

Supplementary Information

Unravelling the intricate photophysical behavior of 3-(pyridin-2-yl)triimidazotriazine AIE and RTP polymorphs

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1. Experimental details

1.1 General information

All reagents and model molecules were purchased from chemical suppliers and used without further purification unless otherwise stated. 3-bromotriimidazo[1,2-*a*:1',2'-*c*:1'',2''-*e*][1,3,5]triazine (TT-Br) was prepared according to literature procedures [S1].

¹H, ¹³C and ¹⁵N NMR spectra were recorded on a Bruker AVANCE-400 instrument (400 MHz). Chemical shifts are reported in parts per million (ppm) and are referenced to the residual solvent peak (DMSO, ¹H 2.50 ppm, ¹³C 39.50 ppm) and to the NH₃ for ¹⁵N resonances. Coupling constants (J) are given in hertz (Hz) and are quoted to the nearest 0.5 Hz. Peak multiplicities are described in the following way: s, singlet; d, doublet; t, triplet; m, multiplet. Mass spectra were recorded on a Thermo Fisher LCQ Fleet Ion Trap Mass Spectrometer equipped with UltiMate™ 3000 HPLC system.

UV-Visible spectra were collected by a Shimadzu UV3600 spectrophotometer. Photoluminescence quantum yields were measured using a C11347 Quantaaurus–Absolute Photoluminescence Quantum Yield Spectrometer (Hamamatsu Photonics K.K), equipped with a 150 W Xenon lamp, an integrating sphere and a multichannel detector. Steady state emission and excitation spectra and photoluminescence lifetimes were obtained using both a FLS 980 (Edinburg Instrument Ltd) and a Nanolog (Horiba Scientific) spectrofluorimeter composed of a iH320 spectrograph equipped with a Synapse QExtra charge-coupled device. The steady state measurements were recorded by excitation with a monochromated 450 W Xenon arc lamp and the spectra are corrected for the instrument response. Phosphorescence spectra are obtained with a PPD-850 single photon detector module with time-gated separation by exciting with a pulsed Xe lamp. Photoluminescence lifetime measurements were performed using: Edinburgh Picosecond Pulsed Diode Laser EPL-375, EPLED-300, (Edinburg Instrument Ltd) and microsecond flash Xe-lamp (60W, 0.1÷100 Hz) with data acquisition devices time correlated single-photon counting (TCSPC) and multi-channel scaling (MCS) methods, respectively. Nanolog TCSPC measurements are performed using DeltaTime series DD-300 DeltaDiode and a DD-405L DeltaDiode Laser, with a PPD-850 single photon detector module and

are analysed with the instrument software DAS6. Average lifetimes are obtained as $\tau_{av} = \frac{\sum A_i \tau_i^2}{\sum A_i \tau_i}$ from bi-exponential or three-exponential fits. Low temperature measurements are performed by immersion of the sample in a LN₂ quartz dewar or with a variable temperature liquid nitrogen cryostat Oxford DN1704. Microscopy fluorescence images were collected with a Nikon Eclipse TE2000-U inverted confocal microscope by exciting with a 100 W Hg lamp with a 330–380 nm band-pass excitation filter

OLED devices were fabricated with the architecture Indium tin oxide (ITO)/poly-(3,4-ethylenedioxythiophene)-poly-(styrenesulfonic acid) (PEDOT:PSS, 35 nm)/polyvinylcarbazole (PVK) 67wt. %:2-(4-tert-Butylphenyl)-5-(4-biphenyl)-1,3,4-oxadiazole (PBD) 23wt. %:TT-Py 10wt. %/2,2',2''-(1,3,5-Benzinetriyl)-tris(1-phenyl-1-H-benzimidazole (TPBI)/Barium(4nm)/Aluminum(100nm). The active layer film (80 nm) was deposited from 15 mg/ml CH₂Cl₂ solution by spin-coating at 3000 rpm in glove box. TPBI (20nm) was thermally evaporated in high vacuum. The device active area is 0.054mm². Photons emitted in forward direction through the glass substrate were collected by a calibrated photodiode. Current density-Luminance-voltage curves are recorded by Keithley 2602 apparatus.

1.2 *Single crystal X-ray crystallographic studies*

X-ray data of the three polymorphs of **TT-Py** were collected on a Bruker Apex II diffractometer using MoK α radiation. The structures were solved using direct methods and refined using a full-matrix least squares procedure based on F^2 using all data [S2]. Hydrogen atoms were placed at geometrically estimated positions. Details relating to the crystals and the structural refinements are presented in Table S1. The structure of **TT-Py-H** includes disordered cocrystallized water molecules. They are characterized by high mobility with three preferential sites in the asymmetric unit (denoted as O1W, O2W and O3W) for a total occupation factor 1.70. They are not self-consistent (owing to too short O1W...O1W', O2W...O2W' and O3W...O3W' distances) but form several possible water chains connecting different oxygen atoms along the *a*-direction, with intermolecular distances compatible with the formation of O-H...O hydrogen bonds between them. This indicates the presence of different and alternating chains of water molecules in the crystal structure, all of them connecting laterally organic molecules through C-H...O and O-H...N hydrogen bonds. Owing to the large disorder of the water molecules it is not possible to include the water hydrogen atoms in the refinement. Full details of crystal data and structure refinement, in CIF format, are available as Supplementary Information. CCDC reference numbers: 1998787 (**TT-Py-A**), 1998788 (**TT-Py-X**) and 1998789 (**TT-Py-H**).

1.3 *Computational details*

DFT and TDDFT calculations on monomeric **TT-Py** were performed with Gaussian 16 program (Revision A.03) [S3] using the 6-311++G(d,p) basis set. The geometries of **TT-Py** and its dimers have been optimized starting from the experimental structures as derived from X-ray studies. Based on previous theoretical results as obtained on the parent cyclic triimidazole and its mono-iodo and mono-, di- and tribromoderivatives, the ω B97X [S4] functional was used owing to its ability in correctly treating at the same time not only ground and excited states properties, but also intermolecular interactions. In fact, the PBE0 and CAM-B3LYP functionals were previously found to accurately reproduce the absorption spectrum of the isolated monomers but failed to provide stable π - π stacked dimers. On the other side, the B97D functional, while providing stable π - π dimers, was found to be unstable for TDDFT calculations.

1.4 *Ultrafast photophysical characterization*

Exc 290 nm: Ultrafast pump-probe measurements were carried out on a pump-probe setup fed by a Ti:sapphire laser (780-nm central wavelength, 100-fs pulses, 1-kHz repetition rate; Libra, Coherent). A fraction of the fundamental wavelength (FW) pulse was used to seed a home-built visible non-collinear optical parametric amplifier (NOPA). The broadband output centered at 580 nm was up-converted by second-harmonic generation in a 20- μ m-thick BBO crystal, providing sub-20-fs pump pulses in the UV range (290 nm). The fluence used was of ~ 0.7 mJ/cm² on the sample. The probe pulses were obtained by focusing a $\sim 1\mu$ J fraction of the FW onto a 3-mm-thick calcium fluoride plate. Through white-light continuum generation we obtained a broadband probe pulse in the range between 320 and 650 nm. The transmitted probe beam was dispersed in a spectrometer (SP2300 Acton, Princeton Instruments) and detected using a linear image sensor driven and read out by a custom-built board (Stresing Entwicklungsbüro, Berlin, Germany).

Exc 390 nm: A Ti:sapphire regenerative amplifier (780-nm central wavelength, 100-fs pulses, 2-kHz repetition rate; Libra, Coherent). As excitation pulses, the second harmonic of the FW has been used ($\lambda = 390$ nm). The excitation density was kept ~ 9 mJ/cm² on the sample. White light generated with

a 2 mm-thick sapphire plate was used as a probe in the visible-near infrared range from 450 to 750 nm. For a spectrally-resolved detection of the probe light, spectrographs and CCD arrays were used. The chirp in the white light pulse was taken into account during the analysis and evaluation of the two-dimensional (wavelength and time) $\Delta T/T$ maps before extraction of the spectral and temporal data using homemade software. Overall, a temporal resolution of at least 100 fs was achieved for all excitation wavelengths.

1.5 Synthesis of 3-(pyridin-2-yl)triimidazo[1,2-*a*:1',2'-*c*:1'',2''-*e*][1,3,5]triazine (TT-Py)

TT-Py is obtained by Stille coupling between 3-bromotriimidazo[1,2-*a*:1',2'-*c*:1'',2''-*e*][1,3,5]triazine (**TT-Br**) and 2-(tributylstannyl)pyridine (Scheme 1). The reaction is performed in a closed cylindrical Pyrex flask (diameter 30 mm, length 300 mm) fitted with a central neck bearing a side neck equipped with a Rotaflo stopcock (previously described for the reductive carbonylation of silica-supported metal chlorides) [S5]. **TT-Br** (500 mg, 1.80 mmol), LiCl (685 mg, 16.16 mmol) and Pd(PPh₃)₂Cl₂ (38 mg, 0.054 mmol) are transferred inside the cylindrical flask, to this mixture 2-(tributylstannyl)pyridine (assay 85%, 1 mL, 2.70 mmol) and anhydrous toluene (15 mL) are added and three freeze-pump-thaw cycles are performed. The mixture is heated at 393 K under static nitrogen for 20 hours. After cooling to room temperature, the reaction mixture is added with NaOH 1M (20 mL) and stirred for 15 minutes. The biphasic solution is diluted with AcOEt (60 mL) and H₂O (40 mL) and separated. The aqueous phase is extracted with an additional 3×10 mL of AcOEt, and the combined organic phases are dried over Na₂SO₄, filtered, and evaporated to dryness. The crude product is purified by column chromatography on silica gel (CH₂Cl₂:MeOH = 95:5; R_f = 0.38) to give **TT-Py** (370 mg, yield 75%). Single crystals of the three different polymorphs are obtained as laminae (**TT-Py-A**), needles (**TT-Py-H**) and rectangular blocks (**TT-Py-X**) by slow evaporation of CH₂Cl₂/CH₃OH, CH₃CN/H₂O and CH₃CN solutions, respectively.

NMR data (9.4 T, DMSO-*d*₆, 298 K, δ , ppm): ¹H NMR 8.65 (ddd, *J* = 4.8, 1.7, 0.9 Hz, 1H), 8.10 (dt, *J* = 8.0, 0.9 Hz, 1H), 8.0 (d, *J* = 1.7 Hz, 1H), 7.98 (d, *J* = 1.6 Hz, 1H), 7.9 (tt, *J* = 5.6, 2.8 Hz, 1H), 7.54 (s, 1H), 7.41 (m, 1H), 7.31 (d, *J* = 1.7 Hz, 1H), 7.22 (d, *J* = 1.6 Hz, 1H); ¹³C NMR: 149.1 (CH), 147.4, 136.8, 135.9 (CH), 135.6, 129.4 (CH), 128.6 (CH), 127.9, 127.8 (CH), 124.8 (CH), 122.9 (CH), 111.8 (CH), 111.4 (CH); ¹⁵N NMR: 308.4, 224.5, 218.8, 218.3, 154.3, 153.1, 149.8 (Fig. S1-S5).

MS (ESI-positive ion mode): *m/z* 276.2 [M+H]⁺ (Fig. S6).

Films of **TT-Py** dispersed in polymethylmethacrylate (PMMA) were prepared by spincoating (2000 rpm, 60 s) a dichloromethane solution (**TT-Py**/PMMA = 5 or 10 wt%; PMMA = 10 wt% with respect to the solvent) on a quartz substrate.

2. NMR and Mass Spectra of TT-Py

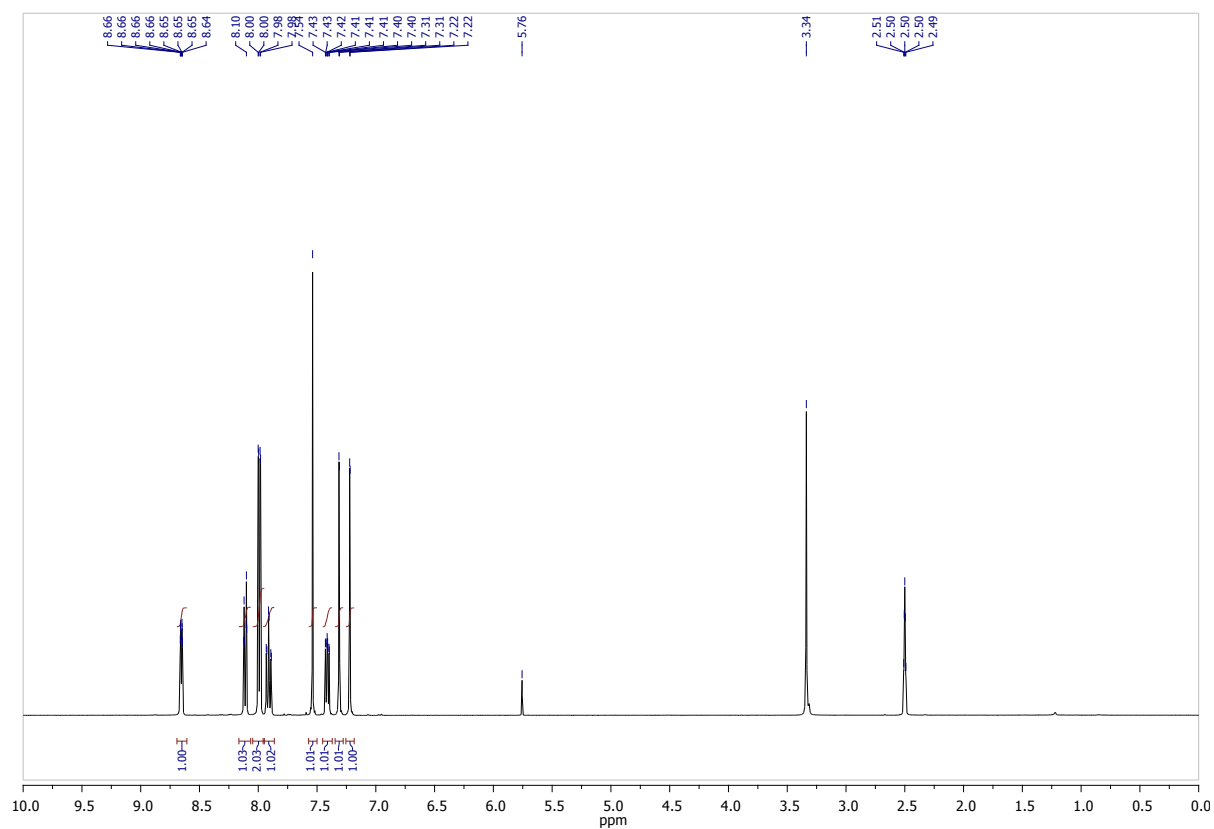


Figure S1. ¹H NMR spectrum of TT-Py (400 MHz, DMSO-d₆, 298 K)

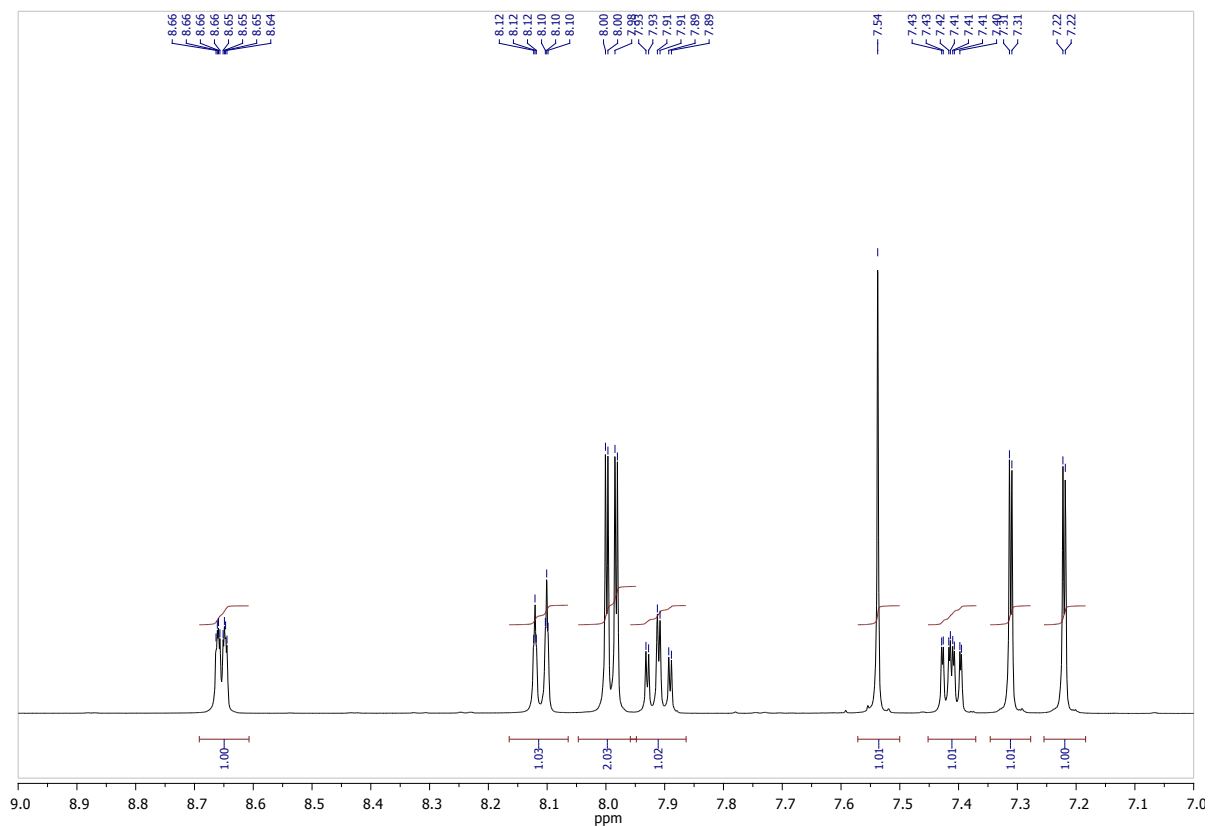


Figure S2. Expanded region of ^1H NMR spectrum of **TT-Py** (400 MHz, DMSO-d_6 , 298 K)

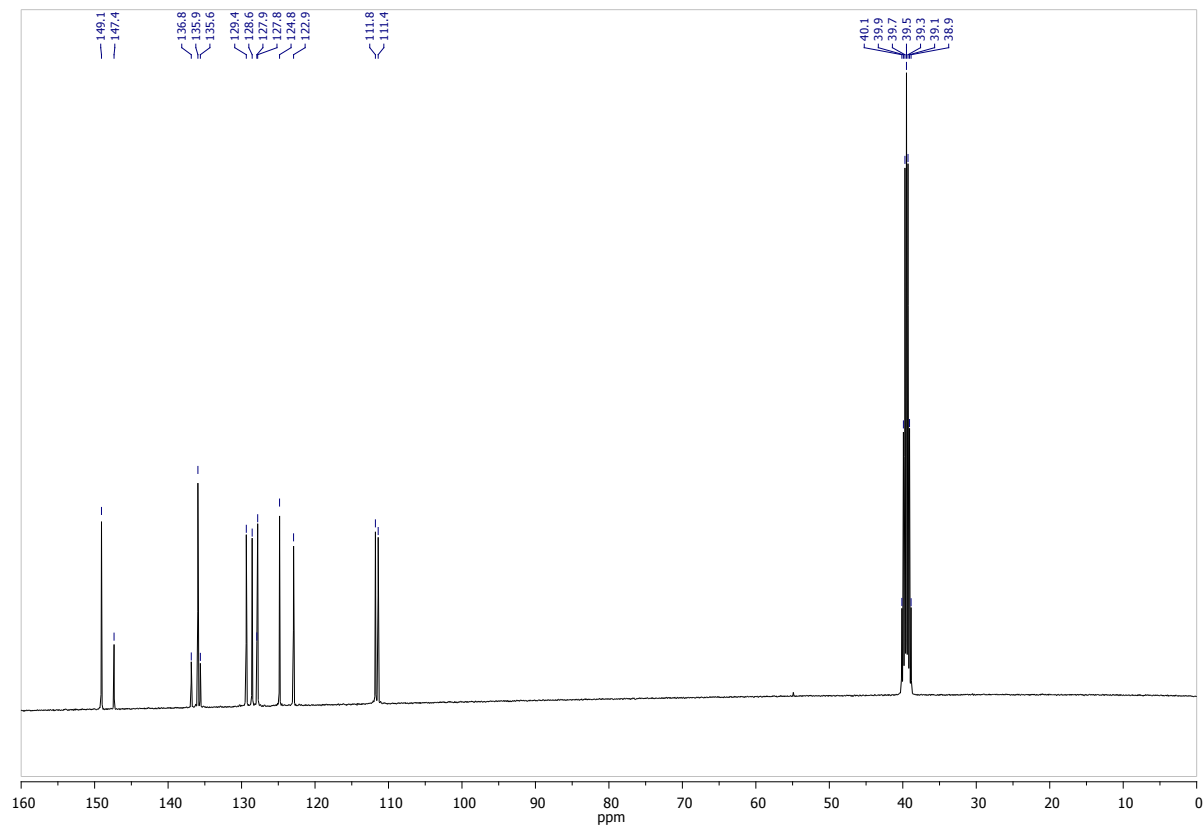


Figure S3. ^{13}C NMR spectrum of **TT-Py** (400 MHz, DMSO-d_6 , 298 K)

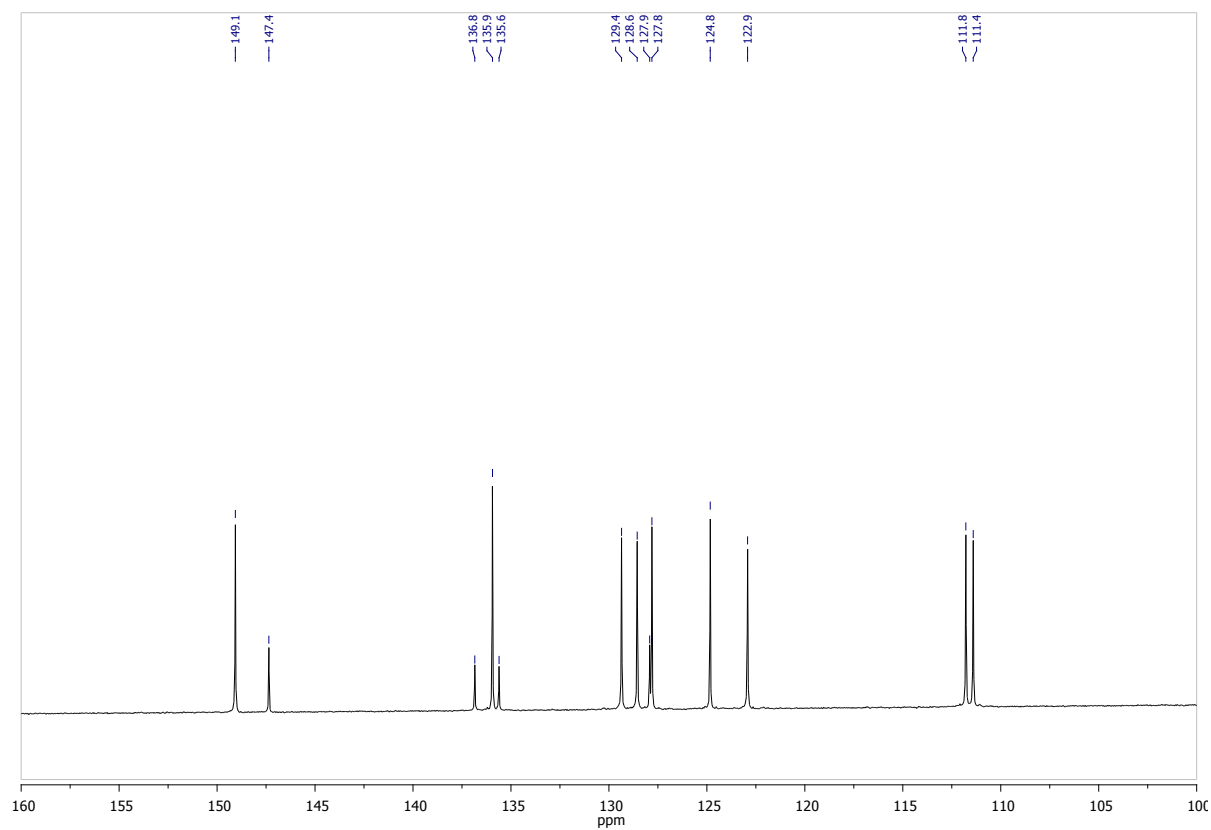


Figure S4. Expanded region of ^{13}C NMR spectrum of **TT-Py** (400 MHz, DMSO-d_6 , 298 K)

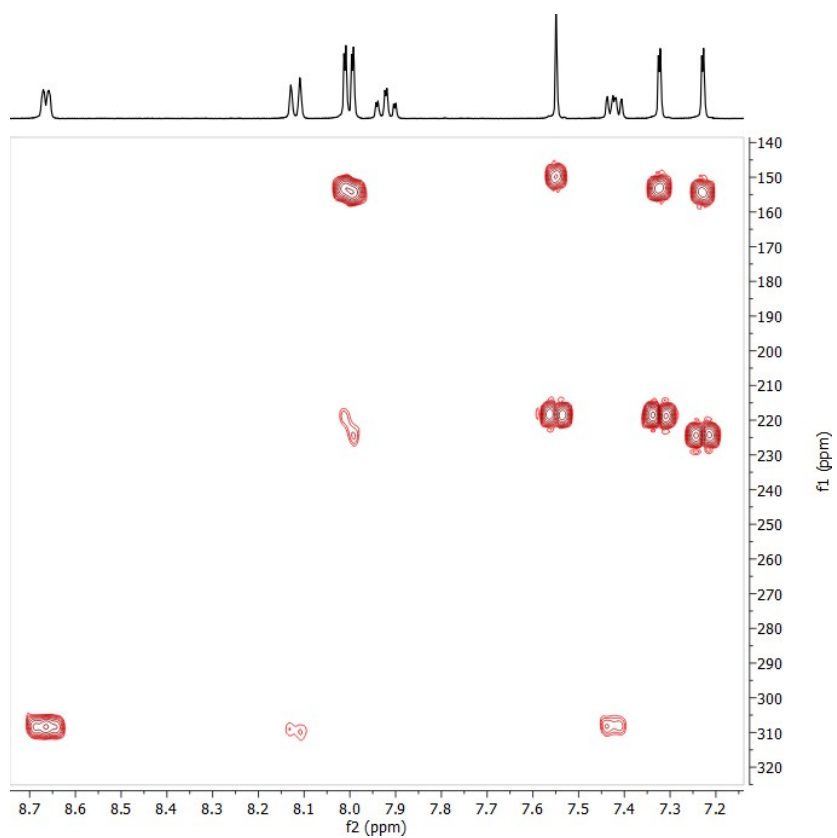
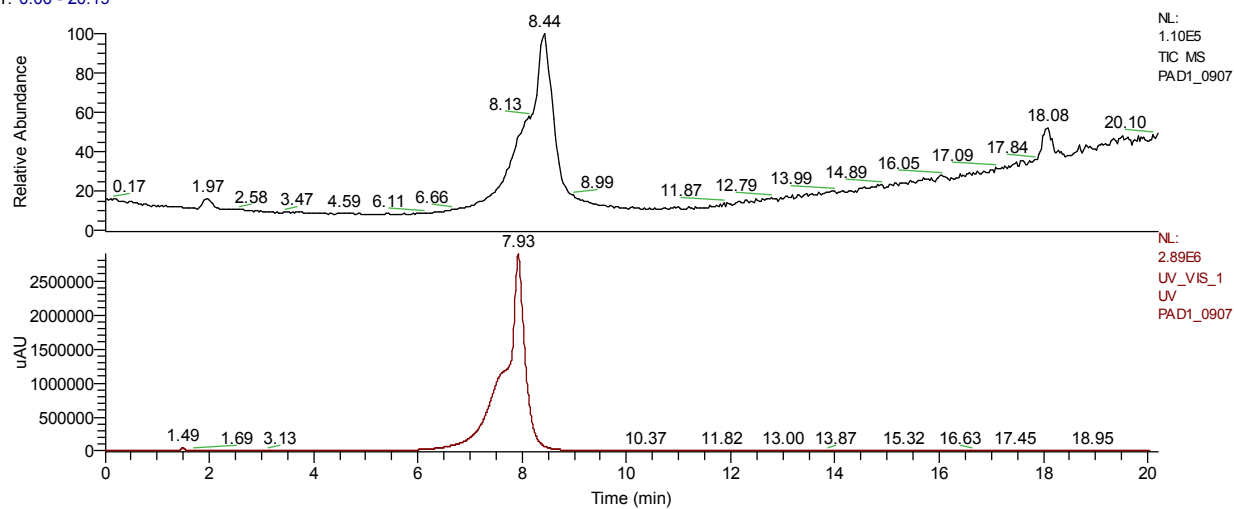


Figure S5. ^{15}N NMR spectrum of **TT-Py** (400 MHz, DMSO-d_6 , 298 K)

RT: 0.00 - 20.19



PAD1_0907 #254 RT: 8.39 AV: 1 NL: 8.05E4
T: ITMS + c ESI Full ms [50.00-2000.00]

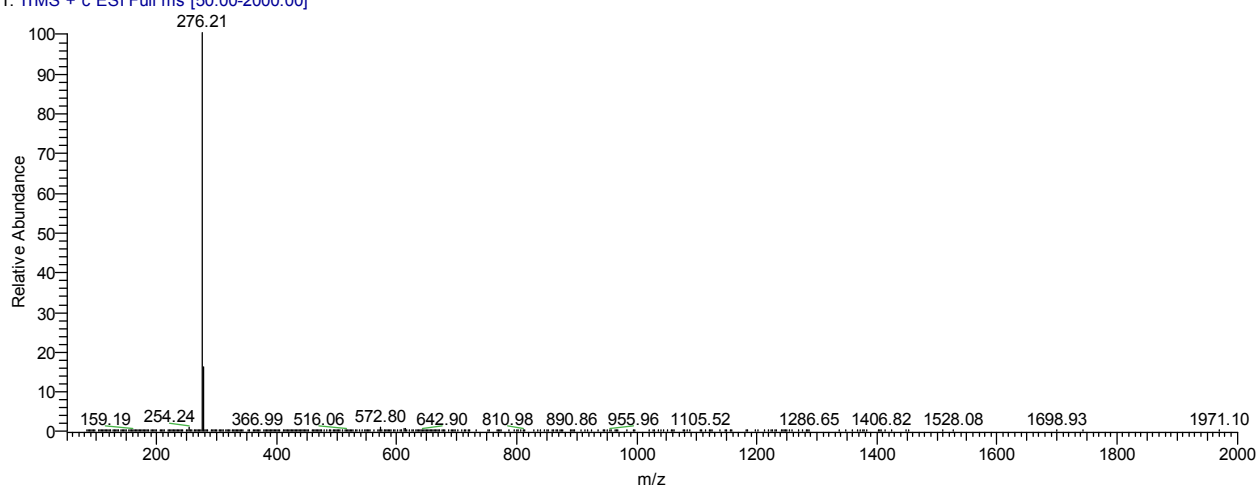


Figure S6. HPLC-MS spectra of TT-Py

3. Photophysical Data and Spectra of TT-Py

3.1 Electronic absorption, Steady State and Time resolved emission Spectra

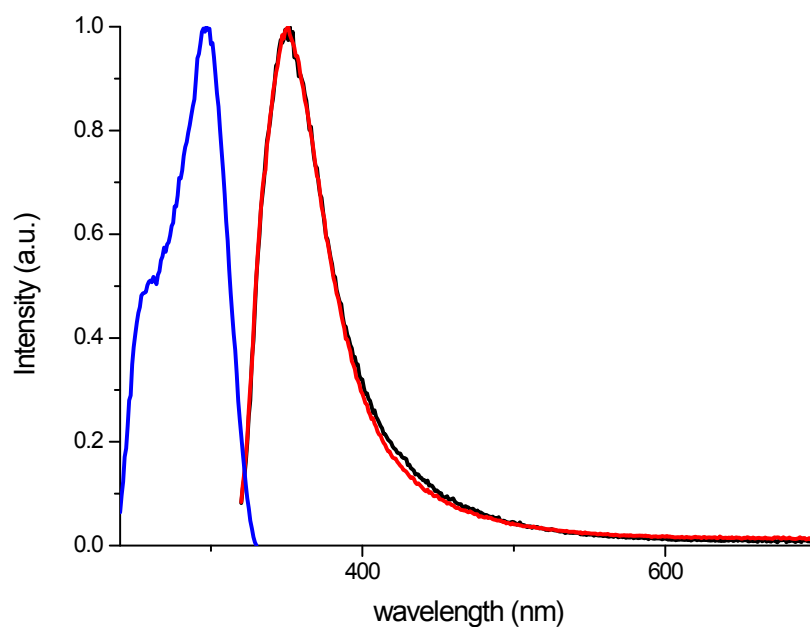


Figure S7. Emission spectra of **TT-Py** in CH_2Cl_2 solution (10^{-5} M) at 298 K: $\lambda_{\text{exc}}=300\text{nm}$ (black line), $\lambda_{\text{exc}}=340\text{nm}$ (red line) and excitation spectra $\lambda_{\text{em}}=350\text{nm}$ (blue line).

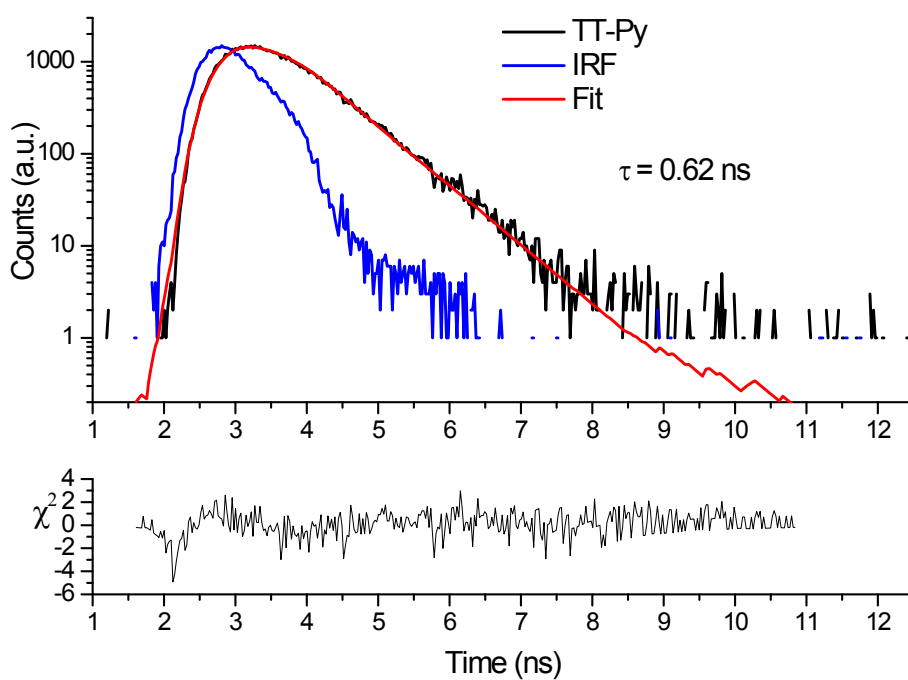


Figure S8. Lifetime measurement ($\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=351\text{nm}$) of **TT Py** in CH_2Cl_2 solution (10^{-5} M) at 298 K.

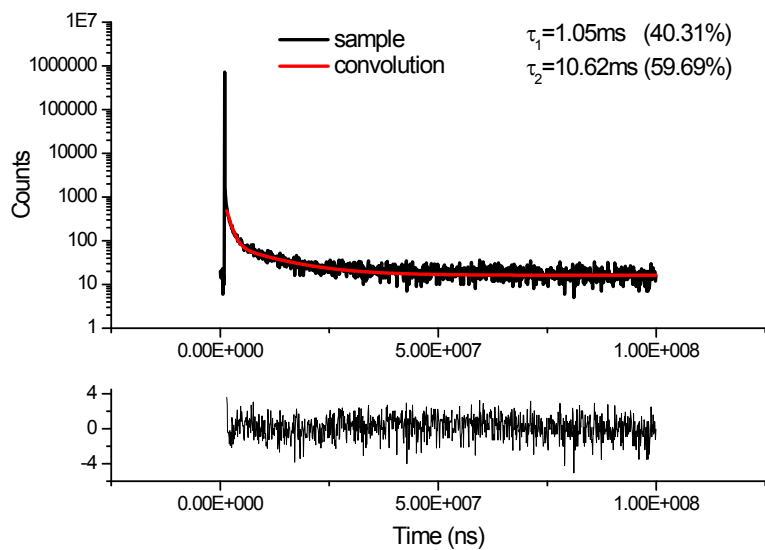


Figure S9. Lifetime measurement ($\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=350\text{nm}$) of **TT-Py** in CH_2Cl_2 solution (10^{-5} M) at 298 K. $\tau_{\text{av}} = 10.02\text{ ms}$

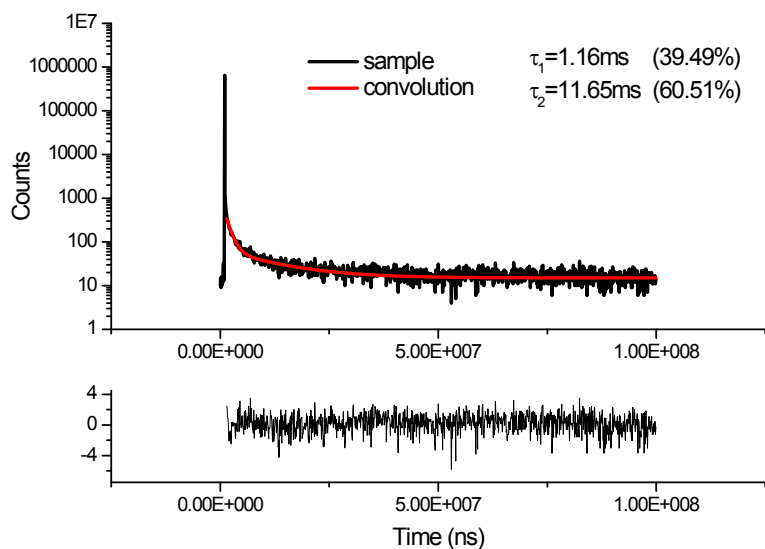


Figure S10. Lifetime measurement ($\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=408\text{nm}$) of **TT-Py** in CH_2Cl_2 solution (10^{-5} M) at 298 K. $\tau_{\text{av}} = 11.01\text{ms}$

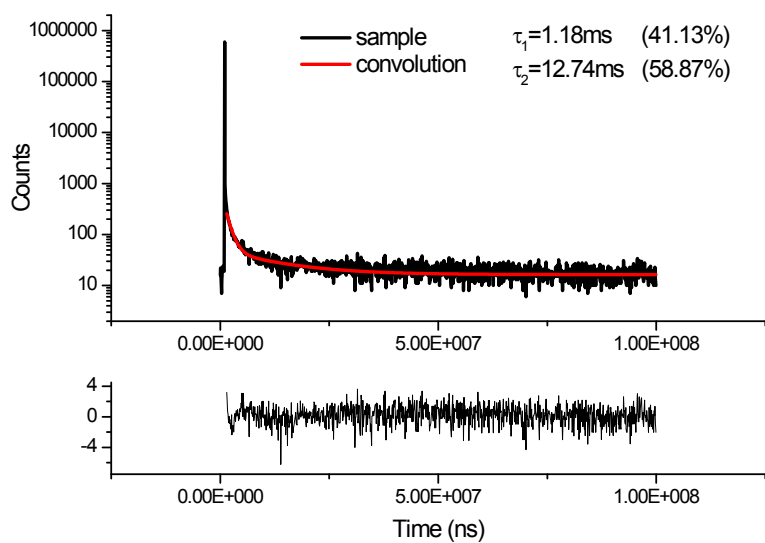


Figure S11. Lifetime measurement ($\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=500\text{nm}$) of **TT-Py** in CH_2Cl_2 solution (10^{-5} M) at 298 K. $\tau_{\text{av}} = 12.04\text{ms}$

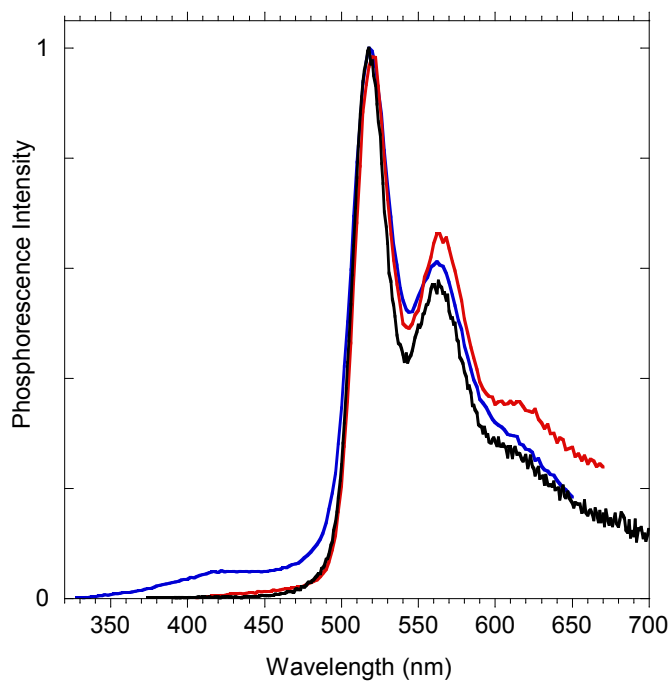


Figure S12. Time evolution of phosphorescence spectra of **TT-Py** (10^{-5}M) in of deaerated CH_2Cl_2 at 298 K, $\lambda_{\text{exc}}=300\text{nm}$ (blue line, delay 50 μs , window 100 μs ; black line, delay 100 μs , window 200 μs ; red line, delay 200 μs , window 5 ms). The spectra are normalized.

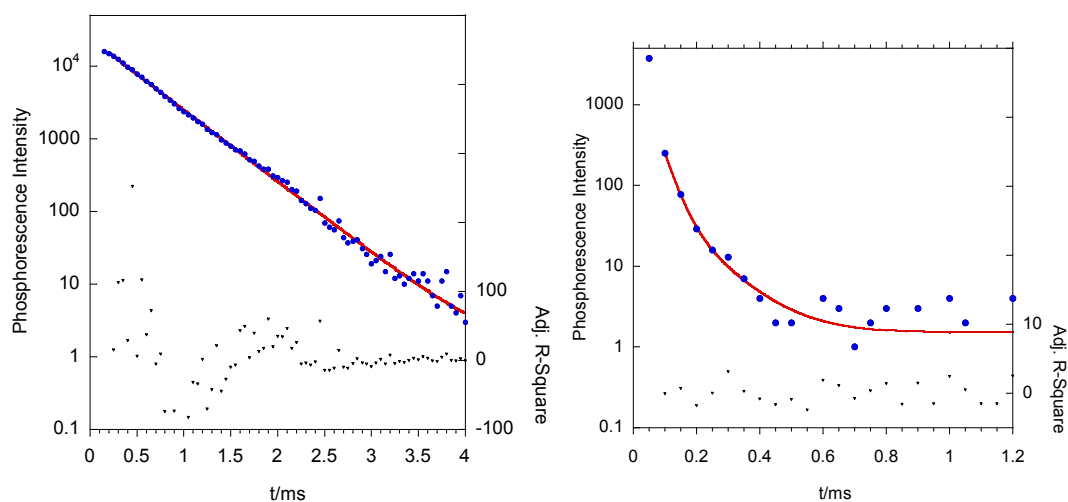


Figure S13. Phosphorescence decays of **TT-Py** (10^{-5}M) in CH_2Cl_2 deaerated solution (blue points) at 298 K, $\lambda_{\text{exc}}=300\text{nm}$. Left, $\lambda_{\text{em}}=520\text{nm}$ (mono-exp fit, red line $\tau=441\mu\text{s}$; Adj. R-Square= 0,99966); Right, $\lambda_{\text{em}}=400\text{nm}$ (bi-exp fit, red line : biexp fit; $\tau_{\text{av}}=40\mu\text{s}$ $A_1=0.03$, $\tau_1=116\mu\text{s}$; $A_2=0.97$, $\tau_2=34\mu\text{s}$, . R-Square= 0,99896)).

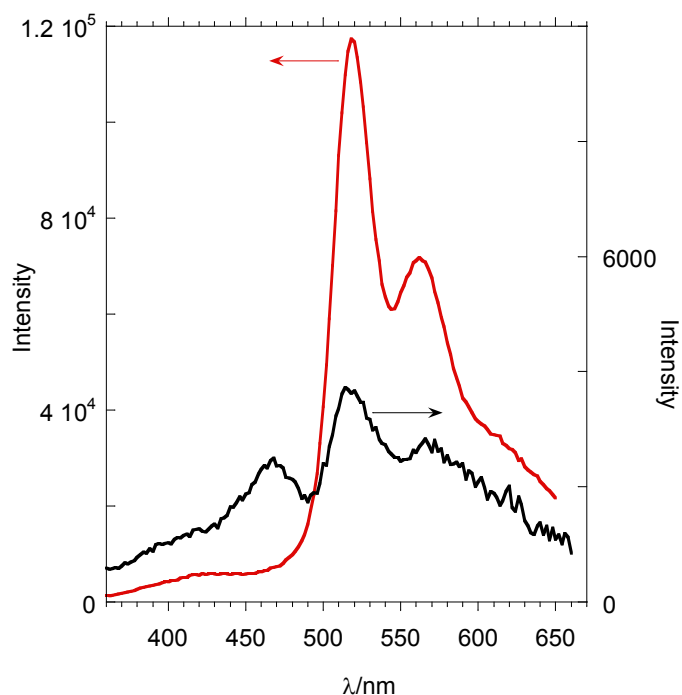


Figure S14. Phosphorescence spectra of **TT-Py** (10^{-5}M) in CH_2Cl_2 deaerated solution before (red) and after (black) air diffusion in the cell, at 298 K. $\lambda_{\text{exc}}=300\text{nm}$; delay $50\mu\text{s}$, window $100\mu\text{s}$.

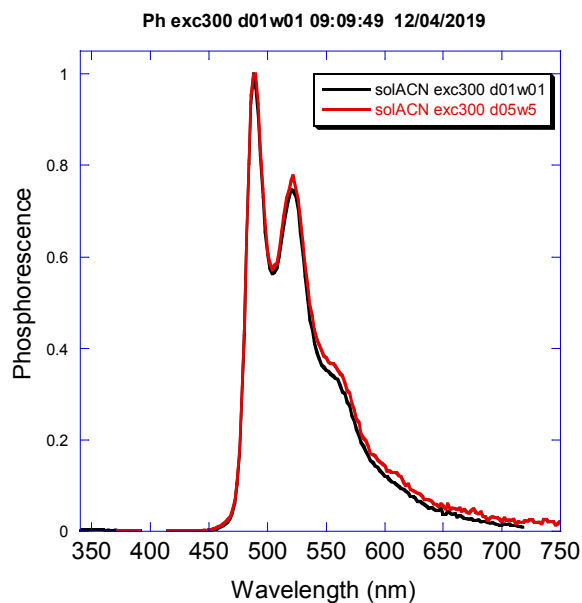


Figure S15. Phosphorescence spectra of **TT-Py** (10^{-5} M) in CH_3CN deaerated solution at 298 K. $\lambda_{\text{exc}}=300\text{nm}$ (black line, delay 100 μs , window 100 μs ; red line, delay 500 μs , window 5ms).

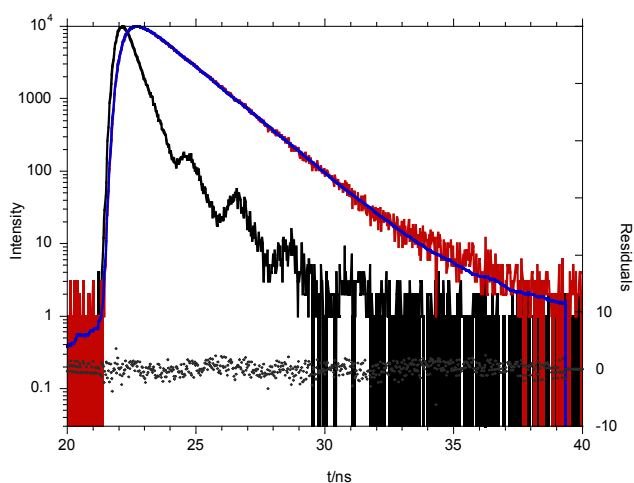


Figure S16 Decay of **TT-Py** (10^{-5} M) in CH_3CN deaerated solution (red line) at 298 K, $\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=350\text{nm}$. Bi-exp fit, blue line: $\tau_{\text{av}}=1.36\text{ns}$, $A_1=0.0815$, $\tau_1=0.6848\text{ns}$; $A_2=0.9185$, $\tau_2=1.3930\text{ns}$, $\text{CHISQ} = 1.084949$). Similar decay is recorded after O_2 diffusion into the cell.

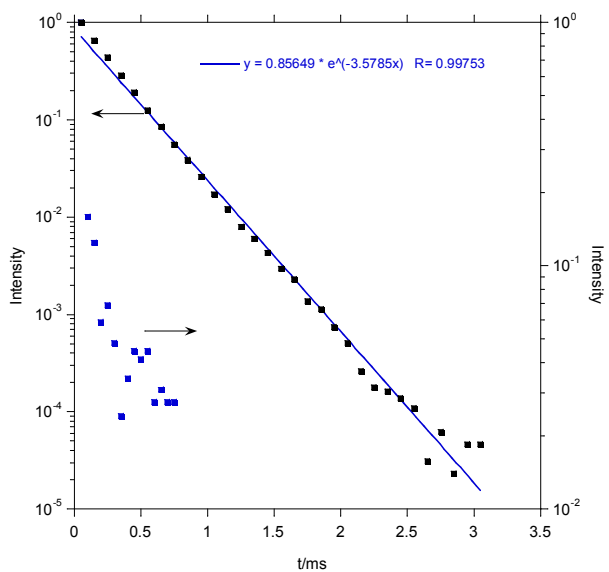


Figure S17 Phosphorescence decay of TT-Py (10⁻⁵M) in CH₃CN deaerated solution (black points) and after O₂ diffusion into the cell (blue points) at 298 K, λ_{exc}=300nm, λ_{em}=490nm. Mono-exp fit (blue line).

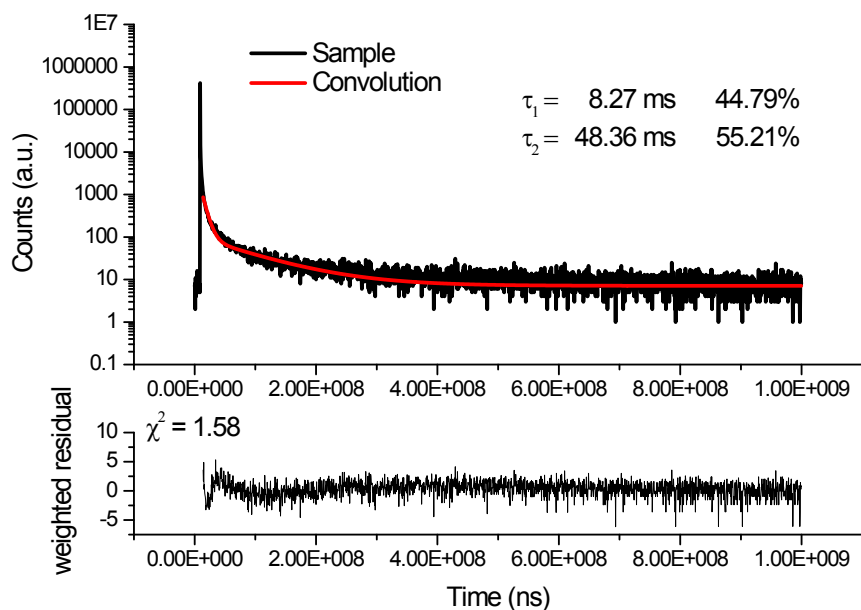


Figure S18. Lifetime measurement (λ_{exc}=300nm, λ_{em}=350nm) of TT-Py in CH₃CN deaerated solution (10⁻⁵ M) at 298 K. τ_{av} = 43.47 ms

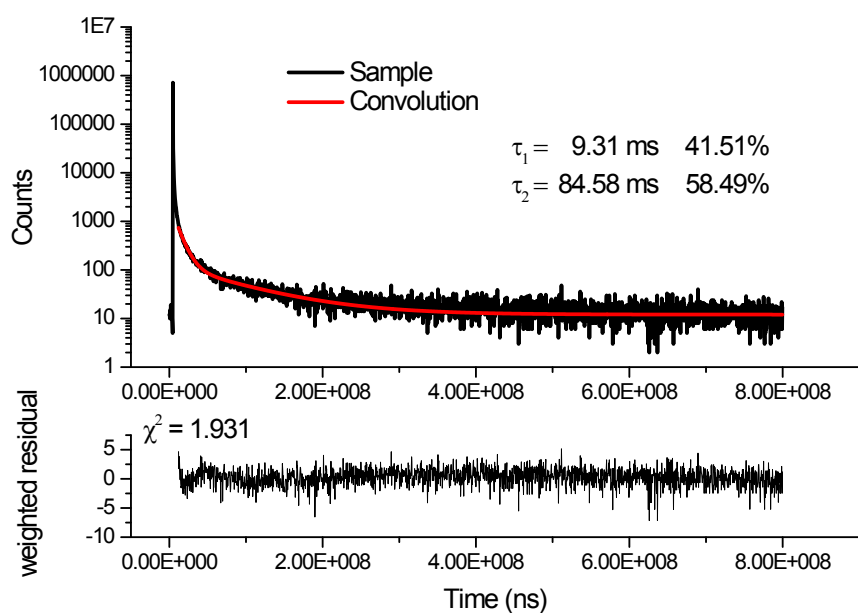


Figure S19. Lifetime measurement ($\lambda_{\text{exc}}=300 \text{ nm}$, $\lambda_{\text{em}}=360 \text{ nm}$) of **TT-Py** (10^{-4}M) in deareated $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ (50/50) solution at 298 K. $\tau_{\text{av}} = 79.13 \text{ ms}$.

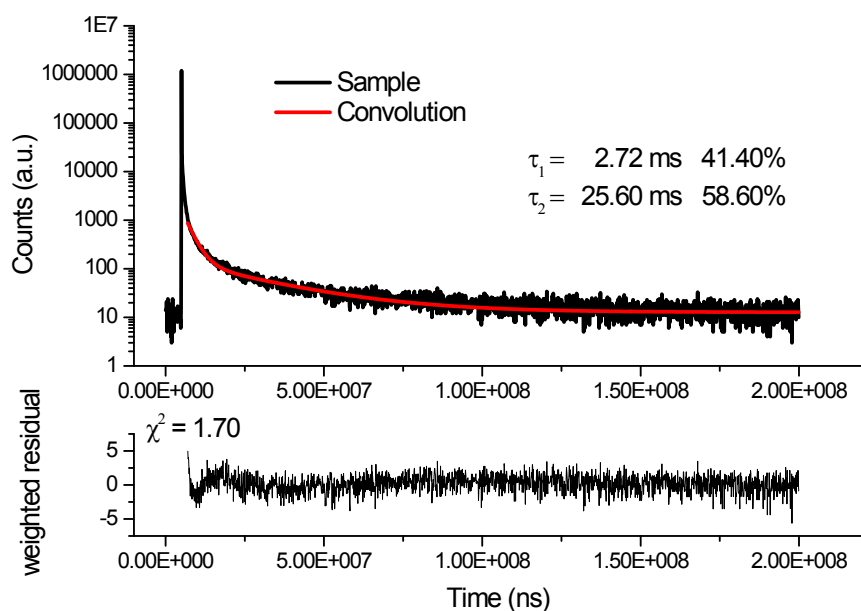


Figure S20. Lifetime measurement ($\lambda_{\text{exc}}=300 \text{ nm}$, $\lambda_{\text{em}}=488 \text{ nm}$) of **TT-Py** (10^{-4}M) in deareated $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ (50/50) solution at 298 K. $\tau_{\text{av}} = 24.00 \text{ ms}$.

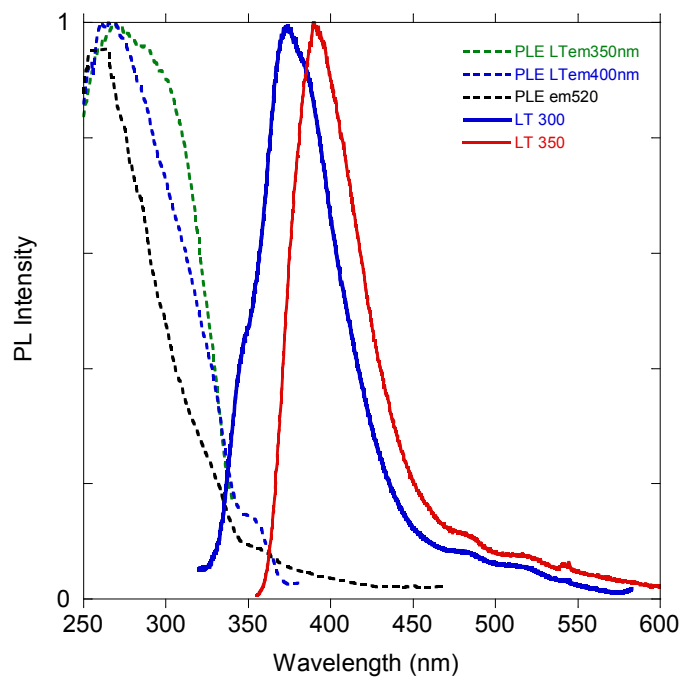


Figure S21. PL (solid line) and PLE (dotted line) of **TT-Py** (10^{-5}M) in CH_2Cl_2 solution at 77 K. $\lambda_{\text{em}}=350\text{nm}$, green line; $\lambda_{\text{em}}=400\text{nm}$, blue line; $\lambda_{\text{em}}=520\text{nm}$, black line; $\lambda_{\text{exc}}=300\text{nm}$, blue line; $\lambda_{\text{exc}}=350\text{nm}$, red line.

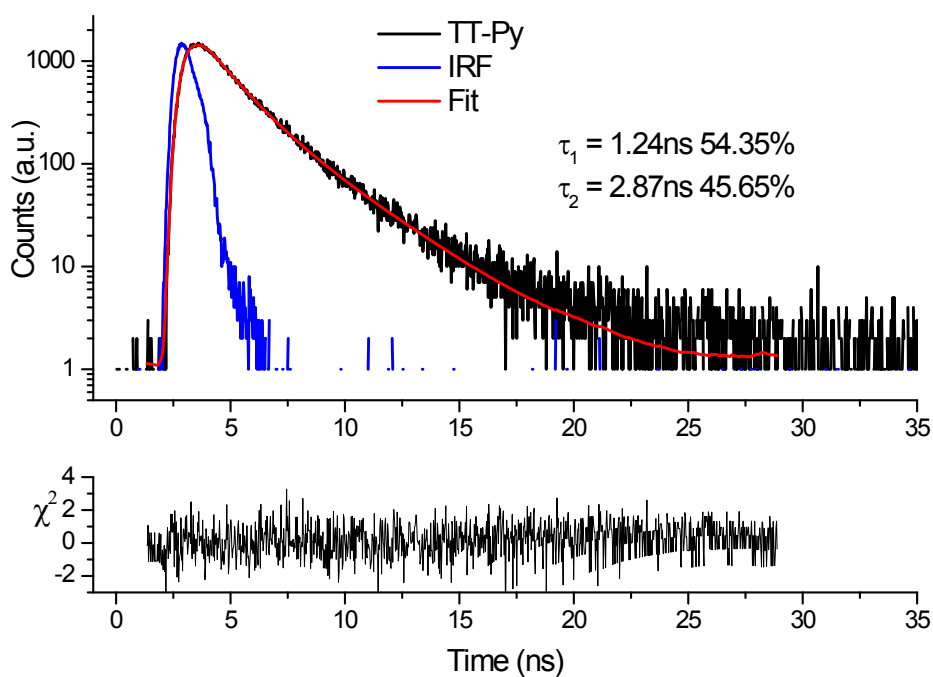


Figure S22. Lifetime measurement ($\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=343\text{nm}$) of **TT-Py** (10^{-5}M) in CH_2Cl_2 solution at 77 K. $\tau_{\text{av}} = 2.32\text{ns}$.

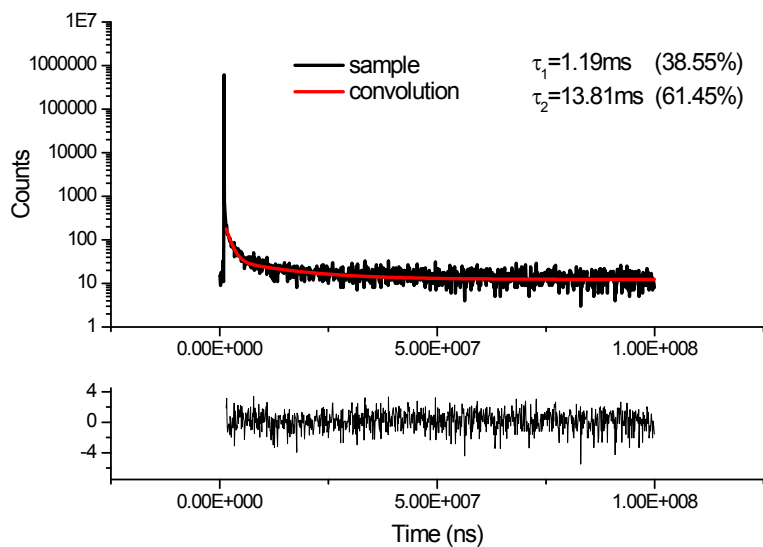


Figure S23. Lifetime measurement ($\lambda_{\text{exc}} = 350\text{ nm}$, $\lambda_{\text{em}} = 388\text{ nm}$) of **TT-Py** (10^{-5}M) in CH_2Cl_2 solution at 77 K. $\tau_{\text{av}} = 13.16\text{ms}$

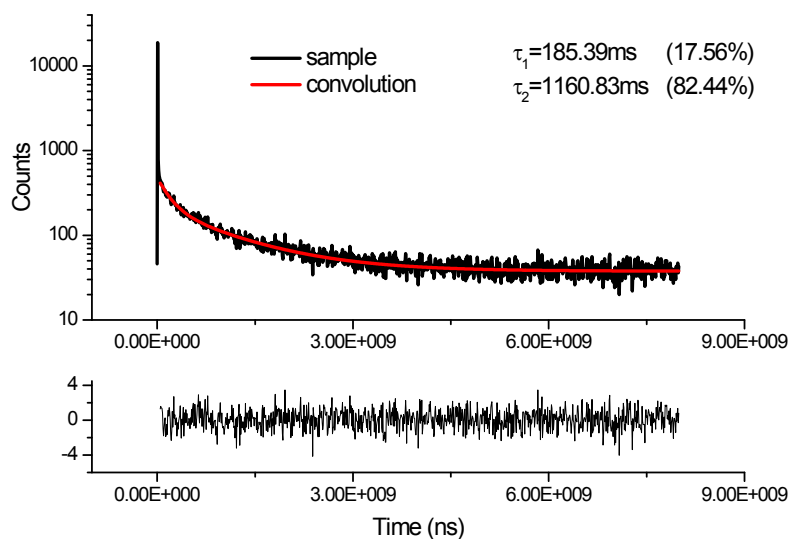


Figure S24. Lifetime measurement ($\lambda_{\text{exc}} = 350\text{ nm}$, $\lambda_{\text{em}} = 550\text{ nm}$) of **TT-Py** (10^{-5}M) in CH_2Cl_2 solution at 77 K. $\tau_{\text{av}} = 1.31\text{ s}$

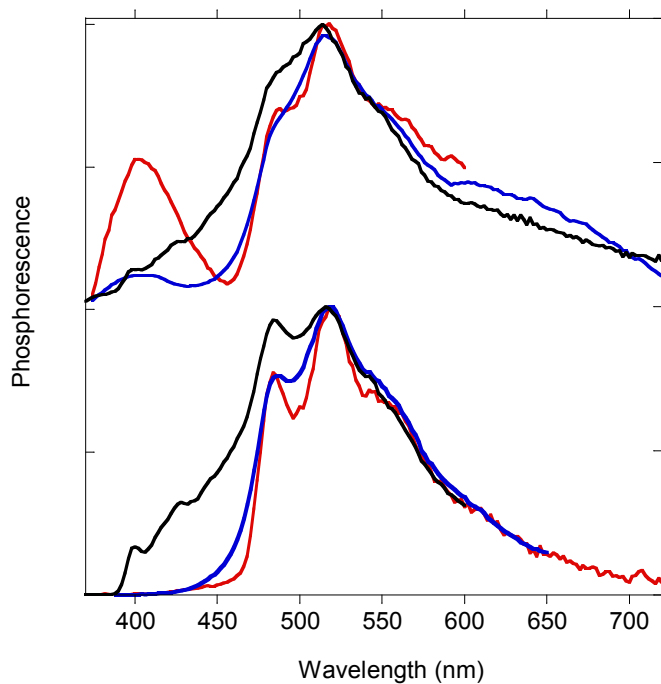


Figure S25. Phosphorescence spectra of **TT-Py** (10^{-5}M) in CH_2Cl_2 solution at 77 K. $\lambda_{\text{exc}}=300\text{nm}$, black line; $\lambda_{\text{exc}}=325\text{nm}$, blue line; $\lambda_{\text{exc}}=350\text{nm}$, red line; Top, delay 200 μs , window 400 μs ; Bottom, delay 10ms, window 20ms.

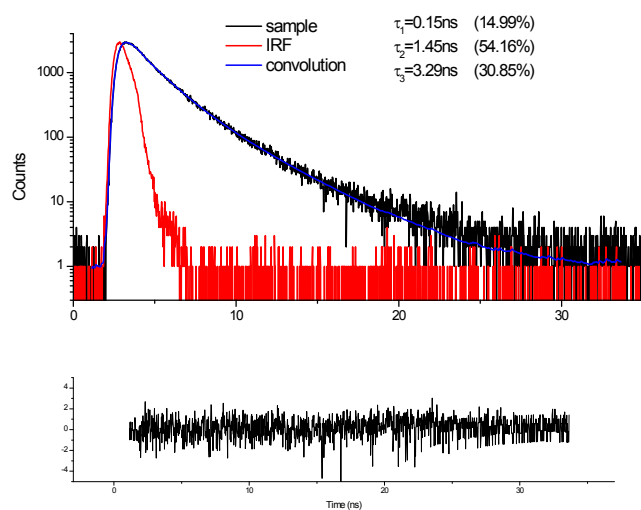


Figure S26. Lifetime measurement ($\lambda_{\text{exc}}=300\text{ nm}$, $\lambda_{\text{em}}=343\text{ nm}$) of **TT-Py** (10^{-5}M) in CH_3CN solution at 77 K. $\tau_{\text{av}} = 2.46\text{ ns}$

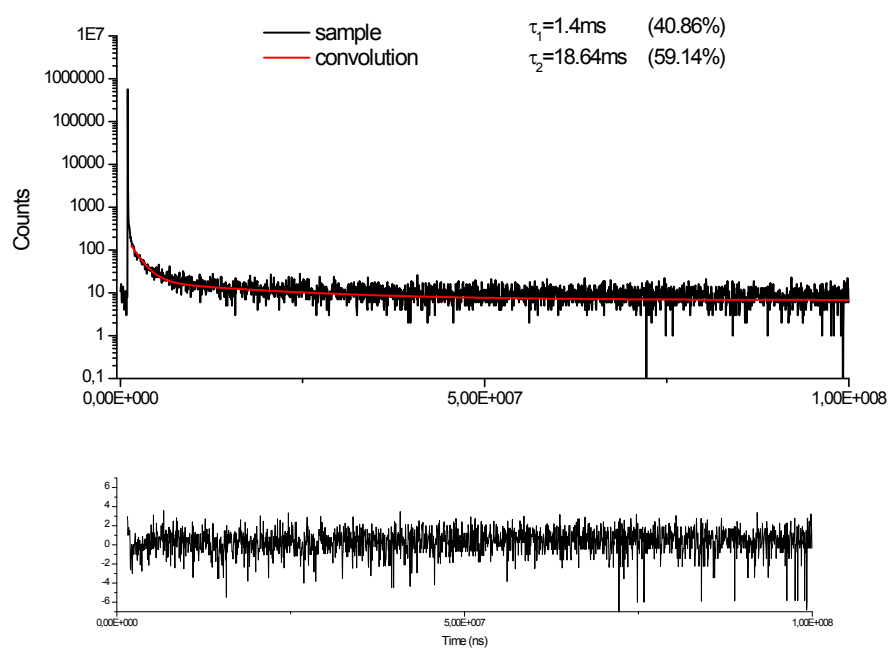


Figure S27. Lifetime measurement ($\lambda_{\text{exc}} = 300 \text{ nm}$, $\lambda_{\text{em}} = 388 \text{ nm}$) of **TT-Py** (10^{-5} M) in CH_3CN solution at 77 K. $\tau_{\text{av}} = 17.79 \text{ ms}$

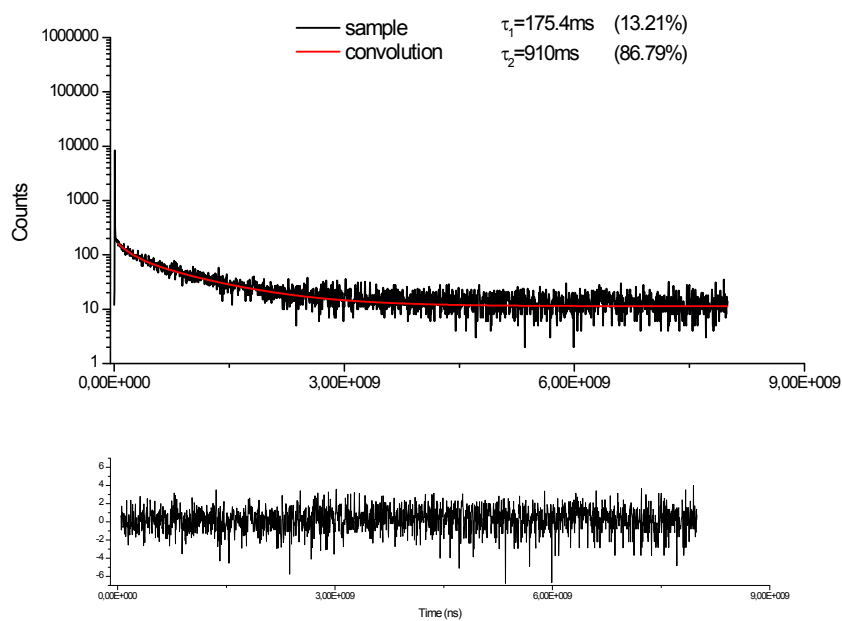


Figure S28. Lifetime measurement ($\lambda_{\text{exc}} = 300 \text{ nm}$, $\lambda_{\text{em}} = 500 \text{ nm}$) of **TT-Py** (10^{-5} M) in CH_3CN solution at 77 K. $\tau_{\text{av}} = 889 \text{ ms}$

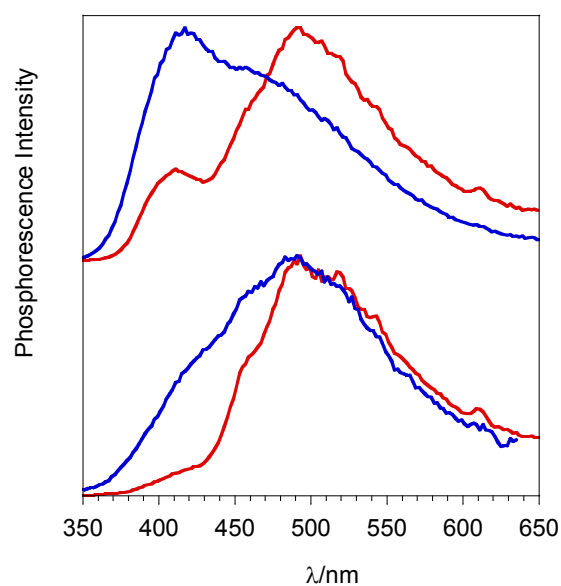


Figure S29. Phosphorescence spectra of **TT-Py** (10^{-5}M) in CH_3CN solution at 77 K. Top, $\lambda_{\text{exc}}=270\text{nm}$,; Bottom, $\lambda_{\text{exc}}=300\text{nm}$ (blue line delay 200 μs , window 500 μs ; red line delay 2ms window 5ms).

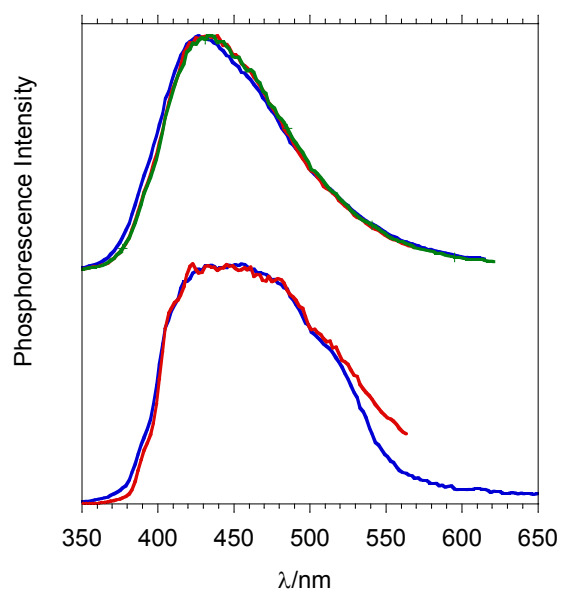


Figure S30. Phosphorescence spectra of **TT-Py** (10^{-5}M) in $\text{CH}_3\text{OH}/\text{CH}_3\text{CH}_2\text{OH}$ ($v/v = 20/80$), solution at 77 K. Top, $\lambda_{\text{exc}}=270\text{nm}$; Bottom, $\lambda_{\text{exc}}=300\text{nm}$; blue line delay 200 μs , window 500 μs ; red line delay 2ms window 5ms; green line delay 20ms, window 10ms.

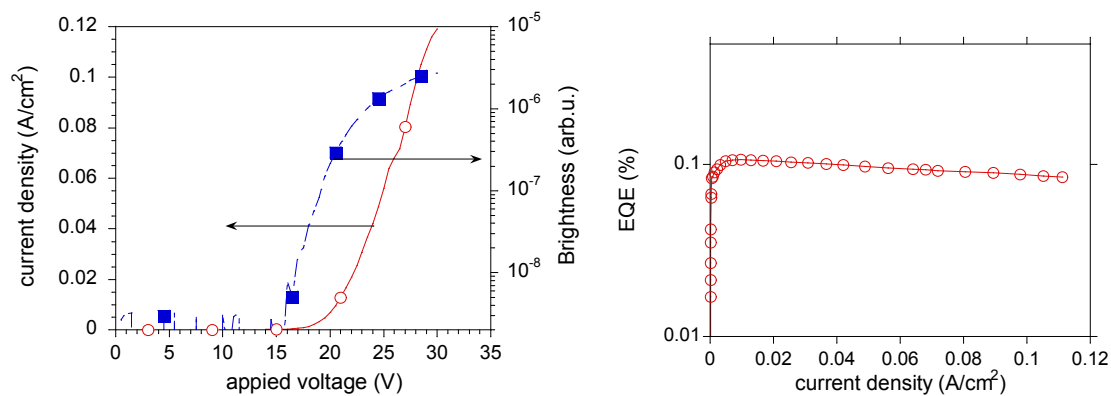


Figure S31. Current-Voltage-Brightness characteristics (left) and external quantum efficiency (right) of a **TT-Py** based OLED with device structure ITO/PEDOT:PSS/PVK:PBD:**TT-Py** (10 wt. %)/TPBI/Ba/Al.

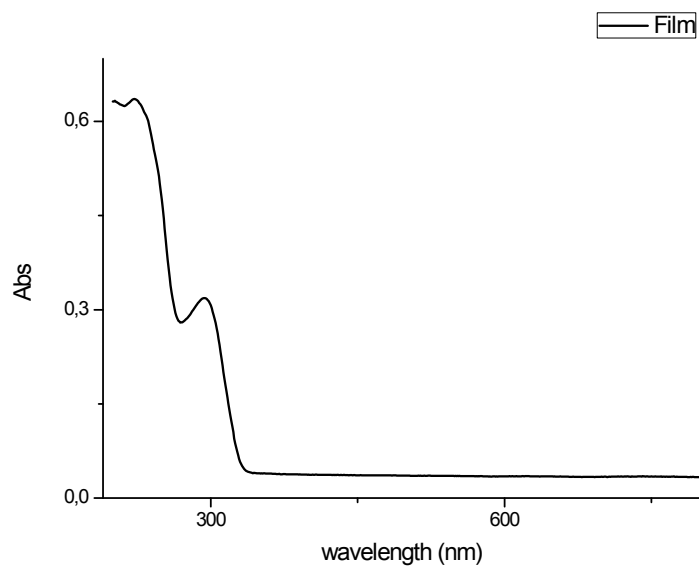


Figure S32. Absorption spectrum of **TT-Py** in PMMA (w/w **TT-Py**/PMMA 10%) at 298 K.

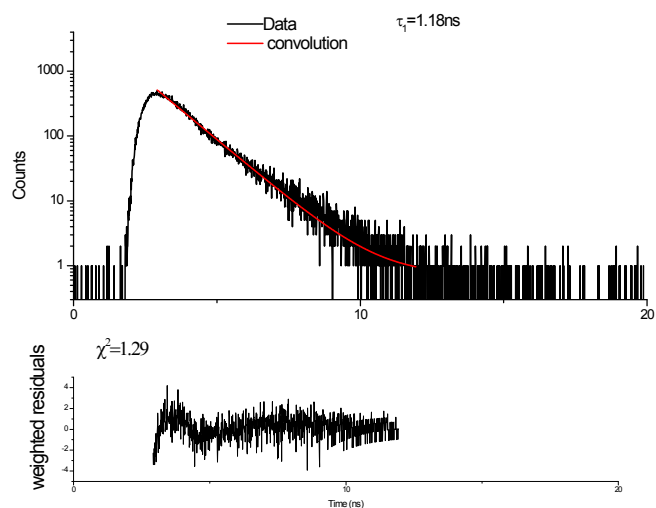


Figure S33. Decay of TT-Py in PMMA (w/w TT-Py/PMMA 10%) at 298 K. $\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=350\text{nm}$

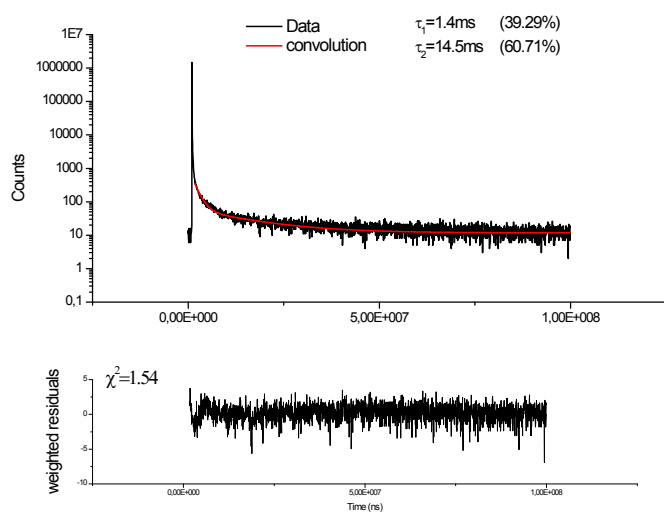


Figure S34. Decay of TT-Py in PMMA (w/w TT-Py/PMMA 10%) at 298 K. $\lambda_{\text{exc}}=350\text{nm}$, $\lambda_{\text{em}}=400\text{nm}$

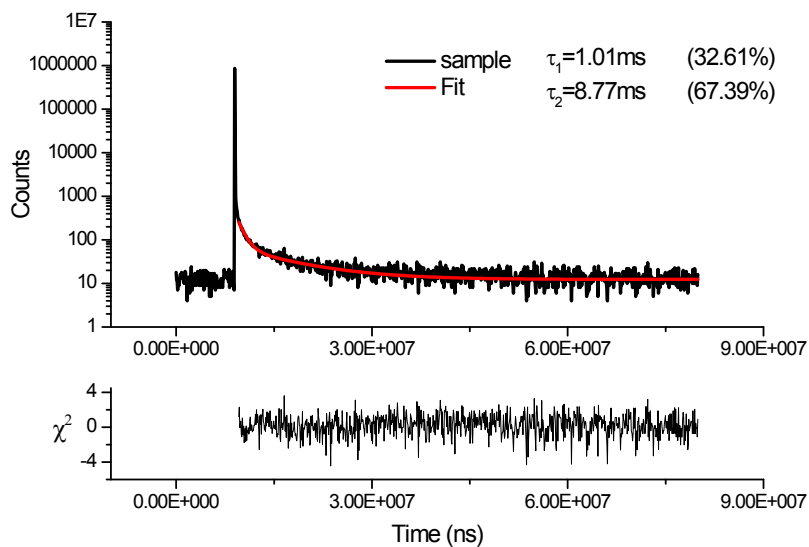


Figure S35-A. TT-Py in PMMA (w/w TT-Py/PMMA 10%) at 298 K. $\lambda_{\text{exc}}=330\text{nm}$; $\lambda_{\text{em}}=450\text{nm}$

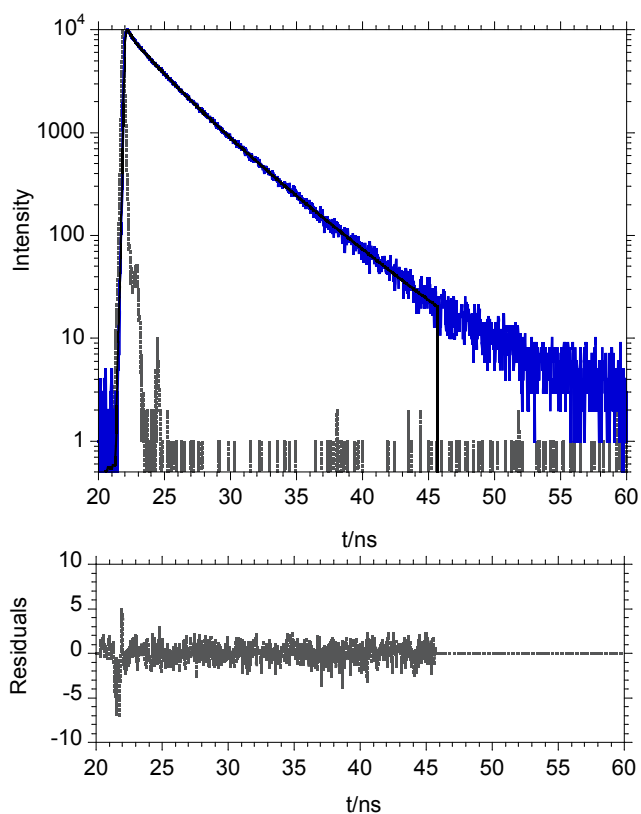


Figure S35-B. Decay of TT-Py in PMMA (w/w TT-Py/PMMA 10%) at 298 K. $\lambda_{\text{exc}}=408\text{nm}$, $\lambda_{\text{em}}=450\text{nm}$. 3-exp fit, black line: $\tau_{\text{av}}=3.47\text{ns}$ $A1=0.4402$, $\tau_1=2.4769\text{ns}$; $A2=0.2345$, $\tau_2=0.330661\text{ns}$, $A3=0.3252$, $\tau_3=4.39950\text{ns}$, CHISQ = 1.1313259)

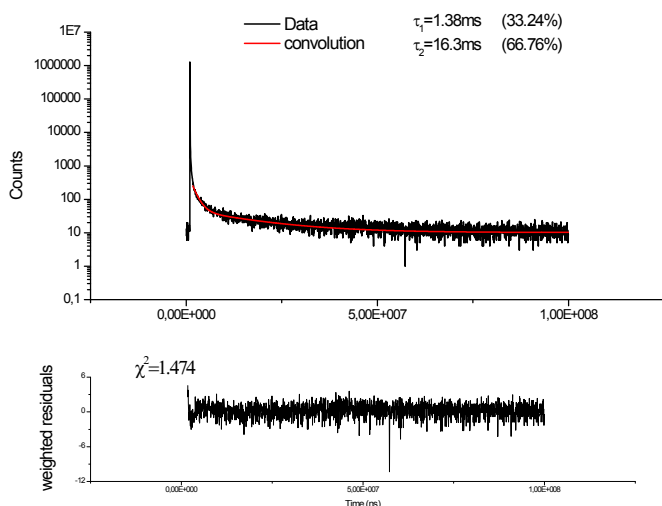


Figure S36. Decay of TT-Py in PMMA (w/w TT-Py/PMMA 10%) at 298 K. $\lambda_{\text{exc}}=400\text{nm}$, $\lambda_{\text{em}}=500\text{nm}$.

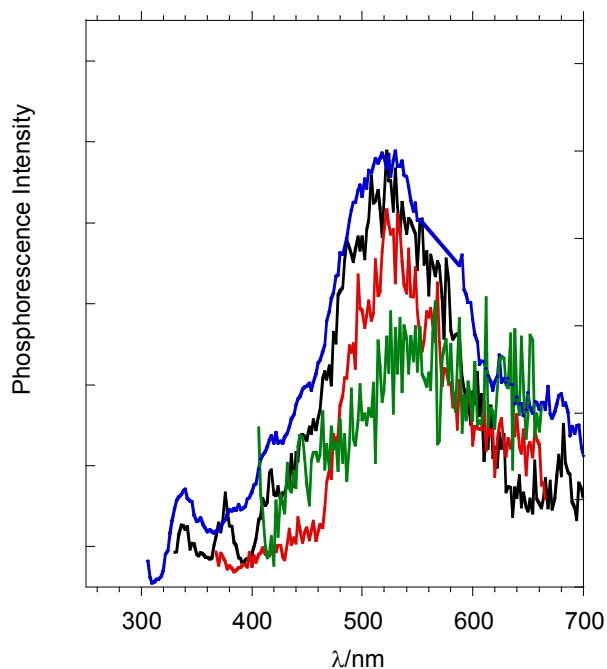


Figure S37. Phosphorescence spectra of TT-Py in PMMA (w/w TT-Py/PMMA 10%) at 298 K. $\lambda_{\text{exc}}=290\text{nm}$ (blue line, delay 100 μs , window 500 μs); $\lambda_{\text{exc}}=300\text{nm}$ (black line, delay 1 ms, window 5 ms); $\lambda_{\text{exc}}=350\text{nm}$ (red line, delay 1 ms, window 5 ms); $\lambda_{\text{exc}}=390\text{nm}$ (green line, delay 1 ms, window 5 ms).

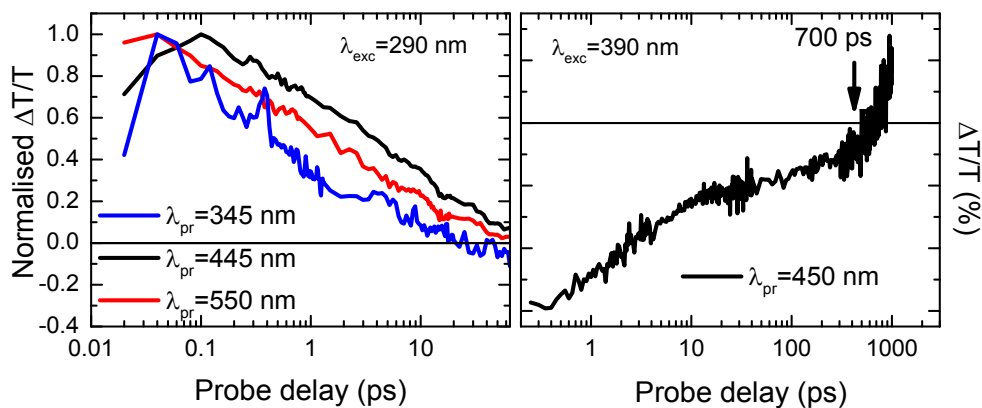


Figure S38. Ultrafast spectroscopy dynamics selected at different probe wavelengths. Left panel: 290 nm excitation. Right panel: 390 nm excitation.

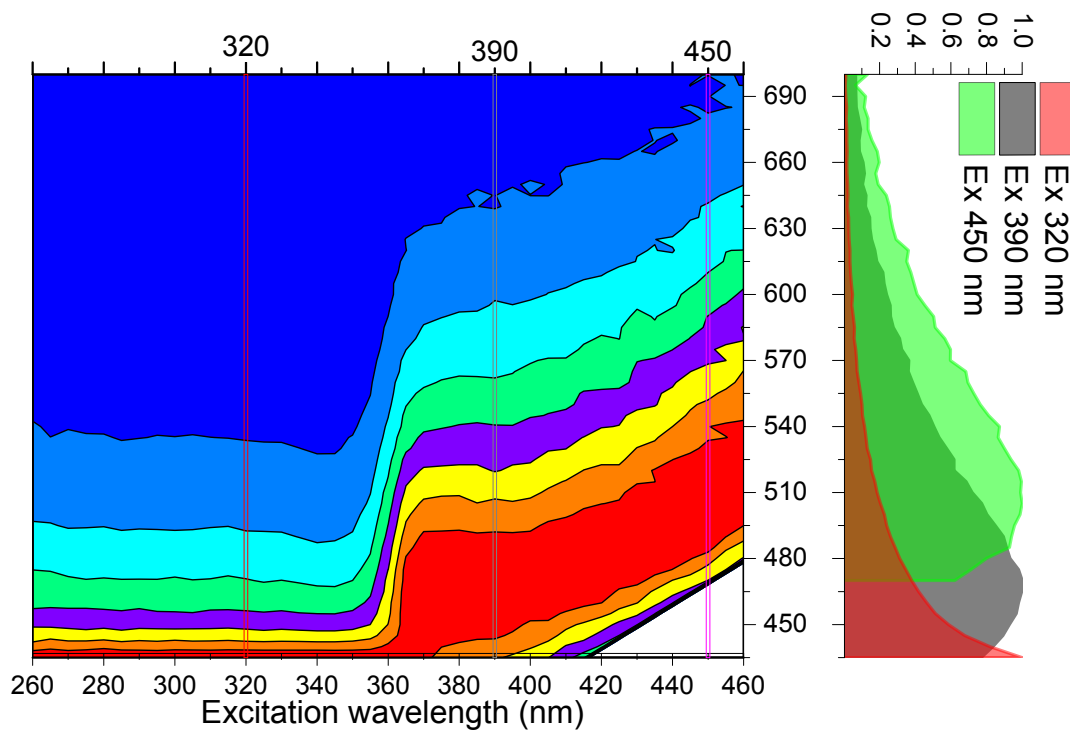


Figure S39. Normalized excitation-emission mapping of TT-Py-A crystals.

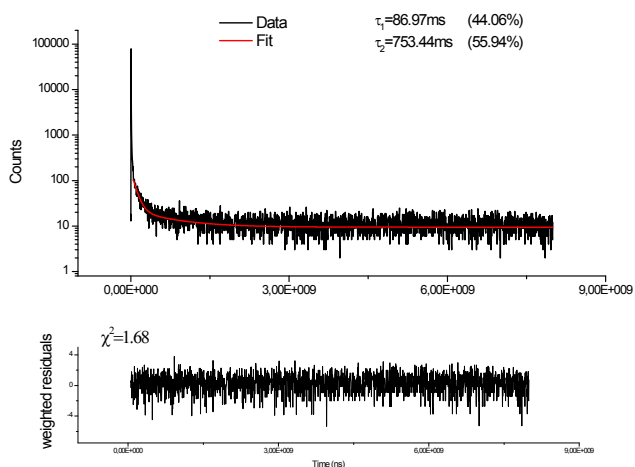
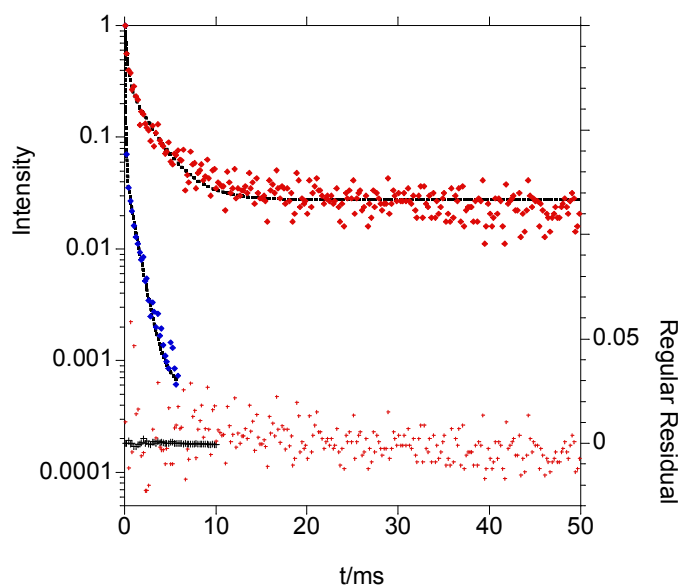


Figure S40. Decay of TT-Py-A crystals at 298 K. $\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=360\text{nm}$



	A1	t1	A2	t2	Adj. R-Square	τ_{av}	$\langle \tau \rangle$
410nm	0,05346	0,92075	2,26981	0,05732	0,99998	0.29	0.08
615nm	0,81543	0,24181	0,30478	2,55486	0,98135	2.09	0.87

Figure S41. Phosphorescence decay of TT-Py-A crystals at 298 K. $\lambda_{\text{exc}}=290\text{nm}$, $\lambda_{\text{em}}=410\text{nm}$ (blue point) with biexponential fit (black dotted); $\lambda_{\text{exc}}=360\text{nm}$, $\lambda_{\text{em}}=615\text{nm}$ (red point) with biexponential fit (black dotted); see parameters above.

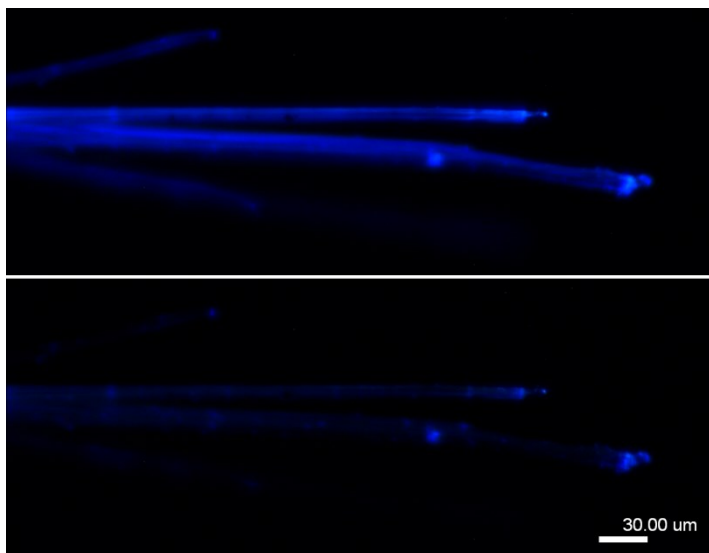


Figure S42. Fluorescence microscopy images of **TT-Py-H** crystals at 298 K. $\lambda_{\text{exc}}=360\text{-}400\text{nm}$, with vertical (top) and horizontal (bottom) polarized emission.

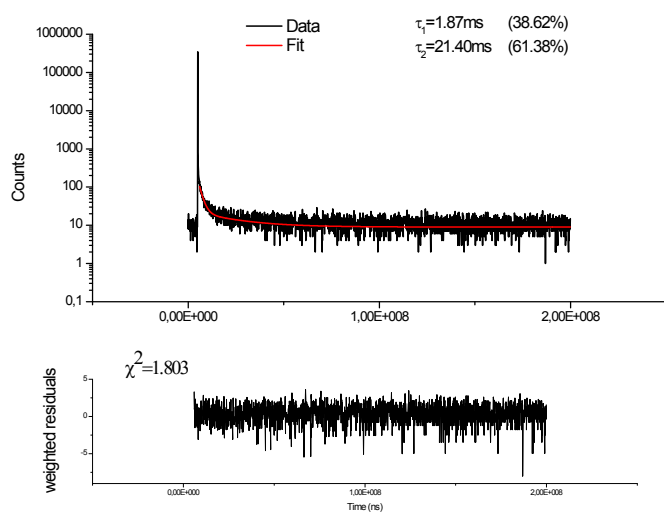


Figure S43. Decay of **TT-Py-H** crystals at 298 K. $\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=360\text{nm}$

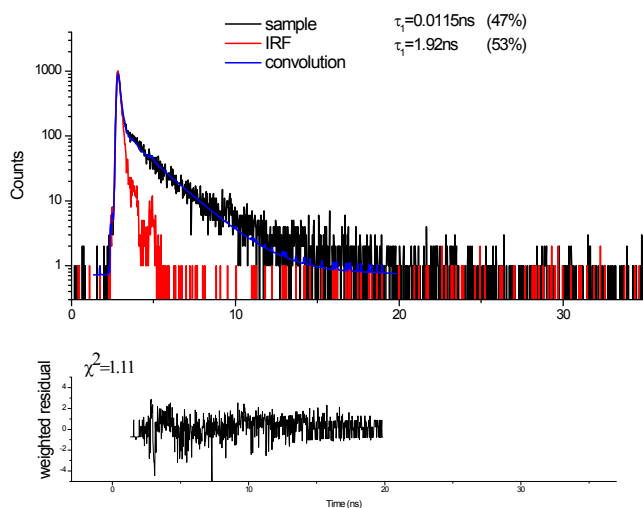


Figure S44. Decay of TT-Py-H crystals at 298 K. $\lambda_{\text{exc}}=404\text{nm}$, $\lambda_{\text{em}}=450\text{nm}$

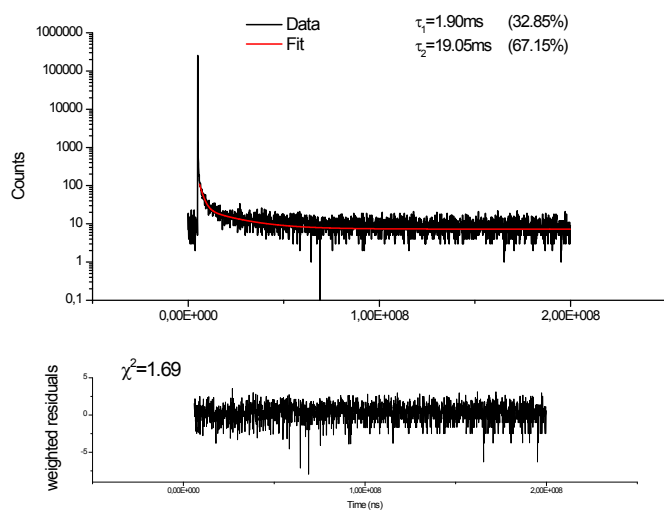


Figure S45. Decay of TT-Py-H crystals at 298 K. $\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=500\text{nm}$

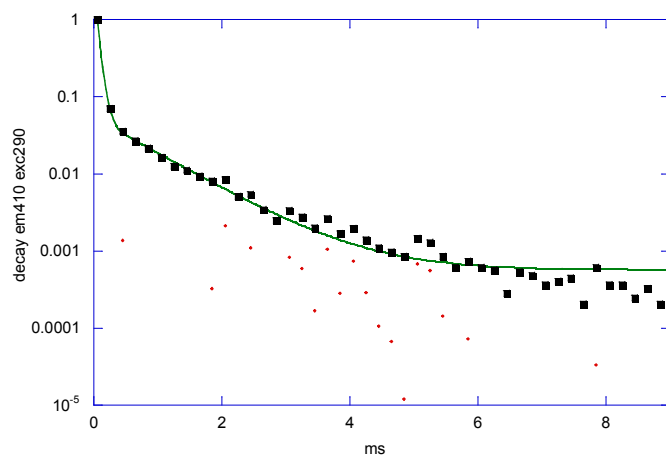


Figure S46. Decay of TT-Py-H crystals at 298 K. $\lambda_{\text{exc}}=290\text{nm}$, $\lambda_{\text{em}}=410\text{nm}$. 2exp fit, $A1=0.0379$, $t1=1.047\text{ms}$, $A2=0.9548$, $t2=0.05846\text{ms}$. $\tau_{\text{av}}=0.47\text{ ms}$. $R= 0,99972$

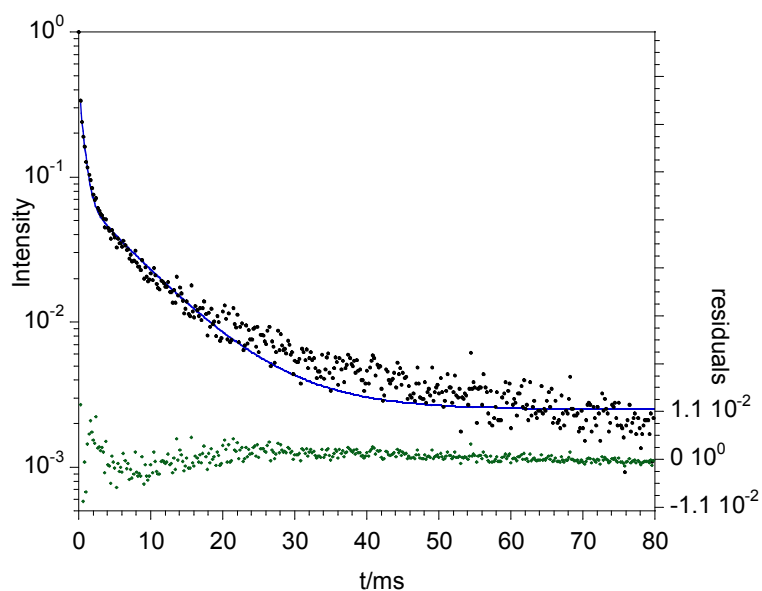


Figure S47. Decay of TT-Py-H crystals at 298 K. $\lambda_{\text{exc}}=272\text{nm}$, $\lambda_{\text{em}}=570\text{nm}$. 2exp fit, $A1=0.8434$, $t1=0.6272\text{ms}$, $A2=0.1566$, $t2=8.189\text{ms}$. $\tau_{\text{av}}=5.98\text{ ms}$. $R= 0,99341$

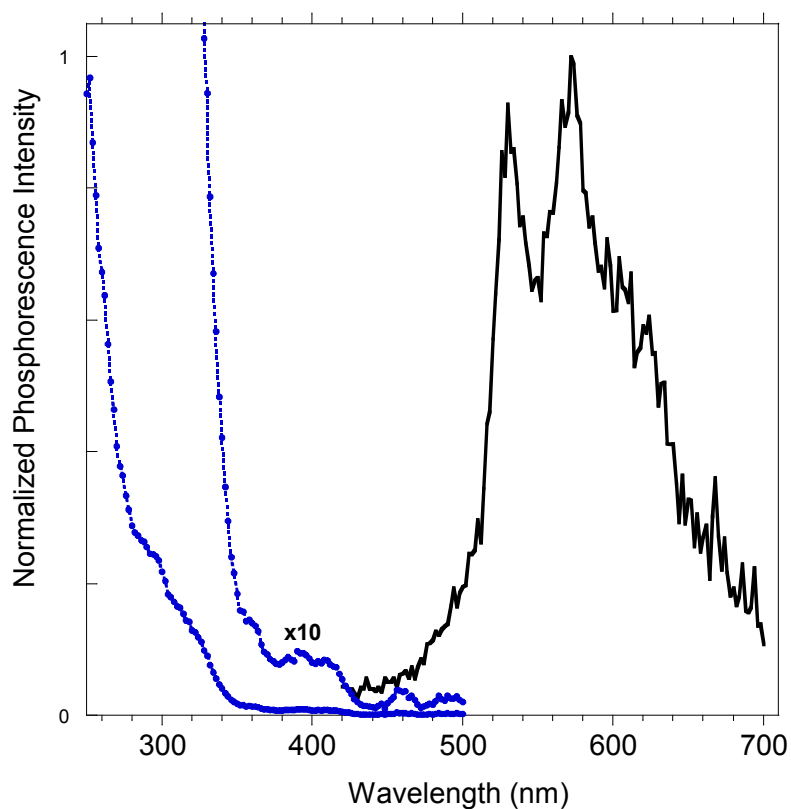


Fig. S48. Phosphorescence of **TT-Py-H** at 86 K. Emission (exc 400nm) and excitation profile of the 530nm emission. (delay 200 μ s, window 500 μ s)

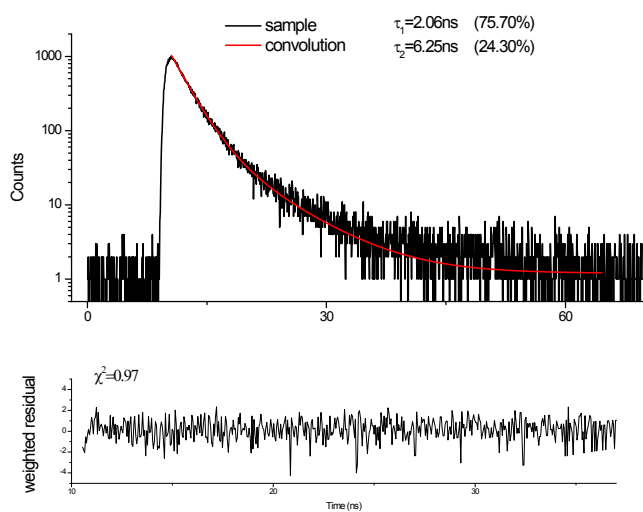


Figure S49. Decay of **TT-Py-X** crystals at 298 K. $\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=450\text{nm}$

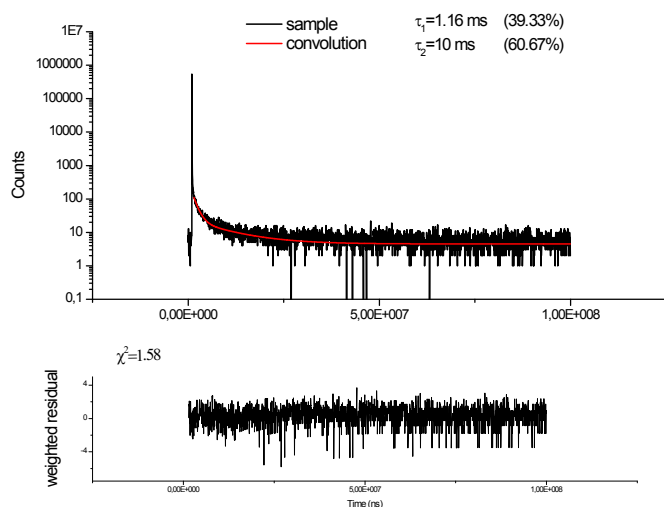


Figure S50. Decay of TT-Py-X crystals at 298 K. $\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=570\text{nm}$

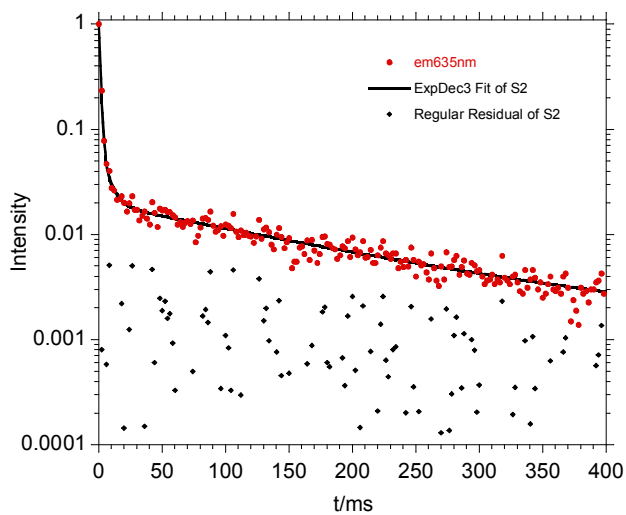


Figure S51. Decay of TT-Py-X crystals at 298 K. $\lambda_{\text{exc}}=370\text{nm}$, $\lambda_{\text{em}}=634\text{nm}$. 3exp fit, $A1=0.0477$, $t1=6.34841\text{ms}$, $A2=0.9363$, $t2=1.19579$, $A3=0.0159$, $t3=168.6882$. $\tau_{\text{av}}=111.07\text{ ms}$. $R=0.99954$

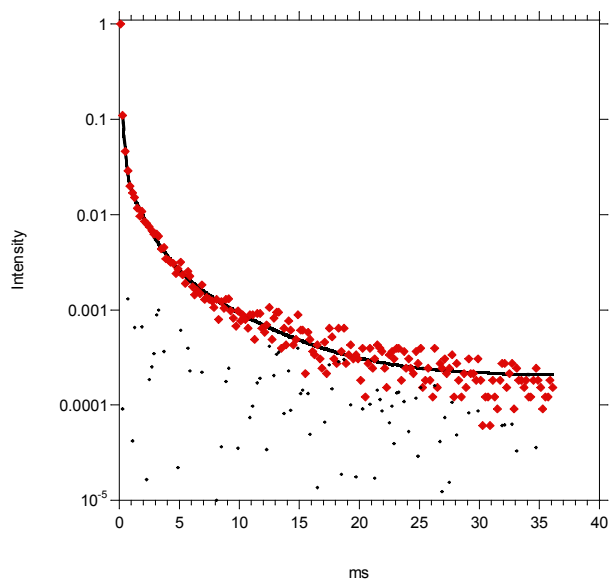


Figure S52. Decay of TT-Py-X crystals at 298 K. $\lambda_{\text{exc}}=270\text{nm}$, $\lambda_{\text{em}}=408\text{nm}$. 3exp fit, $A1=0.94$, $t1=0.1578\text{ms}$, $A2=0.05$, $t2=1.2876$, $A3=0.01$, $t3=5.36276$. $\tau_{\text{av}}=1.31\text{ ms}$. $R=0.99912$

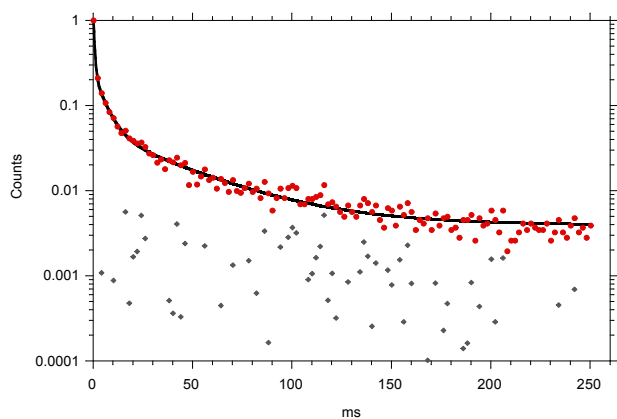


Figure S53. Decay of TT-Py-X crystals at 298 K. $\lambda_{\text{exc}}=360\text{nm}$, $\lambda_{\text{em}}=532\text{nm}$. 3exp fit, $A1=0.1538$, $t1=5.45086\text{ms}$, $A2=0.8098$, $t2=0.61944$, $A3=0.0363$, $t3=40.02115$. $\tau_{\text{av}}=22.60\text{ ms}$. $R=0.99956$

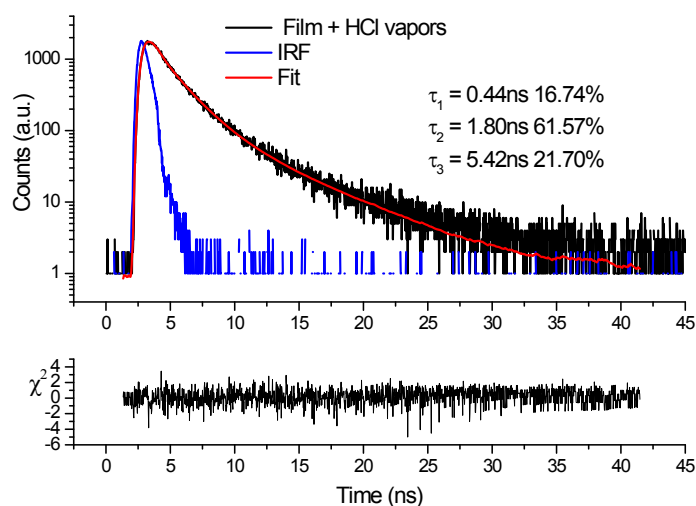


Figure S54. Lifetime measurement ($\lambda_{\text{exc}}=300\text{nm}$, $\lambda_{\text{em}}=421\text{nm}$) of **TT-Py** in PMMA (w/w **TT-Py**/PMMA 10%) at 298 K exposed to HCl vapors for 45min $\tau_{\text{av}}= 3.56$ ns.).

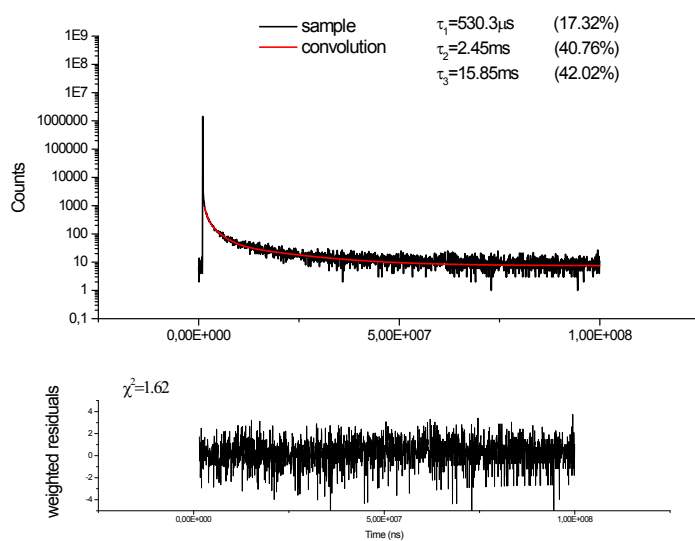


Figure S55. Lifetime measurement ($\lambda_{\text{exc}}=340\text{nm}$, $\lambda_{\text{em}}=426\text{nm}$) of **TT-Py** in PMMA (w/w **TT-Py**/PMMA 10%) at 298 K exposed to HCl vapors for 45min $\tau_{\text{av}}= 13.94$ ms.

4. Crystal structures

Table S1. Crystallographic data and structure refinement details for **TT-Py-A**, **TT-Py-H·1.7H₂O** and **TT-Py-X** polymorphs.

	TT-Py-A	TT-Py-H·1.7H₂O	TT-Py-X
Chemical Formula	C ₁₄ H ₉ N ₇	(C ₁₄ H ₉ N ₇),O _{1.7}	C ₁₄ H ₉ N ₇
Molecular weight	275.28	307.28	275.28
T(K)	293(2)	120(2)	293(2)
Crystal system	Orthorhombic	Monoclinic	Monoclinic
space group	<i>Pbcn</i>	<i>P2₁/c</i>	<i>P2₁/c</i>
a(Å)	33.189(8)	3.7357(18)	18.0949(7)
b(Å)	10.162(3)	14.805(7)	8.8509(4)
c(Å)	7.2434(18)	23.784(11)	17.3684(7)
α(°)	90	90	90
β(°)	90	90.284(7)	117.796(1)
γ(°)	90	90	90
V(Å ³)	2443.0(10)	1315.4(11)	2460.69(18)
Z	8	4	8
D _{calcd} (g cm ⁻³)	1.497	1.552	1.486
μ (mm ⁻¹)	0.099	0.112	0.099
Crystal size (mm)	0.55 x 0.07 x 0.01	0.45 x 0.04 x 0.04	0.35 x 0.27 x 0.18
2θ _{max} , °	55.00	55.32	61.0
No. of measured, independent and observed [I > 2σ(I)] reflections	66937 / 2803 / 1742	19094 / 3031 / 1577	47901 / 7501 / 5146
(R _{int})/(R _σ)	0.1067 / 0.0384	0.1709 / 0.1332	0.0285 / 0.0201
data/restraints/params	2803 / 0 / 190	3031 / 6 / 220	7501 / 0 / 379
R[F ² > 2σ(F ²)], wR(F ²), S	0.0497, 0.1122, 1.059	0.0665, 0.1518, 1.000	0.0468, 0.1245, 1.028
Δρ _{max} , Δρ _{min} (e Å ⁻³)	0.185, -0.204	0.303, -0.349	0.166, -0.220

5. Theoretical studies

In Tables S2-S4 we report the first singlet and triplet excitation energies as computed *in vacuo* at ωB97X/6-311++G(d,p) level for the optimized geometry of **TT-Py** in its absolute (**A**, Table S2) and relative (**D**, Table S3) minima and in the maximum energy state (**E**, Table S4). The simulated absorption spectrum computed at the minimum energy geometry (state **A**) is shown in Figure S52, together with that of the protonated species (**TT-PyH⁺**) in its absolute minimum. The computed maxima positions are slightly blue-shifted with respect to the experimental ones principally as a consequence of the functional choice (see Computational details).

A schematic representation of the computed levels for the optimized monomers of **TT** and **TT-Py** (state **A**) is shown in Figure S53, while Figure S54 displays isodensity surface plots of the frontier orbitals mainly involved in the computed transitions of **TT-Py**.

Tables S5-S7 summarize the first singlet and triplet excitation energies as computed *in vacuo* at ω B97X/6-311++G(d,p) level for the optimized geometries of dimers of **TT-Py** starting from the X-ray structures of **TT-Py-A** (Table S5), **TT-Py-H** (Table S6) and **TT-Py-X** (Table S7).

Table S8 reports the first singlet and triplet excitation energies as computed *in vacuo* at ω B97X/6-311++G(d,p) level for the optimized geometries of **TT-PyH⁺** in its absolute minimum.

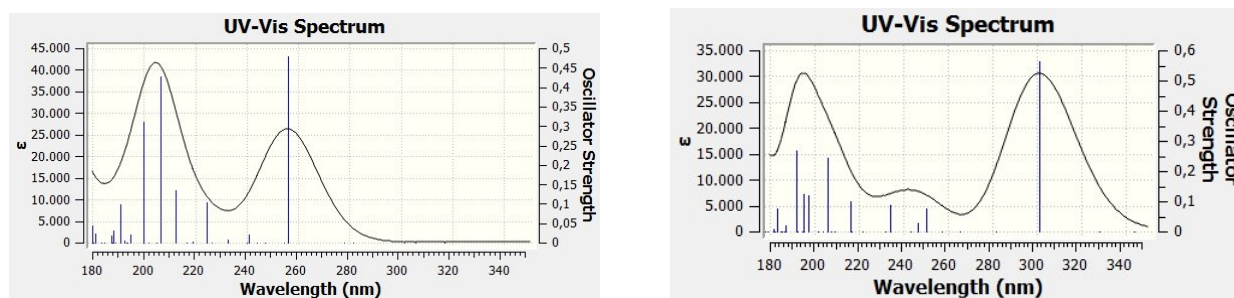


Figure S56. ω B97X/6-311++G(d,p) computed absorption spectrum of **TT-Py** (left) and **TT-PyH⁺** (right) in their absolute minimum, resulting from convolution of the excitation energies (blue sticks) with 0.25 eV of half-bandwidth.

Figure S57. Electronic levels computed for **TT** and **TT-Py** (state A) at molecular level. In blue are reported the singlet levels with oscillator strength $f \geq 0.0001$ and the corresponding values of f (see detailed information in Table S3).

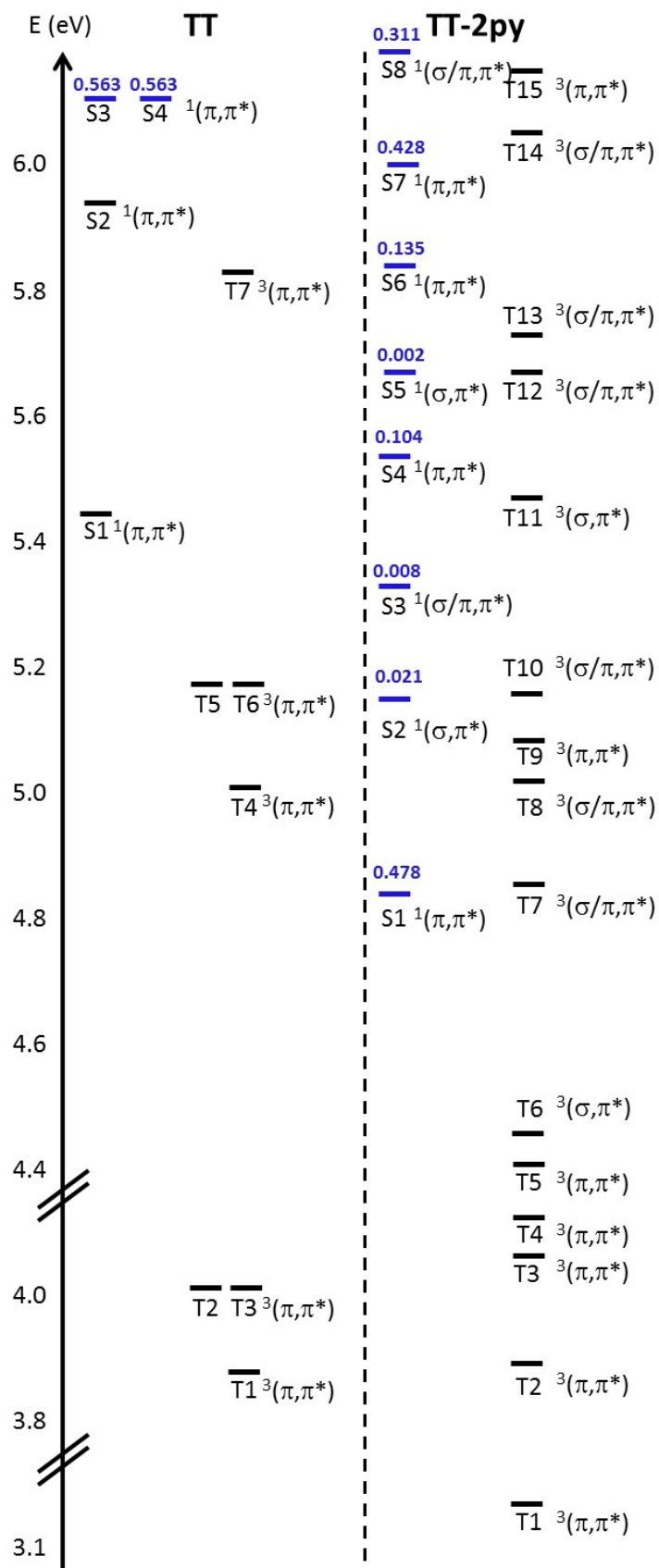
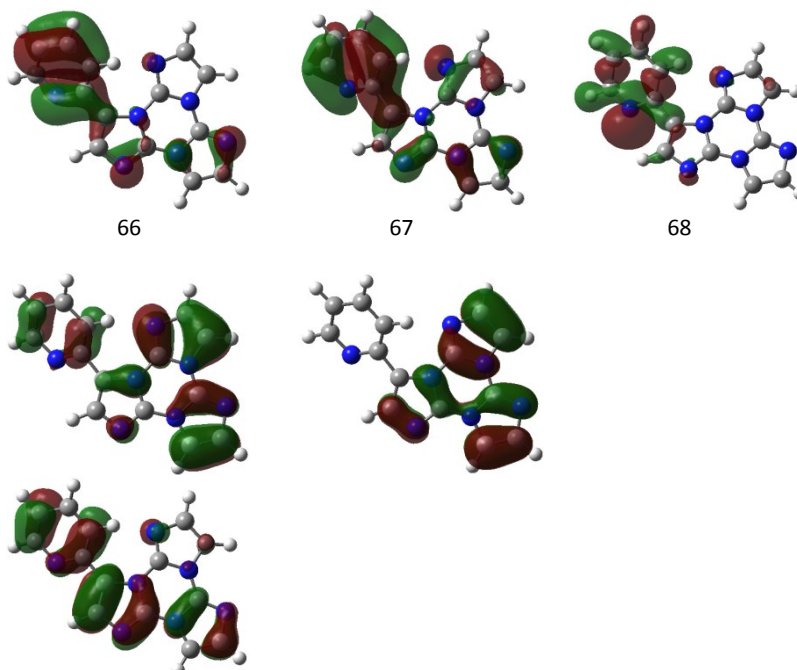


Figure S58. Isodensity surface plots of the frontier orbitals of monomeric **TT-Py** mainly involved in the computed transitions (isosurface values: 0.025, energies in a.u.).

Highest Occupied MOs

MO	E(a.u.)	symmetry
66	-0.39632	π
67	-0.37928	mixed σ/π (σ contribution on one TT's N)
68	-0.37198	σ
69	-0.34920	π
70	-0.33226	π
71	-0.31405	π



Lowest Unoccupied MOs

72	0.02046	π
73	0.04168	π
74	0.04628	π
75	0.04698	π
76	0.04858	mixed σ/π (σ on TT, π on py)
77	0.05238	σ
78	0.06064	σ
79	0.06217	π
80	0.06672	π

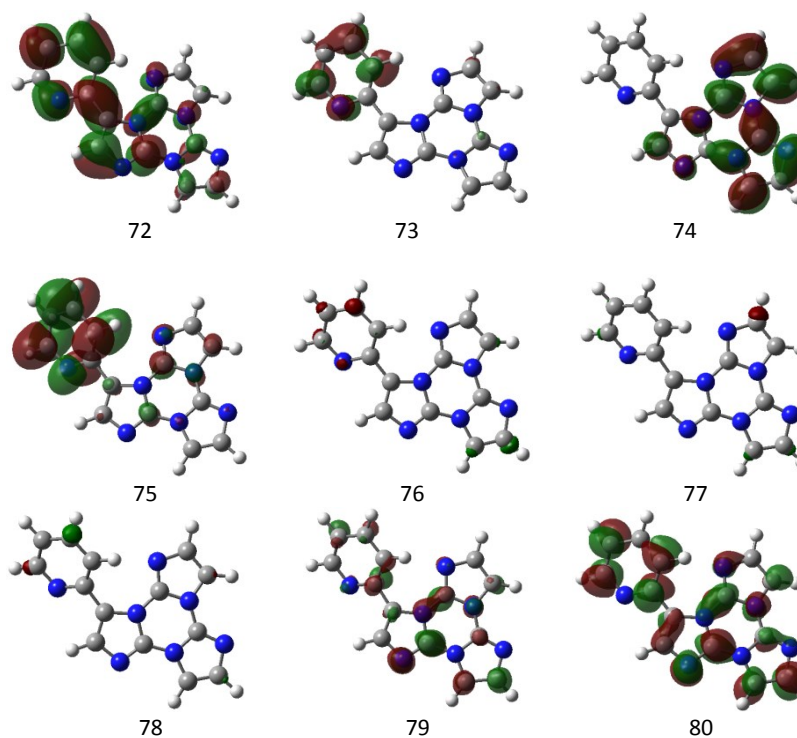


Table S2. First TD- ω B97X/6-311++G(d,p) $S_0 \rightarrow S_n$ and $T_0 \rightarrow T_n$ transitions computed for the optimized geometry of TT-Py (state A).

T1	Excited State	1:	Triplet-A	3.1469 eV	393.98 nm	f=nd	<S**2>=2.000
	66 -> 72		-0.12578				
	66 -> 75		-0.10762				
	67 -> 75		-0.10620				
	67 -> 80		-0.10543				
	69 -> 72		0.11322				
	71 -> 72		0.58886				
	71 -> 75		-0.10734				
T2	Excited State	2:	Triplet-A	3.9108 eV	317.03 nm	f=nd	<S**2>=2.000
	66 -> 72		-0.16448				
	67 -> 75		-0.17145				
	69 -> 72		0.18436				
	69 -> 75		-0.10937				
	70 -> 74		0.42171				
	71 -> 72		-0.11469				
	71 -> 79		0.11748				
	71 -> 80		-0.20702				
T3	Excited State	3:	Triplet-A	4.0599 eV	305.39 nm	f=nd	<S**2>=2.000
	66 -> 72		0.19300				
	67 -> 73		-0.12774				
	67 -> 75		0.26688				
	69 -> 80		0.15806				
	70 -> 74		0.33476				
	71 -> 72		0.17903				
	71 -> 75		0.10862				
	71 -> 80		0.12362				
	71 -> 86		0.13503				
T4	Excited State	4:	Triplet-A	4.1178 eV	301.10 nm	f=nd	<S**2>=2.000
	69 -> 74		0.42389				
	70 -> 72		0.21543				
	70 -> 75		-0.10631				
	70 -> 79		-0.15052				
	70 -> 80		0.22917				
	70 -> 81		-0.10015				
	70 -> 86		-0.14112				
	71 -> 74		-0.17828				
T5	Excited State	5:	Triplet-A	4.4072 eV	281.32 nm	f=nd	<S**2>=2.000
	66 -> 72		-0.15488				
	67 -> 72		0.25881				
	67 -> 73		-0.10053				
	67 -> 75		0.20379				
	69 -> 72		0.12081				
	69 -> 75		0.14212				
	71 -> 73		-0.18919				
	71 -> 75		0.38143				
	71 -> 79		0.14462				
	71 -> 80		-0.12042				
T6	Excited State	6:	Triplet-A	4.4628 eV	277.82 nm	f=nd	<S**2>=2.000
	68 -> 72		0.49664				
	68 -> 73		-0.18725				
	68 -> 75		0.37303				
S1	Excited State	7:	Singlet-A	4.8401 eV	256.16 nm	f=0.4782	<S**2>=0.000
	69 -> 72		0.10809				
	71 -> 72		0.64888				
T7	Excited State	8:	Triplet-A	4.8747 eV	254.34 nm	f=nd	<S**2>=2.000
	66 -> 72		0.25117				
	66 -> 73		-0.15808				
	66 -> 75		0.31080				
	67 -> 72		0.34657				
	69 -> 72		0.10076				
	71 -> 75		-0.20577				
	71 -> 80		-0.17270				
T8	Excited State	9:	Triplet-A	5.0156 eV	247.20 nm	f=nd	<S**2>=2.000
	67 -> 72		-0.15285				
	67 -> 74		-0.11905				
	67 -> 80		0.11782				
	70 -> 72		0.14404				
	70 -> 80		0.14253				
	71 -> 74		0.43334				
	71 -> 80		-0.11016				
	71 -> 97		-0.13366				
T9	Excited State	10:	Triplet-A	5.0810 eV	244.02 nm	f=nd	<S**2>=2.000
	69 -> 72		0.16870				
	69 -> 74		-0.24116				

	69 -> 86	0.18083				
	70 -> 72	0.33677				
	70 -> 75	-0.10153				
	70 -> 80	0.12425				
	70 -> 86	0.18908				
	70 -> 94	0.10296				
	71 -> 80	0.13441				
S2	Excited State 11:	Singlet-A	5.1458 eV	240.94 nm	f=0.0209	<S**2>=0.000
	68 -> 72	0.57418				
	68 -> 73	-0.14131				
	68 -> 75	0.27474				
	68 -> 79	0.10097				
	68 -> 80	-0.10013				
T10	Excited State 12:	Triplet-A	5.1663 eV	239.99 nm	f=nd	<S**2>=2.000
	62 -> 74	0.11105				
	67 -> 72	-0.14923				
	69 -> 72	0.21157				
	69 -> 74	0.17311				
	69 -> 75	-0.12814				
	69 -> 79	-0.13306				
	69 -> 80	0.20532				
	70 -> 74	-0.21725				
	70 -> 86	0.22464				
	70 -> 87	0.11087				
S3	Excited State 13:	Singlet-A	5.3272 eV	232.74 nm	f=0.0081	<S**2>=0.000
	66 -> 75	-0.11748				
	66 -> 80	-0.10059				
	67 -> 72	0.23000				
	69 -> 72	0.17092				
	70 -> 72	0.11492				
	70 -> 74	0.21907				
	71 -> 73	-0.22788				
	71 -> 74	0.17570				
	71 -> 75	0.37599				
	71 -> 79	0.10016				
T11	Excited State 14:	Triplet-A	5.4751 eV	226.45 nm	f=nd	<S**2>=2.000
	68 -> 72	-0.37076				
	68 -> 73	-0.19978				
	68 -> 75	0.40806				
	68 -> 80	0.26248				
	68 -> 81	-0.15390				
S4	Excited State 15:	Singlet-A	5.5278 eV	224.29 nm	f=0.1043	<S**2>=0.000
	67 -> 72	-0.11363				
	69 -> 72	0.17970				
	69 -> 75	-0.11456				
	70 -> 72	0.26849				
	70 -> 74	0.27324				
	70 -> 75	-0.12398				
	70 -> 80	0.13240				
	71 -> 73	0.10750				
	71 -> 74	0.19630				
	71 -> 75	-0.25057				
	71 -> 80	-0.15996				
S5	Excited State 16:	Singlet-A	5.6568 eV	219.18 nm	f=0.0017	<S**2>=0.000
	68 -> 72	-0.23119				
	68 -> 73	-0.24021				
	68 -> 75	0.48370				
	68 -> 80	0.24022				
	68 -> 81	-0.14578				
T12	Excited State 17:	Triplet-A	5.6616 eV	218.99 nm	f=nd	<S**2>=2.000
	66 -> 72	0.26172				
	66 -> 73	0.12084				
	66 -> 75	-0.27633				
	66 -> 80	-0.18137				
	66 -> 81	0.10714				
	67 -> 72	0.23065				
	67 -> 75	-0.14852				
	67 -> 80	-0.14563				
	71 -> 74	0.15499				
	71 -> 80	0.13741				
T13	Excited State 18:	Triplet-A	5.7267 eV	216.50 nm	f=nd	<S**2>=2.000
	60 -> 74	0.16641				
	61 -> 72	0.12322				
	66 -> 72	0.21828				
	66 -> 75	-0.11774				
	66 -> 80	-0.14460				
	67 -> 75	-0.12082				
	69 -> 86	0.18958				
	70 -> 72	-0.13607				
	70 -> 90	0.10369				
	70 -> 92	0.12999				

	70 -> 94	0.10429					
	71 -> 74	-0.16675					
	71 -> 75	0.12567					
	71 -> 94	-0.10878					
	71 -> 97	-0.10927					
S6	Excited State 19:	Singlet-A	5.8424 eV	212.21 nm	f=0.1352	<S**2>=0.000	
	69 -> 72	0.14195					
	69 -> 80	0.10798					
	70 -> 72	-0.27104					
	70 -> 74	0.40095					
	70 -> 80	-0.11212					
	71 -> 74	-0.34906					
S7	Excited State 20:	Singlet-A	6.0009 eV	206.61 nm	f=0.4276	<S**2>=0.000	
	69 -> 74	-0.33687					
	70 -> 72	-0.32804					
	70 -> 74	0.18705					
	70 -> 80	-0.11512					
	71 -> 74	0.34953					
T14	Excited State 21:	Triplet-A	6.0497 eV	204.94 nm	f=nd	<S**2>=2.000	
	62 -> 74	0.15120					
	66 -> 74	0.10305					
	67 -> 74	-0.10057					
	69 -> 74	0.15572					
	69 -> 86	0.16460					
	70 -> 72	-0.21380					
	71 -> 74	0.23609					
	71 -> 75	-0.10703					
	71 -> 79	-0.12484					
	71 -> 80	0.15647					
T15	Excited State 22:	Triplet-A	6.1469 eV	201.70 nm	f=nd	<S**2>=2.000	
	60 -> 72	0.10143					
	62 -> 72	0.14938					
	62 -> 80	0.10497					
	69 -> 72	0.28807					
	69 -> 74	-0.12102					
	69 -> 90	-0.11304					
	69 -> 92	-0.13042					
	70 -> 74	-0.17292					
	70 -> 86	-0.25970					
	70 -> 87	-0.12666					
	70 ->110	-0.10372					
S8	Excited State 23:	Singlet-A	6.1977 eV	200.05 nm	f=0.3114	<S**2>=0.000	
	66 -> 72	0.11703					
	67 -> 75	0.10643					
	70 -> 72	0.18972					
	70 -> 74	0.22332					
	71 -> 73	-0.12727					
	71 -> 74	-0.12173					
	71 -> 79	-0.33491					
	71 -> 80	0.34113					

T1	Excited State	1:	Triplet-A	3.4850 eV	355.76 nm	f=nd	<S**2>=2.000
	66 -> 72		0.18891				
	66 -> 74		0.16568				
	67 -> 72		-0.12062				
	67 -> 74		0.16412				
	67 -> 80		0.19162				
	69 -> 72		-0.20076				
	71 -> 72		0.43137				
	71 -> 74		-0.16506				
T2	Excited State	2:	Triplet-A	3.8757 eV	319.90 nm	f=nd	<S**2>=2.000
	66 -> 72		-0.13285				
	66 -> 74		-0.10361				
	67 -> 72		0.11880				
	67 -> 74		-0.15194				
	69 -> 72		0.11925				
	70 -> 75		0.12600				
	70 -> 76		0.40130				
	71 -> 72		0.16538				
	71 -> 80		0.26658				
T3	Excited State	3:	Triplet-A	4.0664 eV	304.90 nm	f=nd	<S**2>=2.000
	66 -> 72		0.14421				
	66 -> 74		0.10214				
	67 -> 72		-0.13267				
	67 -> 74		0.19754				

	69 -> 80	0.19972					
	70 -> 75	0.10596					
	70 -> 76	0.33450					
	71 -> 72	-0.22519					
	71 -> 80	-0.16920					
	71 -> 86	0.13818					
T4	Excited State	4:	Triplet-A	4.0960 eV	302.70 nm	f=nd	<S**2>=2.000
	69 -> 75	0.11976					
	69 -> 76	0.38329					
	70 -> 72	0.17735					
	70 -> 74	-0.11202					
	70 -> 80	0.26874					
	70 -> 86	0.11174					
	71 -> 76	0.25265					
T5	Excited State	5:	Triplet-A	4.4338 eV	279.64 nm	f=nd	<S**2>=2.000
	68 -> 72	0.53450					
	68 -> 73	-0.14566					
	68 -> 74	0.33539					
T6	Excited State	6:	Triplet-A	4.5196 eV	274.33 nm	f=nd	<S**2>=2.000
	67 -> 72	0.36608					
	67 -> 73	-0.13950					
	67 -> 74	0.31010					
	69 -> 72	0.16123					
	69 -> 74	0.17260					
	71 -> 72	-0.17278					
	71 -> 73	0.13195					
	71 -> 74	-0.28045					
	71 -> 80	0.10966					
T7	Excited State	7:	Triplet-A	4.9251 eV	251.74 nm	f=nd	<S**2>=2.000
	65 -> 72	0.11665					
	66 -> 72	0.37288					
	66 -> 73	-0.11482					
	66 -> 74	0.26487					
	67 -> 72	0.25696					
	67 -> 74	-0.12314					
	67 -> 80	-0.12580					
	69 -> 74	-0.13864					
	71 -> 74	0.16998					
	71 -> 80	0.14962					
T8	Excited State	8:	Triplet-A	5.0528 eV	245.38 nm	f=nd	<S**2>=2.000
	67 -> 72	-0.13959					
	67 -> 76	0.12416					
	67 -> 80	0.10515					
	69 -> 86	-0.10485					
	69 -> 87	0.10760					
	70 -> 72	-0.17291					
	70 -> 80	-0.19144					
	71 -> 75	0.12745					
	71 -> 76	0.40811					
	71 -> 95	-0.10848					
T9	Excited State	9:	Triplet-A	5.1291 eV	241.73 nm	f=nd	<S**2>=2.000
	69 -> 72	0.14767					
	69 -> 76	0.23429					
	69 -> 86	-0.10897					
	70 -> 72	-0.29471					
	70 -> 80	-0.18228					
	70 -> 86	0.15979					
	70 -> 92	-0.11090					
	70 -> 94	0.11384					
	71 -> 80	-0.13367					
	71 -> 87	0.10003					
S1	Excited State	10:	Singlet-A	5.1403 eV	241.20 nm	f=0.0790	<S**2>=0.000
	68 -> 72	0.49261					
	68 -> 73	-0.10078					
	68 -> 74	0.23655					
	69 -> 72	0.18564					
	71 -> 72	-0.30048					

T1	Excited State	1:	Triplet-A	2.5339 eV	489.29 nm	f=nd	<S**2>=2.000
	66 -> 72		0.11051				
	66 -> 80		-0.11232				
	67 -> 77		0.13411				
	68 -> 72		0.14850				
	70 -> 72		0.64064				

	70 <- 72	0.13955					
T2	Excited State	2:	Triplet-A	2.8270 eV	438.58 nm	f=nd	<S**2>=2.000
	71 -> 72	0.61658					
	71 -> 77	0.18820					
	71 -> 80	-0.20458					
S1	Excited State	3:	Singlet-A	3.2264 eV	384.28 nm	f=0.0003	<S**2>=0.000
	71 -> 72	0.66507					
	71 -> 80	-0.16937					

Table S5. First TD- ω B97X/6-311++G(d,p) $S_0 \rightarrow S_n$ and $T_0 \rightarrow T_n$ transitions computed for dimeric **TT-Py** at its optimized geometry starting from the X-ray geometry of **TT-Py-A**.

T1	Excited State	1:	Triplet-A	3.1818 eV	389.67 nm	f=nd	<S**2>=2.000
	132 -> 144	-0.12322					
	138 -> 144	0.11933					
	142 -> 143	0.25955					
	142 -> 144	0.49944					
T2	Excited State	2:	Triplet-A	3.2425 eV	382.37 nm	f=nd	<S**2>=2.000
	131 -> 143	-0.14511					
	134 -> 146	0.10640					
	134 -> 155	0.11834					
	137 -> 143	0.13874					
	140 -> 143	-0.13406					
	141 -> 143	0.48014					
	141 -> 144	-0.21482					
	141 -> 146	-0.13844					
T3	Excited State	3:	Triplet-A	3.8876 eV	318.92 nm	f=nd	<S**2>=2.000
	131 -> 143	0.15316					
	134 -> 145	-0.10995					
	134 -> 146	-0.17875					
	137 -> 143	-0.12380					
	137 -> 146	0.10671					
	139 -> 147	-0.13177					
	139 -> 148	0.25714					
	139 -> 149	-0.10172					
	139 -> 150	0.17559					
	141 -> 143	0.12689					
	141 -> 155	0.23880					
T4	Excited State	4:	Triplet-A	3.8945 eV	318.36 nm	f=nd	<S**2>=2.000
	132 -> 144	-0.15644					
	135 -> 148	-0.10059					
	135 -> 150	0.12171					
	135 -> 151	-0.12771					
	138 -> 144	0.13662					
	140 -> 152	0.31604					
	141 -> 152	0.10000					
	142 -> 144	-0.12097					
	142 -> 157	0.12059					
	142 -> 158	-0.13499					
	142 -> 159	0.10538					
	142 -> 160	0.13458					
T5	Excited State	5:	Triplet-A	4.0310 eV	307.58 nm	f=nd	<S**2>=2.000
	131 -> 143	0.14292					
	134 -> 145	-0.13614					
	134 -> 146	-0.22059					
	137 -> 155	0.17256					
	139 -> 147	0.11995					
	139 -> 148	-0.22794					
	139 -> 150	-0.16327					
	140 -> 152	-0.10574					
	141 -> 143	0.16628					
T6	Excited State	6:	Triplet-A	4.0358 eV	307.21 nm	f=nd	<S**2>=2.000
	132 -> 144	0.12978					
	135 -> 148	0.11367					
	135 -> 150	-0.14030					
	135 -> 151	0.14346					
	140 -> 152	0.29728					
	142 -> 144	0.13381					
	142 -> 171	-0.10344					
T7	Excited State	7:	Triplet-A	4.1054 eV	302.00 nm	f=nd	<S**2>=2.000
	137 -> 147	0.14439					
	137 -> 148	-0.28541					
	137 -> 149	0.11419					
	137 -> 150	-0.19032					
	139 -> 143	0.15702					
	139 -> 155	0.26420					
	141 -> 148	0.13922					
T8	Excited State	8:	Triplet-A	4.1198 eV	300.95 nm	f=nd	<S**2>=2.000
	138 -> 152	0.37487					

139 -> 152	-0.12202					
140 -> 143	0.10620					
140 -> 144	0.15225					
140 -> 157	-0.11786					
140 -> 158	0.14316					
140 -> 160	-0.13526					
140 -> 171	0.11412					
142 -> 152	-0.15490					
T9	Excited State	9:	Triplet-A	4.3818 eV	282.95 nm	f=nd <S**2>=2.000
	131 -> 143	-0.10891				
	134 -> 143	-0.30000				
	134 -> 144	0.11291				
	134 -> 145	-0.13045				
	134 -> 146	-0.17896				
	137 -> 143	0.14291				
	137 -> 146	0.13678				
	141 -> 143	0.10211				
	141 -> 145	0.18874				
	141 -> 146	0.29138				
	141 -> 155	-0.19311				
T10	Excited State	10:	Triplet-A	4.3881 eV	282.55 nm	f=nd <S**2>=2.000
	132 -> 144	-0.11580				
	135 -> 143	-0.10857				
	135 -> 144	-0.21738				
	135 -> 150	-0.11651				
	135 -> 151	0.12259				
	136 -> 144	0.13731				
	138 -> 144	0.14648				
	138 -> 150	0.11290				
	138 -> 151	-0.11819				
	142 -> 148	-0.18197				
	142 -> 150	0.23158				
	142 -> 151	-0.23256				
	142 -> 158	-0.10621				
	142 -> 160	0.11048				
T11	Excited State	11:	Triplet-A	4.4800 eV	276.75 nm	f=nd <S**2>=2.000
	135 -> 144	0.15965				
	136 -> 143	0.21542				
	136 -> 144	0.42061				
	136 -> 148	-0.15051				
	136 -> 150	0.23516				
	136 -> 151	-0.25423				
T12	Excited State	12:	Triplet-A	4.5346 eV	273.42 nm	f=nd <S**2>=2.000
	133 -> 143	0.48210				
	133 -> 144	-0.18056				
	133 -> 145	0.21601				
	133 -> 146	0.30266				
S1	Excited State	13:	Singlet-A	4.8042 eV	258.07 nm	f=0.2231 <S**2>=0.000
	138 -> 144	0.10953				
	141 -> 143	0.29620				
	142 -> 143	0.28030				
	142 -> 144	0.48198				
S2	Excited State	14:	Singlet-A	4.8570 eV	255.27 nm	f=0.3873 <S**2>=0.000
	137 -> 143	0.13002				
	140 -> 143	-0.10827				
	141 -> 143	0.48721				
	141 -> 144	-0.20161				
	142 -> 143	-0.19626				
	142 -> 144	-0.24819				
T13	Excited State	15:	Triplet-A	4.8759 eV	254.28 nm	f=nd <S**2>=2.000
	132 -> 143	0.13849				
	132 -> 144	0.25651				
	132 -> 148	-0.14170				
	132 -> 150	0.20615				
	132 -> 151	-0.22354				
	135 -> 143	-0.12008				
	135 -> 144	-0.25469				
	138 -> 144	0.10314				
	142 -> 148	0.10311				
	142 -> 150	-0.14625				
	142 -> 151	0.14564				
T14	Excited State	16:	Triplet-A	4.8811 eV	254.01 nm	f=nd <S**2>=2.000
	131 -> 143	-0.30113				
	131 -> 144	0.10385				
	131 -> 145	-0.18139				
	131 -> 146	-0.26280				
	134 -> 143	0.25167				
	134 -> 144	-0.13124				
	141 -> 145	0.12478				
	141 -> 146	0.18800				
	141 -> 155	0.13486				

T15	Excited State	17:	Triplet-A	4.9711 eV	249.41 nm	f=nd	<S**2>=2.000
	135 ->	144	0.11839				
	140 ->	144	0.11557				
	142 ->	152	0.35508				
	142 ->	171	-0.10177				
	142 ->	184	0.14361				
T16	Excited State	18:	Triplet-A	5.0091 eV	247.52 nm	f=nd	<S**2>=2.000
	134 ->	143	-0.15635				
	134 ->	155	0.10916				
	137 ->	169	0.11199				
	139 ->	143	-0.17299				
	139 ->	155	-0.19472				
	140 ->	143	-0.12717				
	141 ->	147	-0.12723				
	141 ->	148	0.25926				
	141 ->	150	0.19651				
T17	Excited State	19:	Triplet-A	5.0432 eV	245.84 nm	f=nd	<S**2>=2.000
	138 ->	144	0.10964				
	138 ->	152	-0.21651				
	138 ->	171	-0.12224				
	140 ->	143	0.18120				
	140 ->	144	0.18964				
	140 ->	152	0.11439				
	141 ->	148	0.11857				
	142 ->	158	0.10200				
	142 ->	160	-0.10435				
T18	Excited State	20:	Triplet-A	5.0592 eV	245.07 nm	f=nd	<S**2>=2.000
	137 ->	143	0.14587				
	137 ->	148	0.15444				
	137 ->	150	0.11220				
	137 ->	169	-0.10376				
	139 ->	143	0.24686				
	139 ->	155	0.10251				
	139 ->	169	-0.11251				
	141 ->	155	0.18828				

Table S6. First TD- ω B97X/6-311++G(d,p) $S_0 \rightarrow S_n$ and $T_0 \rightarrow T_n$ transitions computed for dimeric **TT-Py** at its optimized geometry starting from the X-ray geometry of **TT-Py-H**.

T1	Excited State	1:	Triplet-A	3.2072 eV	386.58 nm	f=nd	<S**2>=2.000
	131 ->	143	-0.11193				
	137 ->	143	0.10204				
	141 ->	143	0.44432				
	142 ->	143	-0.34532				
T2	Excited State	2:	Triplet-A	3.3566 eV	369.37 nm	f=nd	<S**2>=2.000
	132 ->	144	0.13230				
	138 ->	144	0.12276				
	141 ->	144	0.25492				
	142 ->	144	0.41860				
	142 ->	147	0.13568				
T3	Excited State	3:	Triplet-A	3.8558 eV	321.56 nm	f=nd	<S**2>=2.000
	132 ->	144	0.14842				
	135 ->	152	0.11913				
	138 ->	144	0.11966				
	139 ->	150	0.11280				
	140 ->	147	0.12159				
	140 ->	150	0.27777				
	141 ->	144	-0.12273				
	142 ->	144	-0.14256				
	142 ->	156	-0.11396				
	142 ->	157	0.11777				
	142 ->	158	0.13054				
T4	Excited State	4:	Triplet-A	3.8833 eV	319.28 nm	f=nd	<S**2>=2.000
	131 ->	143	0.14479				
	133 ->	143	0.13074				
	133 ->	146	-0.10772				
	137 ->	143	-0.16303				
	139 ->	146	-0.19783				
	139 ->	147	-0.18547				
	139 ->	150	0.12731				
	140 ->	146	0.11393				
	141 ->	143	0.10247				
	141 ->	157	-0.14460				
	142 ->	143	-0.11489				
	142 ->	157	0.10804				
T5	Excited State	5:	Triplet-A	4.0217 eV	308.29 nm	f=nd	<S**2>=2.000
	135 ->	152	-0.11791				
	138 ->	157	-0.11490				
	139 ->	146	0.14154				

	139 -> 147	0.17377					
	140 -> 150	0.29675					
	142 -> 144	0.13605					
T6	Excited State 6:	Triplet-A	4.0331 eV	307.42 nm	f=nd	<S**2>=2.000	
	133 -> 152	0.10051					
	137 -> 157	-0.11648					
	139 -> 146	0.20703					
	139 -> 147	0.14273					
	139 -> 150	-0.19341					
	140 -> 146	-0.13313					
	140 -> 147	-0.19267					
	140 -> 150	-0.10085					
	142 -> 143	-0.11556					
T7	Excited State 7:	Triplet-A	4.1010 eV	302.33 nm	f=nd	<S**2>=2.000	
	135 -> 150	0.10023					
	137 -> 150	-0.14478					
	138 -> 147	0.16467					
	138 -> 150	0.30113					
	140 -> 144	0.17882					
	140 -> 150	-0.10401					
	140 -> 156	0.11386					
	140 -> 157	-0.15748					
	140 -> 158	-0.14418					
	142 -> 150	-0.14184					
T8	Excited State 8:	Triplet-A	4.1219 eV	300.79 nm	f=nd	<S**2>=2.000	
	137 -> 146	-0.24954					
	137 -> 147	-0.22341					
	137 -> 150	0.12666					
	138 -> 146	-0.11708					
	138 -> 150	0.11458					
	139 -> 143	-0.18434					
	139 -> 156	0.10748					
	139 -> 157	0.15353					
	140 -> 157	-0.10112					
	141 -> 146	0.11708					
T9	Excited State 9:	Triplet-A	4.3835 eV	282.85 nm	f=nd	<S**2>=2.000	
	133 -> 143	0.26243					
	133 -> 146	0.10746					
	133 -> 147	-0.16200					
	137 -> 143	-0.15247					
	137 -> 146	-0.10211					
	137 -> 147	0.11256					
	141 -> 143	-0.10935					
	141 -> 144	-0.10908					
	141 -> 146	-0.15416					
	141 -> 147	0.18313					
	141 -> 152	0.17819					
	142 -> 146	0.14592					
	142 -> 147	-0.14682					
T10	Excited State 10:	Triplet-A	4.4213 eV	280.42 nm	f=nd	<S**2>=2.000	
	135 -> 144	0.28913					
	135 -> 146	0.10765					
	135 -> 151	-0.12849					
	135 -> 152	0.16691					
	138 -> 144	-0.14225					
	138 -> 152	-0.11049					
	141 -> 146	-0.12202					
	142 -> 144	-0.10291					
	142 -> 146	-0.14810					
	142 -> 151	0.13764					
	142 -> 152	-0.21071					
	142 -> 157	-0.11387					
T11	Excited State 11:	Triplet-A	4.4703 eV	277.35 nm	f=nd	<S**2>=2.000	
	134 -> 143	-0.17968					
	134 -> 147	0.10152					
	136 -> 144	0.44423					
	136 -> 146	0.18564					
	136 -> 151	-0.19605					
	136 -> 152	0.25095					
	136 -> 157	0.10504					
T12	Excited State 12:	Triplet-A	4.4766 eV	276.96 nm	f=nd	<S**2>=2.000	
	134 -> 143	0.43306					
	134 -> 144	0.12019					
	134 -> 146	0.19795					
	134 -> 147	-0.23648					
	134 -> 152	-0.14230					
	135 -> 143	0.14088					
	136 -> 144	0.16668					
	136 -> 152	0.12928					
S1	Excited State 13:	Singlet-A	4.8331 eV	256.53 nm	f=0.1021	<S**2>=0.000	
	137 -> 143	-0.10158					
	141 -> 143	-0.42172					

141 -> 144	0.12627					
141 -> 147	0.10482					
142 -> 143	0.42483					
142 -> 144	0.19054					
T13 Excited State 14:	Triplet-A	4.8642 eV	254.89 nm	f=nd	<S**2>=2.000	
131 -> 143	0.16110					
131 -> 147	-0.13056					
131 -> 152	-0.16174					
132 -> 143	0.22283					
132 -> 144	0.18001					
132 -> 146	0.16885					
132 -> 147	-0.13386					
133 -> 143	-0.15679					
133 -> 144	0.10002					
135 -> 144	0.13183					
141 -> 147	0.12195					
142 -> 147	-0.10418					
142 -> 150	0.12351					
T14 Excited State 15:	Triplet-A	4.8947 eV	253.30 nm	f=nd	<S**2>=2.000	
131 -> 143	-0.11978					
131 -> 144	-0.15148					
131 -> 146	-0.14298					
132 -> 143	-0.15676					
132 -> 144	0.15573					
132 -> 151	-0.12156					
132 -> 152	0.21069					
133 -> 143	0.11004					
133 -> 144	0.12764					
135 -> 144	0.16315					
141 -> 147	-0.10810					
141 -> 150	0.12220					
142 -> 155	0.10296					
142 -> 156	0.11023					
T15 Excited State 16:	Triplet-A	4.9712 eV	249.41 nm	f=nd	<S**2>=2.000	
135 -> 144	-0.13136					
140 -> 144	0.12877					
140 -> 157	-0.10304					
141 -> 150	0.21826					
142 -> 147	0.18407					
142 -> 150	0.24920					
142 -> 152	-0.12600					
142 -> 186	-0.12964					
T16 Excited State 17:	Triplet-A	4.9867 eV	248.63 nm	f=nd	<S**2>=2.000	
133 -> 143	0.12816					
141 -> 146	0.22830					
141 -> 147	0.16822					
142 -> 146	-0.18996					
142 -> 147	-0.10636					
142 -> 150	0.18310					
S2 Excited State 18:	Singlet-A	5.0056 eV	247.69 nm	f=0.5076	<S**2>=0.000	
138 -> 144	0.11120					
141 -> 143	0.20198					
141 -> 144	0.27679					
142 -> 143	-0.12242					
142 -> 144	0.48072					
142 -> 147	0.11216					
T17 Excited State 19:	Triplet-A	5.0562 eV	245.21 nm	f=nd	<S**2>=2.000	
137 -> 143	0.11812					
137 -> 146	-0.12508					
137 -> 168	-0.13174					
138 -> 146	-0.10114					
138 -> 147	-0.10441					
139 -> 143	0.27378					
139 -> 157	-0.11248					
139 -> 168	-0.11313					
139 -> 177	0.11289					
140 -> 143	-0.13419					
T18 Excited State 20:	Triplet-A	5.0828 eV	243.93 nm	f=nd	<S**2>=2.000	
138 -> 147	0.12599					
138 -> 150	0.19711					
138 -> 169	0.12341					
139 -> 144	-0.12085					
140 -> 144	-0.23128					
140 -> 146	0.11699					
140 -> 150	0.10587					
140 -> 157	0.10998					
140 -> 181	0.11829					

Table S7. First TD- ω B97X/6-311++G(d,p) $S_0 \rightarrow S_n$ and $T_0 \rightarrow T_n$ transitions computed for dimeric **TT-Py** at its optimized geometry starting from the X-ray geometry of **TT-Py-X**.

T1	Excited State	1:	Triplet-A	3.1255 eV	396.68 nm	f=0.0000	<S**2>=2.000
	131 -> 143		0.13266				
	137 -> 143		0.13026				
	141 -> 143		0.57049				
T2	Excited State	2:	Triplet-A	3.2302 eV	383.83 nm	f=0.0000	<S**2>=2.000
	132 -> 144		0.13509				
	132 -> 150		0.11558				
	135 -> 150		-0.10649				
	135 -> 160		0.10506				
	138 -> 144		-0.11968				
	142 -> 144		0.54986				
	142 -> 150		-0.10656				
T3	Excited State	3:	Triplet-A	3.8797 eV	319.57 nm	f=0.0000	<S**2>=2.000
	132 -> 144		-0.15355				
	135 -> 150		0.16964				
	138 -> 144		0.16013				
	138 -> 150		-0.11082				
	140 -> 146		-0.11468				
	140 -> 150		0.10556				
	140 -> 151		0.36736				
	142 -> 144		0.12255				
	142 -> 160		-0.19333				
T4	Excited State	4:	Triplet-A	3.9091 eV	317.17 nm	f=0.0000	<S**2>=2.000
	131 -> 143		-0.15115				
	133 -> 146		0.12620				
	137 -> 143		-0.16712				
	139 -> 147		-0.20867				
	139 -> 148		0.37096				
	141 -> 143		0.10480				
	141 -> 155		0.14793				
	141 -> 156		-0.15939				
T5	Excited State	5:	Triplet-A	4.0267 eV	307.91 nm	f=0.0000	<S**2>=2.000
	132 -> 144		-0.17187				
	135 -> 150		0.23493				
	135 -> 151		-0.10912				
	138 -> 160		0.13221				
	140 -> 146		0.10109				
	140 -> 150		-0.11339				
	140 -> 151		-0.27920				
	140 -> 160		0.11887				
	142 -> 144		0.19474				
	142 -> 160		-0.12977				
T6	Excited State	6:	Triplet-A	4.0519 eV	305.99 nm	f=0.0000	<S**2>=2.000
	131 -> 143		0.17992				
	133 -> 145		0.16494				
	133 -> 146		-0.21063				
	134 -> 146		0.11526				
	137 -> 155		-0.10591				
	137 -> 156		0.12915				
	139 -> 143		-0.10343				
	139 -> 147		-0.14341				
	139 -> 148		0.25478				
	141 -> 143		-0.17575				
	141 -> 145		-0.10333				
	141 -> 146		0.12319				
T7	Excited State	7:	Triplet-A	4.1025 eV	302.22 nm	f=0.0000	<S**2>=2.000
	137 -> 147		-0.10985				
	137 -> 148		0.21039				
	138 -> 150		0.10621				
	138 -> 151		0.29897				
	139 -> 143		-0.11412				
	139 -> 155		-0.11916				
	139 -> 156		0.11886				
	140 -> 144		0.16409				
	140 -> 160		-0.16604				
	142 -> 151		0.12839				
T8	Excited State	8:	Triplet-A	4.1060 eV	301.96 nm	f=0.0000	<S**2>=2.000
	137 -> 147		-0.16416				
	137 -> 148		0.28740				
	138 -> 151		-0.20639				
	139 -> 143		-0.16369				
	139 -> 155		-0.16054				
	139 -> 156		0.15422				
	140 -> 144		-0.11948				
	140 -> 160		0.11703				
	141 -> 148		-0.11349				

T9 Excited State	9:	Triplet-A	4.3791 eV	283.13 nm	f=0.0000	<S**2>=2.000
131 -> 143		-0.17071				
133 -> 143		0.23139				
133 -> 145		0.10379				
133 -> 146		-0.12124				
134 -> 143		-0.11515				
137 -> 143		-0.13171				
137 -> 146		0.11995				
141 -> 145		-0.23901				
141 -> 146		0.30258				
141 -> 151		0.12717				
141 -> 155		0.13540				
T10 Excited State	10:	Triplet-A	4.3968 eV	281.98 nm	f=0.0000	<S**2>=2.000
132 -> 144		0.12860				
135 -> 144		0.27877				
135 -> 150		0.19620				
138 -> 144		-0.14612				
138 -> 150		-0.16762				
142 -> 149		0.10997				
142 -> 150		0.35975				
142 -> 160		0.13827				
T11 Excited State	11:	Triplet-A	4.4811 eV	276.68 nm	f=0.0000	<S**2>=2.000
136 -> 144		0.48821				
136 -> 149		0.12654				
136 -> 150		0.37040				
T12 Excited State	12:	Triplet-A	4.5428 eV	272.92 nm	f=0.0000	<S**2>=2.000
133 -> 143		0.18232				
133 -> 146		-0.11299				
134 -> 143		0.44999				
134 -> 144		0.10493				
134 -> 145		0.21316				
134 -> 146		-0.25561				
136 -> 143		0.12022				
S1 Excited State	13:	Singlet-A	4.7592 eV	260.52 nm	f=0.1637	<S**2>=0.000
137 -> 143		0.12785				
141 -> 143		0.60195				
142 -> 143		0.11101				
142 -> 144		-0.18530				
T13 Excited State	14:	Triplet-A	4.8552 eV	255.36 nm	f=0.0000	<S**2>=2.000
131 -> 143		0.24394				
131 -> 145		0.19983				
131 -> 146		-0.24437				
133 -> 143		0.33616				
141 -> 145		0.13484				
141 -> 146		-0.15616				
141 -> 155		0.10040				
141 -> 156		-0.12427				
T14 Excited State	15:	Triplet-A	4.8892 eV	253.59 nm	f=0.0000	<S**2>=2.000
132 -> 144		-0.28011				
132 -> 149		-0.10660				
132 -> 150		-0.31942				
135 -> 144		0.27200				
138 -> 144		-0.10646				
142 -> 150		-0.15077				
142 -> 151		0.20677				
142 -> 160		0.13580				
S2 Excited State	16:	Singlet-A	4.9095 eV	252.54 nm	f=0.5765	<S**2>=0.000
138 -> 144		-0.11652				
141 -> 143		0.19201				
142 -> 144		0.59653				
T15 Excited State	17:	Triplet-A	4.9789 eV	249.02 nm	f=0.0000	<S**2>=2.000
135 -> 144		-0.16156				
135 -> 151		-0.11367				
135 -> 160		-0.11697				
138 -> 144		0.11048				
140 -> 160		0.12882				
142 -> 146		-0.12273				
142 -> 150		0.18544				
142 -> 151		0.36199				
142 -> 171		-0.10343				
T16 Excited State	18:	Triplet-A	5.0008 eV	247.93 nm	f=0.0000	<S**2>=2.000
133 -> 143		-0.15088				
137 -> 148		0.10931				
137 -> 168		0.10561				
139 -> 143		0.29206				
139 -> 155		0.15622				
139 -> 156		-0.14416				
141 -> 147		0.16146				
141 -> 148		-0.29719				

T17	Excited State	19:	Triplet-A	5.0398 eV	246.01 nm	f=0.0000	<S**2>=2.000
	137 -> 143		-0.18609				
	137 -> 148		0.13852				
	139 -> 143		0.22378				
	139 -> 167		0.11501				
	139 -> 168		-0.11307				
	141 -> 147		-0.11531				
	141 -> 148		0.20375				
	141 -> 155		-0.12957				
	141 -> 156		0.15353				
T18	Excited State	20:	Triplet-A	5.0583 eV	245.11 nm	f=0.0000	<S**2>=2.000
	135 -> 144		0.10778				
	138 -> 144		0.16062				
	138 -> 151		-0.12152				
	140 -> 144		0.34680				
	140 -> 150		-0.11359				
	140 -> 160		-0.11633				
	140 -> 167		0.10070				
	140 -> 170		-0.11450				
	140 -> 171		-0.17502				

Table S8. First TD- ω B97X/6-311++G(d,p) $S_0 \rightarrow S_n$ and $T_0 \rightarrow T_n$ transitions computed for **TT-PyH⁺** in its absolute minimum.

T1	Excited State	1:	Triplet-A	2.6923 eV	460.52 nm	f=0.0000	<S**2>=2.000
	67 -> 72		0.16067				
	69 -> 72		0.21961				
	70 -> 72		0.43902				
	70 -> 73		-0.16167				
	71 -> 72		-0.38161				
	71 -> 73		0.13184				
T2	Excited State	2:	Triplet-A	3.6012 eV	344.28 nm	f=0.0000	<S**2>=2.000
	61 -> 73		-0.14203				
	63 -> 72		-0.12145				
	65 -> 72		-0.12278				
	65 -> 73		-0.19104				
	67 -> 72		0.12457				
	67 -> 73		0.29927				
	69 -> 72		0.11823				
	69 -> 73		0.25824				
	70 -> 72		0.11519				
	70 -> 73		0.30801				
	71 -> 72		-0.12249				
	71 -> 73		-0.28988				
T3	Excited State	3:	Triplet-A	3.7777 eV	328.20 nm	f=0.0000	<S**2>=2.000
	69 -> 75		0.14053				
	70 -> 72		0.16095				
	70 -> 74		0.18491				
	70 -> 75		0.19853				
	71 -> 72		0.19571				
	71 -> 74		0.20842				
	71 -> 75		0.43626				
	71 -> 82		-0.11545				
S1	Excited State	4:	Singlet-A	4.1136 eV	301.40 nm	f=0.5642	<S**2>=0.000
	67 -> 72		-0.11587				
	69 -> 72		-0.25874				
	70 -> 72		-0.39177				
	71 -> 72		0.48337				
T4	Excited State	5:	Triplet-A	4.1143 eV	301.35 nm	f=0.0000	<S**2>=2.000
	63 -> 72		0.11440				
	69 -> 72		0.21407				
	69 -> 73		-0.11205				
	69 -> 74		-0.32887				
	70 -> 72		-0.17377				
	70 -> 73		0.11427				
	70 -> 74		0.39366				
	70 -> 75		-0.11277				
	71 -> 72		-0.13294				
T5	Excited State	6:	Triplet-A	4.4009 eV	281.73 nm	f=0.0000	<S**2>=2.000
	61 -> 72		-0.16047				
	63 -> 72		0.23827				
	65 -> 72		-0.11212				
	65 -> 73		0.10740				
	67 -> 72		0.21697				
	69 -> 72		0.16258				
	69 -> 74		0.25630				
	70 -> 72		-0.10438				
	70 -> 74		-0.11698				
	70 -> 75		-0.23679				
	70 -> 82		-0.15377				

	71 -> 74	-0.11633					
	71 -> 75	0.17675					
T6	Excited State	7: Triplet-A	4.6674 eV	265.64 nm	f=0.0000	<S**2>=2.000	
	62 -> 72	-0.18994					
	63 -> 72	-0.18037					
	63 -> 73	-0.13786					
	67 -> 72	-0.10773					
	69 -> 72	0.19323					
	69 -> 74	0.11486					
	69 -> 75	0.17760					
	69 -> 82	0.14984					
	70 -> 72	-0.24489					
	71 -> 72	-0.18711					
	71 -> 73	0.16406					
	71 -> 74	0.29294					
T7	Excited State	8: Triplet-A	4.8172 eV	257.38 nm	f=0.0000	<S**2>=2.000	
	62 -> 72	0.11331					
	62 -> 73	0.11636					
	63 -> 72	0.39379					
	63 -> 73	0.23579					
	65 -> 72	0.23336					
	67 -> 72	-0.12571					
	67 -> 73	0.10626					
	69 -> 75	0.10412					
	70 -> 73	0.13275					
	70 -> 75	0.10501					
	71 -> 72	-0.11258					
	71 -> 74	0.24428					
	71 -> 75	-0.11358					
S2	Excited State	9: Singlet-A	4.9470 eV	250.63 nm	f=0.0767	<S**2>=0.000	
	69 -> 72	-0.22349					
	70 -> 72	0.51517					
	70 -> 73	-0.13774					
	70 -> 75	0.17291					
	71 -> 72	0.28598					
	71 -> 74	-0.13444					
S3	Excited State	10: Singlet-A	5.0279 eV	246.59 nm	f=0.0294	<S**2>=0.000	
	65 -> 72	0.11493					
	67 -> 72	-0.12010					
	67 -> 73	-0.15430					
	69 -> 72	-0.16547					
	69 -> 73	-0.18967					
	70 -> 73	-0.39204					
	71 -> 73	0.40927					
	71 -> 75	-0.12749					
T8	Excited State	11: Triplet-A	5.0973 eV	243.24 nm	f=0.0000	<S**2>=2.000	
	61 -> 72	-0.13938					
	63 -> 72	-0.10177					
	65 -> 74	-0.12147					
	67 -> 72	0.14284					
	69 -> 72	-0.26860					
	69 -> 73	0.10857					
	69 -> 74	0.21714					
	69 -> 75	-0.18907					
	70 -> 74	0.18253					
	71 -> 72	-0.10034					
	71 -> 74	0.27078					
	71 -> 82	0.14190					
	71 -> 91	0.13279					
S4	Excited State	12: Singlet-A	5.2901 eV	234.37 nm	f=0.0881	<S**2>=0.000	
	69 -> 72	0.41729					
	69 -> 73	-0.21077					
	69 -> 75	0.11145					
	70 -> 73	-0.16297					
	70 -> 74	0.16645					
	71 -> 72	0.27827					
	71 -> 73	0.10943					
	71 -> 75	0.28336					
T9	Excited State	13: Triplet-A	5.2939 eV	234.20 nm	f=0.0000	<S**2>=2.000	
	61 -> 72	-0.13253					
	65 -> 72	-0.14503					
	67 -> 75	-0.21887					
	69 -> 75	0.18178					
	69 -> 87	0.11145					
	70 -> 74	0.15195					
	70 -> 75	0.26963					
	70 -> 82	-0.16113					
	70 -> 94	-0.11102					
	71 -> 72	0.12889					
	71 -> 74	-0.11172					
	71 -> 75	-0.25826					
	71 -> 82	0.13483					

T10	Excited State	14:	Triplet-A	5.3507 eV	231.72 nm	f=0.0000	<S**2>=2.000
	69 -> 72		0.19382				
	69 -> 74		0.13234				
	69 -> 82		-0.21071				
	69 -> 83		0.11256				
	69 -> 87		-0.19037				
	69 -> 91		0.10867				
	70 -> 72		-0.21536				
	70 -> 73		-0.10302				
	70 -> 74		-0.17016				
	70 -> 75		0.26340				
	70 -> 82		0.11353				
	70 -> 87		0.20141				
	70 -> 91		-0.12273				
T11	Excited State	15:	Triplet-A	5.5989 eV	221.44 nm	f=0.0000	<S**2>=2.000
	63 -> 73		0.14193				
	65 -> 72		-0.11506				
	69 -> 72		0.21739				
	70 -> 74		-0.15418				
	70 -> 82		0.14641				
	71 -> 72		0.35075				
	71 -> 74		0.20318				
	71 -> 91		0.15607				
	71 -> 94		-0.18983				
T12	Excited State	16:	Triplet-A	5.7181 eV	216.83 nm	f=0.0000	<S**2>=2.000
	61 -> 72		-0.17023				
	63 -> 73		0.14361				
	65 -> 72		-0.11870				
	65 -> 75		0.10236				
	67 -> 72		0.27358				
	69 -> 72		-0.17313				
	69 -> 74		-0.17470				
	71 -> 73		0.10537				
	71 -> 74		0.17275				
	71 -> 75		-0.15836				
	71 -> 91		-0.12533				
	71 -> 94		0.17863				
S5	Excited State	17:	Singlet-A	5.7360 eV	216.15 nm	f=0.0993	<S**2>=0.000
	69 -> 72		-0.32497				
	70 -> 72		0.14179				
	70 -> 73		0.12400				
	70 -> 74		0.33852				
	71 -> 72		-0.10433				
	71 -> 73		0.10795				
	71 -> 74		0.36498				
	71 -> 75		0.20153				
T13	Excited State	18:	Triplet-A	5.9262 eV	209.21 nm	f=0.0000	<S**2>=2.000
	62 -> 73		0.16988				
	62 -> 75		0.11109				
	63 -> 72		-0.24569				
	63 -> 73		0.45162				
	63 -> 74		-0.10556				
	65 -> 73		0.11666				
	67 -> 74		-0.11711				
	70 -> 74		0.10549				
	71 -> 74		-0.13514				
T14	Excited State	19:	Triplet-A	5.9794 eV	207.35 nm	f=0.0000	<S**2>=2.000
	66 -> 72		0.57668				
	66 -> 73		-0.23427				
	66 -> 74		-0.13329				
	66 -> 75		0.12535				
	68 -> 72		0.16700				
S6	Excited State	20:	Singlet-A	6.0127 eV	206.20 nm	f=0.2447	<S**2>=0.000
	69 -> 73		0.15238				
	69 -> 74		0.27891				
	70 -> 74		-0.35314				
	71 -> 74		0.44426				

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