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Supporting Information

A Novel Post-processed Surface Modified Double-network Polymer Layer for

Triboelectric Nanogenerator

Yaqian Liu, ^{*a,b*} Xiumei Wang, ^{*a,b*} Yujie Yan, ^{*a,b*} Zhichao Rao, ^{*a,b*} Huipeng Chen, ^{*a,b*} * Tailiang Guo ^{*a,b*}

^aInstitute of Optoelectronic Display, National & Local United Engineering Lab of Flat Panel Display Technology, Fuzhou University, Fuzhou 350002, China ^bFujian Science & Technology Innovation Laboratory for Optoelectronic Information of China, Fuzhou 350100, China Email: hpchen@fzu.edu.cn

* Corresponding authors.

Email: hpchen@fzu.edu.cn(Huipeng Chen)



Fig. S1 Schematic illustration for DN-PDMS film with different monomer solvents.



Fig. S2 Photograph with three prepared monomer solvents.



Fig. S3 Working mechanism of original state TENG.



Fig. S4 Summarized CPD results presented that the DN-PDMS was apt to attract an even greater amount of charge than P-PDMS.



Fig. S5 Schematic illustration of fabrication micro-patterned DN-PDMS layers.



Fig. S6 Surface profilometry depth profile for a section of a micro-patterned DN-PDMA with (a) lines and (b) cubes.



Fig. S7 V_{oc} of cube DN-PDMS TENG with the size of 2 cm \times 2 cm.



Fig. S8 (a) Voc and (b) Jsc of TENG with different patterned EGDMA DN-PDMS film. (c) Summarized

performances enhancement by monomer treatment and micro-structures.



Fig. S9 (a) Voc and (b) Jsc of TENG with different patterned HEMA DN-PDMS films. (c) Summarized

performances enhancement by monomer treatment and micro-structures.



Fig. S10 Stability test for the TENG with MAA DN-PDMS.



Fig. S11 (a) Sensitivity of the TENG. (b) Hysteresis curve of the TENG.



Fig. S12 The measurement of the detection limit of the TENG through the short-circuit current density (J_{sc}) . The left panel in the plot shows the output J_{sc} with a gentle pressure applied on the TENG, while the right panel shows the measurement of the noise level magnitude. The inset is the enlarged view of the J_{sc} for the noise level with no pressure on. With this measurement the detection limit of the TENG could be deducted with the following equation.¹

$$P_0 = \frac{P}{J_{sc}/J_{sc,noise}} = \frac{0.392}{3.2/0.031} kPa = 3.8 Pa$$

Table S1. The (equilibrium)	surface	concentration	of	monomer	(C _s),	and	diffusivity	of	the
monomer (D_{12}) for several monomer	s in the l	P-PDMS film.							

	EGDMA	НЕМА	MAA
C _s (no units)	0.241	0.256	0.279
D ₁₂ (µm/cm ²)	0.033	0.136	0.295

Materials	Initial voltage (V)	Optimized voltage (V)	Size	Reference
PDMS/ITO	4	18	$3 \text{ cm} \times 3 \text{ cm}$	2
PDMS/Cu	5.34	22.04	N/A	3
PDMS/PTFE/AgNWs	1.5	N/A	$1 \text{ cm} \times 1 \text{ cm}$	4
PTFE/A1	1.2	6.2	$1 \text{ cm} \times 1.5 \text{ cm}$	5
PDMS/Ag	N/A	5	$1 \text{ cm} \times 1 \text{ cm}$	6
PDMS/AgNWs	2.5	18	$2 \text{ cm} \times 2 \text{ cm}$	7
PDMS/ITO	10	N/A	1.5 cm × 1.5 cm	8
PDMS/AgNWs	N/A	75	$2 \text{ cm} \times 2 \text{ cm}$	9
PDMS/A1	32	42.5	$2 \text{ cm} \times 2 \text{ cm}$	10
PDMS/PAMPS ionogel	N/A	2.8	$2 \text{ cm} \times 1.5 \text{ cm}$	11
PDMS/ITO	5	84	1 cm × 1 cm	This work
PDMS/ITO	17.4	301	$2 \text{ cm} \times 2 \text{ cm}$	This work

Table S2. The comparison of the *Voc* with reported work.

Soak time (min)	EGDMA (µm)	HEMA (μm)	MAA (µm)
10	4.5 ± 0.15	6.5 ± 0.1	10.2 ± 0.21

Table S3. Measured channel depths for PDMS micro-patterned substrates with different monomers.

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