## [Electronic Supplementary Information]

## Controlling Nanostructure and Mechanical Properties in Triblock Copolymer/Monomer Blends via Reaction-Induced Phase Transitions

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## Macromolecular Characterization using NMR and SEC

<sup>1</sup>H NMR was used to determine critical molecular characteristics such as PS wt% and ratio of 1,2 vs 1,4 PBD content in the SBS triblock copolymer. The PS wt% was calculated to be 35%, which by using bulk densities of 1.04 g cm<sup>-3</sup> for PS and 0.91 g cm<sup>-3</sup> for PBD, the resulting density of the SBS triblock was calculated to be 0.95 g cm<sup>-3</sup>. Similarly, the dn/dc value of the polymer was approximated using the PS wt% by using dn/dc values of 0.185 for PS and 0.13 for PBD, resulting in a dn/dc to be approximately 0.1495. It was determined that the SBS triblock was 89% 1,4- PBD. By using the above dn/dc value, the number-average molecular weight ( $M_n$ ) of the copolymer was determined to be 62 kg mol<sup>-1</sup>.



Figure S1. 500 MHz <sup>1</sup>H NMR spectrum for  $\phi_{SBS} = 100\%$ . The PS wt% was calculated by comparing the relative mole amounts for PS and PBD. The relative 1,4 vs 1,2- PBD content was calculated by comparing group A and B hydrogens as shown in the spectrum.



**Figure S2**. SEC trace of  $\varphi_{\text{SBS}} = 100\%$  (blue) and  $\varphi_{\text{SBS}} = 2.5\%$  (red). The large shift in elution time indicates the large increase in molecular weight due to grafting of the SBS. The small hump at later elution times in the  $\varphi_{\text{SBS}} = 2.5\%$  SBS trace shows that homopolymer PS is being formed in the reaction. For  $\varphi_{\text{SBS}} = 100\%$ , the  $M_n$  and D were determined to be 62 kg mol<sup>-1</sup> and 1.11, respectively. For  $\varphi_{\text{SBS}} = 2.5\%$ , the  $M_n$  and D were determined to be 6,400 kg mol<sup>-1</sup> and 1.34, respectively.

## Glass Transition Temperature $(T_g)$

Glass transition temperatures ( $T_g$ ) were measured using a TA Instrument DSC 250. 10-20 mg of each sample were pressed into aluminum pans, heated to 120 °C at 20 °C/min, cooled to -160 °C at 20 °C/min, and heated again to 120 °C at 20 °C/min. The  $T_g$  of both the PBD and PS domains were acquired on the second heating cycle.



**Figure S3.** DSC traces for the dog bone samples after polymerization and vacuum drying. a)  $\varphi_{\text{SBS}} = 100\%$ , b)  $\varphi_{\text{SBS}} = 50\%$ , c)  $\varphi_{\text{SBS}} = 40\%$ , d)  $\varphi_{\text{SBS}} = 30\%$ , e)  $\varphi_{\text{SBS}} = 20\%$ , and f)  $\varphi_{\text{SBS}} = 10\%$ .

Sample	$T_{\rm g, PBD}$ (°C)	PS wt% in PBD <sup>a</sup>	$T_{\rm g, PS}(^{\circ}{ m C})$	PBD wt% in PS <sup>a</sup>
$\varphi_{SBS} = 100\%$	-91	9	89	3
$\phi_{SBS}=50\%$	-85	14	95	1
$\phi_{SBS} = 40\%$	-92	8	91	2
$\phi_{SBS}=30\%$	-91	9	90	2
$\phi_{SBS}=20\%$	-90	10	82	4
$\phi_{SBS}=10\%$	-91	9	81	5

Table S1. Results from DSC experiments

<sup>a</sup>Calculated using the Fox equation.



Figure S4. Tensile measurements for all the  $\phi_{SBS} = 100\%$  samples. The applied rate of strain was 5 mm/min.



Figure S5. Tensile measurements for all the  $\phi_{SBS} = 50\%$  samples. The applied rate of strain was 5 mm/min.



Figure S6. Tensile measurements for all the  $\phi_{SBS} = 40\%$  samples. The applied rate of strain was 5



mm/min.

Figure S7. Tensile measurements for all the  $\phi_{SBS} = 30\%$  samples. The applied rate of strain was 5 mm/min.



Figure S8. Tensile measurements for all the  $\varphi_{SBS} = 20\%$  samples. The applied rate of strain was 5 mm/min.



Figure S9. Tensile measurements for all the  $\phi_{SBS} = 10\%$  samples. The applied rate of strain was 5 mm/min.