

Supporting Information of the manuscript entitled:

Transformation of the *cyclo*-P₄ ligand in [Cp'''Co(η⁴-P₄)]

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1. Synthetic procedures and experimental details

Synthetic Procedures: All manipulations were performed under an atmosphere of dry argon using standard glove-box and Schlenk techniques. All solvents were degassed and purified by standard procedures. The compounds $[\text{Cp}^*\text{Co}(\eta^4\text{-P}_4)]$ (**1**)^[1], $\text{LiCH}_2\text{SiMe}_3$ ^[2] and $\text{K}[\text{CpFe}(\text{CO})_2]$ ^[3] were prepared according to literature procedures. 2,2,2-cryptand, NaOH and $t\text{BuLi}$ were purchased commercially.

The NMR spectra were recorded with a Bruker Avance 400 spectrometer (^1H : 400.13 MHz, ^{31}P : 161.976 MHz). The chemical shifts are given in ppm referenced to external SiMe_4 (^1H) and H_3PO_4 (^{31}P). Elemental analyses were determined with an Elementar Vario EL III apparatus. The ESI-MS spectra were acquired on a ThermoQuest Finnigan MAT TSQ 7000 mass spectrometer.

1.1 Synthesis of $[K(dme)]_2[(Cp'''Co)_2(\mu,\eta^3:\eta^3-P_8)]$ (**2a**)

A solution of $K[CpFe(CO)_2]$ (260 mg, 1.201 mmol, 1 eq) in dme was added to a stirred solution of **1** (500 mg, 1.201 mmol, 1 eq) in dme at $-50\text{ }^\circ\text{C}$ while the color changed from dark red to purple. The reaction mixture was warmed to room temperature and after 2 h a color change from purple to brown to green was observed. The solvent was removed *in vacuo*. The residue was washed with n-hexane to remove $[CpFe(CO)_2]_2$ (brown solution). n-Hexane was added and dme was added slowly until complete dissolution. After storage at $-30\text{ }^\circ\text{C}$ overnight, **2a** could be obtained as dark green blocks. The supernatant was decanted off, washed with n-hexane and the crystals dried *in vacuo*.

Yield: 260 mg (40%)

1H NMR (thf- d_8 , $25\text{ }^\circ\text{C}$): δ [ppm] = 3.96 (s, 4H, $C_5H_2^tBu_3$), 3.58 (s, 6H, dme), 3.28 (s, 8.5H, dme), 1.31 (s, 36H, $C_5H_2^tBu_3$), 1.22 (s, 18H, $C_5H_2^tBu_3$).

$^{31}P\{^1H\}$ NMR (thf- d_8 , $25\text{ }^\circ\text{C}$): δ [ppm] = 96.2 (m, 2P, P_A , $P_{A'}$), 83.1 (m, 4P, P_M , $P_{M'}$, $P_{M''}$, $P_{M'''}$), -124.2 (m, 2P, P_X , $P_{X'}$) (coupling constants given in table S1).

^{31}P NMR (thf- d_8 , $25\text{ }^\circ\text{C}$): δ [ppm] = 96.2 (m, 2P, P_A , $P_{A'}$), 83.1 (m, 4P, P_M , $P_{M'}$, $P_{M''}$, $P_{M'''}$), -124.2 (m, 2P, P_X , $P_{X'}$) (coupling constants taken from simulation of $^{31}P\{^1H\}$ NMR spectrum).

ESI-MS (dme): m/z = 1279.14 (50 %, $[(Cp'''Co)_3P_{13}]^-$), 1217.20 (100 %, $[(Cp'''Co)_3P_{11}]^-$), 955.97 (23 %, $[(Cp'''Co)_2P_{12}]^-$), 895.07 (21 %, $[(Cp'''Co)_2P_{10}H]^-$), 833.12 (23%, $[(Cp'''Co)_2P_8H]^-$).

EA $C_{42}H_{78}O_4K_2P_8Co_2$: calc [%]: C 46.24; H 7.21; found [%]: C 46.11; H 7.04.

1.2 Synthesis of $[K(2,2,2\text{-crypt})]_2[(Cp''Co)_2(\mu,\eta^3:\eta^3\text{-P}_8)]$ (**2b**)

A solution of $K[CpFe(CO)_2]$ (101.7 mg, 0.47 mmol, 1 eq) in dme was added to a stirred solution of **I** (196 mg, 0.47 mmol, 1 eq) in dme at $-50\text{ }^\circ\text{C}$ while the color changed from dark red to purple. The reaction mixture was warmed to room temperature and after 2 h a color change from purple to brown to green was observed. The solvent was removed *in vacuo*. The residue was washed with n-hexane to remove $[CpFe(CO)_2]_2$ (brown solution). The residue was dissolved in dme (dark green) and 2,2,2-cryptand (177.2 mg, 0.47 mmol, 1 eq) in dme was added. After stirring for 5 minutes some green precipitate was formed. Acetonitrile was added dropwise until complete dissolution. After storage at $-30\text{ }^\circ\text{C}$, **2b** can be obtained as dark green plates. The supernatant was decanted off, washed with n-hexane and the crystals dried *in vacuo*.

Yield: 94 mg (24 %).

^1H NMR (CD_3CN , $25\text{ }^\circ\text{C}$): δ [ppm] = 3.94 (s, 4H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 3.56 (s, 24H, 2,2,2-cryptand), 3.52 (t, 24H, 2,2,2-cryptand, $^1J_{\text{HH}} = 4.7\text{ Hz}$), 2.53 (t, 24H, 2,2,2-cryptand, $^1J_{\text{HH}} = 4.7\text{ Hz}$), 1.28 (s, 36H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 1.21 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$).

$^{31}\text{P}\{^1\text{H}\}$ NMR (CD_3CN , $25\text{ }^\circ\text{C}$): δ [ppm] = 91.6 (m, 6P, P_8), -124.1 (m, 2P, P_8).

^{31}P NMR (CD_3CN , $25\text{ }^\circ\text{C}$): δ [ppm] = 91.6 (m, 6P, P_8), -124.1 (m, 2P, P_8).

ESI-MS (dme): m/z = 1279.14 (100 %, $[(Cp''Co)_3P_{13}]^-$), 1217.20 (82 %, $[(Cp''Co)_3P_{11}]^-$).

EA $\text{C}_{70}\text{H}_{130}\text{O}_{12}\text{N}_4\text{K}_2\text{P}_8\text{Co}_2$: calc [%]: C 50.54; H 7.88; N 3.37; found [%]: C 49.86; H 7.68; N 3.09.

1.3 Synthesis of [Li(2,2,2-crypt)][Cp^{'''}Co(η³-P₄^tBu)] (4)

a) A solution of ^tBuLi in pentane (0.96 ml, 0.249 mol/L, 0.24 mmol, 1 eq) was added to a stirred solution of **1** (100 mg, 0.24 mmol, 1 eq) in thf at -80 °C while the color changed immediately from red to brown. After five minutes a solution of 2,2,2-cryptand (90.4 mg, 0.24 mmol, 1 eq) in thf was added and the reaction mixture warmed to room temperature. The solvent was reduced *in vacuo* and NMR spectra with a C₆D₆ capillary was recorded revealing the clean formation of **4**. The thf solution was layered with n-hexane at room temperature. After a few days, **4** can be obtained as brown flat rods. The supernatant was decanted off and the obtained crystals dried *in vacuo*.

Yield: 98 mg (48 %).

¹H NMR (thf-d₈, 25 °C): δ [ppm] = 4.90 (s, 2H, C₅H₂^tBu₃), 3.70 (s, 12H, 2,2,2-crypt), 3.64 (pt, ³J_{HH} = 5.1 Hz, 12H, 2,2,2-crypt), 2.72 (pt, ³J_{HH} = 5.1 Hz, 12H, 2,2,2-crypt), 1.47 (s, 9H, P₄^tBu), 1.36, 1.35, 1.34 (three overlapping singlets, 27H, C₅H₂^tBu₃).

³¹P{¹H} NMR (thf-d₈, 25 °C): δ [ppm] = 94.1 (t, ¹J_{PP} = 255.63 Hz, ²J_{PP} = 2.48 Hz, 1P, P_A), 88.7 (t, ¹J_{PP} = 186.23 Hz, ²J_{PP} = 3.97 Hz, 1P, P_M), -25.9 (dd, ¹J_{PP} = 255.63 Hz, ¹J_{PP} = 186.23 Hz, 2P, P_X, P_{X'}) (coupling constants obtained from the simulation, cf. Figure S4).

³¹P NMR (thf-d₈, 25 °C): δ [ppm] = 94.1 (t, ¹J_{PP} = 255.60 Hz, ²J_{PP} = 1.49 Hz, ⁵J_{PH} = 0.77 Hz, 1P, P_A), 88.7 (tdec, ¹J_{PP} = 186.26 Hz, ²J_{PP} = -0.30 Hz, ³J_{PH} = 9.12 Hz 1P, P_M), -26.0 (dd, ¹J_{PP} = 255.60 Hz, ¹J_{PP} = 186.26 Hz, ⁴J_{PH} = -1.12 Hz, 2P, P_X, P_{X'}) (coupling constants obtained from the simulation, cf. Figure S4).

EA C₃₉H₇₄N₂O₆P₄CoLi: calc [%]: C 54.67; H 8.71; N 3.27; found [%]: C 54.35; H 8.42; N 3.27.

Despite of several attempts it was not possible to obtain a proper mass spectrum (ESI), due to the extreme redox sensitivity of **4**.

b) A solution of ^tBuLi in pentane (0.96 ml, 0.249 mol/L, 0.24 mmol, 1 eq) was added to a stirred solution of **1** (100 mg, 0.24 mmol, 1 eq) in thf at -80 °C while the color changed immediately from red to brown. The reaction mixture was warmed to room temperature. The solvent was reduced *in vacuo* and NMR spectra in thf-d₈ were recorded showing a mixture of **3**, **4**, **5** and **6**.

Compound **5**:

³¹P{¹H} NMR (thf-d₈, 25 °C): δ [ppm] = 231.9 (m, ¹J_{PP} = 470.35 Hz, ¹J_{PP} = 441.32 Hz, ²J_{PP} = -1.21 Hz, ²J_{PP} = 13.24 Hz, 1P, P_A), 130.9 (ddd, ¹J_{PP} = 346.33 Hz, ¹J_{PP} = 363.12 Hz, ²J_{PP} = 14.02 Hz, ²J_{PP} = -4.69 Hz, ²J_{PP} = 69.86 Hz, ³J_{PP} = -1.52 Hz, 2P, P_M), 74.7 (dddd, ¹J_{PP} = 470.35 Hz, ¹J_{PP} = 358.96 Hz, ²J_{PP} = -4.69 Hz, ²J_{PP} = 69.86 Hz, ²J_{PP} = -16.59 Hz, ²J_{PP} = -14.40, ³J_{PP} = 1.02 Hz, 2P, P_N), 1.34 (dt, ¹J_{PP} = 441.32 Hz, ¹J_{PP} = 346.33 Hz, ²J_{PP} = -14.40 Hz, ²J_{PP} = 6.80 Hz, 1P, P_O), -123.9 (t, ¹J_{PP} = 363.12 Hz, ¹J_{PP} = 358.96 Hz, ²J_{PP} = 6.80 Hz, ³J_{PP} = -1.52 Hz, ³J_{PP} = 1.02 Hz, ⁴J_{PP} = 20.96 Hz, 2P, P_X) (coupling constants obtained from the simulation, cf. Figure S8).

³¹P NMR (thf-d₈, 25 °C): δ [ppm] = 231.9 (m, 1P, P_A), 130.9 (m, 2P, P_M), 74.7 (m, 2P, P_N), 1.34 (m, 1P, P_O), -123.9 (m, 2P, P_X) (signals 231.9 and 1.34 ppm show further broadening caused by proton coupling, too bad resolved for simulation).

Compound **6**:

³¹P{¹H} NMR (thf-d₈, 25 °C): δ [ppm] = 106.8 (ddd, ¹J_{PP} = 400.71 Hz, ¹J_{PP} = 342.90 Hz, ²J_{PP} = 65.46 Hz, ²J_{PP} = 5.97 Hz, 1P, P_A), 98.9 (tdd, ¹J_{PP} = 352.77 Hz, ¹J_{PP} = 348.39, ²J_{PP} = 65.46 Hz, ²J_{PP} = 12.84 Hz, 1P, P_B), 77.7 (dddd, ¹J_{PP} = 352.77 Hz, ¹J_{PP} = 322.67 Hz, ²J_{PP} = 5.97 Hz, ²J_{PP} = 10.00 Hz, 1P, P_C), 31.3 (dddd, ¹J_{PP} = 400.71 Hz, ¹J_{PP} = 322.67 Hz, ²J_{PP} = 12.84, ²J_{PP} = 7.23 Hz, 1P, P_M), -146.9 (t, ¹J_{PP} = 342.90 Hz, ¹J_{PP} = 348.39 Hz, ²J_{PP} = 10.00 Hz, ²J_{PP} = 7.23 Hz, 1P, P_X) coupling constants obtained from the simulation, cf. Figure S9).

^{31}P NMR (thf- d_8 , 25 °C): δ [ppm] = 106.8 (m, 1P, P_A), 98.9 (m, 1P, P_B), 77.7 (m, 1P, P_C), 31.3 (m, 1P, P_M), -146.9 (m, 1P, P_X) (signals 77.7 and 31.3 ppm show further broadening caused by proton coupling, too bad resolved for simulation).

Despite of several attempts it was not possible to obtain proper mass spectra (ESI) of **5** and **6**, probably due to the extreme redox sensitivity. No elemental analysis could be conducted since compound **5** and **6** were obtained as a mixture of compounds.

1.4 Synthesis of [Li(2,2,2-crypt)][Cp'''Co(η^3 -P₄CH₂SiMe₃)] and [Li(12-c-4)₂][Cp'''Co(η^3 -P₄CH₂SiMe₃)]

a) Reaction with 2,2,2-cryptand: A solution of LiCH₂SiMe₃ in thf (22.6 mg, 0.24 mmol, 1 eq) was added to a stirred solution of **1** (100 mg, 0.24 mmol, 1 eq) in thf at -80 °C while the color changed immediately from red to brown. After five minutes a solution of 2,2,2-cryptand (90.4 mg, 0.24 mmol, 1 eq) in thf was added and the reaction mixture warmed to room temperature. The solvent was reduced *in vacuo* and NMR spectra in thf-d₈ were recorded revealing the formation two isomers **7a** and **7b** in a ratio of 1:0.25 beside other products. The residue was dissolved in a small amount thf and precipitated with pentane (brown oily solid). The NMR spectra in thf-d₈ of the obtained oil reveals a clean mixture of **7a** and **7b** in a ratio of 1:0.36.

Compound [Li(2,2,2-crypt)]**7a**:

¹H NMR (thf-d₈, 25 °C): δ [ppm] = 4.80 (s, 2H, C₅H₂^{*t*}Bu₃), 3.73 (s, 12H, 2,2,2-crypt), 3.67 (t, ³J_{HH} = 4.9 Hz, 12H, 2,2,2-crypt), 2.75 (t, ³J_{HH} = 4.9 Hz, 12H, 2,2,2-crypt), 2.20 (s, 2H, CH₂SiMe₃), 1.48 (s, 9H, C₅H₂^{*t*}Bu₃), 1.34 (s, 18H, C₅H₂^{*t*}Bu₃), 0.02 (s, 9H, CH₂SiMe₃).

³¹P{¹H} NMR (thf-d₈, 25 °C): δ [ppm] = 52.3 (td, ¹J_{PP} = 260.89 Hz, ²J_{PP} = 10.95 Hz, 1P, P_A), 45.9 (td, ¹J_{PP} = 152.24 Hz, ²J_{PP} = 10.95 Hz, 1P, P_M), 18.8 (dd, ¹J_{PP} = 260.89 Hz, ¹J_{PP} = 152.24 Hz, ²J_{PP} = -8.00 Hz, 2P, P_{XX}), (coupling constants obtained from the simulation, cf. Figure S12).

³¹P NMR (thf-d₈, 25 °C): δ [ppm] = 52.3 (td, ¹J_{PP} = 260.89 Hz, ²J_{PP} = 10.95 Hz, 1P, P_A), 45.9 (td, ¹J_{PP} = 152.24 Hz, ²J_{PP} = 10.95 Hz, 1P, P_M), 18.8 (dd, ¹J_{PP} = 260.89 Hz, ¹J_{PP} = 152.24 Hz, ²J_{PP} = -8.00 Hz, 2P, P_{XX}), (coupling constants obtained from the simulation of the ³¹P{¹H}, no further coupling observed, cf. Figure S12).

Compound [Li(2,2,2-crypt)]**7b**:

¹H NMR (thf-d₈, 25 °C): δ [ppm] = 4.75 (s, 2H, C₅H₂^{*t*}Bu₃), 3.73 (s, 12H, 2,2,2-crypt), 3.67 (t, ³J_{HH} = 4.9 Hz, 12H, 2,2,2-crypt), 2.75 (t, ³J_{HH} = 4.9 Hz, 12H, 2,2,2-crypt), 1.36 (s, 18H, C₅H₂^{*t*}Bu₃), 1.31 (s, 2H, CH₂SiMe₃), 1.26 (s, 9H, C₅H₂^{*t*}Bu₃), -0.07 (s, 9H, CH₂SiMe₃).

³¹P{¹H} NMR (thf-d₈, 25 °C): δ [ppm] = 39.4 (td, ¹J_{PP} = 135.82 Hz, ²J_{PP} = 14.38 Hz, 1P, P_A), 3.8 (dd, ¹J_{PP} = 276.36 Hz, ¹J_{PP} = 135.82 Hz, ²J_{PP} = -84.01 Hz, 2P, P_{MM'}), -78.2 (td, ¹J_{PP} = 276.36 Hz, ²J_{PP} = 14.38 Hz, 1P, P_X), (coupling constants obtained from the simulation, cf. Figure S12).

³¹P NMR (thf-d₈, 25 °C): δ [ppm] = 39.4 (td, ¹J_{PP} = 135.82 Hz, ²J_{PP} = 14.38 Hz, 1P, P_A), 3.8 (dd, ¹J_{PP} = 276.36 Hz, ¹J_{PP} = 135.82 Hz, ²J_{PP} = -84.01 Hz, 2P, P_{MM'}), -78.2 (td, ¹J_{PP} = 276.36 Hz, ²J_{PP} = 14.38 Hz, 1P, P_X), (coupling constants obtained from the simulation of the ³¹P{¹H}, no further coupling observed, cf. Figure S12).

Despite of several attempts it was not possible to obtain proper mass spectrum (ESI) of **7a,b** probably due to the extreme redox sensitivity. No elemental analysis could be conducted since compound **7** was only obtained as an oily substance when 2,2,2-cryptand is used.

b) Reaction with 12-c-4: A solution of $\text{LiCH}_2\text{SiMe}_3$ in thf (22.6 mg, 0.24 mmol, 1 eq) was added to a stirred solution of **1** (100 mg, 0.24 mmol, 1 eq) in thf at $-80\text{ }^\circ\text{C}$ while the color changed immediately from red to brown. After five minutes a solution of 12-c-4 in dme (0.60 mL, 0.48 mmol, 0.8102 mol/L, 2 eq) was added and the reaction mixture warmed to room temperature. The solvent was reduced *in vacuo* and NMR spectra in thf-d_8 were recorded revealing the formation two isomers **7a** and **7b** in a ratio of 1:1.33 beside other products. The residue was dissolved in a small amount thf and precipitated with pentane. The NMR spectra of the precipitated solid reveals a clean mixture of **7a** and **7b** in a ratio of 1:1.78. The supernatant was decanted off, the residue dissolved in thf and layered with n-hexane at room temperature. After a few days, very few single crystals of **7a** could be obtained.

Yield: 20 mg (precipitated solid, 10 %).

Compound $[\text{Li}(12\text{-c-4})_2]\mathbf{7a}$:

^1H NMR (thf-d_8 , $25\text{ }^\circ\text{C}$): δ [ppm] = 4.83 (s, 2H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 3.77 (s, 32H, 12-c-4), 2.20 (s, 2H, CH_2SiMe_3), 1.48 (s, 9H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 1.34 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 0.02 (s, 9H, CH_2SiMe_3).

$^{31}\text{P}\{^1\text{H}\}$ NMR (thf-d_8 , $25\text{ }^\circ\text{C}$): δ [ppm] = 50.1 (td, $^1J_{\text{PP}} = 259.18\text{ Hz}$, $^2J_{\text{PP}} = 10.53\text{ Hz}$, 1P, P_A), 46.5 (td, $^1J_{\text{PP}} = 152.95\text{ Hz}$, $^2J_{\text{PP}} = 10.53\text{ Hz}$, 1P, P_M), 17.42 (dd, $^1J_{\text{PP}} = 259.18\text{ Hz}$, $^1J_{\text{PP}} = 152.95\text{ Hz}$, $^2J_{\text{PP}} = -8.00\text{ Hz}$, 2P, P_{XX}), (coupling constants obtained from the simulation, cf. Table S8).

^{31}P NMR (thf-d_8 , $25\text{ }^\circ\text{C}$): δ [ppm] = 50.1 (td, $^1J_{\text{PP}} = 259.18\text{ Hz}$, $^2J_{\text{PP}} = 10.53\text{ Hz}$, 1P, P_A), 46.5 (td, $^1J_{\text{PP}} = 152.95\text{ Hz}$, $^2J_{\text{PP}} = 10.53\text{ Hz}$, 1P, P_M), 17.42 (dd, $^1J_{\text{PP}} = 259.18\text{ Hz}$, $^1J_{\text{PP}} = 152.95\text{ Hz}$, $^2J_{\text{PP}} = -8.00\text{ Hz}$, 2P, P_{XX}), (coupling constants obtained from the simulation of the $^{31}\text{P}\{^1\text{H}\}$, no further coupling observed, cf. Table S8).

Compound $[\text{Li}(12\text{-c-4})_2]\mathbf{7b}$:

^1H NMR (thf-d_8 , $25\text{ }^\circ\text{C}$): δ [ppm] = 4.79 (s, 2H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 3.77 (s, 32H, 12-c-4), 1.36 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 1.32 (s, 2H, CH_2SiMe_3), 1.26 (s, 9H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), -0.07 (s, 9H, CH_2SiMe_3).

$^{31}\text{P}\{^1\text{H}\}$ NMR (thf-d_8 , $25\text{ }^\circ\text{C}$): δ [ppm] = 38.93 (td, $^1J_{\text{PP}} = 135.34\text{ Hz}$, $^2J_{\text{PP}} = 13.76\text{ Hz}$, 1P, P_A), 2.8 (dd, $^1J_{\text{PP}} = 273.40\text{ Hz}$, $^1J_{\text{PP}} = 135.34\text{ Hz}$, $^2J_{\text{PP}} = -84.00\text{ Hz}$, 2P, P_{MM}), -79.9 (td, $^1J_{\text{PP}} = 273.40\text{ Hz}$, $^2J_{\text{PP}} = 13.76\text{ Hz}$, 1P, P_X), (coupling constants obtained from the simulation, cf. Table S8).

^{31}P NMR (thf-d_8 , $25\text{ }^\circ\text{C}$): δ [ppm] = 38.93 (td, $^1J_{\text{PP}} = 135.34\text{ Hz}$, $^2J_{\text{PP}} = 13.76\text{ Hz}$, 1P, P_A), 2.8 (dd, $^1J_{\text{PP}} = 273.40\text{ Hz}$, $^1J_{\text{PP}} = 135.34\text{ Hz}$, $^2J_{\text{PP}} = -84.00\text{ Hz}$, 2P, P_{MM}), -79.9 (td, $^1J_{\text{PP}} = 273.40\text{ Hz}$, $^2J_{\text{PP}} = 13.76\text{ Hz}$, 1P, P_X), (coupling constants obtained from the simulation of the $^{31}\text{P}\{^1\text{H}\}$, no further coupling observed, cf. Table S8).

Precipitated solid **7a/7b**:

EA $\text{C}_{37}\text{H}_{72}\text{O}_8\text{P}_4\text{CoLiSi}$: calc [%]: C 51.51; H 8.41; found [%]: C 51.10; H 8.27.

Despite of several attempts it was not possible to obtain a proper mass spectrum (ESI) of **7a,b**, probably due to the extreme redox sensitivity.

1.5 Synthesis of $[\text{Li}(\text{dme})_3][(\text{Cp}'''\text{Co})_2(\mu, \eta^3: \eta^3\text{-P}_8\text{CH}_2\text{SiMe}_3)]$ (**8**)

A solution of $\text{LiCH}_2\text{SiMe}_3$ (11.3 mg, 0.12 mmol, 1 eq) in thf was added to a stirred solution of **1** (100 mg, 0.24 mmol, 1 eq) in thf at -80°C while the color changed from red to brown. The reaction mixture was warmed to room temperature, while the color changed to dark green. The solvent was removed *in vacuo*. n-Hexane was added to the residue (partial dissolution) and dme dropwise until complete dissolution. After storage at -30°C **8** can be obtained as dark green blocks. The supernatant was decanted off and washed with cold (-50°C) n-hexane. The crystals were dried *in vacuo*.

Yield: 47 mg (33%).

^1H NMR (thf- d_8 , 300 K): δ [ppm] = 4.53 (s, 2H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 4.15 (s, 2H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 3.43 (s, 12H, $\text{Li}(\text{dme})_3$), 3.27 (s, 12H, $\text{Li}(\text{dme})_3$), 1.80 (br, 2H, $\text{P}_8\text{CH}_2\text{SiMe}_3$), 1.34 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 1.31 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 1.29 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 0.32 (s, 9H, $\text{P}_8\text{CH}_2\text{SiMe}_3$).

$^1\text{H}\{^{31}\text{P}\}$ NMR (thf- d_8 , 300 K, decoupled at -114 ppm): δ [ppm] = 4.53 (d, $^3J_{\text{HH}} = 2.2$ Hz, 2H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 4.15 (d, d, $^3J_{\text{HH}} = 2.2$ Hz, 2H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 3.43 (s, 12H, $\text{Li}(\text{dme})_3$), 3.27 (s, 12H, $\text{Li}(\text{dme})_3$), 1.80 (d, $^2J_{\text{PH}} = 8.5$ Hz, 2H, $\text{P}_8\text{CH}_2\text{SiMe}_3$), 1.34 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 1.31 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 1.29 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 0.32 (s, 9H, $\text{P}_8\text{CH}_2\text{SiMe}_3$).

$^1\text{H}\{^{31}\text{P}\}$ NMR (thf- d_8 , 300 K, decoupled at +71 ppm): δ [ppm] = 4.53 (d, $^3J_{\text{HH}} = 2.2$ Hz, 2H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 4.15 (d, d, $^3J_{\text{HH}} = 2.2$ Hz, 2H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 3.43 (s, 12H, $\text{Li}(\text{dme})_3$), 3.27 (s, 12H, $\text{Li}(\text{dme})_3$), 1.80 (s, 2H, $\text{P}_8\text{CH}_2\text{SiMe}_3$), 1.34 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 1.31 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 1.29 (s, 18H, $\text{C}_5\text{H}_2^t\text{Bu}_3$), 0.32 (s, 9H, $\text{P}_8\text{CH}_2\text{SiMe}_3$).

$^{31}\text{P}\{^1\text{H}\}$ NMR (thf- d_8 , 193 K): δ [ppm] = 187.9 (q, $^1J_{\text{PP}} = 448.32$ Hz, $^1J_{\text{PP}} = 424.01$ Hz, $^2J_{\text{PP}} = -2.44$ Hz, $^2J_{\text{PP}} = 12.49$ Hz, 1P, P_A), 102.8 (td, $^1J_{\text{PP}} = 351.60$ Hz, $^1J_{\text{PP}} = 366.96$ Hz, $^2J_{\text{PP}} = -2.44$ Hz, $^2J_{\text{PP}} = 6.65$ Hz, $^2J_{\text{PP}} = 62.41$ Hz, $^2J_{\text{PP}} = -12.85$ Hz, $^3J_{\text{PP}} = 0.20$ Hz, 2P, P_M), 72.5 (ddd, $^1J_{\text{PP}} = 448.32$ Hz, $^1J_{\text{PP}} = 369.36$ Hz, $^2J_{\text{PP}} = 62.41$ Hz, $^2J_{\text{PP}} = -12.85$ Hz, $^2J_{\text{PP}} = -3.66$ Hz, $^2J_{\text{PP}} = -17.34$ Hz, $^3J_{\text{PP}} = 0.87$ Hz, 2P, P_N), 40.5 (dt, $^1J_{\text{PP}} = 424.01$ Hz, $^1J_{\text{PP}} = 351.60$ Hz, $^2J_{\text{PP}} = -17.34$ Hz, $^2J_{\text{PP}} = 7.69$ Hz, 1P, P_O), -118.8 (t, $^1J_{\text{PP}} = 366.96$ Hz, $^1J_{\text{PP}} = 369.36$ Hz, $^2J_{\text{PP}} = 12.49$ Hz, $^2J_{\text{PP}} = 7.69$ Hz, $^3J_{\text{PP}} = 0.20$ Hz, $^3J_{\text{PP}} = 0.87$ Hz, $^4J_{\text{PP}} = 19.47$ Hz, 2P, P_X). (coupling constants obtained from the simulation, cf. Figure S13).

^{31}P NMR (thf- d_8 , 193 K): δ [ppm] = 187.9 (m, 1P, P_A), 102.8 (m, 2P, P_M), 72.5 (m, 2P, P_N), 40.5 (m, 1P, P_O), -118.8 (m, 2P, P_X). (no further broadening or splitting of the signals detected).

ESI-MS (anion, dme): $m/z = 919.3$ (100 %, $[\text{M}]^-$).

No proper elemental analysis could be obtained despite of several attempts, even if the EA tube was filled in a glove box with the compound. Maybe the extreme oxidation sensibility of **8** is the reason.

1.6 Synthesis of [Na(dme)₂][Cp^{'''}Co(η³-P₄(O)H)] (10)

NaOH (9.6 mg, 0.24 mmol, 1 eq) and **1** (100 mg, 0.24 mmol, 1 eq) were weighed in together, dme was added and stirred for three days. The solvent was reduced *in vacuo*, layered with n-hexane and stored at -30 °C. After a few days, **10** can be obtained as dark brown blocks. The supernatant was decanted off, washed with n-hexane and dried *in vacuo*.

Yield: 84 mg (55 %).

¹H NMR (thf-d₈, 300 K): δ [ppm] = 7.78 (dtd, ¹J_{PH} = 356.77 Hz, ²J_{PH} = 20.56 Hz, ³J_{PH} = 5.12 Hz, 1H, P₄(O)H), 4.95 (s, 2H, C₅H₂^tBu₃), 3.43 (s, 8H, dme), 3.27 (s, 12H, dme), 1.36 (s, 18H, C₅H₂^tBu₃), 1.34 (s, 9H, C₅H₂^tBu₃).

¹H{³¹P} NMR (thf-d₈, 300 K): δ [ppm] = 7.78 (s, 1H, P₄(O)H), 4.95 (s, 2H, C₅H₂^tBu₃), 3.43 (s, 8H, dme), 3.27 (s, 12H, dme), 1.36 (s, 18H, C₅H₂^tBu₃), 1.34 (s, 9H, C₅H₂^tBu₃).

³¹P{¹H} NMR (thf + C₆D₆ capillary, 25 °C): δ [ppm] = 74.8 (dd, ¹J_{PP} = 262.67 Hz, ¹J_{PP} = 246.57 Hz, ²J_{PP} = -9.00 Hz, 2P, P_A), 29.0 (ddd, ¹J_{PP} = 262.67 Hz, ²J_{PP} = 4.92 Hz, 1P, P_M), -21.8 (t, ¹J_{PP} = 246.57 Hz, ²J_{PP} = 4.92 Hz, 1P, P_X) (coupling constants obtained from the simulation, cf. Figure S17).

³¹P NMR (thf + C₆D₆ capillary, 25 °C): δ [ppm] = 74.8 (dd, ¹J_{PP} = 262.71 Hz, ¹J_{PP} = 246.64 Hz, ²J_{PH} = 20.56 Hz, 2P, P_A), 29.0 (dddd, ¹J_{PP} = 262.67 Hz, ²J_{PP} = 4.92 Hz, ¹J_{PH} = 356.77 Hz, 1P, P_M), -21.8 (t, ¹J_{PP} = 246.57 Hz, ²J_{PP} = 4.92 Hz, ³J_{PH} = 5.12 Hz, 1P, P_X) (coupling constants obtained from the simulation, cf. Figure S17).

ESI-MS (anion, dme): *m/z* = 867.2 (100 %, [2M+H]⁻), 433.0 (12 %, [M]⁻).

EA C₂₅H₅₀O₅P₄CoNa: calc [%]: C 47.18; H 7.92; found [%]: C 46.94; H 7.87.

2. NMR spectroscopic investigations

2.1 $[\text{K}(\text{dme})]_2[(\text{Cp}'''\text{Co})_2(\mu, \eta^3:\eta^3\text{-P}_8)]$ (**2a**)

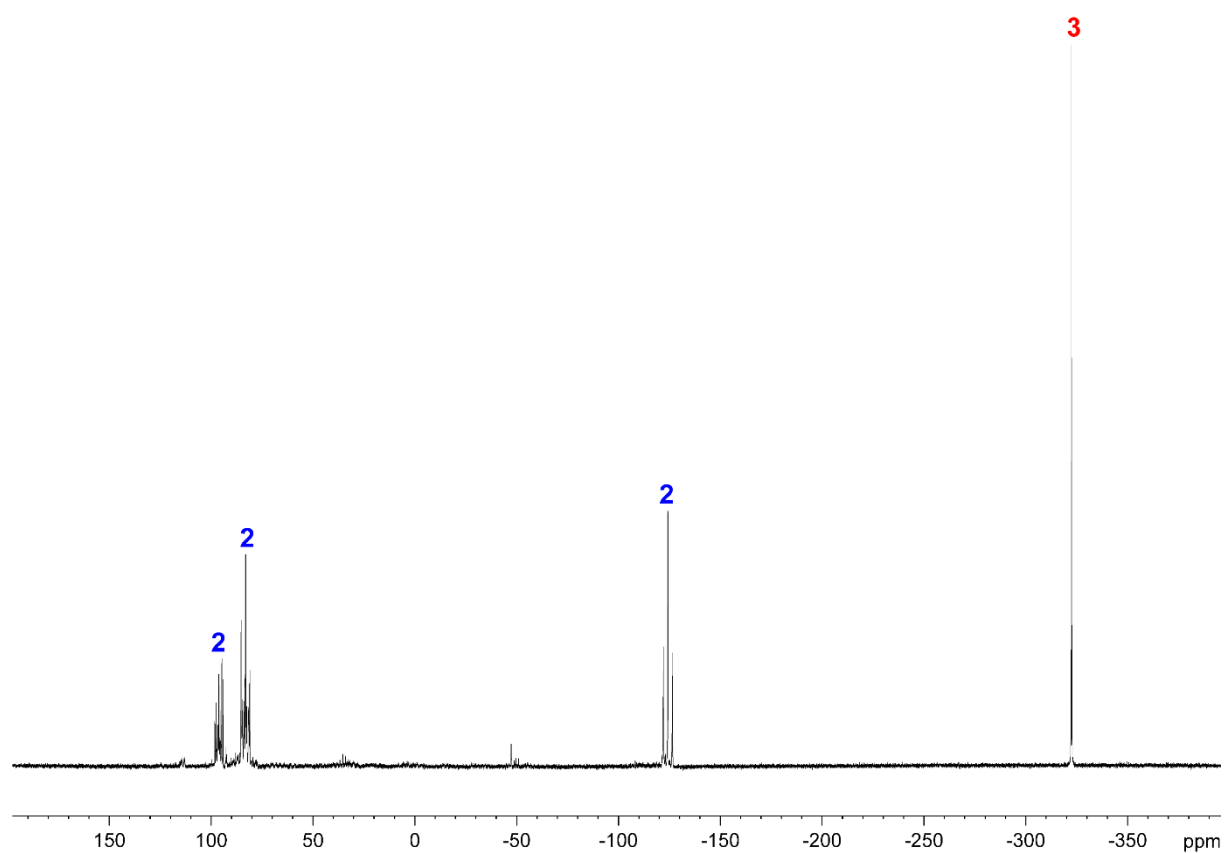


Figure S1: $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the crude reaction mixture from the reaction of **1** with KC_8 in thf-d_8 at room temperature.

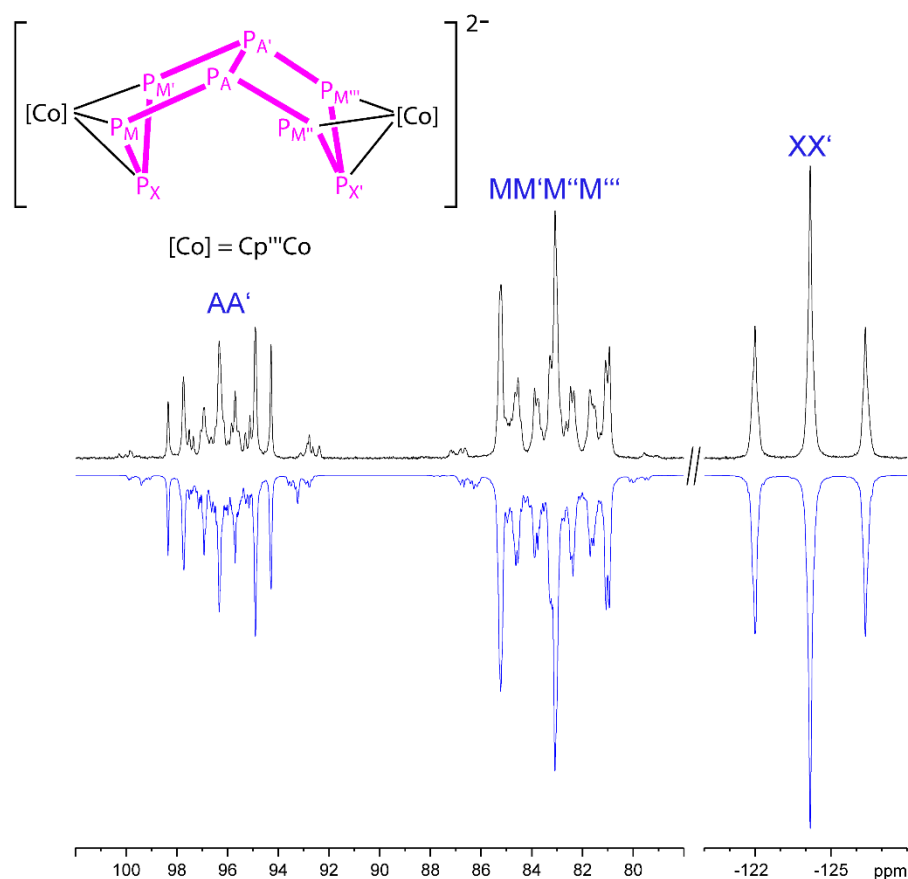


Figure S2: $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of **2a** in thf-d_8 at room temperature (experimental (black) and simulated (blue)).

Table S1: Coupling constants and chemical shifts obtained from the simulation.

coupling constants [Hz]			
$^1J_{P_A P_{A'}}$	309.44	$^2J_{P_{A'} P_{M''}}$	23.96
$^1J_{P_A P_M}$	330.75	$^2J_{P_{A'} P_X}$	-4.96
$^1J_{P_A P_{M''}}$	301.97	$^2J_{P_{A'} P_{X'}}$	4.28
$^1J_{P_{A'} P_{M'}}$	292.35	$^2J_{P_M P_{M'}}$	-57.81
$^1J_{P_{A'} P_{M'''}}$	343.58	$^2J_{P_M P_{M''}}$	-20.81
$^1J_{P_M P_X}$	351.21	$^2J_{P_{M'} P_{M'''}}$	-11.67
$^1J_{P_{M'} P_X}$	359.89	$^2J_{P_{M''} P_{M'''}}$	44.81
$^1J_{P_{M''} P_{X'}}$	355.82	$^3J_{P_M P_{M'''}}$	-11.83
$^1J_{P_{M'''}} P_{X'}$	362.15	$^3J_{P_M P_{X'}}$	8.65
$^2J_{P_A P_{M'}}$	21.49	$^3J_{P_{M'} P_{M''}}$	19.83
$^2J_{P_A P_{M'''}}$	-10.88	$^3J_{P_{M'} P_{X'}}$	-5.40
$^2J_{P_A P_X}$	9.55	$^3J_{P_{M''} P_X}$	-1.67
$^2J_{P_A P_{X'}}$	-0.66	$^3J_{P_{M'''}} P_X$	-15.65
$^2J_{P_{A'} P_M}$	11.96	$^4J_{P_X P_{X'}}$	21.50
chemical shifts [ppm]			
$P_{AA'}$	96.16	R value	1.56 %
$P_{MM'M''M'''}$	83.32		
$P_{XX'}$	-124.17		

2.2 [K(2,2,2-crypt)]₂[(Cp''Co)₂(μ,η³:η³-P₈)] (**2b**)

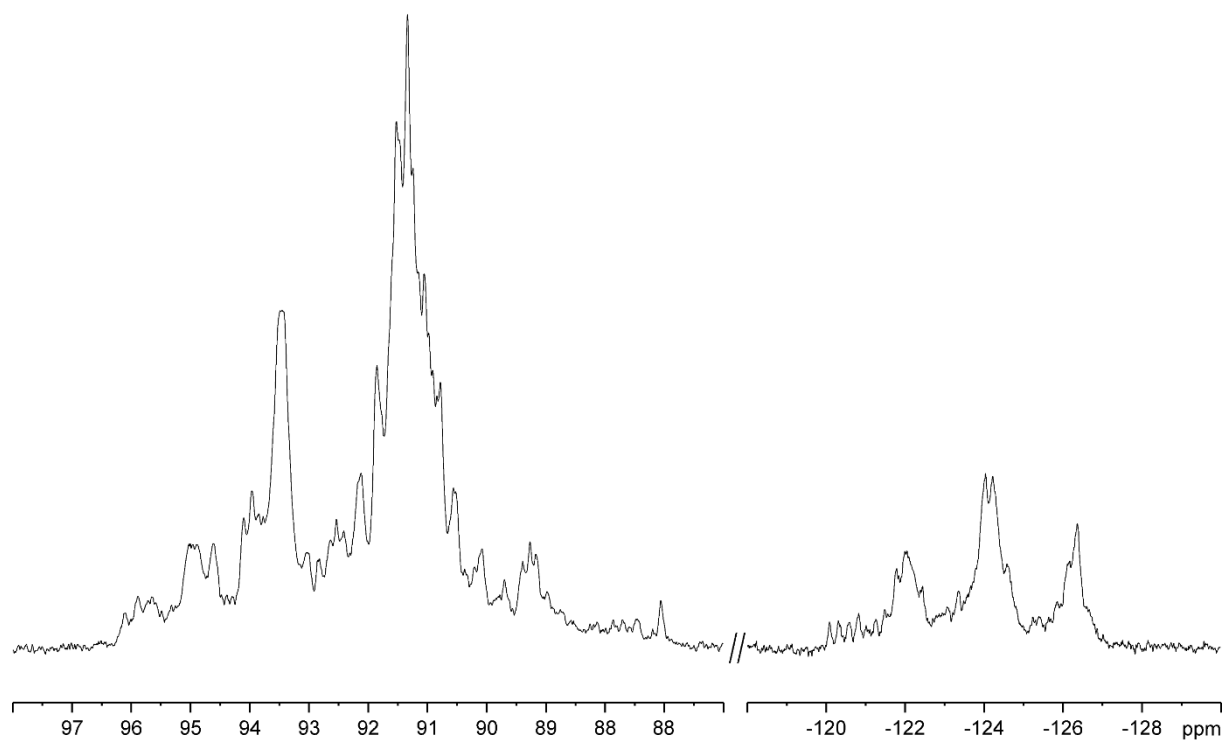


Figure S3: ³¹P{¹H} NMR spectrum of **2b** in CD₃CN at room temperature.

2.3 [Li(2,2,2-crypt)][Cp'''Co(η^3 -P₄^tBu)] (4)

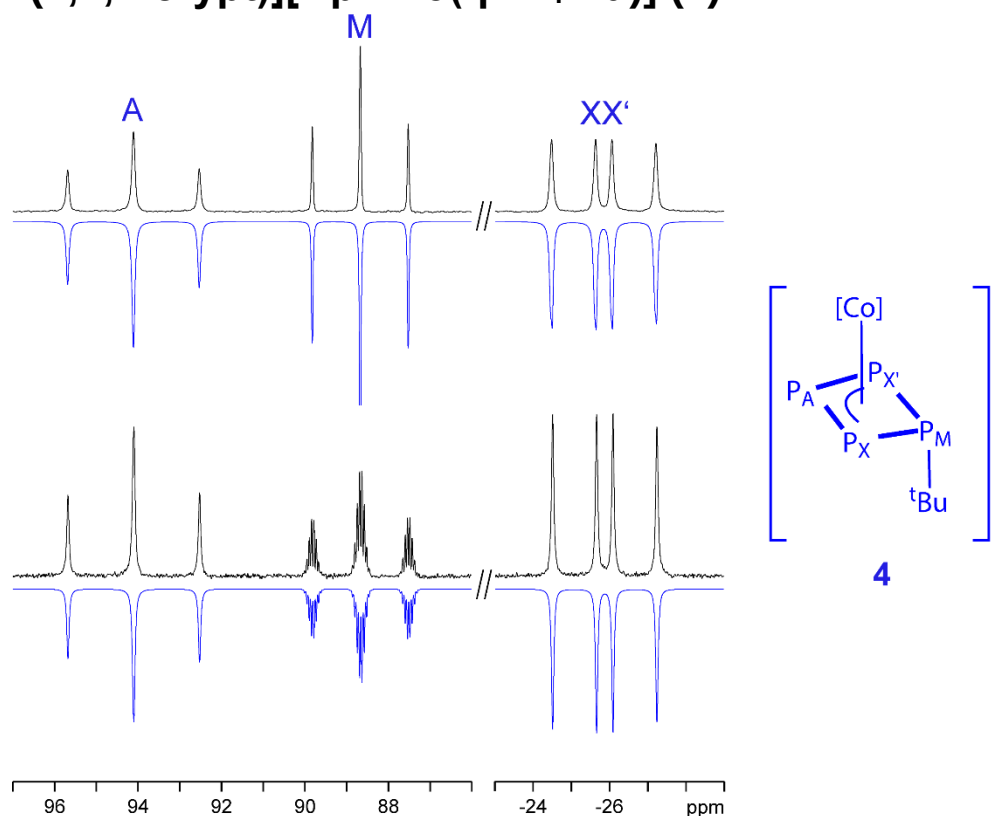


Figure S4: $^{31}\text{P}\{^1\text{H}\}$ (top) and ^{31}P (bottom) NMR spectrum of **4** thf with C₆D₆ capillary at room temperature (experimental (black) and simulated (blue)).

Table S2: Coupling constants and chemical shifts obtained from the simulation of the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum.

coupling constants [Hz]		chemical shifts [ppm]	
$^1J_{P_AP_X} = ^1J_{P_AP_{X'}}$	255.63	P _A	94.13
$^1J_{P_MP_X} = ^1J_{P_MP_{X'}}$	186.23	P _M	88.64
$^2J_{P_AP_M}$	2.48	P _X	-25.92
$^2J_{P_XP_{X'}}$	3.97	P _{X'}	-25.92
R value		0.84 %	

Table S3: Coupling constants and chemical shifts obtained from the simulation of the ^{31}P NMR spectrum.

coupling constants [Hz]		chemical shifts [ppm]	
$^1J_{P_AP_X} = ^1J_{P_AP_{X'}}$	255.60	P _A	94.11
$^1J_{P_MP_X} = ^1J_{P_MP_{X'}}$	186.26	P _M	88.62
$^2J_{P_AP_M}$	1.49	P _X	-25.95
$^2J_{P_XP_{X'}}$	-0.30	P _{X'}	-25.95
$^3J_{P_MH}$	9.12		
$^4J_{P_XH} = ^4J_{P_{X'}H}$	-1.12		
$^5J_{P_AH}$	0.77		
R value		0.24%	

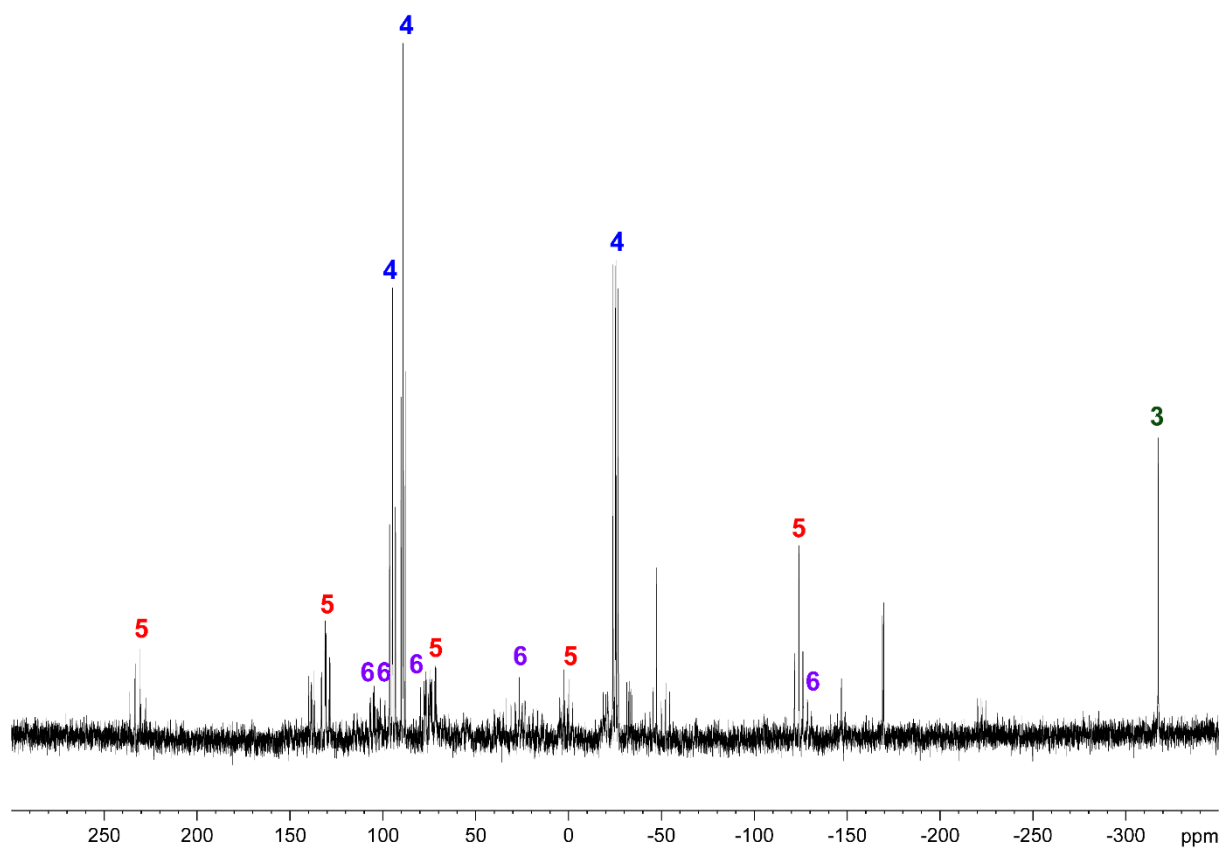


Figure S5: $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the reaction of **1** with $^t\text{BuLi}$ at room temperature thf-d_8 at room temperature (large sweep).

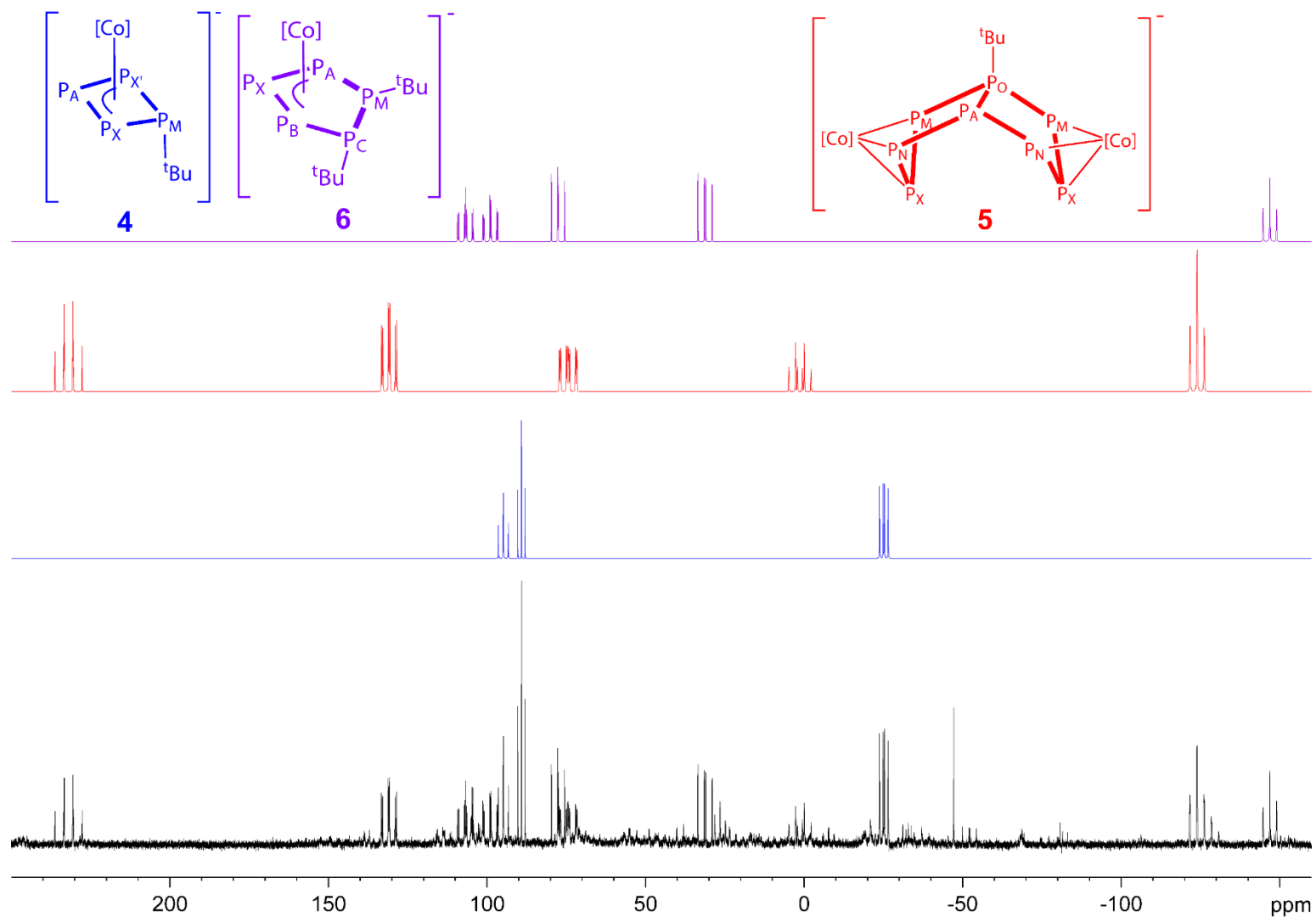


Figure S6: Part of the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the reaction of **1** with $t\text{BuLi}$ at room temperature thf-d_8 at room temperature and the simulated spectra of **4** (blue), **5** (red) and **6** (purple).

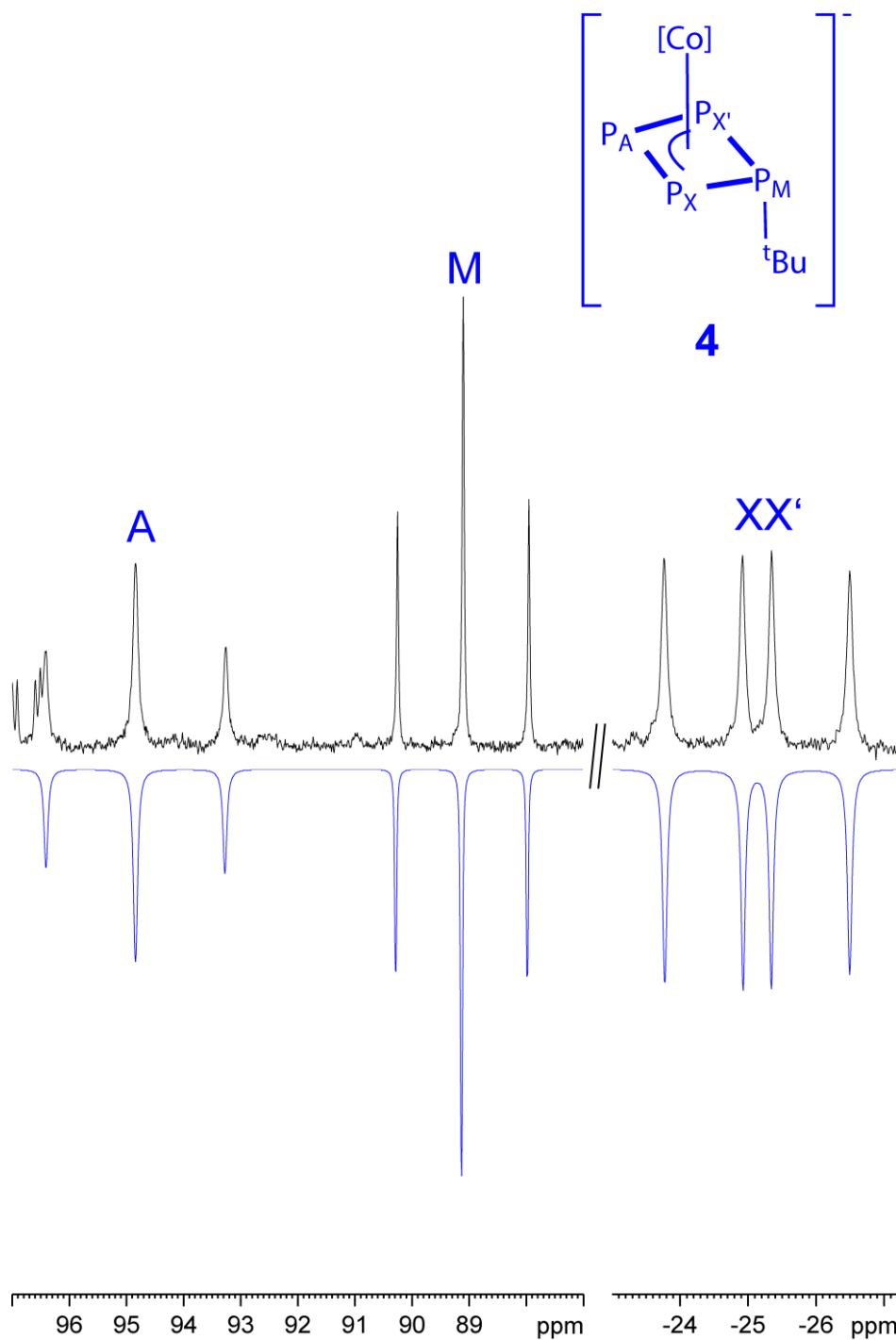


Figure S7: Part of the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the reaction of **1** with $^t\text{BuLi}$ at room temperature thf-d_8 at room temperature and the simulated spectrum of **4** (blue).

Table S4: Coupling constants and chemical shifts obtained from the simulation of **4** from the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum.

coupling constants [Hz]		chemical shifts [ppm]	
$^1J_{P_A P_X} = ^1J_{P_A P_{X'}}$	253.71	P _A	94.83
$^1J_{P_M P_X} = ^1J_{P_M P_{X'}}$	186.61	P _M	89.13
$^2J_{P_A P_M}$	2.99	P _X	-25.13
$^2J_{P_X P_{X'}}$	4.01	P _{X'}	-25.13

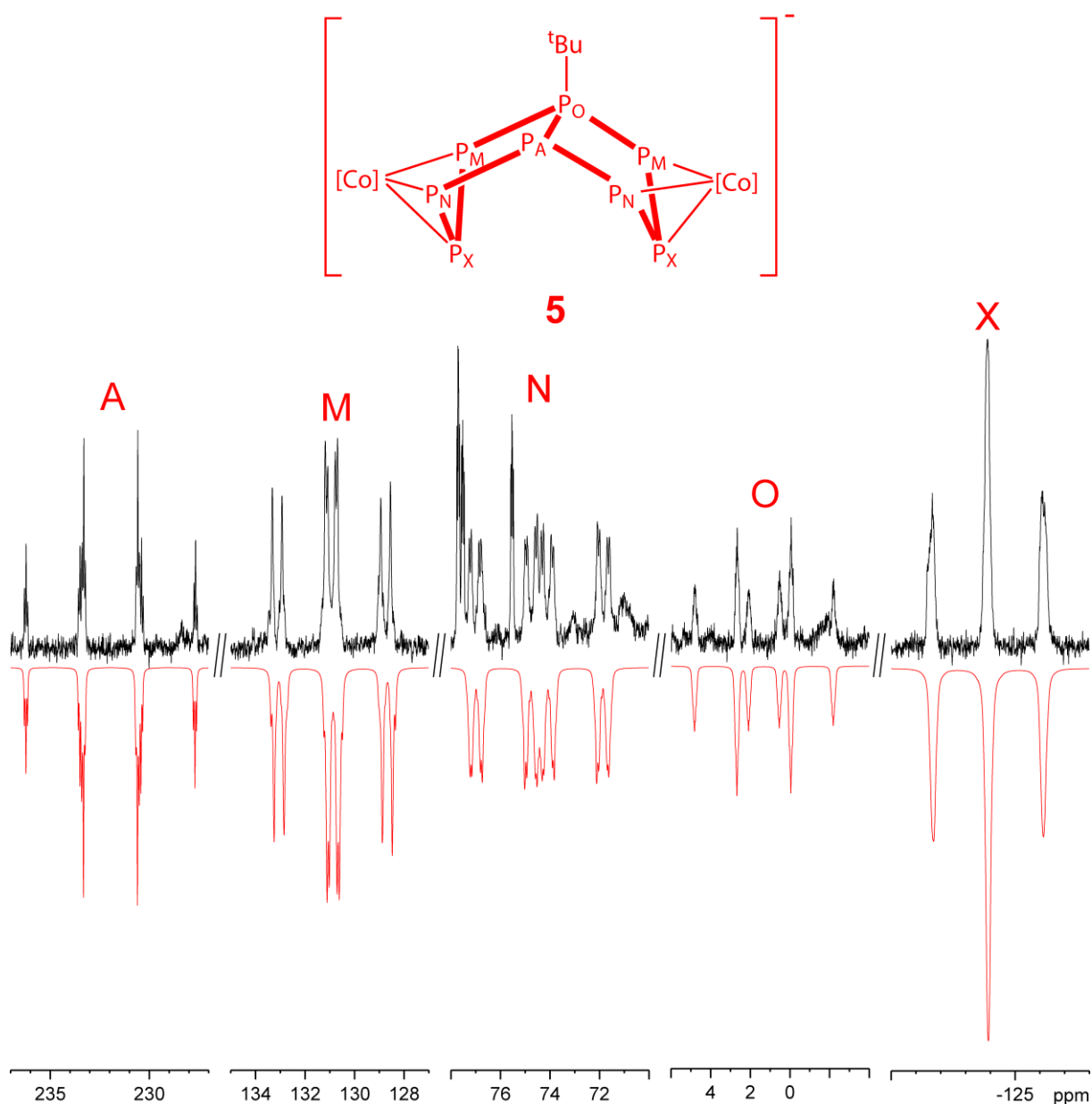


Figure S8: Part of the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the reaction of **1** with tBuLi at room temperature thf-d_8 at room temperature and the simulated spectrum of **5** (red).

Table S5: Coupling constants and chemical shifts obtained from the simulation of **5** from the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum.

coupling constants [Hz]			
$^1J_{\text{P}_\text{A}\text{P}_\text{N}}$	470.35	$^2J_{\text{P}_\text{M}\text{P}_\text{N}}$	-4.69
$^1J_{\text{P}_\text{A}\text{P}_\text{O}}$	441.32	$^2J_{\text{P}_\text{M}\text{P}_\text{N}}$	69.86
$^1J_{\text{P}_\text{M}\text{P}_\text{O}}$	346.33	$^2J_{\text{P}_\text{N}\text{P}_\text{N}}$	-16.59
$^1J_{\text{P}_\text{M}\text{P}_\text{X}}$	363.12	$^2J_{\text{P}_\text{N}\text{P}_\text{O}}$	-14.40
$^1J_{\text{P}_\text{N}\text{P}_\text{X}}$	358.96	$^2J_{\text{P}_\text{O}\text{P}_\text{X}}$	6.80
$^2J_{\text{P}_\text{A}\text{P}_\text{M}}$	-1.21	$^3J_{\text{P}_\text{M}\text{P}_\text{X}}$	-1.52
$^2J_{\text{P}_\text{A}\text{P}_\text{X}}$	13.24	$^3J_{\text{P}_\text{N}\text{P}_\text{X}}$	1.02
$^2J_{\text{P}_\text{M}\text{P}_\text{M}}$	14.02	$^4J_{\text{P}_\text{X}\text{P}_\text{X}}$	20.96

chemical shifts [ppm]	
P _A	231.93
P _M	130.85
P _N	74.73
P _O	1.34
P _X	-123.91

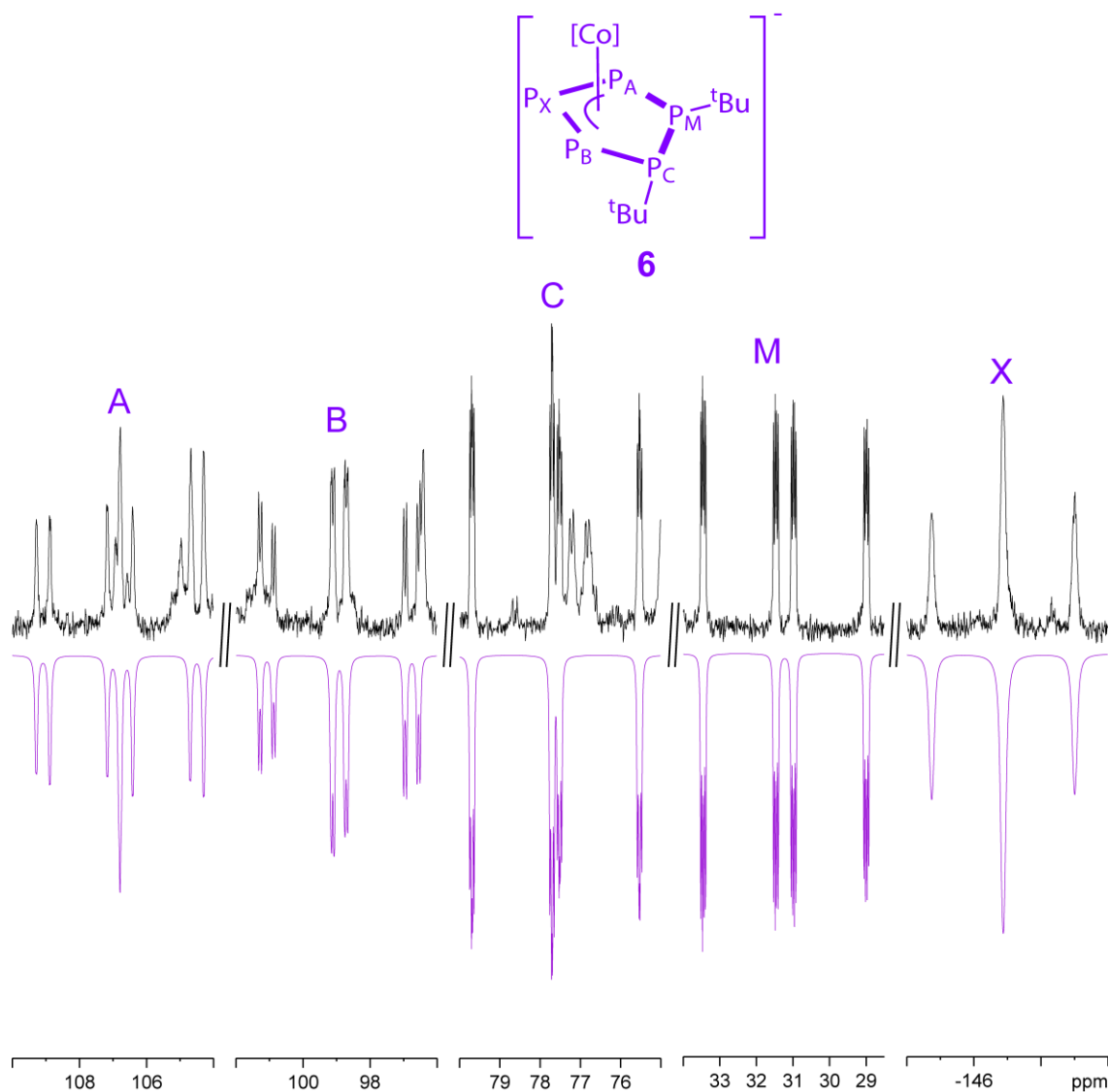


Figure S9: Part of the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the reaction of **1** with $^t\text{BuLi}$ at room temperature thf-d_8 at room temperature and the simulated spectrum of **6** (purple).

Table S6: Coupling constants and chemical shifts obtained from the simulation of **6** from the $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum.

coupling constants [Hz]			
$^1J_{P_A P_M}$	400.71	$^2J_{P_A P_B}$	65.46
$^1J_{P_A P_X}$	342.90	$^2J_{P_A P_C}$	5.97
$^1J_{P_B P_C}$	352.77	$^2J_{P_B P_M}$	12.84
$^1J_{P_B P_X}$	348.39	$^2J_{P_C P_X}$	10.00
$^1J_{P_C P_M}$	322.67	$^2J_{P_M P_X}$	7.23
chemical shifts [ppm]			
P_A	106.76		
P_B	98.86		
P_C	77.65		
P_M	31.27		
P_X	-146.87		

2.4 [Li(2,2,2-crypt)][Cp'''Co(η^3 -P₄CH₂SiMe₃)] (7)

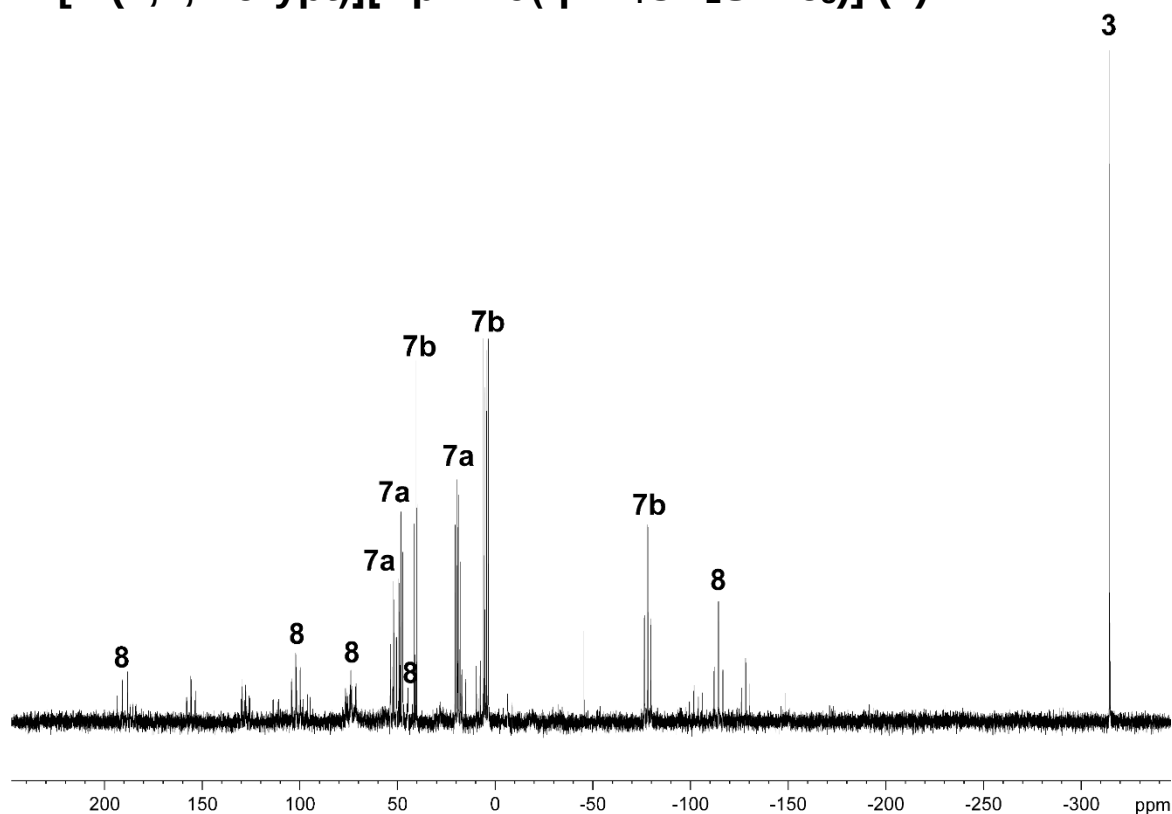


Figure S10: $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum in thf- d_8 of the reaction of $\text{LiCH}_2\text{SiMe}_3$ with **1** after addition of 12-c-4.

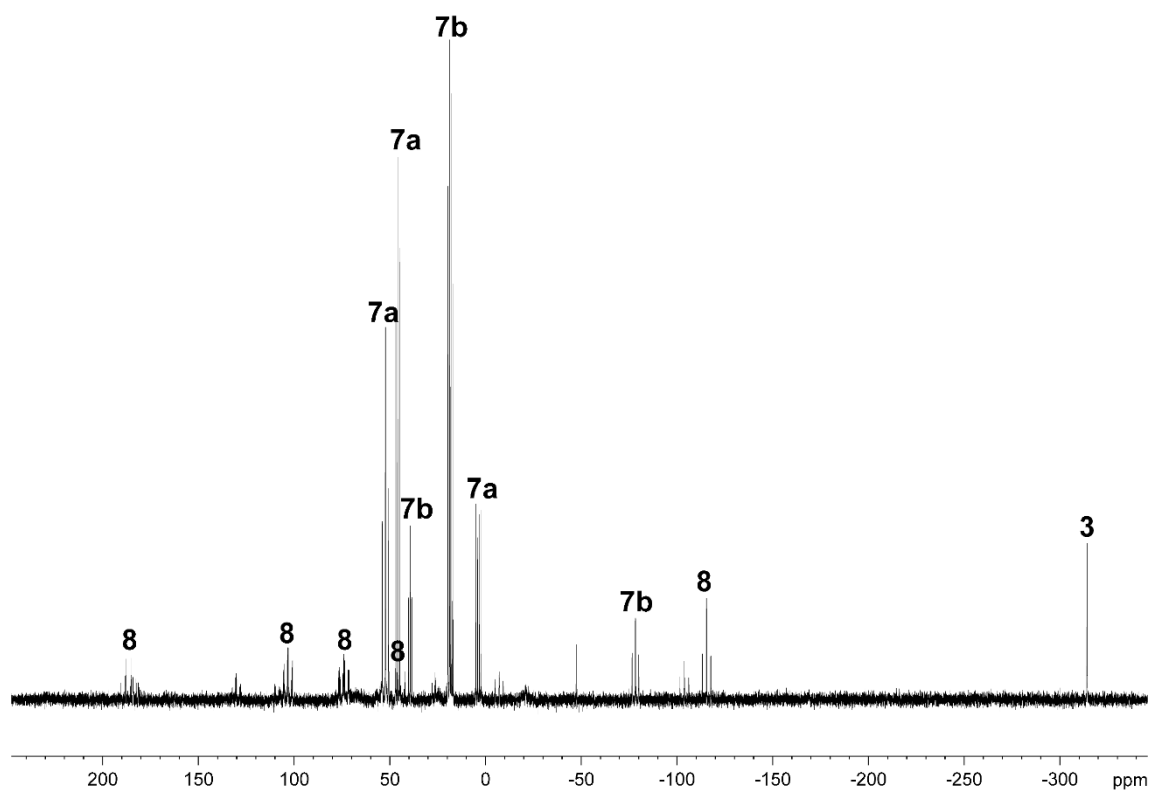


Figure S11: $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum in thf- d_8 of the reaction of $\text{LiCH}_2\text{SiMe}_3$ with **1** after addition of 2,2,2-cryptand.

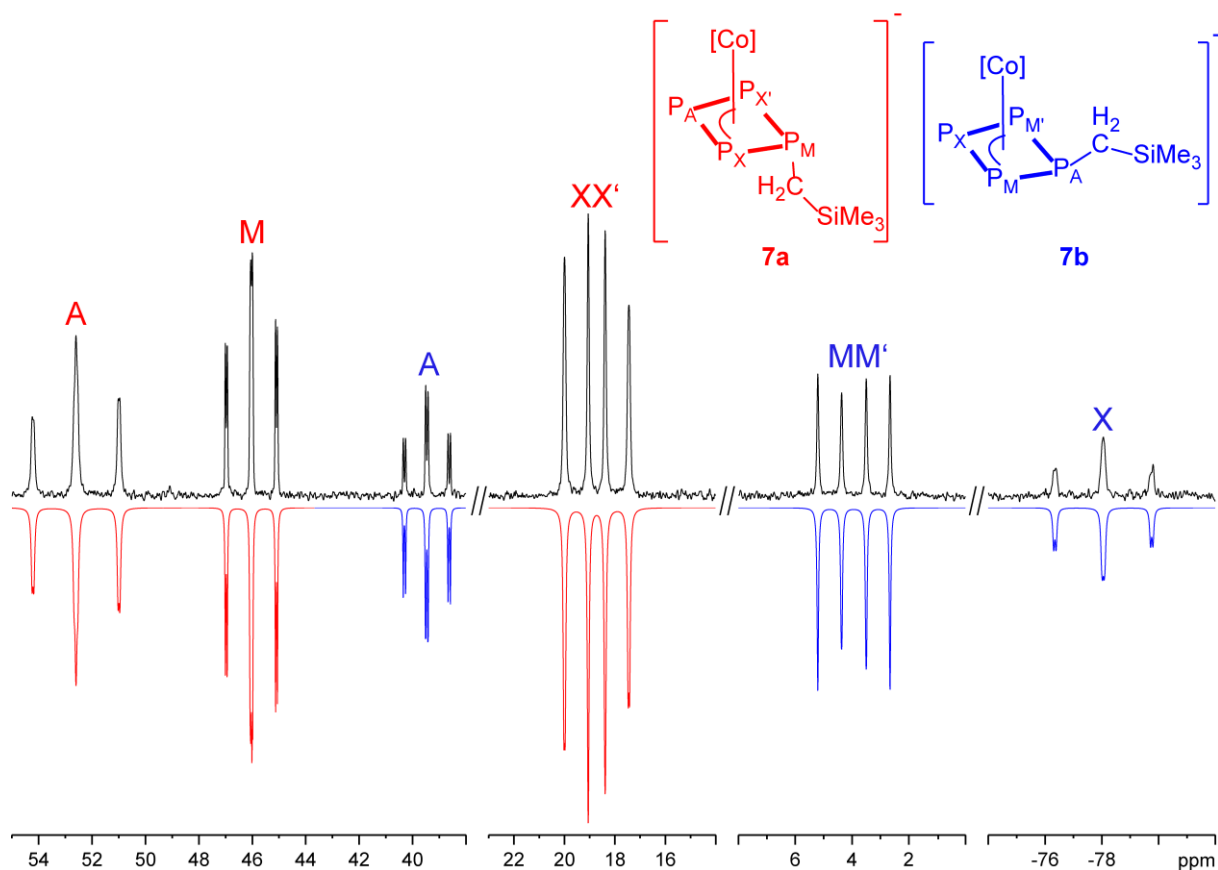


Figure S12: $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the $[\text{Li}(2,2,2\text{-crypt})]\mathbf{7a/7b}$ in thf-d_8 (experimental (black) and simulated (blue, red)).

Table S7: Coupling constants and chemical shifts obtained from the simulation ($[\text{Li}(2,2,2\text{-crypt})]\mathbf{7a/7b}$).

coupling constants [Hz]			
7a		7b	
$^1J_{P_A P_X} = ^1J_{P_A P_{X'}}$	260.89	$^1J_{P_A P_M} = ^1J_{P_A P_{M'}}$	135.82
$^1J_{P_M P_X} = ^1J_{P_M P_{X'}}$	152.24	$^1J_{P_M P_X} = ^1J_{P_{M'} P_X}$	276.36
$^2J_{P_A P_M}$	10.95	$^2J_{P_A P_X}$	14.38
$^2J_{P_X P_{X'}}$	-8.00	$^2J_{P_M P_{M'}}$	-84.01
chemical shifts [ppm]		chemical shifts [ppm]	
P_A	52.29	P_A	39.39
P_M	45.93	P_M	3.78
P_X	18.48	$P_{M'}$	3.78
$P_{X'}$	18.48	P_X	-78.24
R value	0.53 %		

Table S8: Coupling constants and chemical shifts obtained from the simulation ($[\text{Li}(12\text{-c-4})_2]\mathbf{7a/7b}$).

coupling constants [Hz]			
7a		7b	
$^1J_{P_A P_X} = ^1J_{P_A P_{X'}}$	259.18	$^1J_{P_A P_M} = ^1J_{P_A P_{M'}}$	135.34
$^1J_{P_M P_X} = ^1J_{P_M P_{X'}}$	152.95	$^1J_{P_M P_X} = ^1J_{P_{M'} P_X}$	273.40

$$\begin{array}{l|l} {}^2J_{P_AP_M} & 10.53 \\ {}^2J_{P_XP_{X'}} & -8.00 \end{array}$$

$$\begin{array}{l|l} {}^2J_{P_AP_X} & 13.76 \\ {}^2J_{P_MP_{M'}} & -84.00 \end{array}$$

chemical shifts [ppm]		chemical shifts [ppm]	
P _A	50.07	P _A	38.93
P _M	46.54	P _M	2.84
P _X	17.42	P _{M'}	2.84
P _{X'}	17.42	P _X	-79.93
R value	1.69 %		

2.5 [Li(dme)₃][(Cp'''Co)₂(μ,η³:η³-P₈CH₂SiMe₃)] (8)

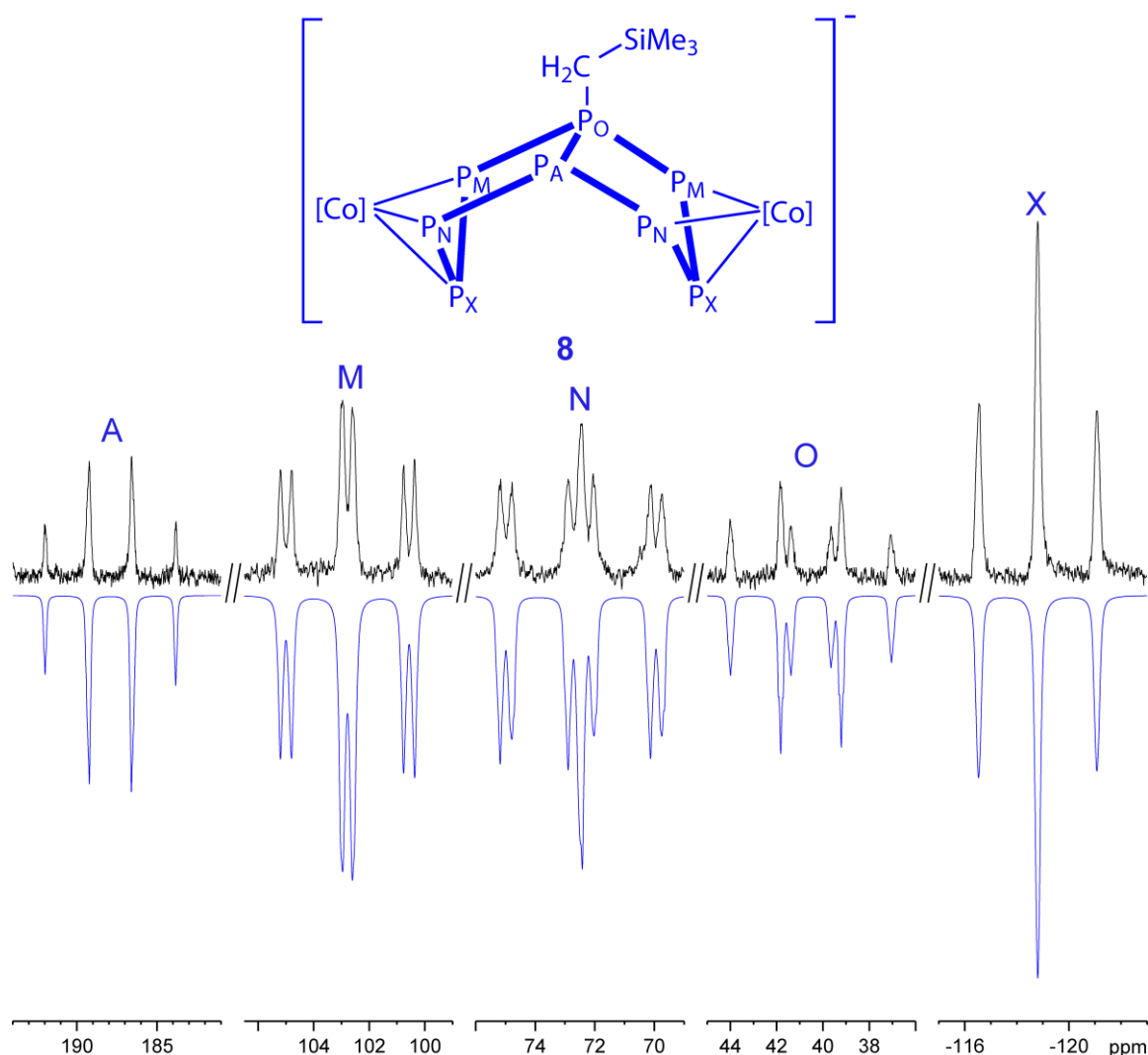


Figure S13: ³¹P{¹H} NMR spectrum of **8** in thf-d₈ at 193 K (experimental (black) and simulated (blue)).

Table S9: Coupling constants and chemical shifts obtained from the simulation.

coupling constants [Hz]			
¹ J _{P_AP_N}	448.32	² J _{P_MP_N}	62.41
¹ J _{P_AP_O}	424.01	² J _{P_MP_N}	-12.85
¹ J _{P_MP_O}	351.60	² J _{P_NP_N}	-3.66
¹ J _{P_MP_X}	366.96	² J _{P_NP_O}	-17.34
¹ J _{P_NP_X}	369.36	² J _{P_OP_X}	7.69
² J _{P_AP_M}	-2.44	³ J _{P_MP_X}	0.20
² J _{P_AP_X}	12.49	³ J _{P_NP_X}	0.87
² J _{P_MP_M}	6.65	⁴ J _{P_XP_X}	19.47
chemical shifts [ppm]			
P _A	187.87	R value	1.53 %
P _M	102.75		
P _N	74.48		
P _O	40.57		
P _X	-118.80		

2.6 **1** + LiCH₂SiMe₃ VT-NMR

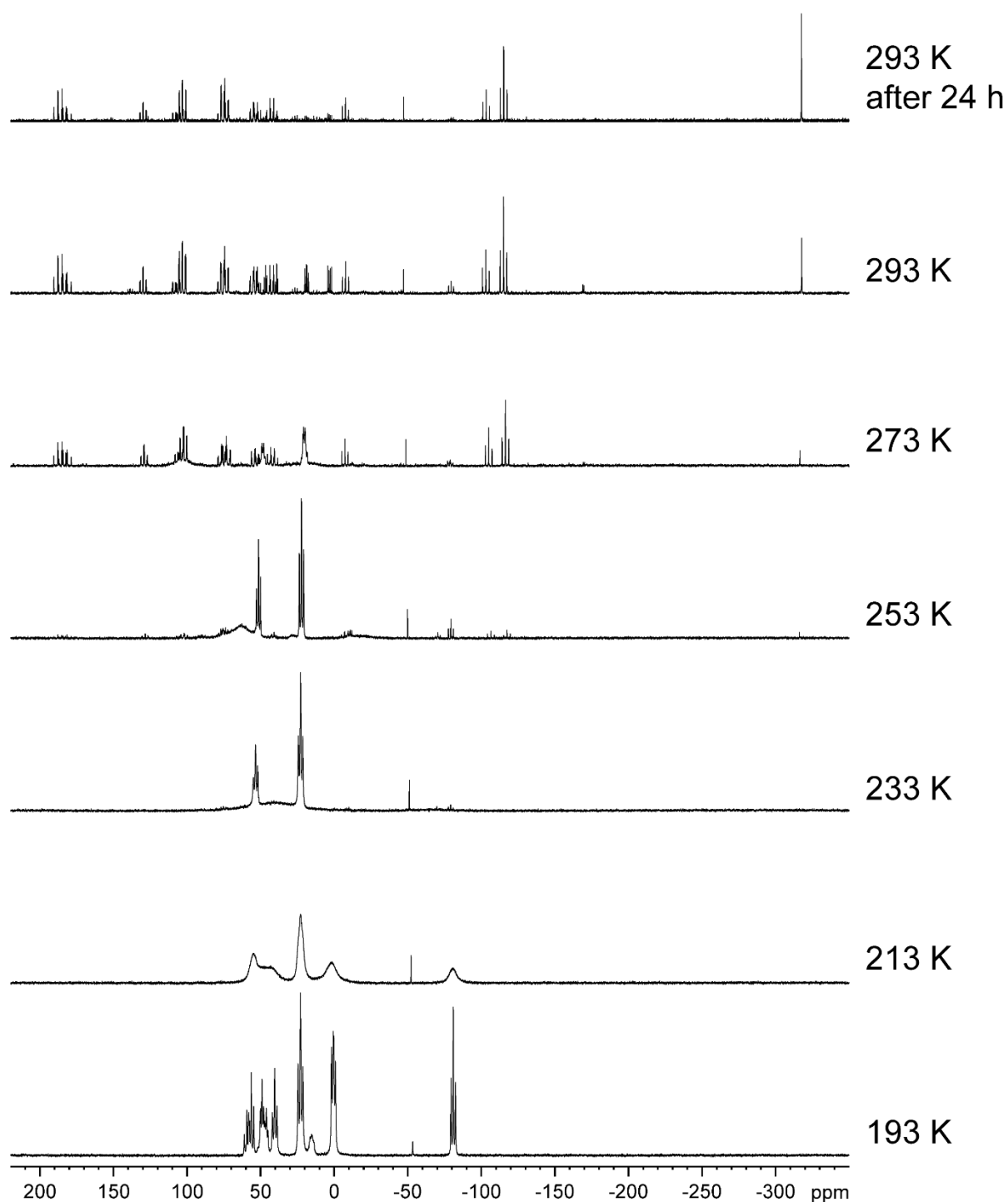


Figure S14: ³¹P{¹H} VT NMR spectra of the reaction of **1** with LiCH₂SiMe₃ in thf-d₈ between 193 and 293 K and after 24 h at room temperature.

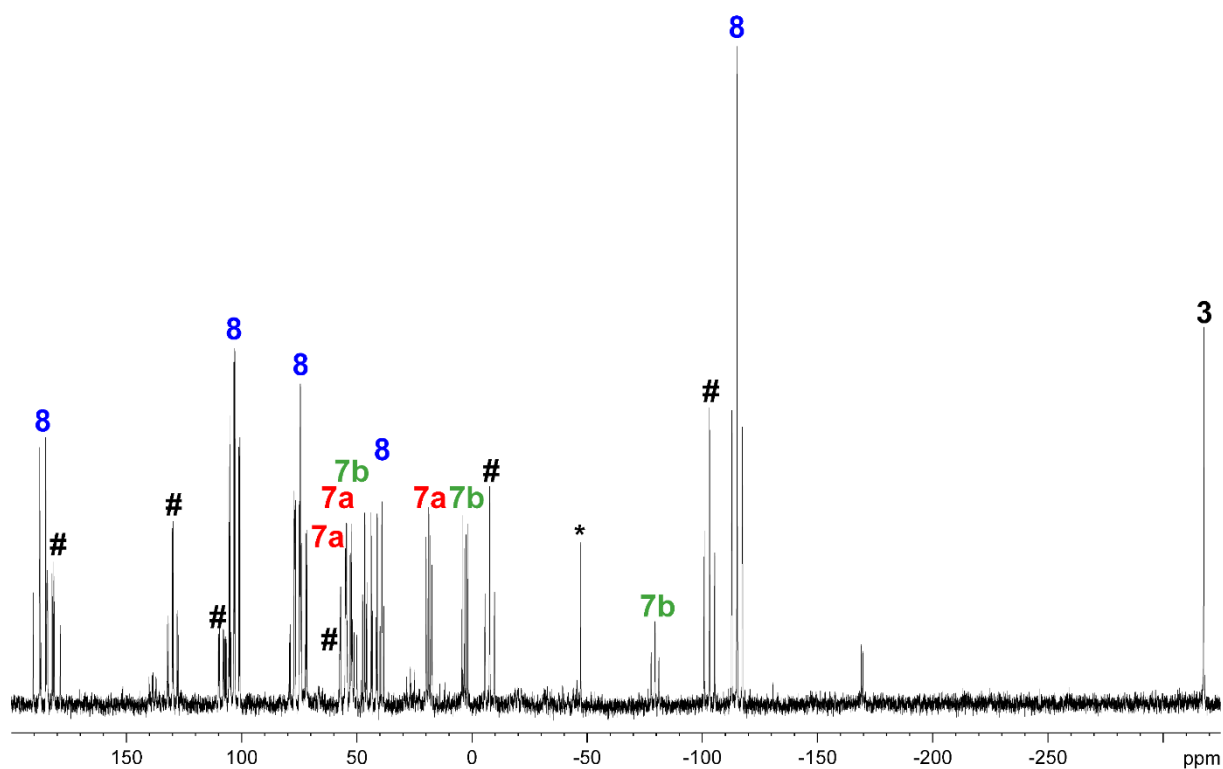


Figure S15: $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the VT-NMR (S14) at 293 K (# unknown compound, * $[(\text{Cp}'''\text{Co})_2(\text{P}_2)_2]$).

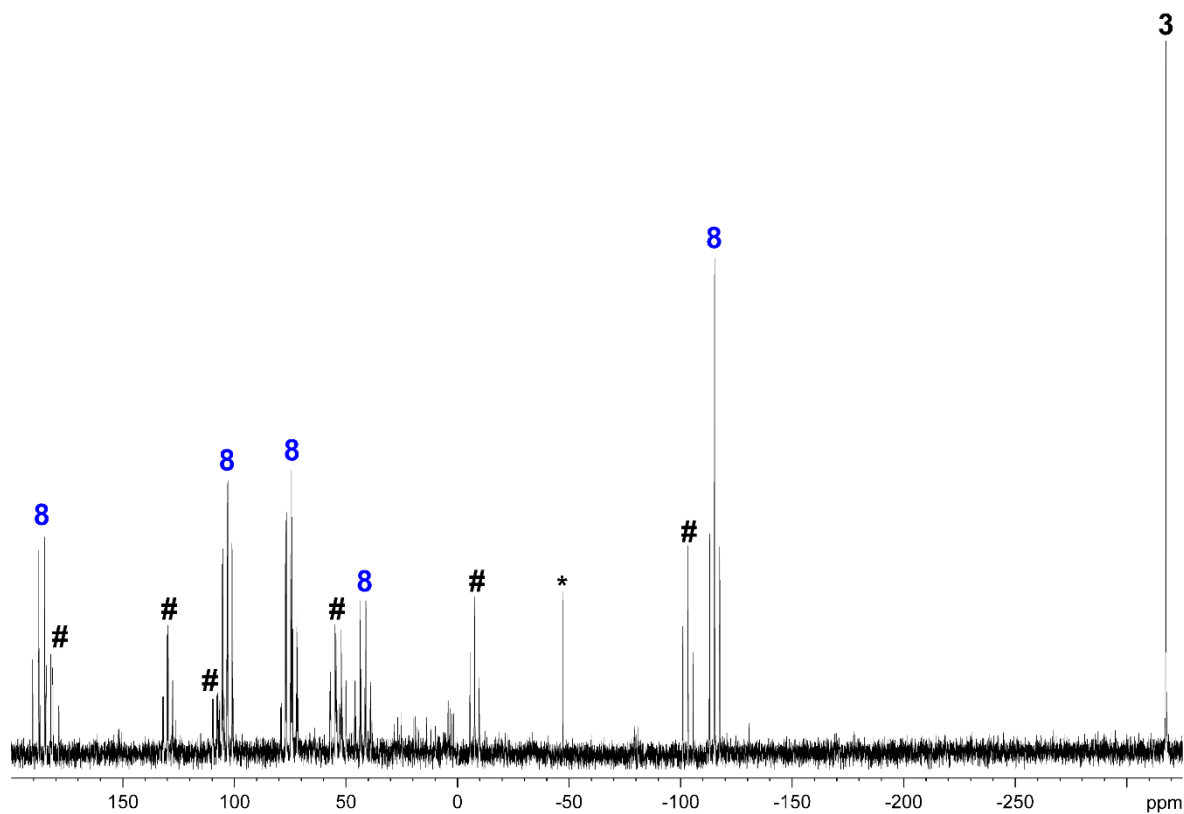


Figure S16: $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of the VT-NMR (S14) at 293 K after 24 h (# unknown compound, * $[(\text{Cp}'''\text{Co})_2(\text{P}_2)_2]$).

2.7 [Na(dme)₂][Cp'''Co(η³-P₄(O)H)] (10)

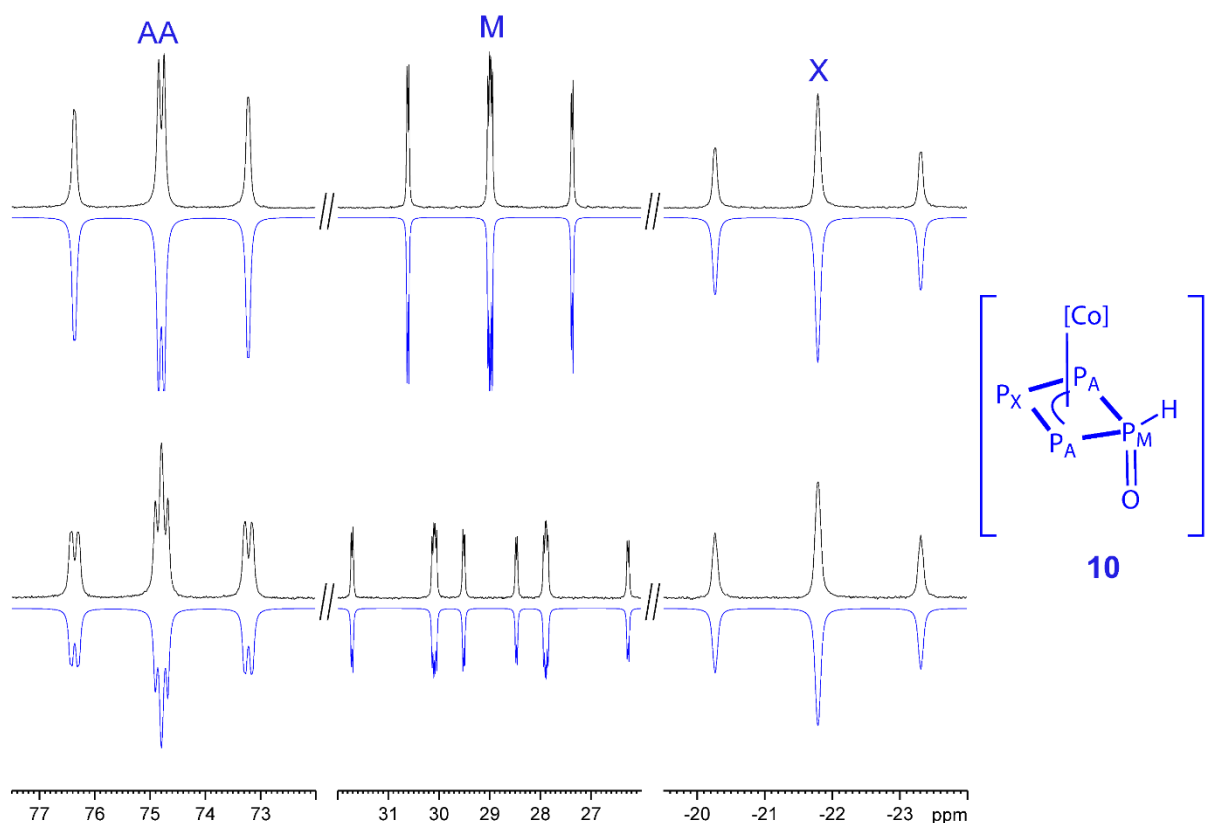


Figure S17: ³¹P{¹H} (top) and ³¹P (bottom) NMR spectrum of **10** thf-d₈ at room temperature (experimental (black) and simulated (blue)).

Table S10: Coupling constants and chemical shifts obtained from the simulation of the ³¹P{¹H} NMR spectrum.

coupling constants [Hz]		chemical shifts [ppm]	
¹ J _{P_AP_M}	262.67	P _A	74.78
¹ J _{P_AP_X}	246.57	P _M	29.02
² J _{P_AP_A}	-9.00	P _X	-21.77
² J _{P_MP_X}	4.92		
R value		1.23 %	

Table S11: Coupling constants and chemical shifts obtained from the simulation of the ³¹P NMR spectrum.

coupling constants [Hz]		chemical shifts [ppm]	
¹ J _{P_AP_M}	262.71	P _A	74.77
¹ J _{P_AP_X}	246.64	P _M	29.02
² J _{P_AP_A}	0.00	P _X	-21.78

$^2J_{PM^P_X}$	4.98	
$^1J_{PM^H}$	356.77	
$^2J_{PA^H}$	20.56	
$^3J_{PX^H}$	5.12	
R value		0.41 %

3. Cyclic voltammogram of **1**

The measurement was conducted in 5 ml dme with 500 mg conducting salt [n Bu₄N][PF₆] and 0.01 mmol of **1** were used. A platinum working electrode, a platinum counter electrode and a silver wire as a reference electrode were used. Ferrocene was added as an internal standard for referencing. The weak oxidation process at +275 mV against [Cp₂Fe]/[Cp₂Fe]⁺ is a feature of **1** and could not be detected within the blank measurement of the pure solvent. A measurement in CH₂Cl₂ (5 ml, 750 mg conducting salt [n Bu₄N][PF₆], 0.01 mmol **1**, cf. Figure S18_2) reveals the oxidation process more clearly separated from the solvent window, while the reduction process can be detected in the shoulder of the solvent window.

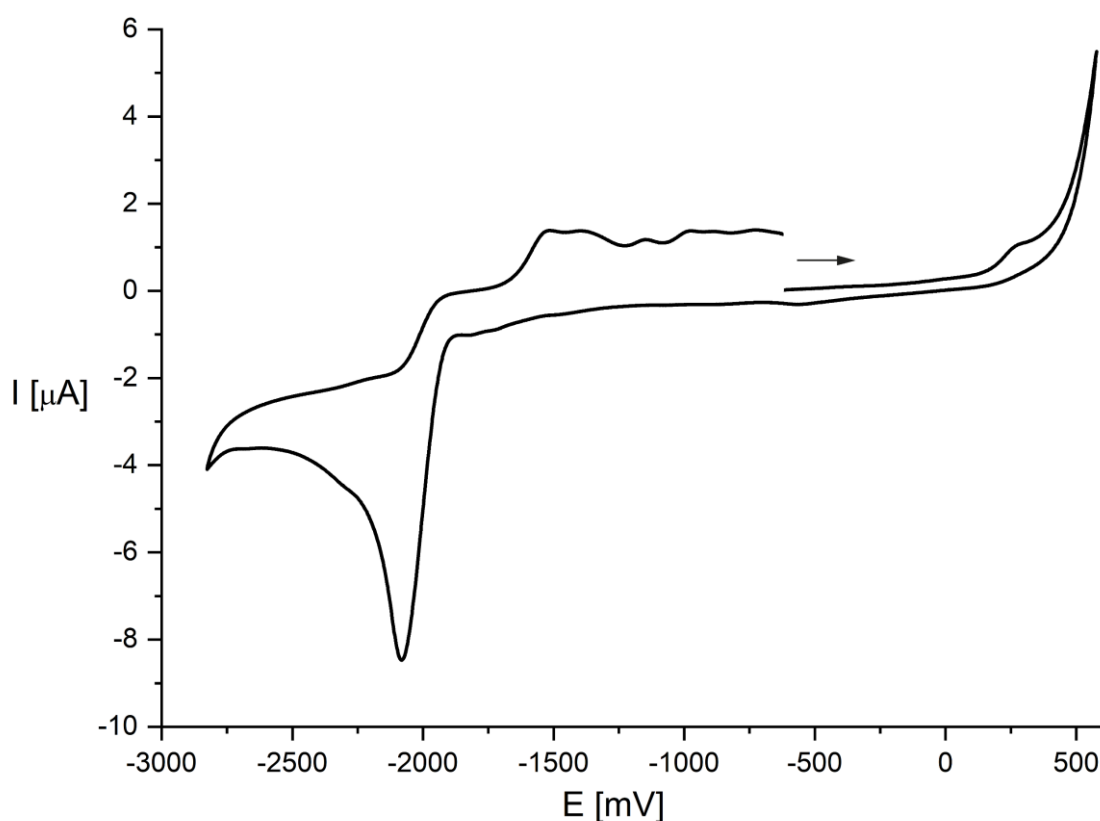


Figure S18_1. Cyclic voltammogram of **1** in dme against [Cp₂Fe]/[Cp₂Fe]⁺ (electrolyte n Bu₄NPF₆, scan rate: 100 mV/s. temperature: r.t.).

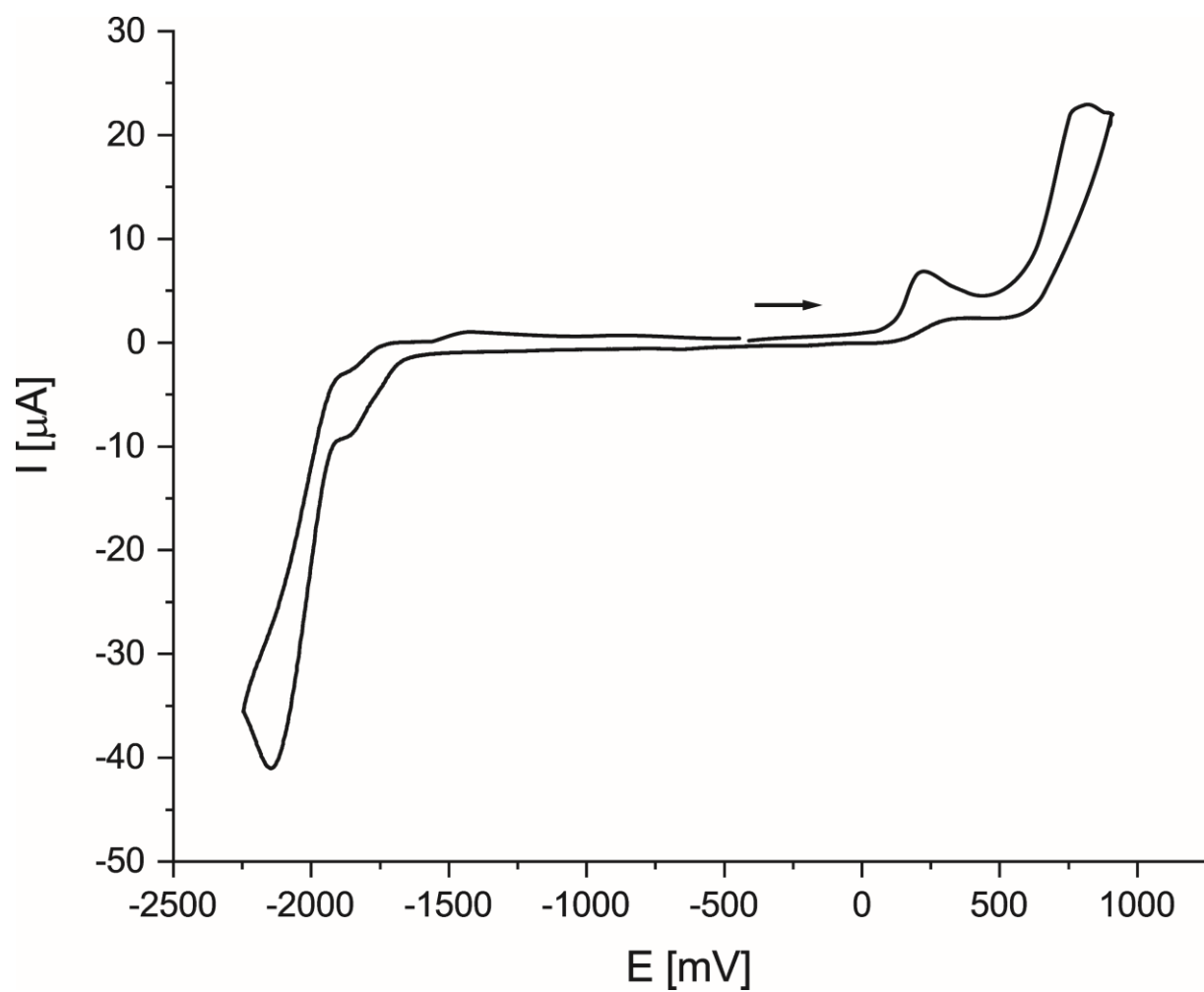


Figure S18_2. Cyclic voltammogram of **1** in CH₂Cl₂ against [Cp₂Fe]/[Cp₂Fe]⁺ (electrolyte ⁿBu₄NPF₆, scan rate: 100 mV/s. temperature: r.t.).

4. Details on single crystal X-ray structure analysis

The X-ray diffraction experiments were performed on either an Gemini Ultra diffractometer (Oxford diffraction) with an AtlasS2 detector applying Mo radiation ($\lambda = 0.71073 \text{ \AA}$) (**2a**, **5**, **7a**) or Cu-K α radiation ($\lambda = 1.54178 \text{ \AA}$) (**2b**, **4**), a GV 50 diffractometer (Rigaku, formerly Agilent Technologies) with TitanS2 detector from applying Cu-K α radiation ($\lambda = 1.54178 \text{ \AA}$) (**6**, **9**) or SuperNova (Agilent Technologies) with an Atlas detector applying Cu-K α radiation ($\lambda = 1.54178 \text{ \AA}$) (**8**). All measurements were performed at 123 K. An analytical numeric absorption correction^[4] using a multifaceted crystal model based on expressions derived by R.C. Clark & J.S. Reid using spherical harmonics as implemented in SACLE3 ABSPACK was applied (**2a**, **2b**, **4**, **5**, **7**).^[5] A numerical absorption correction based on gaussian integration over a multifaceted crystal model using *CrysAlisPro* using spherical harmonics as implemented in SACLE3 ABSPACK was applied (**8**, **9**). All structures were solved by direct methods with ShelXT^[6] and Olex2^[7] and refined by full-matrix least-squares method against F^2 in anisotropic approximation using ShelXL.^[8] Hydrogen atoms were refined in calculated positions using riding on pivot atom model.

CCDC-1984450 (**2b**), CCDC-1984451 (**4**), CCDC-1984452 (**5**), CCDC-1984453 (**7a**), CCDC-1984454 (**8**) and CCDC-1984455 (**10**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge at www.ccdc.cam.ac.uk/conts/retrieving.html (or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; Fax: + 44-1223-336-033; e-mail: deposit@ccdc.cam.ac.uk).

4.1 [K(dme)]₂[(Cp^{'''}Co)₂(μ,η³:η³-P₈)] (2a)

Crystals suitable for X-ray single crystal structure analysis can be obtained from a concentrated solution in a mixture of n-hexane and dme at -30 °C. Compound **2a** crystallizes in the orthorhombic space group *Cmce* as dark green blocks. Although a very good measurement was conducted (up to a resolution of 0.67 Å) a final refinement of the data was not possible. The anion can be modeled quite well, but the potassium counterions are located over several positions (partially site occupancies below 8 %) surrounded by dme molecules which cannot be modeled properly. The structure in solid state including the electron density map is depicted in Figure S19.

Current level: 0.6

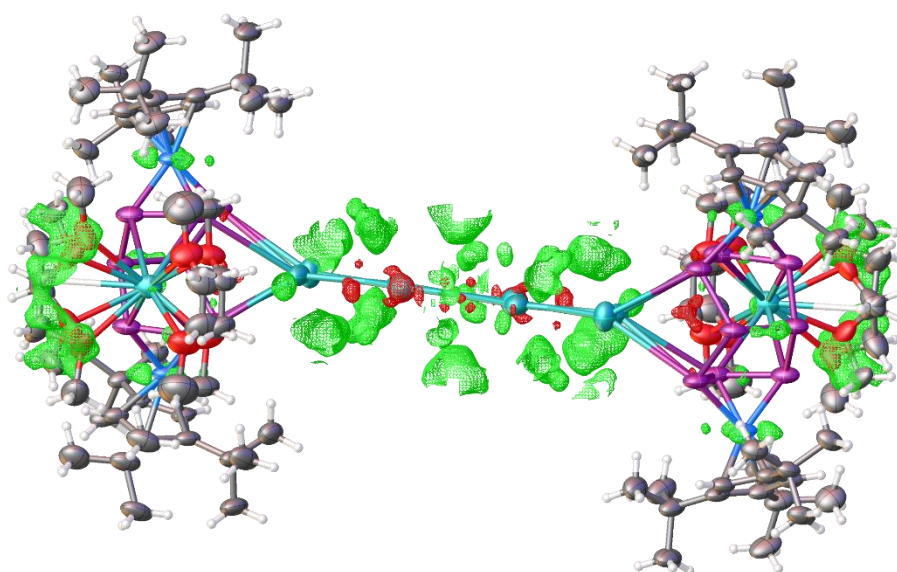


Figure S19. Part of the molecular structure of **2a** in solid state. Depicted is the unfinished refinement with the electron density map.

4.2 [K(2,2,2-crypt)]₂[(Cp^{'''}Co)₂(μ,η³:η³-P₈)] (2b)

Crystals suitable for X-ray single crystal structure analysis can be obtained from a concentrated solution of **2b** in a mixture of dme and MeCN at -30 °C. Compound **2b** crystallizes in form of dark green plates in the triclinic space group $P\bar{1}$. The asymmetric unit contains one dianion of **2b**, two potassium counterions chelated by 2,2,2-cryptand and 1.5 molecules dme. The structure in solid state is depicted in Figure S20 and S21.

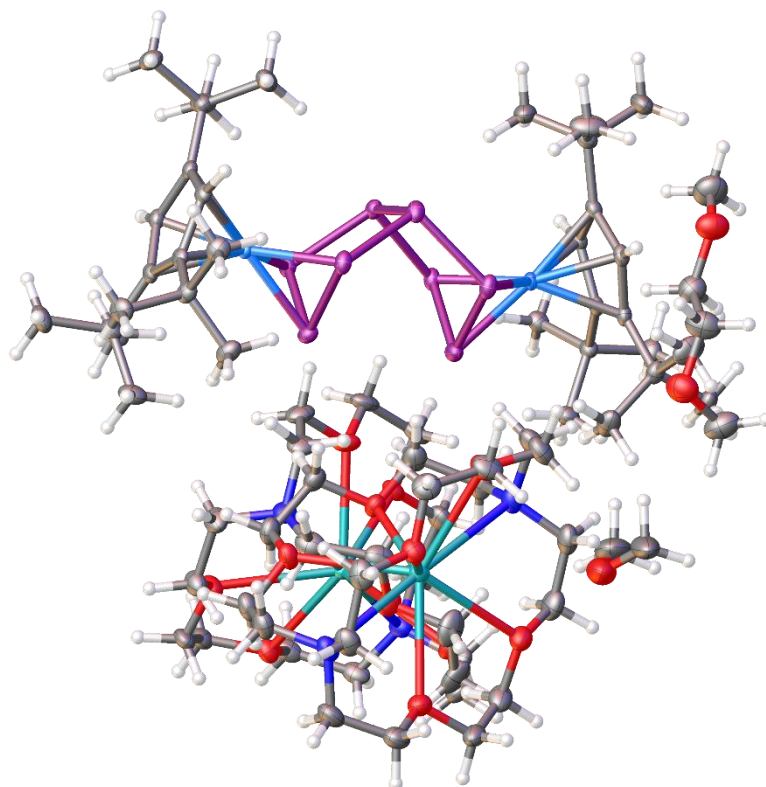


Figure S20: Molecular structure of **2b** in the solid state. Thermal ellipsoids are drawn with 50 % probability level.

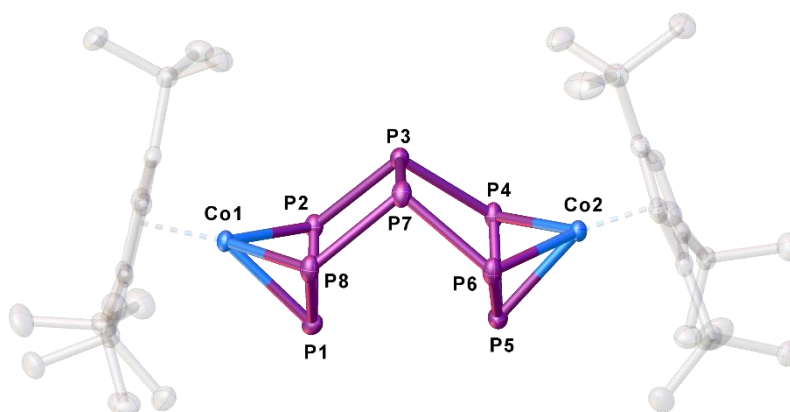


Figure S21: Molecular structure of the dianion in **2b** in the solid state. Thermal ellipsoids are drawn with 50 % probability level. Selected bond lengths [Å] and angles [°]: P1-P2 2.1554(6), P1-P5 3.4607(6), P1-P8 2.1577(6), P2-P3 2.2152(6), P3-P4 2.2151(6), P3-P7 2.1947(6), P4-P5 2.1519(6), P5-P6 2.1580(6), P6-P7 2.2247(6), P7-P8 2.2133(6).

4.5 [Li(2,2,2-crypt)][Cp'''Co(η^3 -P₄^tBu)] (4)

Crystals suitable for X-ray single crystal structure analysis can be obtained from a concentrated solution of **4** in a in thf layered with n-hexane at room temperature. Compound **4** crystallizes in form of brown flat rods in the orthorhombic space group *Pbca*. The asymmetric unit contains one anion of **4**, one lithium counterion chelated by 2,2,2-cryptand and one molecule thf. Since the crystals were only poorly diffracting, only a measurement until a resolution of 1.20 Å was conducted. The structure in solid state is depicted in Figure S22.

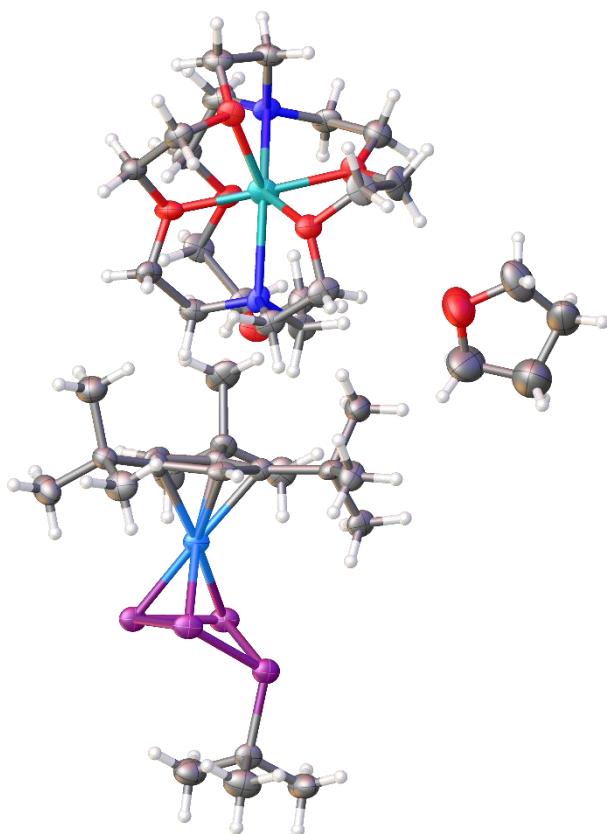


Figure S22: Molecular structure of **4** in the solid state. Thermal ellipsoids are drawn with 50 % probability level.

4.6 [Li(thf)₃][(Cp^{'''}Co)₂(μ,η³:η³-P₈^tBu)] (**5**)

Crystals suitable for X-ray single crystal structure analysis can be obtained from a concentrated solution of **5** in pentane at -30 °C. Compound **5** crystallizes in form of dark brown blocks in the monoclinic space group $P2_1/c$. The asymmetric unit contains one anion of **5**, one lithium counterion chelated by three thf molecules, 0.25 free thf and 0.5 pentane molecules. One tBu group of one Cp^{'''} ligand is disordered over two positions. Two of the coordinating thf molecules are disordered over two positions. The restraints DFIX, SIMU and SADI were applied to describe the disorder and the free solvent molecules properly. The measured crystal was a twin and therefore a HKLF5 refinement was applied. The structure in solid state is depicted in Figure S23 and S24.

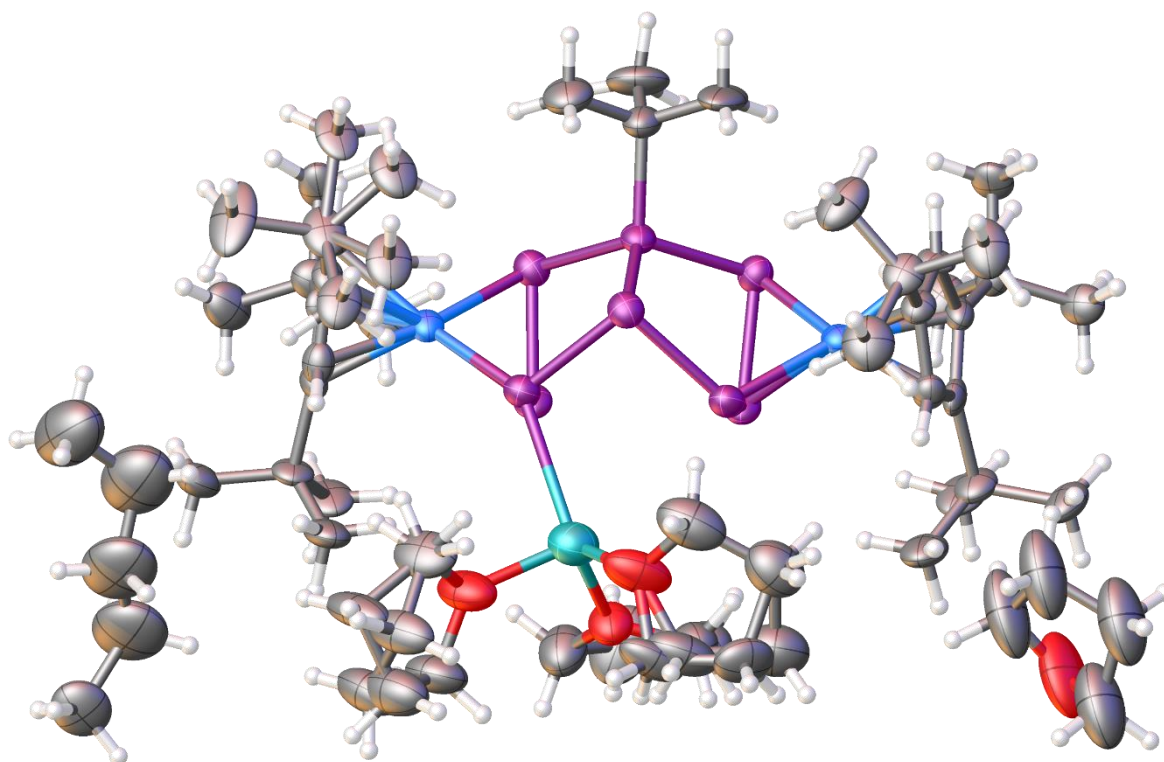


Figure S23: Molecular structure of **5** in the solid state. Thermal ellipsoids are drawn with 50 % probability level.

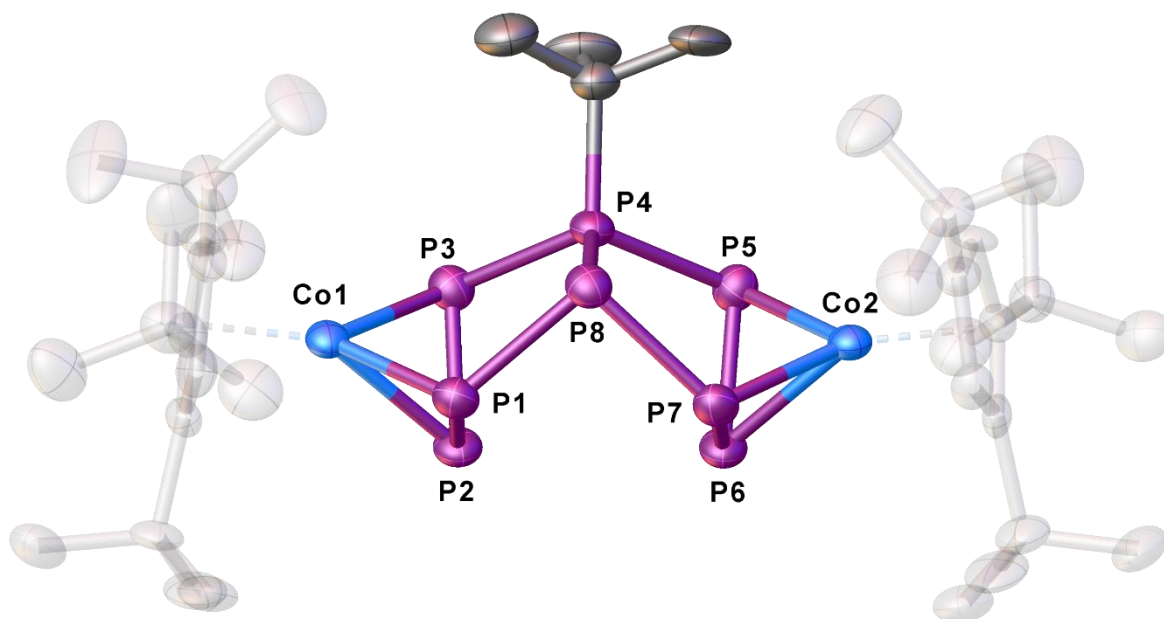


Figure 24 Molecular structure of the anion in **5** in the solid state. Thermal ellipsoids are drawn with 50 % probability level. Selected bond lengths [Å] and angles [°]: P1-P2 2.165(3), P1-P8 2.232(3), P2-P3 2.161(3), P2-P6 3.385(3), P3-P4 2.205(3), P4-P5 2.190(3), P4-P8 2.166(3), P5-P6 2.155(3), P6-P7 2.166(3), P7-P8 2.233(3).

4.6 [Li(12-c-4)₂][Cp^{'''}Co(η³-P^tBu₂)] (6)

Crystals of very poor quality were obtained from the reaction mixture of **1** and ^tBuLi in thf layered with n-hexane at -30 °C. A low resolution measurement gives a hint on the connectivity of **6**.

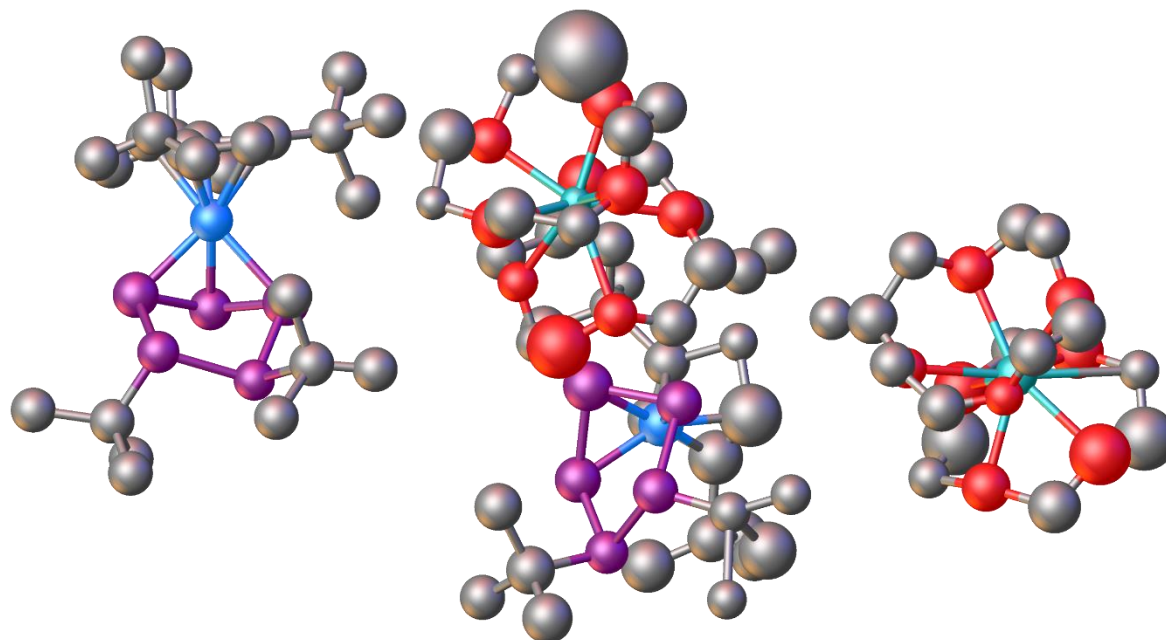


Figure S25: Molecular structure of **7** in solid state. Thermal ellipsoids are drawn with 50 % probability level.

4.3 [Li(2,2,2-crypt)][Cp'''Co(η^3 -P₄CH₂SiMe₃)] (7a)

Crystals suitable for X-ray single crystal structure analysis can be obtained from a concentrated solution of **7a** in thf layered with n-hexane -30 °C. Compound **7a** crystallizes in form of dark brown plates in the triclinic space group $P\bar{1}$. The asymmetric unit contains two anions of **7a**, two lithium atoms each chelated by two 12-c-4 molecules and 1.5 molecules thf. One thf molecule and all 12-c-4 are disordered over several positions. The restraints SIMU and SADI were applied to describe the disorder. The measured crystals was twinned and a HKLF5 refinement was applied. The structure in the solid state is depicted in Figure S26 and S27.

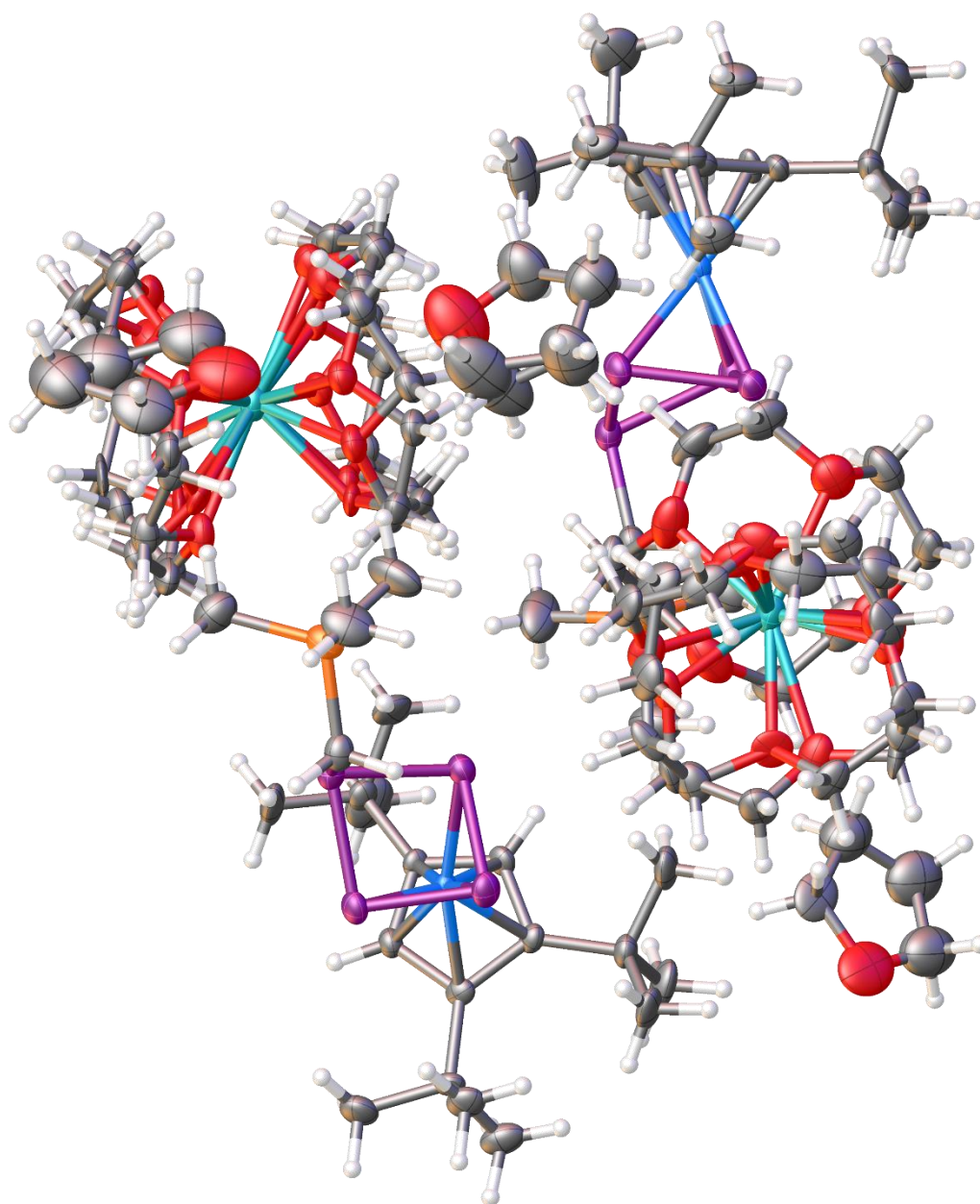


Figure S26: Molecular structure of **7a** in solid state. Thermal ellipsoids are drawn with 50 % probability level.

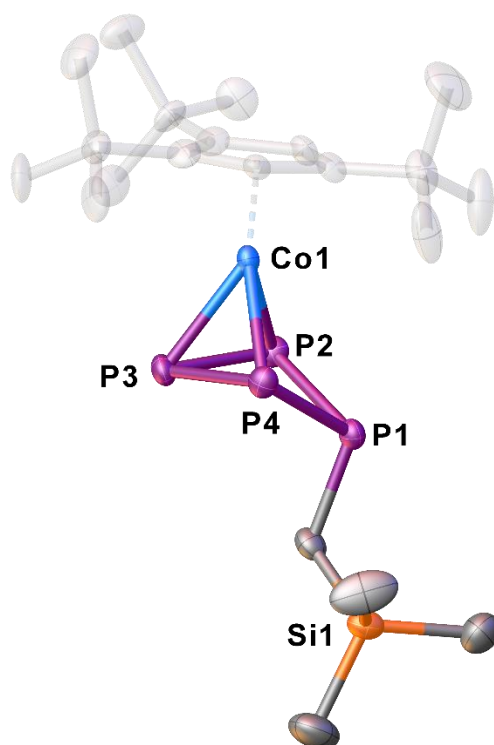


Figure S27: Molecular structure of one anion in **7a** in solid state. Thermal ellipsoids are drawn with 50 % probability level. Hydrogen atoms, cations and solvent molecules are omitted for clarity. Selected bond lengths [Å] and angles [°] of both anions: P1-P2 2.2283(16), P1-P4 2.2229(19), P2-P3 2.1877(17), P3-P4 2.1895(16), P5-P6 2.240(2), P5-P8 2.2286(16), P6-P7 2.1914(16), P7-P8 2.1898(17).

4.4 [Li(dme)₃][(Cp'''Co)₂(μ,η³:η³-P₈CH₂SiMe₃)] (**8**)

Crystals suitable for X-ray single crystal structure analysis can be obtained from a concentrated solution of **8** in a mixture of n-hexane and dme at -30 °C. Compound **8** crystallizes in form of dark green blocks in the triclinic space group P $\bar{1}$. The asymmetric unit contains one anion of **8**, one lithium counterion chelated by three molecules dme and 0.5 molecules n-hexane. The structure in solid state is depicted in Figure S28 and S29.

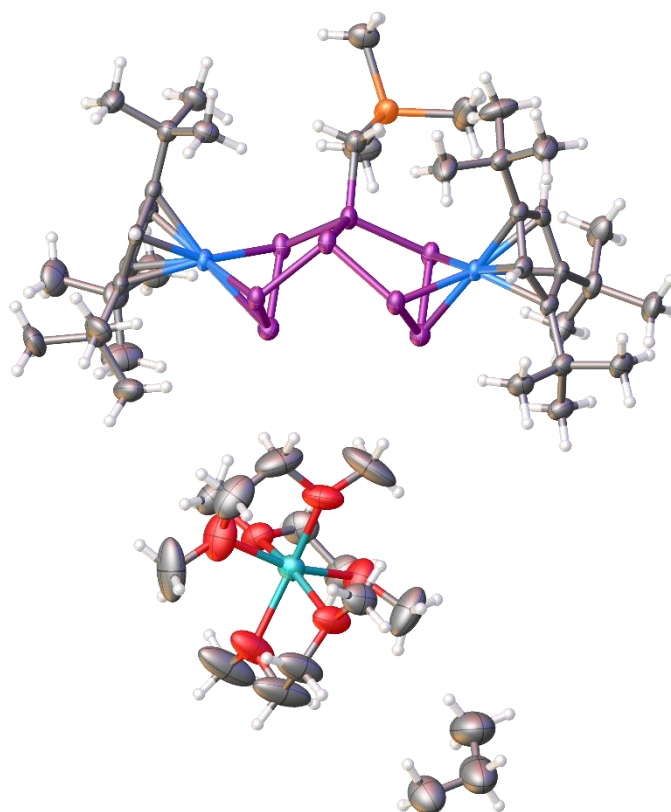


Figure S28. Asymmetric unit of **8** in the solid state. Thermal ellipsoids are drawn with 50 % probability level.

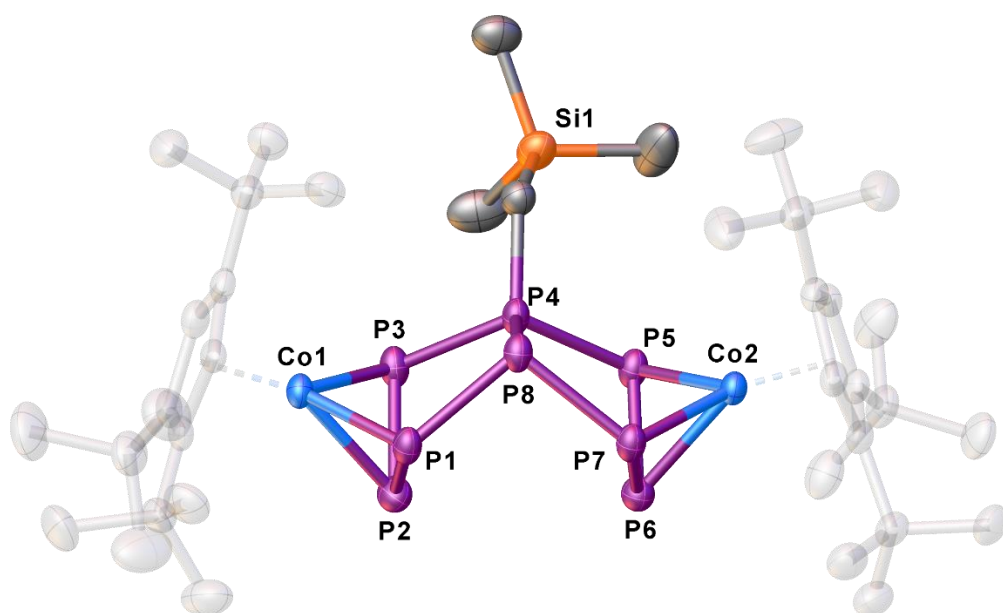


Figure S29: Molecular structure of the anion in **8** in the solid state. Thermal ellipsoids are drawn with 50 % probability level. Selected bond lengths [Å] and angles [°]: P1-P2 2.1643(11), P1-P8 2.2176(10), P2-P3 2.1574(11), P2-P6 3.7294(11), P3-P4 2.1948(10), P4-P5 2.1829(10), P4-P8 2.1738(10), P5-P6 2.1494(11), P6-P7 2.1575(11), P7-P8 2.2174(10).

3.6 [Na(dme)₂][Cp^{'''}Co(η³-P₄(O)H)] (10)

Crystals suitable for X-ray single crystal structure analysis can be obtained from a concentrated solution of **10** in a in dme layered with n-hexane at -30 °C. Compound **10** crystallizes in form of dark brown blocks in the triclinic space group P $\bar{1}$. The asymmetric unit contains one anion of **10**, one sodium counterion chelated by two molecules dme. The structure in solid state is depicted in Figure S30 and S31.

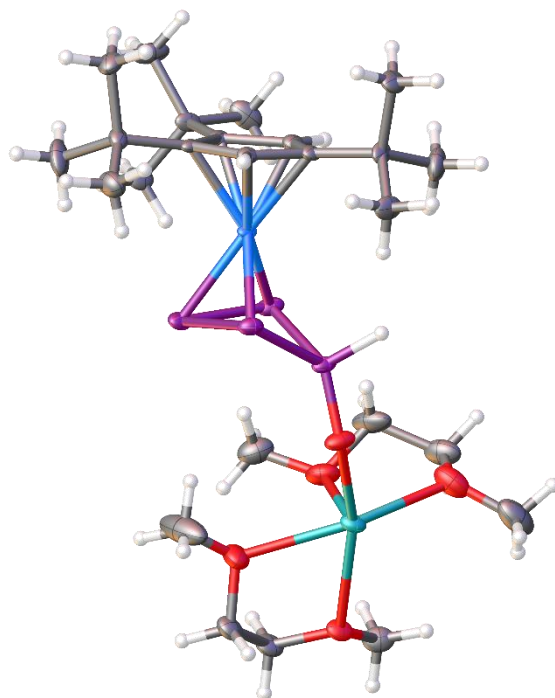


Figure S30: Molecular structure of **10** in the solid state. Thermal ellipsoids are drawn with 50 % probability level.

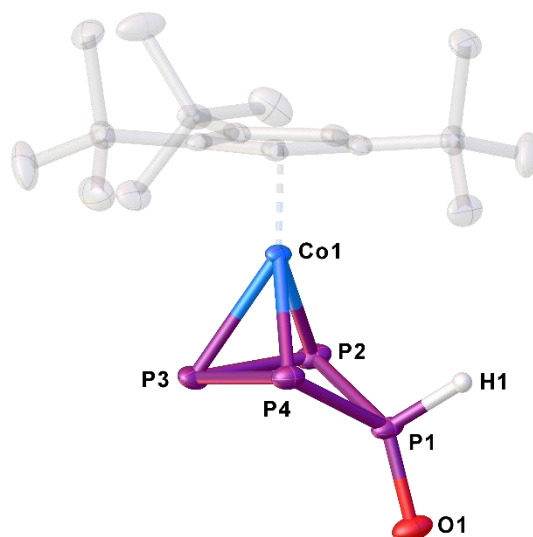


Figure S31: Molecular structure of the anion in **10** in the solid state. Thermal ellipsoids are drawn with 50 % probability level. Selected bond lengths [Å] and angles [°]: P1-P2 2.1659(7), P2-P3 2.1984(8), P3-P4 2.1938(8), P1-P4 2.1676(7), P1-H1 1.30(3), P1-O1 1.5200(15).

3.7 Crystallographic information

Table S12: Crystallographic data for all compounds

	2a	2b	4	5
CCDC		1984450	1984451	1984452
Formula		C ₇₆ H ₁₄₄ Co ₂ K ₂ N ₄ O ₁₅ P ₈	C ₄₃ H ₈₂ CoLiN ₂ O ₇ P ₄	C _{53.5} H ₉₉ Co ₂ LiO _{3.25} P ₈
<i>D</i> _{calc.} / g cm ⁻³		1.287	1.229	1.231
μ /mm ⁻¹		5.379	4.248	0.768
Formula Weight		1797.76	928.85	1166.88
Colour	dark green	clear dark green	clear dark brown	dark brown
Shape	block	plate	needle	block
Size/mm ³	0.63×0.38×0.32	0.23×0.16×0.05	0.67×0.12×0.06	0.32×0.13×0.10
<i>T</i> /K	123	123	123(1)	123(1)
Crystal System	orthorhombic	triclinic	orthorhombic	monoclinic
Space Group	<i>Cmce</i>	<i>P</i> -1	<i>Pbca</i>	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> /Å	27.1353(5)	13.7131(2)	17.8715(3)	31.947(3)
<i>b</i> /Å	26.3358(7)	15.2042(2)	22.5218(4)	15.5710(12)
<i>c</i> /Å	21.7358(6)	25.3794(3)	24.9460(5)	12.6636(10)
α /°	90	74.8590(10)	90	90
β /°	90	78.8180(10)	90	91.670(8)
γ /°	90	65.9260(10)	90	90
<i>V</i> /Å ³	15533.0(7)	4640.57(11)	10040.7(3)	6296.8(9)
<i>Z</i>		2	8	4
<i>Z</i> '		1	1	1
Wave length/Å	0.71073	1.54184	1.54184	0.71073
Radiation type		CuK α	Cu K α	Mo K α
θ_{min} /°		3.386	3.544	3.242
θ_{max} /°		73.012	40.147	25.156
Measured Refl.		34504	13776	19403
Independent Refl.		17809	3009	19403
Reflections with <i>I</i> > 2(<i>I</i>)		16566	2723	9021
<i>R</i> _{int}		0.0252	0.0414	.
Parameters		985	535	747
Restraints		0	0	277
Largest Peak		0.490	0.433	1.250
Deepest Hole		-0.360	-0.352	-0.620
GooF		1.022	1.066	0.876
<i>wR</i> ₂ (all data)		0.0794	0.1273	0.1886
<i>wR</i> ₂		0.0773	0.1223	0.1727
<i>R</i> ₁ (all data)		0.0344	0.0503	0.1377
<i>R</i> ₁		0.0311	0.0459	0.0745

	6	7a	8	10
CCDC		1984453	1984454	1984455
Formula		C ₈₀ H ₁₅₆ Co ₂ Li ₂ O _{17.5} P ₈ Si ₂	C ₅₃ H ₁₀₆ Co ₂ LiO ₆ P ₈ Si	C ₂₅ H ₅₀ O ₅ NaP ₄ Co
$D_{calc.}/\text{g cm}^{-3}$		1.196	1.203	1.299
μ/mm^{-1}		0.531	6.043	6.376
Formula Weight		1833.72	1240.02	636.45
Colour	dark brown	clear dark brown	clear dark green	clear dark brown
Shape	block	plate	plate	block
Size/mm ³	0.066×0.082×0.111	0.39×0.13×0.04	0.13×0.05×0.04	0.51×0.20×0.14
T/K	122.9 (3)	123(1)	123.01(10)	123.0(2)
Crystal System	monoclinic	triclinic	triclinic	triclinic
Space Group	<i>Pn</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a</i> /Å	13.520(3)	14.4740(7)	14.2675(3)	8.8103(3)
<i>b</i> /Å	26.875(4)	18.2968(4)	15.0423(4)	10.6602(3)
<i>c</i> /Å	15.467(5)	19.7326(6)	18.2337(5)	17.7138(5)
$\alpha/^\circ$	90	87.037(2)	102.047(2)	78.926(2)
$\beta/^\circ$	99.51(3)	77.660(4)	94.044(2)	89.741(2)
$\gamma/^\circ$	90	86.306(3)	114.602(2)	85.158(2)
<i>V</i> /Å ³	5543(2)	5090.5(3)	3424.69(16)	1626.75(9)
<i>Z</i>		2	2	2
<i>Z'</i>		1	1	1
Wave length/Å	1.54184	0.71073	1.54184	1.54184
Radiation type		Mo K α	Cu K α	Cu K α
$\theta_{min}/^\circ$		3.307	3.462	2.542
$\theta_{max}/^\circ$		25.125	73.444	74.291
Measured Refl.		26773	24262	10786
Independent Refl.		26773	13133	6333
Reflections with <i>I</i> > 2(<i>I</i>)		15803	10990	6143
<i>R</i> _{int}		.	0.0488	0.0425
Parameters		1430	668	342
Restraints		1102	0	0
Largest Peak		0.589	0.700	0.686
Deepest Hole		-0.437	-0.486	-0.609
GooF		0.908	1.018	1.038
<i>wR</i> ₂ (all data)		0.1123	0.1321	0.1183
<i>wR</i> ₂		0.1051	0.1219	0.1169
<i>R</i> ₁ (all data)		0.0967	0.0608	0.0441
<i>R</i> ₁		0.0526	0.0494	0.0432

4. Computational Details

Gaussian 09 program package was used throughout.^[9] Density functional theory (DFT) in form of Becke's three-parameter hybrid functional B3LYP^[10] with def2-TZVP all electron basis set^[11] was employed. For solvents effects has been accounted by using continuous polarizable continuum model (CPM).^[12] The dielectric constant of thf ($\epsilon = 7.4257$) has been used in the calculations. The Natural Bond Orbital (NBO) analysis has been performed with the NBO6 program.^[13] The long range dispersion correction GD3BJ was applied.^[14] The figures for the energy schemata were created using OLEX2.^[7] The figures for the supporting information concerning the DFT calculations were created with Chemcraft.^[15]

Table S13: Total energies for all optimized geometries (B3LYP/def2-TZVP level of theory).

	total energy [Ha]
1 [Cp ^{'''} Co(η^4 -P ₄)]	-3414.10100308
2 [(Cp ^{'''} Co) ₂ (μ , η^3 : η^3 -P ₈)] ²⁻	-6328.46309287
3 [Cp ^{'''} Co(η^3 -P ₃)] ⁻	-3072.8032350
4a [Cp ^{'''} Co(η^3 -P ₄ ^t Bu)] ⁻ (axial isomer)	-3572.14117871
4b [Cp ^{'''} Co(η^3 -P ₄ ^t Bu)] ⁻ (equatorial isomer)	-3572.13664123
5 [(Cp ^{'''} Co) ₂ (μ , η^3 : η^3 -P ₈ ^t Bu)] ⁻	-6986.30353930
6 [Cp ^{'''} Co(η^3 -P ₅ ^t Bu ₂)] ⁻	-4071.49911149
7a [Cp ^{'''} Co(η^3 -P ₄ CH ₂ SiMe ₃)] ⁻ (axial isomer)	-3862.92310537
7b [Cp ^{'''} Co(η^3 -P ₄ CH ₂ SiMe ₃)] ⁻ (equatorial isomer)	-3862.92275672
8 [(Cp ^{'''} Co) ₂ (μ , η^3 : η^3 -P ₈ CH ₂ SiMe ₃)] ⁻	-7277.08151112
I-8 [Cp ^{'''} Co(η^3 -P ₅ (CH ₂ SiMe ₃) ₂)] ⁻	-4653.06490984
10 [Cp ^{'''} Co(η^3 -P ₄ (O)H)] ⁻	-3490.11025036

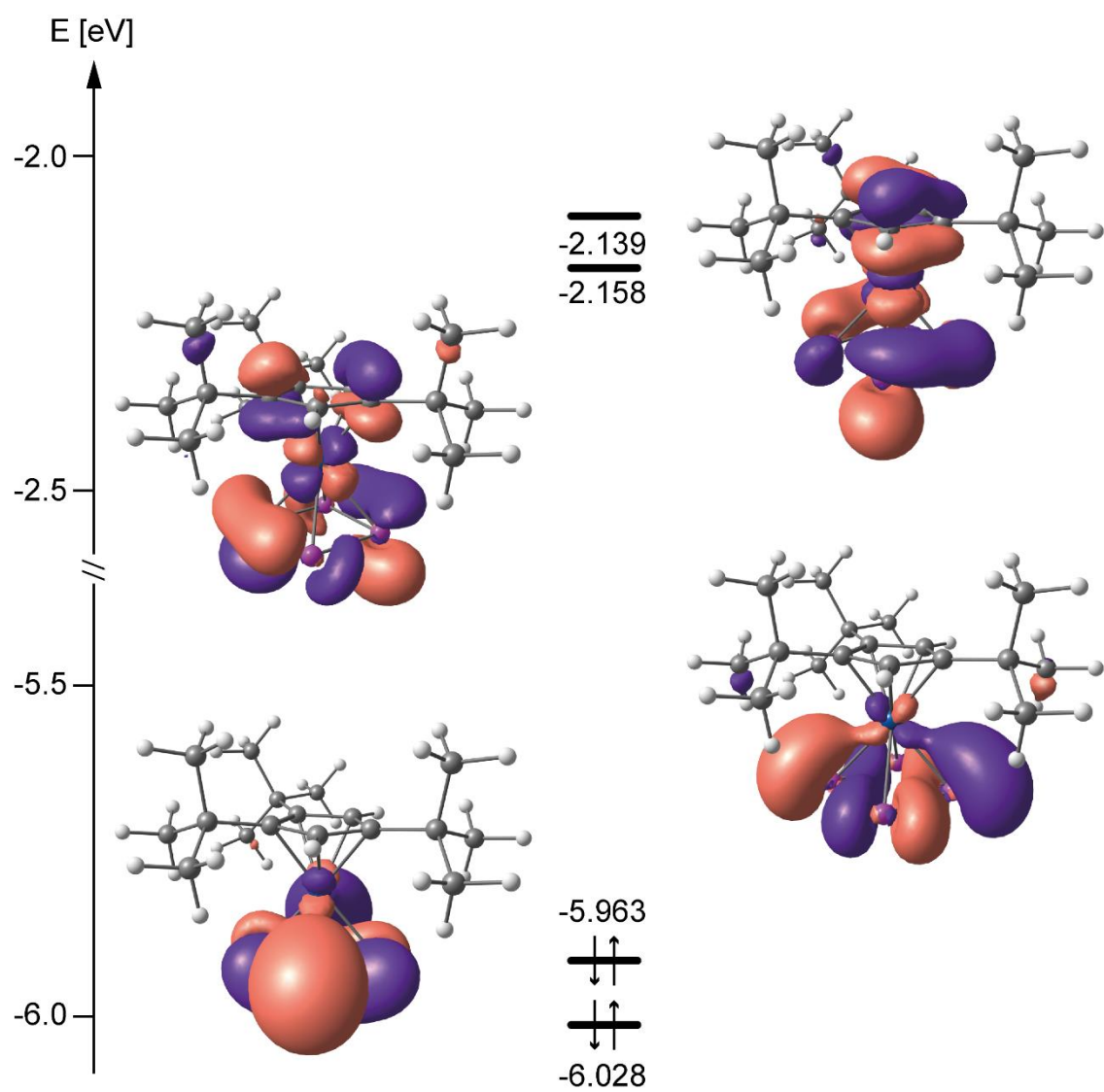
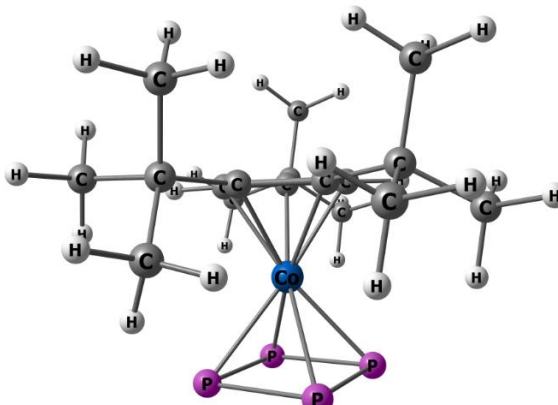
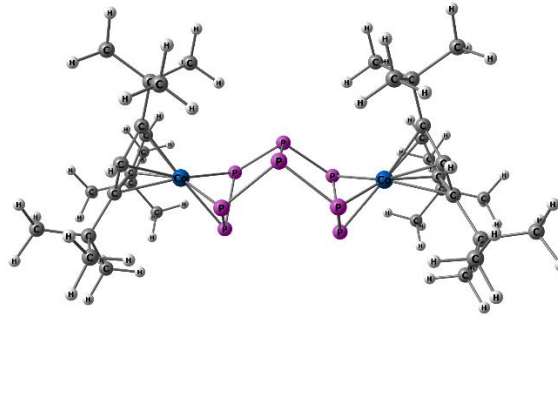


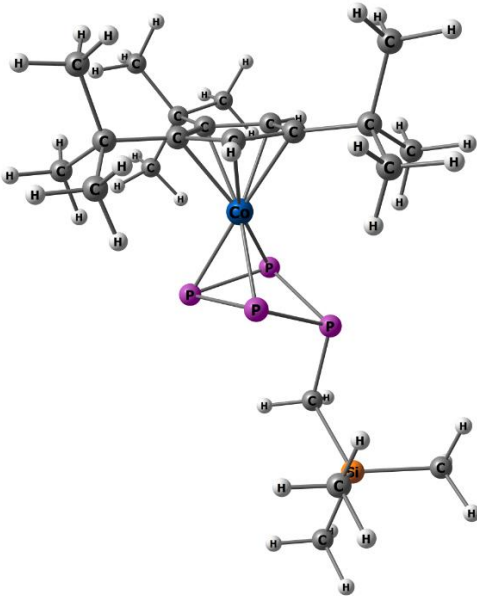
Figure S32: Molecular frontier orbitals of **1** (B3LYP/def2-TZVP level of theory).

Table S14: Optimized geometries of **1** (left), **2** (right). XYZ coordinated in angstroms. B3LYP/def2-TZVP level of theory.

Co	-0.408168000	-0.538547000	-0.006352000	H	-1.830566000	3.780242000	-1.019022000	Co	-3.312766000	-0.098603000	-0.053425000	H	-3.913640000	-2.647455000	1.898689000
P	-2.337380000	-1.800346000	0.372129000	H	-2.866529000	1.660822000	-2.025728000	Co	3.312381000	-0.101957000	-0.041461000	C	-5.546422000	-1.008106000	3.185789000
P	-0.515162000	-2.411411000	1.364220000	H	-4.308544000	1.998929000	-1.070804000	P	-1.746035000	-0.106960000	-1.677834000	H	-4.493251000	-0.825748000	3.398501000
P	0.458579000	-2.618151000	-0.563909000	H	-3.524587000	0.420284000	-0.953829000	P	0.017755000	1.010734000	-0.940406000	H	-5.898903000	-1.780338000	3.872499000
P	-1.365614000	-1.980853000	-1.552651000	H	-3.315378000	0.673198000	1.595764000	P	1.727521000	-0.209689000	-1.644879000	H	-6.108237000	-0.094483000	3.388018000
C	-1.101610000	1.425695000	0.083090000	H	-4.197911000	2.193416000	1.409185000	P	-1.944336000	-1.883204000	-0.453992000	C	-5.129825000	0.016330000	0.473416000
C	-0.283568000	1.113922000	1.197778000	H	-2.661424000	2.174354000	2.269712000	P	-1.743203000	-0.795805000	1.414439000	H	-5.807183000	3.922292000	-0.378177000
C	1.033190000	0.739624000	0.774553000	H	1.205984000	1.649673000	3.418163000	P	1.756994000	-0.721101000	1.482858000	H	-4.834272000	5.065544000	0.560146000
C	1.028263000	0.805612000	-0.680731000	H	2.388615000	0.453636000	3.941663000	P	-0.016459000	0.582211000	1.225777000	H	-5.681247000	3.741266000	1.374943000
C	-0.290187000	1.219186000	-1.058320000	H	0.790053000	-0.075582000	3.425335000	P	1.945466000	-1.911916000	-0.321124000	C	5.780995000	-1.369673000	1.797055000
C	-2.490896000	2.020085000	0.098716000	H	2.041631000	-1.652956000	2.068764000	C	-4.370499000	1.684132000	0.178202000	C	7.278957000	-1.660930000	1.597439000
C	-2.310609000	3.544405000	-0.067907000	H	3.624139000	-0.953968000	2.410182000	C	5.341243000	-0.310357000	-0.691722000	H	7.477728000	-2.205185000	0.678985000
C	-3.343566000	1.489436000	-1.060441000	H	3.098376000	-1.186951000	0.745906000	C	-4.861684000	0.795642000	1.241166000	H	7.653412000	-2.26984000	2.426206000
C	-3.203834000	1.744160000	1.429576000	H	3.786258000	1.486308000	0.689520000	H	-4.437712000	0.990529000	2.271161000	H	7.851962000	-0.731725000	1.569746000
C	2.142548000	0.494508000	1.802036000	H	3.951723000	1.502158000	2.437386000	C	5.955771000	-1.220365000	-1.759498000	C	-5.993624000	-1.101623000	-1.798157000
C	1.585083000	0.644575000	3.229595000	H	2.773681000	2.576462000	1.675371000	C	4.681824000	0.878773000	1.171211000	C	3.209669000	3.502178000	-1.241809000
C	2.765784000	-0.909096000	1.736642000	H	3.079865000	2.559767000	-1.072830000	H	4.468169000	1.144577000	2.189374000	H	3.891894000	3.400000000	-2.087711000
C	3.221191000	1.581998000	1.630553000	H	1.811175000	2.820230000	-2.268772000	C	-5.262051000	-0.411910000	0.755946000	H	2.875854000	4.542156000	-1.208688000
C	2.105389000	0.720236000	-1.766013000	H	3.374779000	2.169083000	-2.770359000	C	-4.800836000	1.012176000	-0.996628000	H	2.342020000	2.867699000	-1.421632000
C	2.624373000	2.159882000	-1.977602000	H	3.951069000	-0.198254000	-2.351825000	H	-4.696475000	1.405580000	-1.990593000	C	-5.368338000	-0.745951000	-3.161826000
C	3.289607000	-0.207270000	-1.484116000	H	2.953231000	-1.231732000	-1.330312000	C	4.783386000	0.945581000	-1.077894000	H	-5.582423000	0.281116000	-3.456870000
C	1.492290000	0.234507000	-3.095074000	H	3.981021000	0.100562000	-0.625319000	H	4.658699000	1.270881000	-2.093680000	H	-5.789412000	-1.398363000	-3.929390000
H	-0.613699000	1.135785000	2.218566000	H	2.280255000	0.180624000	-3.848039000	C	5.278874000	-0.356663000	0.758910000	H	-4.286405000	-0.876642000	-3.149072000
H	-0.621333000	1.351209000	-2.070833000	H	0.726619000	0.904717000	-3.478843000	C	-5.352406000	-0.267288000	-0.685756000	C	-7.493676000	-0.747680000	-1.880318000
H	-1.696146000	3.954249000	0.735443000	H	1.058225000	-0.758704000	-2.986451000	C	5.306533000	-0.945705000	-3.131171000	H	-8.019285000	-0.978940000	-0.955910000
H	-3.282948000	4.039853000	-0.042681000					H	4.223118000	-1.057299000	-3.087379000	H	-7.970930000	-1.303850000	-2.692082000
								H	5.700822000	-1.653940000	-3.862592000	H	-7.619727000	0.318430000	-2.079084000
								H	5.529883000	0.055950000	-3.497895000	C	-2.966509000	3.288115000	1.501129000
								C	3.892197000	3.128355000	0.078950000	H	-3.475386000	0.303927000	2.432745000
								C	4.372391000	1.694419000	0.055340000	H	-2.636313000	4.327271000	1.575322000
								C	-3.8952004000	3.114203000	0.291527000	H	-2.087363000	2.650365000	1.412264000
								H	-7.247292000	-1.757329000	1.563800000	C	5.121413000	4.037779000	0.281009000
								H	-7.811945000	-0.824741000	1.627294000	H	5.618905000	3.813949000	1.227032000
								H	-7.600709000	-2.422638000	2.354769000	H	4.826428000	5.090706000	0.291114000
								H	-7.480833000	-2.228289000	0.612218000	H	5.846078000	3.894038000	-0.523466000
								C	7.456678000	-0.886633000	-1.893288000	C	4.980761000	-2.683798000	1.801529000
								H	7.587523000	0.165965000	-2.151944000	H	3.947825000	-2.487533000	2.085016000
								H	7.910728000	-1.490177000	-2.684297000	H	5.415189000	-3.377859000	2.527313000
								H	8.001257000	-1.072874000	-0.969894000	H	4.963367000	-3.167605000	0.832557000
								C	-3.144057000	3.549584000	-0.973862000	C	-5.830293000	-2.617363000	-1.638686000
								H	-2.261649000	2.931289000	-1.135161000	H	-4.776995000	-2.878892000	-1.536478000
								H	-2.826192000	4.590979000	-0.880584000	H	-6.226108000	-3.118825000	-2.525196000
								C	3.779086000	3.474619000	-1.858415000	H	-6.367714000	-3.008470000	-0.780716000
								H	5.781655000	-2.721366000	-1.500852000	C	2.905631000	3.359373000	1.229376000
								H	6.328500000	-3.062129000	-0.627258000	H	2.015625000	2.743500000	1.107028000
								H	6.159068000	-3.282625000	-2.358984000	H	2.600895000	4.408596000	1.258671000
								H	4.727852000	-2.965248000	-1.364799000	H	3.353325000	3.114988000	2.194064000
								C	-5.741246000	-1.492810000	1.736760000	C	5.638988000	-0.782731000	3.214227000
								C	-4.956302000	-2.812079000	1.631603000	H	6.217536000	3.3381000	3.331068000
								H	-4.971812000	-3.233961000	0.634413000	H	6.007316000	-1.509104000	3.941404000
								H	-5.79679000	-3.546589000	2.323127000	H	4.596889000	-0.571056000	3.452930000

Table S15: Optimized geometries of **7a** (left), **7b** (right). XYZ coordinated in angstroms. B3LYP/def2-TZVP level of theory.

											
Co	-0.669106000	-0.047351000	-0.081384000	H	6.892250000	-2.089224000	-0.957118000	Co	0.588261000	-0.289588000	-0.319910000
P	0.417185000	-1.994302000	-0.017222000	C	-2.055103000	4.258415000	0.234348000	P	0.568812000	-2.429688000	-0.925538000
P	0.721325000	-0.685682000	-1.765458000	H	-2.723350000	4.175456000	-0.625065000	P	-0.903372000	-1.752966000	0.566166000
P	2.446838000	0.324146000	-0.764081000	H	-1.655324000	5.275573000	0.259941000	P	-2.360910000	-1.654171000	-1.140221000
P	1.287570000	-0.275137000	1.048524000	H	-2.646457000	4.100868000	1.138623000	P	-0.493262000	-0.984535000	-2.195092000
Si	5.369635000	-0.436958000	0.126659000	C	-3.454595000	-0.198446000	-3.009866000	Si	-4.640146000	-0.100799000	0.453967000
C	-2.074253000	1.172977000	-0.996390000	H	-3.941716000	0.776893000	-2.962800000	C	0.835633000	1.118693000	1.177437000
H	-1.898262000	1.519476000	-2.009140000	H	-3.999537000	-0.812841000	-3.729152000	H	0.330297000	1.201971000	2.060079000
C	-1.534829000	1.846498000	0.124253000	H	-2.440986000	-0.063444000	-3.387948000	C	0.678986000	1.804020000	-0.032063000
C	3.842555000	-0.814448000	-0.859686000	C	0.014689000	3.399765000	1.348490000	C	-3.323348000	-0.081196000	-0.885275000
H	3.473902000	-1.897687000	-0.551220000	H	0.525882000	3.280379000	2.289128000	H	-2.628882000	0.734925000	-0.684247000
H	4.135078000	-0.899410000	-1.912639000	H	0.453865000	4.400690000	1.341897000	C	-3.821724000	0.143023000	-1.835893000
C	-2.557663000	-0.144604000	0.836095000	H	0.818838000	2.664534000	1.324277000	C	2.573902000	0.445460000	-0.302390000
C	-1.844639000	1.025349000	1.241781000	C	-3.167600000	-2.578855000	1.466035000	C	1.691299000	1.373135000	-0.931284000
H	-1.547539000	1.239341000	2.251419000	H	-3.900172000	-2.743782000	0.602596000	H	1.768931000	1.693949000	-1.953915000
C	-2.711662000	-0.049674000	-0.607946000	H	-3.465879000	-3.189266000	2.321468000	C	2.089314000	0.277784000	1.059289000
C	-4.933561000	-1.066117000	-1.241209000	H	-2.197443000	-2.928428000	1.113987000	C	4.064648000	-0.029446000	2.596501000
H	-5.059866000	-1.641362000	-0.329165000	C	6.158000000	1.090864000	-0.638305000	H	4.769510000	-0.284753000	1.810635000
H	-5.482230000	-1.578363000	-2.035297000	H	6.437307000	0.906972000	-1.678883000	H	4.410893000	-0.501724000	3.519251000
H	-5.394748000	-0.088537000	-1.086195000	H	7.060374000	1.382514000	-0.094807000	H	4.098308000	1.053229000	2.735345000
C	-0.913739000	3.225756000	0.141017000	H	5.464857000	1.934602000	-0.623615000	C	-0.293645000	2.939672000	-0.251763000
C	-2.802782000	-2.277696000	-1.879008000	C	-4.465054000	-0.626720000	2.366998000	C	2.564545000	-2.013035000	2.124418000
H	-1.805819000	-2.142919000	-2.298564000	H	-4.402424000	0.392726000	2.751978000	H	1.527000000	-2.333760000	2.028320000
H	-3.400573000	-2.852696000	-2.592114000	H	-4.847630000	-1.269698000	3.163889000	H	2.989735000	-2.491022000	3.011701000
C	-2.697900000	-2.860318000	-0.972246000	H	-5.186583000	-0.632028000	1.551015000	H	3.099216000	-2.373555000	1.254874000
C	-2.139805000	-1.100023000	3.127666000	C	4.936254000	-0.074772000	1.918299000	C	3.506783000	-0.095727000	-2.577386000
H	-1.120055000	-1.353765000	2.837333000	H	4.239186000	0.762257000	1.990598000	H	2.645733000	-0.734818000	-2.774724000
H	-2.486785000	-1.836166000	3.855309000	H	5.833990000	0.176149000	2.489644000	H	4.370809000	-0.502905000	-3.106146000
H	-2.123234000	-0.132802000	3.627913000	H	4.459602000	-0.936282000	2.390113000	H	3.304817000	0.889095000	-2.996553000
C	-3.455821000	-0.905064000	-1.640703000	C	-0.110295000	3.485038000	-1.140175000	C	2.628426000	-0.483664000	2.277101000
C	-3.076615000	-1.112985000	1.903104000	H	0.720705000	2.784746000	-1.224646000	C	3.810330000	-0.039467000	-1.066686000
C	6.609938000	-1.954560000	0.072223000	H	0.292344000	4.500981000	-1.129489000	C	-5.28624000	1.561601000	0.435774000
H	6.187673000	-2.758930000	0.517086000	H	-0.733394000	3.384593000	-2.030260000	H	-4.824425000	2.377489000	0.622978000
H	7.521231000	-1.603737000	0.621713000					H	-6.303227000	1.603437000	1.204236000

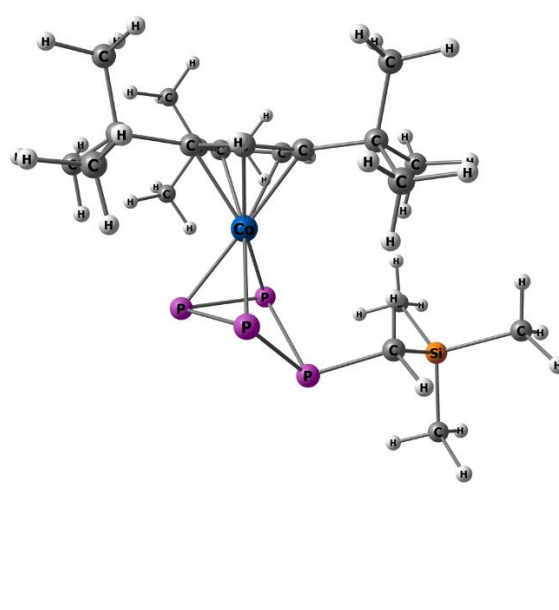


Table S16: Optimized geometries of **8** (left), **4a** (right). XYZ coordinated in angstroms. B3LYP/def2-TZVP level of theory.

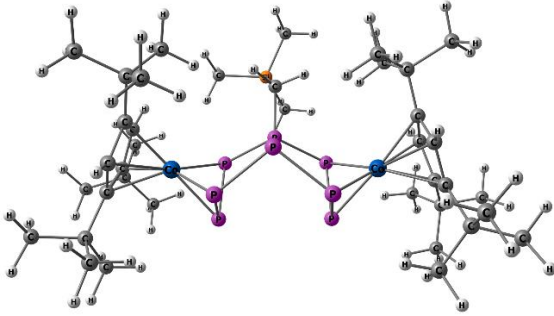
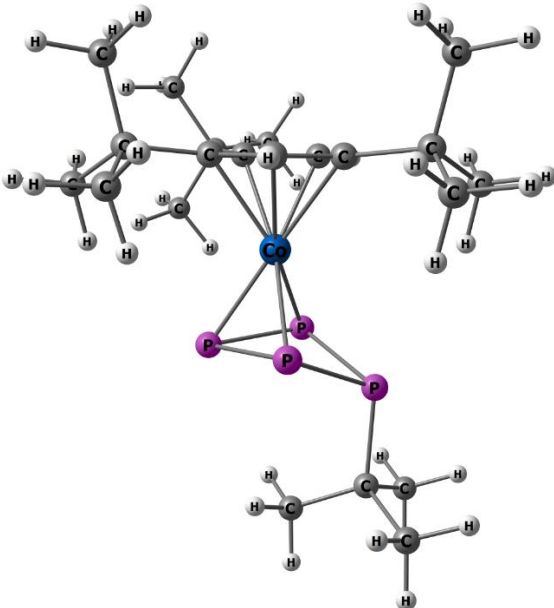
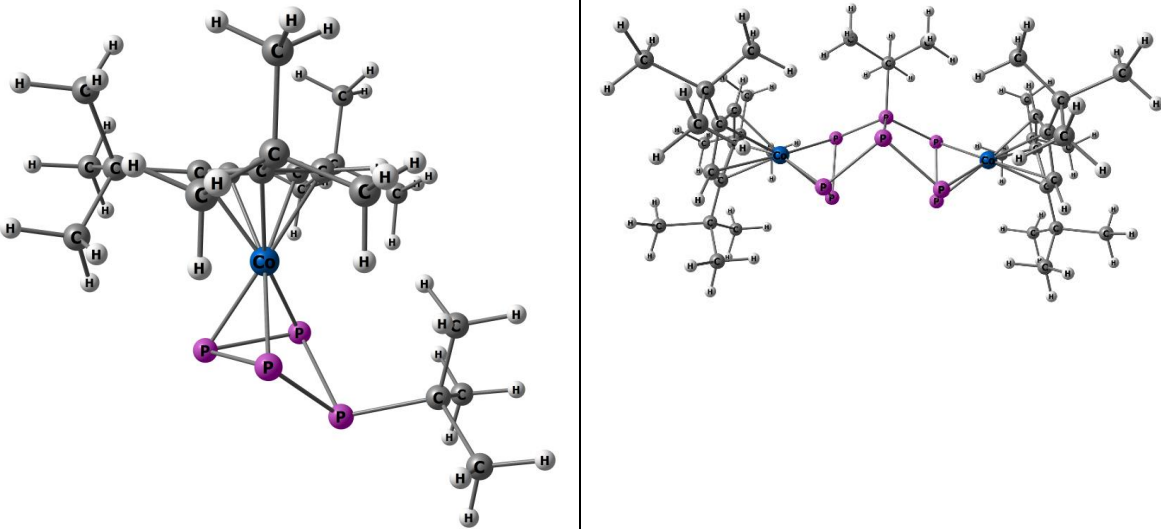
					
Co	-3.359829000	-0.512317000	0.140542000	H	3.124363000
Co	3.447489000	-0.356550000	0.027198000	C	5.717591000
P	0.001954000	0.872673000	-0.249667000	C	6.020644000
P	0.061479000	-0.374700000	1.583816000	C	-7.287224000
P	1.861258000	-1.626012000	1.051570000	H	-7.469509000
P	1.872749000	0.466951000	-1.409020000	H	-7.454589000
P	-1.895711000	-1.793448000	-1.101041000	H	-7.899801000
P	2.000419000	-1.708503000	-1.149531000	C	-4.967029000
P	-1.880148000	0.390656000	-1.340870000	H	-3.960022000
P	-1.677586000	-1.727370000	1.087201000	H	-5.419325000
Si	0.336352000	0.038495000	-1.086255000	H	-4.846589000
C	4.713379000	1.057225000	0.942222000	C	-5.704500000
C	-4.610011000	0.463885000	1.527458000	H	-6.423550000
C	-5.333016000	-0.199090000	-0.625828000	H	-5.577899000
C	-4.891093000	0.995223000	0.201183000	H	-6.150520000
H	-4.779910000	1.916679000	-0.146083000	C	-5.828394000
C	5.063498000	0.886165000	-0.431850000	H	-6.476218000
H	5.026368000	1.673949000	-1.177167000	H	-6.200498000
C	-4.855891000	-0.942768000	1.519668000	H	-4.813250000
H	-4.713151000	-1.594788000	2.378865000	C	5.545642000
C	5.452625000	-0.466753000	-0.704402000	H	6.187179000
C	4.871280000	-0.226808000	1.530720000	H	5.827540000
H	4.662590000	-0.452067000	2.571254000	H	4.501414000
C	4.432662000	2.385658000	1.672159000	C	4.946772000
C	5.331019000	-1.191422000	0.563411000	H	3.805948000
C	-0.023905000	2.672615000	0.216314000	H	5.226536000
H	0.942817000	2.875375000	0.705694000	H	4.833264000
H	0.783349000	2.781844000	0.995370000	C	0.534402000
C	-5.304275000	-1.386423000	0.230001000	H	1.718708000
C	-5.545079000	-3.457605000	-1.332373000	H	0.445840000
H	-6.133671000	-2.963966000	-2.114788000	H	0.340305000
H	-5.945854000	-4.518560000	-1.324875000	C	-3.382101000
H	-4.482285000	-3.403510000	-1.611337000	H	-3.759825000
C	5.740607000	2.822112000	2.363953000	H	-3.255370000
H	6.079204000	2.076448000	3.100807000	H	-2.393282000
H	5.594565000	3.781087000	2.891208000	C	7.213290000
H	6.550218000	2.956509000	1.628410000	H	7.431852000
C	-4.347736000	1.336806000	2.750614000	H	7.520087000
C	-5.777677000	-2.849415000	0.063608000	H	7.846532000
C	3.340887000	2.176811000	2.742863000	C	-3.758741000
H	2.401416000	1.818784000	2.296722000	H	-2.821222000
H	3.145585000	3.126881000	3.269431000	H	-3.546537000
H	3.642032000	1.435054000	3.498976000	H	-4.453156000
C	-5.838697000	0.056558000	-2.068686000	C	5.402776000
C	-7.309484000	-0.399762000	-2.213919000	H	4.304224000
H	-7.423663000	-1.487955000	-2.134896000	H	5.765351000
H	-7.704068000	-0.099894000	-3.199760000	H	5.685629000
H	-7.943225000	0.063137000	-1.440089000	C	7.548004000
C	0.230290000	5.663150000	-0.278682000	H	7.758937000
H	-0.312230000	5.850123000	0.663534000	H	7.983369000
H	0.045337000	6.521914000	-0.946900000	H	8.069771000
H	1.380697000	5.643469000	-0.047267000	C	-2.180645000
C	-5.036091000	-3.717555000	1.062620000	H	-2.552620000
H	-3.944341000	-3.687171000	0.959704000	H	-2.368711000
H	-5.322331000	-4.819632000	0.877410000	H	-2.768469000
H	-5.294653000	-3.550104000	2.108639000	C	5.755999000
C	3.995830000	3.458991000	0.680428000	H	6.261495000
H	4.801627000	3.706848000	-0.029012000	H	6.134798000
H	3.732834000	4.386627000	1.217263000	H	4.678012000
Co	0.183195000	-0.025855000	-0.063508000	H	2.905830000
P	-0.835209000	-1.921200000	-0.598456000	H	2.836201000
P	-1.435415000	-0.841347000	1.283013000	H	1.357619000
P	-2.914145000	0.355889000	0.243174000	H	-0.335889000
P	-1.598568000	-0.065154000	-1.499012000	H	0.285089000
C	1.401826000	1.049864000	1.221712000	H	-0.736708000
H	1.143422000	1.263618000	2.241462000	H	-1.164895000
C	1.062948000	1.871122000	0.123489000	C	2.934407000
C	2.202106000	-0.031280000	-0.656065000	H	3.507264000
C	1.555910000	1.189264000	-1.022933000	H	3.395594000
H	1.433502000	1.536663000	-2.031246000	H	1.919653000
C	2.102247000	-0.127028000	0.791099000	C	4.357363000
C	4.189412000	-1.253557000	1.657909000	H	4.358906000
H	4.476982000	-1.712219000	0.716566000	H	4.882159000
H	4.595113000	-1.670028000	2.463810000	H	4.919099000
H	4.664836000	-0.271790000	1.709805000	C	-0.447514000
C	0.469093000	3.261830000	0.168323000	H	-1.293037000
C	1.990799000	-2.504427000	1.747963000	H	-0.833819000
H	0.939735000	-2.416739000	2.023407000	H	0.087271000
H	2.474333000	-3.186317000	2.453462000	C	-5.444842000
C	2.032070000	-2.945670000	0.760070000	C	-5.351831000
C	2.210611000	-0.658334000	-3.094035000	H	-5.514672000
H	1.155746000	-0.925354000	-3.027157000	H	-4.834998000
H	2.684086000	-1.298343000	-3.840929000	H	-6.331434000
H	2.283030000	0.367185000	-3.453501000	C	-5.267280000
C	2.663373000	-1.122288000	1.814170000	H	-6.246325000
C	2.911729000	-0.850340000	-1.734836000	H	-4.692876000
C	1.634809000	4.266472000	0.270782000	H	-5.425081000
H	2.218860000	4.089577000	1.176170000	C	-4.369711000
H	1.259514000	5.292820000	0.300098000	H	-5.349588000
H	2.304222000	4.171894000	-0.586729000	H	-3.836022000
C	2.421610000	-0.605112000	3.245100000	H	-3.801753000

Table S17: Optimized geometries of **4b** (left), **5** (right). XYZ coordinated in angstroms. B3LYP/def2-TZVP level of theory.



Co	-0.112579000	0.0540951000	-0.506496000	H	-4.170450000	1.354418000	-1.745252000	C	7.310250000	-1.746266000	-1.873725000
P	0.418883000	-0.736931000	-2.513824000	H	-4.577232000	-0.063367000	-2.713443000	H	7.499301000	-0.793763000	-2.371588000
P	1.766942000	0.737444000	-1.597718000	H	-2.956459000	0.625029000	-2.809807000	H	7.682022000	-2.547596000	-2.517105000
P	3.037660000	-1.025433000	-1.080453000	C	1.210066000	3.169407000	1.226399000	H	7.883735000	-1.760552000	-0.947994000
P	0.999030000	-1.915377000	-0.742473000	H	1.190758000	2.689776000	2.206064000	C	-4.746663000	-2.558722000	2.153656000
C	-1.734022000	1.402031000	-0.493921000	H	1.586395000	4.186293000	1.361379000	H	-4.709068000	-3.233128000	1.300667000
H	-2.036403000	1.983046000	-1.344356000	H	1.907686000	2.623582000	0.593259000	H	-5.159066000	-3.125044000	2.999502000
C	-0.781294000	1.819974000	0.463722000	C	-2.114114000	-2.824007000	1.389097000	H	-3.723059000	-2.289970000	2.397185000
C	-1.535561000	-0.318555000	1.029510000	H	-3.085894000	-2.810363000	0.907506000	C	-7.297252000	-1.797593000	-1.839402000
C	-0.656424000	0.759042000	1.391145000	H	-2.152508000	-3.581472000	2.175345000	H	-7.505501000	-0.847890000	-2.334858000
H	-0.024079000	0.774343000	2.256888000	H	-1.365025000	-3.132859000	0.660406000	H	-7.858836000	-1.819352000	-0.906573000
C	-2.226501000	0.095007000	-0.174418000	C	-2.820278000	-1.050709000	3.021807000	H	-7.666089000	-2.604175000	-2.477897000
C	-4.603198000	-0.755932000	-0.126729000	H	-2.516806000	-0.142908000	3.546491000	C	-5.781619000	-1.901985000	-1.597177000
H	-4.449294000	-1.495876000	0.652795000	H	-2.978484000	-1.837592000	3.764288000	C	4.893172000	-2.547635000	2.147957000
H	-5.437338000	-1.095794000	-0.745551000	H	-3.771291000	-0.850082000	2.531484000	H	5.220004000	-3.098785000	2.995028000
H	-4.896993000	0.179464000	0.354126000	C	-0.093369000	3.899713000	-0.774288000	H	4.767295000	-3.215368000	1.297327000
C	-0.184226000	3.206217000	0.590007000	H	0.536941000	3.328814000	-1.455949000	H	3.778695000	-2.272472000	2.392885000
C	-2.947649000	-1.954126000	-1.694685000	H	0.336927000	4.897216000	-0.658366000	C	-5.083974000	-1.951396000	-2.973019000
H	-2.202925000	-1.647816000	-2.462313000	H	-1.078662000	4.014652000	-1.228701000	H	-5.389543000	-2.834324000	-3.537306000
H	-3.822101000	-2.304188000	-2.173426000	C	3.887028000	-0.656601000	0.584270000	H	-4.000596000	-1.971860000	-2.857677000
H	-2.520630000	-2.581213000	-1.016073000	C	4.727276000	-1.890175000	0.934041000	H	-5.350271000	-1.079634000	-3.568692000
C	-0.443960000	-1.745608000	2.811806000	H	4.095420000	-2.771338000	1.070070000	C	5.085329000	-1.928952000	-2.979955000
H	0.362455000	-2.044183000	2.145164000	H	5.452635000	-2.114303000	0.149258000	H	4.003810000	-1.965990000	-2.851678000
H	-0.625048000	-2.557181000	3.519873000	H	5.276311000	-1.725552000	1.867793000	H	5.396951000	-2.806467000	-3.549293000
H	-0.116563000	-0.880256000	3.385045000	C	2.932181000	-0.347923000	1.731723000	H	5.309303000	-1.052538000	-3.577601000
C	-3.352977000	-0.541743000	-1.001076000	H	3.496030000	-0.055797000	2.625341000	C	-5.513357000	-3.345247000	-0.999548000
C	-1.732619000	1.477259000	2.012608000	H	2.245814000	0.454331000	1.474281000	H	6.074387000	-3.525168000	-0.087477000
C	-2.127423000	4.023641000	1.502448000	C	2.330955000	-1.219553000	1.985147000	H	-4.451166000	-3.476508000	-0.794132000
H	-2.127423000	4.070275000	1.080272000	C	4.808086000	0.543558000	0.338548000	H	-5.815552000	-4.08174000	-1.719974000
H	-0.749594000	5.045091000	1.618354000	H	5.418353000	0.740477000	1.226521000	C	5.556792000	-3.318731000	-1.014284000
H	-1.191560000	3.569450000	2.492715000	H	5.484090000	0.365065000	-0.501325000	H	6.126345000	-3.491469000	-0.106529000
C	-3.783592000	0.410745000	-2.132824000	H	4.231605000	1.444464000	0.120169000	H	5.865462000	-4.076253000	-1.737718000
								H	4.479994000	-3.466023000	-0.802211000
								C	-4.094028000	3.193570000	-2.266030000
								H	-3.165606000	2.624394000	-2.483161000
								H	-3.941575000	4.217246000	-2.460358000
								H	-4.859276000	2.796043000	-2.957047000
								C	5.496112000	-0.423150000	3.162130000
								H	4.462988000	-0.106210000	3.307112000
								H	6.131665000	0.463109000	3.130399000
								H	5.789074000	-1.013763000	4.031923000
								C	-3.455794000	3.575890000	0.104070000
								H	-3.707631000	3.451526000	1.157463000
								H	-3.737597000	4.446595000	-0.097368000
								H	-2.485665000	3.113909000	-0.069275000
								C	-1.270423000	2.351659000	2.551054000
								H	-1.335560000	2.944514000	3.468029000
								H	-2.185388000	1.769874000	2.448621000
								H	-1.210255000	3.035836000	1.707459000
								C	4.021570000	3.155011000	-2.258938000
								H	3.843215000	4.215089000	-2.450847000
								H	3.103533000	2.605292000	-2.468857000
								H	4.788843000	2.815657000	-2.956841000
								C	3.394807000	3.561892000	0.116873000
								H	3.655269000	3.434978000	1.167929000
								H	2.431196000	3.084869000	-0.054598000
								H	3.292870000	4.632268000	-0.077232000

Table S18: Optimized geometries of **6** (left), **3** (right). XYZ coordinated in angstroms. B3LYP/def2-TZVP level of theory.

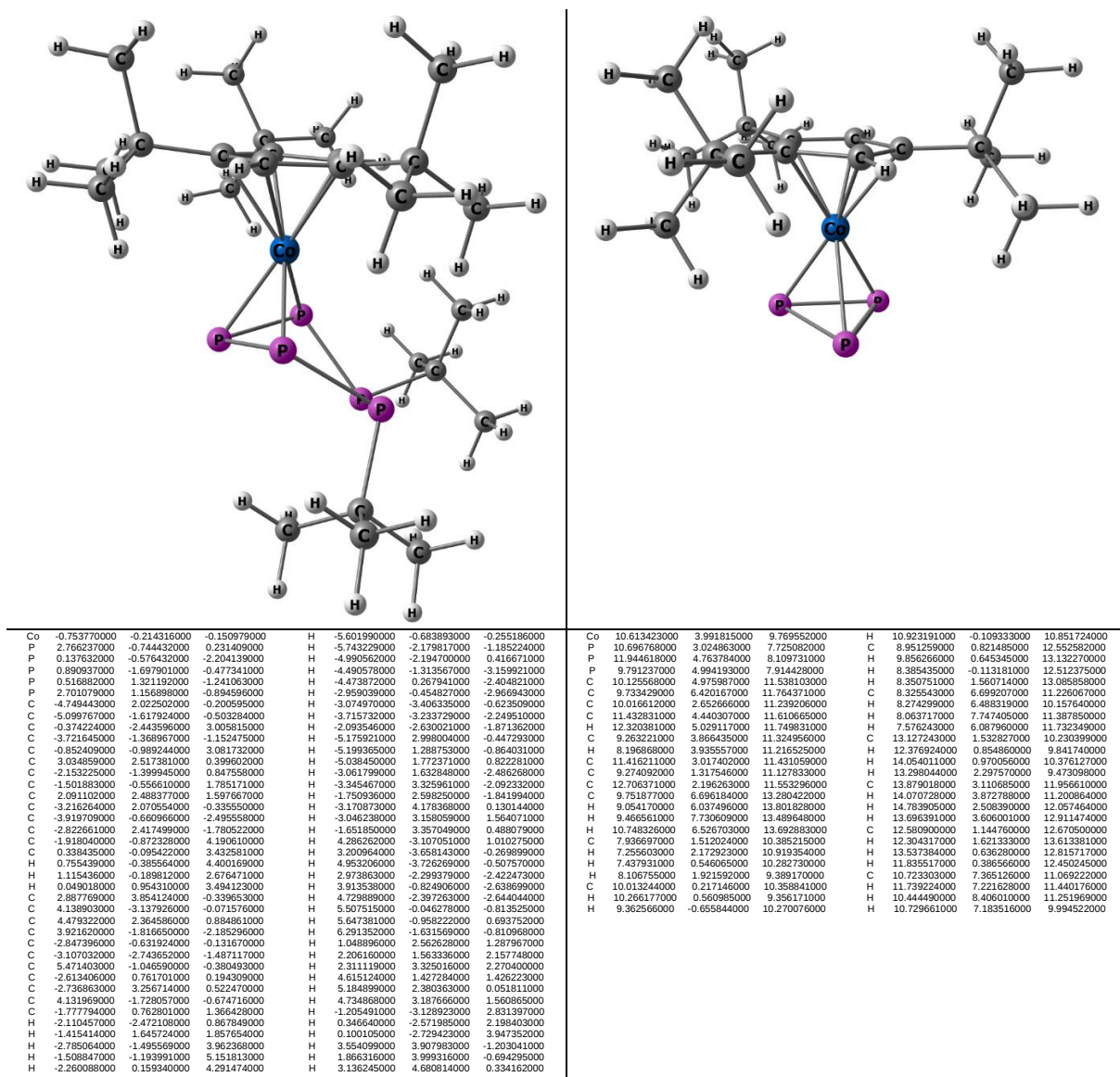
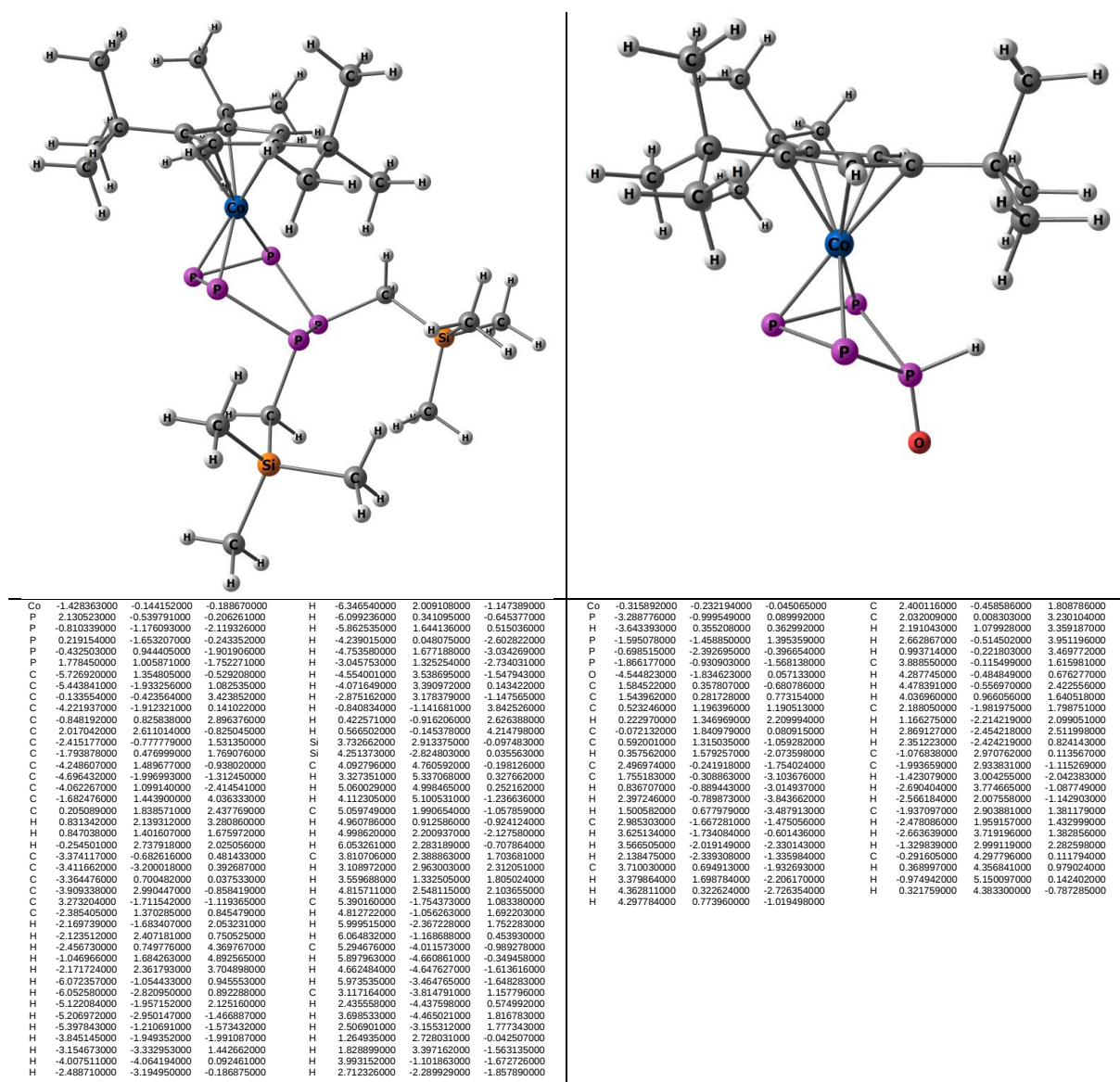


Table S19: Optimized geometries of **1-8** (left), **10** (right). XYZ coordinated in angstroms. B3LYP/def2-TZVP level of theory.



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