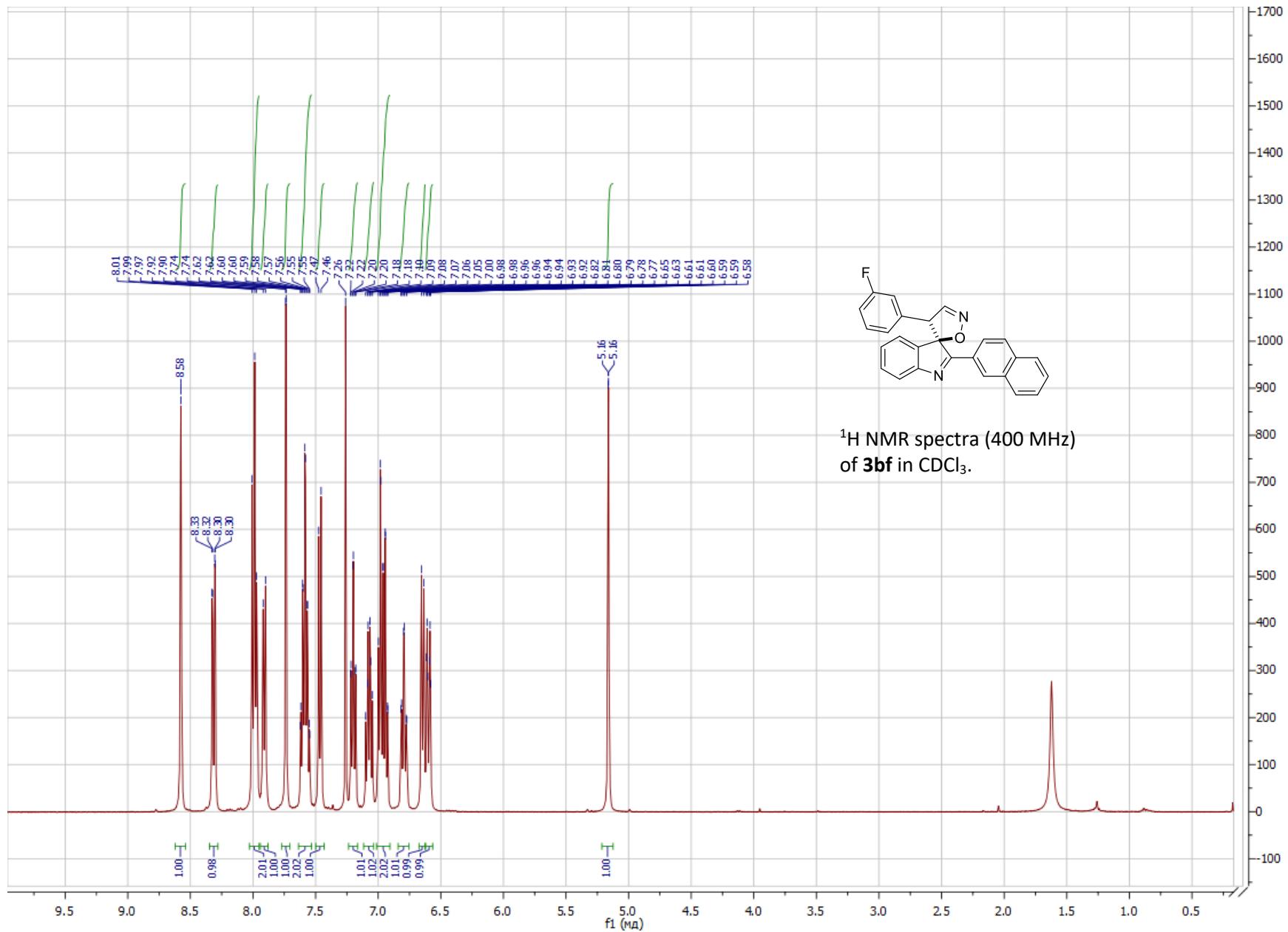


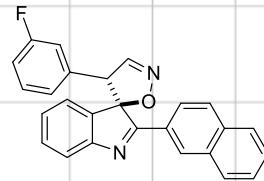
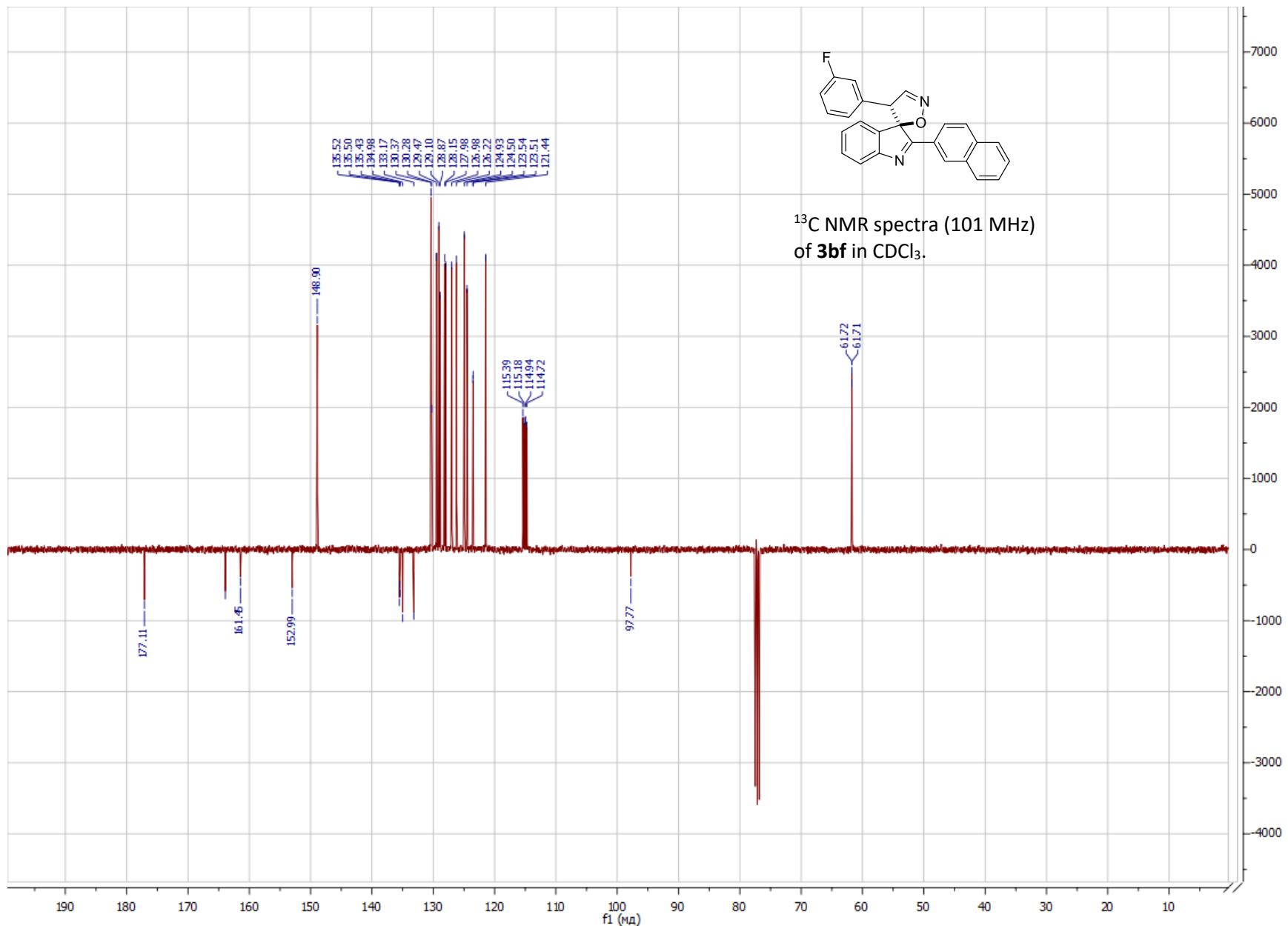


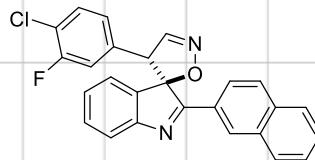
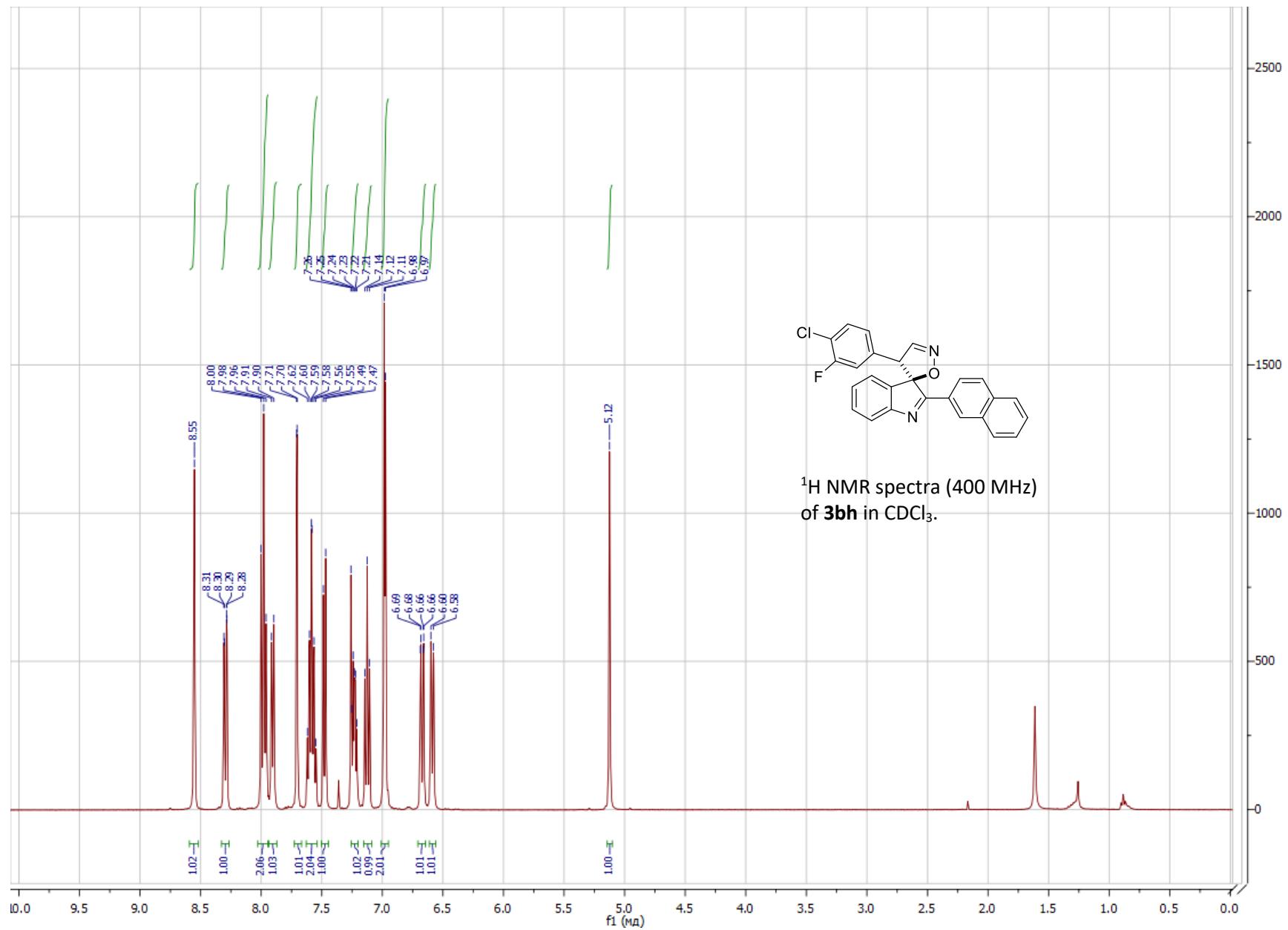




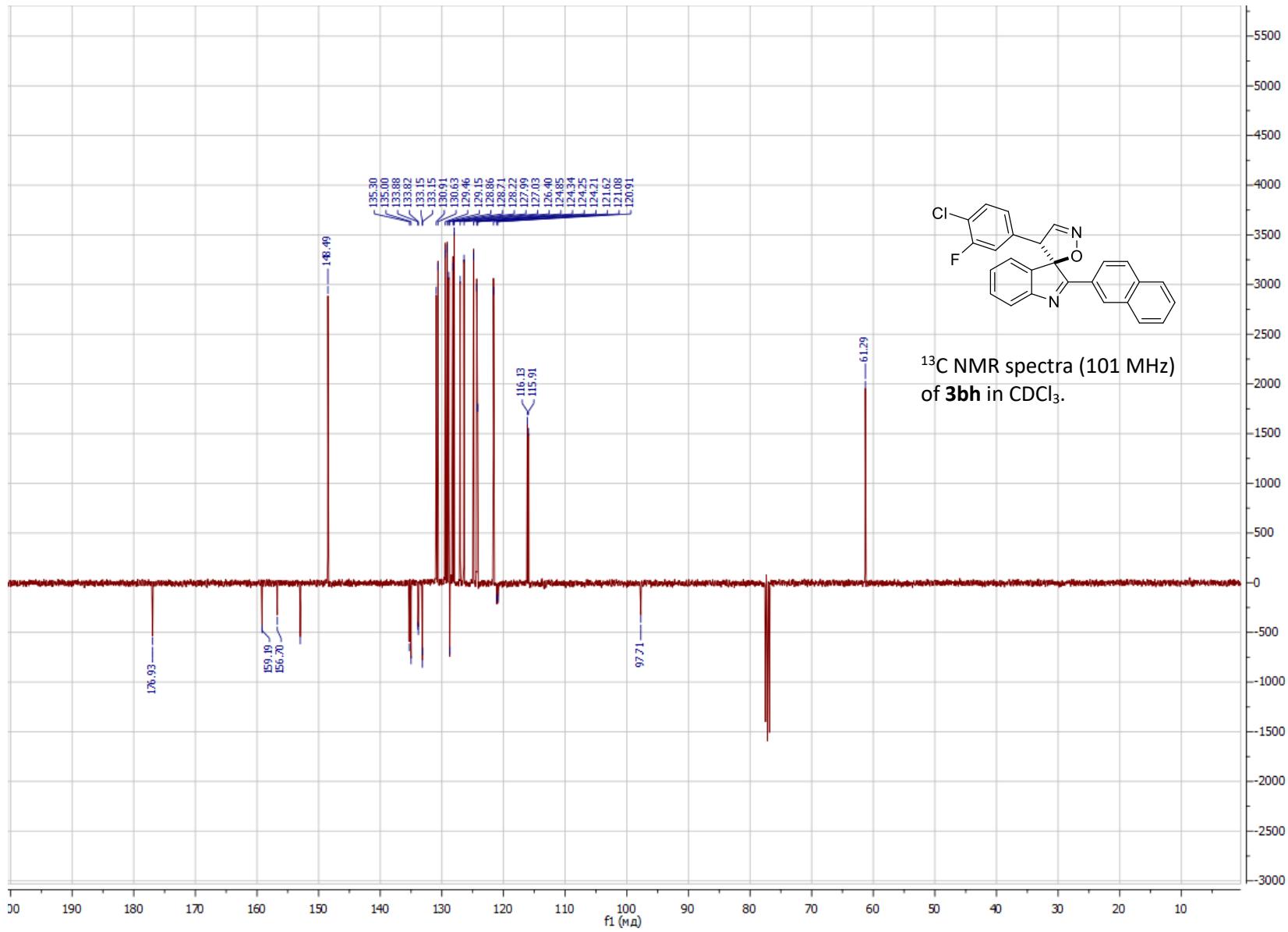
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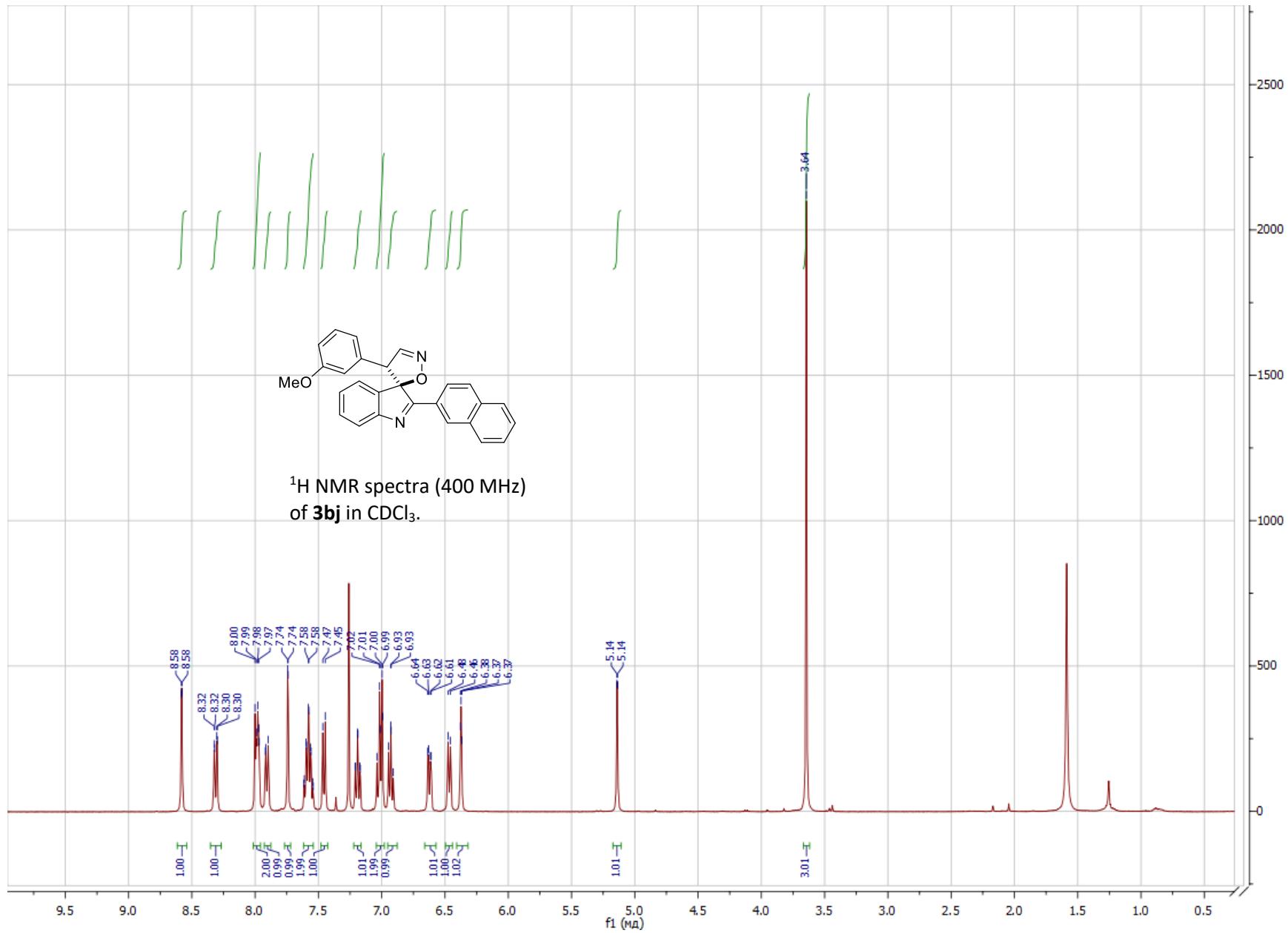


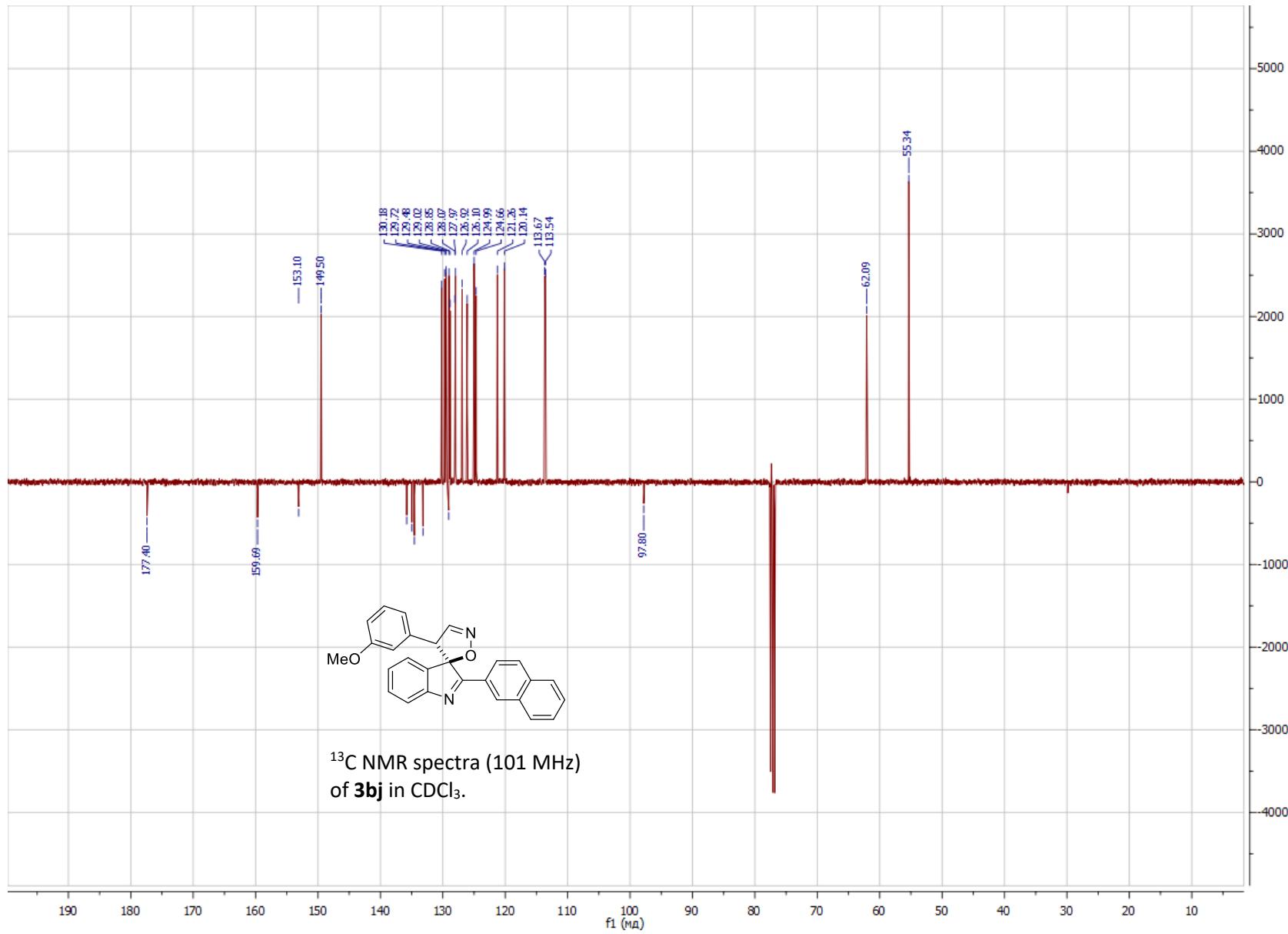


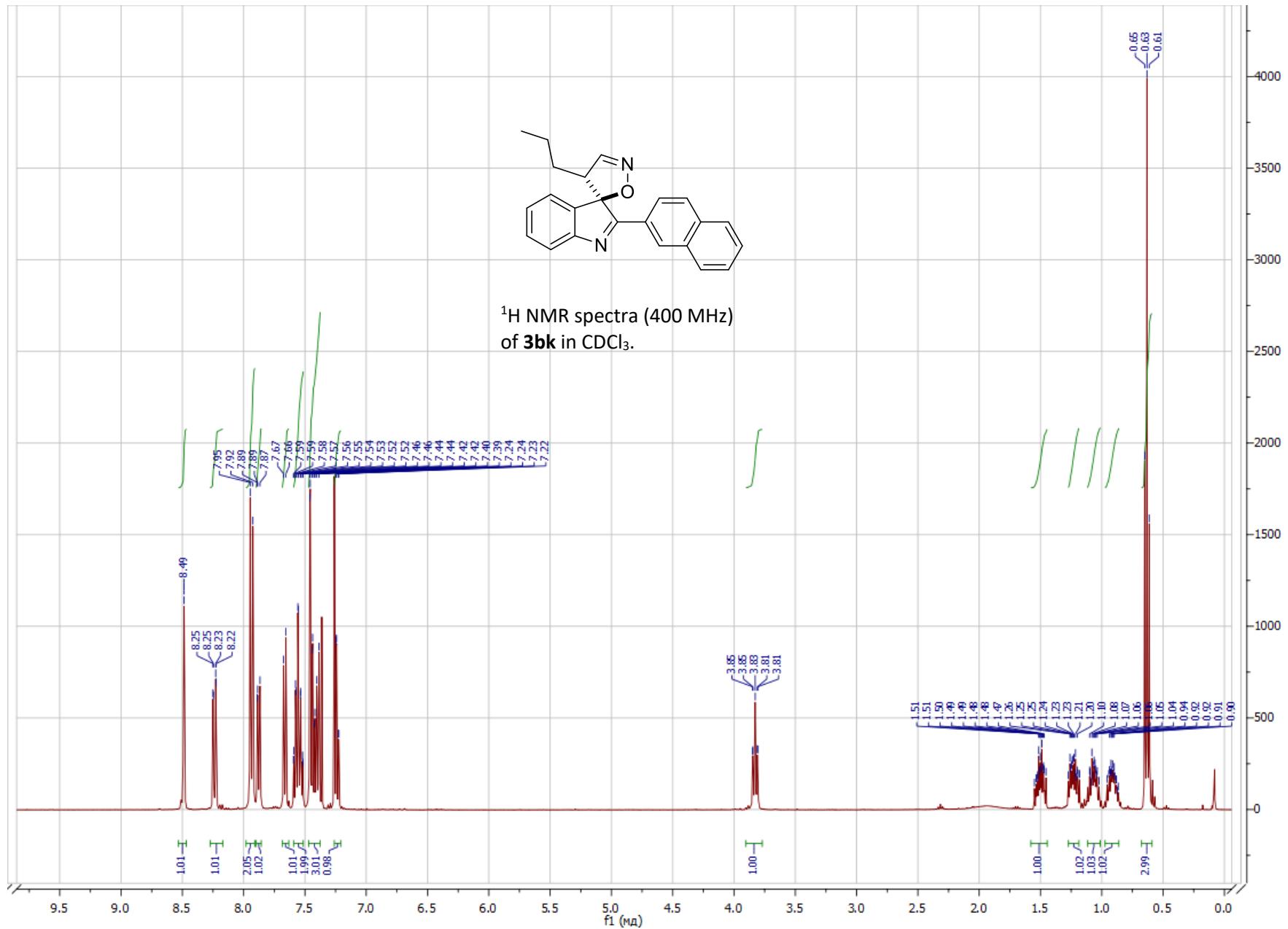


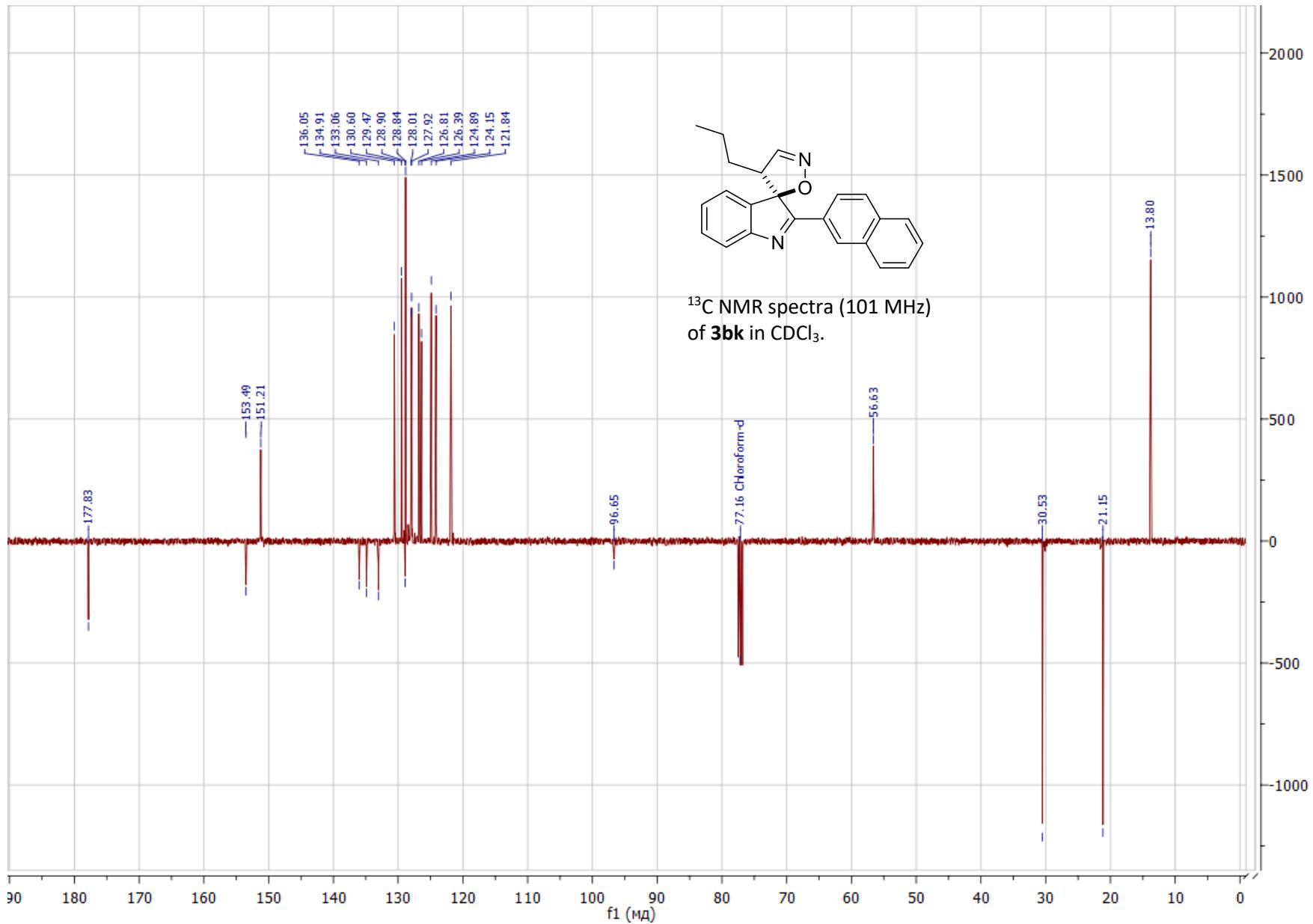
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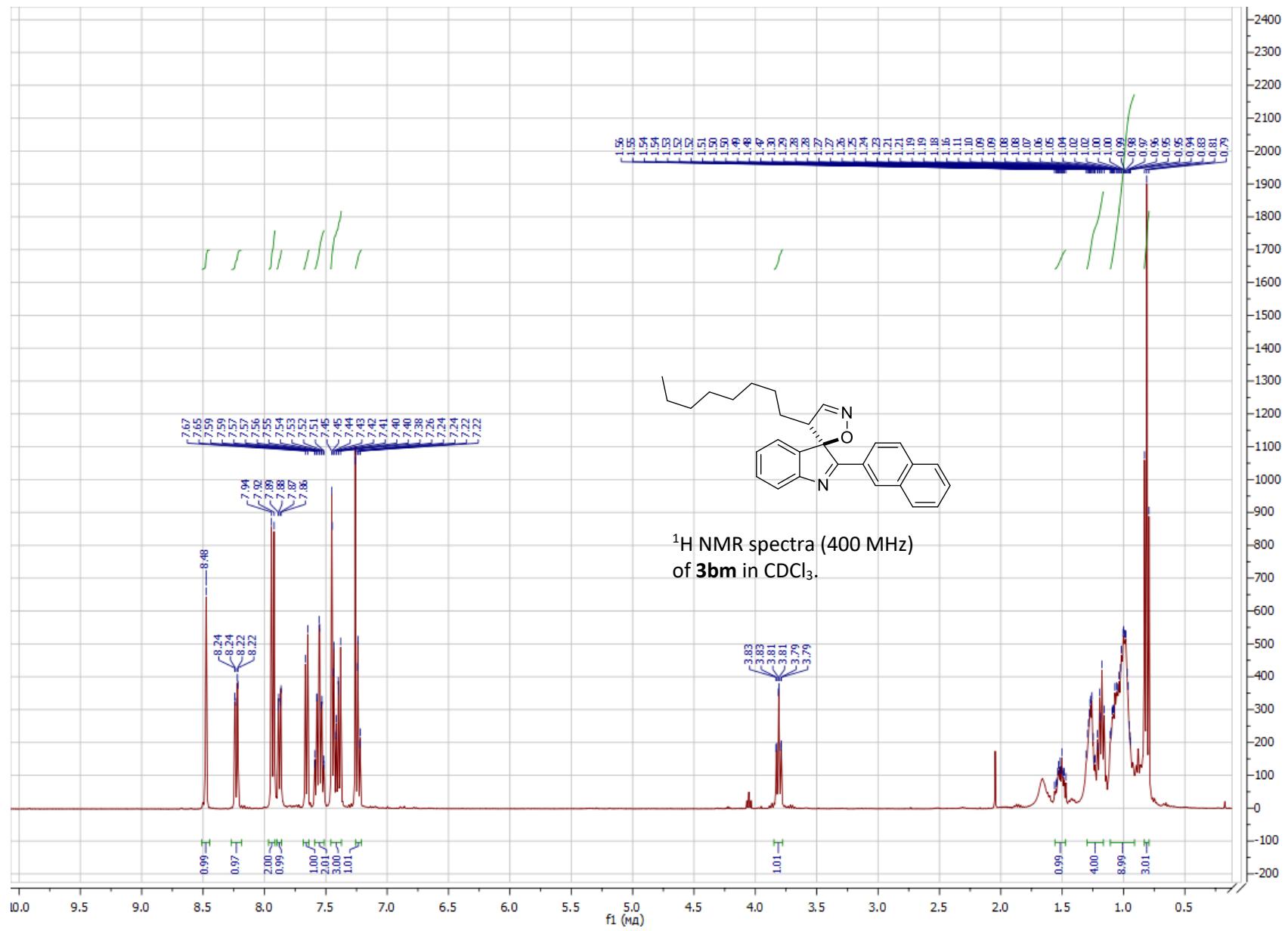


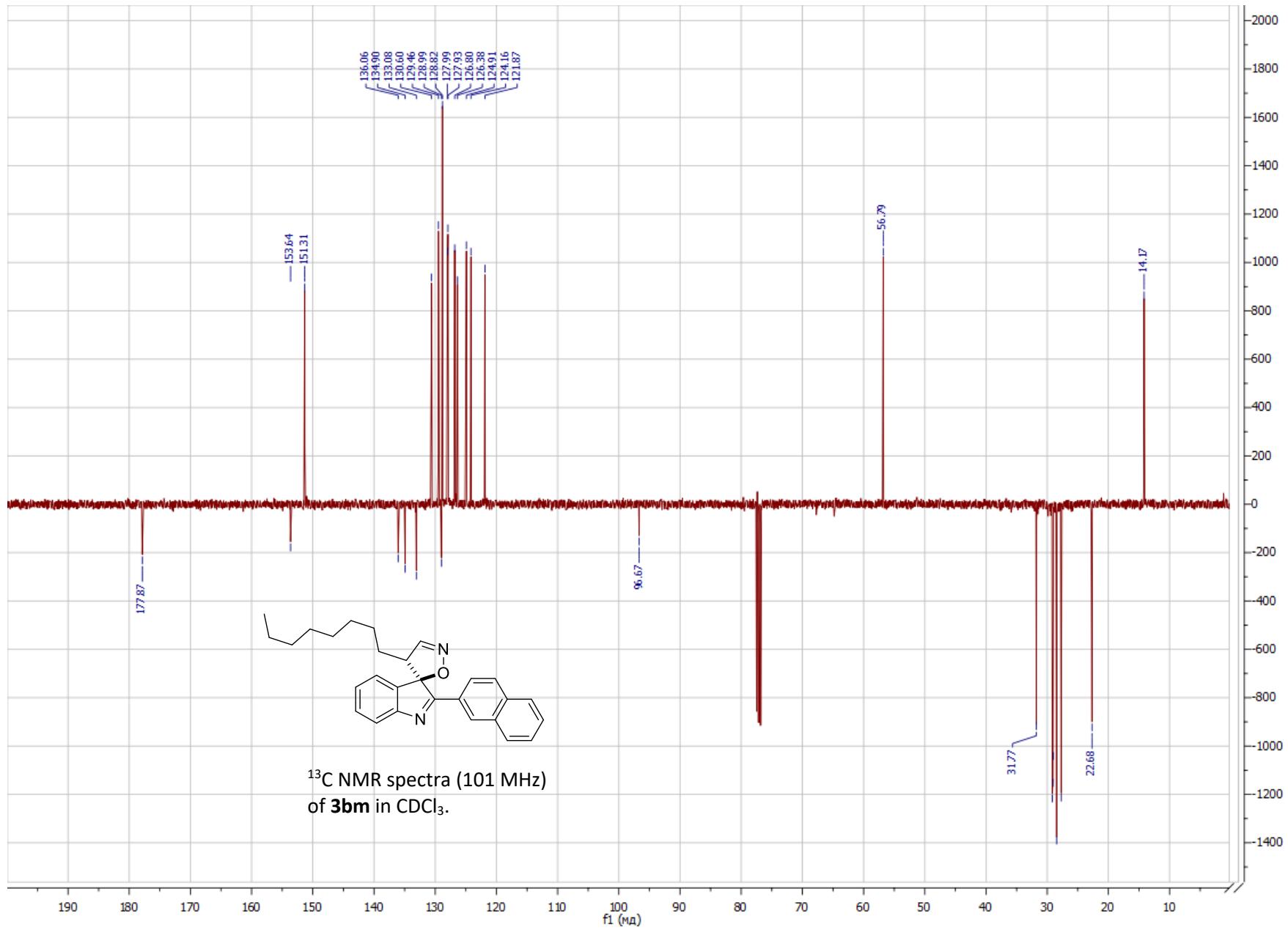


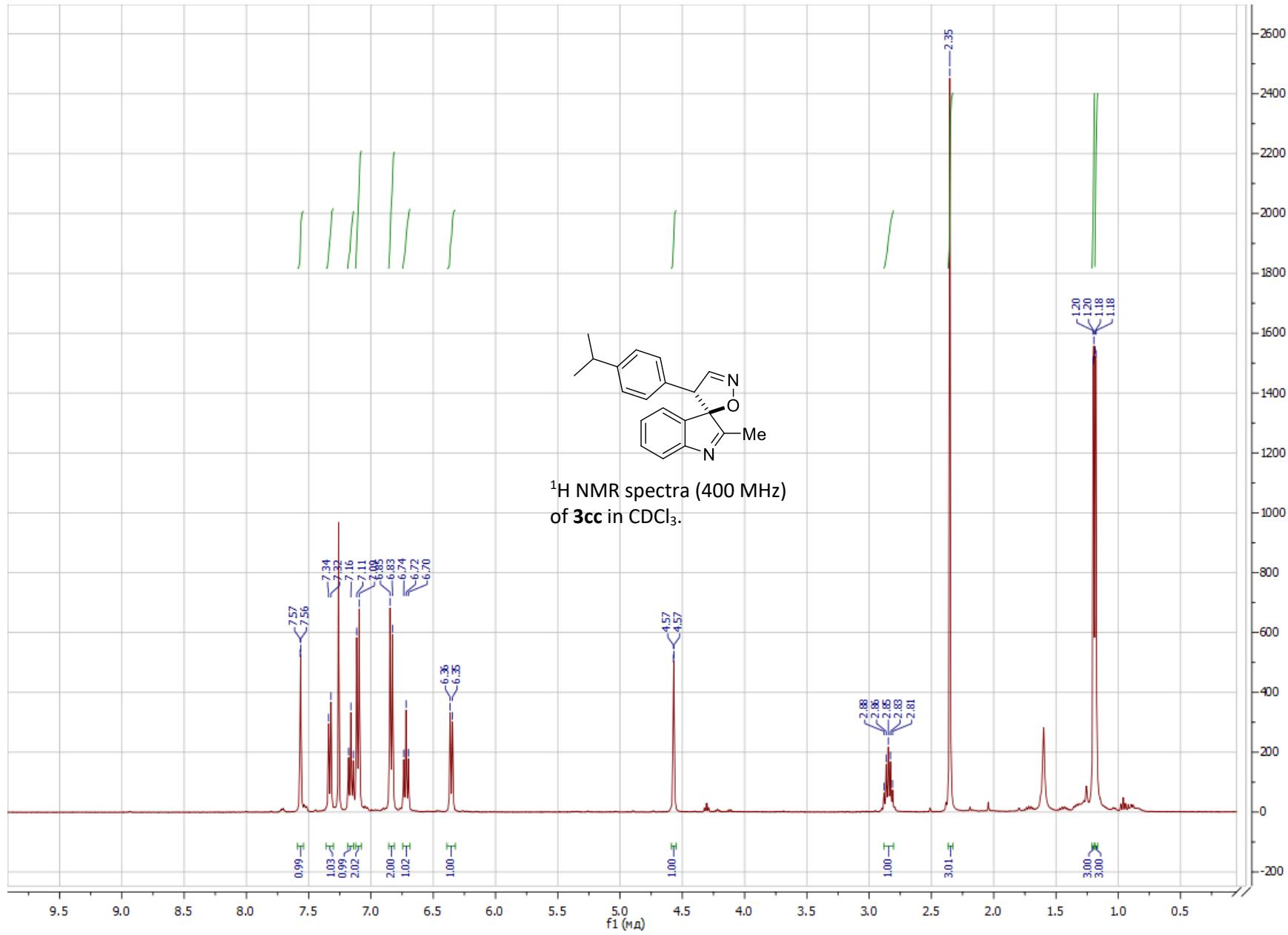


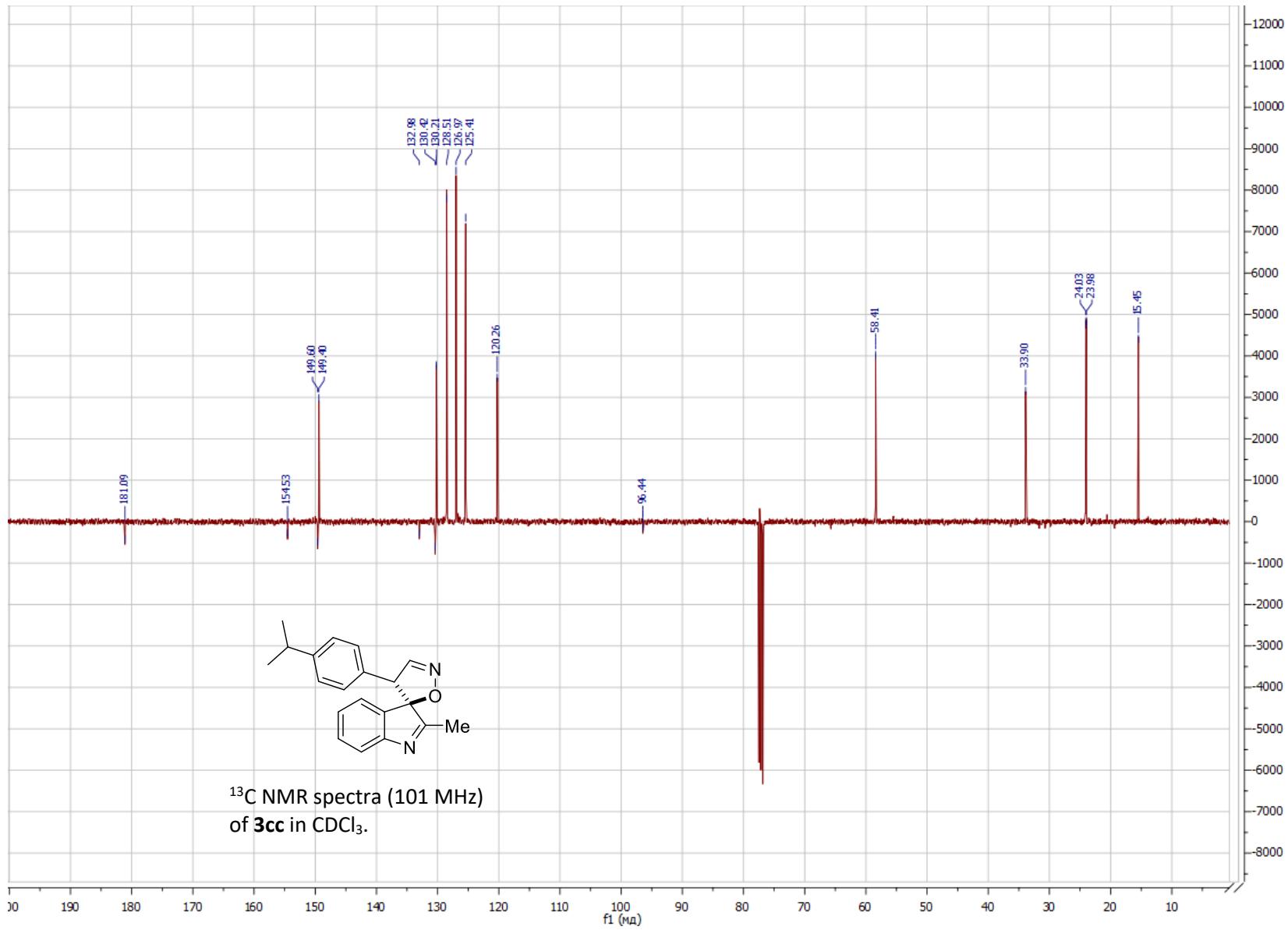


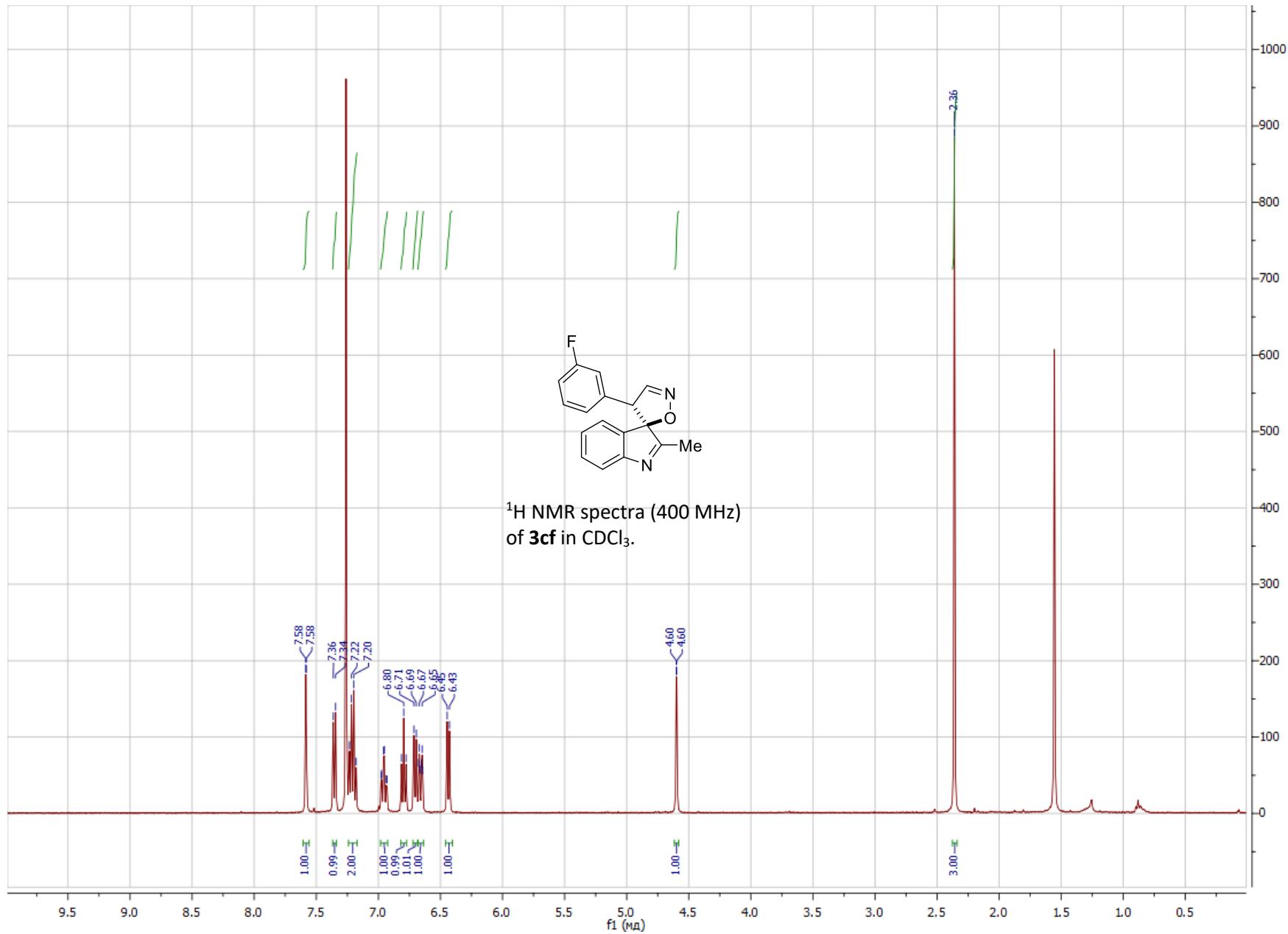


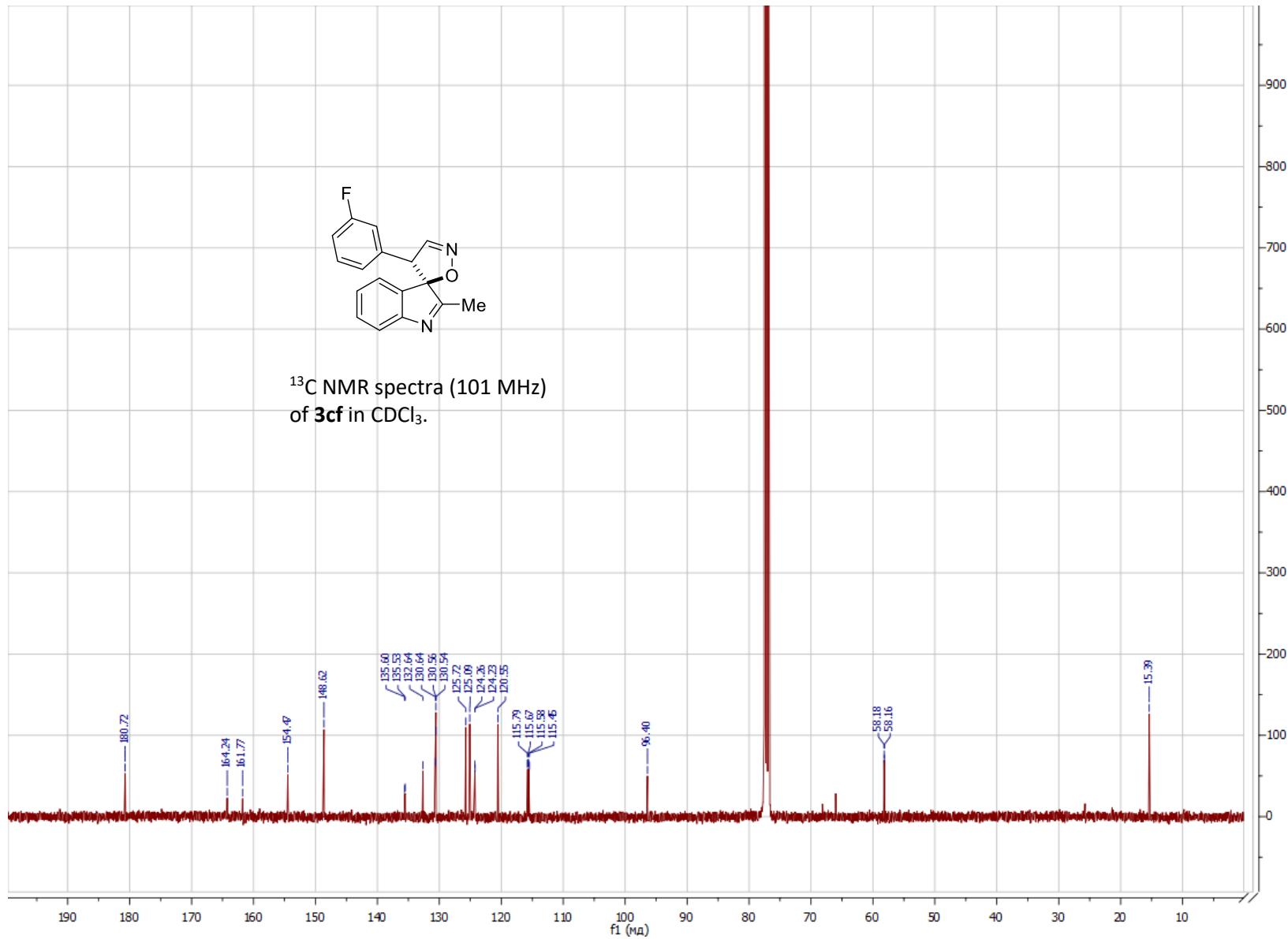


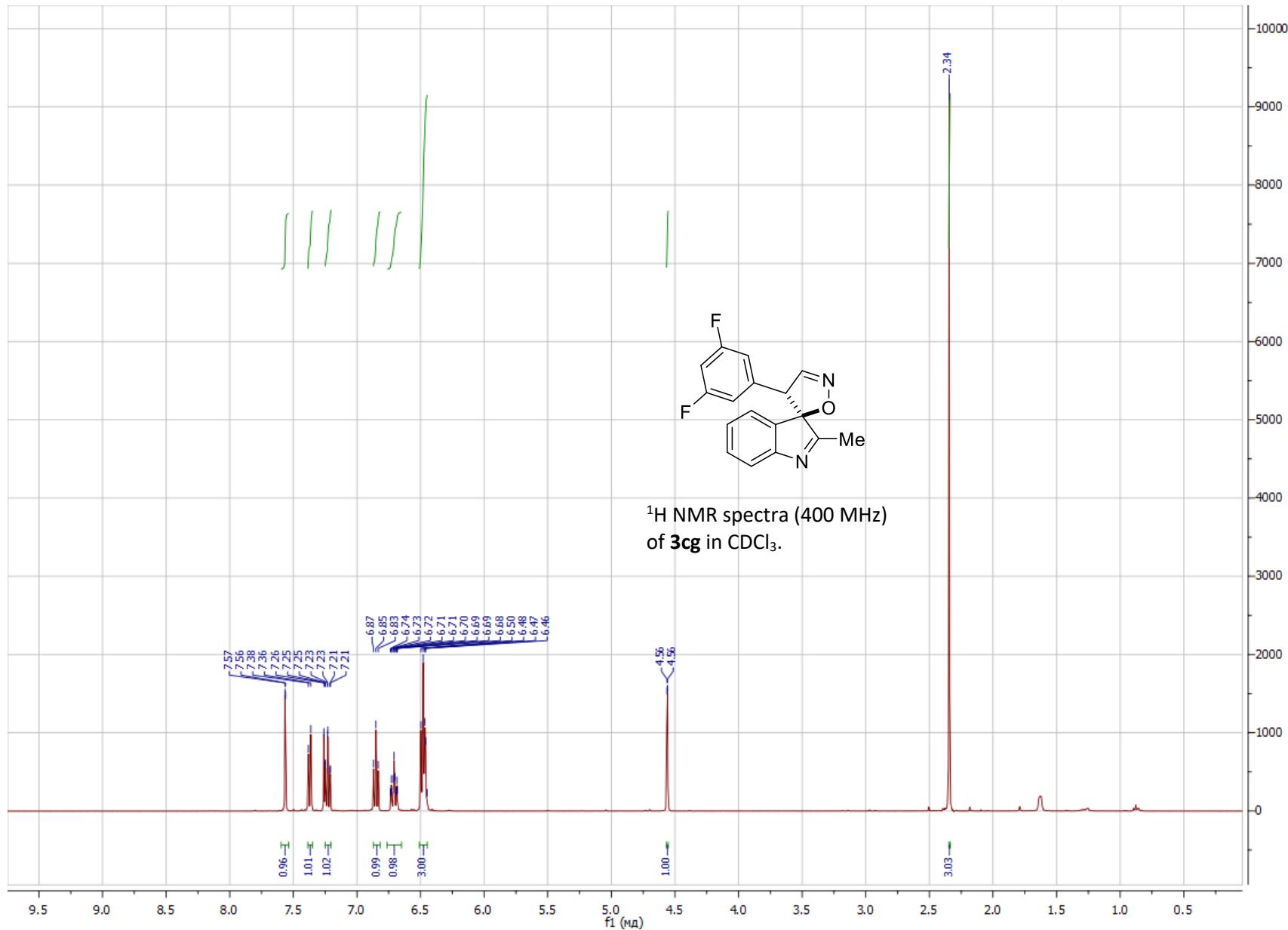


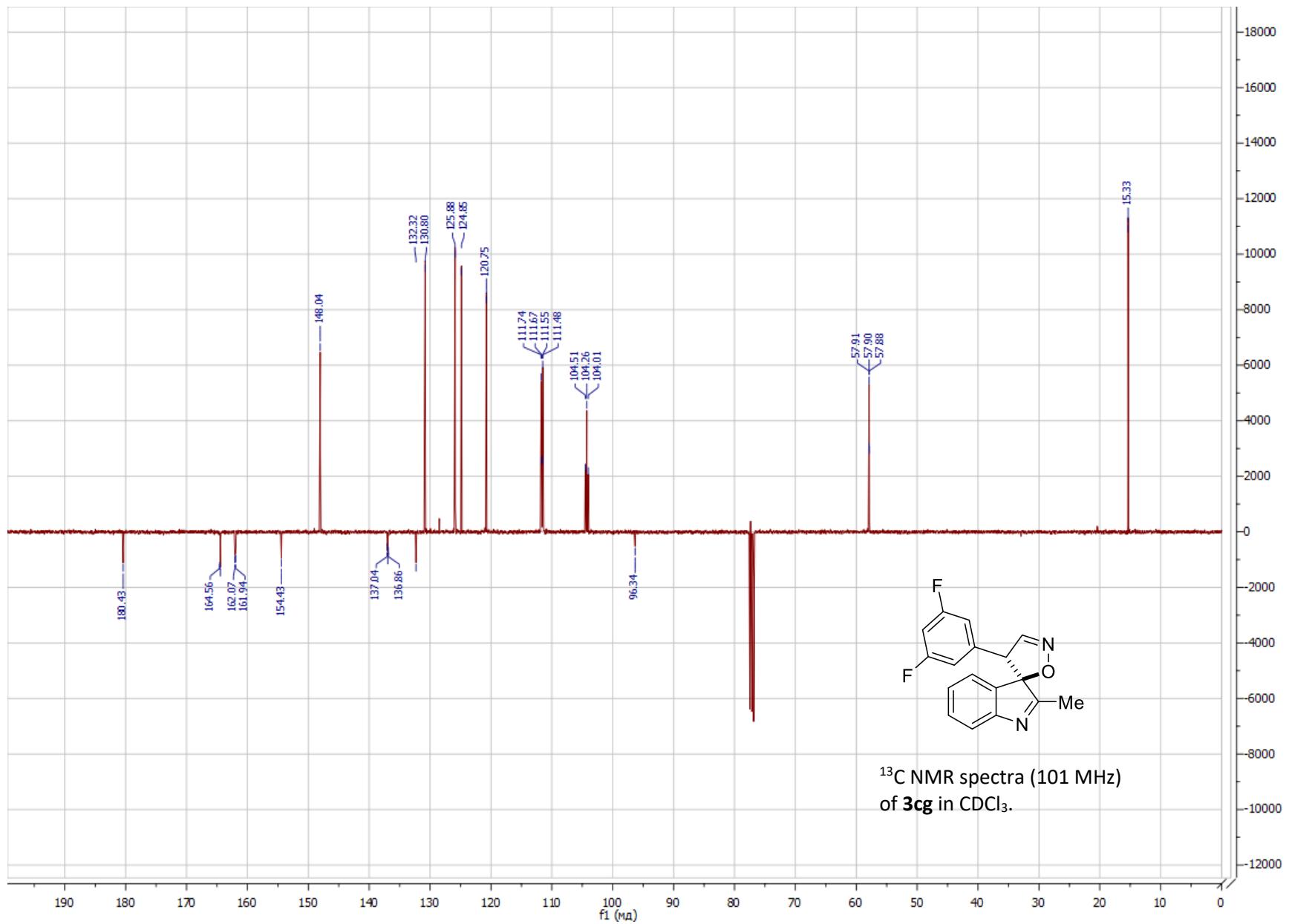


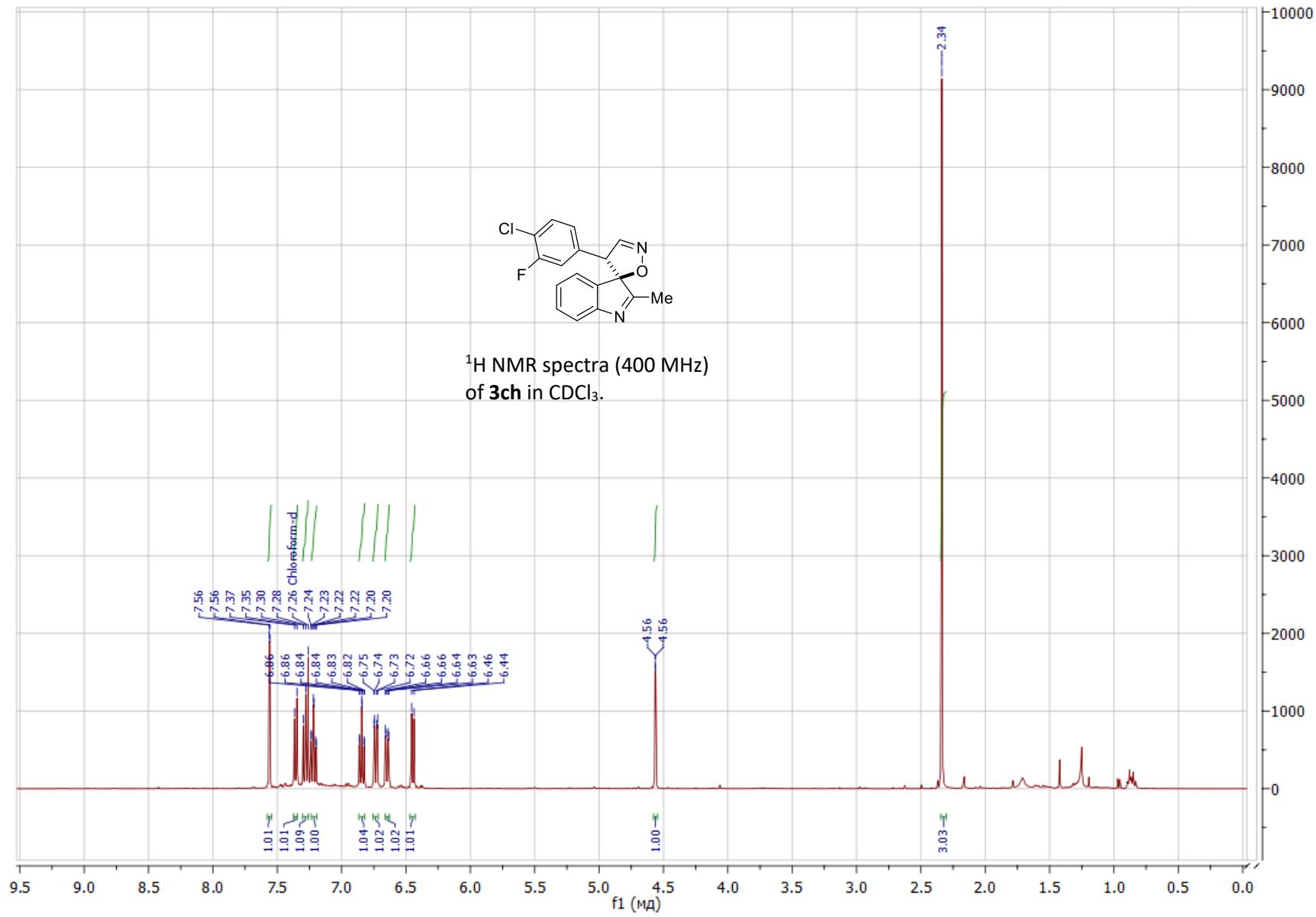


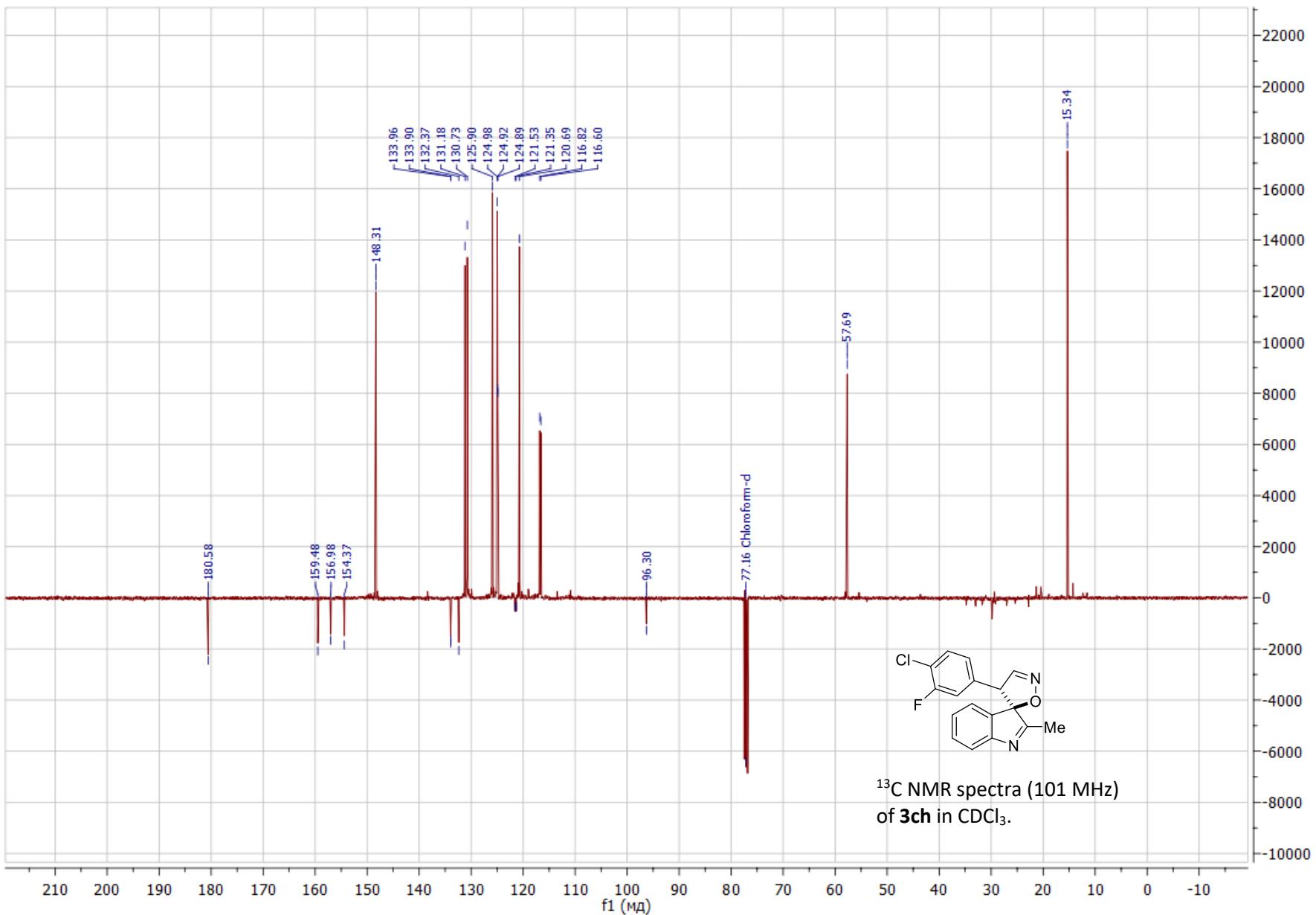


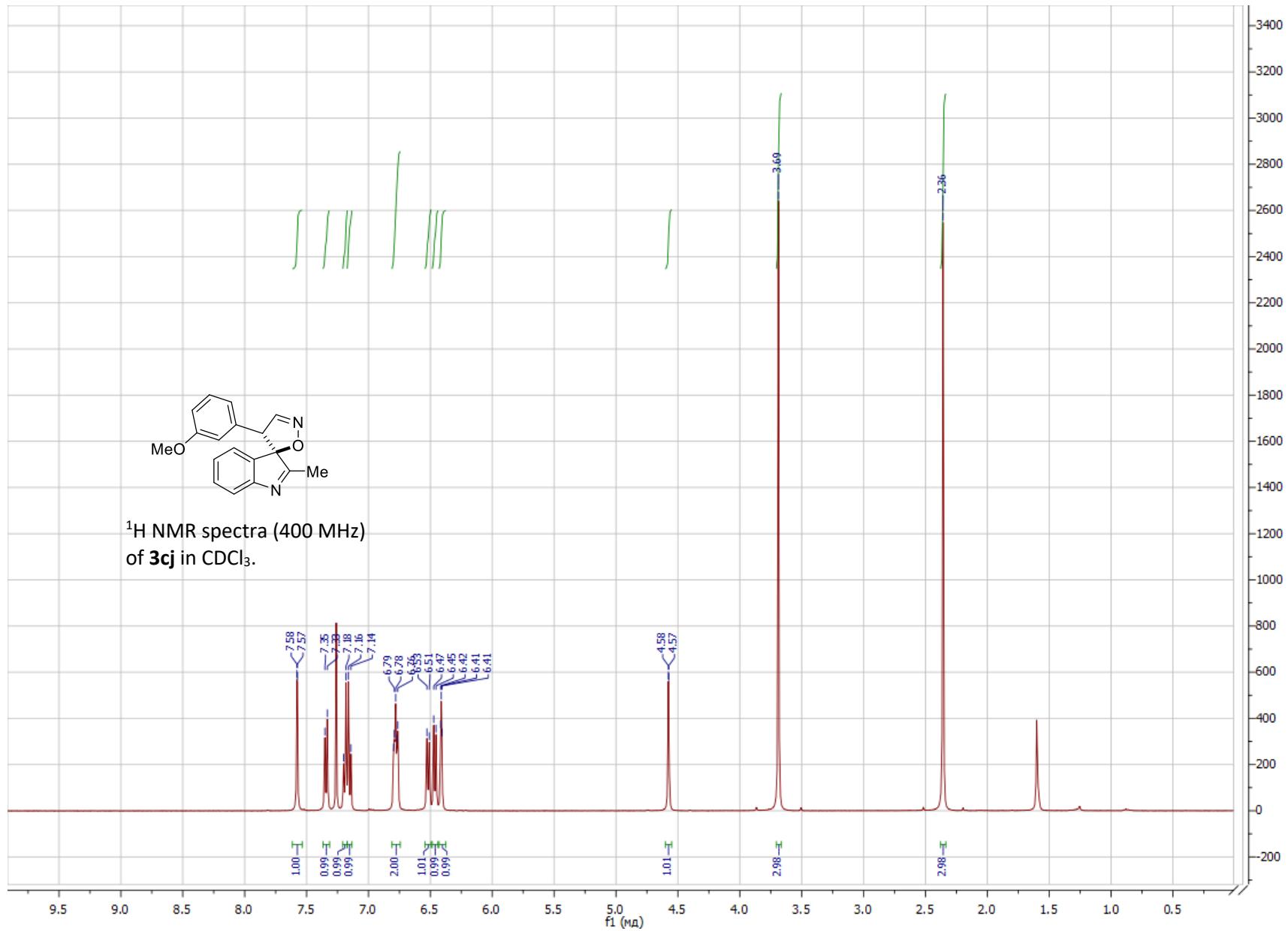


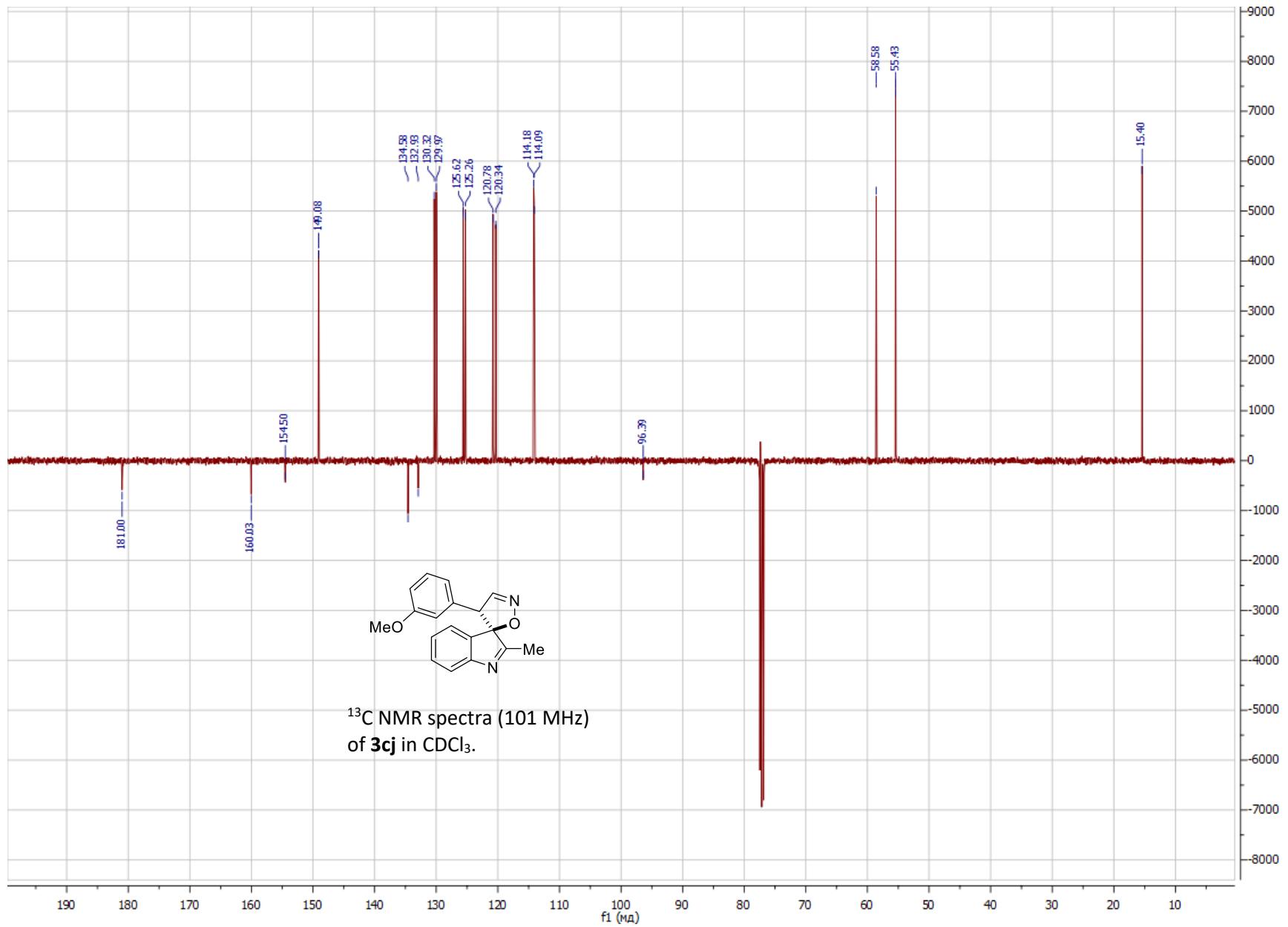


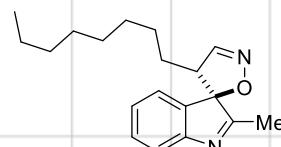




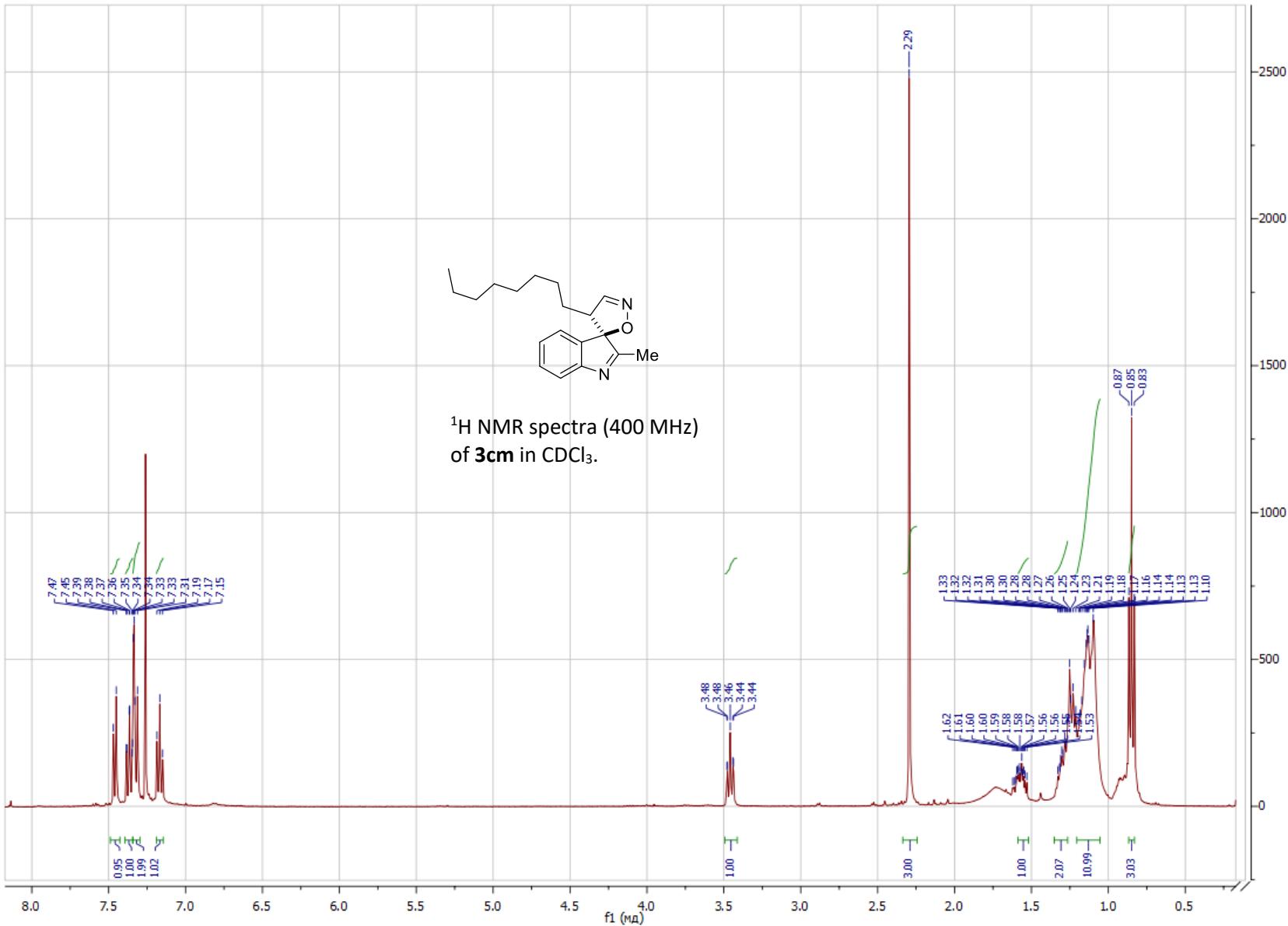


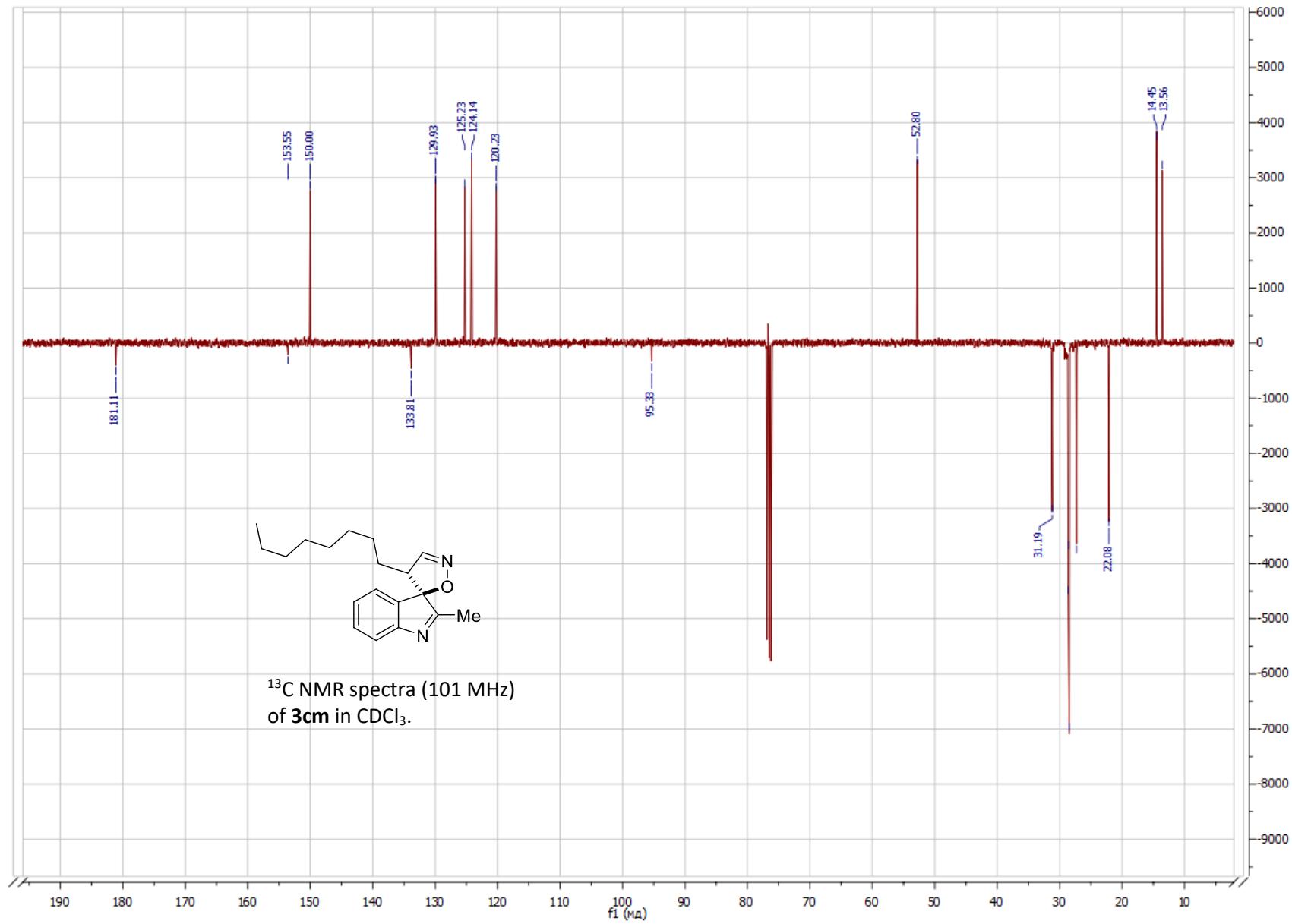


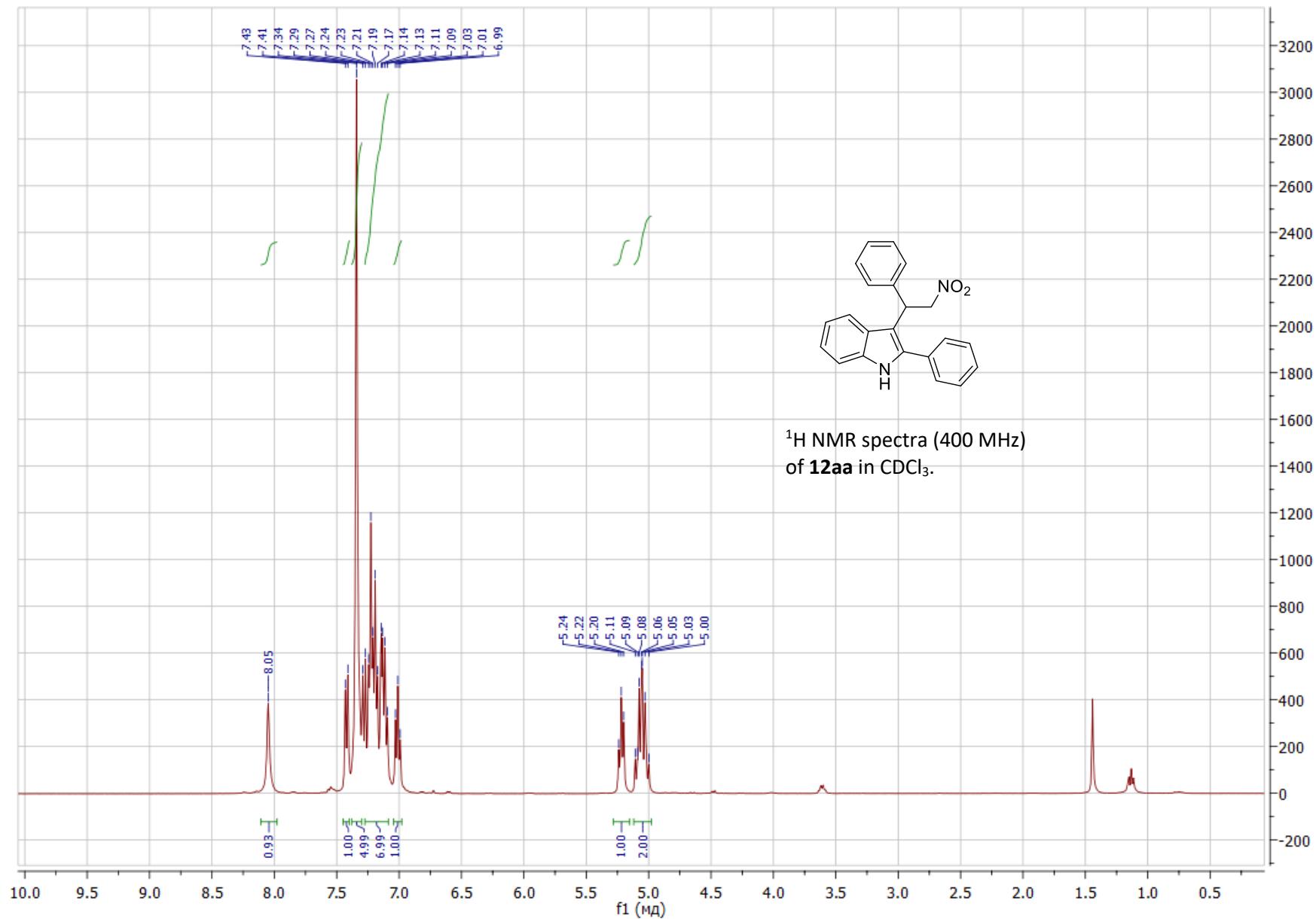


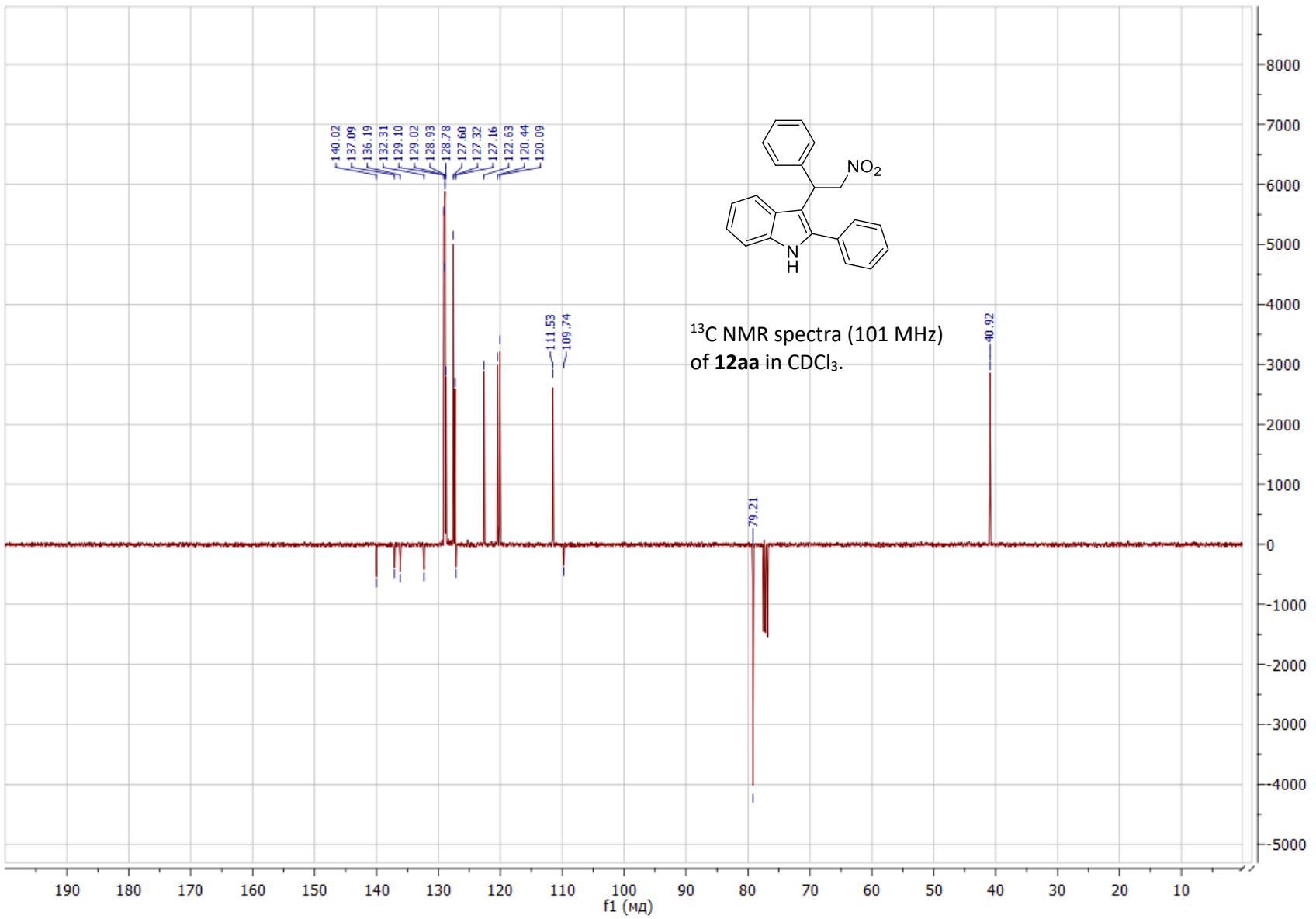


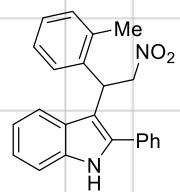
<sup>1</sup>H NMR spectra (400 MHz) of **3cm** in CDCl<sub>3</sub>.



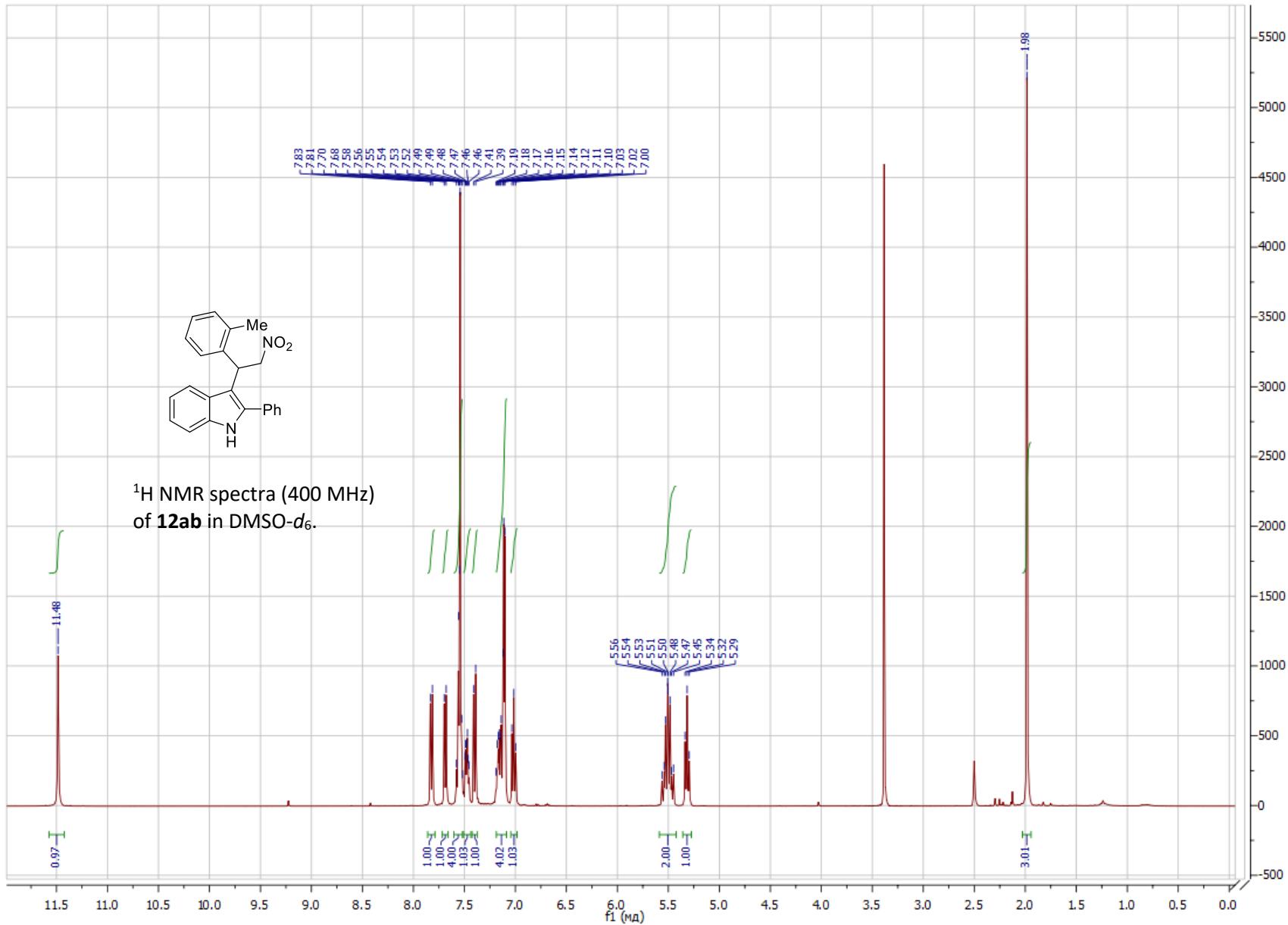


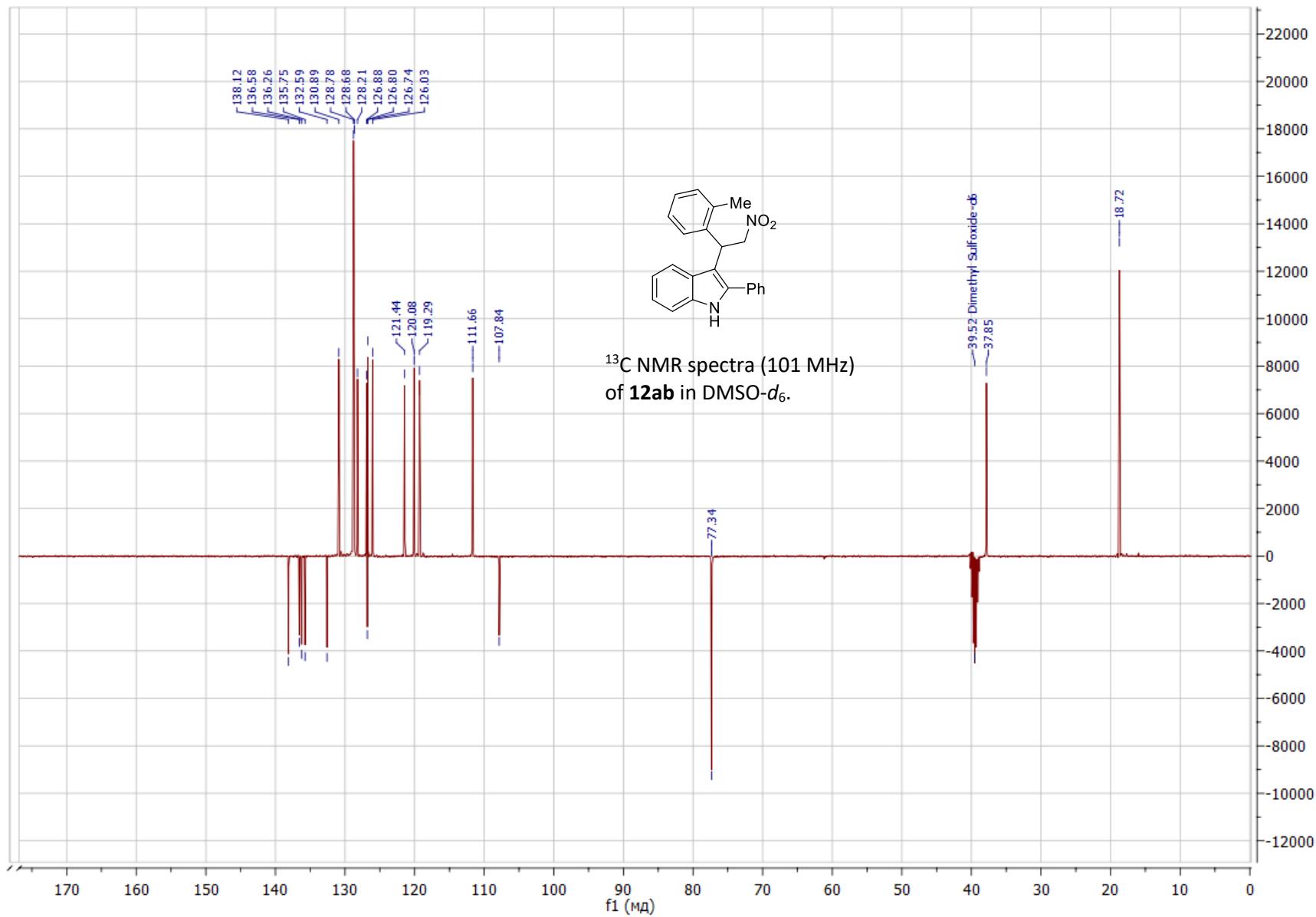


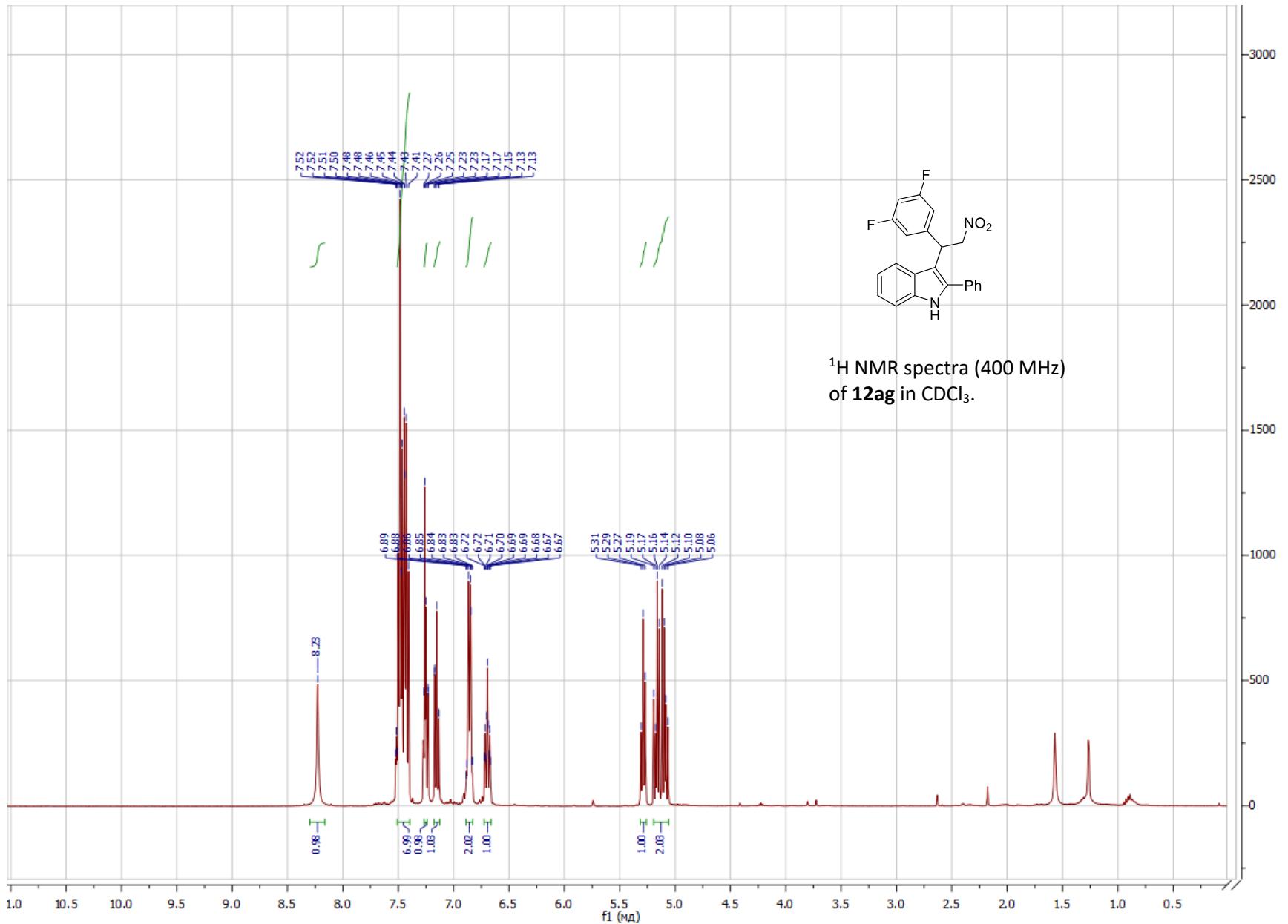


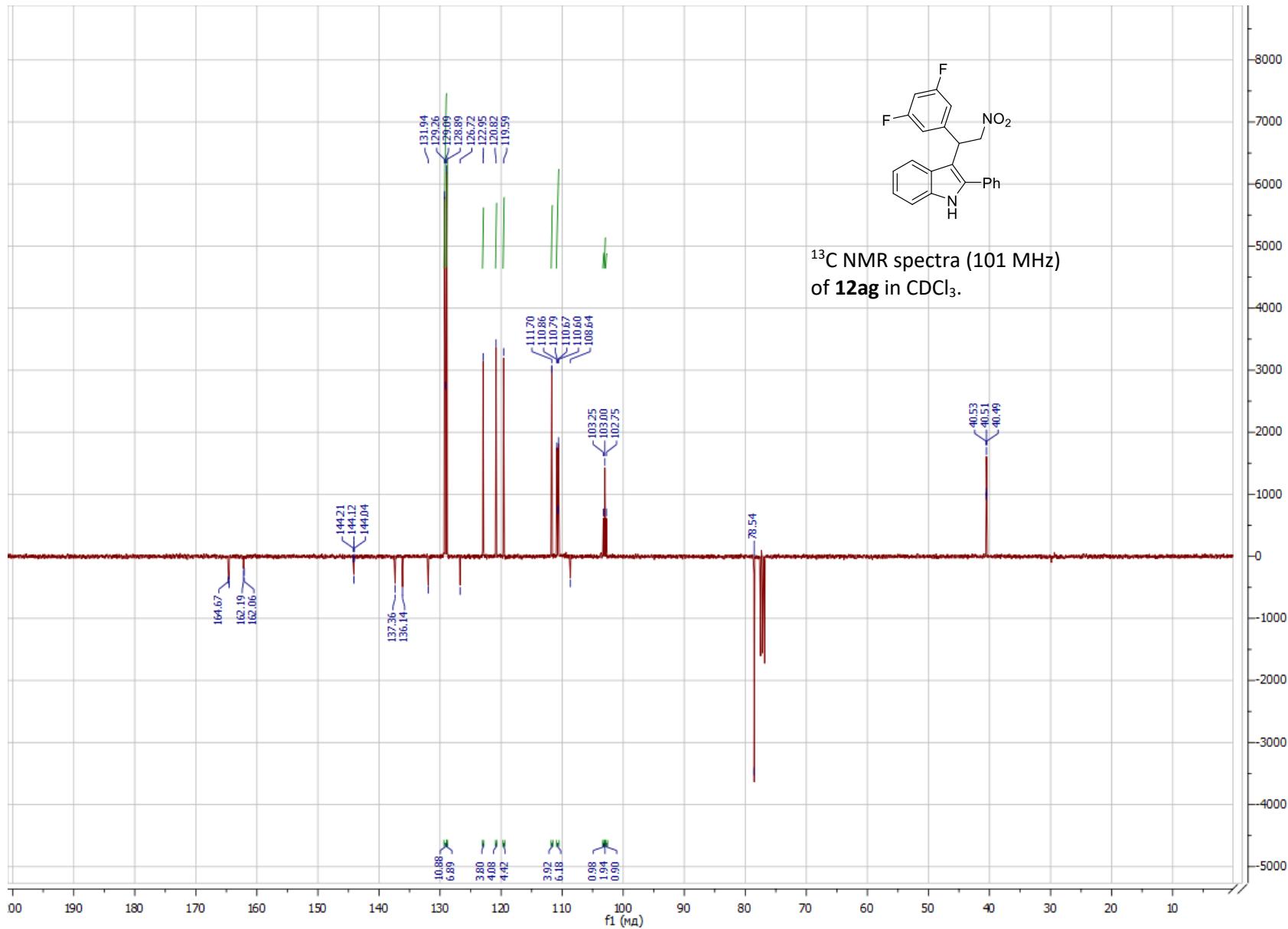


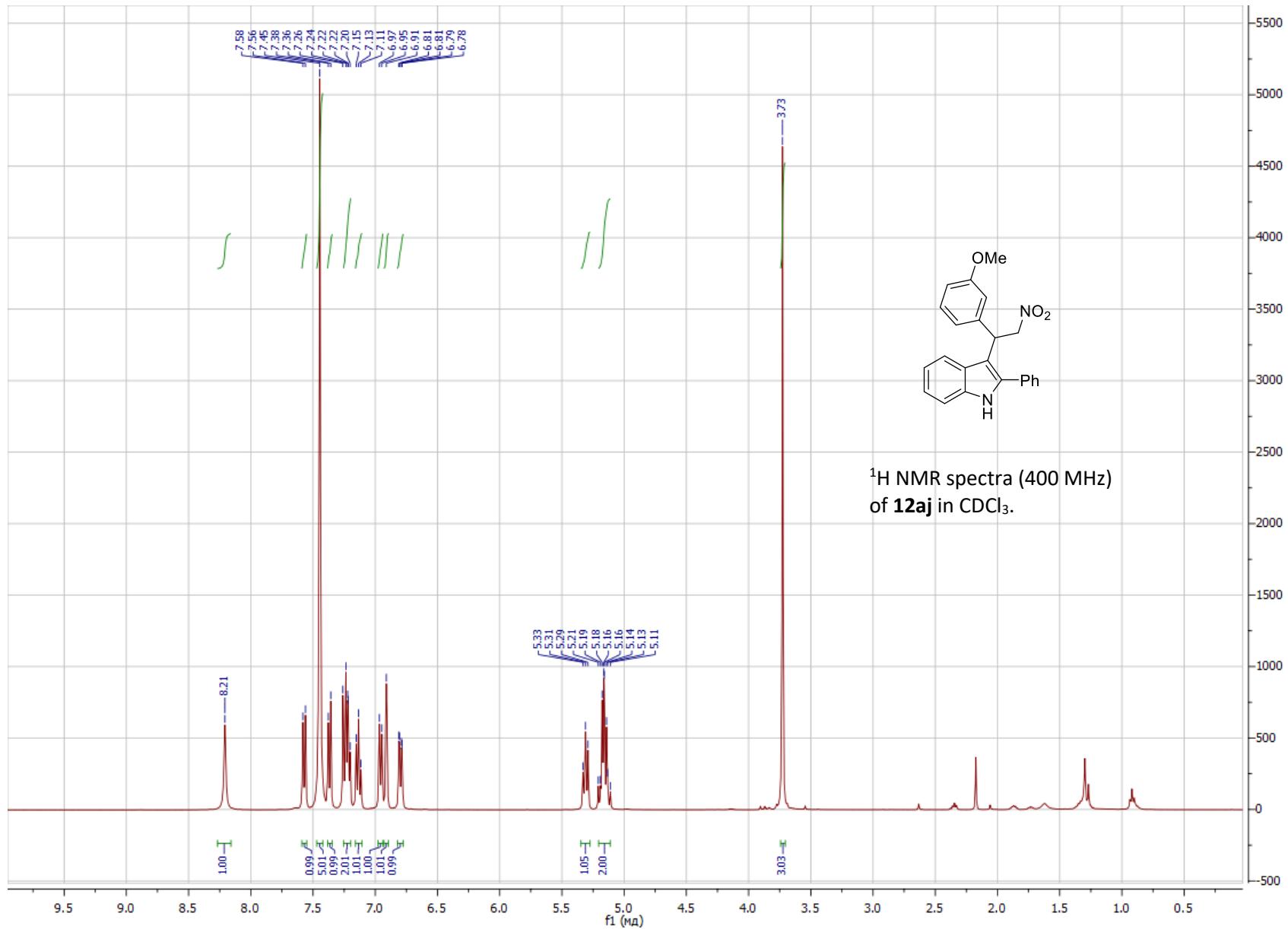
<sup>1</sup>H NMR spectra (400 MHz) of **12ab** in DMSO-*d*<sub>6</sub>.

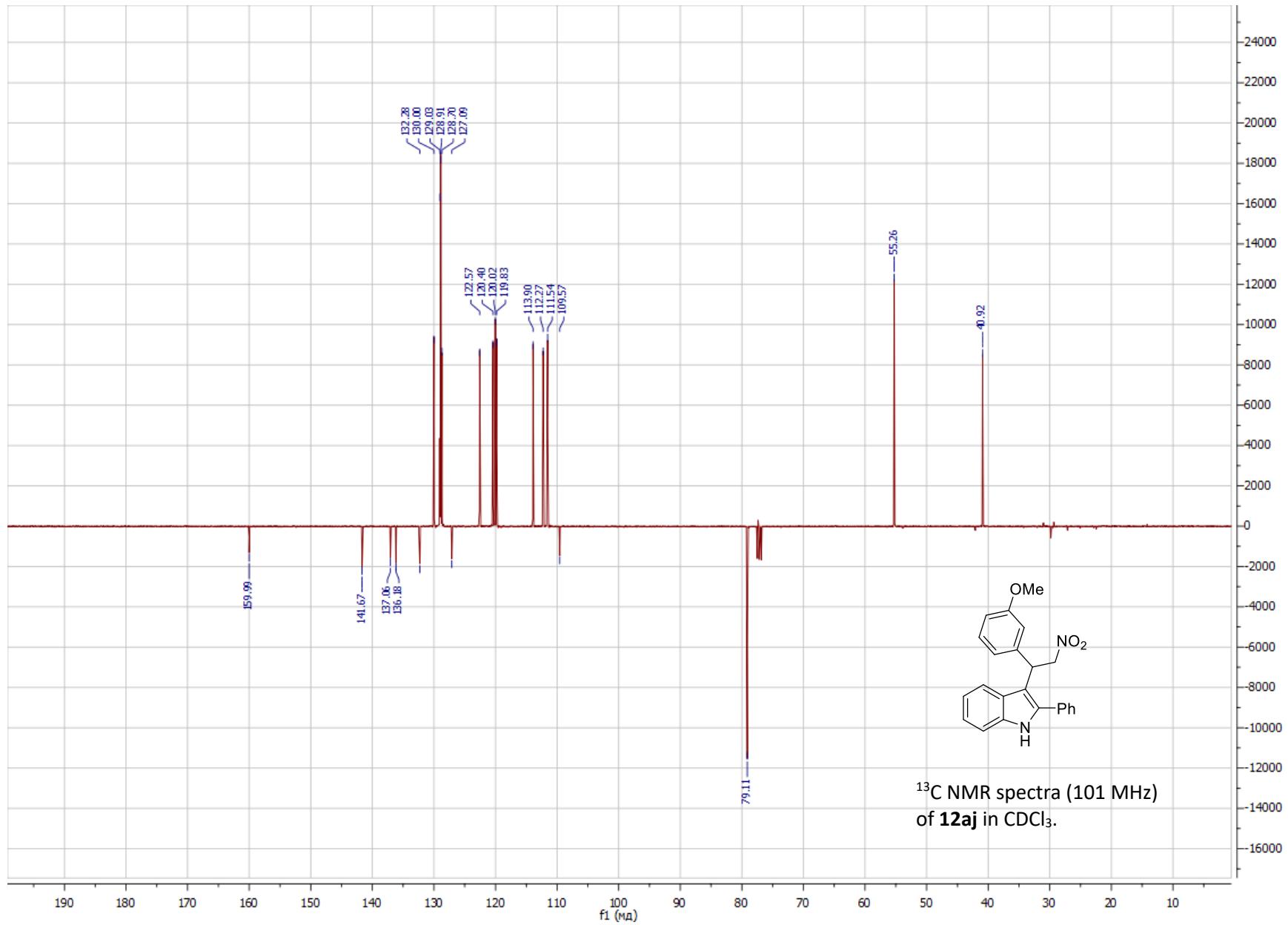


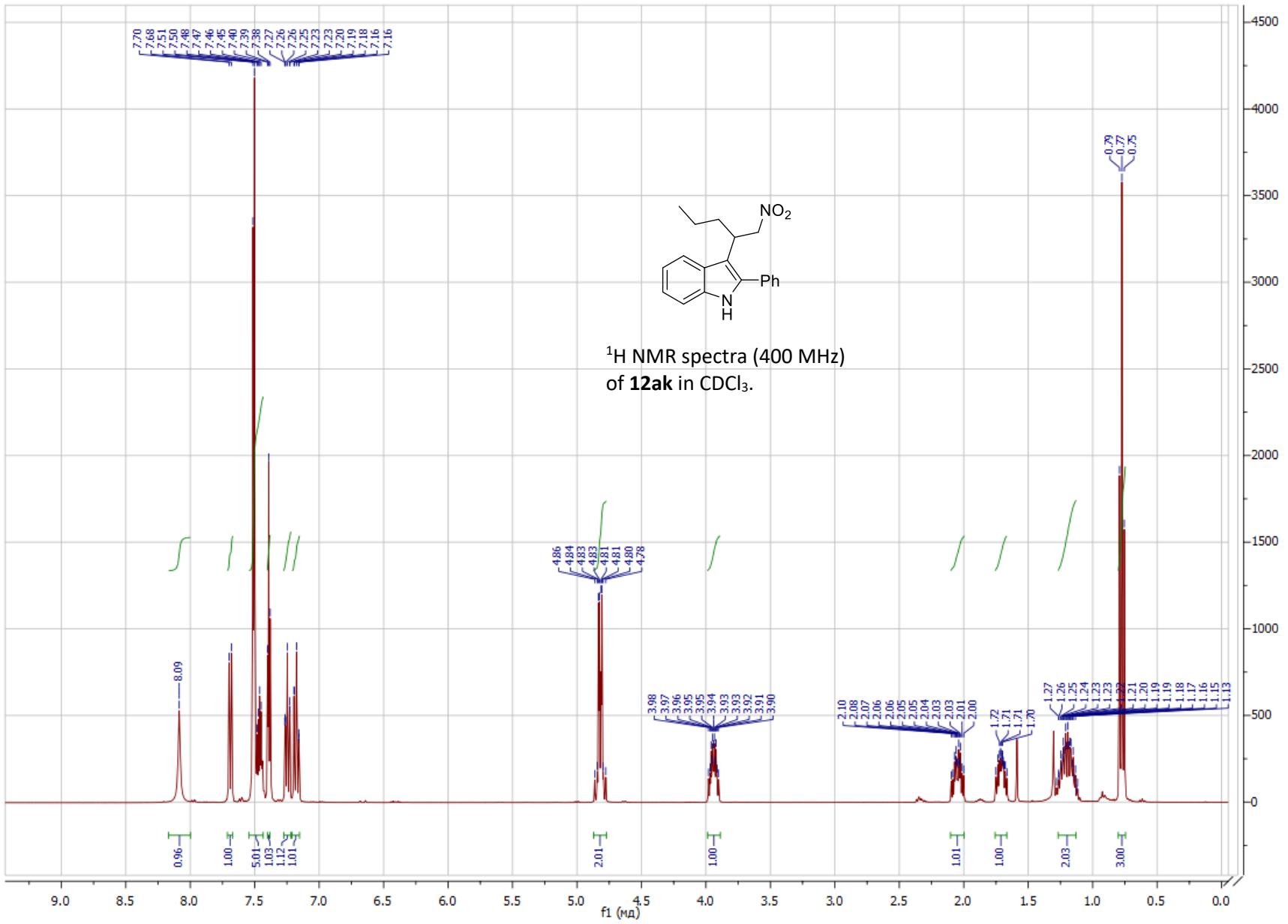


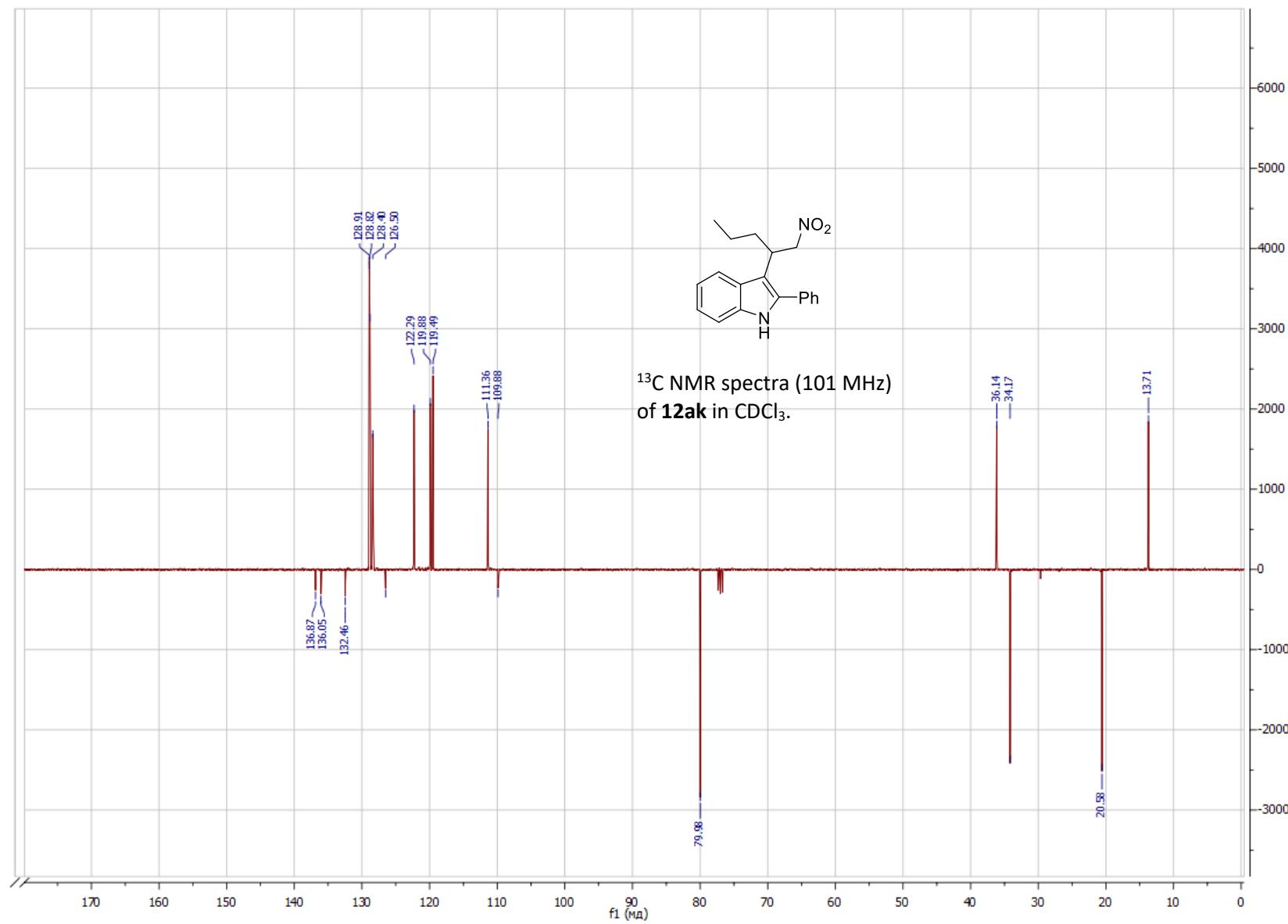


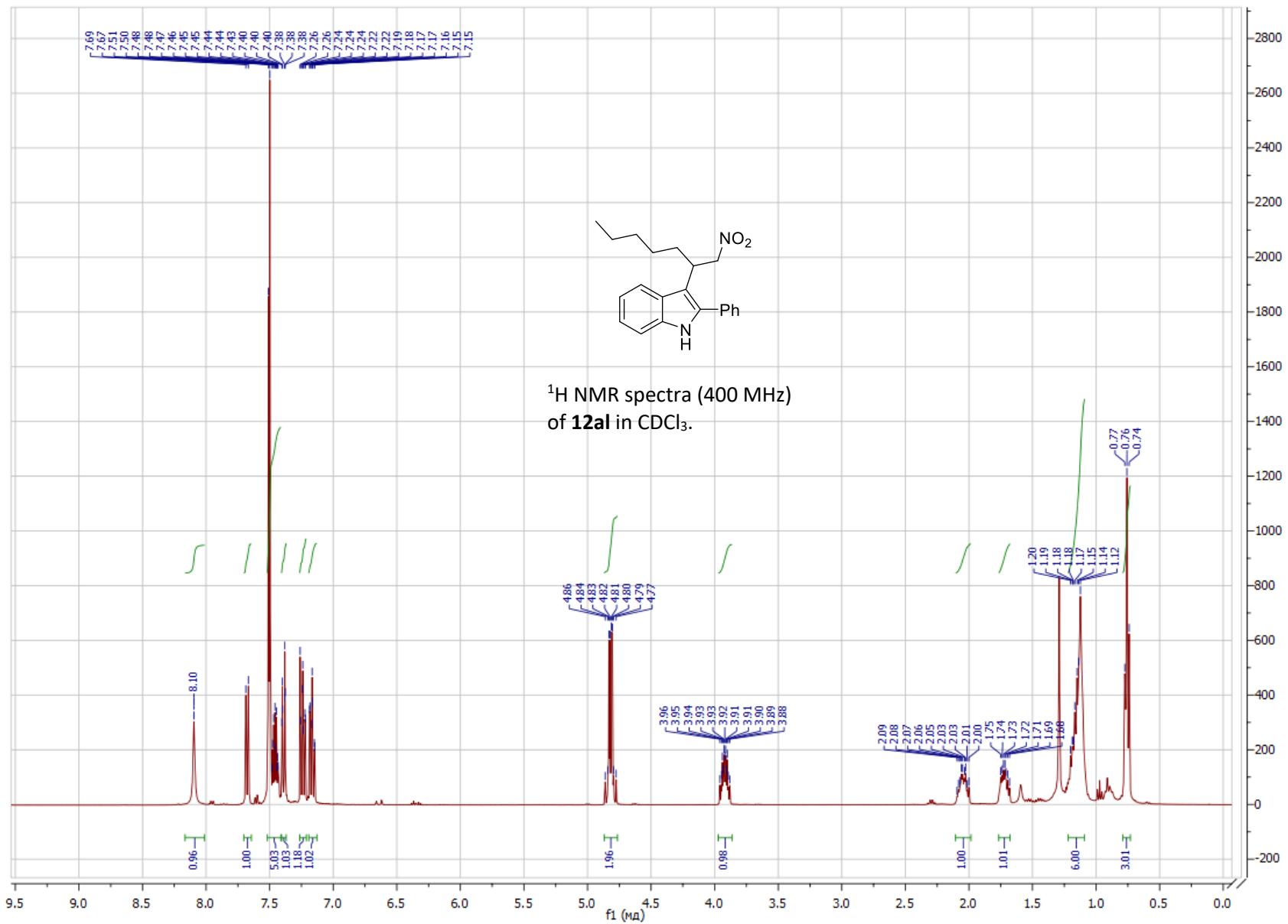


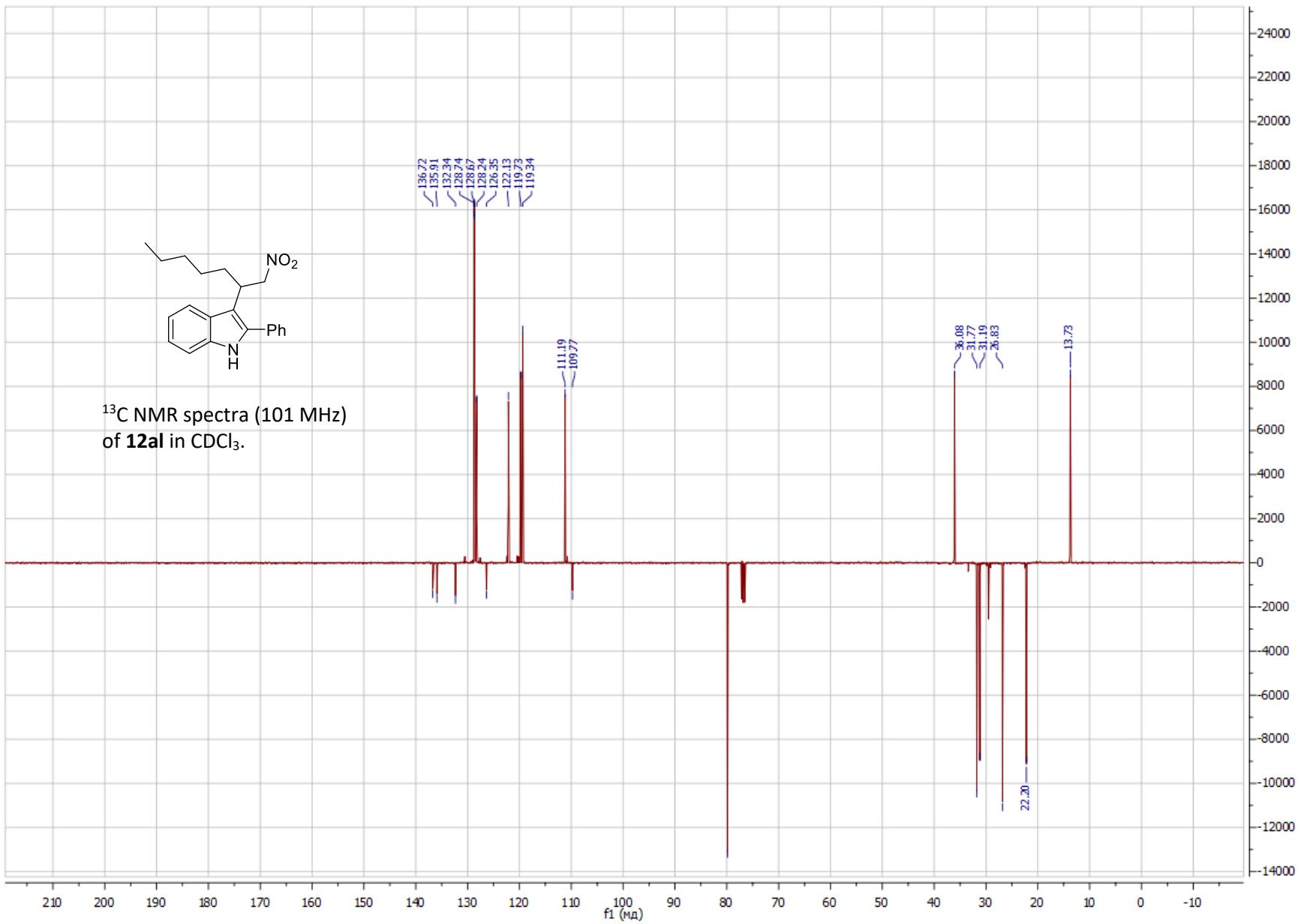




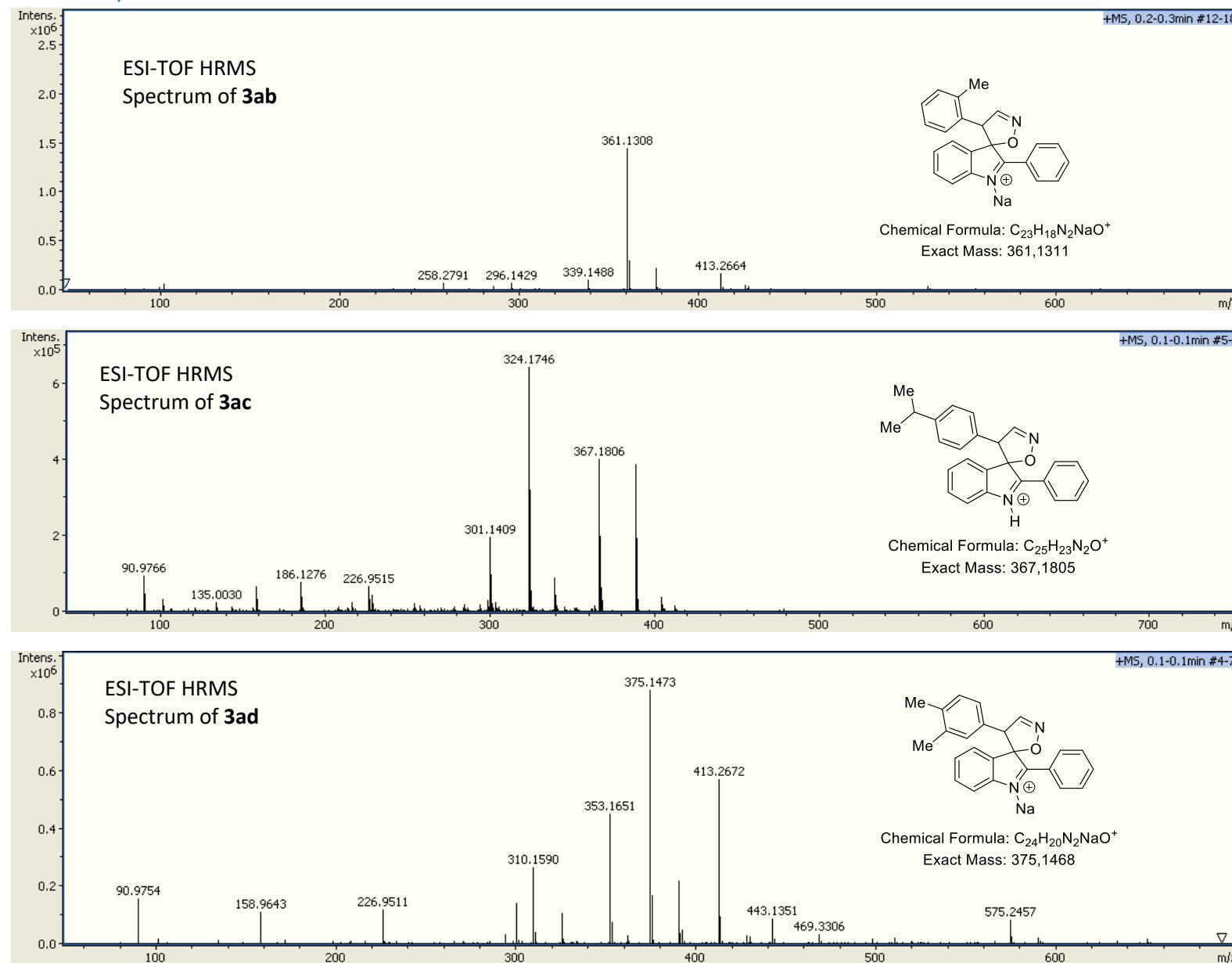


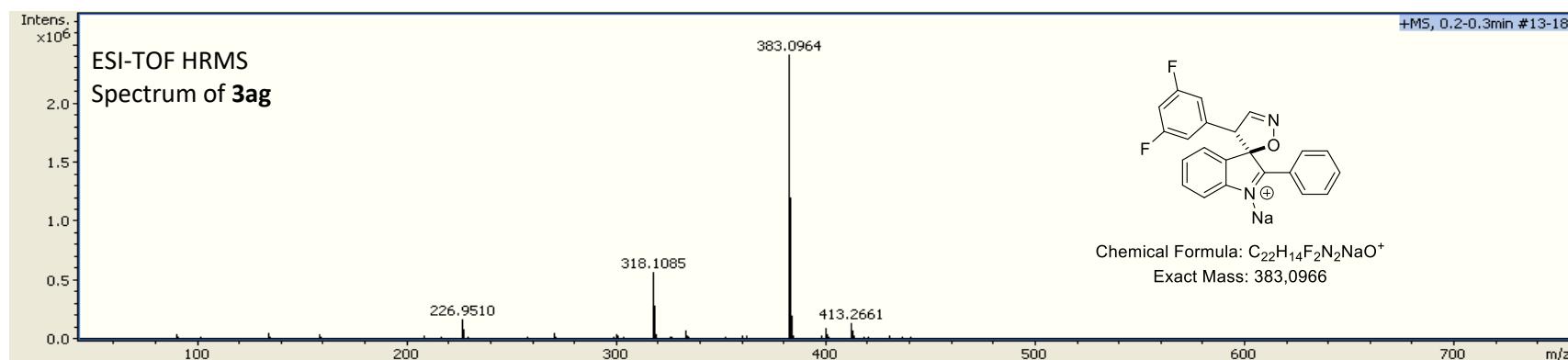
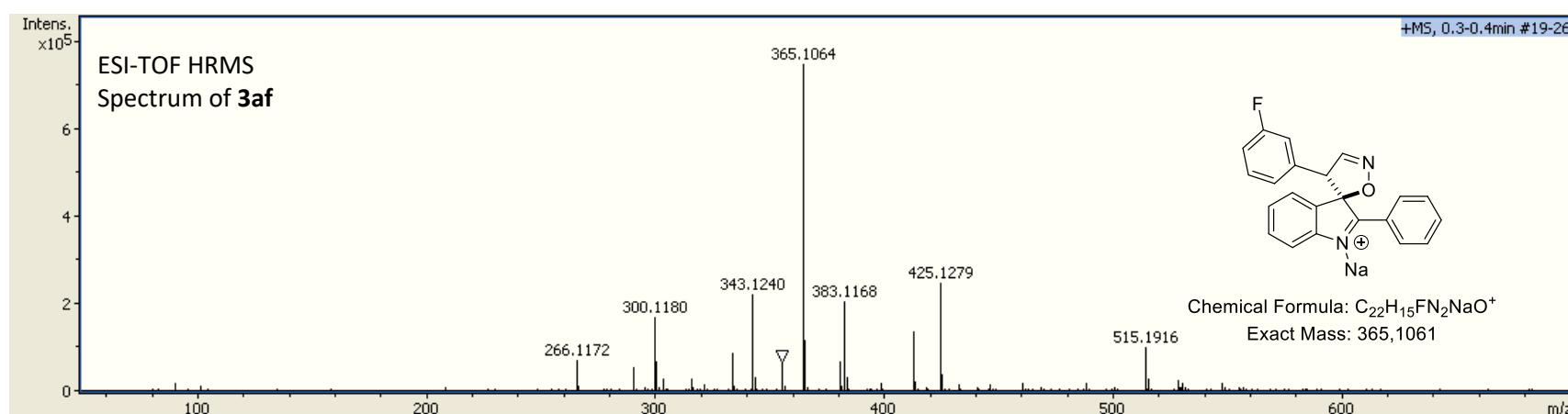
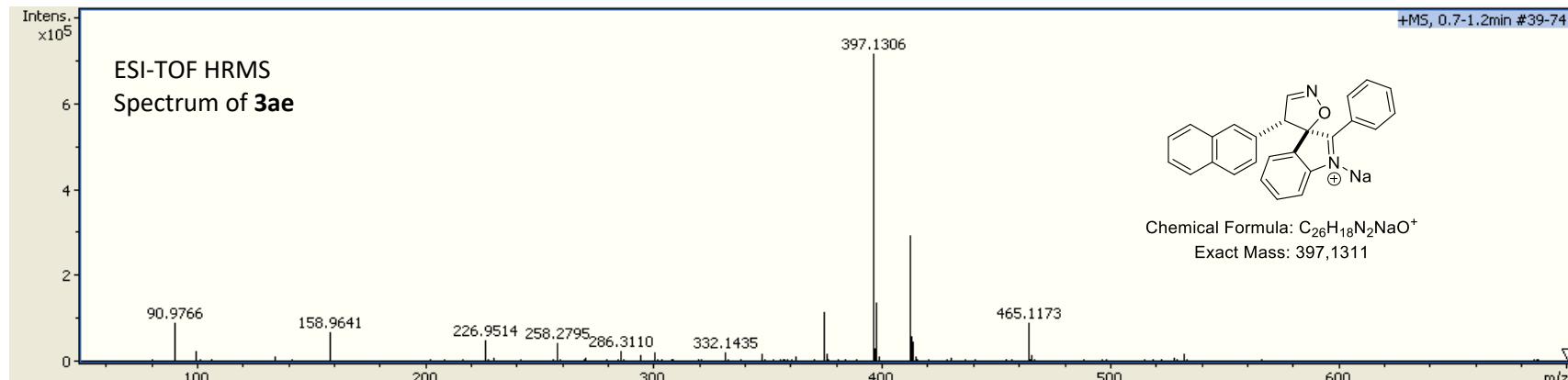


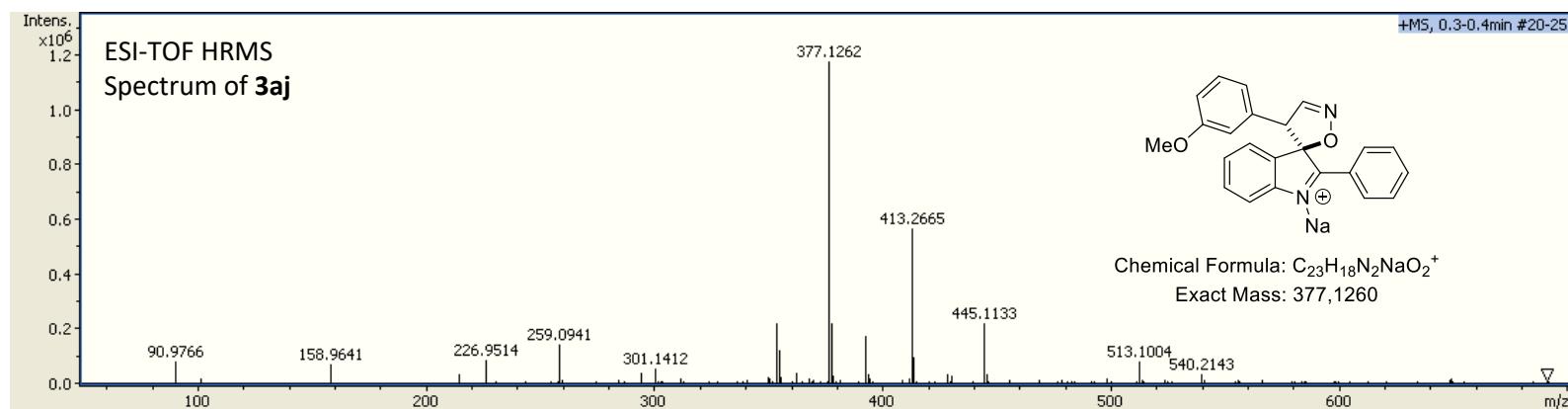
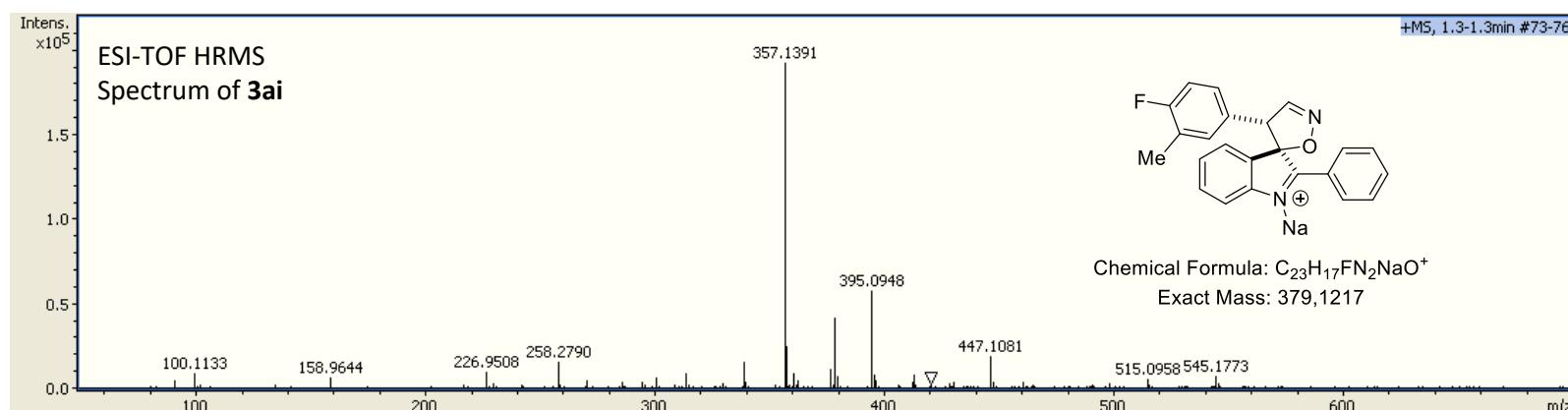
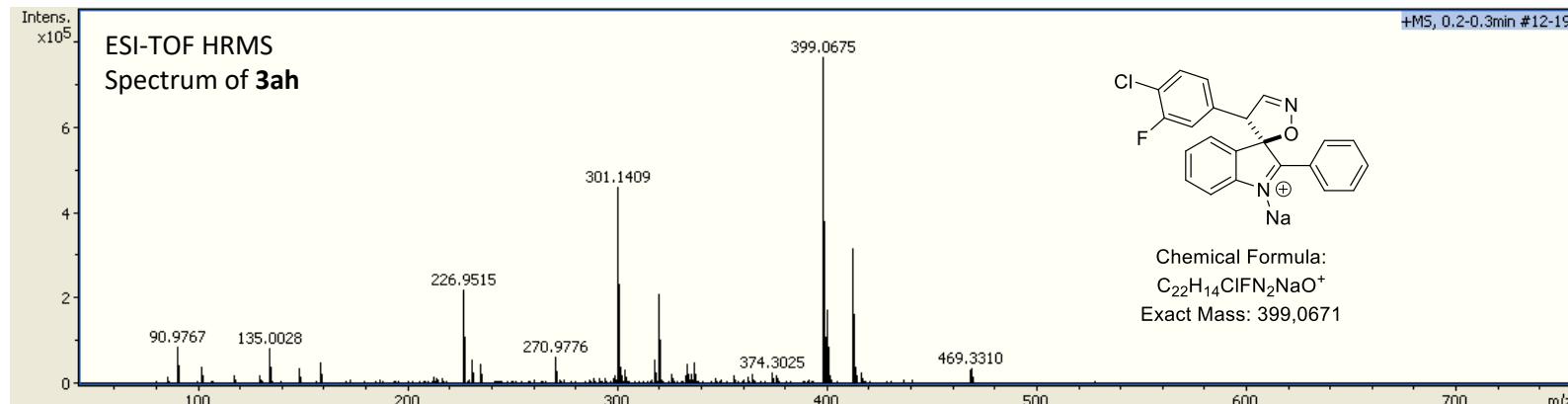


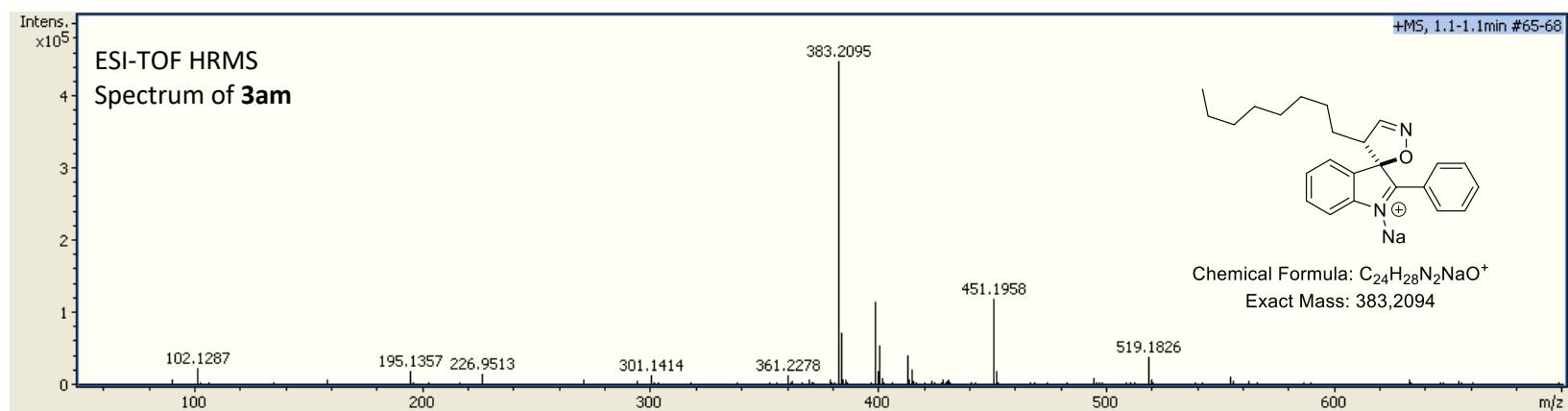
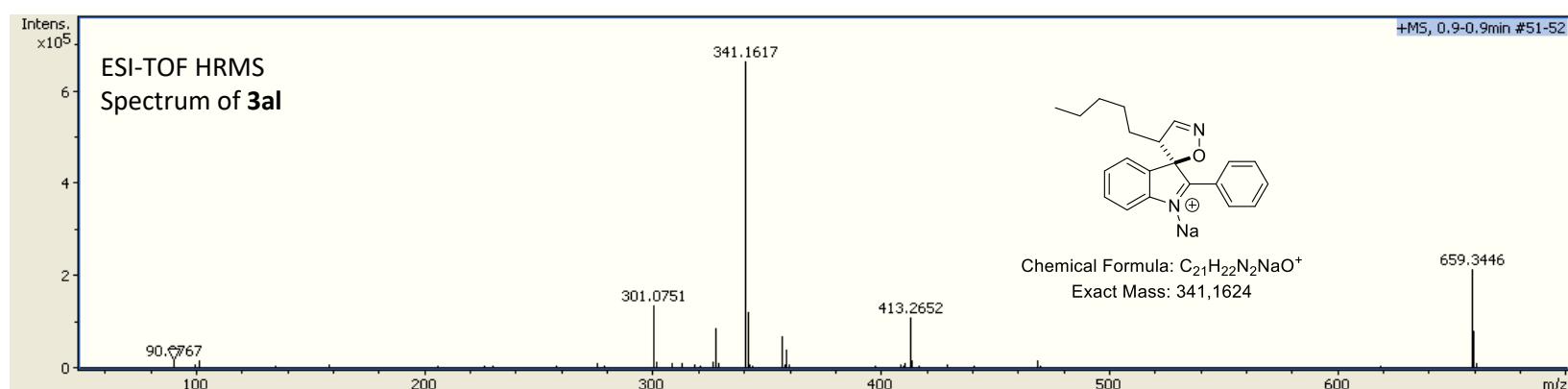
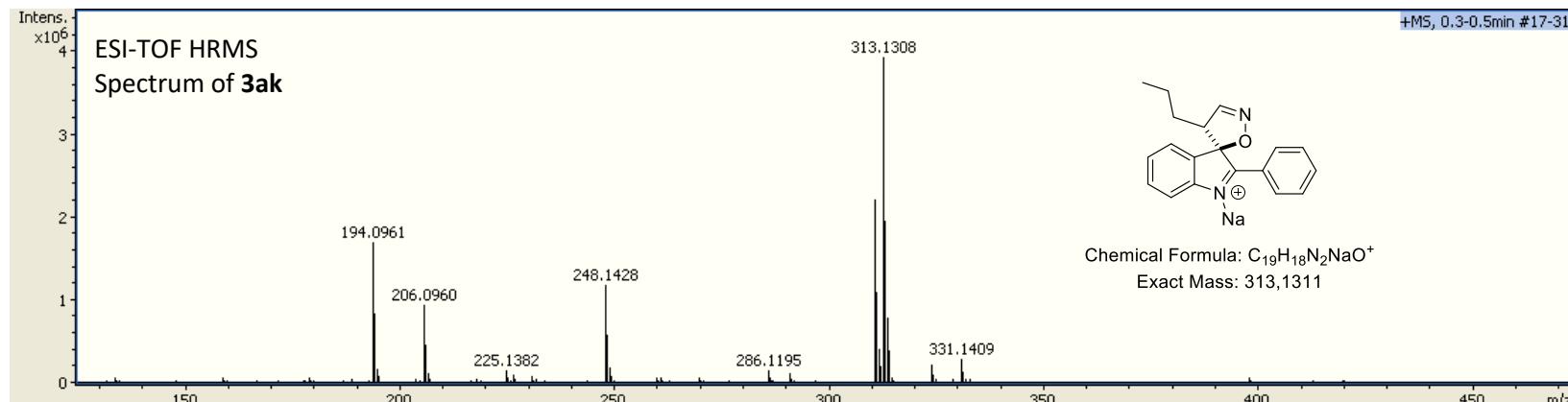


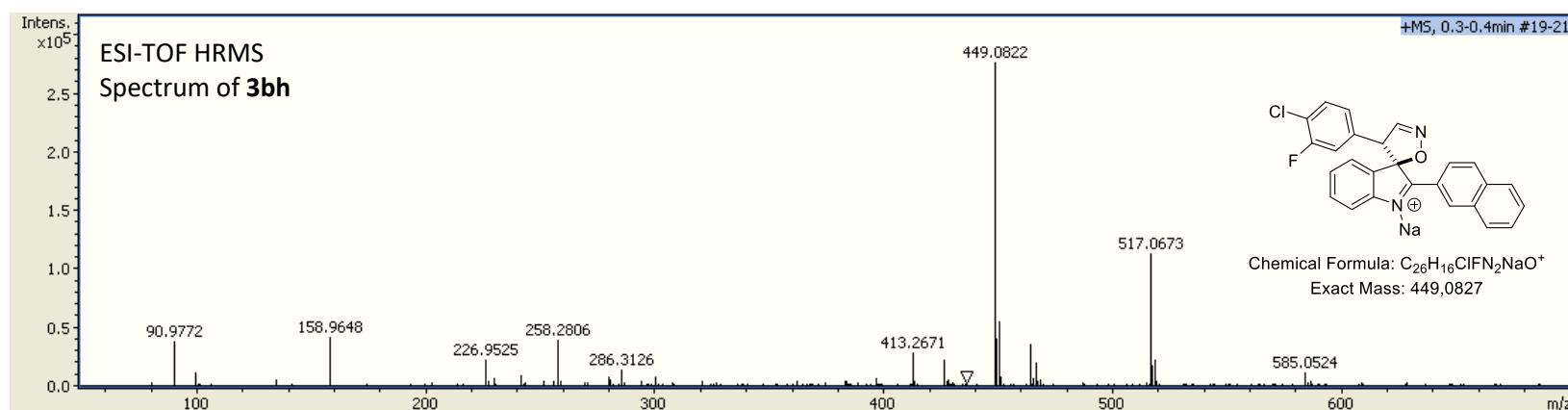
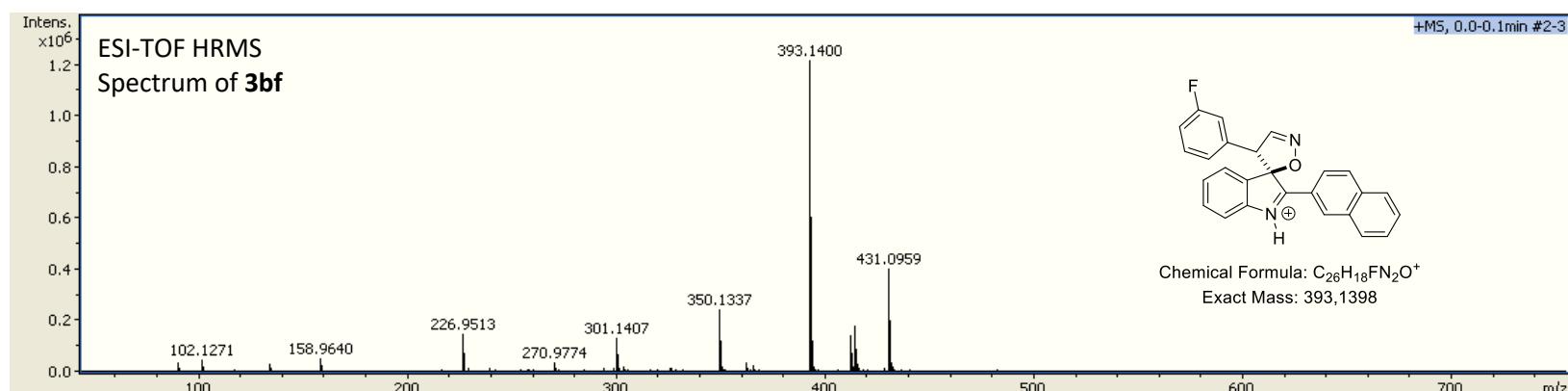
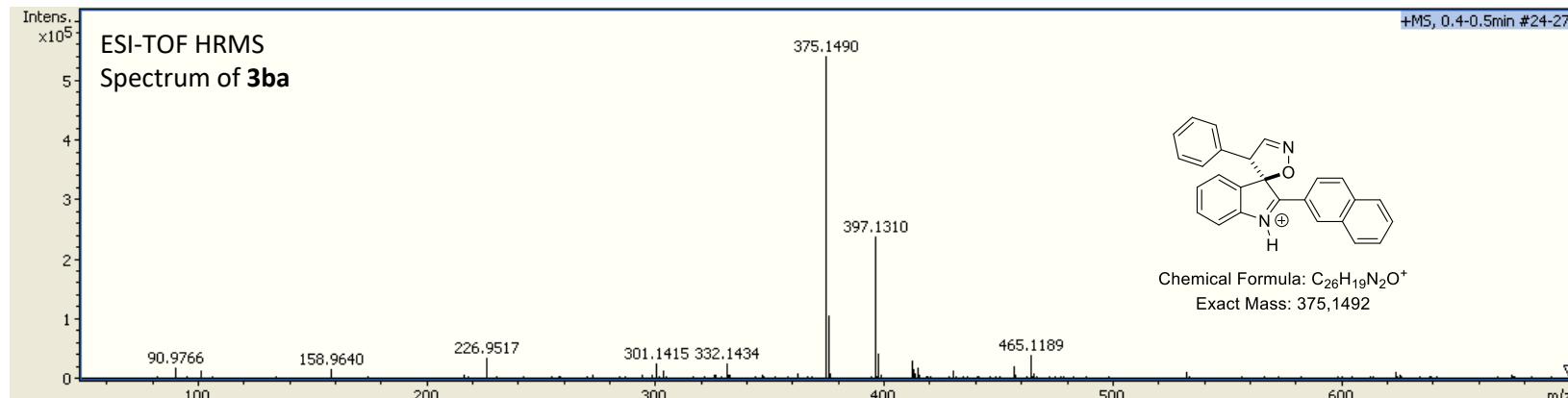
## HRMS Spectral Charts

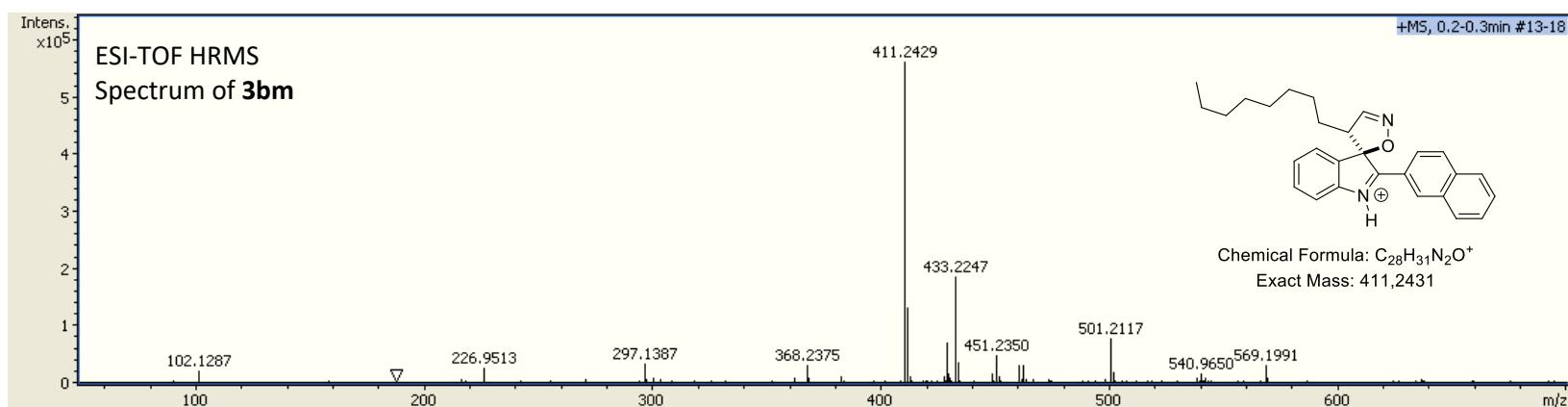
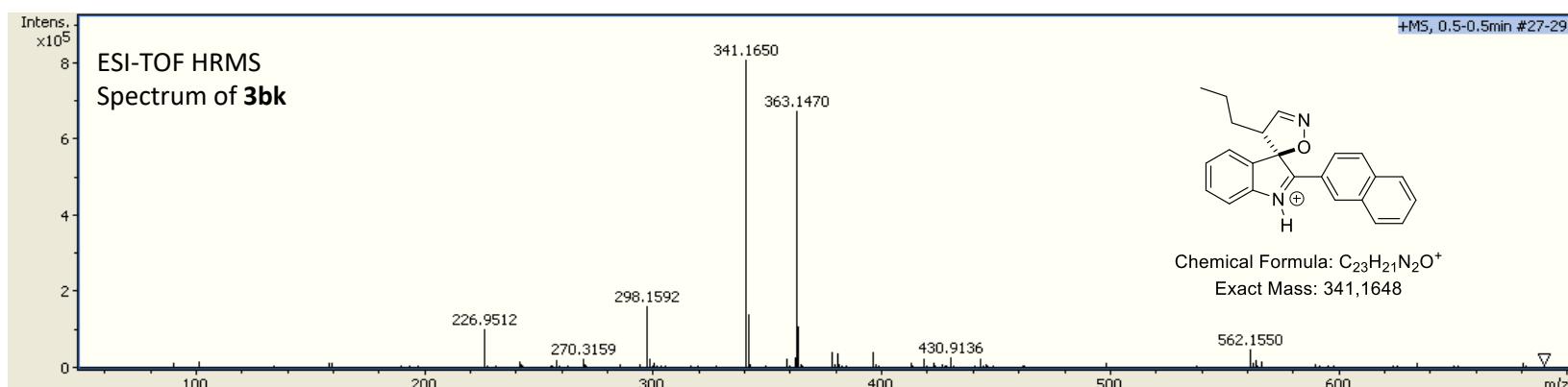
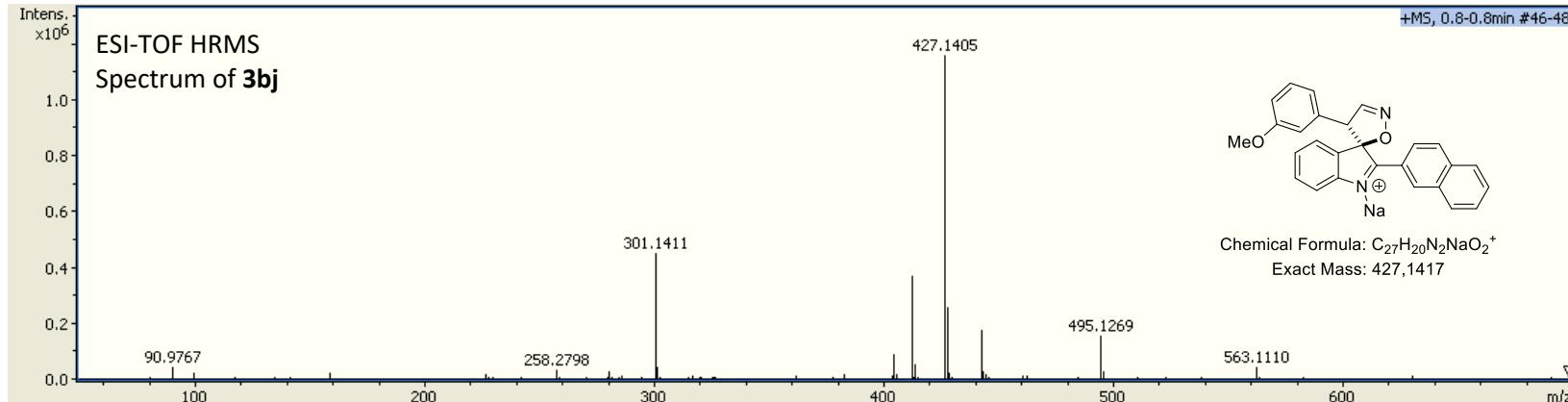


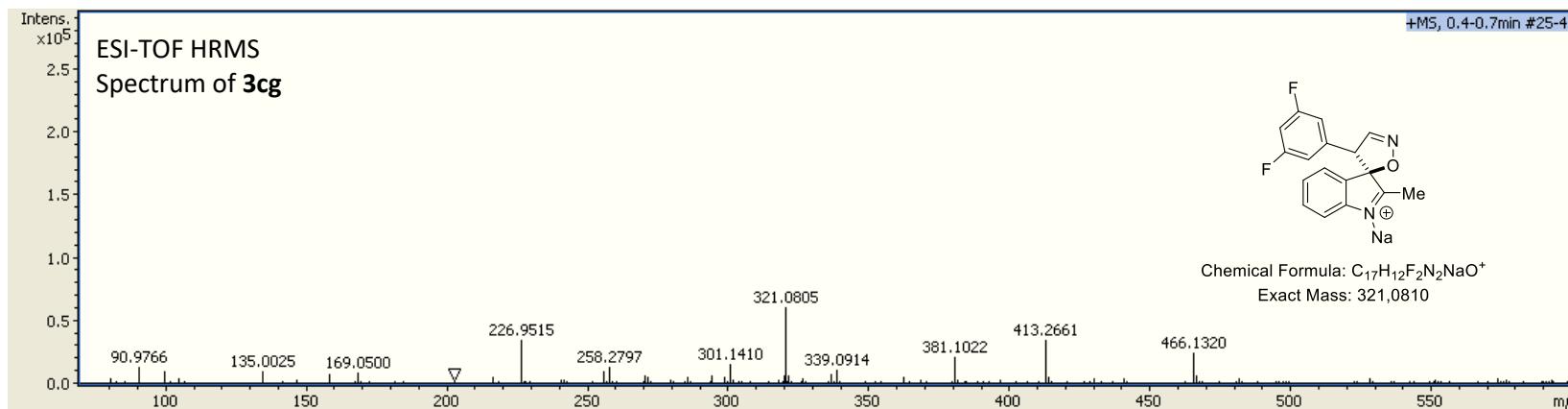
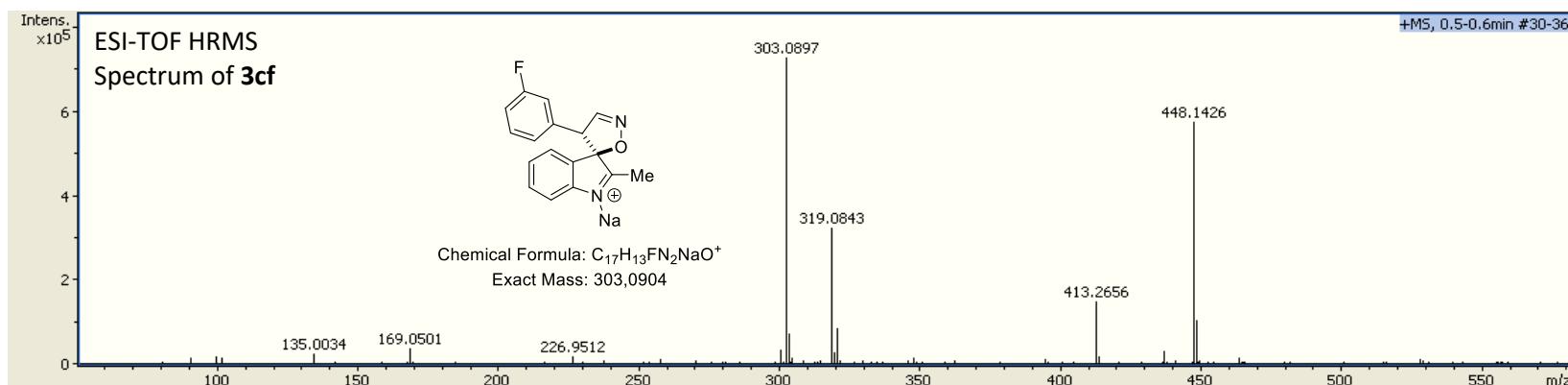
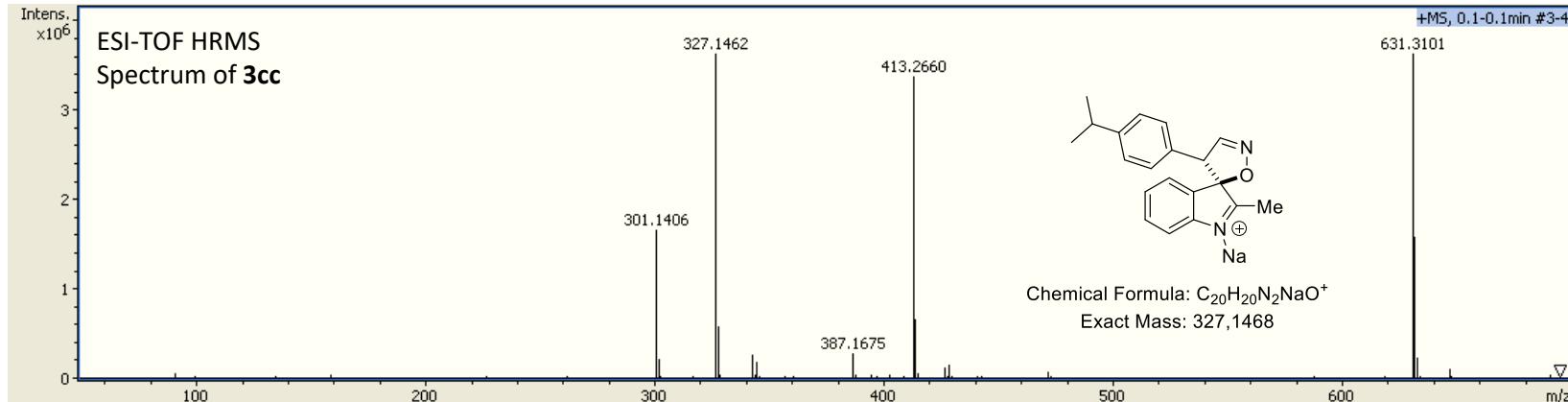


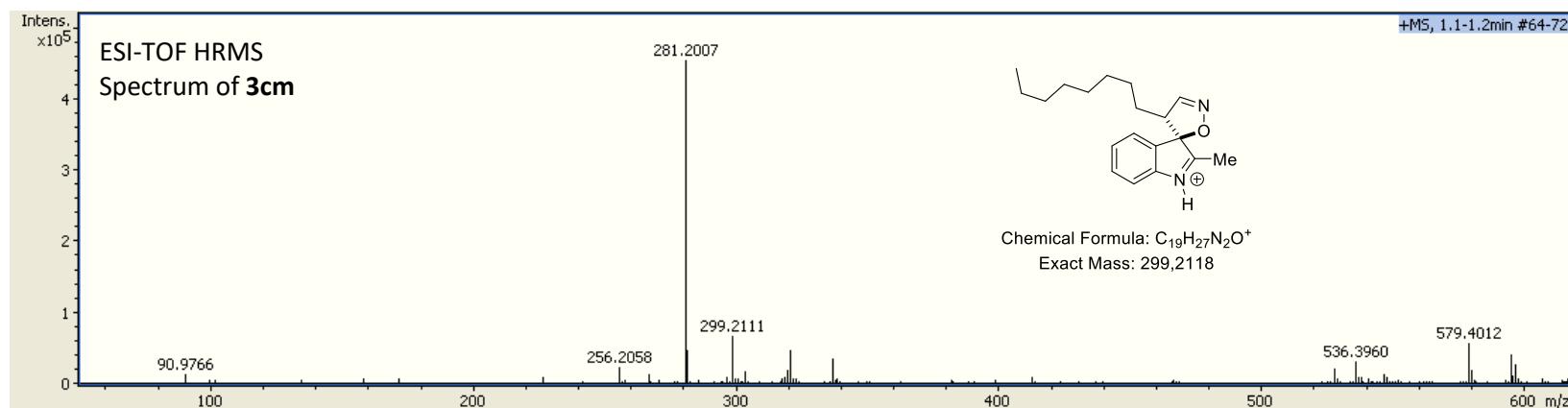
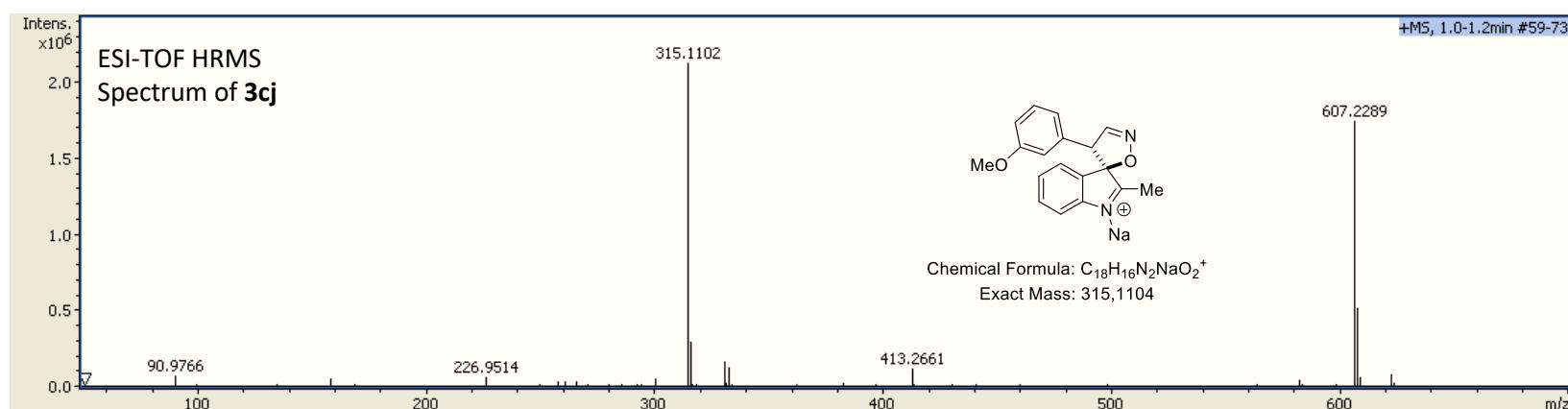
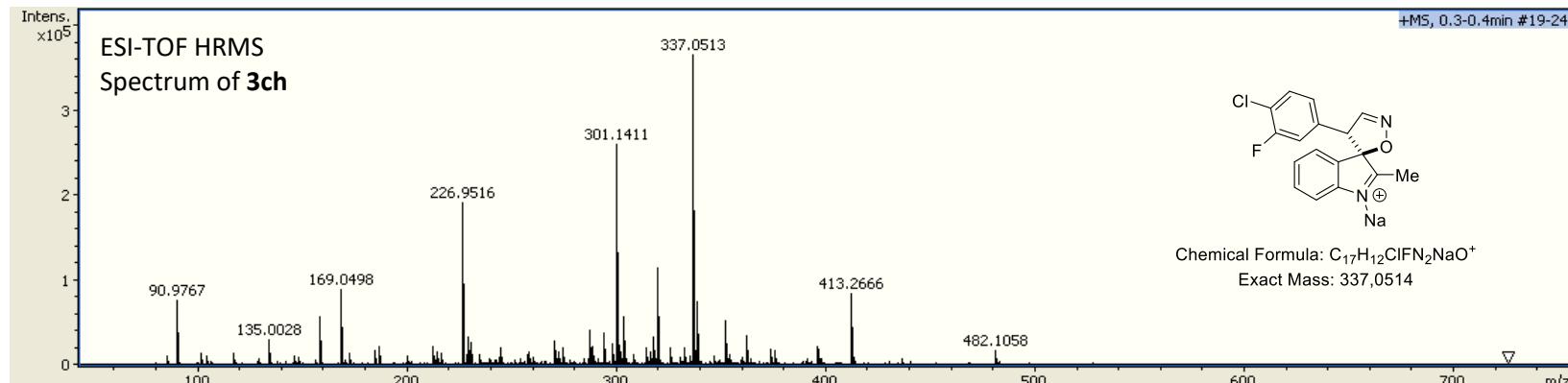


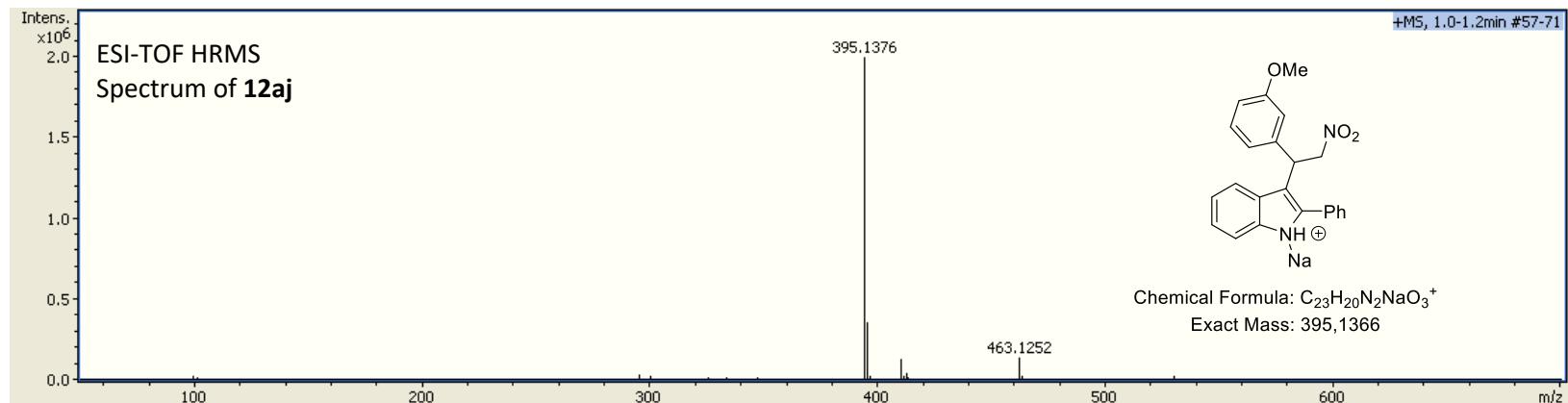
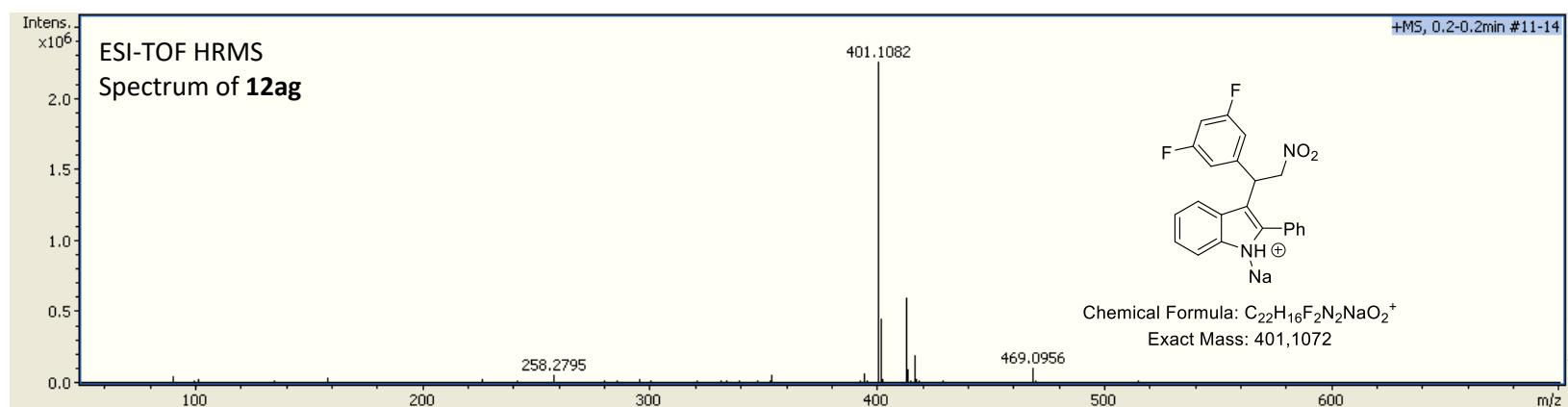
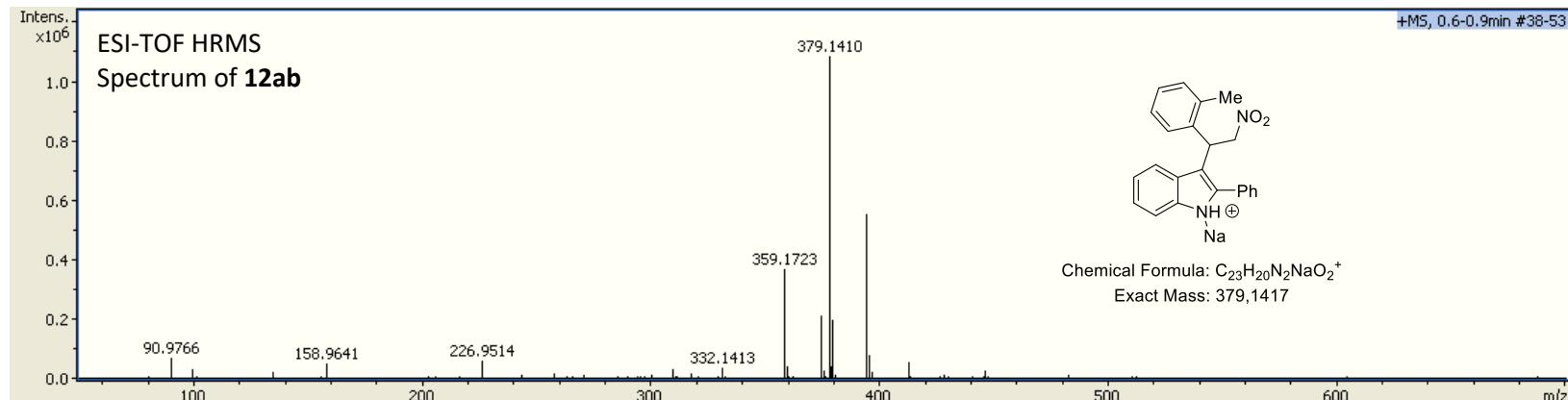


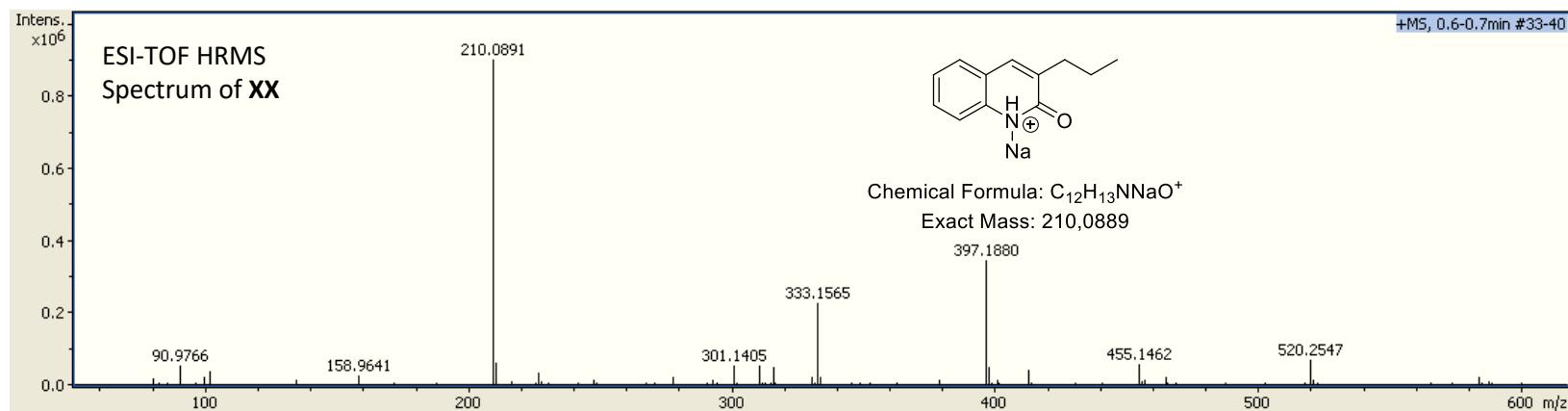
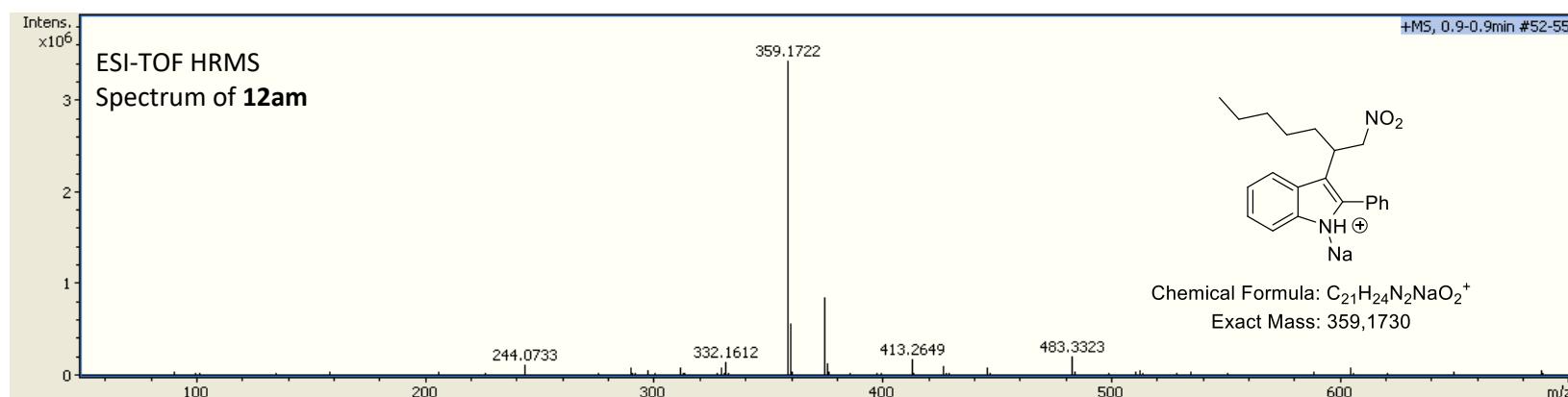
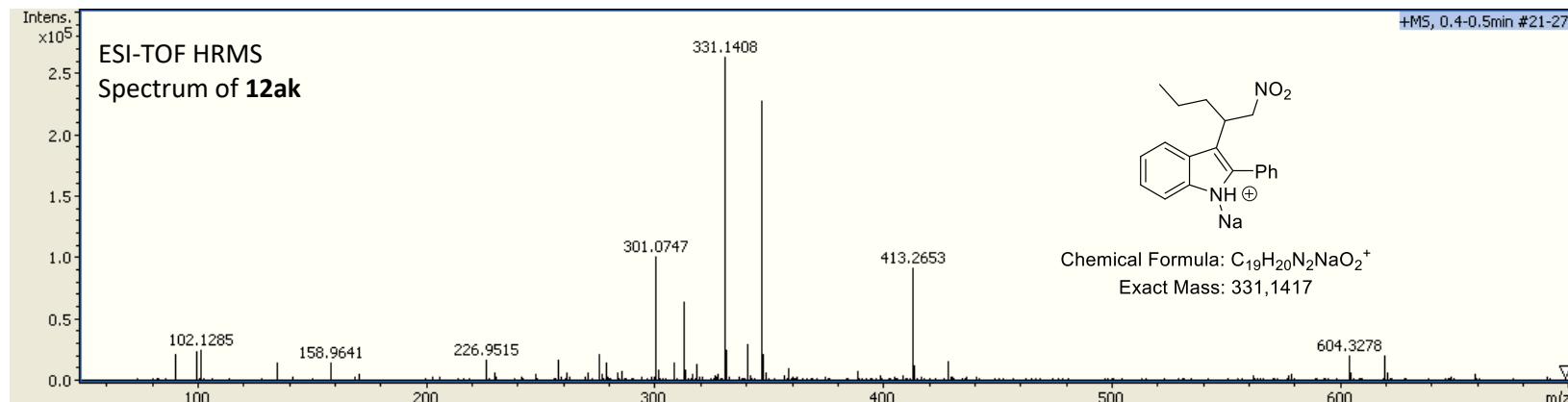




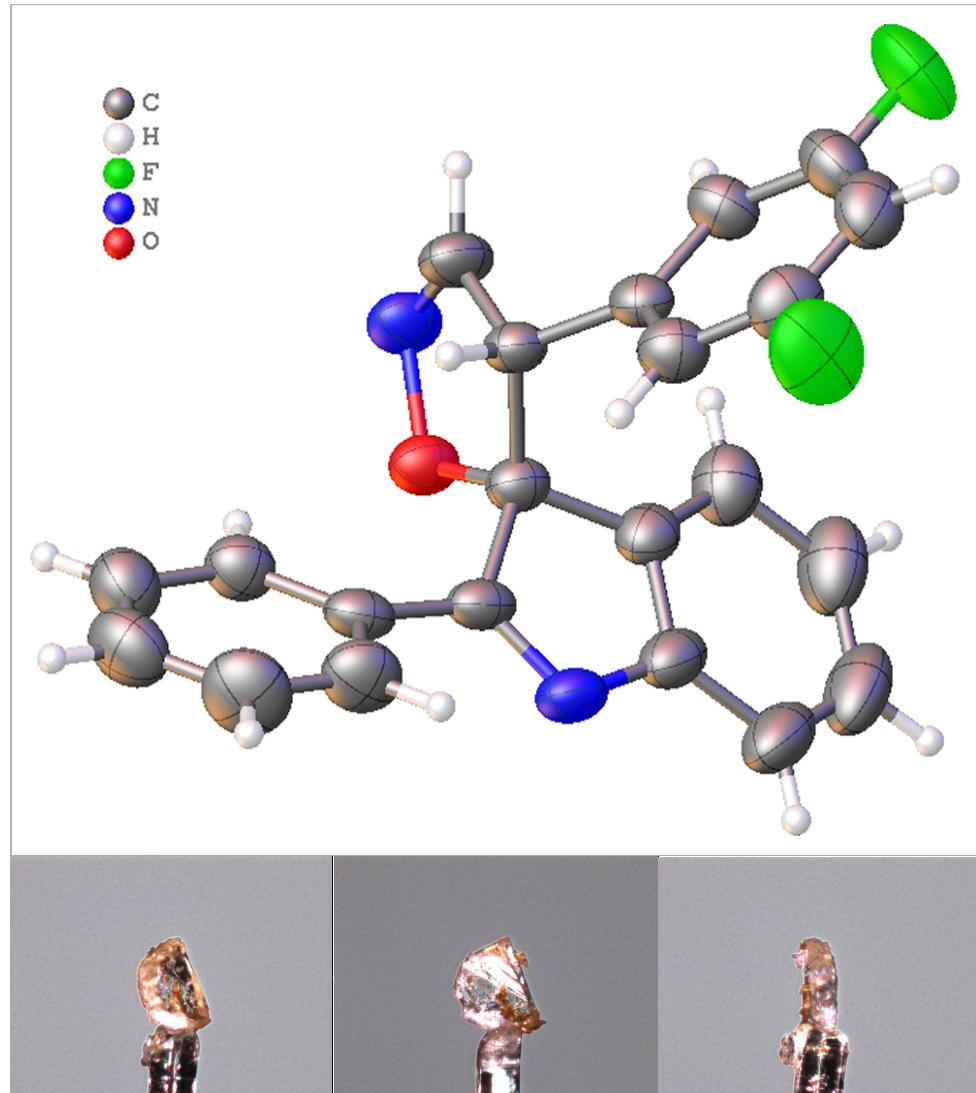








X-Ray Crystallography Data for compound 3ag



**Figure S1.** ORTEP drawing and sample microphotographs of the sample crystal

**Table S1. Crystal data and structure refinement for 3ag.**

Identification code	ANNA_ROS534_11
Empirical formula	C <sub>22</sub> H <sub>14</sub> F <sub>2</sub> N <sub>2</sub> O
Formula weight	360.35
Temperature/K	104(6)
Crystal system	monoclinic
Space group	P2 <sub>1</sub> /c
a/Å	20.4243(4)
b/Å	15.3838(2)
c/Å	11.4158(2)
α/°	90
β/°	103.3576(18)
γ/°	90
Volume/Å <sup>3</sup>	3489.86(10)
Z	8
ρ <sub>calc</sub> g/cm <sup>3</sup>	1.372
μ/mm <sup>-1</sup>	0.833
F(000)	1488.0
Crystal size/mm <sup>3</sup>	0.359 × 0.266 × 0.122
Radiation	CuKα ( $\lambda = 1.54184$ )
2Θ range for data collection/°	4.446 to 153.222
Index ranges	-25 ≤ h ≤ 25, -14 ≤ k ≤ 19, -14 ≤ l ≤ 13
Reflections collected	37744
Independent reflections	7322 [R <sub>int</sub> = 0.0317, R <sub>sigma</sub> = 0.0196]
Data/restraints/parameters	7322/0/542
Goodness-of-fit on F <sup>2</sup>	1.025
Final R indexes [I>=2σ (I)]	R <sub>1</sub> = 0.0439, wR <sub>2</sub> = 0.1158
Final R indexes [all data]	R <sub>1</sub> = 0.0507, wR <sub>2</sub> = 0.1230
Largest diff. peak/hole / e Å <sup>-3</sup>	0.15/-0.21

**Table S2 Fractional Atomic Coordinates ( $\times 10^4$ ) and Equivalent Isotropic Displacement Parameters ( $\text{\AA}^2 \times 10^3$ ) for 3ag.  $U_{\text{eq}}$  is defined as 1/3 of the trace of the orthogonalised  $U_{IJ}$  tensor.**

Atom	x	y	z	U(eq)
O <sub>(1)</sub>	7055.1(5)	4446.3(6)	1974.4(9)	47.3(2)
F <sub>(4)</sub>	5042.7(6)	3310.2(7)	7844.3(12)	82.6(3)
F <sub>(3)</sub>	4900.2(6)	6218.3(8)	8941.4(12)	84.2(3)
F <sub>(1)</sub>	9784.8(6)	6672.4(8)	4443.0(12)	89.1(4)
O <sub>(2)</sub>	7876.3(6)	5305.1(7)	7647.1(12)	64.3(3)
F <sub>(2)</sub>	9533.2(7)	4607.9(9)	7296.7(11)	94.2(4)
N <sub>(4)</sub>	7115.6(7)	7225.4(7)	7983.9(11)	53.1(3)
N <sub>(1)</sub>	7914.4(6)	2778.3(7)	3624.5(13)	54.1(3)
N <sub>(2)</sub>	6975.5(7)	5364.3(8)	1958.0(13)	56.2(3)
N <sub>(3)</sub>	7924.8(8)	4423.5(8)	8031.3(15)	65.7(4)
C <sub>(9)</sub>	7699.0(7)	5013.8(7)	3805.8(12)	41.5(3)
C <sub>(2)</sub>	7614.4(7)	4210.2(8)	2965.9(11)	41.0(3)
C <sub>(1)</sub>	7476.1(7)	3380.4(8)	3620.1(12)	43.9(3)
C <sub>(33)</sub>	6343.1(7)	4929.0(8)	8701.9(11)	44.4(3)
C <sub>(11)</sub>	8402.4(7)	5230.2(8)	4502.8(12)	41.1(3)
C <sub>(17)</sub>	6926.6(7)	3276.1(8)	4241.4(12)	44.8(3)
C <sub>(18)</sub>	6294.3(8)	3635.7(9)	3779.3(13)	50.8(3)
C <sub>(19)</sub>	5772.8(9)	3480.3(11)	4341.5(16)	60.8(4)
C <sub>(20)</sub>	5880.6(10)	2985.8(12)	5373.0(17)	68.3(4)
C <sub>(21)</sub>	6505.6(11)	2636.5(14)	5836.0(18)	76.2(5)
C <sub>(22)</sub>	7027.3(9)	2776.9(11)	5281.8(15)	62.0(4)
C <sub>(3)</sub>	8207.1(7)	3917.6(9)	2490.7(13)	46.8(3)
C <sub>(25)</sub>	6781.2(7)	5964.9(8)	6874.8(12)	46.0(3)
C <sub>(26)</sub>	6688.7(7)	6856.2(9)	6939.7(13)	48.4(3)
C <sub>(31)</sub>	7093.5(7)	5059.7(8)	8872.2(12)	45.8(3)
C <sub>(16)</sub>	8786.7(8)	5860.8(9)	4108.8(13)	50.5(3)
C <sub>(24)</sub>	7306.4(7)	5713.1(8)	7980.4(12)	45.4(3)
C <sub>(23)</sub>	7488.8(8)	6614.0(8)	8544.7(13)	48.0(3)
C <sub>(34)</sub>	6030.0(8)	4140.8(9)	8334.5(13)	51.1(3)
C <sub>(4)</sub>	8354.0(7)	3073.0(9)	2902.2(15)	53.7(3)

C <sub>(38)</sub>	5954.9(8)	5630.5(9)	8915.7(13)	52.2(3)
C <sub>(12)</sub>	8657.5(8)	4802.1(9)	5582.9(13)	50.9(3)
C <sub>(10)</sub>	7333.9(8)	5670.9(9)	2920.8(15)	55.2(4)
C <sub>(37)</sub>	5272.2(9)	5529.4(11)	8740.3(14)	58.0(4)
C <sub>(32)</sub>	7505.1(8)	4290.3(9)	8667.7(16)	58.9(4)
C <sub>(15)</sub>	9411.9(8)	6052.0(11)	4817.6(15)	57.7(4)
C <sub>(35)</sub>	5342.0(9)	4082.8(10)	8192.3(14)	57.5(4)
C <sub>(13)</sub>	9287.7(9)	5023.7(11)	6239.2(14)	59.5(4)
C <sub>(36)</sub>	4941.6(9)	4759.9(12)	8384.7(14)	60.7(4)
C <sub>(14)</sub>	9683.2(8)	5650.4(12)	5888.9(16)	62.1(4)
C <sub>(27)</sub>	6260.0(9)	7302.4(11)	6025.5(16)	61.6(4)
C <sub>(39)</sub>	7991.4(4)	6853(6)	9693(8)	40.1(13)
C <sub>(40)</sub>	7885(4)	7577(5)	10335(6)	46.7(12)
C <sub>(41)</sub>	8372(4)	7820(4)	11334(6)	59.7(14)
C <sub>(42)</sub>	8965(5)	7343(6)	11684(9)	69.5(17)
C <sub>(43)</sub>	9053(5)	6610(4)	11016(8)	71.3(18)
C <sub>(44)</sub>	8562(5)	6362(5)	10024(9)	62.1(15)
C <sub>(30)</sub>	6451.5(9)	5494.5(11)	5895.8(14)	60.3(4)
C <sub>(8)</sub>	8534.0(8)	4297.7(12)	1696.8(15)	60.8(4)
C <sub>(28)</sub>	5937.2(9)	6826.4(13)	5021.6(16)	69.8(5)
C <sub>(29)</sub>	6029.0(10)	5939.9(14)	4959.5(15)	71.8(5)
C <sub>(5)</sub>	8835.8(9)	2584.0(12)	2529(2)	73.1(5)
C <sub>(7)</sub>	9013.7(9)	3800.7(16)	1304.9(18)	76.8(5)
C <sub>(6)</sub>	9159.6(9)	2961.6(16)	1718(2)	82.0(6)
C <sub>(50)</sub>	8054(7)	7312(11)	10387(11)	71(3)
C <sub>(49)</sub>	8591(9)	7498(14)	11364(9)	88(5)
C <sub>(48)</sub>	9153(10)	7018(16)	11470(12)	98(5)
C <sub>(47)</sub>	9224(7)	6367(13)	10664(15)	97(4)
C <sub>(46)</sub>	8682(7)	6198(11)	9712(16)	78(3)
C <sub>(45)</sub>	8067(9)	6665(10)	9511(14)	60(3)

**Table S3. Anisotropic Displacement Parameters ( $\text{\AA}^2 \times 10^3$ ) for 3ag. The Anisotropic displacement factor exponent takes the form: -  
 $2\pi^2[\mathbf{h}^2\mathbf{a}^{*2}\mathbf{U}_{11}+2\mathbf{hka}^{*}\mathbf{b}^{*}\mathbf{U}_{12}+\dots]$ .**

Atom	$\mathbf{U}_{11}$	$\mathbf{U}_{22}$	$\mathbf{U}_{33}$	$\mathbf{U}_{23}$	$\mathbf{U}_{13}$	$\mathbf{U}_{12}$
O <sub>(1)</sub>	52.4(5)	37.4(5)	47.0(5)	2.4(4)	1.2(4)	2.9(4)
F <sub>(4)</sub>	78.8(7)	65.0(6)	94.6(8)	-2.6(5)	0.4(6)	-26.5(5)
F <sub>(3)</sub>	80.2(7)	80.5(7)	95.7(8)	-2.6(6)	28.2(6)	27.7(6)
F <sub>(1)</sub>	78.2(7)	94.5(8)	95.4(8)	5.5(7)	21.4(6)	-43.3(6)
O <sub>(2)</sub>	73.6(7)	49.3(6)	77.8(7)	12.6(5)	33.2(6)	16.2(5)
F <sub>(2)</sub>	88.5(8)	110.4(10)	67.3(7)	26.4(6)	-15.8(6)	0.3(7)
N <sub>(4)</sub>	67.4(8)	32.4(5)	54.9(7)	0.8(5)	4.6(6)	-4.4(5)
N <sub>(1)</sub>	55.7(7)	31.2(5)	70.8(8)	-2.3(5)	5.4(6)	4.8(5)
N <sub>(2)</sub>	59.0(7)	37.8(6)	65.6(8)	10.8(5)	1.5(6)	5.3(5)
N <sub>(3)</sub>	71.9(9)	41.3(7)	84.9(10)	1.9(6)	20.1(8)	13.6(6)
C <sub>(9)</sub>	46.8(7)	28.3(5)	48.9(7)	-1.1(5)	9.9(5)	1.3(5)
C <sub>(2)</sub>	45.1(6)	31.9(6)	43.2(6)	-0.6(5)	4.2(5)	2.4(5)
C <sub>(1)</sub>	50.4(7)	27.7(5)	47.7(7)	-2.2(5)	-0.7(5)	-0.3(5)
C <sub>(33)</sub>	57.8(8)	36.6(6)	36.8(6)	1.9(5)	6.8(5)	-0.8(5)
C <sub>(11)</sub>	47.6(7)	31.3(5)	44.6(6)	-4.7(5)	11.1(5)	0.2(5)
C <sub>(17)</sub>	54.3(7)	28.5(5)	47.1(7)	-1.2(5)	2.6(6)	-5.2(5)
C <sub>(18)</sub>	54.8(8)	44.6(7)	50.1(7)	4.8(6)	6.5(6)	-1.7(6)
C <sub>(19)</sub>	55.8(8)	61.8(9)	63.4(9)	0.0(7)	11.0(7)	-6.0(7)
C <sub>(20)</sub>	75.4(11)	66.5(10)	67.1(10)	3.7(8)	24.7(9)	-15.6(8)
C <sub>(21)</sub>	91.5(13)	73.8(11)	63.7(10)	24.8(9)	18.7(10)	-3.1(10)
C <sub>(22)</sub>	70.4(10)	52.4(8)	58.4(9)	15.2(7)	5.2(7)	2.6(7)
C <sub>(3)</sub>	47.1(7)	42.8(7)	48.5(7)	-10.5(5)	6.9(6)	0.9(5)
C <sub>(25)</sub>	58.3(8)	36.5(6)	43.0(6)	1.4(5)	11.7(6)	-2.9(5)
C <sub>(26)</sub>	54.2(7)	37.9(6)	50.5(7)	2.7(5)	6.9(6)	0.3(5)
C <sub>(31)</sub>	58.2(8)	30.8(6)	45.6(7)	3.1(5)	6.1(6)	-0.1(5)
C <sub>(16)</sub>	56.9(8)	44.5(7)	50.5(7)	0.8(6)	13.1(6)	-7.9(6)
C <sub>(24)</sub>	53.8(7)	32.9(6)	49.2(7)	2.1(5)	11.4(6)	2.5(5)
C <sub>(23)</sub>	56.8(8)	36.6(6)	48.1(7)	3.3(5)	7.2(6)	-7.3(5)
C <sub>(34)</sub>	63.6(9)	38.1(6)	48.5(7)	0.6(5)	6.4(6)	-1.7(6)
C <sub>(4)</sub>	49.2(7)	41.5(7)	65.2(9)	-13.1(6)	2.5(6)	6.0(6)
C <sub>(38)</sub>	66.4(9)	41.8(7)	48.0(7)	-2.3(5)	12.6(6)	2.8(6)

C <sub>(12)</sub>	57.6(8)	44.6(7)	49.1(7)	1.8(6)	9.3(6)	-0.2(6)
C <sub>(10)</sub>	59.1(8)	31.0(6)	69.6(9)	2.3(6)	2.5(7)	4.8(6)
C <sub>(37)</sub>	65.1(9)	60.3(9)	49.1(8)	4.6(6)	14.2(7)	13.5(7)
C <sub>(32)</sub>	63.6(9)	32.9(6)	77.5(10)	7.0(6)	11.1(8)	5.3(6)
C <sub>(15)</sub>	56.1(8)	55.2(8)	64.7(9)	-7.6(7)	19.9(7)	-15.4(7)
C <sub>(35)</sub>	64.5(9)	53.2(8)	49.1(7)	4.4(6)	1.7(7)	-12.4(7)
C <sub>(13)</sub>	62.6(9)	62.5(9)	48.2(8)	1.6(6)	2.2(7)	7.4(7)
C <sub>(36)</sub>	56.1(8)	73.9(10)	49.8(8)	11.9(7)	7.5(6)	-1.2(7)
C <sub>(14)</sub>	51.6(8)	68.5(10)	61.9(9)	-15.2(8)	4.4(7)	-2.5(7)
C <sub>(27)</sub>	63.2(9)	50.8(8)	66.3(10)	13.1(7)	5.8(8)	6.3(7)
C <sub>(39)</sub>	54(2)	29(3)	35(3)	-4(2)	4.4(17)	-9.8(19)
C <sub>(40)</sub>	54(2)	41(2)	45.2(18)	-4.4(16)	10.7(16)	-5.9(15)
C <sub>(41)</sub>	72(3)	55(2)	52.0(18)	-5.3(18)	13.9(19)	-16.0(19)
C <sub>(42)</sub>	71(4)	72(3)	55(3)	4(3)	-7(2)	-23(3)
C <sub>(43)</sub>	64(3)	63(3)	78(4)	7(3)	-5(2)	2(2)
C <sub>(44)</sub>	66(4)	46(2)	67(3)	4.4(19)	0(2)	5.8(19)
C <sub>(30)</sub>	81.4(11)	50.3(8)	47.4(8)	-5.6(6)	11.2(7)	-10.3(7)
C <sub>(8)</sub>	56.4(8)	71.7(10)	54.9(8)	-4.8(7)	13.8(7)	0.8(7)
C <sub>(28)</sub>	61.4(9)	82.0(12)	58.0(9)	19.2(8)	-2.5(7)	-2.3(8)
C <sub>(29)</sub>	80.5(12)	81.4(12)	47.3(8)	-2.1(8)	1.8(8)	-18.0(9)
C <sub>(5)</sub>	59.2(9)	59.8(10)	95.0(13)	-24.9(9)	6.9(9)	14.7(8)
C <sub>(7)</sub>	58.0(10)	108.8(16)	66.5(10)	-19.5(10)	20.2(8)	2.9(10)
C <sub>(6)</sub>	54.2(9)	99.7(15)	90.4(14)	-39.5(12)	13.2(9)	15.1(10)
C <sub>(50)</sub>	71(6)	69(7)	69(4)	9(5)	9(4)	-15(4)
C <sub>(49)</sub>	98(10)	107(10)	52(4)	-14(6)	6(5)	-50(9)
C <sub>(48)</sub>	97(9)	126(14)	59(6)	18(7)	-8(5)	-53(8)
C <sub>(47)</sub>	71(5)	114(9)	91(7)	34(6)	-13(4)	-10(5)
C <sub>(46)</sub>	58(5)	74(7)	88(8)	23(5)	-14(4)	-6(4)
C <sub>(45)</sub>	92(6)	38(5)	49(4)	-7(3)	17(4)	-21(4)

**Table S4. Bond Lengths for 3ag.**

<b>Atom</b>	<b>Atom</b>	<b>Length/Å</b>	<b>Atom</b>	<b>Atom</b>	<b>Length/Å</b>
O <sub>(1)</sub>	N <sub>(2)</sub>	1.4211(15)	C <sub>(25)</sub>	C <sub>(30)</sub>	1.371(2)
O <sub>(1)</sub>	C <sub>(2)</sub>	1.4562(15)	C <sub>(26)</sub>	C <sub>(27)</sub>	1.380(2)
F <sub>(4)</sub>	C <sub>(35)</sub>	1.3534(18)	C <sub>(31)</sub>	C <sub>(24)</sub>	1.5622(18)
F <sub>(3)</sub>	C <sub>(37)</sub>	1.3542(19)	C <sub>(31)</sub>	C <sub>(32)</sub>	1.501(2)
F <sub>(1)</sub>	C <sub>(15)</sub>	1.3506(18)	C <sub>(16)</sub>	C <sub>(15)</sub>	1.376(2)
O <sub>(2)</sub>	N <sub>(3)</sub>	1.4218(17)	C <sub>(24)</sub>	C <sub>(23)</sub>	1.5374(18)
O <sub>(2)</sub>	C <sub>(24)</sub>	1.4484(18)	C <sub>(23)</sub>	C <sub>(39)</sub>	1.511(8)
F <sub>(2)</sub>	C <sub>(13)</sub>	1.3559(19)	C <sub>(23)</sub>	C <sub>(45)</sub>	1.420(18)
N <sub>(4)</sub>	C <sub>(26)</sub>	1.4226(18)	C <sub>(34)</sub>	C <sub>(35)</sub>	1.380(2)
N <sub>(4)</sub>	C <sub>(23)</sub>	1.2845(19)	C <sub>(4)</sub>	C <sub>(5)</sub>	1.381(2)
N <sub>(1)</sub>	C <sub>(1)</sub>	1.2872(18)	C <sub>(38)</sub>	C <sub>(37)</sub>	1.371(2)
N <sub>(1)</sub>	C <sub>(4)</sub>	1.426(2)	C <sub>(12)</sub>	C <sub>(13)</sub>	1.374(2)
N <sub>(2)</sub>	C <sub>(10)</sub>	1.263(2)	C <sub>(37)</sub>	C <sub>(36)</sub>	1.377(2)
N <sub>(3)</sub>	C <sub>(32)</sub>	1.262(2)	C <sub>(15)</sub>	C <sub>(14)</sub>	1.368(3)
C <sub>(9)</sub>	C <sub>(2)</sub>	1.5495(16)	C <sub>(35)</sub>	C <sub>(36)</sub>	1.373(3)
C <sub>(9)</sub>	C <sub>(11)</sub>	1.5099(18)	C <sub>(13)</sub>	C <sub>(14)</sub>	1.375(3)
C <sub>(9)</sub>	C <sub>(10)</sub>	1.5002(18)	C <sub>(27)</sub>	C <sub>(28)</sub>	1.391(3)
C <sub>(2)</sub>	C <sub>(1)</sub>	1.5377(17)	C <sub>(39)</sub>	C <sub>(40)</sub>	1.379(7)
C <sub>(2)</sub>	C <sub>(3)</sub>	1.5052(19)	C <sub>(39)</sub>	C <sub>(44)</sub>	1.367(11)
C <sub>(1)</sub>	C <sub>(17)</sub>	1.468(2)	C <sub>(40)</sub>	C <sub>(41)</sub>	1.381(11)
C <sub>(33)</sub>	C <sub>(31)</sub>	1.513(2)	C <sub>(41)</sub>	C <sub>(42)</sub>	1.393(9)
C <sub>(33)</sub>	C <sub>(34)</sub>	1.3896(19)	C <sub>(42)</sub>	C <sub>(43)</sub>	1.396(9)
C <sub>(33)</sub>	C <sub>(38)</sub>	1.394(2)	C <sub>(43)</sub>	C <sub>(44)</sub>	1.383(12)
C <sub>(11)</sub>	C <sub>(16)</sub>	1.3866(19)	C <sub>(30)</sub>	C <sub>(29)</sub>	1.389(3)
C <sub>(11)</sub>	C <sub>(12)</sub>	1.3888(19)	C <sub>(8)</sub>	C <sub>(7)</sub>	1.396(3)
C <sub>(17)</sub>	C <sub>(18)</sub>	1.392(2)	C <sub>(28)</sub>	C <sub>(29)</sub>	1.381(3)
C <sub>(17)</sub>	C <sub>(22)</sub>	1.389(2)	C <sub>(5)</sub>	C <sub>(6)</sub>	1.384(3)
C <sub>(18)</sub>	C <sub>(19)</sub>	1.385(2)	C <sub>(7)</sub>	C <sub>(6)</sub>	1.383(3)
C <sub>(19)</sub>	C <sub>(20)</sub>	1.376(3)	C <sub>(50)</sub>	C <sub>(49)</sub>	1.402(17)
C <sub>(20)</sub>	C <sub>(21)</sub>	1.373(3)	C <sub>(50)</sub>	C <sub>(45)</sub>	1.414(12)
C <sub>(21)</sub>	C <sub>(22)</sub>	1.376(3)	C <sub>(49)</sub>	C <sub>(48)</sub>	1.346(16)
C <sub>(3)</sub>	C <sub>(4)</sub>	1.391(2)	C <sub>(48)</sub>	C <sub>(47)</sub>	1.390(14)

C <sub>(3)</sub>	C <sub>(8)</sub>	1.374(2)	C <sub>(47)</sub>	C <sub>(46)</sub>	1.385(17)
C <sub>(25)</sub>	C <sub>(26)</sub>	1.3884(19)	C <sub>(46)</sub>	C <sub>(45)</sub>	1.419(17)
C <sub>(25)</sub>	C <sub>(24)</sub>	1.5060(19)			

**Table S5. Bond Angles for 3ag.**

Atom	Atom	Atom	Angle/ <sup>°</sup>	Atom	Atom	Atom	Angle/ <sup>°</sup>
N <sub>(2)</sub>	O <sub>(1)</sub>	C <sub>(2)</sub>	108.97(9)	N <sub>(4)</sub>	C <sub>(23)</sub>	C <sub>(24)</sub>	113.13(12)
N <sub>(3)</sub>	O <sub>(2)</sub>	C <sub>(24)</sub>	109.59(11)	N <sub>(4)</sub>	C <sub>(23)</sub>	C <sub>(39)</sub>	117.4(4)
C <sub>(23)</sub>	N <sub>(4)</sub>	C <sub>(26)</sub>	107.67(11)	N <sub>(4)</sub>	C <sub>(23)</sub>	C <sub>(45)</sub>	129.6(7)
C <sub>(1)</sub>	N <sub>(1)</sub>	C <sub>(4)</sub>	107.40(12)	C <sub>(39)</sub>	C <sub>(23)</sub>	C <sub>(24)</sub>	129.2(4)
C <sub>(10)</sub>	N <sub>(2)</sub>	O <sub>(1)</sub>	108.55(11)	C <sub>(45)</sub>	C <sub>(23)</sub>	C <sub>(24)</sub>	116.9(7)
C <sub>(32)</sub>	N <sub>(3)</sub>	O <sub>(2)</sub>	108.93(12)	C <sub>(35)</sub>	C <sub>(34)</sub>	C <sub>(33)</sub>	118.50(14)
C <sub>(11)</sub>	C <sub>(9)</sub>	C <sub>(2)</sub>	117.29(11)	C <sub>(3)</sub>	C <sub>(4)</sub>	N <sub>(1)</sub>	112.13(13)
C <sub>(10)</sub>	C <sub>(9)</sub>	C <sub>(2)</sub>	98.85(11)	C <sub>(5)</sub>	C <sub>(4)</sub>	N <sub>(1)</sub>	126.40(16)
C <sub>(10)</sub>	C <sub>(9)</sub>	C <sub>(11)</sub>	118.08(11)	C <sub>(5)</sub>	C <sub>(4)</sub>	C <sub>(3)</sub>	121.21(18)
O <sub>(1)</sub>	C <sub>(2)</sub>	C <sub>(9)</sub>	103.72(9)	C <sub>(37)</sub>	C <sub>(38)</sub>	C <sub>(33)</sub>	119.15(14)
O <sub>(1)</sub>	C <sub>(2)</sub>	C <sub>(1)</sub>	112.67(10)	C <sub>(13)</sub>	C <sub>(12)</sub>	C <sub>(11)</sub>	118.57(14)
O <sub>(1)</sub>	C <sub>(2)</sub>	C <sub>(3)</sub>	110.25(11)	N <sub>(2)</sub>	C <sub>(10)</sub>	C <sub>(9)</sub>	115.62(12)
C <sub>(1)</sub>	C <sub>(2)</sub>	C <sub>(9)</sub>	111.63(11)	F <sub>(3)</sub>	C <sub>(37)</sub>	C <sub>(38)</sub>	118.55(15)
C <sub>(3)</sub>	C <sub>(2)</sub>	C <sub>(9)</sub>	118.53(11)	F <sub>(3)</sub>	C <sub>(37)</sub>	C <sub>(36)</sub>	117.83(16)
C <sub>(3)</sub>	C <sub>(2)</sub>	C <sub>(1)</sub>	100.37(10)	C <sub>(38)</sub>	C <sub>(37)</sub>	C <sub>(36)</sub>	123.62(15)
N <sub>(1)</sub>	C <sub>(1)</sub>	C <sub>(2)</sub>	113.19(13)	N <sub>(3)</sub>	C <sub>(32)</sub>	C <sub>(31)</sub>	116.16(13)
N <sub>(1)</sub>	C <sub>(1)</sub>	C <sub>(17)</sub>	121.86(12)	F <sub>(1)</sub>	C <sub>(15)</sub>	C <sub>(16)</sub>	118.64(15)
C <sub>(17)</sub>	C <sub>(1)</sub>	C <sub>(2)</sub>	124.88(11)	F <sub>(1)</sub>	C <sub>(15)</sub>	C <sub>(14)</sub>	117.42(15)
C <sub>(34)</sub>	C <sub>(33)</sub>	C <sub>(31)</sub>	122.45(13)	C <sub>(14)</sub>	C <sub>(15)</sub>	C <sub>(16)</sub>	123.93(15)
C <sub>(34)</sub>	C <sub>(33)</sub>	C <sub>(38)</sub>	119.20(14)	F <sub>(4)</sub>	C <sub>(35)</sub>	C <sub>(34)</sub>	118.03(15)
C <sub>(38)</sub>	C <sub>(33)</sub>	C <sub>(31)</sub>	118.35(12)	F <sub>(4)</sub>	C <sub>(35)</sub>	C <sub>(36)</sub>	117.91(16)
C <sub>(16)</sub>	C <sub>(11)</sub>	C <sub>(9)</sub>	121.63(12)	C <sub>(36)</sub>	C <sub>(35)</sub>	C <sub>(34)</sub>	124.06(15)
C <sub>(16)</sub>	C <sub>(11)</sub>	C <sub>(12)</sub>	119.70(13)	F <sub>(2)</sub>	C <sub>(13)</sub>	C <sub>(12)</sub>	118.39(16)
C <sub>(12)</sub>	C <sub>(11)</sub>	C <sub>(9)</sub>	118.64(12)	F <sub>(2)</sub>	C <sub>(13)</sub>	C <sub>(14)</sub>	117.93(15)
C <sub>(18)</sub>	C <sub>(17)</sub>	C <sub>(1)</sub>	121.29(12)	C <sub>(12)</sub>	C <sub>(13)</sub>	C <sub>(14)</sub>	123.68(15)
C <sub>(22)</sub>	C <sub>(17)</sub>	C <sub>(1)</sub>	119.65(14)	C <sub>(35)</sub>	C <sub>(36)</sub>	C <sub>(37)</sub>	115.47(15)

C <sub>(22)</sub>	C <sub>(17)</sub>	C <sub>(18)</sub>	119.00(15)	C <sub>(15)</sub>	C <sub>(14)</sub>	C <sub>(13)</sub>	115.64(14)
C <sub>(19)</sub>	C <sub>(18)</sub>	C <sub>(17)</sub>	120.08(14)	C <sub>(26)</sub>	C <sub>(27)</sub>	C <sub>(28)</sub>	117.27(15)
C <sub>(20)</sub>	C <sub>(19)</sub>	C <sub>(18)</sub>	120.24(16)	C <sub>(40)</sub>	C <sub>(39)</sub>	C <sub>(23)</sub>	119.9(7)
C <sub>(21)</sub>	C <sub>(20)</sub>	C <sub>(19)</sub>	119.76(17)	C <sub>(44)</sub>	C <sub>(39)</sub>	C <sub>(23)</sub>	118.1(6)
C <sub>(20)</sub>	C <sub>(21)</sub>	C <sub>(22)</sub>	120.75(16)	C <sub>(44)</sub>	C <sub>(39)</sub>	C <sub>(40)</sub>	121.9(7)
C <sub>(21)</sub>	C <sub>(22)</sub>	C <sub>(17)</sub>	120.16(16)	C <sub>(39)</sub>	C <sub>(40)</sub>	C <sub>(41)</sub>	119.1(5)
C <sub>(4)</sub>	C <sub>(3)</sub>	C <sub>(2)</sub>	106.64(13)	C <sub>(40)</sub>	C <sub>(41)</sub>	C <sub>(42)</sub>	120.4(5)
C <sub>(8)</sub>	C <sub>(3)</sub>	C <sub>(2)</sub>	131.62(14)	C <sub>(41)</sub>	C <sub>(42)</sub>	C <sub>(43)</sub>	118.9(6)
C <sub>(8)</sub>	C <sub>(3)</sub>	C <sub>(4)</sub>	121.36(15)	C <sub>(44)</sub>	C <sub>(43)</sub>	C <sub>(42)</sub>	120.6(6)
C <sub>(26)</sub>	C <sub>(25)</sub>	C <sub>(24)</sub>	106.78(12)	C <sub>(39)</sub>	C <sub>(44)</sub>	C <sub>(43)</sub>	119.1(6)
C <sub>(30)</sub>	C <sub>(25)</sub>	C <sub>(26)</sub>	121.17(14)	C <sub>(25)</sub>	C <sub>(30)</sub>	C <sub>(29)</sub>	117.89(15)
C <sub>(30)</sub>	C <sub>(25)</sub>	C <sub>(24)</sub>	131.93(13)	C <sub>(3)</sub>	C <sub>(8)</sub>	C <sub>(7)</sub>	117.55(18)
C <sub>(25)</sub>	C <sub>(26)</sub>	N <sub>(4)</sub>	111.98(12)	C <sub>(29)</sub>	C <sub>(28)</sub>	C <sub>(27)</sub>	121.21(15)
C <sub>(27)</sub>	C <sub>(26)</sub>	N <sub>(4)</sub>	126.44(13)	C <sub>(28)</sub>	C <sub>(29)</sub>	C <sub>(30)</sub>	121.01(16)
C <sub>(27)</sub>	C <sub>(26)</sub>	C <sub>(25)</sub>	121.42(14)	C <sub>(4)</sub>	C <sub>(5)</sub>	C <sub>(6)</sub>	117.52(18)
C <sub>(33)</sub>	C <sub>(31)</sub>	C <sub>(24)</sub>	115.34(11)	C <sub>(6)</sub>	C <sub>(7)</sub>	C <sub>(8)</sub>	120.9(2)
C <sub>(32)</sub>	C <sub>(31)</sub>	C <sub>(33)</sub>	117.53(12)	C <sub>(7)</sub>	C <sub>(6)</sub>	C <sub>(5)</sub>	121.44(17)
C <sub>(32)</sub>	C <sub>(31)</sub>	C <sub>(24)</sub>	98.96(12)	C <sub>(49)</sub>	C <sub>(50)</sub>	C <sub>(45)</sub>	124.6(11)
C <sub>(15)</sub>	C <sub>(16)</sub>	C <sub>(11)</sub>	118.47(14)	C <sub>(48)</sub>	C <sub>(49)</sub>	C <sub>(50)</sub>	116.9(9)
O <sub>(2)</sub>	C <sub>(24)</sub>	C <sub>(25)</sub>	110.57(12)	C <sub>(49)</sub>	C <sub>(48)</sub>	C <sub>(47)</sub>	123.6(10)
O <sub>(2)</sub>	C <sub>(24)</sub>	C <sub>(31)</sub>	104.71(10)	C <sub>(46)</sub>	C <sub>(47)</sub>	C <sub>(48)</sub>	117.9(10)
O <sub>(2)</sub>	C <sub>(24)</sub>	C <sub>(23)</sub>	111.66(12)	C <sub>(47)</sub>	C <sub>(46)</sub>	C <sub>(45)</sub>	123.3(11)
C <sub>(25)</sub>	C <sub>(24)</sub>	C <sub>(31)</sub>	117.39(12)	C <sub>(50)</sub>	C <sub>(45)</sub>	C <sub>(23)</sub>	116.1(12)
C <sub>(25)</sub>	C <sub>(24)</sub>	C <sub>(23)</sub>	100.13(10)	C <sub>(50)</sub>	C <sub>(45)</sub>	C <sub>(46)</sub>	113.7(13)
C <sub>(23)</sub>	C <sub>(24)</sub>	C <sub>(31)</sub>	112.58(11)	C <sub>(46)</sub>	C <sub>(45)</sub>	C <sub>(23)</sub>	130.1(9)

**Table S6. Torsion Angles for 3ag.**

A	B	C	D	Angle/ <sup>°</sup>	A	B	C	D	Angle/ <sup>°</sup>
O <sub>(1)</sub>	N <sub>(2)</sub>	C <sub>(10)</sub>	C <sub>(9)</sub>	-3.0(2)	C <sub>(25)</sub>	C <sub>(24)</sub>	C <sub>(23)</sub>	C <sub>(39)</sub>	179.9(5)
O <sub>(1)</sub>	C <sub>(2)</sub>	C <sub>(1)</sub>	N <sub>(1)</sub>	122.10(12)	C <sub>(25)</sub>	C <sub>(24)</sub>	C <sub>(23)</sub>	C <sub>(45)</sub>	-168.7(8)
O <sub>(1)</sub>	C <sub>(2)</sub>	C <sub>(1)</sub>	C <sub>(17)</sub>	-60.95(16)	C <sub>(25)</sub>	C <sub>(30)</sub>	C <sub>(29)</sub>	C <sub>(28)</sub>	-0.9(3)
O <sub>(1)</sub>	C <sub>(2)</sub>	C <sub>(3)</sub>	C <sub>(4)</sub>	-121.21(12)	C <sub>(26)</sub>	N <sub>(4)</sub>	C <sub>(23)</sub>	C <sub>(24)</sub>	-5.25(18)

O <sub>(1)</sub> C <sub>(2)</sub> C <sub>(3)</sub> C <sub>(8)</sub>	51.64(19)	C <sub>(26)</sub> N <sub>(4)</sub> C <sub>(23)</sub> C <sub>(39)</sub>	179.8(4)
F <sub>(4)</sub> C <sub>(35)</sub> C <sub>(36)</sub> C <sub>(37)</sub>	179.83(14)	C <sub>(26)</sub> N <sub>(4)</sub> C <sub>(23)</sub> C <sub>(45)</sub>	168.2(9)
F <sub>(3)</sub> C <sub>(37)</sub> C <sub>(36)</sub> C <sub>(35)</sub>	179.96(14)	C <sub>(26)</sub> C <sub>(25)</sub> C <sub>(24)</sub> O <sub>(2)</sub>	-121.47(13)
F <sub>(1)</sub> C <sub>(15)</sub> C <sub>(14)</sub> C <sub>(13)</sub>	179.93(15)	C <sub>(26)</sub> C <sub>(25)</sub> C <sub>(24)</sub> C <sub>(31)</sub>	118.53(13)
O <sub>(2)</sub> N <sub>(3)</sub> C <sub>(32)</sub> C <sub>(31)</sub>	-1.3(2)	C <sub>(26)</sub> C <sub>(25)</sub> C <sub>(24)</sub> C <sub>(23)</sub>	-3.60(15)
O <sub>(2)</sub> C <sub>(24)</sub> C <sub>(23)</sub> N <sub>(4)</sub>	122.70(14)	C <sub>(26)</sub> C <sub>(25)</sub> C <sub>(30)</sub> C <sub>(29)</sub>	1.0(3)
O <sub>(2)</sub> C <sub>(24)</sub> C <sub>(23)</sub> C <sub>(39)</sub>	-63.1(5)	C <sub>(26)</sub> C <sub>(27)</sub> C <sub>(28)</sub> C <sub>(29)</sub>	1.7(3)
O <sub>(2)</sub> C <sub>(24)</sub> C <sub>(23)</sub> C <sub>(45)</sub>	-51.7(8)	C <sub>(31)</sub> C <sub>(33)</sub> C <sub>(34)</sub> C <sub>(35)</sub>	-179.44(13)
F <sub>(2)</sub> C <sub>(13)</sub> C <sub>(14)</sub> C <sub>(15)</sub>	-179.82(16)	C <sub>(31)</sub> C <sub>(33)</sub> C <sub>(38)</sub> C <sub>(37)</sub>	178.57(13)
N <sub>(4)</sub> C <sub>(26)</sub> C <sub>(27)</sub> C <sub>(28)</sub>	173.49(16)	C <sub>(31)</sub> C <sub>(24)</sub> C <sub>(23)</sub> N <sub>(4)</sub>	-119.85(14)
N <sub>(4)</sub> C <sub>(23)</sub> C <sub>(39)</sub> C <sub>(40)</sub>	23.7(9)	C <sub>(31)</sub> C <sub>(24)</sub> C <sub>(23)</sub> C <sub>(39)</sub>	54.4(5)
N <sub>(4)</sub> C <sub>(23)</sub> C <sub>(39)</sub> C <sub>(44)</sub>	-152.7(7)	C <sub>(31)</sub> C <sub>(24)</sub> C <sub>(23)</sub> C <sub>(45)</sub>	65.8(8)
N <sub>(4)</sub> C <sub>(23)</sub> C <sub>(45)</sub> C <sub>(50)</sub>	37.4(16)	C <sub>(16)</sub> C <sub>(11)</sub> C <sub>(12)</sub> C <sub>(13)</sub>	0.5(2)
N <sub>(4)</sub> C <sub>(23)</sub> C <sub>(45)</sub> C <sub>(46)</sub>	-138.6(14)	C <sub>(16)</sub> C <sub>(15)</sub> C <sub>(14)</sub> C <sub>(13)</sub>	-0.1(3)
N <sub>(1)</sub> C <sub>(1)</sub> C <sub>(17)</sub> C <sub>(18)</sub>	-145.31(14)	C <sub>(24)</sub> O <sub>(2)</sub> N <sub>(3)</sub> C <sub>(32)</sub>	-7.51(19)
N <sub>(1)</sub> C <sub>(1)</sub> C <sub>(17)</sub> C <sub>(22)</sub>	31.74(19)	C <sub>(24)</sub> C <sub>(25)</sub> C <sub>(26)</sub> N <sub>(4)</sub>	1.02(17)
N <sub>(1)</sub> C <sub>(4)</sub> C <sub>(5)</sub> C <sub>(6)</sub>	172.78(16)	C <sub>(24)</sub> C <sub>(25)</sub> C <sub>(26)</sub> C <sub>(27)</sub>	176.77(15)
N <sub>(2)</sub> O <sub>(1)</sub> C <sub>(2)</sub> C <sub>(9)</sub>	19.56(14)	C <sub>(24)</sub> C <sub>(25)</sub> C <sub>(30)</sub> C <sub>(29)</sub>	-174.50(17)
N <sub>(2)</sub> O <sub>(1)</sub> C <sub>(2)</sub> C <sub>(1)</sub>	140.44(12)	C <sub>(24)</sub> C <sub>(31)</sub> C <sub>(32)</sub> N <sub>(3)</sub>	8.55(18)
N <sub>(2)</sub> O <sub>(1)</sub> C <sub>(2)</sub> C <sub>(3)</sub>	-108.33(12)	C <sub>(24)</sub> C <sub>(23)</sub> C <sub>(39)</sub> C <sub>(40)</sub>	-150.4(5)
N <sub>(3)</sub> O <sub>(2)</sub> C <sub>(24)</sub> C <sub>(25)</sub>	-114.91(13)	C <sub>(24)</sub> C <sub>(23)</sub> C <sub>(39)</sub> C <sub>(44)</sub>	33.3(10)
N <sub>(3)</sub> O <sub>(2)</sub> C <sub>(24)</sub> C <sub>(31)</sub>	12.43(16)	C <sub>(24)</sub> C <sub>(23)</sub> C <sub>(45)</sub> C <sub>(50)</sub>	-149.3(9)
N <sub>(3)</sub> O <sub>(2)</sub> C <sub>(24)</sub> C <sub>(23)</sub>	134.53(13)	C <sub>(24)</sub> C <sub>(23)</sub> C <sub>(45)</sub> C <sub>(46)</sub>	34.7(18)
C <sub>(9)</sub> C <sub>(2)</sub> C <sub>(1)</sub> N <sub>(1)</sub>	-121.66(13)	C <sub>(23)</sub> N <sub>(4)</sub> C <sub>(26)</sub> C <sub>(25)</sub>	2.67(19)
C <sub>(9)</sub> C <sub>(2)</sub> C <sub>(1)</sub> C <sub>(17)</sub>	55.29(16)	C <sub>(23)</sub> N <sub>(4)</sub> C <sub>(26)</sub> C <sub>(27)</sub>	-172.82(16)
C <sub>(9)</sub> C <sub>(2)</sub> C <sub>(3)</sub> C <sub>(4)</sub>	119.55(13)	C <sub>(23)</sub> C <sub>(39)</sub> C <sub>(40)</sub> C <sub>(41)</sub>	-175.5(6)
C <sub>(9)</sub> C <sub>(2)</sub> C <sub>(3)</sub> C <sub>(8)</sub>	-67.6(2)	C <sub>(23)</sub> C <sub>(39)</sub> C <sub>(44)</sub> C <sub>(43)</sub>	174.8(7)
C <sub>(9)</sub> C <sub>(11)</sub> C <sub>(16)</sub> C <sub>(15)</sub>	177.07(13)	C <sub>(34)</sub> C <sub>(33)</sub> C <sub>(31)</sub> C <sub>(24)</sub>	114.23(14)
C <sub>(9)</sub> C <sub>(11)</sub> C <sub>(12)</sub> C <sub>(13)</sub>	-177.52(13)	C <sub>(34)</sub> C <sub>(33)</sub> C <sub>(31)</sub> C <sub>(32)</sub>	-2.0(2)
C <sub>(2)</sub> O <sub>(1)</sub> N <sub>(2)</sub> C <sub>(10)</sub>	-11.17(17)	C <sub>(34)</sub> C <sub>(33)</sub> C <sub>(38)</sub> C <sub>(37)</sub>	-1.0(2)
C <sub>(2)</sub> C <sub>(9)</sub> C <sub>(11)</sub> C <sub>(16)</sub>	98.18(15)	C <sub>(34)</sub> C <sub>(35)</sub> C <sub>(36)</sub> C <sub>(37)</sub>	0.0(2)
C <sub>(2)</sub> C <sub>(9)</sub> C <sub>(11)</sub> C <sub>(12)</sub>	-83.86(16)	C <sub>(4)</sub> N <sub>(1)</sub> C <sub>(1)</sub> C <sub>(2)</sub>	-5.39(15)
C <sub>(2)</sub> C <sub>(9)</sub> C <sub>(10)</sub> N <sub>(2)</sub>	14.51(18)	C <sub>(4)</sub> N <sub>(1)</sub> C <sub>(1)</sub> C <sub>(17)</sub>	177.55(12)
C <sub>(2)</sub> C <sub>(1)</sub> C <sub>(17)</sub> C <sub>(18)</sub>	37.99(18)	C <sub>(4)</sub> C <sub>(3)</sub> C <sub>(8)</sub> C <sub>(7)</sub>	0.8(2)

C <sub>(2)</sub> C <sub>(1)</sub> C <sub>(17)</sub> C <sub>(22)</sub>	-144.96(14)	C <sub>(4)</sub> C <sub>(5)</sub> C <sub>(6)</sub> C <sub>(7)</sub>	0.8(3)
C <sub>(2)</sub> C <sub>(3)</sub> C <sub>(4)</sub> N <sub>(1)</sub>	-0.65(16)	C <sub>(38)</sub> C <sub>(33)</sub> C <sub>(31)</sub> C <sub>(24)</sub>	-65.32(16)
C <sub>(2)</sub> C <sub>(3)</sub> C <sub>(4)</sub> C <sub>(5)</sub>	173.86(14)	C <sub>(38)</sub> C <sub>(33)</sub> C <sub>(31)</sub> C <sub>(32)</sub>	178.49(13)
C <sub>(2)</sub> C <sub>(3)</sub> C <sub>(8)</sub> C <sub>(7)</sub>	-171.16(15)	C <sub>(38)</sub> C <sub>(33)</sub> C <sub>(34)</sub> C <sub>(35)</sub>	0.1(2)
C <sub>(1)</sub> N <sub>(1)</sub> C <sub>(4)</sub> C <sub>(3)</sub>	3.83(16)	C <sub>(38)</sub> C <sub>(37)</sub> C <sub>(36)</sub> C <sub>(35)</sub>	-0.9(2)
C <sub>(1)</sub> N <sub>(1)</sub> C <sub>(4)</sub> C <sub>(5)</sub>	-170.34(15)	C <sub>(12)</sub> C <sub>(11)</sub> C <sub>(16)</sub> C <sub>(15)</sub>	-0.9(2)
C <sub>(1)</sub> C <sub>(2)</sub> C <sub>(3)</sub> C <sub>(4)</sub>	-2.19(13)	C <sub>(12)</sub> C <sub>(13)</sub> C <sub>(14)</sub> C <sub>(15)</sub>	-0.4(3)
C <sub>(1)</sub> C <sub>(2)</sub> C <sub>(3)</sub> C <sub>(8)</sub>	170.67(15)	C <sub>(10)</sub> C <sub>(9)</sub> C <sub>(2)</sub> O <sub>(1)</sub>	-19.16(13)
C <sub>(1)</sub> C <sub>(17)</sub> C <sub>(18)</sub> C <sub>(19)</sub>	175.82(13)	C <sub>(10)</sub> C <sub>(9)</sub> C <sub>(2)</sub> C <sub>(1)</sub>	-140.74(12)
C <sub>(1)</sub> C <sub>(17)</sub> C <sub>(22)</sub> C <sub>(21)</sub>	-176.54(16)	C <sub>(10)</sub> C <sub>(9)</sub> C <sub>(2)</sub> C <sub>(3)</sub>	103.41(13)
C <sub>(33)</sub> C <sub>(31)</sub> C <sub>(24)</sub> O <sub>(2)</sub>	-138.18(12)	C <sub>(10)</sub> C <sub>(9)</sub> C <sub>(11)</sub> C <sub>(16)</sub>	-19.95(19)
C <sub>(33)</sub> C <sub>(31)</sub> C <sub>(24)</sub> C <sub>(25)</sub>	-15.15(17)	C <sub>(10)</sub> C <sub>(9)</sub> C <sub>(11)</sub> C <sub>(12)</sub>	158.00(13)
C <sub>(33)</sub> C <sub>(31)</sub> C <sub>(24)</sub> C <sub>(23)</sub>	100.32(14)	C <sub>(32)</sub> C <sub>(31)</sub> C <sub>(24)</sub> O <sub>(2)</sub>	-11.85(14)
C <sub>(33)</sub> C <sub>(31)</sub> C <sub>(32)</sub> N <sub>(3)</sub>	133.36(16)	C <sub>(32)</sub> C <sub>(31)</sub> C <sub>(24)</sub> C <sub>(25)</sub>	111.18(13)
C <sub>(33)</sub> C <sub>(34)</sub> C <sub>(35)</sub> F <sub>(4)</sub>	-179.44(13)	C <sub>(32)</sub> C <sub>(31)</sub> C <sub>(24)</sub> C <sub>(23)</sub>	-133.35(12)
C <sub>(33)</sub> C <sub>(34)</sub> C <sub>(35)</sub> C <sub>(36)</sub>	0.4(2)	C <sub>(27)</sub> C <sub>(28)</sub> C <sub>(29)</sub> C <sub>(30)</sub>	-0.5(3)
C <sub>(33)</sub> C <sub>(38)</sub> C <sub>(37)</sub> F <sub>(3)</sub>	-179.44(13)	C <sub>(39)</sub> C <sub>(40)</sub> C <sub>(41)</sub> C <sub>(42)</sub>	0.5(10)
C <sub>(33)</sub> C <sub>(38)</sub> C <sub>(37)</sub> C <sub>(36)</sub>	1.5(2)	C <sub>(40)</sub> C <sub>(39)</sub> C <sub>(44)</sub> C <sub>(43)</sub>	-1.5(14)
C <sub>(11)</sub> C <sub>(9)</sub> C <sub>(2)</sub> O <sub>(1)</sub>	-147.22(11)	C <sub>(40)</sub> C <sub>(41)</sub> C <sub>(42)</sub> C <sub>(43)</sub>	-0.8(11)
C <sub>(11)</sub> C <sub>(9)</sub> C <sub>(2)</sub> C <sub>(1)</sub>	91.20(14)	C <sub>(41)</sub> C <sub>(42)</sub> C <sub>(43)</sub> C <sub>(44)</sub>	0.0(12)
C <sub>(11)</sub> C <sub>(9)</sub> C <sub>(2)</sub> C <sub>(3)</sub>	-24.65(17)	C <sub>(42)</sub> C <sub>(43)</sub> C <sub>(44)</sub> C <sub>(39)</sub>	1.1(13)
C <sub>(11)</sub> C <sub>(9)</sub> C <sub>(10)</sub> N <sub>(2)</sub>	142.02(15)	C <sub>(44)</sub> C <sub>(39)</sub> C <sub>(40)</sub> C <sub>(41)</sub>	0.7(12)
C <sub>(11)</sub> C <sub>(16)</sub> C <sub>(15)</sub> F <sub>(1)</sub>	-179.31(14)	C <sub>(30)</sub> C <sub>(25)</sub> C <sub>(26)</sub> N <sub>(4)</sub>	-175.48(14)
C <sub>(11)</sub> C <sub>(16)</sub> C <sub>(15)</sub> C <sub>(14)</sub>	0.7(3)	C <sub>(30)</sub> C <sub>(25)</sub> C <sub>(26)</sub> C <sub>(27)</sub>	0.3(2)
C <sub>(11)</sub> C <sub>(12)</sub> C <sub>(13)</sub> F <sub>(2)</sub>	179.61(14)	C <sub>(30)</sub> C <sub>(25)</sub> C <sub>(24)</sub> O <sub>(2)</sub>	54.5(2)
C <sub>(11)</sub> C <sub>(12)</sub> C <sub>(13)</sub> C <sub>(14)</sub>	0.2(3)	C <sub>(30)</sub> C <sub>(25)</sub> C <sub>(24)</sub> C <sub>(31)</sub>	-65.5(2)
C <sub>(17)</sub> C <sub>(18)</sub> C <sub>(19)</sub> C <sub>(20)</sub>	1.4(2)	C <sub>(30)</sub> C <sub>(25)</sub> C <sub>(24)</sub> C <sub>(23)</sub>	172.38(17)
C <sub>(18)</sub> C <sub>(17)</sub> C <sub>(22)</sub> C <sub>(21)</sub>	0.6(2)	C <sub>(8)</sub> C <sub>(3)</sub> C <sub>(4)</sub> N <sub>(1)</sub>	-174.40(13)
C <sub>(18)</sub> C <sub>(19)</sub> C <sub>(20)</sub> C <sub>(21)</sub>	-0.8(3)	C <sub>(8)</sub> C <sub>(3)</sub> C <sub>(4)</sub> C <sub>(5)</sub>	0.1(2)
C <sub>(19)</sub> C <sub>(20)</sub> C <sub>(21)</sub> C <sub>(22)</sub>	0.2(3)	C <sub>(8)</sub> C <sub>(7)</sub> C <sub>(6)</sub> C <sub>(5)</sub>	0.2(3)
C <sub>(20)</sub> C <sub>(21)</sub> C <sub>(22)</sub> C <sub>(17)</sub>	0.0(3)	C <sub>(50)</sub> C <sub>(49)</sub> C <sub>(48)</sub> C <sub>(47)</sub>	0.6(16)
C <sub>(22)</sub> C <sub>(17)</sub> C <sub>(18)</sub> C <sub>(19)</sub>	-1.2(2)	C <sub>(49)</sub> C <sub>(50)</sub> C <sub>(45)</sub> C <sub>(23)</sub>	-176.0(9)
C <sub>(3)</sub> C <sub>(2)</sub> C <sub>(1)</sub> N <sub>(1)</sub>	4.85(14)	C <sub>(49)</sub> C <sub>(50)</sub> C <sub>(45)</sub> C <sub>(46)</sub>	0.7(19)
C <sub>(3)</sub> C <sub>(2)</sub> C <sub>(1)</sub> C <sub>(17)</sub>	-178.20(12)	C <sub>(49)</sub> C <sub>(48)</sub> C <sub>(47)</sub> C <sub>(46)</sub>	-0.8(18)

C <sub>(3)</sub> C <sub>(4)</sub> C <sub>(5)</sub> C <sub>(6)</sub>	-0.9(2)	C <sub>(48)</sub> C <sub>(47)</sub> C <sub>(46)</sub> C <sub>(45)</sub>	1(2)
C <sub>(3)</sub> C <sub>(8)</sub> C <sub>(7)</sub> C <sub>(6)</sub>	-1.0(3)	C <sub>(47)</sub> C <sub>(46)</sub> C <sub>(45)</sub> C <sub>(23)</sub>	175.1(12)
C <sub>(25)</sub> C <sub>(26)</sub> C <sub>(27)</sub> C <sub>(28)</sub>	-1.6(3)	C <sub>(47)</sub> C <sub>(46)</sub> C <sub>(45)</sub> C <sub>(50)</sub>	-1(2)
C <sub>(25)</sub> C <sub>(24)</sub> C <sub>(23)</sub> N <sub>(4)</sub>	5.63(17)	C <sub>(45)</sub> C <sub>(50)</sub> C <sub>(49)</sub> C <sub>(48)</sub>	-0.5(17)

**Table S7. Hydrogen Atom Coordinates ( $\text{\AA} \times 10^4$ ) and Isotropic Displacement Parameters ( $\text{\AA}^2 \times 10^3$ ) for 3ag.**

Atom	x	y	z	U(eq)
H <sub>(9)</sub>	7426.03	4911.75	4394.59	50
H <sub>(18)</sub>	6221.75	3980.7	3092.39	61
H <sub>(19)</sub>	5348.36	3711.05	4020.74	73
H <sub>(20)</sub>	5531.45	2888.57	5755.06	82
H <sub>(21)</sub>	6577.07	2301.28	6531.94	91
H <sub>(22)</sub>	7448.3	2537	5604.52	74
H <sub>(31)</sub>	7273.06	5270.99	9693.53	55
H <sub>(16)</sub>	8625.75	6148	3382.43	61
H <sub>(34)</sub>	6278.65	3663.27	8187.75	61
H <sub>(38)</sub>	6156.37	6160.48	9174.11	63
H <sub>(12)</sub>	8407.18	4374.59	5856.25	61
H <sub>(10)</sub>	7369.11	6265.25	3071.46	66
H <sub>(32)</sub>	7453.76	3746.63	8991.4	71
H <sub>(36)</sub>	4478.52	4702.69	8281.89	73
H <sub>(14)</sub>	10107.55	5791.91	6350.2	75
H <sub>(27)</sub>	6189.84	7897.02	6078.38	74
H <sub>(40)</sub>	7490.92	7898.46	10098.53	56
H <sub>(41)</sub>	8304.22	8305.5	11777.31	72
H <sub>(42)</sub>	9295.92	7509.69	12350.54	83
H <sub>(43)</sub>	9446.82	6285.71	11241.04	86
H <sub>(44)</sub>	8618.35	5867.69	9588.88	74
H <sub>(30)</sub>	6508.77	4896.2	5859.36	72
H <sub>(8)</sub>	8438.99	4865.18	1430.61	73
H <sub>(28)</sub>	5654.36	7110.47	4380.59	84
H <sub>(29)</sub>	5804.55	5636.2	4280.63	86

H <sub>(5)</sub>	8938.6	2022.04	2812.97	88
H <sub>(7)</sub>	9238.49	4037.01	758.07	92
H <sub>(6)</sub>	9482.28	2644.2	1445.31	98
H <sub>(50)</sub>	7661.94	7635.48	10308.26	85
H <sub>(49)</sub>	8560.61	7933.52	11914.39	105
H <sub>(48)</sub>	9513.2	7128.78	12116.6	118
H <sub>(47)</sub>	9622.47	6054.4	10762.35	117
H <sub>(46)</sub>	8723.54	5756.92	9176.63	94

**Table S8. Atomic Occupancy for ANNA\_ROS534\_11.**

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
C <sub>(39)</sub>	0.58(2)	C <sub>(40)</sub>	0.58(2)	H <sub>(40)</sub>	0.58(2)
C <sub>(41)</sub>	0.58(2)	H <sub>(41)</sub>	0.58(2)	C <sub>(42)</sub>	0.58(2)
H <sub>(42)</sub>	0.58(2)	C <sub>(43)</sub>	0.58(2)	H <sub>(43)</sub>	0.58(2)
C <sub>(44)</sub>	0.58(2)	H <sub>(44)</sub>	0.58(2)	C <sub>(50)</sub>	0.42(2)
H <sub>(50)</sub>	0.42(2)	C <sub>(49)</sub>	0.42(2)	H <sub>(49)</sub>	0.42(2)
C <sub>(48)</sub>	0.42(2)	H <sub>(48)</sub>	0.42(2)	C <sub>(47)</sub>	0.42(2)
H <sub>(47)</sub>	0.42(2)	C <sub>(46)</sub>	0.42(2)	H <sub>(46)</sub>	0.42(2)
C <sub>(45)</sub>	0.42(2)				

## Experimental

Single crystals of C<sub>22</sub>H<sub>14</sub>F<sub>2</sub>N<sub>2</sub>O (3ag) were prepared by slow evaporation of saturated solution in EtOAc/hexane. A suitable crystal was selected and mounted with acrylic glue on a glass stick in a SuperNova, Dual, Cu at home/near, AtlasS2 diffractometer. The crystal was kept at 104(6) K during data collection. Using Olex2,<sup>S1</sup> the structure was solved with the ShelXT<sup>S2</sup> structure solution program using Intrinsic Phasing and refined with the ShelXL<sup>S3</sup> refinement package using Least Squares minimization.

## Crystal structure determination of [ANNA\_ROS534\_11]

**Crystal Data** for C<sub>22</sub>H<sub>14</sub>F<sub>2</sub>N<sub>2</sub>O ( $M=360.35$  g/mol): monoclinic, space group P2<sub>1</sub>/c (no. 14),  $a = 20.4243(4)$  Å,  $b = 15.3838(2)$  Å,  $c = 11.4158(2)$  Å,  $\beta = 103.3576(18)^\circ$ ,  $V = 3489.86(10)$  Å<sup>3</sup>,  $Z = 8$ ,  $T = 104(6)$  K,  $\mu(\text{CuK}\alpha) = 0.833$  mm<sup>-1</sup>,  $D_{\text{calc}} = 1.372$  g/cm<sup>3</sup>, 37744 reflections measured ( $4.446^\circ \leq 2\Theta \leq 153.222^\circ$ ), 7322 unique ( $R_{\text{int}} = 0.0317$ ,  $R_{\text{sigma}} = 0.0196$ ) which were used in all calculations. The final  $R_1$  was 0.0439 ( $I > 2\sigma(I)$ ) and  $wR_2$  was 0.1230 (all data).

## Refinement model description

Number of restraints - 0, number of constraints - unknown.

Details:

1. Fixed Uiso

At 1.2 times of:

All C(H) groups

2. Others

Sof(C50)=Sof(H50)=Sof(C49)=Sof(H49)=Sof(C48)=Sof(H48)=Sof(C47)=Sof(H47)=  
Sof(C46)=Sof(H46)=Sof(C45)=1-FVAR(1)  
Sof(C39)=Sof(C40)=Sof(H40)=Sof(C41)=Sof(H41)=Sof(C42)=Sof(H42)=Sof(C43)=  
Sof(H43)=Sof(C44)=Sof(H44)=FVAR(1)

3.a Ternary CH refined with riding coordinates:

C9(H9), C31(H31)

3.b Aromatic/amide H refined with riding coordinates:

C18(H18), C19(H19), C20(H20), C21(H21), C22(H22), C16(H16), C34(H34),  
C38(H38), C12(H12), C10(H10), C32(H32), C36(H36), C14(H14), C27(H27), C40(H40),  
C41(H41), C42(H42), C43(H43), C44(H44), C30(H30), C8(H8), C28(H28), C29(H29),  
C5(H5), C7(H7), C6(H6), C50(H50), C49(H49), C48(H48), C47(H47), C46(H46)

This report has been created with Olex2, compiled on 2018.05.29 svn.r3508 for OlexSys. Please [let us know](#) if there are any errors or if you would like to have additional features.

## References Cited

(S1) Dolomanov, O.V., Bourhis, L.J., Gildea, R.J., Howard, J.A.K. & Puschmann, H. OLEX2: a Complete Structure Solution, Refinement and Analysis Program. *J. Appl. Cryst.* **2009**, *42*, 339-341.

(S2) Sheldrick, G.M. SHELXT - Integrated Space-Group and Crystal-Structure Determination. *Acta Cryst.* **2015**, *A71*, 3-8.

(S3) Sheldrick, G.M. Crystal Structure Refinement with SHELXL. *Acta Cryst.* **2015**, *C71*, 3-8.