

Supporting Information

In situ construction of a MOF-derived carbon-encapsulated LiCoO₂ heterostructure as a superior cathode for elevated-voltage lithium storage: from experimental to theoretical study

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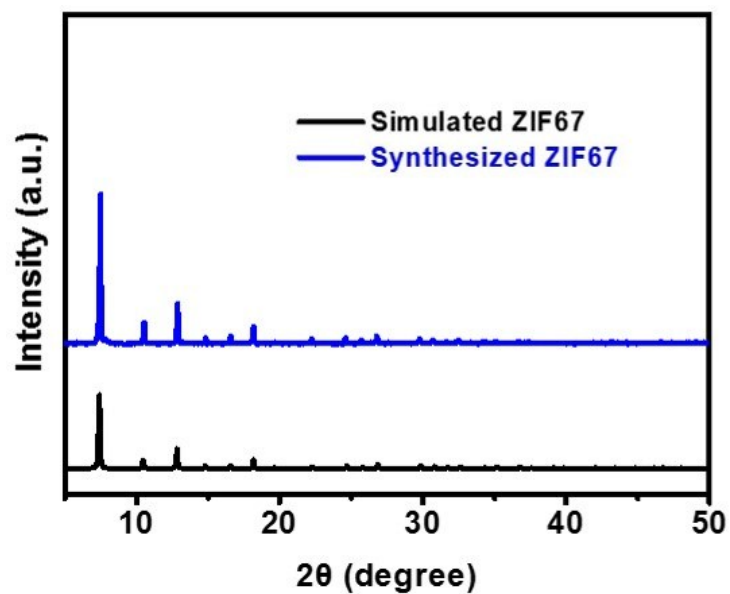


Fig. S1 XRD patterns of as-synthesized and simulated ZIF67.

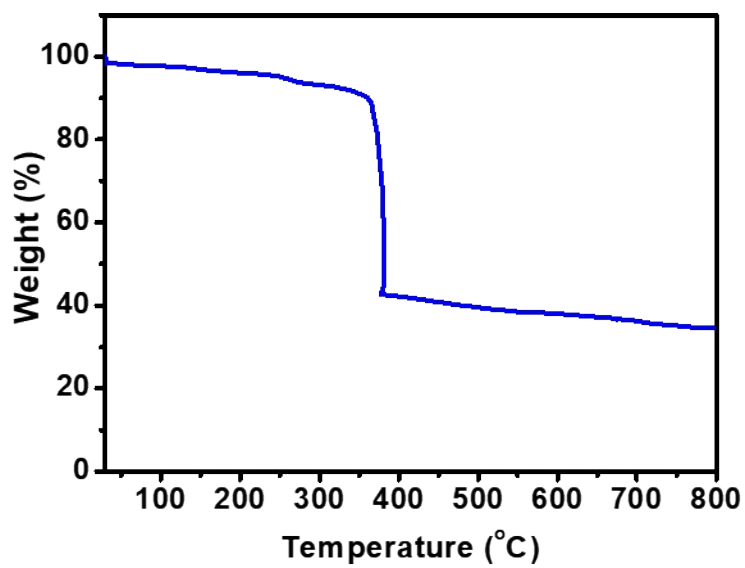


Fig. S2 TGA plot of the ZIF67 under air atmosphere.

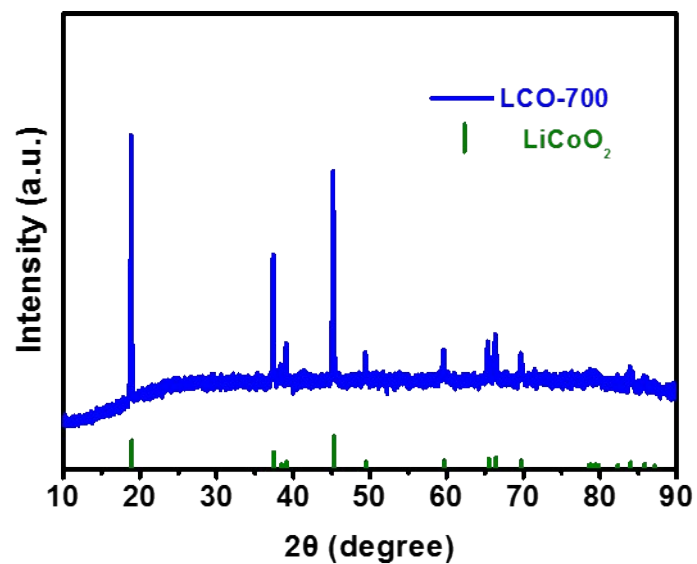


Fig. S3 XRD pattern of MOF-derived LCO-700 sample.

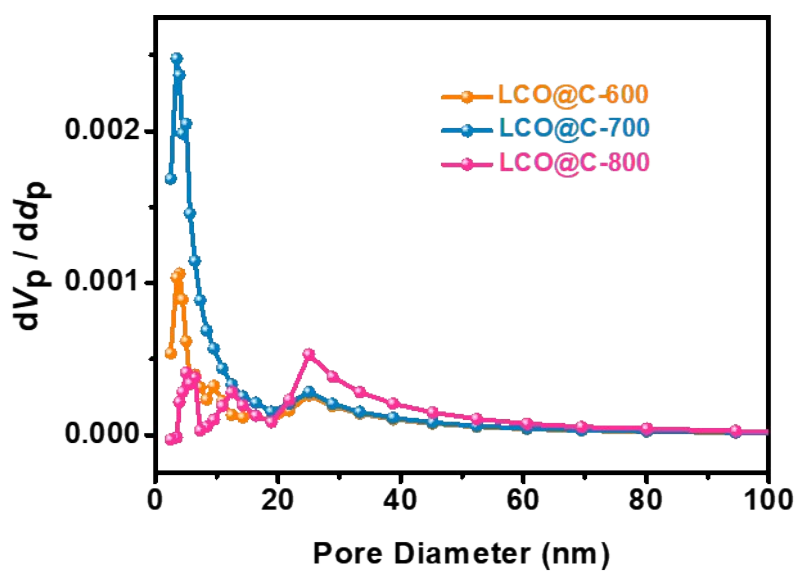


Fig. S4 Pore-size distribution curves of the LCO@C-600, -700, and -800, respectively.

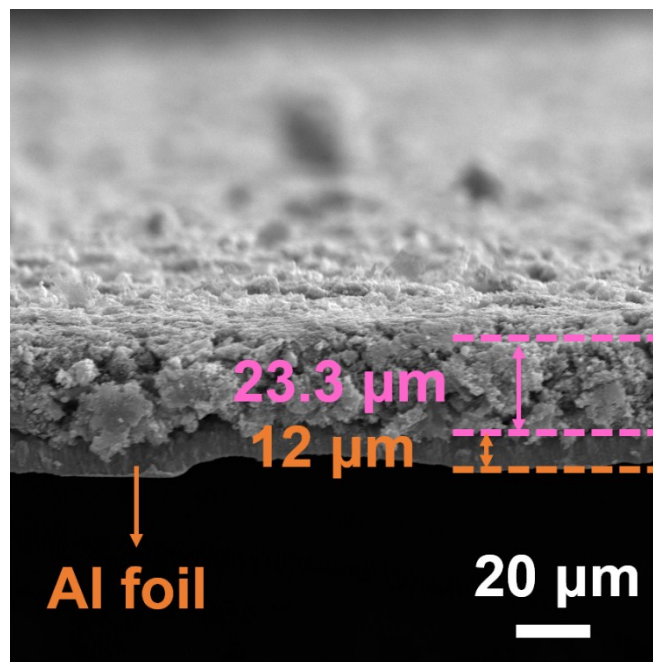


Fig. S5 The cross-section view SEM image of the LCO@C-700 electrode film.

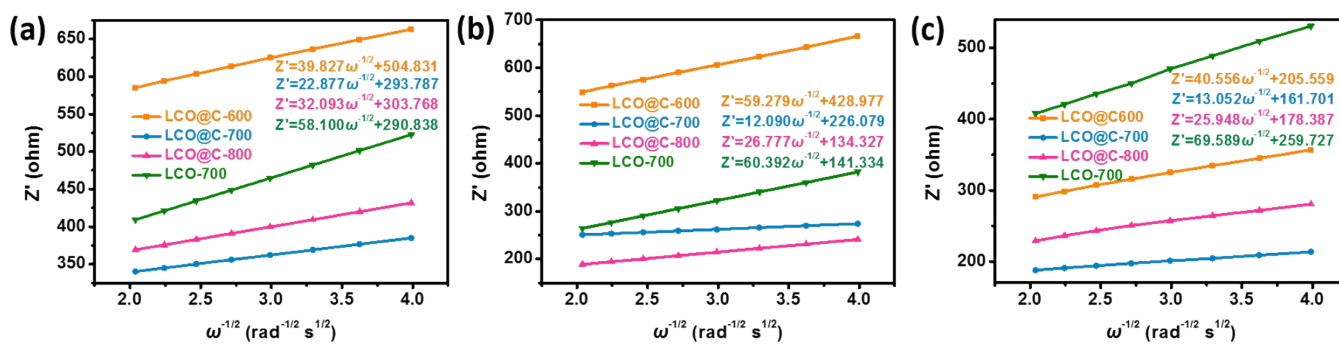


Fig. S6 The relationship of Z' and $\omega^{-1/2}$ of LCO@C-600, -700, -800, and LCO-700 after (a) 1st cycle, (b) 100th cycle, and (c) 200th cycle.

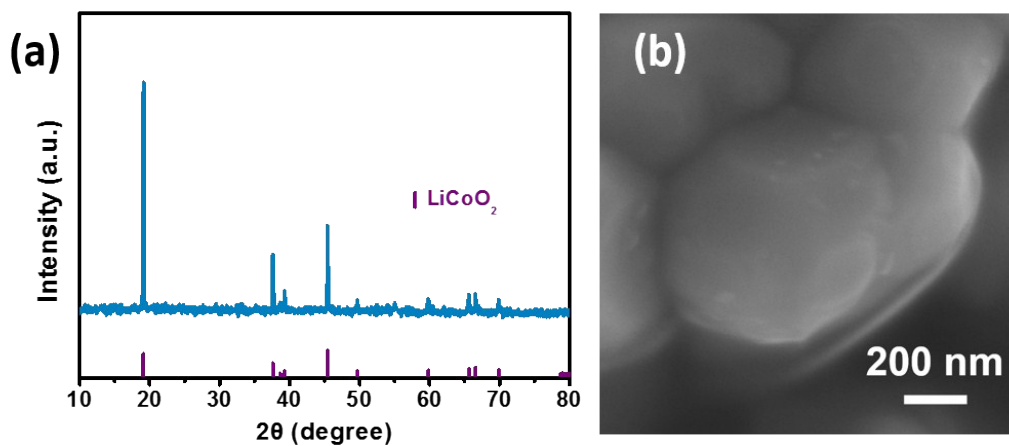


Fig. S7 (a) XRD pattern and (b) SEM image of the LCO@C-700 electrode material scraped down of the electrode films after 200 cycles, respectively.

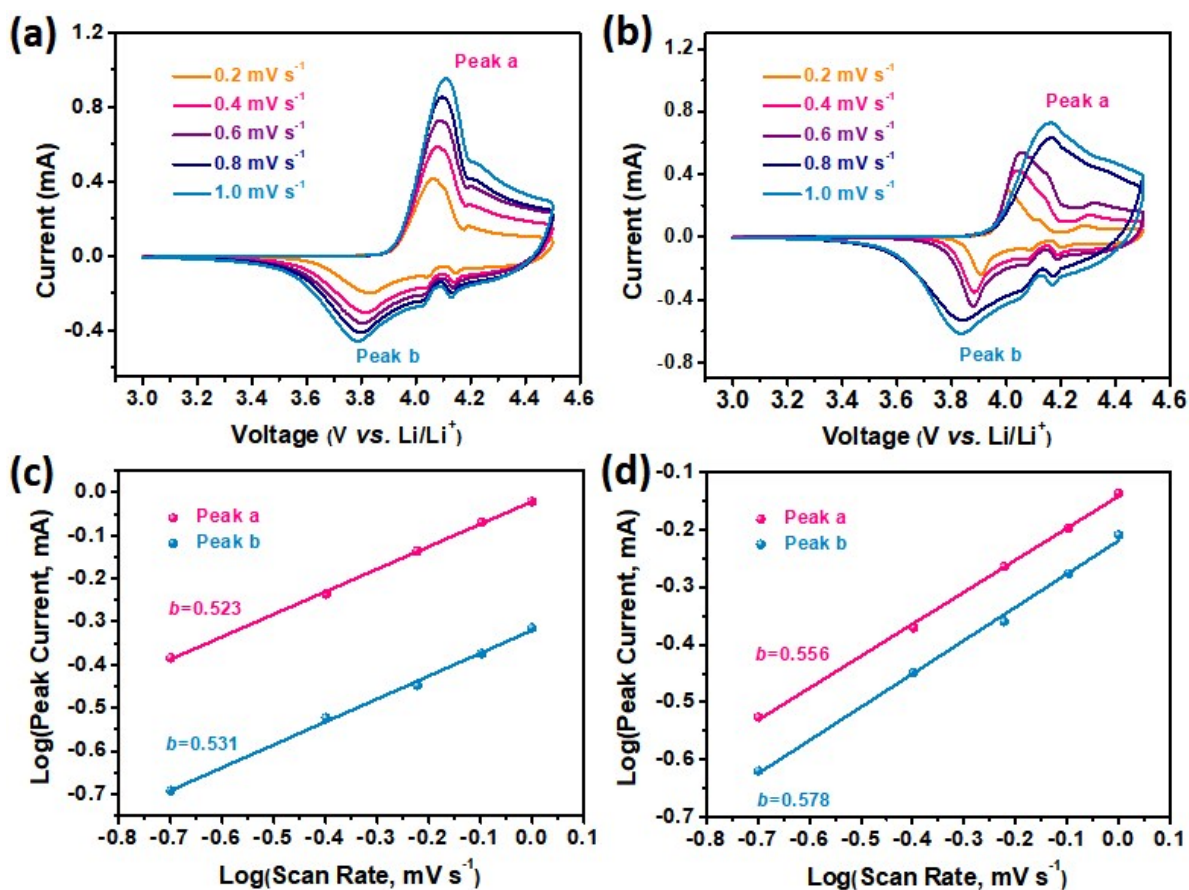


Fig. S8 CV curves and the liner relationships of the main cathodic and anodic peaks at the different scan rates from 0.2 to 1.0 mV s^{-1} for (a and c) LCO-700 and (b and d) LCO@C-600, respectively.

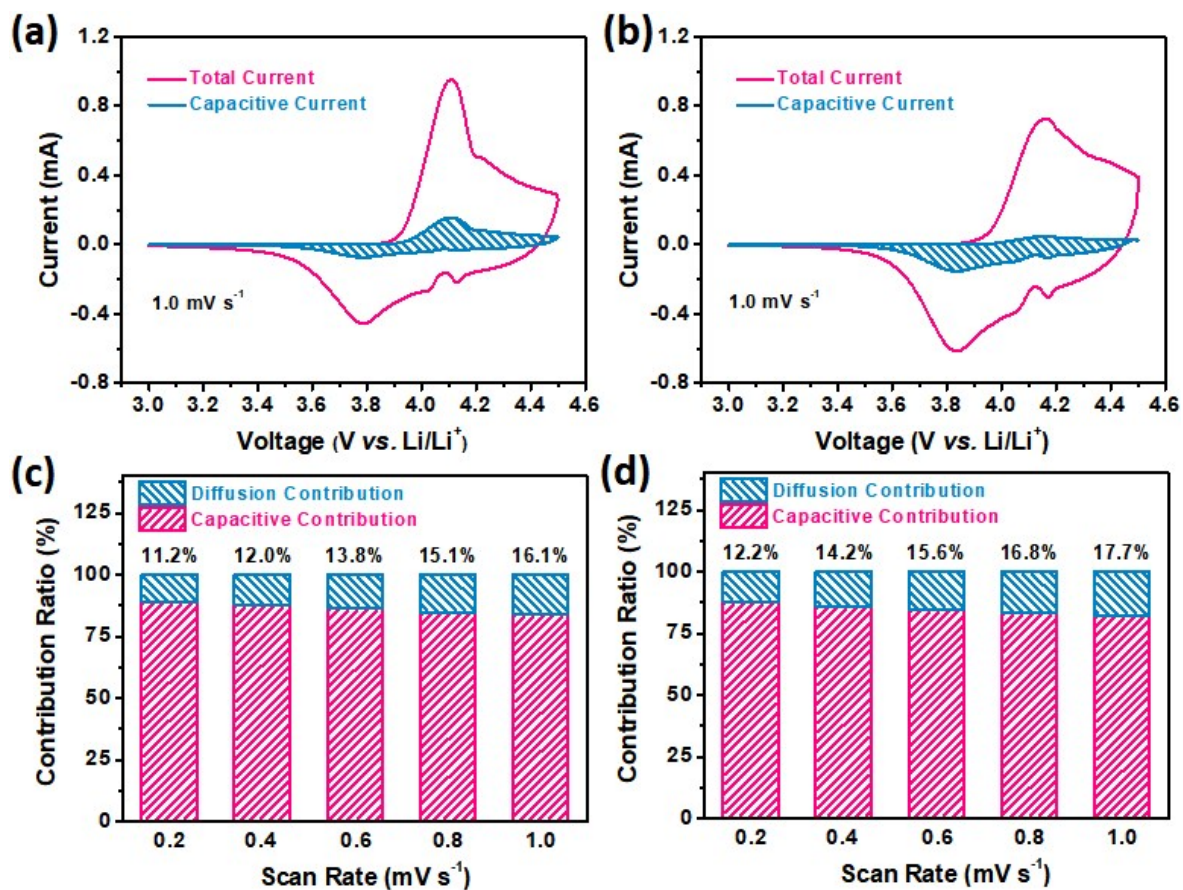


Fig. S9 The CV curve separation of the pseudocapacitive current and total current contribution at 1.0 mV s⁻¹, and contribution ratio of the diffusion- and pseudocapacitive-controlled capacities at different scan rates for (a and c) LCO-700 and (b and d) LCO@C-600, respectively.

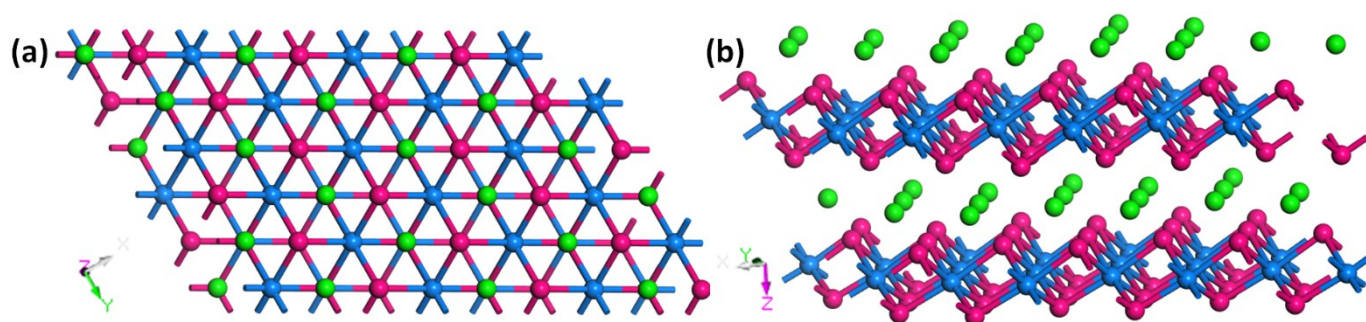


Fig. S10 Cluster structure of sole LiCoO₂ from the (a) top and (b) side view. The big green, pink, and blue spheres represent for Li, O, and Co atoms, respectively.

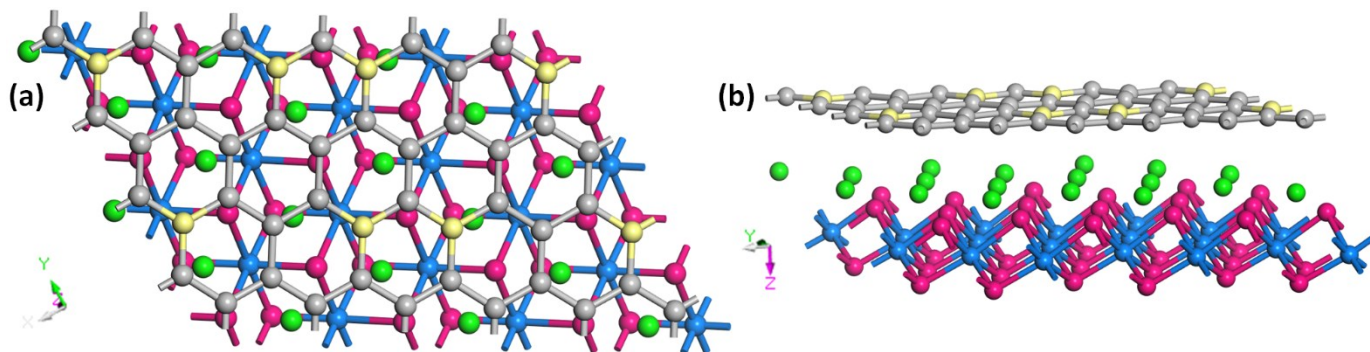


Fig. S11 Cluster structure model of the LCO@C heterostructure from the (a) top and (b) side view. The big green, grey, yellow, pink, and blue spheres represent for Li, C, N, O, Co and atoms, respectively.

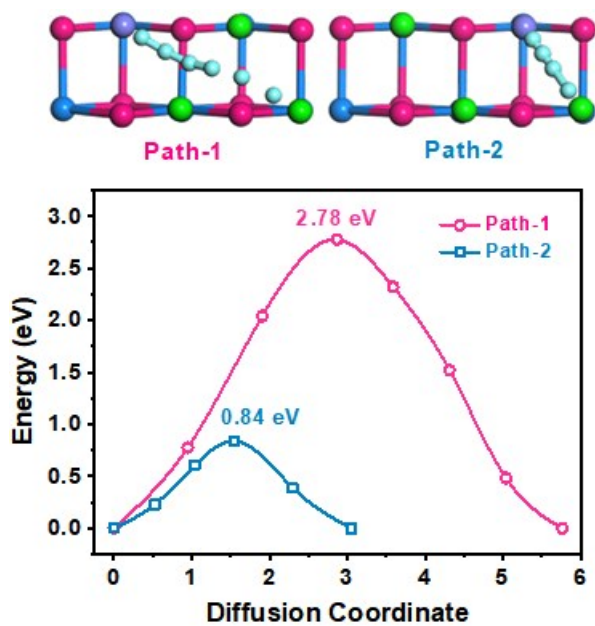


Fig. S12 The Li-vacancy migration paths and corresponding calculated diffusion energy barrier profiles of sole LiCoO_2 from DFT calculations.

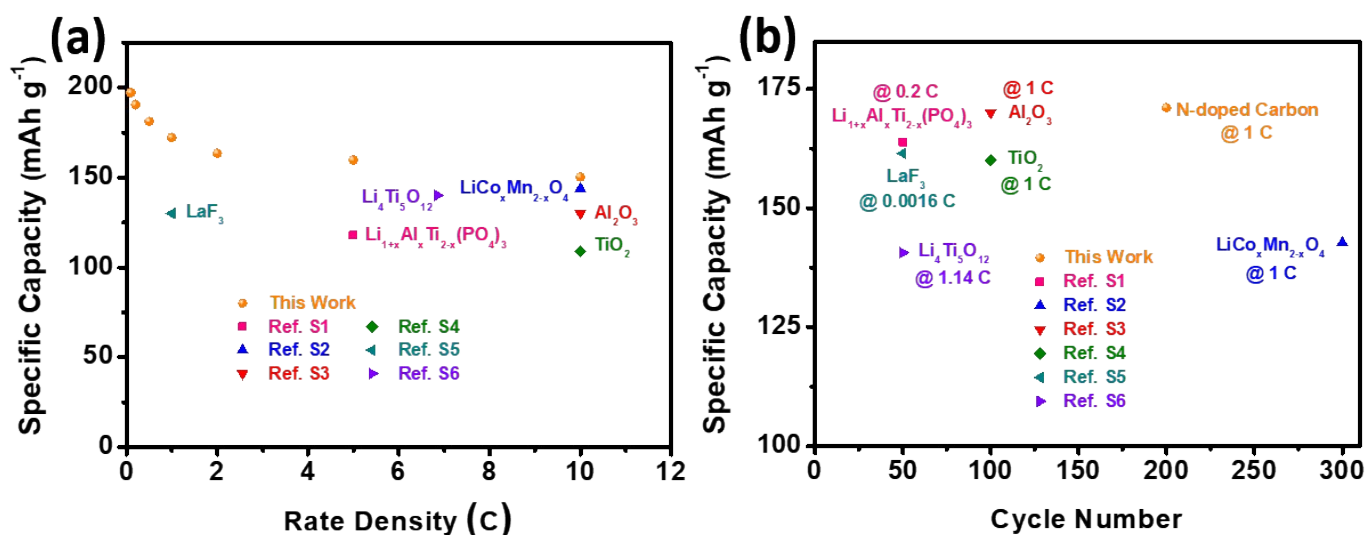


Fig. S13 Comparison of the rate capability and long-cycling performance of the surface-coating LCO at an elevated cut-off voltage of 4.5 V.

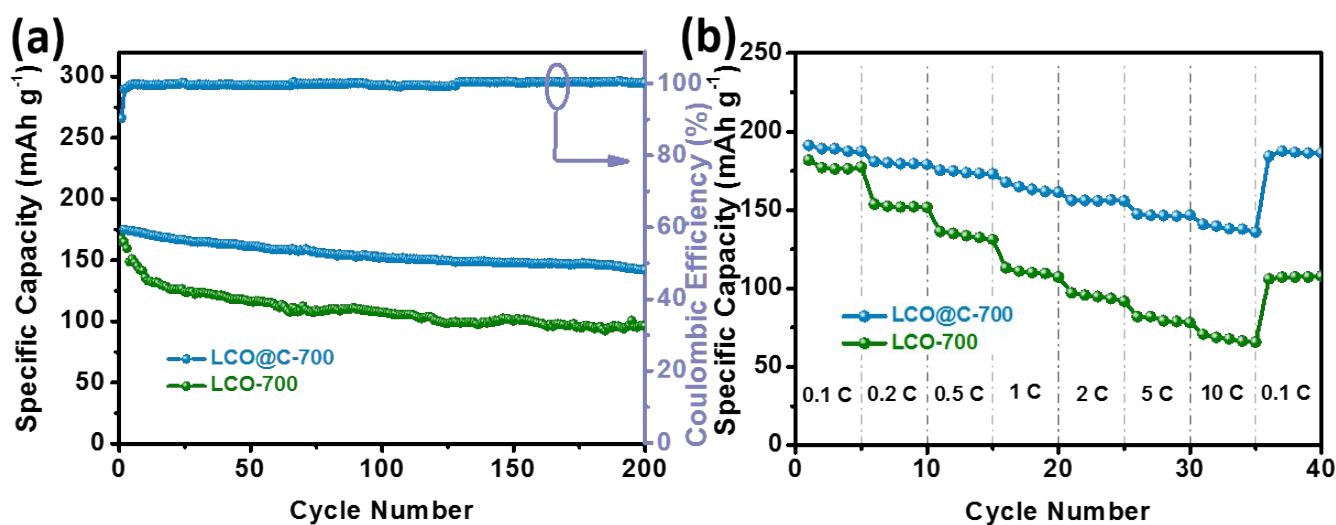


Fig. S14 (a) Cyclability at the current density of 2 C, (b) rate capability at different current rates ranging from 0.1 C to 10 C of the LCO@C-700 and LCO-700 cathode films which includes 92 wt% active material, 4 wt% PVDF, and 4 wt% Super P.

Table S1 Crystallographic data obtained from detailed Rietveld refinement result of the LCO@C-600, -700, and -800 samples.

	<i>a</i> -axis (Å)	<i>c</i> -axis (Å)	V (Å ³)	R _{wp} (%)	R _p (%)
LCO@C-600	2.8300	14.12519	97.73	4.47	3.38
LCO@C-700	2.8167	14.08632	96.85	2.35	1.36
LCO@C-800	2.8205	14.10774	97.08	3.25	2.21

Table S2 The fitted results of impedance spectra by the equivalent circuit as inset in Fig. 4g.

	Impedance (Ω)											
	LCO@C-600			LCO@C-700			LCO@C-800			LCO-700		
	R _e	R _{sf}	R _{ct}	R _e	R _{sf}	R _{ct}	R _e	R _{sf}	R _{ct}	R _e	R _{sf}	R _{ct}
1 st cycle	2.9	151.9	384.8	2.59	111.7	163.3	2.8	107.9	241.4	8.3	146.9	166.2
100 th cycle	5.1	146.2	228.2	2.8	87.2	106.1	3.3	174.9	127.2	9.0	109.1	393.6
200 th cycle	2.3	102.8	852.1	2.4	63.4	65.2	4.0	118.7	92.2	7.7	158.2	768.9

Table S3 Lithium ion diffusion coefficients at various stages after discharged cycling calculated by the EIS results.

	Lithium Ion Diffusion Coefficient (D_{Li} , cm ² s ⁻¹)			
	LCO@C-600	LCO@C-700	LCO@C-800	LCO-700
1 st cycle	6.54*10 ⁻⁹	1.98*10 ⁻⁸	1.01*10 ⁻⁸	3.07*10 ⁻⁹
100 th cycle	2.95*10 ⁻⁹	7.10*10 ⁻⁸	1.45*10 ⁻⁸	2.84*10 ⁻⁹
200 th cycle	6.31*10 ⁻⁹	6.09*10 ⁻⁸	1.54*10 ⁻⁸	2.14*10 ⁻⁹

References:

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S2 R. Gu, Z. Ma, T. Cheng, Y. Lyu, A. Nie and B. Guo, *ACS Appl. Mater. Interfaces*, 2018, **10**, 31271–31279.

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