

Supporting Information

Co Nanoparticles Combined with Nitrogen-Doped Graphitic Carbon Anchored in Carbon Fibers as a Self-standing Air Electrode for Flexible Zinc-Air Batteries

Yangshen Chen^a, Wenhui Zhang^a, Zeyu Zhu^a, Lulu Zhang^a, Jiayi Yang^a, Huanhuan Chen^a, Bing Zheng^a, Sheng Li^a, Weina Zhang^a, Jiansheng Wu^{*a}, Fengwei Huo^{*a}

a. Key Laboratory of Flexible Electronics (KLOFE) & Institute of Advanced Materials (IAM), Nanjing Tech University (NanjingTech), 30 South Puzhu Road, Nanjing 211816, China

Emails: iamjswu@njtech.edu.cn, iamfwhuo@njtech.edu.cn

The potentials (vs. Ag/AgCl) in this work were converted to RHE by using the Nernst equation¹:
$$(E_{\text{RHE}} = E_{\text{Ag/AgCl}} + 0.197 + 0.059 \text{ pH})$$

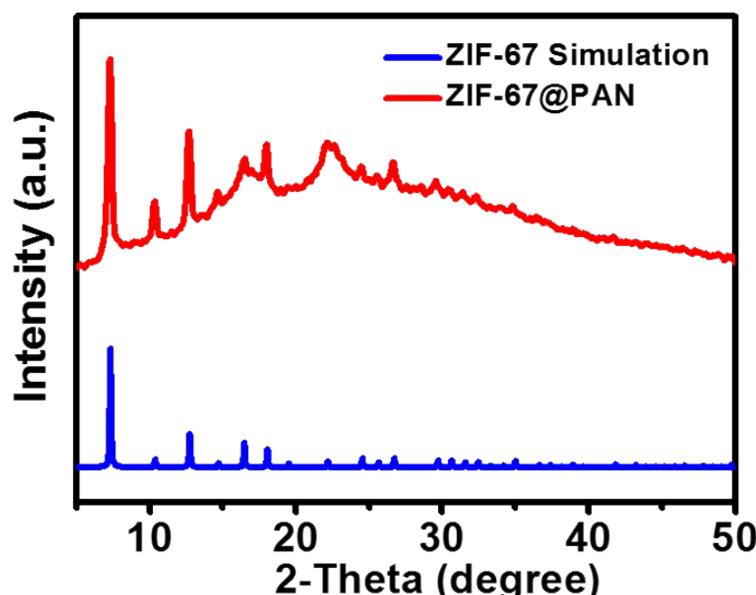


Fig. S1. The XRD pattern of ZIF-67@PAN and ZIF-67 simulation.

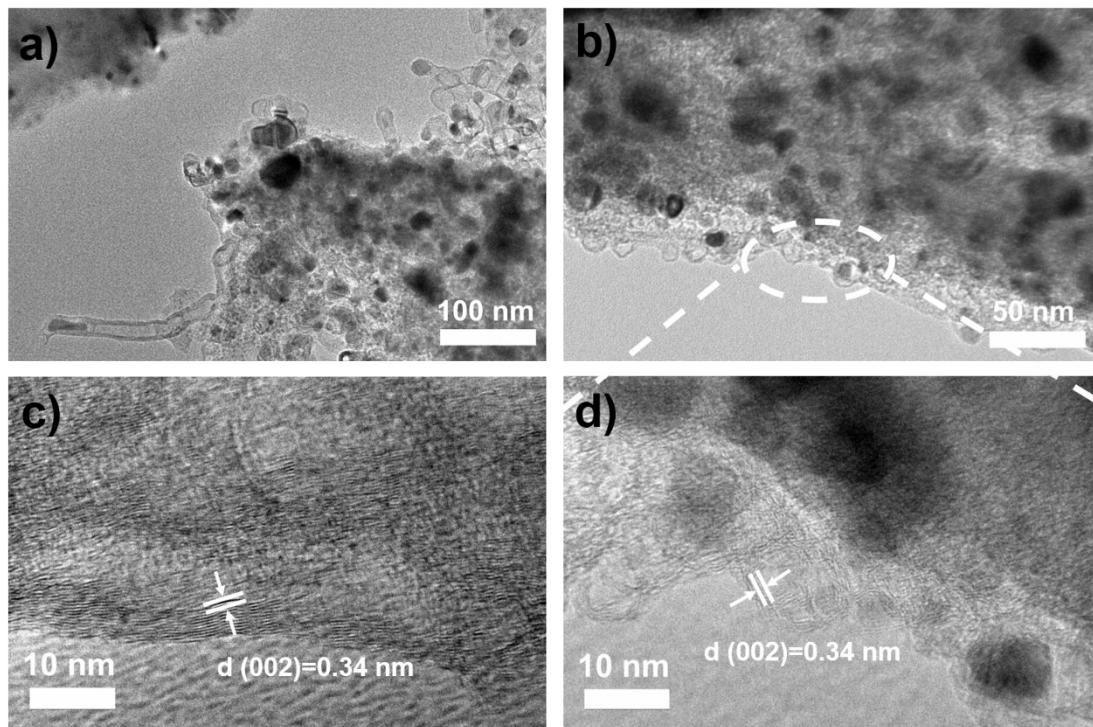


Fig. S2. (a-b) TEM images and (c-d) HRTEM images of Co@NPCFs

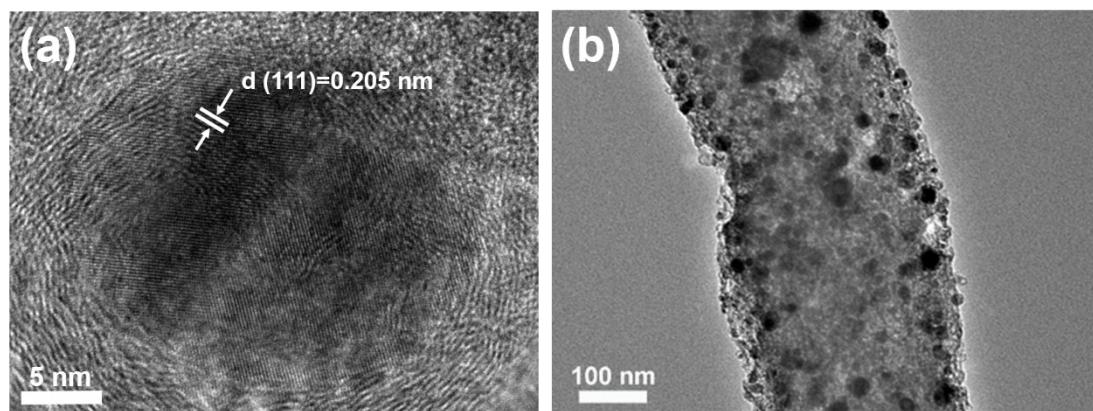


Fig. S3. (a) HRTEM images of Co@NCFs and (b) TEM image of Co@NCFs.

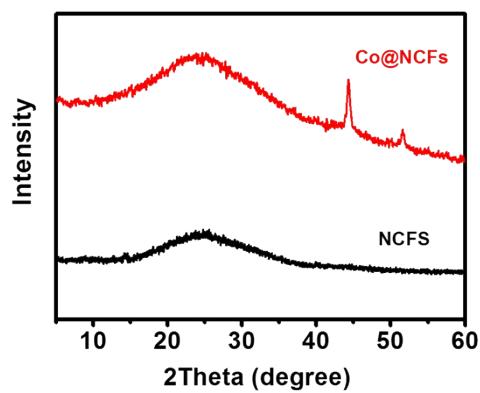


Fig. S4. XRD patterns of NCFs and Co@NCFs.

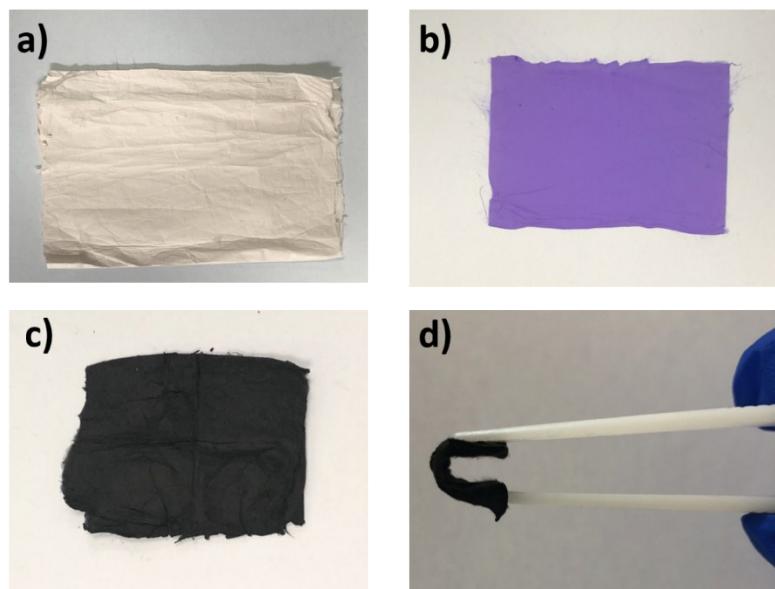


Fig. S5. The photographs of (a) $\text{Co}(\text{AC})_2/\text{PAN}$ (b) ZIF-67@PAN (c-d) Co@NPCFs.

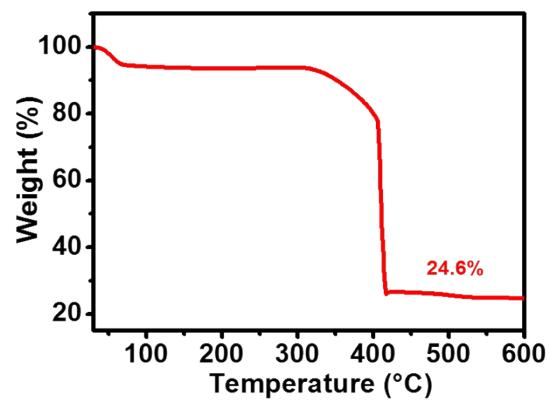


Fig. S6. TGA data of the Co@NPCFs.

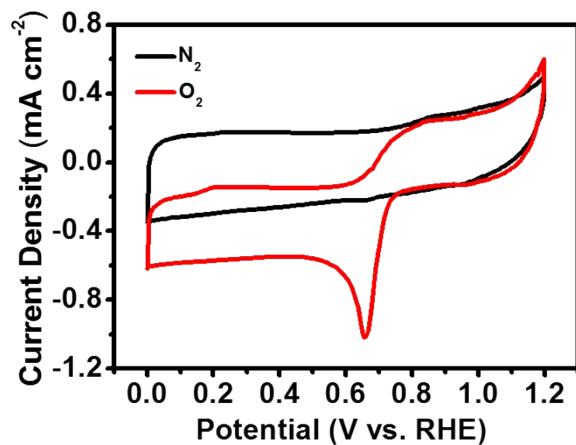


Fig. S7. CV curves of Co@NPCFs in O_2 - and N_2 - saturated 0.1 M KOH solution.

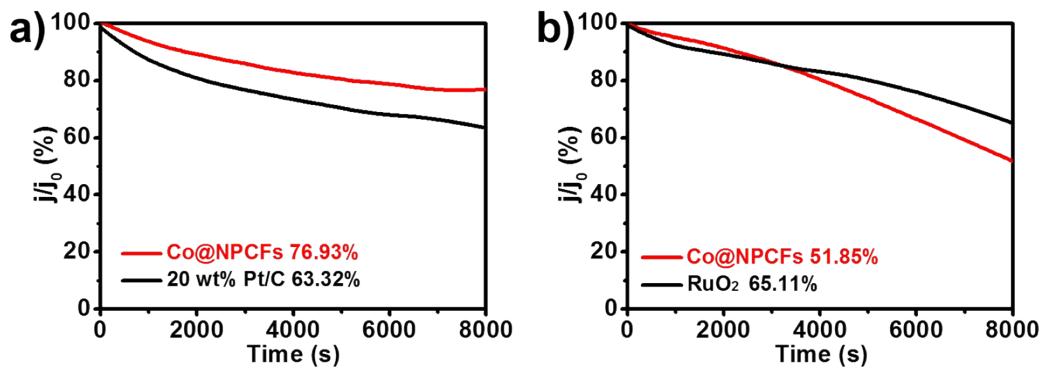


Fig. S8. (a) ORR and (b) OER chronoamperometric responses of Co@NPCFs at a constant potential of 0.7 V and 1.6 V, respectively.

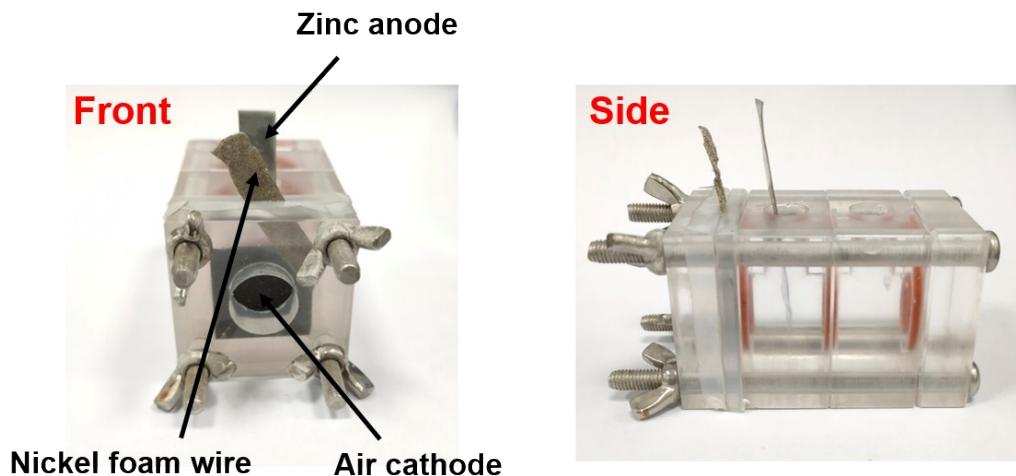


Fig. S9. The photographs of liquid-state ZABs mould.

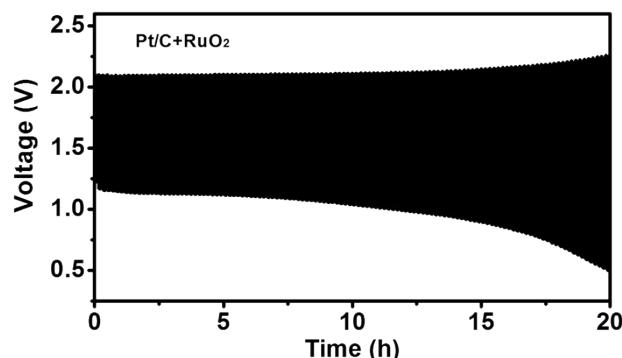


Fig. S10. Galvanostatic cycling stability of the liquid-state ZAB based upon Pt/C+RuO₂ at 5 mA cm⁻².

Table 1

Catalyst	ΔE (V)	Power density (mW cm ⁻²)	Discharge stability (liquid state)	Discharge stability (solid state)	Ref
----------	----------------	--------------------------------------	------------------------------------	-----------------------------------	-----

Co@NPCFs	0.97	91.87	80 h at 5 mA cm⁻²	5 h at 3 mA cm⁻²	Present work
NCNF-1000	1.02	185	83 h at 10 mA cm⁻²	6 h at 2 mA cm⁻²	[2]
Co₃FeS_{1.5}(OH)₆	0.87	113.1	36 h at 2 mA cm⁻²	-	[3]
Co₃O₄/N-rGO nanosheets	0.93	-	-	25 h at 3 mA cm⁻²	[4]
DN-CP@G	0.99	135	250 cycles at 5 mA cm⁻²	180 cycles at 1 mA cm⁻²	[5]
Ni₃Fe/N-C sheets	0.84	-	420 h at 10 mA cm⁻²	-	[6]
PCN-CFP	0.96	-	50 h at 20 mA cm⁻²	-	[7]
Fe-N_x-C	0.92	96.4	300 h at 5 mA cm⁻²	120 h at 1 mA cm⁻²	[8]

Reference

1. B. Chen, X. He, F. Yin, H. Wang, D.-J. Liu, R. Shi, J. Chen and H. Yin, *Adv. Funct. Mater.*, 2017, **27**, 1700795.
2. Q. Liu, Y. Wang, L. Dai and J. Yao, *Adv Mater*, 2016, **28**, 3000-3006.
3. H. F. Wang, C. Tang, B. Wang, B. Q. Li and Q. Zhang, *Adv Mater*, 2017, **29**, 1702327.
4. Y. Li, C. Zhong, J. Liu, X. Zeng, S. Qu, X. Han, Y. Deng, W. Hu and J. Lu, *Adv. Mater.*, 2018, **30**, 1703657.
5. C. Hang, J. Zhang, J. Zhu, W. Li, Z. Kou and Y. Huang, *Adv. Energy Mater.*, 2018, **8**, 1703539.
6. G. Fu, Z. Cui, Y. Chen, Y. Li, Y. Tang and J. B. Goodenough, *Adv. Energy Mater.*, 2017, **7**, 1601172.
7. T. Y. Ma, J. Ran, S. Dai, M. Jaroniec and S. Z. Qiao, *Angew. Chem. Int. Ed.*, 2015, **54**, 4646-4650.
8. J. Han, X. Meng, L. Lu, J. Bian, Z. Li and C. Sun, *Adv. Funct. Mater.*, 2019, 1808872.