## **Supporting Information**

## Reversible Gas Capture by Ferroelectric Switch and 2D Molecule Multiferroics in In<sub>2</sub>Se<sub>3</sub> Monolayer

Xiao Tang, Jing Shang, Yuantong Gu, Aijun Du, and Liangzhi Kou\*

School of Mechanical, Medical and Process Engineering, Queensland University of Technology, Brisbane, QLD 4001, Australia

\* Corresponding author E-mail: liangzhi.kou@qut.edu.au



Fig. S1 Adsorption energy of NH<sub>3</sub> adsorbed on In<sub>2</sub>Se<sub>3</sub> surfaces under different electric field.



Fig. S2 The most stable configurations of H<sub>2</sub>, O<sub>2</sub>, N<sub>2</sub> onto P  $\downarrow$  and P  $\uparrow$  sides of In<sub>2</sub>Se<sub>3</sub> monolayer. Corresponding adsorption distance and adsorption energy were also shown.



Fig. S3 Electrostatic potential of (a) pure  $In_2Se_3$ , (b)  $NH_3$ , (c) NO and (d)  $NO_2$ . Green, orange, white, blue and red balls present the Se, In, H, N and O atoms, respectively. The arrow represents the direction of electric polarization.



Fig. S4 Energy-level with respect to the vacuum level for  $In_2Se_3$  (P  $\downarrow$ ),  $In_2Se_3$  (P  $\uparrow$ ), CO<sub>2</sub>, CO and SO<sub>2</sub>.



Fig. S5 Charge density difference and adsorption energies for NH<sub>3</sub> (a-d), NO (e-h) and NO<sub>2</sub> (i-l) on In<sub>2</sub>Se<sub>3</sub> (P $\uparrow\uparrow$ ), In<sub>2</sub>Se<sub>3</sub> (P $\uparrow\downarrow$ ), In<sub>2</sub>Se<sub>3</sub> (P $\uparrow\downarrow$ ) and In<sub>2</sub>Se<sub>3</sub> (P $\downarrow\uparrow\uparrow$ ) surfaces, respectively. The yellow (cyan) region represents charge accumulation (depletion) and the isosurface value is 0.0004 e Å<sup>-3</sup>. The purple arrow represents the direction of polarization. Green, orange, white, blue and red balls present the Se, In, H, N and O atoms, respectively. E<sub>ads</sub> =E<sub>total</sub>-E<sub>gas</sub>-E<sub>In2Se3-bilayer</sub>, where E<sub>total</sub>, E<sub>gas</sub> and E<sub>In2Se3-bilayer</sub> are energies of bilayer In<sub>2</sub>Se<sub>3</sub> with adsorbed gas molecule, isolated gas molecule and pure In<sub>2</sub>Se<sub>3</sub> bilayer.



Fig. S6 Calculated projected density of states (PDOS) of NH<sub>3</sub> adsorption on (a)  $In_2Se_3$  (P $\uparrow$   $\uparrow$ ) (b)  $In_2Se_3$  (P $\uparrow$  $\downarrow$ ) (c)  $In_2Se_3$  (P $\downarrow$  $\downarrow$ ) and (d)  $In_2Se_3$  (P $\downarrow$  $\uparrow$ ) bilayer. Black, magenta and blue lines denote PDOS of NH<sub>3</sub>, top layer  $In_2Se_3$  and bottom layer  $In_2Se_3$ , respectively. The Fermi level is set to zero.



Fig. S7 Calculated projected density of states (PDOS) of NO adsorption on (a)  $In_2Se_3$  (P $\uparrow\uparrow$ ) (b)  $In_2Se_3$  (P $\uparrow\downarrow$ ) (c)  $In_2Se_3$  (P $\downarrow\downarrow$ ) and (d)  $In_2Se_3$  (P $\downarrow\uparrow$ ) bilayer. Black, magenta and blue lines denote PDOS of NO, top layer  $In_2Se_3$  and bottom layer  $In_2Se_3$ , respectively. The Fermi level is set to zero.



Fig. S8 Calculated projected density of states (PDOS) of NO<sub>2</sub> adsorption on (a)  $In_2Se_3$  (P $\uparrow\uparrow$ ) (b)  $In_2Se_3$  (P $\uparrow\downarrow$ ) (c)  $In_2Se_3$  (P $\downarrow\downarrow$ ) and (d)  $In_2Se_3$  (P $\downarrow\uparrow$ ) bilayer. Black, magenta and blue lines denote PDOS of NO<sub>2</sub>, top layer  $In_2Se_3$  and bottom layer  $In_2Se_3$ , respectively. The Fermi level is set to zero.



Fig. S9 Upper panel: PDOS for the NO<sub>2</sub> adsorbed  $In_2Se_3$  (P  $\downarrow$ ). Lower panel: PDOS for the NO<sub>2</sub> adsorbed  $In_2Se_3$  (P  $\uparrow$ ).

Table S1.	The calculated	magnetic	moment of NO	and NO <sub>2</sub>	adsorbed	In <sub>2</sub> Se <sub>3</sub> .
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	Magnetic		Magnetic	
System	Moment	System	Moment	
	$(\mu_B)$		$(\mu_B)$	
NO+In <sub>2</sub> Se <sub>3</sub>	0.88	NO <sub>2</sub> +In <sub>2</sub> Se <sub>3</sub>	1.00	
$(\downarrow)$	0.000	$(\downarrow)$		
NO+In <sub>2</sub> Se <sub>3</sub>	1.00	NO <sub>2</sub> +In <sub>2</sub> Se <sub>3</sub>	0.89	
(†)	1.00	(†)		
NO+In <sub>2</sub> Se <sub>2</sub>		NO2+In2Se2		
( <sup>†</sup> <sup>†</sup> )	1.07	( <u></u> <u></u> <u></u> <u></u> <u></u> <u></u> ( <u></u> <u></u> <u></u> <u></u> <u></u> <u></u> <u></u> <u></u> <u></u> )	0.79	
NO+In <sub>2</sub> Se <sub>3</sub>	1.00	NO <sub>2</sub> +In <sub>2</sub> Se <sub>3</sub>	0.86	
$(\uparrow\downarrow)$		(↑↓)		
NO+In <sub>2</sub> Se <sub>3</sub>	0.72	NO <sub>2</sub> +In <sub>2</sub> Se <sub>3</sub>	1.00	
$(\downarrow\uparrow)$	0.73	$(\downarrow\uparrow)$		
NO+In <sub>2</sub> Se <sub>2</sub>		NO2+In2Se2	0.98	
(11)	0.70	(11)		
(++)		(++)		



Fig. S10 (a-b) Geometric structures and energy of different magnetic coupling (ferromagnetic, antiferromagnetic and nonmagnetic coupling) between two NO<sub>2</sub> molecules on  $In_2Se_3$  (P $\downarrow$ ) surface and (c-d)  $In_2Se_3$  (P $\uparrow$ ) surface. The purple arrows indicate the directions of electric polarization. Green, orange, blue and red balls present the Se, In, N, and O atoms, respectively.



Fig. S11 (a-b) Geometric structures and energy of different magnetic coupling (ferromagnetic, ferrimagnetic and nonmagnetic coupling) between three NO molecules on  $In_2Se_3(P\downarrow)$  surface and  $In_2Se_3(P\uparrow)$  surface. The purple arrows indicate the directions of electric polarization. Green, orange, blue and red balls present the Se, In, N, and O atoms, respectively.