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Supporting Information

Synthesize Ternary and Microspheres Structure of SiO_x@SnO₂@C by

Hydrothermal Method as Anode for High Performance Lithium-Ion

Batteries

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Figure S1. Color of the mixed solution at initial (a) and after aging for 4 h at 60 °C (b).



Figure S2. The XRD pattern of SiO_x@SnO₂@C carbonized at 650 °C for 3 h (a), and the corresponding cyclic

performance (b).



Figure S3. SEM images of pristine Si nanoparticles (a), and Si@C composites (b).



Figure S4. N2 adsorption/desorption isotherms (a), DFT pore size distribution curves (b), TGA plots (c) of

SnO₂@C and SiO_x@SnO₂@C composites, respectively.



Figure S5. CV-curves of SnO₂@C (a), SiO_x@SnO₂@C (b) and Si@C (c) composites from 0.001 to 3 V at a scan

rate of 0.1 mV s⁻¹ for the first three cycles, Charge and discharge voltage profiles of SnO₂@C for first cycle (d).



Figure S6. SEM image (a) and TEM image of SiO_x@SnO₂@C electrode after 300th charge/discharge.

		cycles.		
Sample	R_{ct} -1 (Ω)	R _{ct} -100 (Ω)	Charge state-1 (V)	Charge state-100
				(V)
SnO ₂	62	28	2.76	2.46
SiO _x @SnO ₂ @C	163	45	2.98	2.54
Si@C	70	42	3.03	1.62

 $Table \ S1 \ the \ R_{ct} \ value \ and \ corresponding \ charge \ state \ of \ SnO_2, \ SiO_x @SnO_2 @C \ and \ Si@C \ of \ initial \ and \ after \ 100$