

Supporting Information

Engineering oxygen vacancies on dendrite-like IrO₂ for oxygen evolution reaction in acid solution

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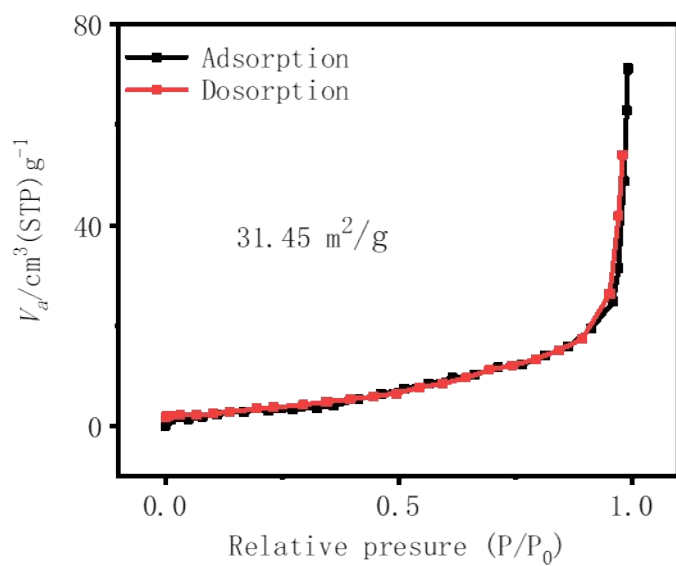


Figure S1. N₂ adsorption-desorption isotherms of IrO₂ DLNs (650 °C, 24 h).

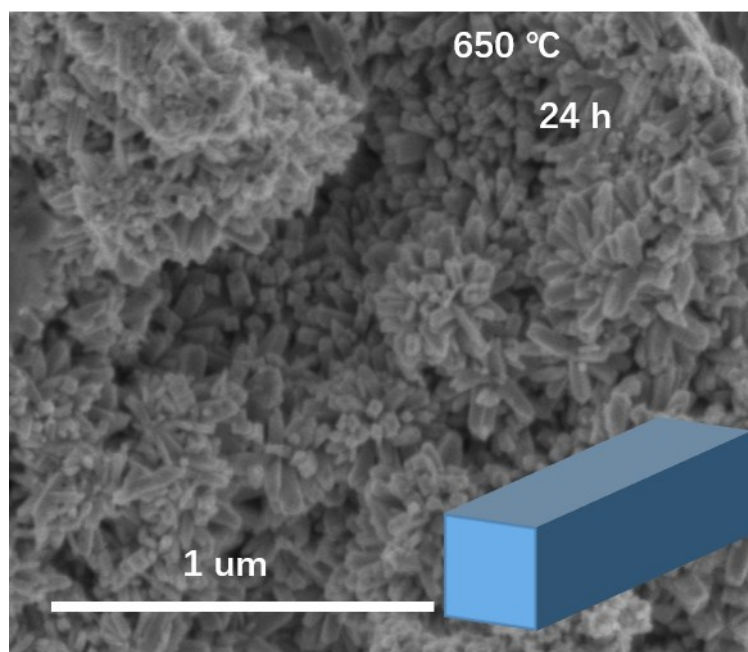


Figure S2. IrO₂ DLNs prepared at 650 °C with ageing time of 24 h.

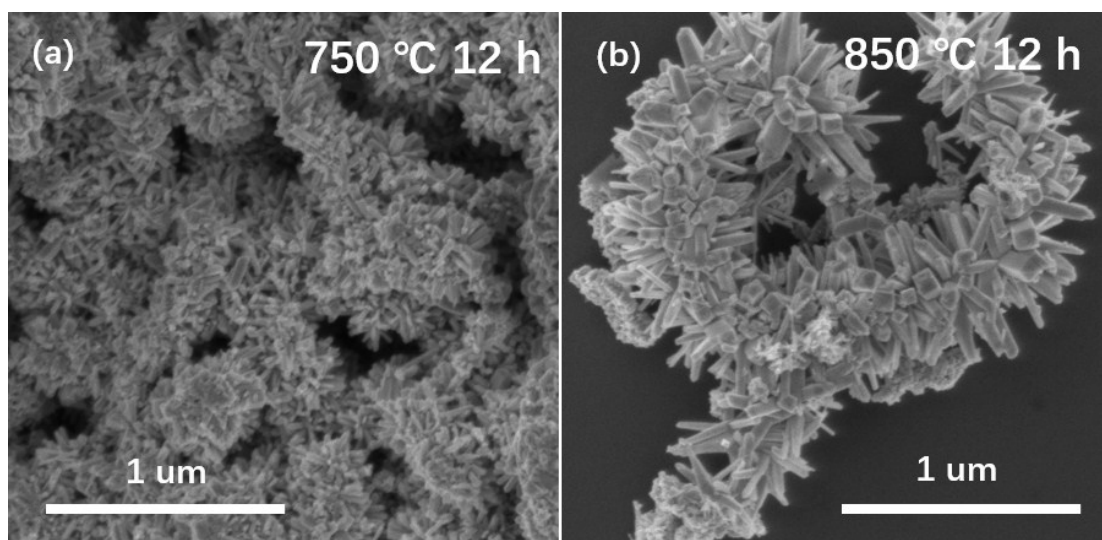


Figure S3. IrO₂ DLNs prepared at different temperature with ageing time of 12 h.

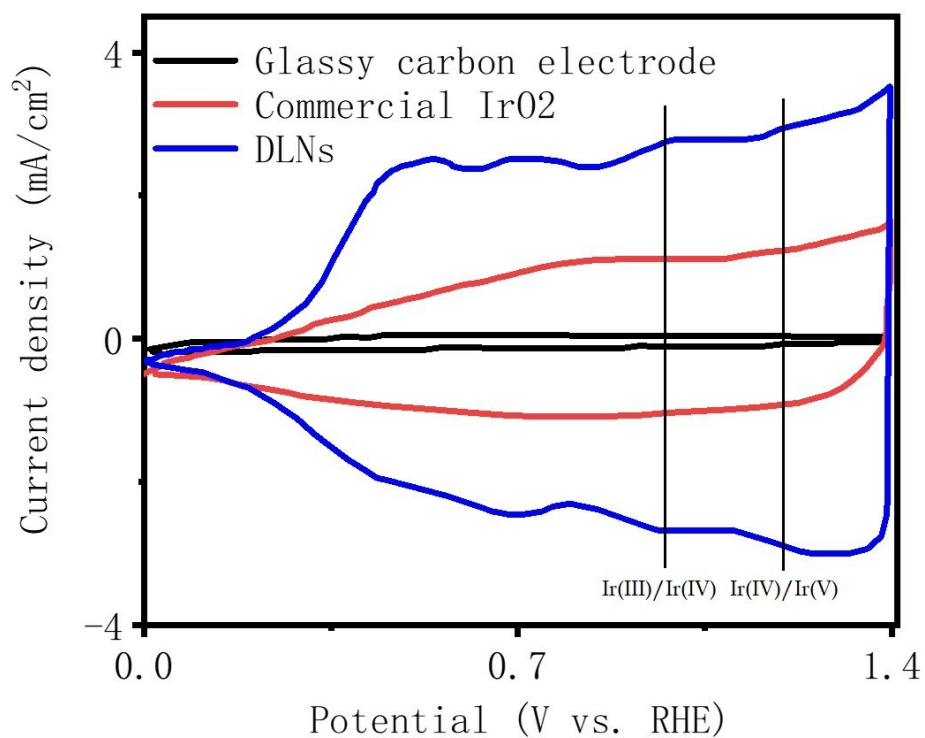


Figure S4. CV curve of support, Commercial IrO₂ and IrO₂ DLNs (650 12 h). IrO₂ DLNs show two redox peaks, i.e., Ir(III)/Ir(IV) and Ir(IV)/Ir(V), located at +0.95 V and +1.23 V vs. RHE, respectively.

Table S1. Ir-based materials for OER in acid solution.

Materials	Electrolyte	Overpotential at 10 mA/cm ² (mV)	Tafel slopes (mV/dec)	Mass activity (A/g)	Ref.
IrRuO ₃	0.5 M H ₂ SO ₄	308	-	27.45 A/g at 1.48 V vs. RHE	[1]
Ir/Fe ₄ N	0.5 M H ₂ SO ₄	316	-	77.6 A/g at 1.54 V vs. RHE	[2]
Ir nanowires	0.5 M HClO ₄	270	43.6	--	[3]
IrCoNi/C	0.5 M H ₂ SO ₄	305	53.8	--	[4]
IrCu frame	0.1 M HClO ₄	293	-	--	[5]
Y ₂ Ru ₂ O _{7-x}	0.1 M HClO ₄	2.23 mA/cm ² at 1.5 V vs. RHE	55	--	[6]
IrO ₂ needles	0.5 M H ₂ SO ₄	313	57	--	[7]
IrO ₂ DLNs	0.5 M H ₂ SO ₄	270	43	820 A/g	This

				at 1.55 V vs. RHE	work
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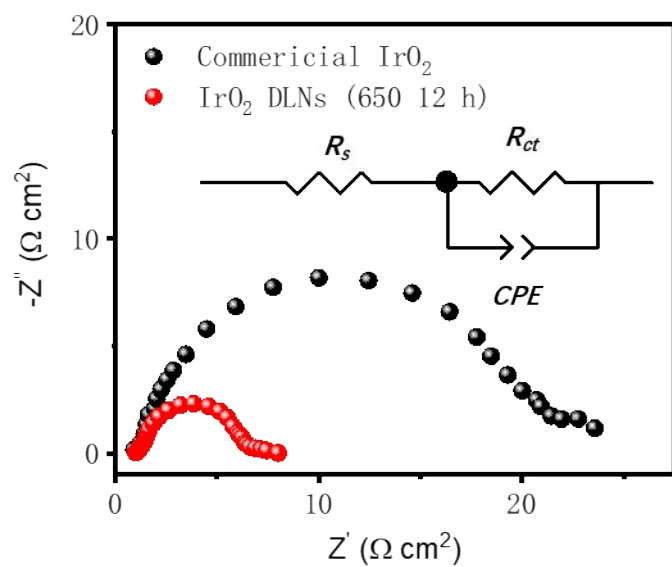


Figure S5. Nyquist plots of Commercial IrO₂ and IrO₂ DLNs (650 12 h). The EIS is measured at 1.5 V vs. RHE.

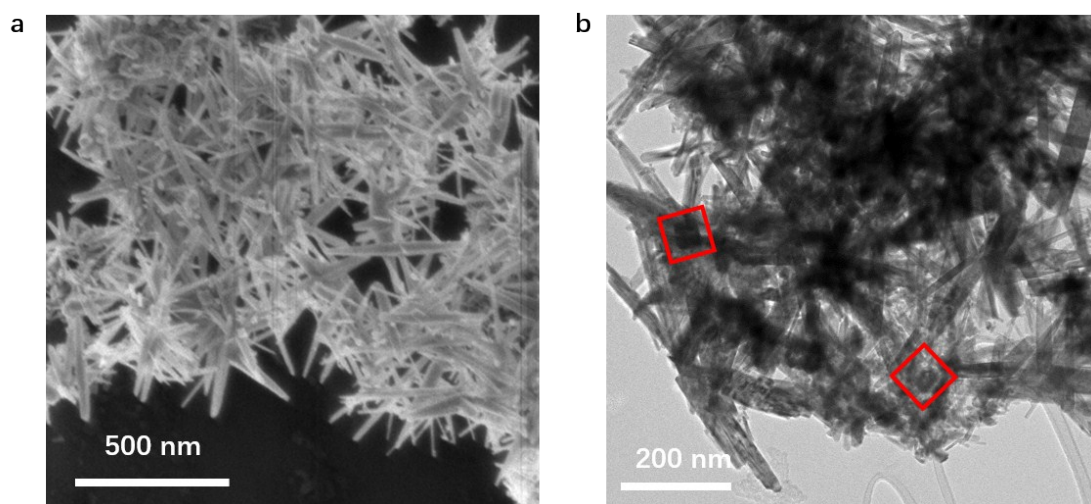


Figure S6. SEM (a) and TEM (b) image of IrO₂ DLNs (650 12 h) after use.

References

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