

Electronic Supplementary Information

Understanding of A-site non-stoichiometry in perovskites: promotion of exsolution of metallic nanoparticles and hydrogen oxidation process in solid oxide fuel cells

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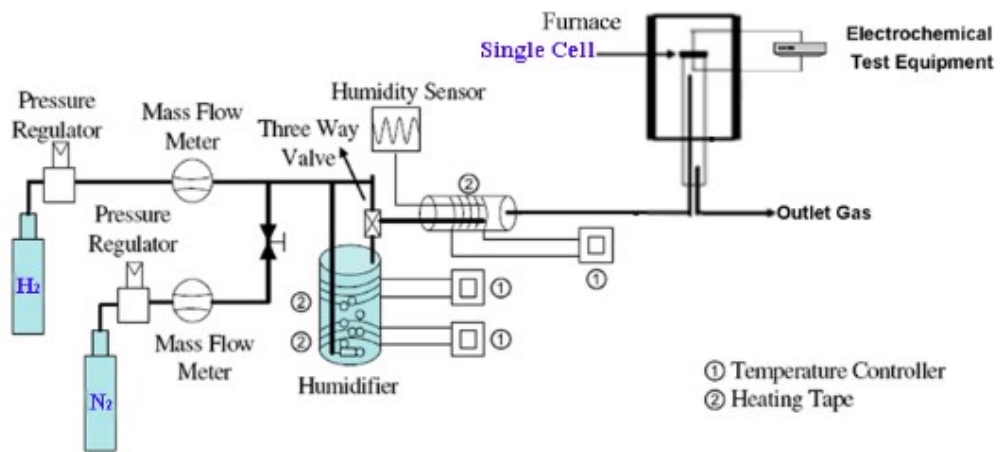


Fig. S1 Schematic diagram of the home-made setup for the measurement of the electrochemical performance of a single cell.

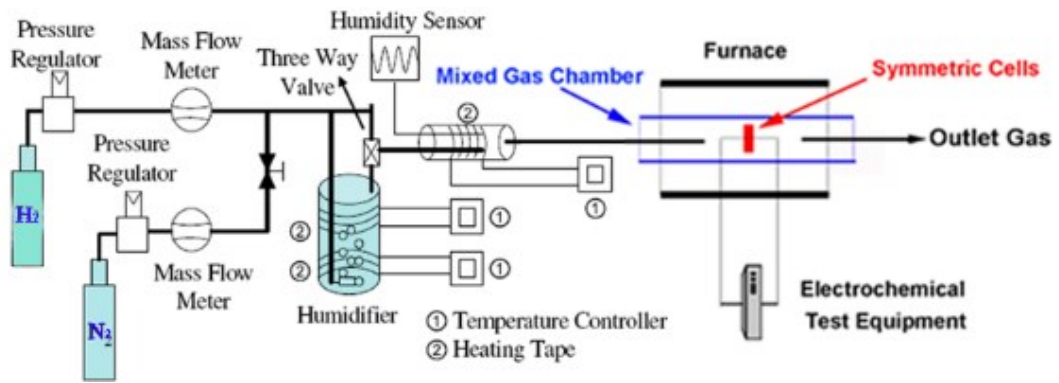


Fig. S2 Schematic diagram of the home-made setup for the measurement of the electrode polarization resistance via a symmetrical cell.

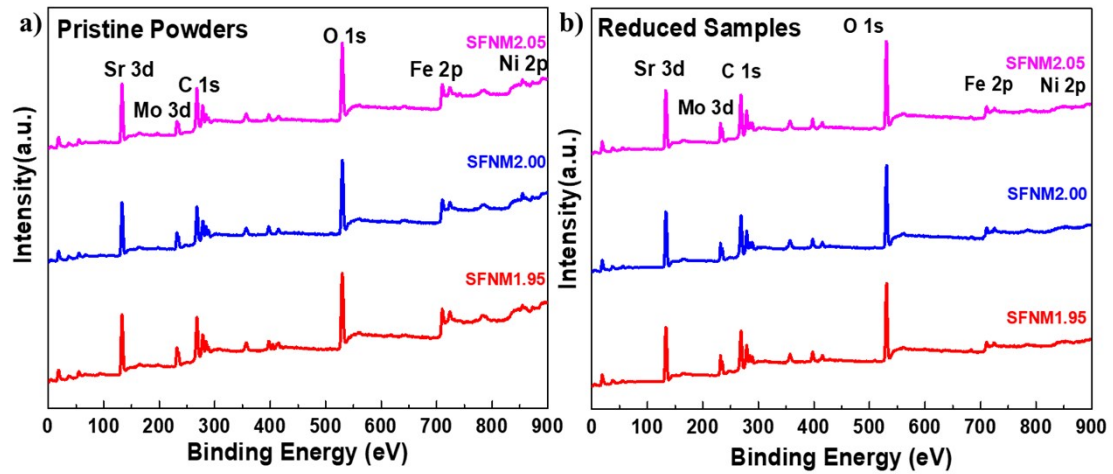


Fig. S3 XPS spectra of pristine SFNM_x powders and reduced SFNM_x powders ($x=1.95, 2.00,$ and 2.05), a) pristine SFNM_x powders, and b) reduced SFNM_x powders.

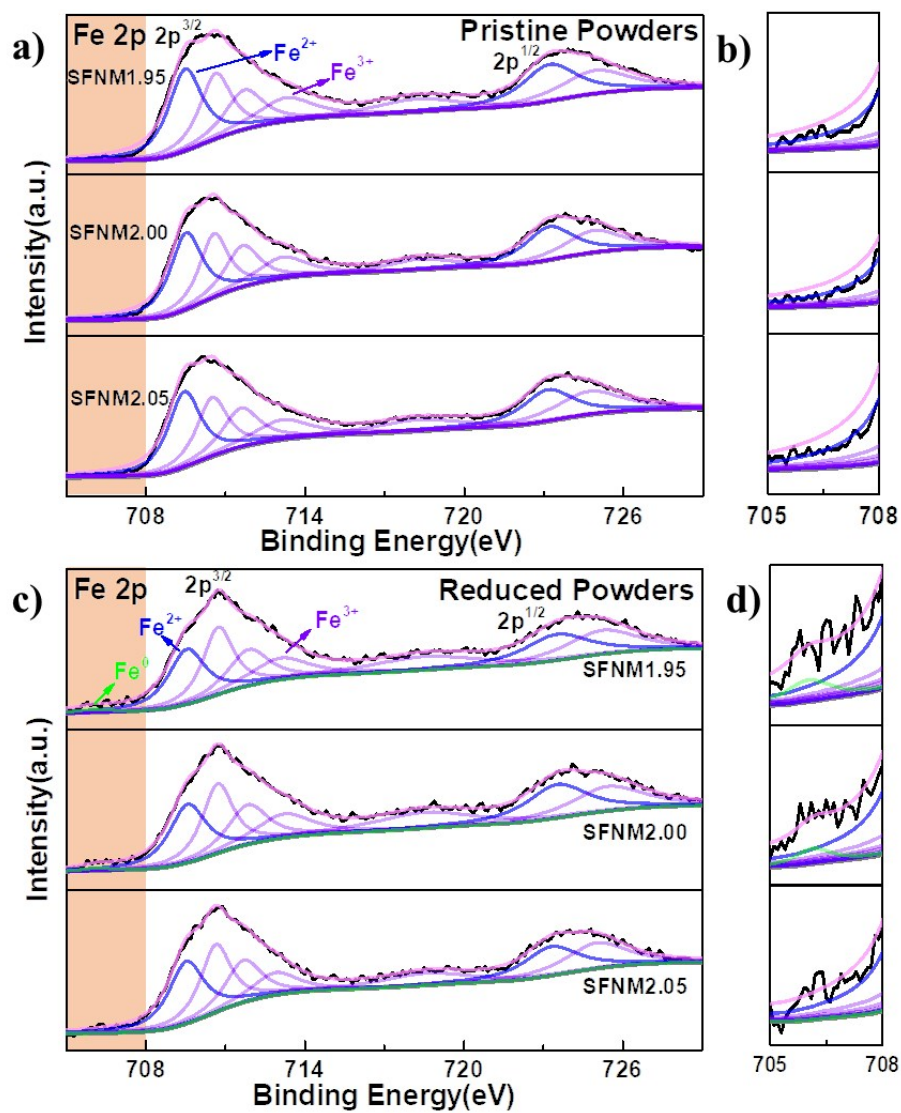


Fig. S4. XPS spectrum of Fe 2p core level regions of a-b) the pristine SFNM_x powders and c-d) the reduced SFNM_x powders (x=1.95, 2.00, and 2.05) in the binding energy range of a) and c) 705-729 eV and b) and d) 705-708 eV.

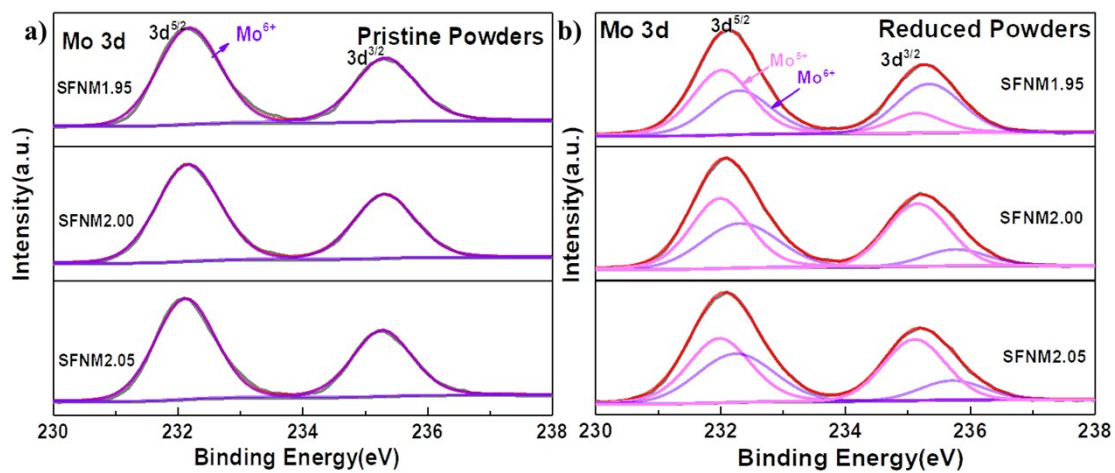


Fig. S5 XPS of Mo 3d of the core level regions of a) pristine SFNM_x powders and b) reduced SFNM_x powders ($x=1.95, 2.00, \text{ and } 2.05$).

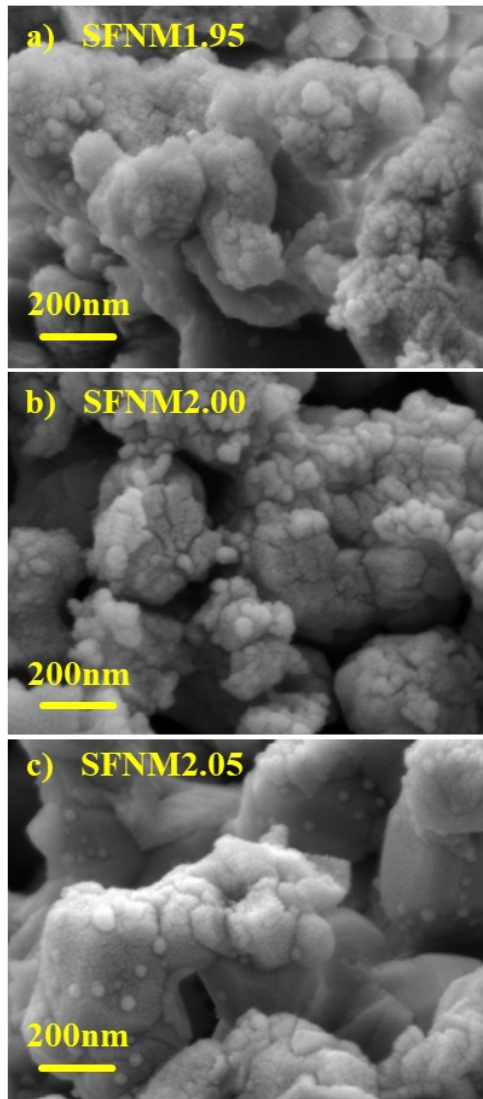


Fig. S6 Cross-sectional SEM images of a) SFNM1.95 electrode, b) SFNM2.00 electrode, and c) SFNM2.05 electrode in the symmetric cells after testing in 97% H_2 -3% H_2O .

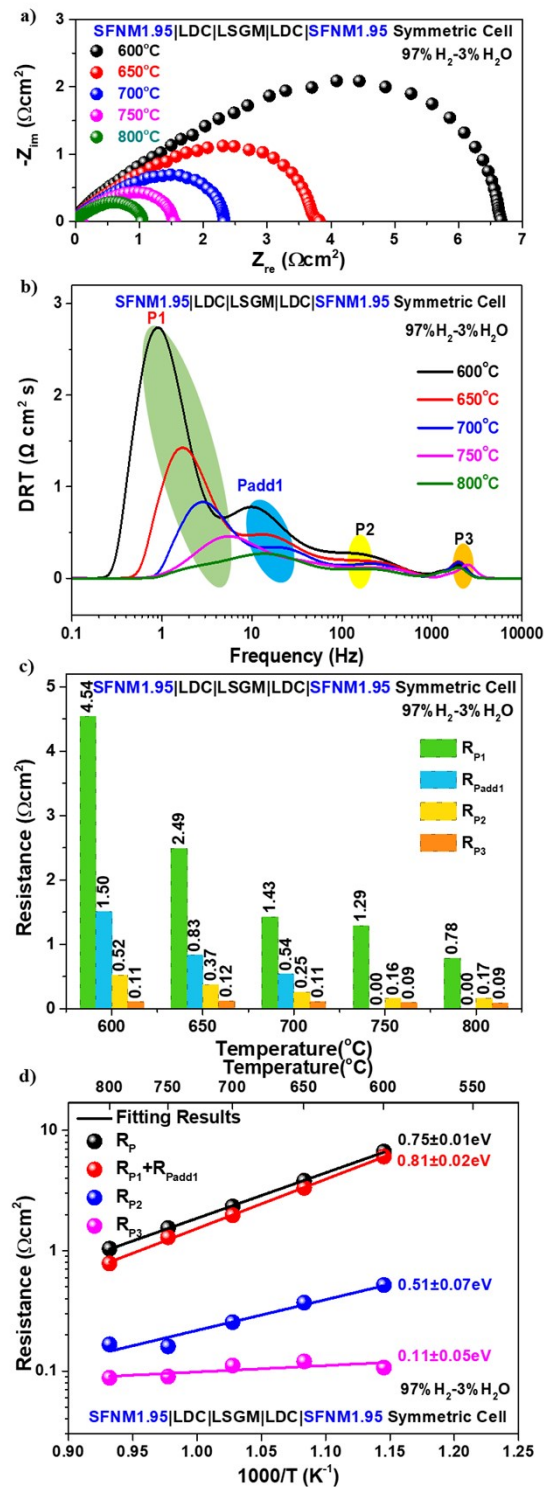


Fig. S7. a) The Nyquist plots, b) the corresponding DRT analysis results, c) resistances of the resolved peaks calculated from the DRT analysis results shown in Fig. S7b, and d) Arrhenius plots of R_p s and resistances of the resolved peaks shown in Fig. S7c of the symmetrical cell with SFMNi1.95 electrodes in 97% H₂-3% H₂O atmosphere measured at 600-800 °C.

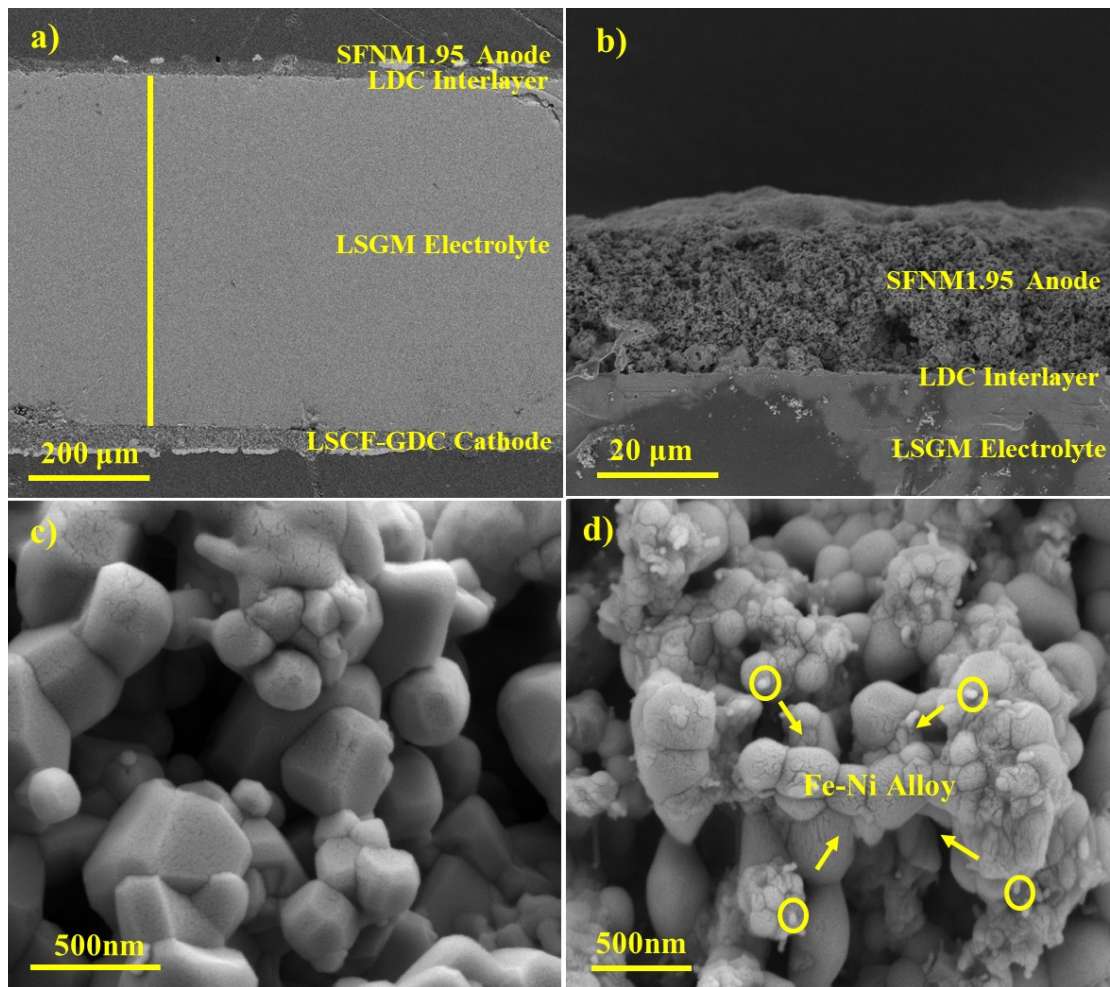


Fig. S8 Cross-sectional SEM images of a) the single cell and b) the interface of anode/interlayer/electrolyte, and SEM images of SFNM1.95 anode c) before and b) after testing.

Table S1 A table of cell performance comparison obtained with various hydrogen electrodes at 800°C using 97%H₂-3%H₂O as the feeding gas

Anode	Cathode	Electrolyte(μm)	R _p (Ωcm ²)	Ref.
SFNM1.95	LSCF-GDC	LSGM(580)	0.25	This work
Ba ₂ FeMoO ₆ (BFMO)	SCF	LSGM(300)	1.2	[64]
Sr ₂ FeMoO ₆ (SFMO)	SCF	LSGM(300)	0.91	[64]
Ni-YSZ	LSCF	YSZ(20)	0.81	[65]
SrMo _{0.9} Ga _{0.1} O _{3-δ}	SCFO	LSGM(300)	0.66	[66]
Sr _{1.9} MgMoO _{6-δ2}	BSCF	LSGM(280)	0.31	[50]
Sr ₂ Fe _{1.5} Mo _{0.5} O _{6-δ}	SFM	LSGM	0.27	[71]
La _{0.75} Sr _{0.25} Cr _{0.5} Mn _{0.5} O _{3-δ}	LSM	LSGM(120)	0.16	[72]
MDC infiltrated Ni-YSZ	LSM-YSZ	YSZ(30)	0.42	[69]
La _{0.4} Sr _{0.6} Fe _{0.7} Ti _{0.1} Co _{0.2} O _{3-δ} (CoFe@LSFTC-20)	LSFTC-20	LSGM(300)	1.47	[63]
La _{0.95} Fe _{0.8} Ni _{0.05} Ti _{0.15} O ₃ (Ni@LFNT)	LSC	LSGM(350)	0.51	[67]
(NiO) _{0.05} -(SrTi _{0.8} Nb _{0.2} O _{3-δ}) _{0.95} (Ni@STN)	LSCF	LSGM(300)	0.38	[29]
Pr _{0.8} Sr _{1.2} (Co,Fe) _{0.8} Nb _{0.2} O _{4+δ} (CoFe@K-PSCFN-CFA)	P-PSCFN	LSGM(30)	0.44	[68]
La _{1.2} Sr _{0.8} Mn _{0.4} Fe _{0.6} O ₄₋ GDC(Fe@LSMF)	LSCF-GDC	LSGM(280)	0.31	[70]

