An efficient and practical aerobic oxidation of benzylic

methylenes by recyclable N-hydroxyimide

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Table S1. Screen of iron salts^a



Entry	Catalyst	Conversion (%) ^b
1	Iron tristearate	78
2	Fe(acac) ₃	73
3	$Fe(NH_4)_2(SO_4)_2$	trace
4	FeOH(CH ₃ COO) ₂	80
5	FeCl ₃	trace
6	Fe(NO ₃) ₃ ·9H ₂ O	83
7	$Fe_2(SO4)_3^c$	69

^aReaction conditions: 2 mmol ethylbenzene and 10 mol % metal salts, 10 mol % NHPI in 2 mL AcOH are stirred at 100°C oil bath for 10 h.

^bConversion is determined by GC.

^c5 mol% Fe₂(SO₄)₃ was added.

Table S2. Screen of solvents^a

	Fe(I	NO ₃) ₃ ●9H ₂ O, NHPI O ₂ , 80 ^o C	
Entry	Solvent		Conversion (%)b
1	АсОН		23
2	Chlorobenzene		trace
3	Cyclohexanone		9
4	EtOAc ^c		11
5	EtOH ^c		<5
6	CH ₃ CN		41
7	PhCN		57
8	DMF		trace
9	DMSO		<5
10	Toluene		19
11	n-Butanol		<5
12	t-Butanol		<5

^a2 mmol ethylbenzene, 5 mol % metal salts, 5 mol % NHPI in 2 mL solvents are stirred at 80 °C oil bath for 10 h.

^bConversion is determined by GC.

^cRecation mixture was stirred at refluxing condition.



Figure S1. Structure of screened NHIs

Table S3. Screen of NHIs^a

	Fe(NC	$D_{3}_{3} \bullet 9H_{2}O, \text{ NHIs}$		_
Prici, 0 ₂ , 80 °C				
Entry	NHIs	Conversion (%) ^b		
1	NHPI	99		
2	NHNI	44		
3	NHND	11		
4	NHSI	97		
5	HOAT	<5		
6	THICA	38		

^a Reaction conditions: 2 mmol ethylbenzene and 5 mol % metal salts, 5 mol % NHIs in 2 mL PhCN are stirred in 90 °C oil bath for 10 h.

^b Conversion is determined by GC with diphenyl ether as internal standard.



Figure S2. Linear regression line and regression equation of NHPI/internal standard when molar raion of NHPI : PhOPh = 0.02 - 0.50)



Figure S3. Linear regression line and regression equation of NHSI/internal standard when molar ratio of NHSI : PhOPh = 0.00 - 0.80)

Table S4. Further optimization of Fe(NO₃)₃/NHSI/O₂ system^a



Entry	$Fe(NO_3)_3 \cdot 9H_2O$	NHSI	PhCN	Т	Oxygen	Time	Yield ^c
	(mol%)	(mol%)	(mL)	(°C)	Source ^b	(h)	(%)
1	5	5	2	90	A	10	93
2	3	3	2	90	А	18	93
3	1	1	2	90	А	24	57
4	1	1	2	100	А	48	68
5	1	2	2	100	А	48	77
6	2	2	2	100	А	24	83
7	1	3	2	100	А	12	86
8	1	3	2	90	А	18	86
9	1	3	2	80	А	48	63
10	1	3	1	90	А	16	86
11	1	3	0.5	90	А	48	61
12	1	3	1	90	В	14	90
13	1	3	1	90	C	16	90
14 ^d	1	3	1	90	В	16	89

^aReaction conditions: 2 mmol ethylbenzene, $Fe(NO_3)_3 \cdot 9H_2O$ and NHSI in PhCN are stirred in 90 °C oil bath.

^bOxygen source condition: A: Air (open system); B: O_2 Balloon without purging; C: O_2 balloon with purging (replacing original air in system with pure O_2).

^cDetermined by GC with diphenyl ether as internal standard.

^d20 mmol% KPF₆ was added as additive.

Table S5. Controlled experiments of EB oxidation^a



Entry	Catalyst	Additive	Conversion ^b
1	5 mol% Fe(NO ₃) ₃	\	98
	5 mol% NHSI		
2°	5 mol% Fe(NO ₃) ₃	\	98
	5 mol% NHSI		
3	2 mol% Fe(NO ₃) ₃	\	79

	5 mol% NHSI		
4	5 mol% NHSI	\	54
5	5 mol% Fe(NO ₃) ₃	\	trace
6	5 mol% Fe(NO ₃) ₃	\	21
	1 mol% NHSI		
7	2.5 mol% Fe ₂ (SO ₄) ₃	\	71
	5 mol% NHSI		
8	15 mol% NaNO ₃	\	57
	5 mol% NHSI		
9	5 mol% Fe(NO ₃) ₃	5 mol% BHT	79
	5 mol% NHSI		
10	5 mol% Fe(NO ₃) ₃	20 mol% BHT	53
	5 mol% NHSI		
11	5 mol% Fe(NO ₃) ₃	100 mol% BHT	trace
	5 mol% NHSI		

^aReaction conditions: 2 mmol ethylbenzene, catalyst, additive in 1 mL PhCN with O_2 balloon are stirred in 90 °C oil bath for 24 h.

^bConversion was determined by GC.

^cReaction flask was tightly wrapped by aluminum foil to avoid any visible light.

Product characterization

Benzophenone^[1] (2a). Compound 2a was obtain as white solid (357 mg, 98% yield). m. p. 48-49 °C (lit.^[1] 47-49 °C) ¹H NMR (500 MHz, Chloroform-*d*) δ 7.84 – 7.82 (m, 4H), 7.60 (tt, 2H, *J* = 7.5, 1.5 Hz), 7.52 – 7.48 (m, 4H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 196.8, 137.6, 132.4, 130.1, 128.3.

4-Chlorobenzophenone^[2] **(2b).** Compound **2b** was obtain as white solid (359 mg, 83% yield). m. p. 75-76 °C (lit.^[2] 75-77 °C) ¹H NMR (500 MHz, Chloroform-*d*) δ 7.80-7.76 (m, 4H), 7.62 (tt, J =7.5 Hz, 1.5 Hz, 1H), 7.53-7.47 (m, 4H).¹³C NMR (125 MHz, Chloroform-*d*) δ 195.5, 138.9, 137.3, 135.9, 132.7, 131.5, 129.9, 128.7, 128.4.

9-Fluorenone^[1] (2c). Compound 2c was obtain as yellow solid (349 mg, 97% yield). m. p. 83-84 °C (lit.^[5] 84-86 °C) ¹H NMR (500 MHz, Chloroform-*d*) δ 7.66 (dt, *J* = 7.5 Hz, 1 Hz, 2H), 7.51 (dt, *J* = 7.5 Hz, 1 Hz, 2H), 7.48 (dt, *J* = 7.5, 1.5 Hz, 2H), 7.29 (td, *J* = 7.5 Hz, 1.5 Hz, 2H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 193.9, 144.4, 134.7, 134.1, 129.0, 124.3, 120.3.

9-Xanthenone^{[3][4]} (2d). Compound 2d was obtain as white solid (384 mg, 98% yield). m. p. 170-172 °C (lit.^[3] 168-170 °C) ¹H NMR (500 MHz, Chloroform-*d*) δ 8.33 (dd, *J* = 8.0 Hz, 1.5 Hz, 2H), 7.72 – 7.69 (m, 2H), 7.47 (d, *J* = 8.5 Hz, 2H), 7.38 – 7.35 (m, 2H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 177.1, 156.1, 134.7, 126.7, 123.9, 121.8, 117.9.

1-[4-(4-Pyridinyl)phenyl]ethenone^[5] (**2e).** Compound **2e** was obtain as off-white solid (362 mg, 99% yield). m. p. 71-72 °C (lit.^[5] 71-72 °C) ¹H NMR (600 MHz, Chloroform-*d*) δ 8.82 (d, *J* = 6 Hz, 2H), 7.83 (d, *J* = 7.2 Hz, 2H), 7.66 (t, *J* = 7.8 Hz, 1H), 7.59 (d, *J* = 6 Hz, 2H), 7.53 (t, *J* = 7.8 Hz, 2H). ¹³C NMR (150 MHz, Chloroform-*d*) δ 195.1, 150.3, 144.4, 135.9, 133.6, 130.1, 128.7, 122.9.

1-Tetralone^[1] (2f). Compound 2f was obtain as light yellow liquid (289 mg, 99% yield). ¹H NMR (500 MHz, Chloroform-*d*) δ 7.98 (dd, J = 8.0 Hz, 1.5Hz, 1H), 7.40 (td, J = 7.5 Hz, 1.5Hz, 1H), 7.24(t, J = 7.5 Hz, 1H), 7.19 (d, J = 7.5 Hz, 1H), 2.90 (t, J = 6.5 Hz, 1H), 2.59 (t, J = 6.5 Hz, 1H), 2.10-2.03 (m,1H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 197.9, 144.2, 133.1, 132.4, 128.6, 126.8, 126.3, 38.9, 29.4, 23.1.

1-Indanone^[1] (**2g**). Compound **2g** was obtain as light yellow solid (230 mg, 87% yield). m. p. 39-41 °C (lit.^[1] 40-42 °C) ¹H NMR (500 MHz, Chloroform-*d*) δ 7.76 (d, *J* = 7.5Hz, 1H), 7.59 (td, *J* = 7.5 Hz, 1.0 Hz, 1H), 7.49 (dt, *J* = 8 Hz, 1.0 Hz, 1H), 7.39-7.36 (m, 1H), 3.15 (t, *J* = 6 Hz, 2H), 2.70 (t, *J* = 6 Hz, 3H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 207.1, 155.1, 137.0, 134.6, 127.2, 126.7, 123.7, 36.2, 25.8.

3,4-dihydro-1H-2-benzopyran-1-one^[6] (**2h**). Compound **2h** was obtain as light yellow liquid. ¹H NMR (500 MHz, Chloroform-*d*) δ 8.04 (d, *J* = 7.5 Hz, 1H), 7.51 (tt, *J* = 7.5 Hz, 1.2 Hz, 1H), 7.35 (t, *J* = 7.5Hz, 1H), 7.24 (d, *J* = 7.5 Hz, 1H), 4.50 (t, *J* = 5.5 Hz, 2H), 3.03 (t, *J* = 6.0 Hz, 2H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 165.0, 139.5, 133.5, 130.1, 127.5, 127.2, 125.1, 67.2, 27.6.

Acetophenone^[1] (2i). Compound 2i was obtain as light yellow liquid (204 mg, 85% yield). ¹H NMR (500 MHz, Chloroform-*d*) δ 7.93-7.91 (m, 2H), 7.52 (tt, *J* = 7.5, 1.5 Hz, 1H), 7.43 – 7.40 (m, 2H), 2.55 (s, 3H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 198.06, 137.09, 133.09, 128.55, 128.28, 26.54.

n-Butyrophenone^[6] (**2j**). Compound **2j** was obtain as light yellow liquid (269 mg, 91% yield). ¹H NMR (500 MHz, Chloroform-*d*) δ 7.95-7.93 (m, 2H), 7.51 (tt, *J* = 7.5 Hz, 1.5 Hz, 1H), 7.42 (t, *J* = 7.5 Hz, 2H), 2.91 (t, *J* = 7.5 Hz, 2H), 1.75 (sext, *J* = 7.5 Hz, 2H), 0.98 (t, *J* = 7.5 Hz, 3H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 200.02, 136.92, 132.61, 129.31, 40.23, 17.53, 13.64.

2'-Bromoacetophenone^[7] (**2k**). Compound **2k** was obtain as light yellow liquid (394 mg, 99% yield). ¹H NMR (500 MHz, Chloroform-*d*) δ 7.57 (dd, J = 8 Hz, 1 Hz, 1H), 7.43 (dd, J = 7.5 Hz, 2 Hz, 1H), 7.33 (td, J = 7.5 Hz, 1 Hz, 1H), 7.26 (td, J = 8 Hz, 1.5Hz, 1H), 2.59 (s, 3H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 201.1, 141.3, 133.7, 131.7,128.8, 127.3, 118.7, 30.1.

4'-Bromoacetophenone^[1] **(2I).** Compound **2I** was obtain as light yellow solid (386 mg, 97% yield). m. p. 51-52 °C (lit.^[1] 48-50 °C) ¹H NMR (500 MHz, Chloroform*d*) δ 7.83 (d, *J* = 8.5 Hz, 1H), 7.61 (d, *J* = 8.5 Hz, 1H), 2.59 (s, 3H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 197.0, 135.9, 131.9, 129.8, 128.3, 26.5.

2-Bromoacetophenone^[8] **(2m).** Compound **2m** was obtain as off-white solid (394 mg, 99% yield). m. p. 46-47 °C (lit.^[8] 46-48 °C) ¹H NMR (500 MHz, Chloroform-*d*) δ 8.01-7.99 (m, 2H), 7.62 (tt, *J* = 7.5 Hz, 1.5Hz, 1H), 7.53-7.49 (m, 2H), 4.47 (s, 2H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 191.3, 134.0, 133.9, 128.9, 30.9.

4'-Cyanoacetophenone^[9] **(2n).** Compound **2n** was obtain as white solid (200 mg, 69% yield). m. p. 57-59 °C (lit.^[21] 59-60 °C) ¹H NMR (500 MHz, Chloroform-*d*) δ 8.05 (d, J = 8 Hz, 2H), 7.78 (d, J = 8.5 Hz, 2H), 2.65 (s, 3H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 196.5, 139.9, 132.5, 128.7, 117.9, 116.4, 26.7.

4'-Nitroacetophenone^[1] **(20).** Compound **20** was obtain as yellow solid (280 mg, 85 % yield). m. p. 78-79 °C (lit.^[1] 80-81 °C) ¹H NMR (500 MHz, Chloroform-*d*) δ

8.30 (dt, J = 9, 2 Hz, 2H), 8.10 (dt, J = 9, 2 Hz, 2H), 2.68 (s, 3H). ¹³C NMR (125 MHz, Chloroform-d) δ 196.3, 150.3, 141.4, 129.3, 123.8, 26.9.

4'-Ethylacetophenone^[10] **(2p).** Compound **2p** was obtain as light yellow liquid (97 mg, 33 % yield).¹H NMR (500 MHz, Chloroform-*d*) δ 7.89 (d, J = 8.0 Hz, 2H), 7.28 (d, J = 8.0 Hz, 2H), 2.71 (q, J = 7.5 Hz, 2H), 2.58 (s, 3H), 1.26 (t, J = 7.5 Hz, 3H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 197.8, 150.0, 134.9, 128.5, 128.0, 28.9, 26.5, 15.1.

4'-Methylacetophenone^[11] (**2q).** Compound **2q** was obtain as light yellow liquid (99 mg, 37 % yield). ¹H NMR (500 MHz, Chloroform-d) δ 7.84 (d, J = 8.5 Hz, 2H), 7.24 (d, J = 8 Hz, 2H), 2.55, (s, 3H), 2.39 (s, 3H). ¹³C NMR (150 MHz, Chloroform-d) δ 197.6, 143.7, 134.6, 129.1, 128.3, 26.3, 21.5.

4'-Acetoxyacetophenone^{[12][13]} **(2r).** Compound **2r** was obtain as off-white solid (238 mg, 67 % yield). m. p. 51-52 °C (lit.^[12] 52-54 °C). ¹H NMR (500 MHz, Chloroform-*d*) δ 7.89-7.83 (m, 2H), 7.10-7.05 (m, 2H), 2.45 (s, 1H), 2.19 (s, 1H). ¹³C NMR (125 MHz, Chloroform-*d*) δ 196.4, 168.5, 154.1, 134.4, 129.6, 121.5, 26.6, 20.7.

4'-Methoxyacetophenone^[11] **(2s).** Compound **2s** was obtain as light yellow liquid (105 mg, 35 % yield). ¹H NMR (600 MHz, Chloroform-*d*) δ 7.75-7.72 (m, 2H), 6.75-6.72 (m, 2H), 3.66-3.65 (m, 3H), 2.35-2.34 (m, 3H). ¹³C NMR (150 MHz, Chloroform-*d*) δ 196.4, 163.3, 130.4, 113.5, 55.2, 26.1.

NMR Spectrum of products

2a:



2b:



2c:



2d:



2e:



2f:



2g:



2h:



2i:



2j:



2k:



2l:



2m:















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