

Sustainable development of vanadium pentoxide@Carbon composites derived hibiscus sabdariffa family's for supercapacitor applications

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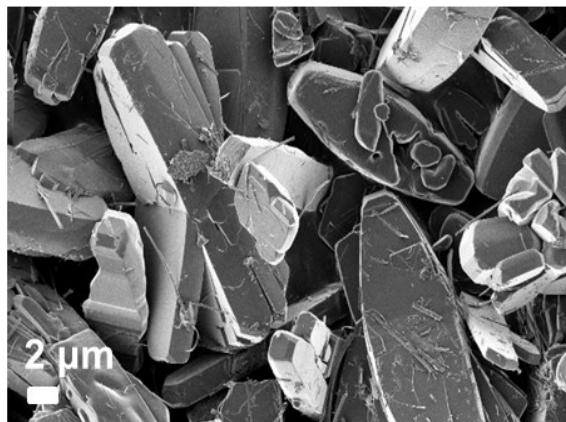


Figure S1: The SEM images of the V₂O₅-DI powder obtained with DI-Ionized water as a solvent

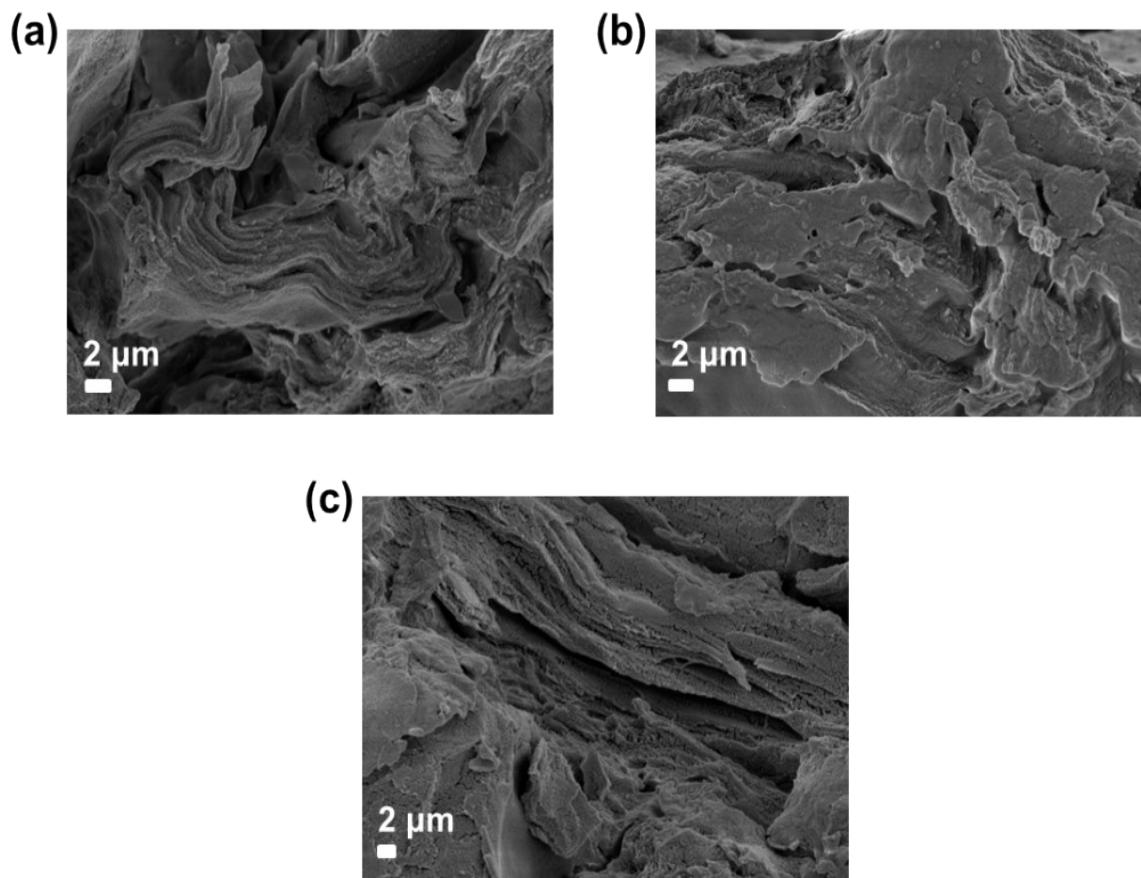


Figure S2: The SEM images of the as-prepared materials without H₂O₂ adding (a) V₂O₅@C-WHS, (b) V₂O₅@C-Red₁HS and (c) V₂O₅@C-Red₂HS nanostructures

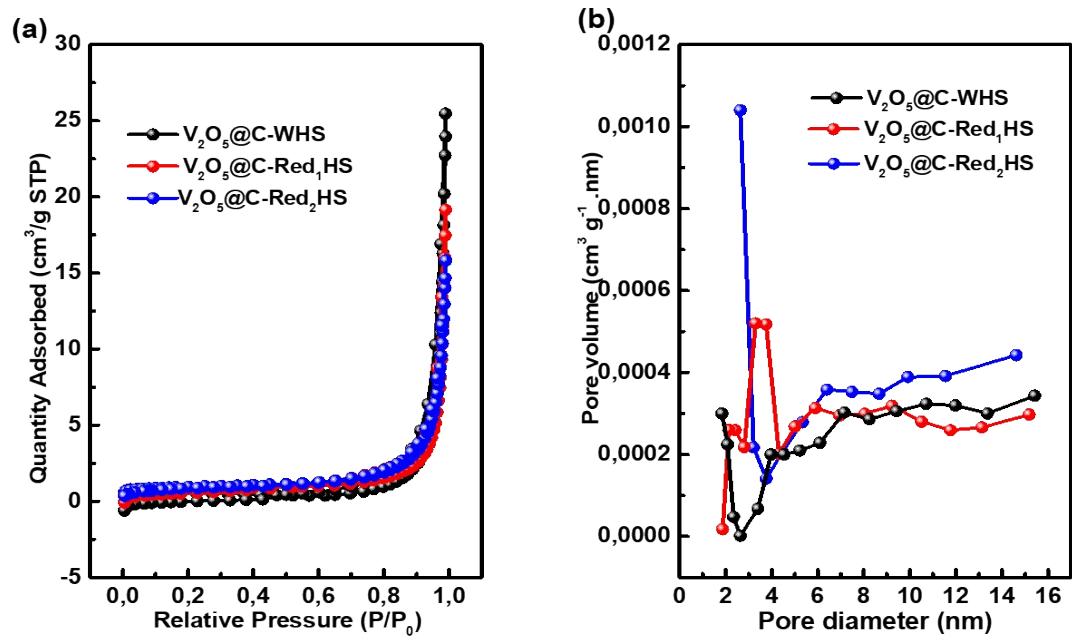


Figure S3: (a) N_2 absorption/desorption isotherms and (b) pore size distribution of $\text{V}_2\text{O}_5@\text{C-WHS}$, $\text{V}_2\text{O}_5@\text{C-R}_1\text{HS}$ and $\text{V}_2\text{O}_5@\text{C-R}_2\text{HS}$ composites

Table S1: chemical composition of all the $\text{V}_2\text{O}_5@\text{C}$ composites

Materials	V at%	O at%	C at%
$\text{V}_2\text{O}_5@\text{C-WHS}$	30.90	54.11	14.28
$\text{V}_2\text{O}_5@\text{C-R}_1\text{HS}$	31.61	52.50	15.89
$\text{V}_2\text{O}_5@\text{C-R}_2\text{HS}$	31.66	50.90	17.44

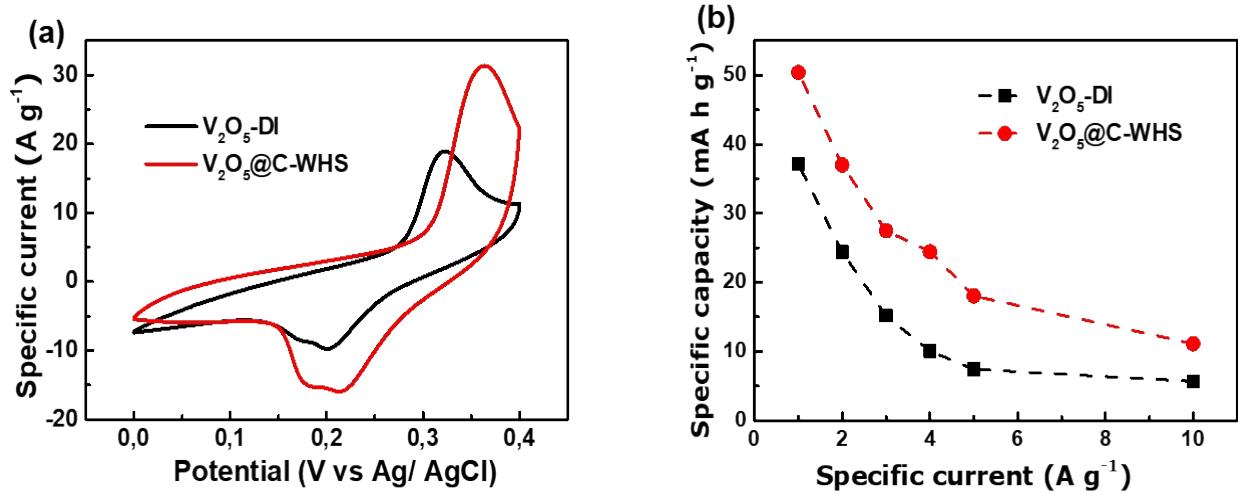


Figure S4: (a) CV plots at 50 mV/s and (b) specific capacities curves at different specific currents of the V_2O_5 -DI and $\text{V}_2\text{O}_5@\text{C-WHS}$ electrodes.

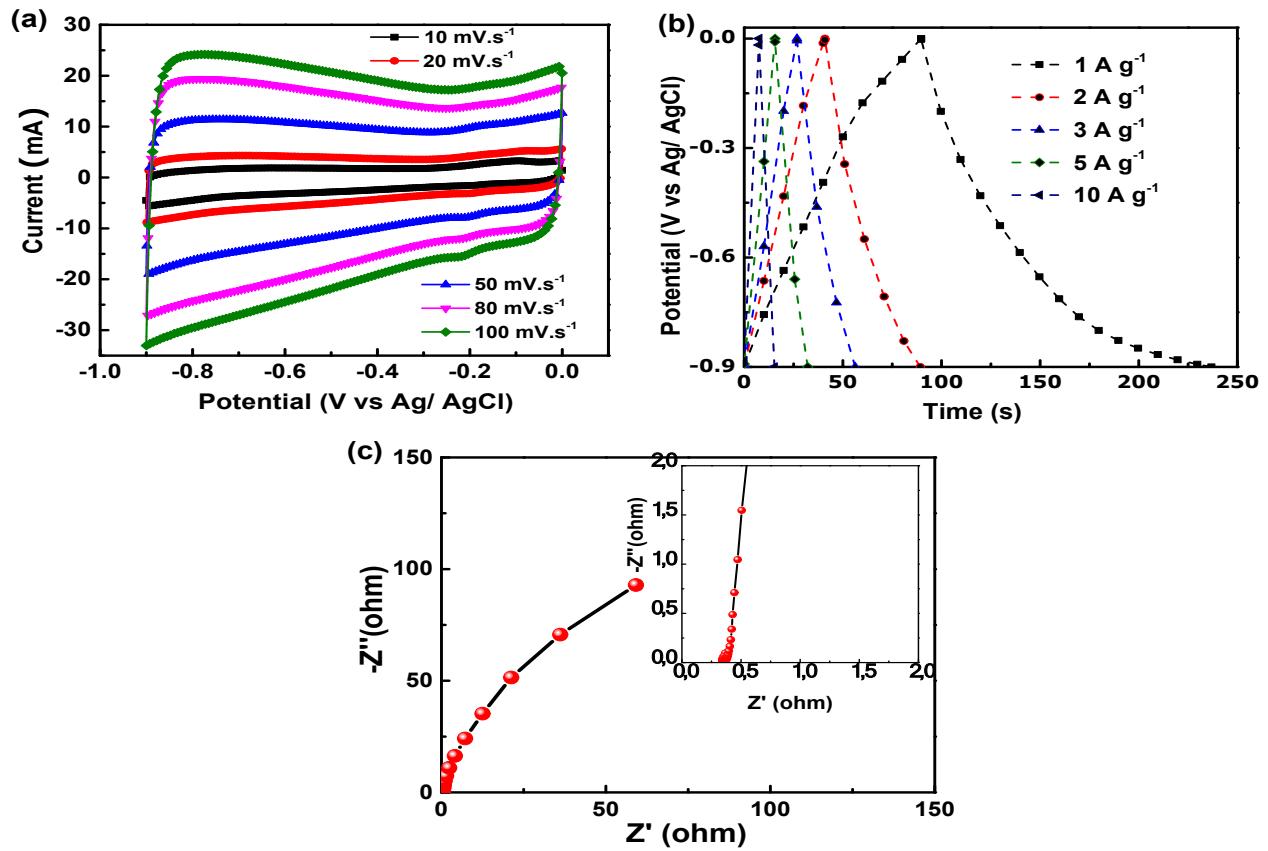


Figure S5: (a) CV curves at different scan rates, (b) GCD profiles at various specific current and (c) Nyquist plot (the inset showed the magnified plot) of the activated carbon electrode.