

Hierarchical Nickel Cobalt Sulfide Nanosheet Arrays Supported on CuO/Cu Hybrid Foam as a Rational-Designed Core-Shell Dendrite Electrocatalyst for Efficient Oxygen Evolution Reaction

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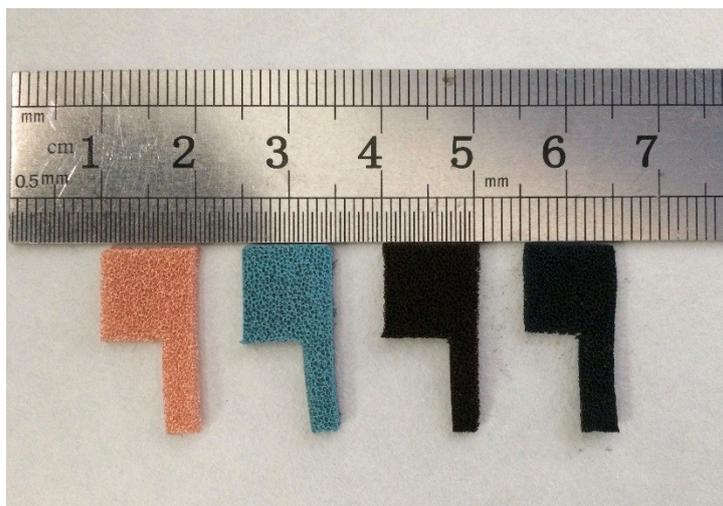


Figure S1. Optical pictures of as-prepared samples.

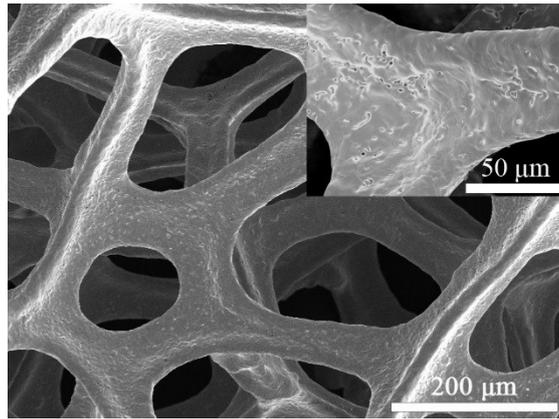


Figure S2. Typical SEM images of the blank Cu foam.

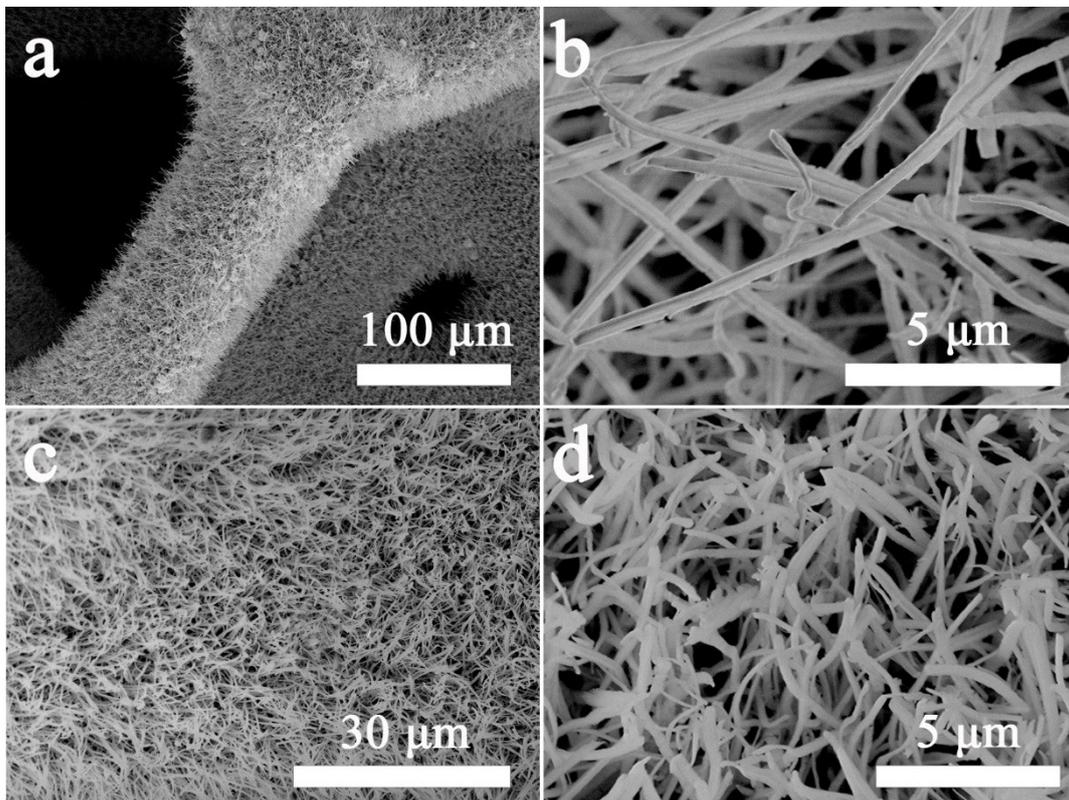


Figure S3. a, b) the SEM images of $\text{Cu}(\text{OH})_2$ and c, d) CuO nanowires.

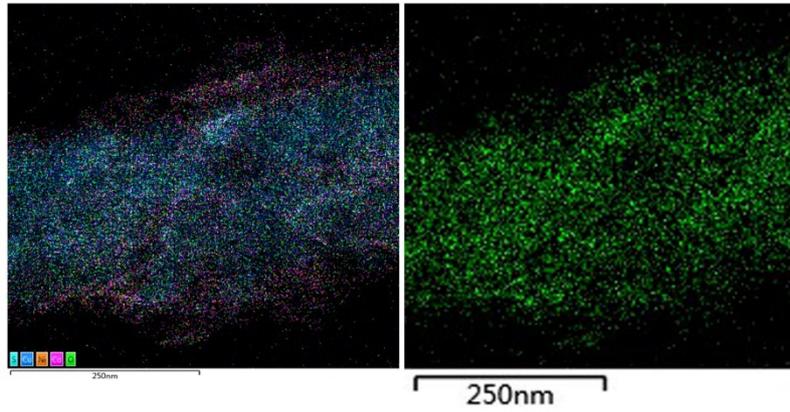


Figure S4. EDS elemental mapping image of O in the sample of Ni-Co-S@CuO/CF.

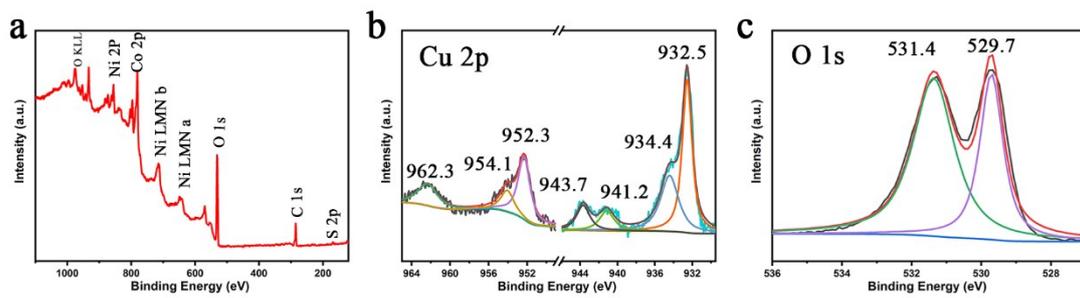


Figure S5. a) XPS full spectrum of Ni-Co-S@CuO/CF and high-resolution XPS profiles of b) Cu 2p and c) O 1s.

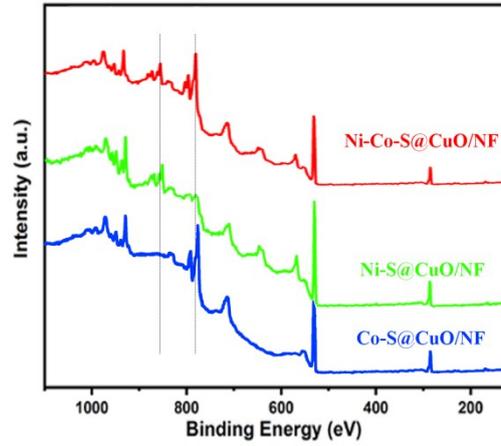


Figure S6. The XPS full spectra of Ni-Co-S@CuO/CF, Ni-S@CuO/CF and Co-S@CuO/CF.

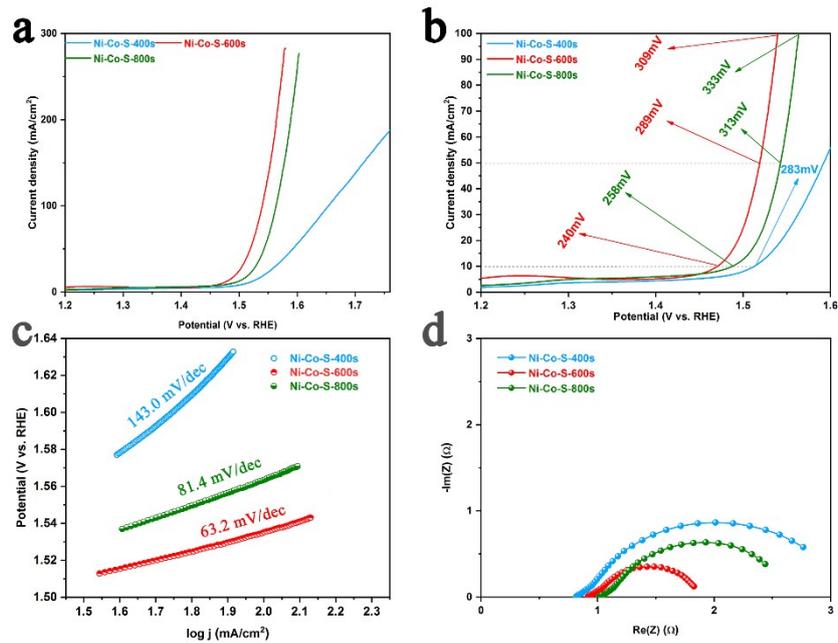


Figure S7. OER performance of Ni-Co-S-400s, Ni-Co-S-600s and Ni-Co-S-800s in 1.0 M KOH. (a, b) Polarization curves, (c) corresponding Tafel plots, and (d) EIS Nyquist plots (overpotential = 300 mV) for Ni-Co-S-400s, Ni-Co-S-600s and Ni-Co-S-800s.

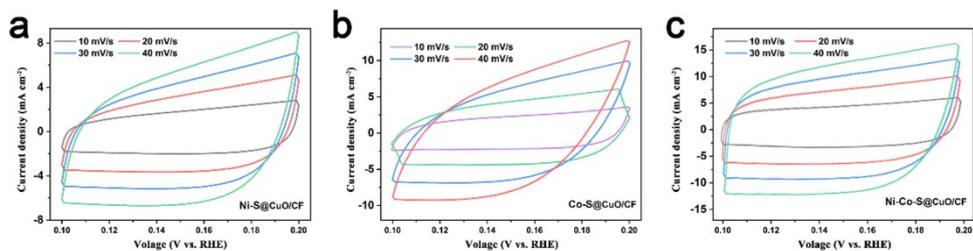


Figure S8. Typical cyclic voltammograms at different scan rate. a) Ni-S@CuO/CF, b) Co-S@CuO/CF and c) Ni-Co-S@CuO/CF at various scan rates from 5 to 40 mV/s in the potential range of 0.10-0.20 V.

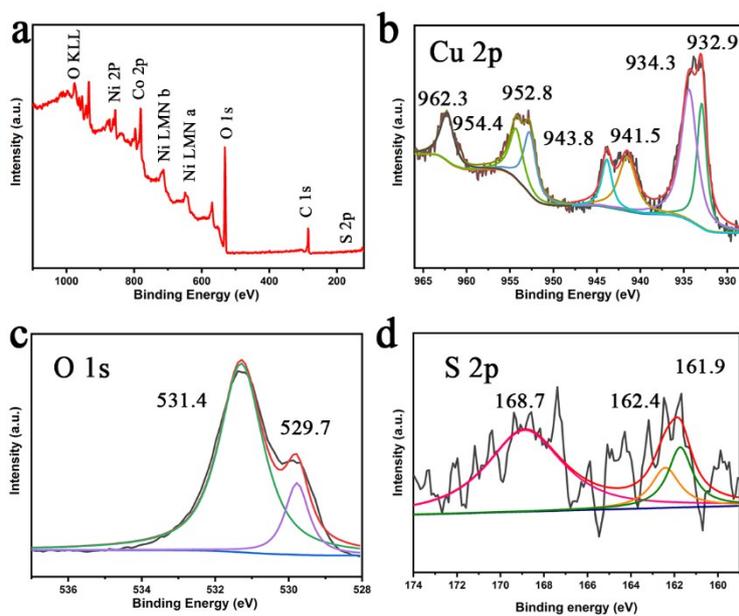


Figure S9. a) The XPS full spectrum of Ni-Co-S@CuO/CF after the OER stability test and high-resolution XPS profiles of b) Cu 2p, c) O 1s and d) S 2p.

Table S1. OER performances in KOH electrolyte: this work vs. literature.

Catalyst	Electrolyte (KOH M)	$\eta_{10 \text{ mA/cm}^2}$ (mV)	Reference
Ni-Co-S@CuO/CF	1.0	240	This work
NiFe ₂ O ₄ NF	0.1	440	1
NiCo-LDH	1.0	335	2
Co ₃ O ₄ /CNFs	1.0	330	3
CoO	1.0	269	4
Ni-Co-S	1.0	363	5
NiO-Vo@2D/3D NS@DSE	1.0	230	6
Ni-Co-mixed oxide nanocages	1.0	380	7
Ni ₃ S ₂ /NF	1.0	260	8

Reference

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