Electronic Supplementary Information for:

Polymer Reinforced Carbon Fiber Interfaces for High Energy Density Structural Lithium-Ion Batteries

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Fig. S1: Galvanostatic half-cell cycling at 0.1 C for 100 cycles. Every 10\textsuperscript{th} cycle is shown for a) PAN GR\textbar Li and b) GR\textbar Li electrodes. Capacity is with respect to GR active material.
**Fig. S2:** Half-cell capacity vs cycle number for 100 cycles at 0.1 C for PAN GR|Li and GR|Li electrodes. Capacity is with respect to GR active material.
**Figure S3**

**Fig. S3**: Coulombic efficiency vs cycle number for 100 cycles at 0.1 C of a) PAN GR|Li and b) GR|Li half-cells.
Fig. S4: Galvanostatic charge-discharge curves of every 10th cycle for half-cells cycled at 0.1 C for 100 cycles for a) PAN LFP|LI and b) LFP|Li electrodes. Capacity is with respect to active material of LFP.
**Fig. S5:** Capacity with respect to LFP active material vs. cycle number for 100 cycles at 0.1C for PAN LFP|Li and LFP|Li half-cells.
Fig. S6: Coulombic efficiency for half-cells tested at 0.1 C for 100 cycles for a) PAN LFP|Li and b) LFP|Li electrodes.
Fig. S7: Coulombic efficiency for half-cells tested at 0.1 C for 100 cycles for a) PAN GR|LFP and b) GR|LFP electrodes.
**Figure S8**

**Fig. S8:** Coulombic efficiency vs cycle number for galvanostatic cycling of a) PAN GR\|LFP and b) GR\|LFP electrodes at 0.1 C for 100 cycles.
Fig. S9: Galvanostatic charge-discharge curves for second cycle of each rate of rate study for a) PAN GR|LFP and b) GR|LFP electrodes.
**Fig. S10:** Rate-study for PAN GR|LFP and GR|LFP full-cell electrodes at 0.1 C, 0.2 C, 0.5 C, 0.7 C, and 1.0 C. Capacity with respect to all mass, inactive and active material, (left y-axis) and with respect to LFP active material (right y-axis).
Fig. S11: Coulombic efficiency vs cycle number for rate study of a) PAN GR|LFP and b) GR|LFP full-cell electrodes.
**Fig. S12:** Composite layup process: a) GR anode on CF current collector with nickel tab and LFP cathode on CF current collector with aluminum tab, b) addition of separator, c) carbon fiber battery, d) epoxy impregnation, e) vacuum infusion process, f) carbon fiber composite structural battery.
Fig. S13: Tensile testing of composite structural battery.
Fig. S14: Cross-section SEMs of a) graphite and b) lithium iron phosphate electrode materials after 100 cycles at 0.1 C with and without PAN.

Figure S15
Fig. S15: Analysis of electrochemical impedance spectroscopy data from Fig. 4d, e in the manuscript a) PAN GR|LFP and b) GR|LFP full-cells before and after 100 cycles at 0.1 C, and f) capacity retention over 100 cycles at 0.1 C.1-3

|                    | PAN GR|LFP | GR|LFP |
|--------------------|-------|------|------|
| $R_S$              |       | 9.9  | 10.8 |
| $R_S + R_{CT \text{Before}}$ | 63.1  | 45.3 |
| $R_{CT \text{Before}}$ | 53.2  | 14.4 |
| $R_S + R_{CT \text{After}}$ | 247.4 | 1151.6 |
| $R_{CT \text{After}}$ | 237.5 | 1140.8 |

Modified Multifunctional Efficiency Calculations
Modified Multifunctional Efficiency ($\hat{\sigma}^{mf}$):

$$\hat{\sigma}^{mf} = 2\sqrt{\sigma^e \times \sigma^s}$$

Where $\sigma^e$ is the ratio of energy density:

$$\sigma^e = \frac{\Gamma}{\Gamma_{typ}}$$

And $\sigma^s$ is the ratio of specific strength:

$$\sigma^s = \frac{S}{S_{typ}}$$

Here, the typically energy density of a graphite|lithium iron phosphate was assumed to be 90 Wh/kg at 1C.

$$\sigma^e_{PAN \text{ coated}} = \frac{\Gamma}{\Gamma_{typ}} = \frac{39.96 \text{ Wh/kg}}{90 \text{ Wh/kg}} = 0.33$$

$$\sigma^e_{Uncoated} = \frac{\Gamma}{\Gamma_{typ}} = \frac{15.41 \text{ Wh/kg}}{90 \text{ Wh/kg}} = 0.17$$

The theoretical specific strength of the Fibre Glast System 1000 Laminating Epoxy Resin Standard Part Kit 1000/1025 as reported by the manufacturer is 257 MPa. Upon testing the strength of the epoxy in a tensile test with the full-cell carbon fiber composite, the experimental specific strength was 228 MPa.

$$\sigma^s = \frac{S}{S_{typ}} = \frac{228}{257} = 0.887$$

And the modified multifunctional advantage for each type of electrode is:

$$\hat{\sigma}^{mf}_{PAN \text{ coated}} = 2\sqrt{\sigma^e \times \sigma^s} = 2\sqrt{0.33 \times 0.887} = 1.08 \ (> 1)$$

$$\hat{\sigma}^{mf}_{Uncoated} = 2\sqrt{\sigma^e \times \sigma^s} = 2\sqrt{0.17 \times 0.887} = 0.78 \ (< 1)$$

REFERENCES
1. X.-Y. Q. Quan-Chao Zhuang, Shou-Dong Xu, Ying-Huai Qiang, Shi-Gang Sun, 

*Diagnosis of electrochemical impedance spectroscopy in lithium-ion batteries*, 2012.

