Supporting Information

Hexanedioic Acid Mediated in-situ Functionalized Interconnected

Graphitic 3D Carbon Nanofibers as Pt Support Catalysts for

Trifunctional Electrocatalysis

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Fig. S1 FTIR spectra of ACNF-0 (a), ACNF-I (b), ACNF-II (c), and ACNF-III (d).



Fig. S2 EDAX spectra of Pt/ACNFs.



Fig. S3 Elemental mapping images of Pt/ACNF-II.



Fig. S4 EIS analysis: Nyquist plot of prepared Pt/ACNF electrodes.



Fig. S5 LSV polarization curves of Pt/ACNF-0, I, and II at various rpms.



Fig. S6 K-L plots of Pt/ACNF-0, I, and II.

	XRD									
ACNEs	(0 0 2)			(1 0 0)					RAMAN	
i i i i i i i i i i i i i i i i i i i	2 θ	d ₀₀₂	FWHM	L _c	2 θ	d ₁₀₀	FWHM	La	D	I_D/I_G
	(deg)	(nm)	(2 θ)	(nm)	(deg)	(nm)	(2 θ)	(nm)	K	
ACNF-0	24.76	0.359	8.637	0.95	43.70	0.206	5.215	3.35	3.71	0.90
ACNF-I	24.87	0.357	8.562	0.96	43.90	0.206	5.198	3.36	3.72	0.89
ACNF-II	25.07	0.354	8.371	0.98	44.10	0.205	4.826	3.63	3.92	0.88
ACNF-III	24.68	0.360	8.655	0.94	43.61	0.207	5.353	3.27	3.55	0.92

Table S1. Structural parameters from XRD and Raman analysis.

ACNFs	C1s (At. %)	N1s (At. %)	O1s (At. %)	N/C	O/C
ACNF-0	85.5	9.1	5.4	0.10	0.06
ACNF-I	84.8	9.6	5.6	0.11	0.07
ACNF-II	81.9	11.2	6.9	0.14	0.09
ACNF-III	81.7	9.7	8.6	0.12	0.11

Table S2. Elemental composition of prepared ACNFs obtained using XPS analysis.

 Table S3. Comparative performance of ORR with other related recent reports on Pt based
 electrocatalysts.

Electrocatalysts	Electrolyte (H ₂ SO ₄)	Onset Potential (mV)	Half-Wave Potential (mV)	Tafel Slope (mV dec ⁻¹)	Ref
Fe@Pt/C	0.5 M	-	-	119	[S1]
2Pt- 35TiO ₂ /MWCNT	0.05 M	0.92	0.75	62 133	[S2]
Pt/MWCNT-RT	0.05 M	-	0.82	62 119	[S3]
Pt/MWCNT-B1500	0.05 M	-	0.88	68 118	[S4]
Pt/rGO-N	0.05 M	-	0.85	63 121	[85]
PtCo/10PAN-CNT	0.5 M	-	-	59.5	[S6]
Pt/mPHCNFs	0.5 M	0.998	0.917	-	[S7]
TiH ₂ S60	0.5 M	0.8	-	93	[S8]
Pt/S-MC	0.5 M	-	0.886	84	[S9]
Pt/CFx	0.5 M	-	-	109	[S10]
Pt/Ni ₃ P/CNT-CNF	0.5 M	0.729	0.499	-	[S11]
Pt/CNF 700	0.5 M	-	-	65 120	[S12]
Pt-C (Mo ₂ C) 800 C	0.5 M	0.98	0.83	125	[S13]
Pt/ACN3F-II	0.5 M	0.90	0.79	69 129	Present work

Electrocatalysts	Electrolyte	Scan rate	ECSA	Mass Activity	Ref
	$(0.5M H_2 SO_4 +)$	(mV s ⁻¹)	(m ² g ⁻¹)	(A g ⁻¹)	
Pt/C-OT-30	1M CH ₃ OH	50	72.150	-	[S14]
Pt/MWCNTs-U	1M CH ₃ OH	50	36.0	-	[S15]
Pt/Co-coal-CF	0.5M CH ₃ OH	50	-	78.5	[S16]
Pt/CXG-3s	2M CH ₃ OH	20	59	-	[S17]
Pt/Lg-CDs-800	0.5M CH ₃ OH	50	40.6	-	[S18]
Pt (10cycles)-CQD	0.5M CH ₃ OH	50	49.61	-	[S19]
PtPd SAANs	0.5M CH ₃ OH	50	-	376.0	[S20]
PtAu PNCs	0.5M CH ₃ OH	50	-	85.2	[S21]
PtNPs/TPANI-MWCNTs	0.5M CH ₃ OH	50	42.53	173	[S22]
Pt/ATO NF	0.5M CH ₃ OH	50	33	102	[S23]
Pt/PVA-CuO-Co ₃ O ₄ /CH	1.83M CH ₃ OH	100	54.56	-	[S24]
PtCu NFs	0.5M CH ₃ OH	50	63.7	1.64 (A mg ⁻¹)	[S25]
PtRu/GS-CNTs	1M CH ₃ OH	20	118.69	-	[S26]
Pt/3D-SPG	0.5M CH ₃ OH	50	79.65	-	[S27]
Pt/Ti _{0.9} Cu _{0.1} N	0.5M CH ₃ OH	50	57.5	1.56 (A mg ⁻¹)	[S28]
Pt/ACNF-II	1M CH ₃ OH	20	119.21	684.57	Present work

 Table S4. Comparative performance of MOR with other related recent reports on Pt based
 electrocatalysts.

Electrocatalysts	Electrolyte	Overpotential,	Tafel Slope	Ref	
		η ₁₀ (mV)	(mV dec ⁻¹)		
PtNi/CNFs	0.5 M H ₂ SO ₄	34	31	[S29]	
PtPd NSs	0.5 M H ₂ SO ₄	22	37	[S30]	
Pt ₆₆ Ni ₃₄ NFs	0.5 M H ₂ SO ₄	43	33	[S31]	
Pt-12	0.5 M H ₂ SO ₄	50(n ₆₀)	31	[\$32]	
Pt NPs/rGO	0.5 M H ₂ SO ₄	42	36	[S33]	
PtCu RDNFs	0.5 M H ₂ SO ₄	40	35.51	[S34]	
Pt NPs/CNFs	0.5 M H ₂ SO ₄	175	50	[\$35]	
Pt ₁₃ Cu ₇₃ Ni ₁₄ /CNF@CF	1M H ₂ SO ₄	70	38	[S36]	
Pt/HPC-14.1	0.5 M H ₂ SO ₄	24	33	[S37]	
PtNi ₂ @CNS-600	0.5 M H ₂ SO ₄	68	35.27	[S38]	
Pt ₇₅ Co ₂₅ NDAs	0.5 M H ₂ SO ₄	34	30	[S39]	
H-AgPt NCs	0.5 M H ₂ SO ₄	51	40	[S40]	
Pt@HN-BC	0.5 M H ₂ SO ₄	47	35	[S41]	
Pt/BCF	0.5 M H ₂ SO ₄	55	32	[S42]	
AC Pt-NG/C	0.5 M H ₂ SO ₄	35.28	27	[S43]	
Pt/rGO/GCE	0.5 M H ₂ SO ₄	-	33	[S44]	
Pt/ACNF-II	0.5 M H ₂ SO ₄	50	35	Present work	

 Table S5. Comparative performance of HER with other related recent reports on Pt based
 electrocatalysts.

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