

E-supplementary data 2

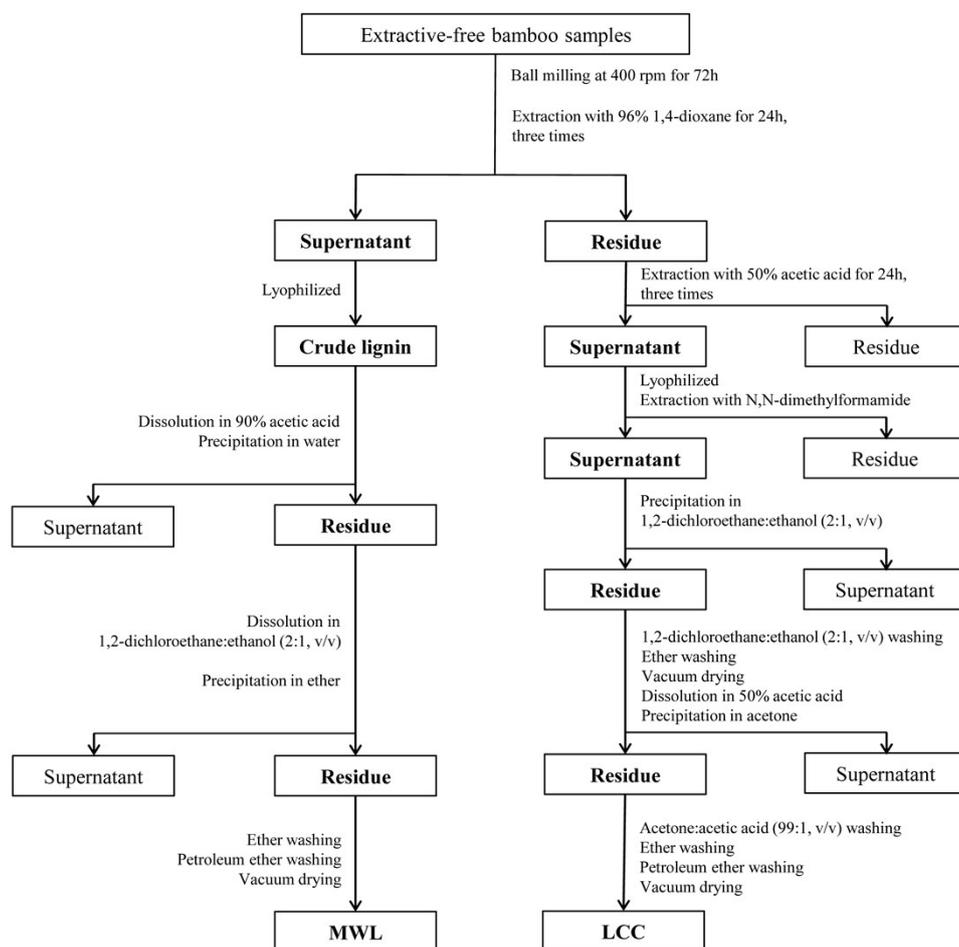


Fig. S1. The main isolation procedures of MWLs and LCCs from raw and treated bamboo.

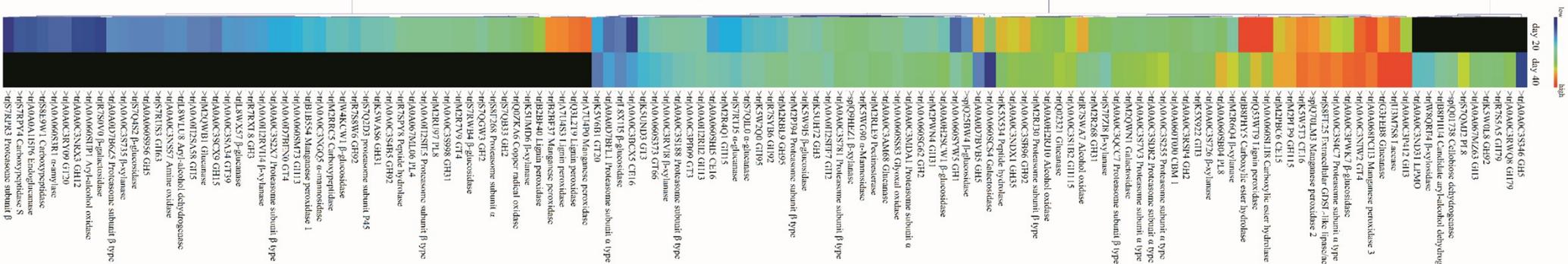


Fig. S2. Expression abundance of all secreted proteins on day 20 and day 40. Unexpressed proteins are colored black.

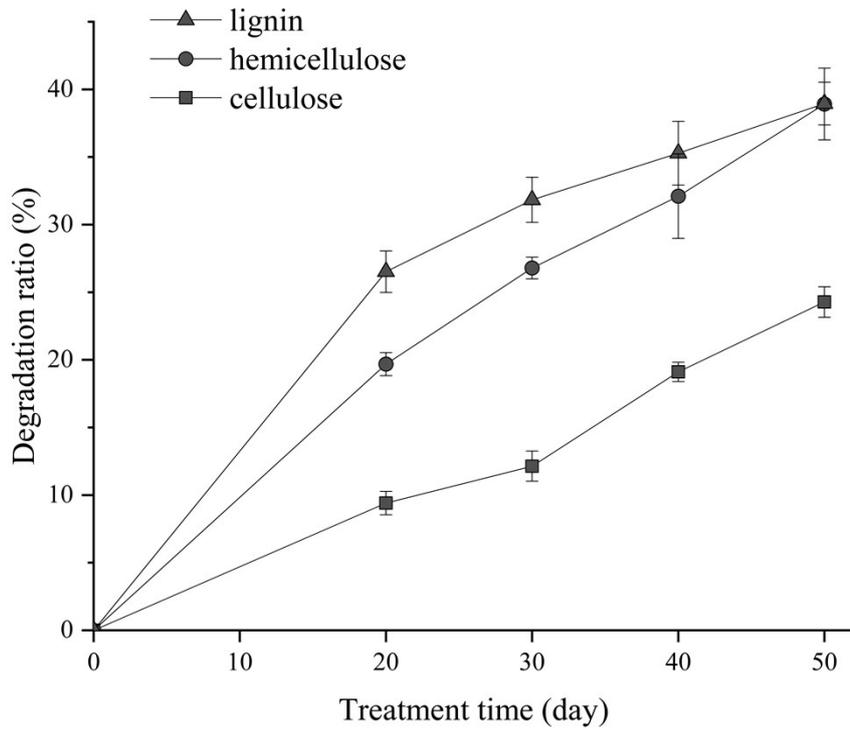


Fig. S3. The degradation ratio of lignin, cellulose and hemicellulose during fungal treatment of bamboo samples. Chemical composition including lignin, cellulose and hemicellulose composition was determined based on procedure of National Renewable Energy Laboratory (NREL, Golden, CO) [1]. The degradation ratio of the composition (%) was calculated as the formula:

$$\text{degradation ratio of day } X(\%) = \frac{\text{content of day } X(\%) \times \text{solid recovery of day } X(\%)}{\text{content of day } 0(\%)}$$

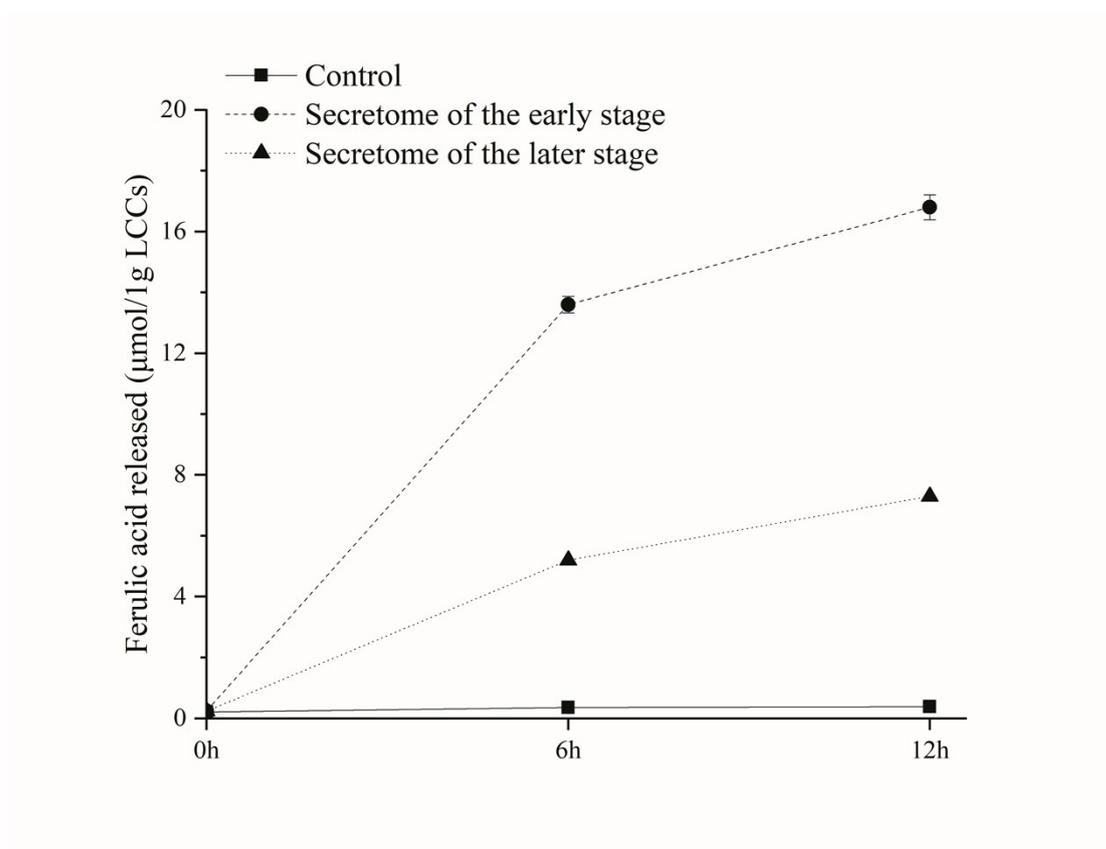


Fig. S4. FA esterase enzymatic activities of 2538 on bamboo LCCs polymers

LCCs isolated from original bamboo was used as substrate to test FA esterase enzymatic activities of extracted secretomes. 1 U of enzymatic activity was defined as 1 μmol of ferulic acid released from 1g LCCs per minute. With 13.6 μmol ferulic acid released in 6 h, 37.8 mU of FA esterase enzymatic activities was detected in fungal secretome of the early stage. At the later stage, the FA esterase enzymatic activities in fungal secretome was lower than early stage, in which only 5.2 μmol ferulic acid released in 6 h.

The reaction mixture contained 2 ml fungal secretome extract (0.05 mol/L pH 7.0 citrate-phosphate buffer as control), 1g LCCs and 0.5 ml 250 mM sodium malonate, pH 4.5. After 6 h at 25 °C, the ferulic acid of the supernatants were determined by HPLC

using standard calibration curves. The ferulic acid of secretome was also detected and excluded in the calculation. Samples were subjected to HPLC analysis with C18 column (4.6 × 250 mm) (Waters) under the condition of 1 ml/min a flow rate with methanol-10 mM phosphoric acid as the eluent at 40 °C. The vertical line on each point indicates the standard deviation for three replicates (SD, n = 3).

Table S2. The isolation yields of all MWL and LCC fractions.

Fraction	Fungal treated time (day)	Isolation yield (%)
MWL ^a	Untreated	15.42
	20	13.79
	30	13.31
	40	12.64
	50	9.52
LCC ^b	Untreated	8.16
	20	8.93
	30	7.75
	40	7.41
	50	6.32

^a % of Klason lignin

^b % of extraction-free bamboo

Table S2. The assignments of ^{13}C - ^1H peaks in HSQC spectrum from the isolated MWLs.

Label	$\delta \text{ C} / \delta \text{ H}$ (ppm)	Assignments
PCA $_{\alpha}$	144.8/7.41	C $_{\alpha}$ -H $_{\alpha}$ in <i>p</i> -coumarate (PCA) and ferulate (FA)
PCA $_{2,6}$	130.1/7.45	C $_2$ -H $_2$ and C $_6$ -H $_6$ in <i>p</i> -coumarate (PCA)
I $_{\alpha}$	128.4/6.44	C $_{\alpha}$ -H $_{\alpha}$ in cinnamyl alcohol end-groups (I)
I $_{\beta}$	128.4/6.23	C $_{\beta}$ -H $_{\beta}$ in cinnamyl alcohol end-groups(I)
H $_{2,6}$	127.8/7.22	C $_{2,6}$ -H $_{2,6}$ in <i>p</i> -hydroxyphenyl units (H)
J $_{\beta}$	126.3/6.76	C $_{\beta}$ -H $_{\beta}$ in cinnamyl aldehyde end-groups(J)
FA $_6$	123.2/7.15	C $_6$ -H $_6$ in ferulate (FA)
G' $_6$	123.2/7.33	C $_6$ -H $_6$ in guaiacyl units (G)
G $_6$	119.0/6.78	C $_6$ -H $_6$ in guaiacyl units (G)
G $_5$	115.1/6.92	C $_5$ -H $_5$ in etherified guaiacyl units (G)
PCA $_{3,5}$	115.5/6.77	C $_3$ -H $_3$ and C $_5$ -H $_5$ in <i>p</i> -coumarate (PCA)
G $_5$	114.9/6.70	C $_5$ -H $_5$ in guaiacyl units (G)
PCA $_{\beta}$	113.5/6.27	C $_{\beta}$ -H $_{\beta}$ in <i>p</i> -coumarate (PCA) and ferulate (FA)
FA $_2$	111.0/7.32	C $_2$ -H $_2$ in ferulate (FA)
G $_2$	110.9/6.99	C $_2$ -H $_2$ in guaiacyl units (G)
S' $_{2,6}$	106.3/7.32	C $_{2,6}$ -H $_{2,6}$ in oxidized S units (S')
S $_{2,6}$	103.8/6.69	C $_2$ -H $_2$ and C $_6$ -H $_6$ in etherified syringyl units (S)
T' $_{2,6}$	103.9/7.34	C' $_{2,6}$ - H' $_{2,6}$ in triclin (T)
T $_6$	98.9/6.23	C $_{2,6}$ - H $_{2,6}$ in triclin (T)
T $_8$	94.2/6.60	C $_8$ - H $_8$ in triclin (T)
B $_{\alpha}$	86.8/5.43	C $_{\alpha}$ -H $_{\alpha}$ in phenylcoumaran substructures(B)
A $_{\beta}$ (S)	85.9/4.10	C $_{\beta}$ -H $_{\beta}$ in β -O-4'-substructures linked (A) to a S unit
D $_{\beta}$	85.3/3.85	C $_{\beta}$ -H $_{\beta}$ in dibenzodioxocin substructures(D)
F $_{\alpha}'$	84.6/4.75	C $_{\alpha}'$ -H $_{\alpha}'$ in spirodienone substructures (F)
C $_{\alpha}$	84.8/4.65	C $_{\alpha}$ -H $_{\alpha}$ in β - β' resinol substructures (C)

$A_{\beta}(G)$	83.4/4.27	$C_{\beta}-H_{\beta}$ in β -O-4'substructures (A) linked to a G unit
D_{α}	83.3/4.81	$C_{\alpha}-H_{\alpha}$ in dibenzodioxocin substructures(D)
$A_{\beta}(H)$	82.9/4.48	$C_{\beta}-H_{\beta}$ in β -O-4'substructures (A) linked to a H-unit
Aox_{β}	82.7/5.22	$C_{\beta}-H_{\beta}$ in α -oxidized β -O-4' substructures(Aox)
F_{α}	81.2/5.10	$C_{\alpha}-H_{\alpha}$ in spirodienone substructures (F)
$A'_{\beta}(G)$	80.8/4.52	$C_{\beta}-H_{\beta}$ in γ -acylated β -O-4'substructures linked to a G-unit (A')
E_{α}	79.5/5.59	$C_{\alpha}-H_{\alpha}$ in α -O-4' substructures (E)
$A_{\alpha}(S)$	71.8/4.83	$C_{\alpha}-H_{\alpha}$ in β -O-4'substructures (A) linked to a S-unit
$A_{\alpha}(G)$	70.9/4.71	$C_{\alpha}-H_{\alpha}$ in β -O-4' substructures (A) linked to a G-unit
C_{γ}	71.0/3.81 and 4.17	$C_{\gamma}-H_{\gamma}$ in β - β' resinol substructures (C)
A'_{γ}	63.5/3.83 and 4.30	$C_{\gamma}-H_{\gamma}$ in γ -acylated β -O-4' substructures(A')
B_{γ}	62.6/3.67	$C_{\gamma}-H_{\gamma}$ in phenylcoumaran substructures (B)
I_{γ}	61.3/4.08	$C_{\gamma}-H_{\gamma}$ in cinnamyl alcohol end-groups (I)
F_{β}	59.5/2.75	$C_{\beta}-H_{\beta}$ in spirodienone substructures (F)
A_{γ}	59.4/3.35 – 3.80	$C_{\gamma}-H_{\gamma}$ in γ -hydroxylated β -O-4'substructures (A)
-OMe	55.6/3.73	C-H in methoxyls
C_{β}	53.5/3.05	$C_{\beta}-H_{\beta}$ in β - β' resinol substructures (C)
B_{β}	53.1/3.43	$C_{\beta}-H_{\beta}$ in phenylcoumaran substructures (B)

Table S4. The assignments of ^{13}C - ^1H peaks in HSQC spectrum from the isolated LCCs

Label	$\delta\text{ C} / \delta\text{ H}$ (ppm)	Assignments
X ₅	62.6/3.30 and 3.95	C ₅ -H ₅ in β -D-xylopyranoside
X ₂	72.6/3.2	C ₂ -H ₂ in β -D-xylopyranoside
X2 ₂	73.2/4.64	C ₂ -H ₂ in 2-O-acetyl- β -D-xylopyranoside
X ₃	73.9/3.41	C ₃ -H ₃ in β -D-xylopyranoside
X3 ₃	74.7/4.96	C ₃ -H ₃ in 3-O-acetyl- β -D-xylopyranoside
X ₄	75.4/3.68	C ₄ -H ₄ in β -D-xylopyranoside
U ₄	81.3/3.25	C ₄ -H ₄ in 4-O-methyl- α -D-GlcUA
BE _{α}	81.3/4.672	C _{α} -H _{α} in benzyl ether LCC structures
α X1(R)	92.2/5.07	(1-4)- α -D-xylopyranoside (R)
β X1(R)	97.4/4.26	(1-4)- β -D-xylopyranoside (R)
X2 ₁	99.2/4.5	2-O-acetyl- β -D-xylopyranoside
PhGlc ₂	101.2/4.65	phenyl glycoside linkages
PhGlc ₃	101.45/4.89	phenyl glycoside linkages
X3 ₁	101.5/4.27	3-O-acetyl- β -D-xylopyranoside
X ₁ /Glc ₁	103.0/4.16	β -D-xylopyranoside/ β -D-glucopyranoside
GlcAE	99.1/4.68	esterified 4-O-methyl- α -D-glucuronic acid units
esterified FA _{β}	116.5/6.33	C _{β} -H _{β} in esterified ferulate (FA)

Table S5. Sugar and Lignin Analysis of MWL and LCC Preparations.

Preparation	Chemical Composition ^a (% of Relative Content)		Carbohydrate Content ^b (% of Relative Molar Content)			
	Lignin ^b	Carbohydrate	Glc	Xyl	Ara	Gal
MWL-untreated	96.4	3.6	84.7	10.2	2.9	2.2
MWL-20d	95.6	4.4	79.9	12.4	4.6	3.1
MWL-30d	98.2	1.8	83.5	9.4	4.2	2.9
MWL-40d	97.1	2.9	89.2	6.7	3	1.1
MWL-50d	96.7	3.3	80.5	11.9	5.6	2
LCC-untreated	29.4	70.6	32.8	62.3	3.7	1.2
LCC-20d	26.4	73.6	27.3	68.4	2.9	1.4
LCC-30d	28.2	71.8	30.3	64.9	4.1	0.7
LCC-40d	25.7	74.3	36.1	59.7	2.5	1.7
LCC-50d	26.1	73.9	28.6	65.1	5	1.3

^a Relative to MWL or LCC sample (%);

^b including acid-soluble lignin and Klason lignin;

^c Glc, glucose; Xyl, xylose Ara, arabinose; Gal, galactose.

Chemical composition was determined based on reference [2].

Reference

1. Sluiter A, Hames B, Ruiz R, Scarlata C, Sluiter J, Templeton D, Crocker D. Determination of structural carbohydrates and lignin in biomass. *Lab. Anal. Proced.* 2008.
2. Yue P-P, Hu Y-J, Fu G-Q, Sun C-X, Li M-F, Peng F, Sun R-C: Structural Differences between the Lignin-Carbohydrate Complexes (LCCs) from 2- and 24-Month-Old Bamboo (*Neosinocalamus affinis*). *Int. J. Mol. Sci.* vol. 19; 2017.