

Supporting Information

In-situ Electro-Organic Synthesis of Hydroquinone Using Anisole on MWCNT/Nafion Modified Electrode Surface and Its Heterogenous Electrocatalytic Reduction of Toxic Cr(VI) Species

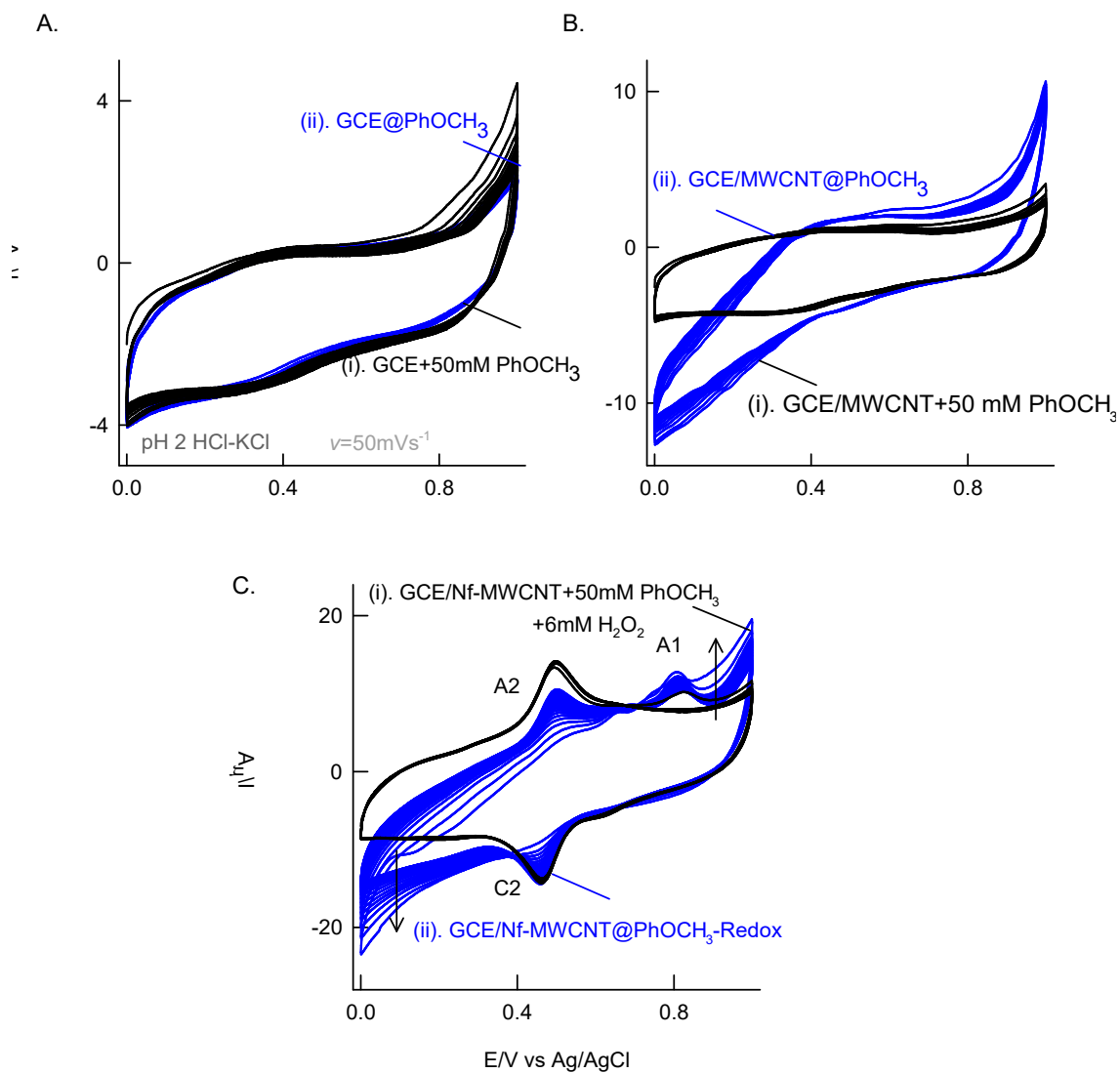
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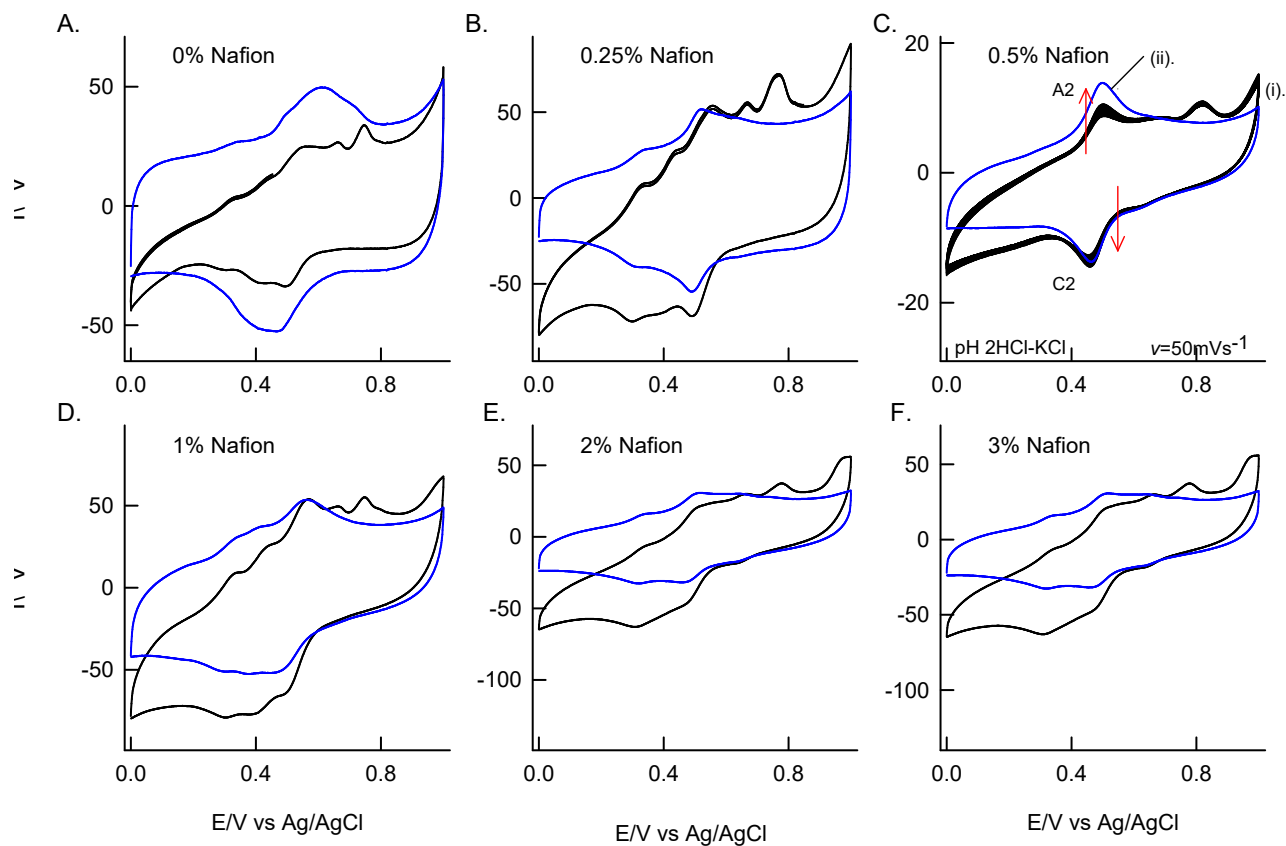
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Vellore Institute of Technology University, Vellore-632014, Tamil Nadu, India*

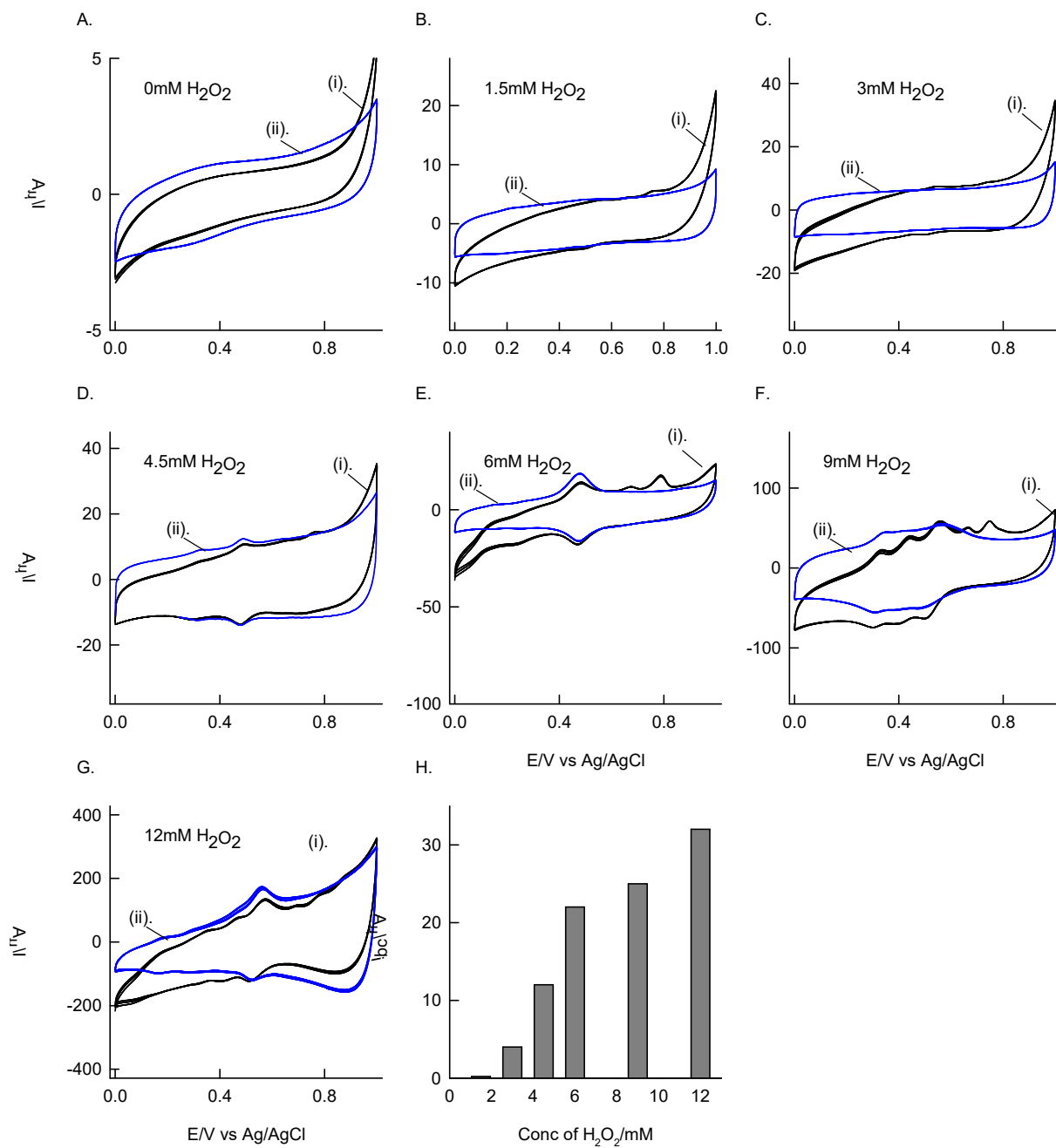
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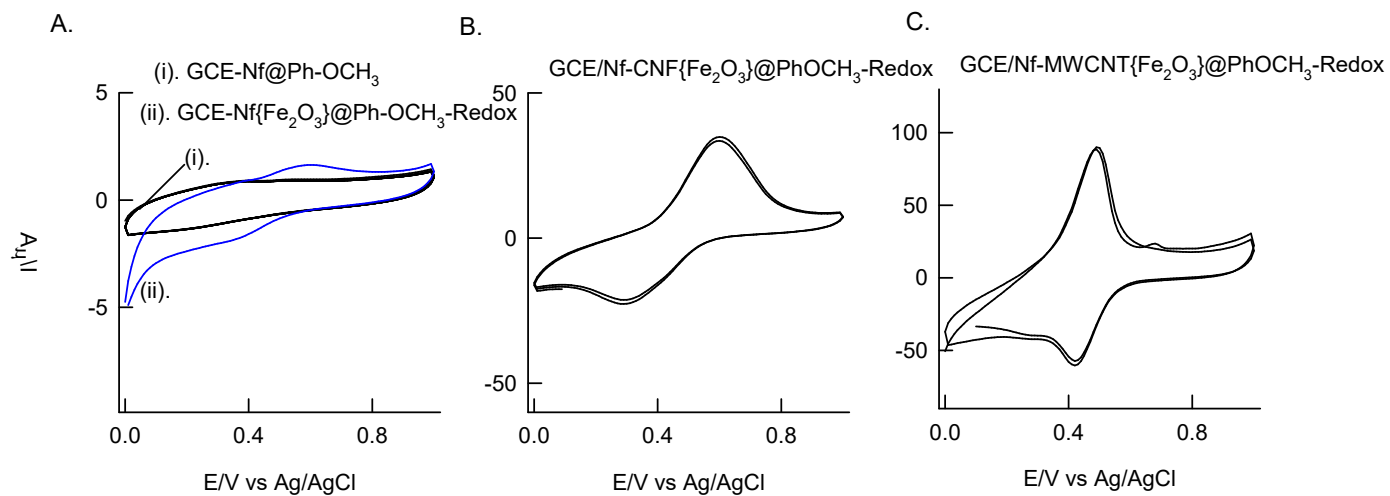
Supplementary Fig. S1. Twenty continuous CV segments of (A) GCE and (B) GCE/MWCNT in 50mM $PhOCH_3$ dissolved in 10 mL pH 2 KCl-HCl buffer (curve (i)) and its medium transferred CV responses (curve (ii)). CV responses of (C) GCE/Nf-MWCNT in a mixture of 50mM $PhOCH_3$ + 6mM H_2O_2 dissolved in 10mL of pH 2 KCl-HCl buffer (curve (i)) and its medium (blank pH 2) transferred response (curve (ii)) for at $\nu=50mVs^{-1}$. Note: Nf=Nafion.



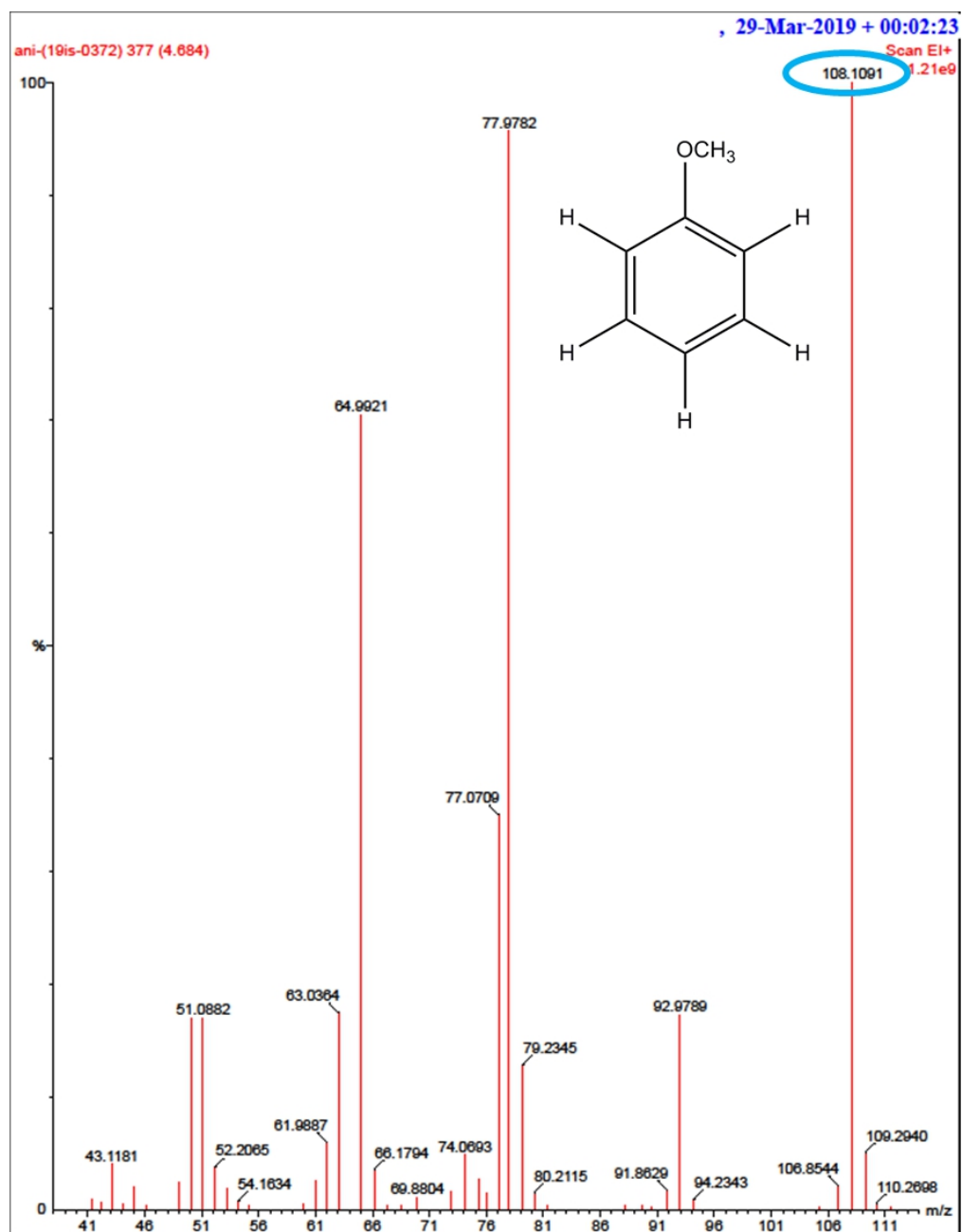
Supplementary Fig. S2. (A-E) CV responses of GCE/MWCNT in 50mM Ph-OCH₃-Redox+9mM H₂O₂ with various concentrations of Nafion (0% to 3%) (i) and its medium transferred responses ((ii); GCE/MWCNT@Ph-OCH₃-Redox) in pH 2 KCl-HCl buffer at $v= 50 \text{ mV s}^{-1}$.



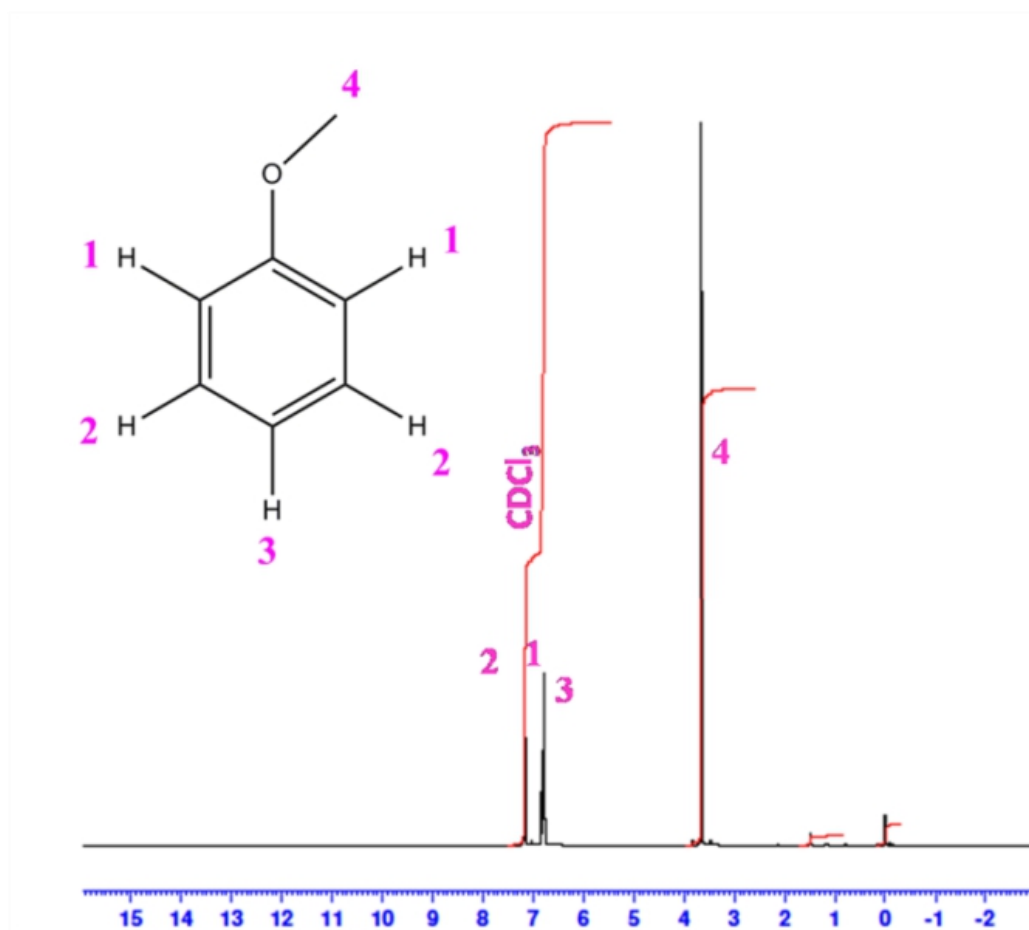
Supplementary Fig. S3. (A-G) CV responses of GCE/Nf-MWCNT exposed 50mM $\text{PhOCH}_3\text{-Redox} + x \text{ mM } \text{H}_2\text{O}_2$ (curve (i)) and its medium transferred responses (curve (ii); GCE/Nf-MWCNT@ $\text{PhOCH}_3\text{-Redox}$) in pH 2 KCl-HCl buffer at $v = 50 \text{ mV s}^{-1}$ and (H) its comparative plot of reduction peak current vs different concentration of H_2O_2 .



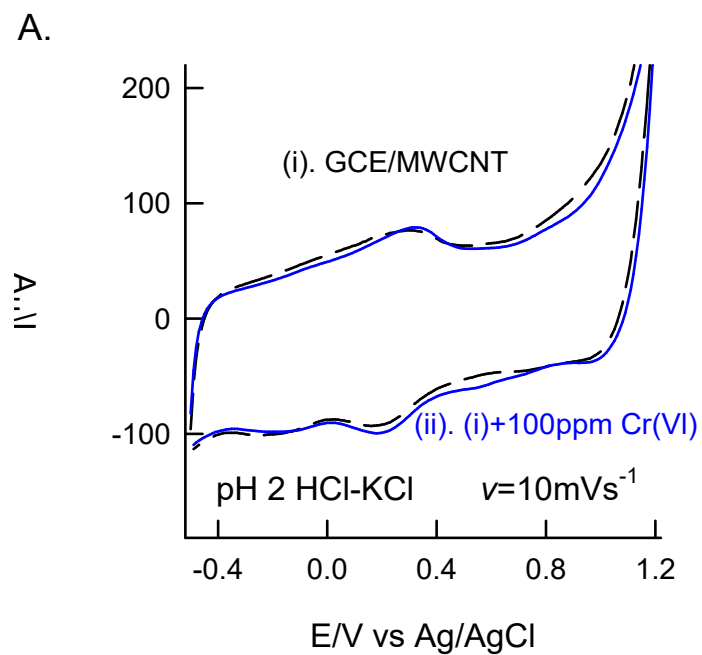
Supplementary Fig. S4. CV responses of (A) GCE-Nf@PhOCH₃-Redox (curve (i)) and GCE-Nf/Fe₂O₃{Fe₂O₃}@PhOCH₃-Redox (curve (ii)), (B) GCE/Nf-CNT{Fe₂O₃}@PhOCH₃-Redox and (C) GCE/N-MWCNT{Fe₂O₃}@PhOCH₃-Redox in KCl-HCl buffer at $\nu = 50 \text{ mV s}^{-1}$. {Fe₂O₃} = Deliberately added iron oxide to the carbon matrix.



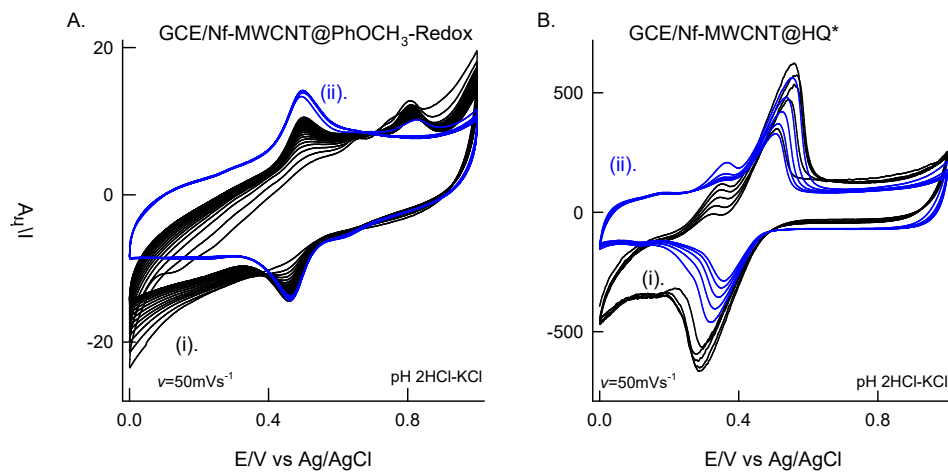
Supplementary Fig. S5A. Control GC-MS response of 108.14, PhOCH₃ (calculated molecular weight 108.14).



Supplementary Fig. S5B. Control NMR spectra of the PhOCH_3 (B, control) using CDCl_3 as a solvent.



Supplementary Fig. S6. Control Experiments for CV response with various systems of GCE/MWCNT (A) without (curve (i)) and with 100ppm Cr(VI) (curve (ii)) dissolved in pH 2 KCl-HCl buffer at $\nu=10\text{ mV s}^{-1}$.



Supplementary Fig. S7. Comparative control experiments for ten continuous CV responses of GCE/Nf-MWCNT with 50 mM PhOCH₃-Redox (A), 10mM Hydroquinone (B) in pH buffer (curve (i)) and its medium transferred responses (curve (ii)) in pH 2 HCl-KCl at $v= 50 \text{ mV s}^{-1}$.