

# Metal Cation Responsive Anionic Micogels: Behaviour Towards Biologically Relevant Divalent and Trivalent Ions

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## Nuclear Magnetic resonance (NMR)

<sup>1</sup>H NMR were recorded in deuterated solvents (CDCl<sub>3</sub> or D<sub>2</sub>O) on a Bruker 400 MHz spectrometer at room temperature, utilizing 256 scans with a 0.1 wt% microgel concentration. The chemical shifts ( $\delta$ ) are reported in ppm and are calibrated to the residual peaks of the main solvent. The collected spectra were analysed using MestReNova (v 9.1) (Mestrelab Research S.L.).

**Table S1. Synthetic conditions for the polymerization of neutral microgels with various *N*-vinylcaprolactam (VCL) and dimethylitaconate (IADME) ratios.**

Sample ID	VCL/IADME ratio	VCL		IADME		AMPA		BIS		CTAB	
		g	mmol	g	mmol	g	mmol	g	mmol	g	mmol
N0	100:0	2.087	15.00	-	-	0.053	0.19	0.060	0.38	0.010	0.027
N1	95:5	1.983	14.24	0.118	0.75	0.053	0.19	0.060	0.38	0.010	0.027
N2	90:10	7.879	13.50	0.237	1.50	0.053	0.19	0.060	0.38	0.010	0.027
N3	80:20	1.670	12.00	0.474	3.0	0.053	0.19	0.060	0.38	0.010	0.027

**Table S2. Amount of dimethylitaconate (IADME) incorporated and itaconic acid (IA) groups.**

Sample ID	IADME mol%	Theoretical value	NMR value	ATR-FTIR value	Sample	TITRATION COOH mol%
<b>N1</b>	5	5	5	5.775	<b>M1</b>	5.8
<b>N2</b>	10		10	9.395	<b>M2</b>	10
<b>N3</b>	20		20	22.950	<b>M3</b>	19

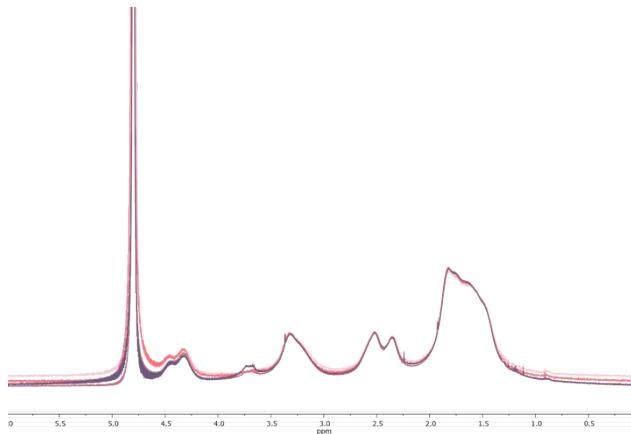


Figure S1. <sup>1</sup>H NMR spectrum of Mn microgels recorded in CDCl<sub>3</sub>. **M1** (light pink), **M2** (pink) and **M3** (purple).

**Table S3. R<sub>g</sub>/R<sub>H</sub> ratio of M1, M2 and M3 taken at 20°C, at pH 7, 1 mM Mg<sup>2+</sup> and pH=2.**

Microgel	pH 7	1 mM [M <sup>2+</sup> ]	pH 3
<b>M1</b>	0.44	0.59	0.56
<b>M2</b>	0.49	0.64	0.64
<b>M3</b>	0.45	0.66	0.65

**Table S4.**  $D_H$  of homopolymeric VCL microgel N0 in pure water and in the presence of 1mM  $M^{2+/3+}$  ions at 10 °C and neutral pH.

Sample	$H_2O$	$Mg^{2+}$	$Sr^{+2}$	$Cu^{(II)}$	$Fe^{(III)}$
N0	$375 \pm 19$	$355 \pm 20$	$355 \pm 22$	$360 \pm 27$	$340 \pm 35$

**Table S5.** Amount of  $Mg^{2+}$  ion bonded at ambient condition in PBS

Microgel	% $Mg^{2+}$	% $Fe^{3+}$
M1	0	100
M3	0	100

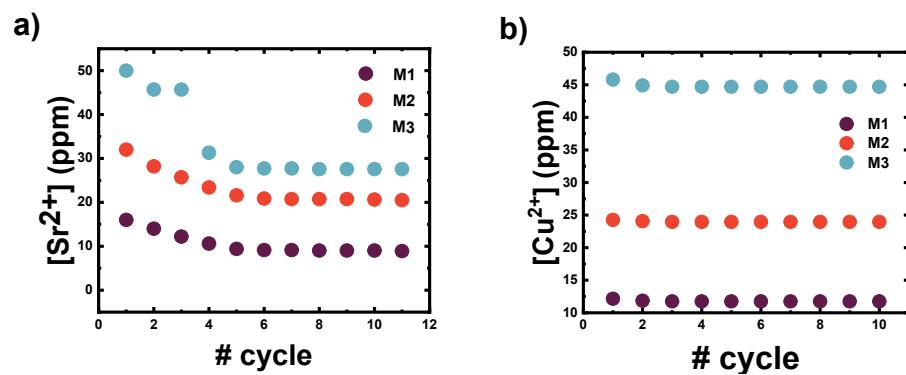


Figure S2. Amount in ppm of (a)  $Sr^{2+}$  and (b)  $Cu^{2+}$  retained by a 0.1 wt% microgel solution at room temperature and neutral pH. Microgels used ae M1 (5.8 mol% COOH), M2 (10 mol% COOH) and M3 (18 mol% COOH).