

Electronic Supplementary Information

Preparation of Hybrid Paper Electrode Based on Hexagonal Boron Nitride Integrated Graphene Nanocomposite for Free-standing Flexible Supercapacitors

Jerome Rajendran^a, Anatoly N. Reshetilov^b and Ashok K. Sundramoorthy^{a*}

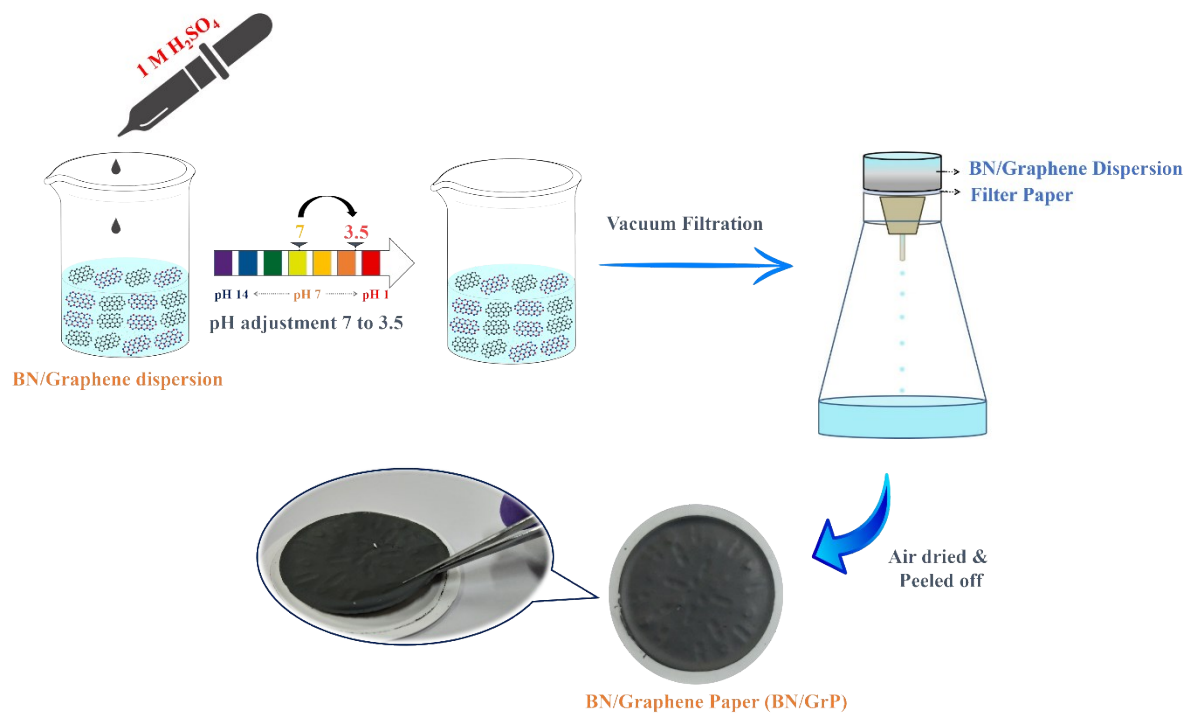
^aDepartment of Chemistry, SRM Institute of Science and Technology, Kattankulathur-603
203, Tamil Nadu, India

^bG.K. Skryabin Institute of Biochemistry and Physiology of Microorganisms of the Russian Academy of Sciences (IBPM RAS),-Subdivision of “Federal Research Center Pushchino Biological Research Center of the Russian Academy of Sciences”(FRC PBRC RAS),
142290, Pushchino, Moscow oblast, Russia

***Corresponding author**

Email: ashokkus@srmist.edu.in

Supporting Figures/Scheme/Table



Scheme S1. Schematic illustration for the preparation of freestanding BN/GrP flexible paper using BN/graphene dispersion.

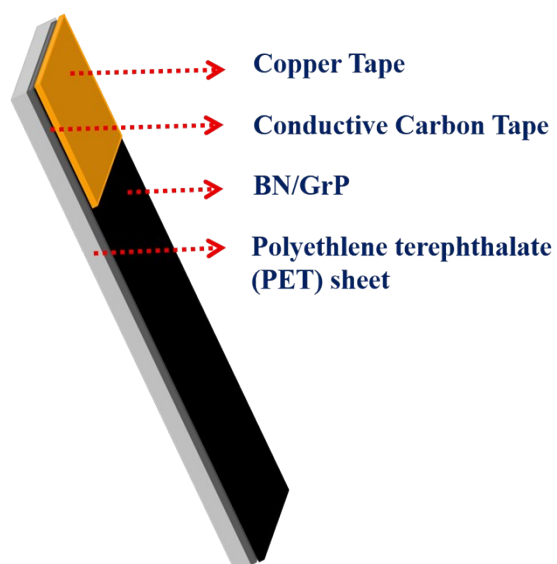


Fig. S1. Preparation of BN/GrP electrode for electrochemical measurements of supercapacitor.

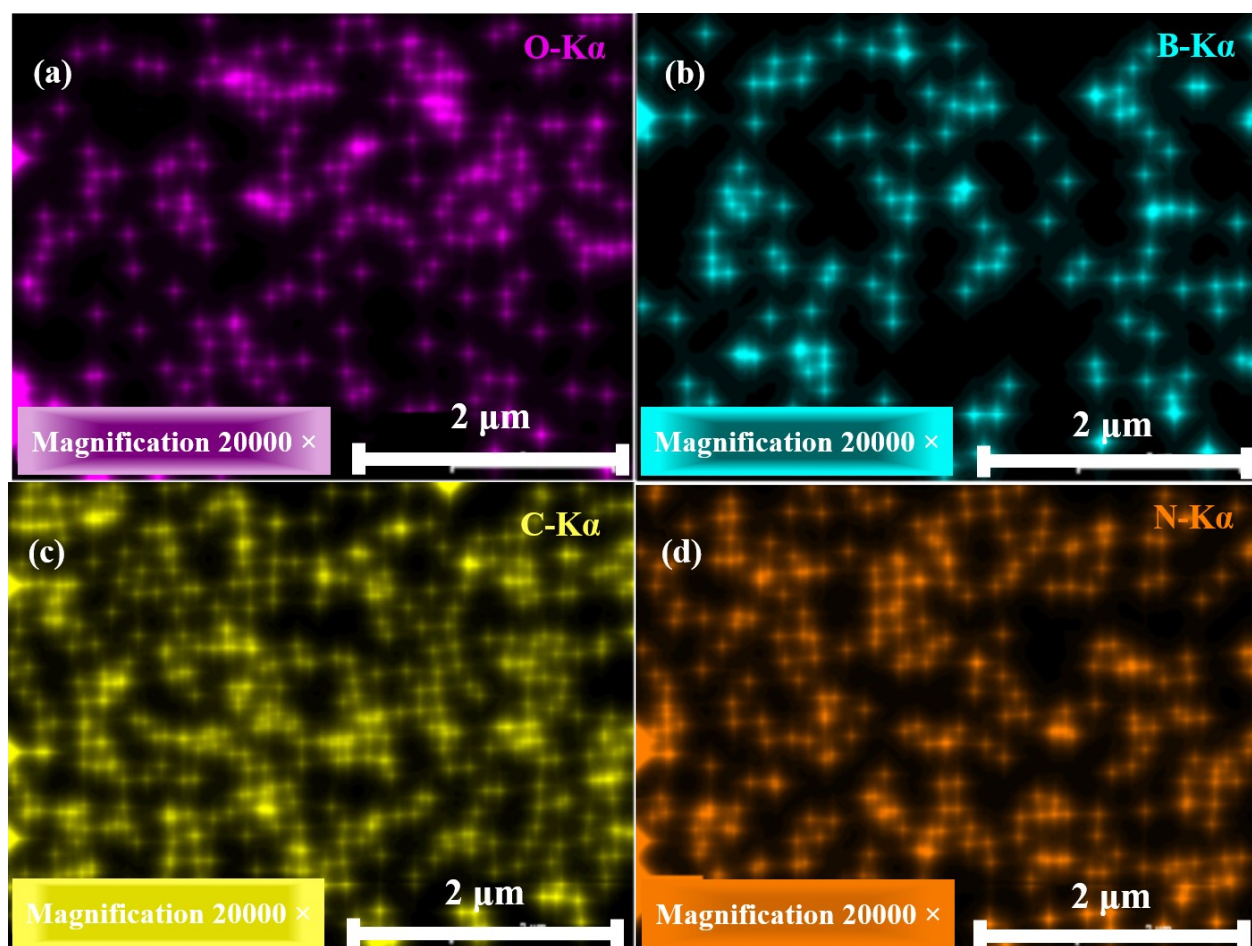


Fig. S2. EDX mapping analysis of BN/GrP: (a) oxygen, (b) boron, (c) carbon and (d) nitrogen.

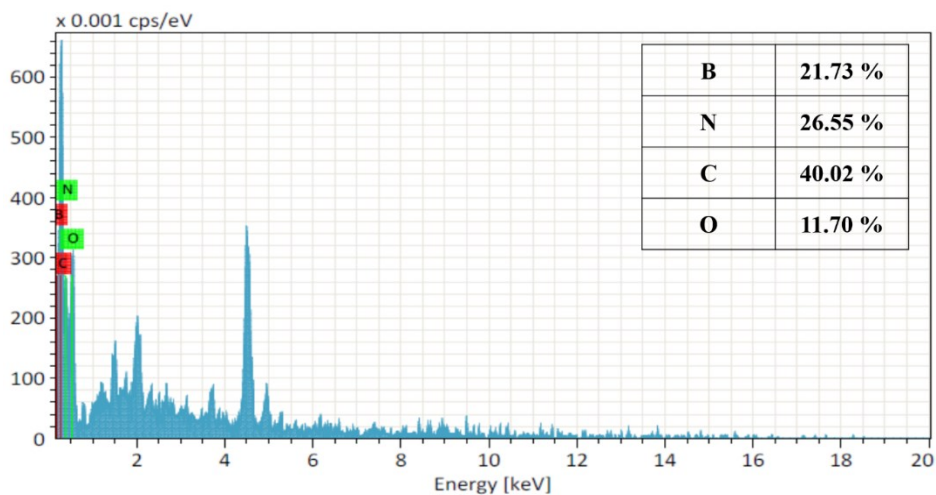


Fig. S3. EDX spectrum of BN/GrP (inset: atomic percentages of elements present in BN/GrP).

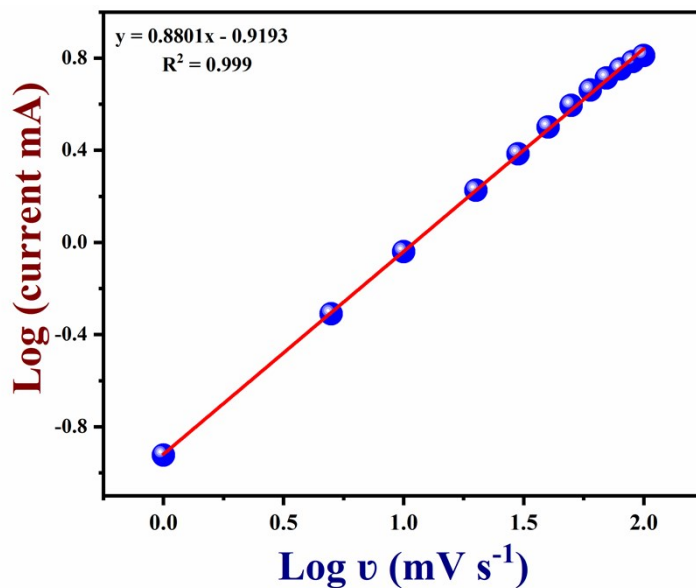


Fig. S4. The log of peak current was plotted against the log of scan rate to determine the “b” value of the BN/GrP from the anodic curve of different scan rates.

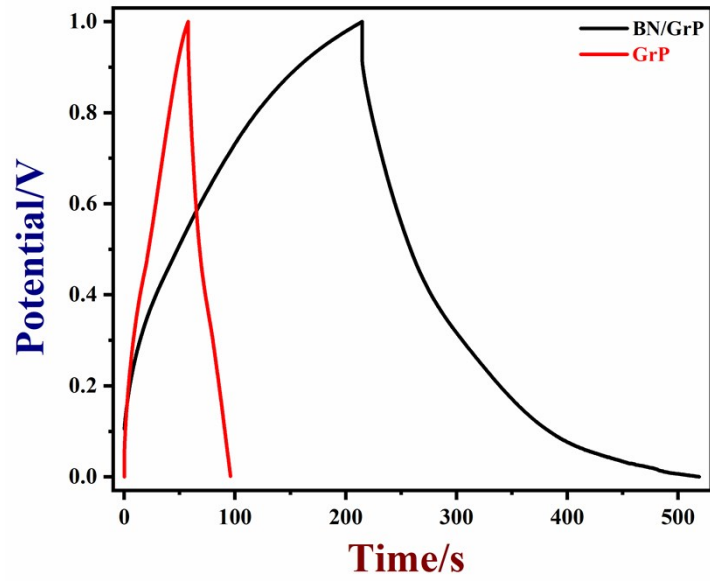


Fig. S5. Galvanostatic charge/discharge (GCD) curves of GrP and BN/GrP electrodes at a current density of 1 Ag^{-1} .

Table S1. Comparison of specific capacitances, retention and power density of the reported supercapacitors.

Active Material	Method of preparation	Electrolyte solution	Capacitance	Power density	Cycles/Retention	Ref.
rGO film	Brush-coating/annealing	6 M KOH	81.7 Fg ⁻¹ at 10 mV s ⁻¹	1.5 kW.kg ⁻¹ at 3 Ag ⁻¹	1500 - 93.8%	¹
Graphene paper	freeze-drying	1 M H ₂ SO ₄	172 Fg ⁻¹ at 1 Ag ⁻¹	-	5000 – 99%	²
MnO ₂ -coated 3D graphene	Electrochemical deposition on pressed Ni foam	0.5 M Na ₂ SO ₄	130 Fg ⁻¹ at 1 mV s ⁻¹	62 W.kg-1	5000 – 82%	³
GrP/10-MnO ₂	Vacuum filtration and electrochemical deposition	0.1 M Na ₂ SO ₄	385.2 Fg ⁻¹ at 1 mV s ⁻¹	3.72 kW.kg ⁻¹ at 4 Ag ⁻¹	5000 – 85.4%	⁴
GrP	Vacuum filtration	1 M H ₂ SO ₄	195.8 Fg ⁻¹ at 1 mV s ⁻¹	-	10000 - 93.2%	⁵
BN/GrP	Vacuum filtration method	1M H ₂ SO ₄	321.95 Fg ⁻¹ at 0.5 Ag ⁻¹	3.5 kW.kg ⁻¹ at 9 Ag ⁻¹	6000 - 96.3%	This work

Supporting references

- 1 W. Liu, X. Yan, J. Lang, C. Peng and Q. Xue, *J. Mater. Chem.*, 2012, **22**, 17245–17253.
- 2 F. Liu, S. Song, D. Xue and H. Zhang, *Adv. Mater.*, 2012, **24**, 1089–1094.
- 3 Y. He, W. Chen, X. Li, Z. Zhang, J. Fu, C. Zhao and E. Xie, *ACS Nano*, 2013, **7**, 174–182.
- 4 O. Sadak, W. Wang, J. Guan, A. K. Sundramoorthy and S. Gunasekaran, *ACS Appl. Nano Mater.*, 2019, **2**, 4386–4394.
- 5 O. Sadak, A. K. Sundramoorthy and S. Gunasekaran, *Carbon*, 2018, **138**, 108–117.